

Facile synthesis of mesoporous polymeric carbon nitride nanosheets anchored by Pt with ultralow loading for high-efficiency photocatalytic H₂ evolution

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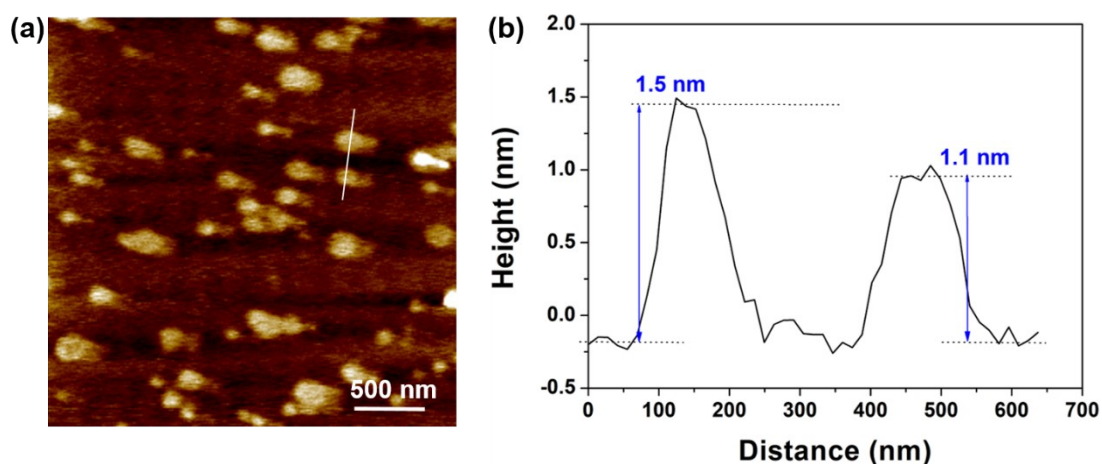


Fig. S1 (a) AFM image and (b) corresponding height profiles along the white line in figure (a) of CNNS.

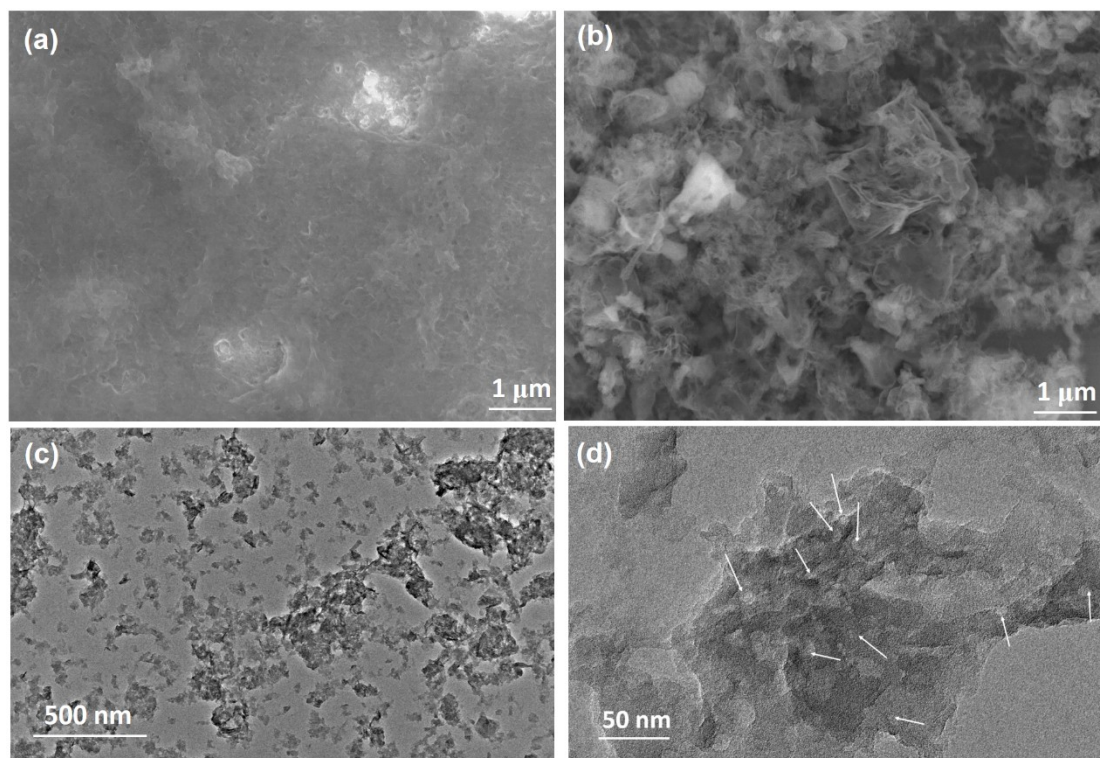


Fig. S2 SEM images of (a) bulk CNU, (b) CNNSu, and TEM images of CNNSu (c, d)

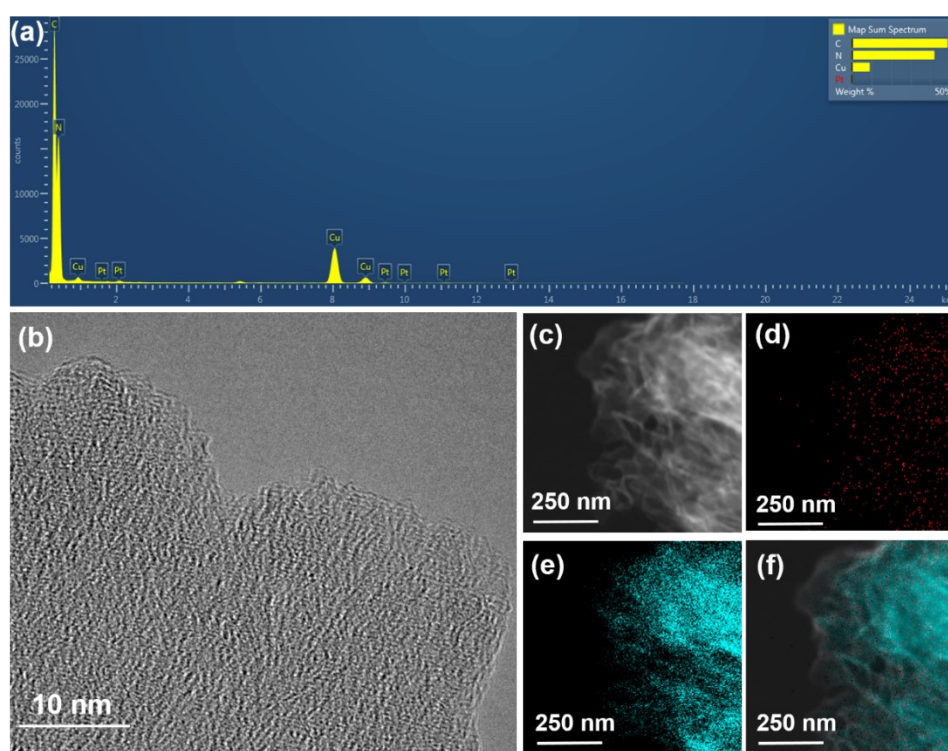


Fig. S3 (a) Energy dispersive spectrum, (b) high-resolution TEM and (c) TEM image of Pt/CNNS-5H ($\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}$ solution: 1mL) and its (d) Pt, (e) N and (f) Pt, N merged element mappings.

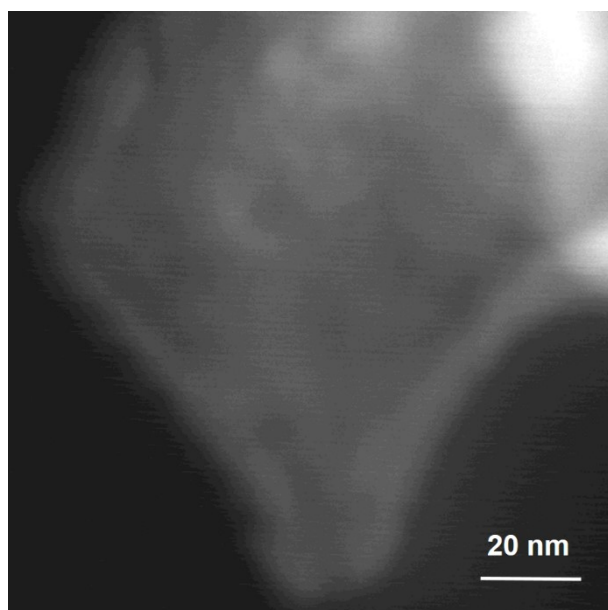


Fig. S4 STEM image of Pt/CNNS-5H ($\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}$ solution: 1mL).

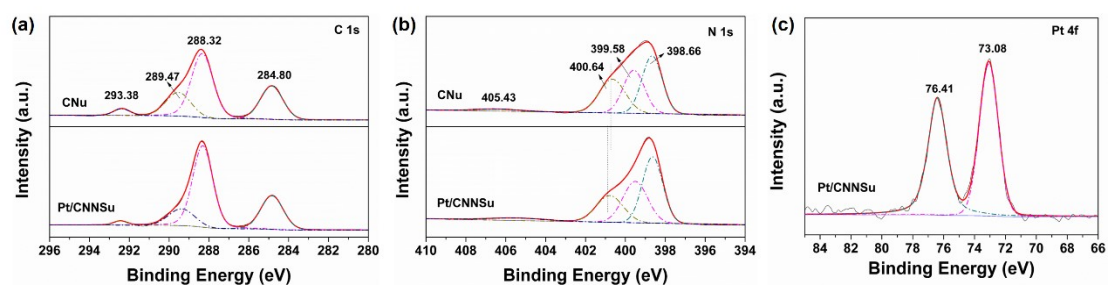


Fig. S5 High-resolution XPS of (a) C 1s, (b) N 1s, and (c) Pt 4f in CNU and CNNSu as-obtained samples.

Table S1 Physicochemical properties and photocatalytic activities of the CN and CNNS samples for H₂ evolution reduction under visible light.

Sample	Specified surface area (m ² /g)	Binding energy (eV)	H ₂ resolution (μmol/h)
CN	9.96	2.76	16.1
CNNS-2.5H	61.63	2.77	118.68
CNNS-5H	91.77	2.83	172.79
CNNS-10	80.60	2.86	150.28
CNNS-15	64.02	2.88	97.24

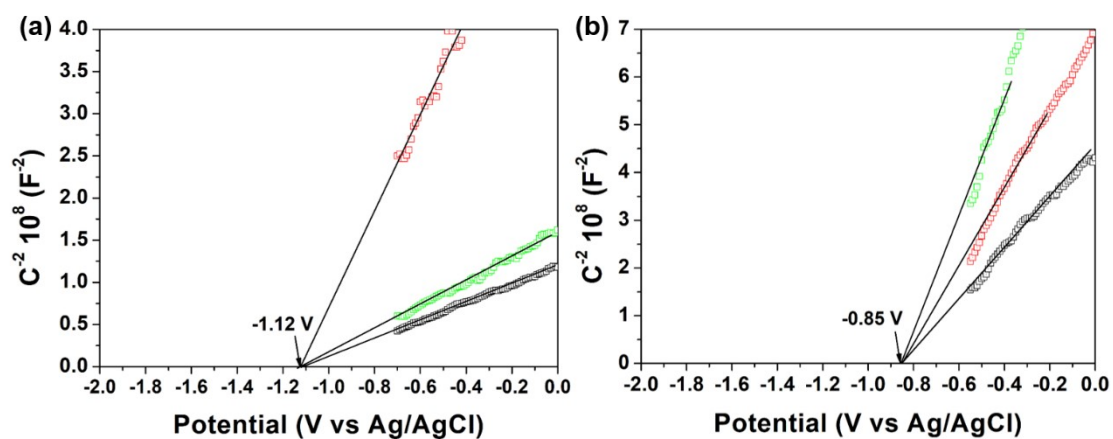


Fig. S6 Mott-Schottky plots of CN (a) and CNNS-5H (b) collected at various frequencies¹ versus saturated Ag/AgCl reference electrode

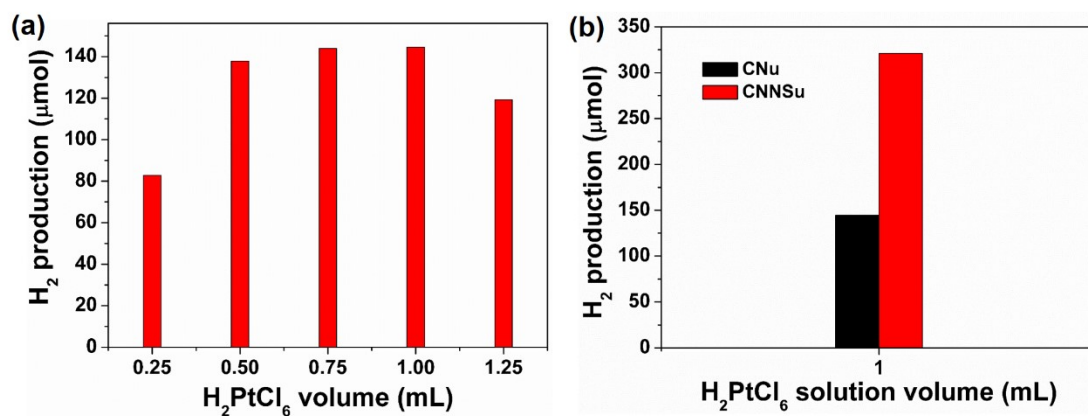


Fig. S7 (a) H₂ evolution rate under visible light irradiation over CNU with varied H₂PtCl₆·6H₂O solution (2 mg/mL in H₂O) addition, (b) H₂ yield rate under visible light irradiation over CNU and CNNSu with Pt as cocatalyst (Pt content, 0.28 wt%).

Table S2 Comparison of hydrogen evolution rate on the basis of co-catalyst Pt under visible light irradiation.

Catalyst	Light Source	Cocatalyst (wt%)	H ₂ evolution rate	H ₂ evolution rate	References
			($\mu\text{mol/g/h}$)	(mmol/g/h, per Pt)	
CNNS	300 W Xe lamp	Pt, 0.23	3455.8 ($\lambda > 400$ nm)	1502.5 ($\lambda > 400$ nm)	This work
			1888.4 ($\lambda > 420$ nm)	821.0 ($\lambda > 420$ nm)	
CNNSu	300 W Xe lamp	Pt, 0.28	6412 ($\lambda > 400$ nm)	2290 ($\lambda > 400$ nm)	This work
			3379 ($\lambda > 420$ nm)	1206.8 ($\lambda > 420$ nm)	
Atomically thin CNNS	300 W Xe lamp ($\lambda > 420$ nm)	Pt, 3	3809.6	126.99	[2]
g-C ₃ N ₄	AM 1.5	Pt (II), 0.24	138	57.5	[3]
g-C ₃ N ₄	300 W Xe lamp	Pt single atom, 0.91	11472	1260.6	[4]
N-vacancy g-C ₃ N ₄	300 W Xe lamp ($\lambda > 420$ nm)	Pt single atom, 1.72	3020	175.58	[5]
MOF	300 W Xe lamp ($\lambda > 420$ nm)	Pt single atom, 12	11320	94.3	[6]
Defective ultra-thin g-C ₃ N ₄	300 W Xe lamp	Pt, 1	4300	430	[7]
Onion-ring g-C ₃ N ₄	300 W Xe lamp ($\lambda > 420$ nm)	Pt, 3	1900	63.3	[8]

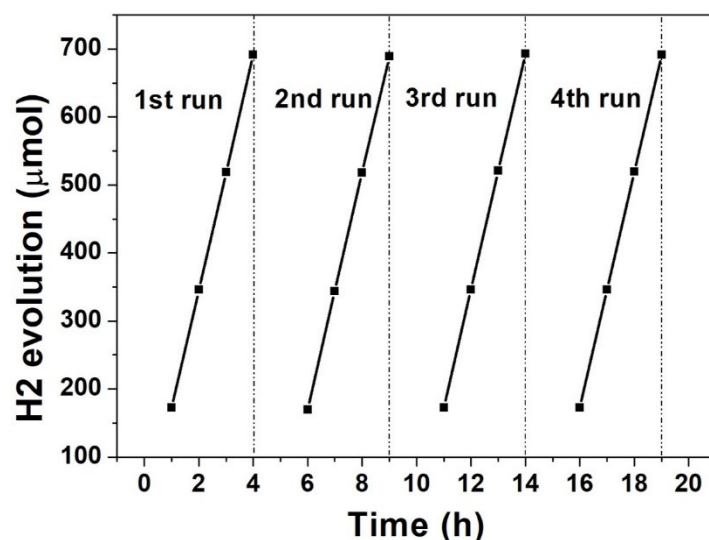


Fig. S8 Cycling runs for photostability test of photocatalytic H₂ evolution on CNNS-5H.

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