

## **Organic solvent-assisted co-precipitation synthesis of a red-emitting K<sub>2</sub>TiF<sub>6</sub>:Mn phosphors of improved quantum efficiency and optimized morphology**

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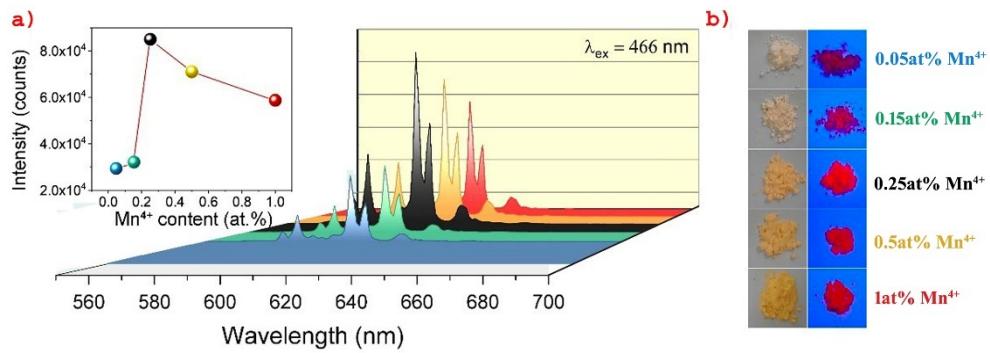
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**Table S1.** Quantities of chemicals added to the reaction mixture to prepare  $K_2TiF_6$ :xat.%  $Mn^{4+}$  ( $x = 0.05, 0.15, 0.25, 0.5$  and  $1$ ).

	<b>Mn<sup>4+</sup> content (at. %)</b>				
	<b>0.05</b>	<b>0.15</b>	<b>0.25</b>	<b>0.5</b>	<b>1.0</b>
$K_2MnF_6$	0.0014 g	0.0042 g	0.0069 g	0.0139 g	0.0278 g
$TiO_2$			0.1 g		
48% HF			3.0 ml		
KF			0.1452 g		
Ethanol /					
1-propanol /			1 ml		
n-butanol					



**Fig. S1** Room temperature emission spectra of  $\text{K}_2\text{TiF}_6:\text{xat.\% Mn}^{4+}(x = 0.05, 0.15, 0.25, 0.5 \text{ and } 1)$  phosphors and concentration-dependent emission intensity (b). Appearance of phosphors under daylight and 365 nm UV-light excitation (c).