

Supporting Information For:

Self-assembly of non-macrocyclic triangular Ni₃Ln clusters

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Table S1. Crystal structure and refinement details for complexes **C1 – C4**. CCDC deposition numbers 2116396 – 2116399.

Complex	C1	C2	C3	C4
Deposition number	2116396	2116398	2116399	2116397
Molecular formula	C ₃₈ H ₄₆ LaN ₃ Ni ₃ O ₃₃	C _{37.5} H ₃₆ EuN ₃ Ni ₃ O _{28.5}	C ₄₅ H ₅₄ GdN ₃ Ni ₃ O ₃₀	BaC ₄₃ H ₅₂ N ₂ Ni ₃ O ₂₈
Formula weight	1387.82	1312.78	1450.29	1358.33
Temperature /K	100	100	100	100
Crystal system	monoclinic	monoclinic	monoclinic	triclinic
Space group	<i>P2₁/n</i>	<i>C2/c</i>	<i>P2₁/n</i>	<i>P1</i>
a /Å	10.4295(8)	12.7986(12)	17.0721(10)	11.0862(6)
b /Å	19.2819(15)	16.5467(16)	18.7861(11)	15.5290(9)
c /Å	24.0313(18)	39.9720(4)	17.1302(10)	16.8872(9)
α /°	90	90	90	115.311(4)
β /°	102.277(3)	90.203(3)	106.716(3)	101.491(4)
γ /°	90	90	90	97.235(4)
Volume /Å³	4722.2(6)	8464.9(14)	5261.8(5)	2501.3(3)
Z/Z'	4 / 1	8 / 1	4 / 1	2 / 1
ρ_{calc} /g cm⁻³	1.952	2.06	1.831	1.804
μ /mm⁻¹	9.197	12.934	10.111	8.159
F(000)	2800	5256	2932	1376
Crystal size /mm	0.24 × 0.18 × 0.04	0.18 × 0.14 × 0.10	0.23 × 0.10 × 0.10	0.21 × 0.20 × 0.16
Radiation	CuK _α (λ = 1.54178)	CuK _α (λ = 1.54178)	CuK _α (λ = 1.54178)	CuK _α (λ = 1.54178)
2θ range /°	5.93 to 136.49	4.42 to 136.46	6.44 to 136.48	6.03 to 136.48
Reflections collected	28526	129662	143215	68979
Independent reflections	8337	7697	9641	8850
Restraints/parameters	4 / 730	14 / 682	70 / 799	21 / 678
Goof on F²	1.081	1.046	1.025	1.056
Final R indices [I ≥ 2σ(I)]	R ₁ = 0.0580, wR ₂ = 0.1659	R ₁ = 0.0249, wR ₂ = 0.0674	R ₁ = 0.0232, wR ₂ = 0.0607	R ₁ = 0.0771, wR ₂ = 0.1548
Final R indices [all data]	R ₁ = 0.0607, wR ₂ = 0.1719	R ₁ = 0.0250, wR ₂ = 0.0675	R ₁ = 0.0242, wR ₂ = 0.0613	R ₁ = 0.1034, wR ₂ = 0.1672
Residual density / e⁻ Å⁻³	1.92/-2.27	0.68/-0.72	0.76/-0.57	1.23/-1.28

Table S2. Selected distances and angles for C1 – C4.

Distance / Å	C1	C2	C3	C4	Ni ₃ DyL' ₃	Ni ₃ YbL' ₃
Ln ^{III} – O _{phenol}	2.544(4) –	2.473(2) –	2.493(2) –	2.686(6) –	2.478(3) –	2.449(5) –
	2.597(4)	2.543(2)	2.525(2)	2.752(6)	2.500(4)	2.478(6)
Ln ^{III} – O _{nitrate}	2.614(4) –	2.487(2) –	2.455(2) –	2.824(8) –	2.459(5) –	2.404(7) –
	2.658(4)	2.555(2)	2.516(2)	3.306(7)	2.514(5)	2.463(7)
Ln ^{III} – O _{water}	2.515(4)	—	—	2.881(9)	—	—
Ni ^{II} – O _{phenol}	1.978(4) –	1.963(2) –	1.962(2) –	1.990(6) –	1.961(4) –	1.961(6) –
	2.003(4)	1.999(2)	1.993(2)	2.015(6)	1.988(4)	1.982(5)
Ni ^{II} – O _{formyl}	1.983(4) –	1.981(2) –	1.975(2) –	1.990(6) –	2.004(4) –	2.000(7) –
	2.016(4)	2.008(2)	2.023(2)	2.013(6)	2.017(4)	2.011(6)
Ni ^{II} – O _{water}	2.073(4) –	2.076(2) –	—	2.067(6) –	—	1.983(16) –
	2.169(4)	2.167(2)	—	2.112(6)	—	2.080(30)
Ni ^{II} – O _{methanol}	2.098(4)	—	2.061(2) –	2.068(7) –	2.050(5) –	2.055(7) –
			2.120(2)	2.099(7)	2.098(6)	2.237(15)
Av. Ni ^{II} ... Ni ^{II}	6.339(2)	6.276(1)	6.296(1)	6.404(3)	6.364(3)	6.342(3)
Av. Ni ^{II} ... Ln ^{III}	3.670(2)	3.625(1)	3.636(1)	3.761(2)	3.675(3)	3.662(3)
Min. Ln ^{III} ... Ln ^{III}	9.160(1)	9.767(1)	10.216(1)	10.430(1)	9.912(2)	9.900(2)
Angle / °						
Ni ^{II} – O – Ln ^{III}	105.6(2) –	106.2(1) –	106.1(1) –	102.6(2) –	109.8(2) –	110.2(2) –
	107.3(2)	108.7(1)	108.9(1)	105.4(2)	110.9(2)	111.2(2)

Table S3. List of π – π interactions with centroid-centroid distances and angles, as calculated by Olex2. The second plane in each row corresponds to a symmetry generated molecule.

Complex	Planes	d _{centroid-centroid} / Å	Angle / °
C1	<i><C1, C2, C3, C4, C5, C6></i> <i><C25, C26, C27, C28, C29, C30></i>	3.519	5.130
	<i><C41, C42, C43, C44, C45, C50></i> <i><C25, C26, C27, C28, C29, C30></i>	3.597	4.422
C2	<i><C41, C42, C43, C44, C45, C50></i> <i><C45, C46, C47, C48, C49, C50></i>	3.668	8.272
	<i><C21, C22, C23, C24, C25, C30></i> <i><C25, C26, C27, C28, C29, C30></i>	3.820	3.426
	<i><C1, C2, C3, C4, C5, C10></i> <i><C25, C26, C27, C28, C29, C30></i>	3.885	18.729
	<i><C41, C42, C43, C44, C45, C50></i> <i><C41, C42, C43, C44, C45, C50></i>	3.960	8.900
C3	<i><C1, C2, C3, C4, C5, C10></i> <i><C5, C6, C7, C8, C9, C10></i>	3.606	5.855
	<i><C5, C6, C7, C8, C9, C10></i> <i><C5, C6, C7, C8, C9, C10></i>	3.723	0.000
	<i><C1, C2, C3, C4, C5, C10></i> <i><C25, C26, C27, C28, C29, C30></i>	3.580	3.674
C4	<i><C21, C22, C23, C24, C25, C30></i> <i><C25, C26, C27, C28, C29, C30></i>	3.714	4.300

Table S4. Calculated high-temperature-limit values per ion and per molecule for **C1 – C3**.

		Per ion			
M	<i>J</i> (S)	<i>g_J</i> (calc.)	$\chi_m T / \text{cm}^3 \text{K mol}^{-1}$ (calc.)	M_{sat} / μ_B (calc.)	
Ni ^{II}	1	2	1.000	2	
Eu ^{III}	0	-	0	0	
Gd ^{III}	7/2	2	7.875	7	

		Per molecule			
Complex	$\chi_m T / \text{cm}^3 \text{K mol}^{-1}$ (calc.)	$\chi_m T$ (calc.)	M_{sat} / μ_B (exp. at 7 T)	M_{sat} / μ_B (calc.)	
C2	7.57 (5000 Oe)	3.00	8.07 (1.8 K, in eicosane)	6	
	7.49 (500 Oe)				
	7.49 (500 Oe, in eicosane)				
C3	12.68 (5000 Oe)	10.88	13.66 (1.8 K, in eicosane)	14	
	12.45 (500 Oe)				
	12.40 (500 Oe, in eicosane)				

Fitting parameters, *A* and *B*, for the van Vleck equation used to fit $\chi_m T$ vs *T* plot for C3

$$\begin{aligned}
 A &= 3\exp(-48J/k_B) + 30\exp(-45J/k_B) + 105\exp(-40J/k_B) + 60 \\
 &+ 315\exp(-30J/k_B) + 504\exp(-27J/k_B) + 495\exp(-24J/k_B) + \\
 &+ 990\exp(-18J/k_B) + 1485\exp(-14J/k_B) + 858\exp(-13J/k_B)
 \end{aligned} \quad (1)$$

$$\begin{aligned}
 B &= \exp(-48J/k_B) + 2\exp(-45J/k_B) + 3\exp(-40J/k_B) + 4\exp(-30J/k_B) \\
 &+ 8\exp(-27J/k_B) + 5\exp(-24J/k_B) + 12\exp(-23J/k_B) + 4\exp(-14J/k_B) \\
 &+ 6\exp(-13J/k_B) + 12\exp(-7J/k_B) + 7
 \end{aligned} \quad (2)$$

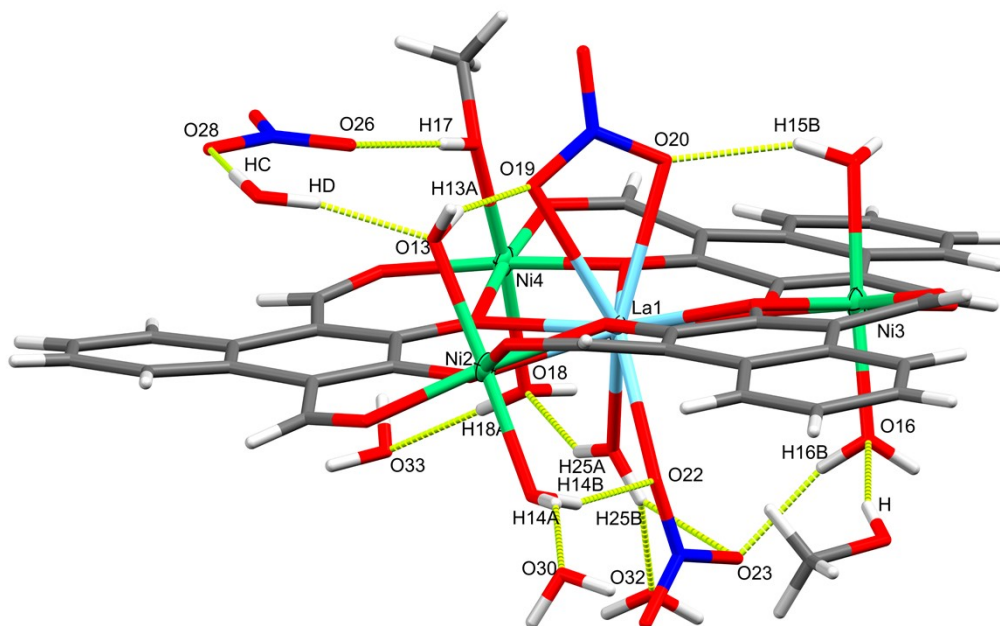


Figure S1. X-ray crystal structure showing the full asymmetry unit of **C1**. Thermal ellipsoids of metal atoms shown at the 70% probability level. C = grey, O = red, N = blue, H-bonds = yellow.

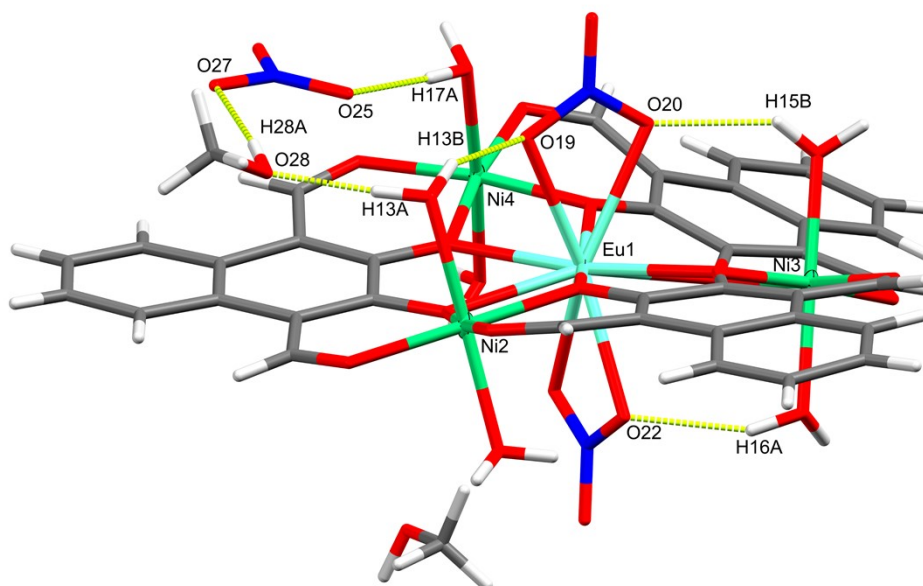


Figure S2. X-ray crystal structure showing the full asymmetry unit of **C2**. Thermal ellipsoids of metal atoms shown at the 70% probability level. C = grey, O = red, N = blue, H-bonds = yellow.

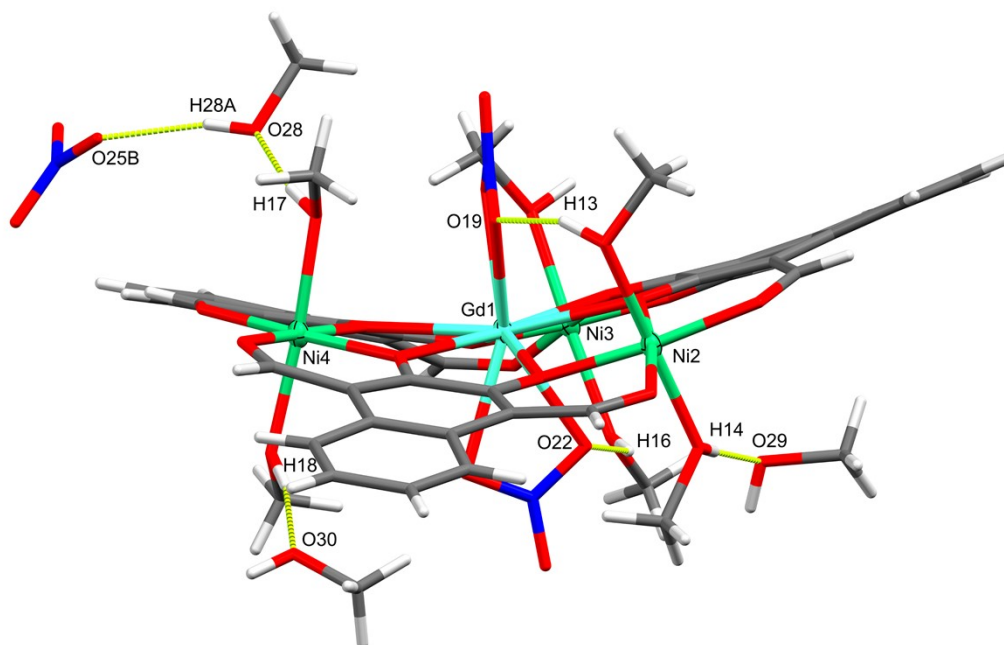


Figure S3. X-ray crystal structure showing the full asymmetry unit of **C3**. Thermal ellipsoids of metal atoms shown at the 70% probability level. C = grey, O = red, N = blue, H-bonds = yellow.

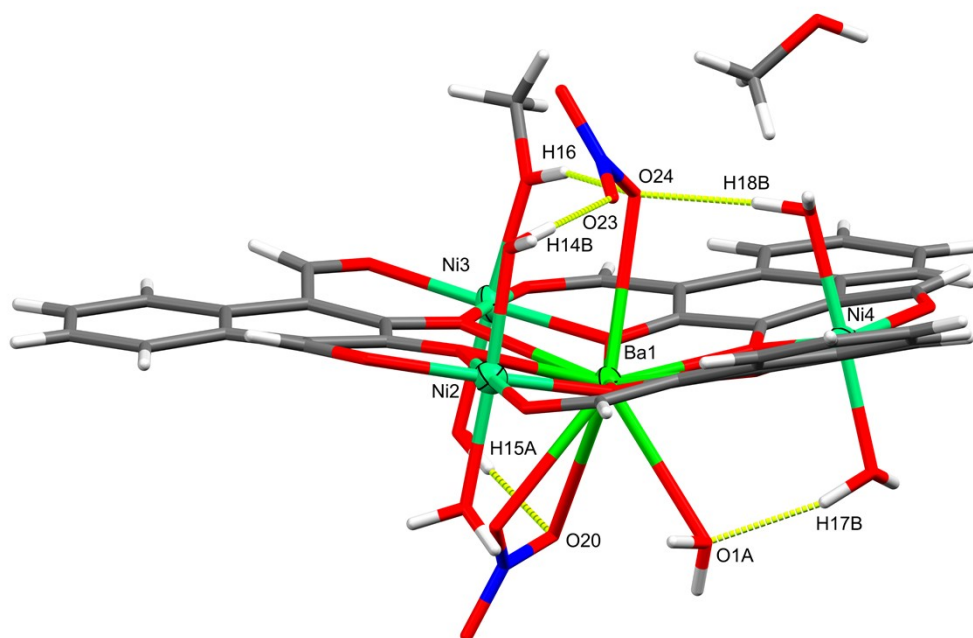


Figure S4. X-ray crystal structure showing the full asymmetry unit of **C4**. Thermal ellipsoids of metal atoms shown at the 70% probability level. C = grey, O = red, N = blue, H-bonds = yellow.

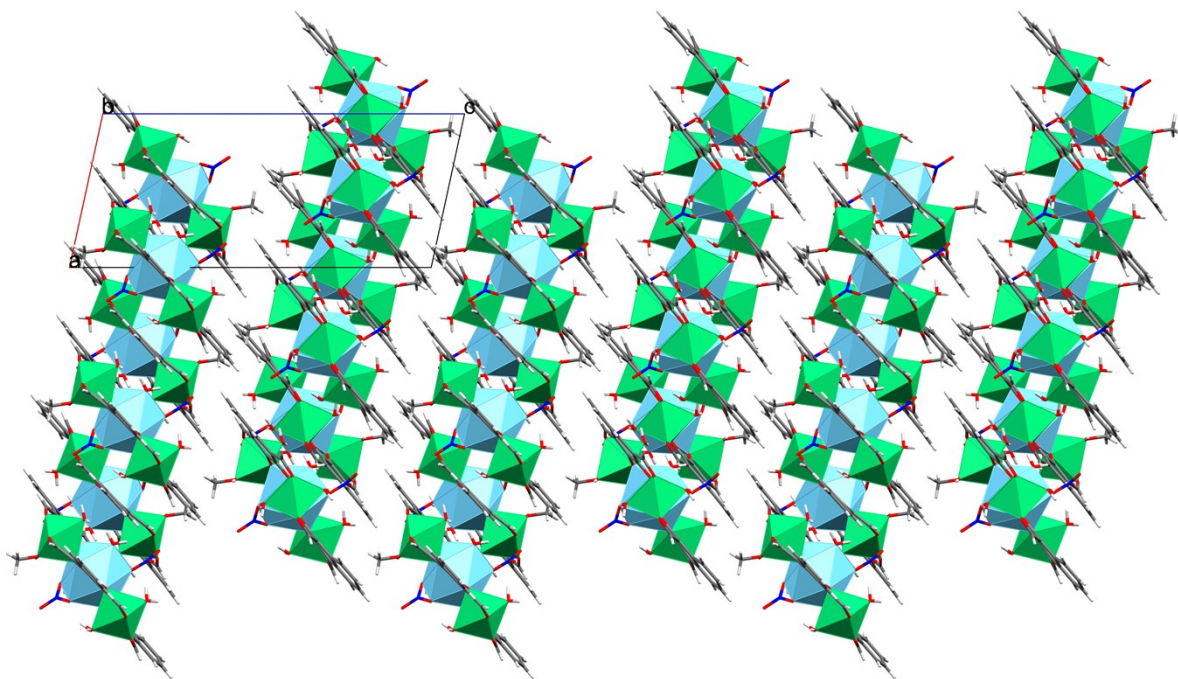


Figure S5. X-ray crystal packing diagram (3x3x3) of C1.

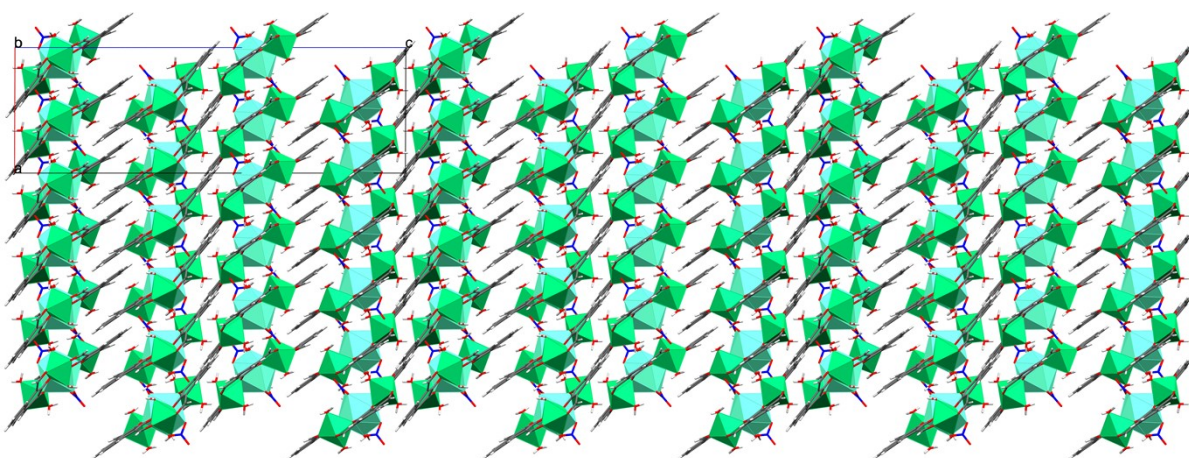


Figure S6. X-ray crystal packing diagram (3x3x3) of C2.

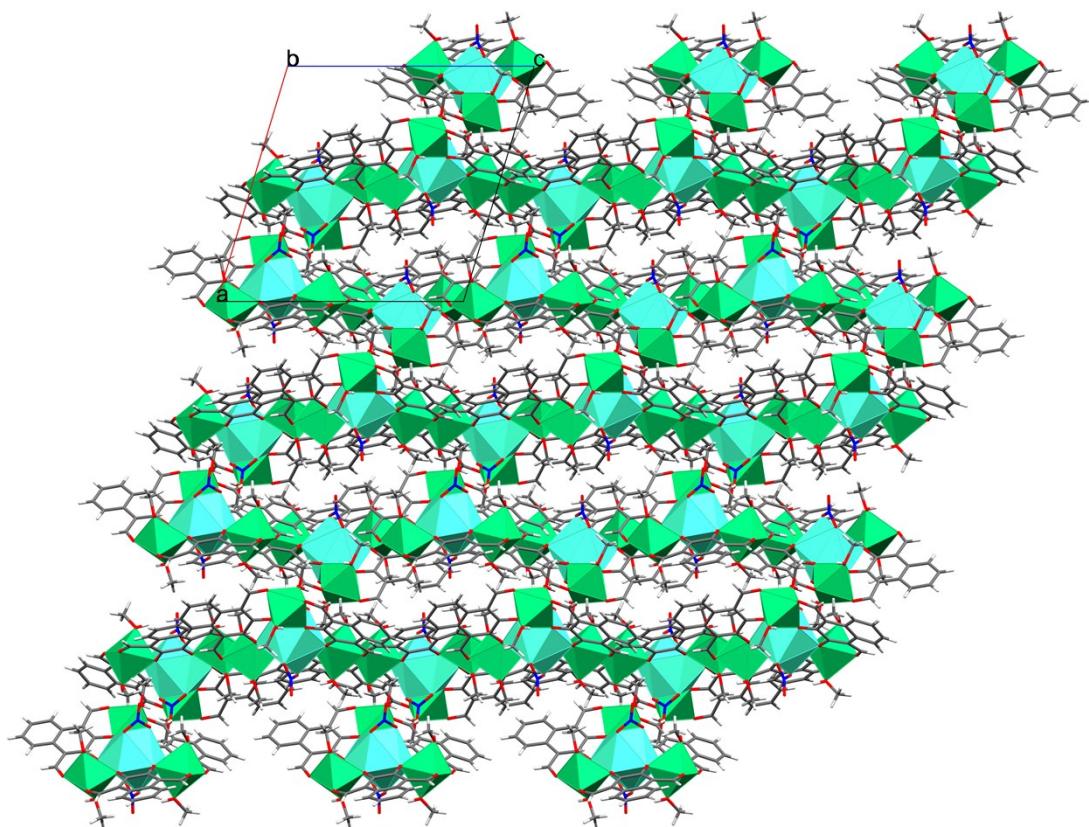


Figure S7. X-ray crystal packing diagram (3x3x3) of C3.

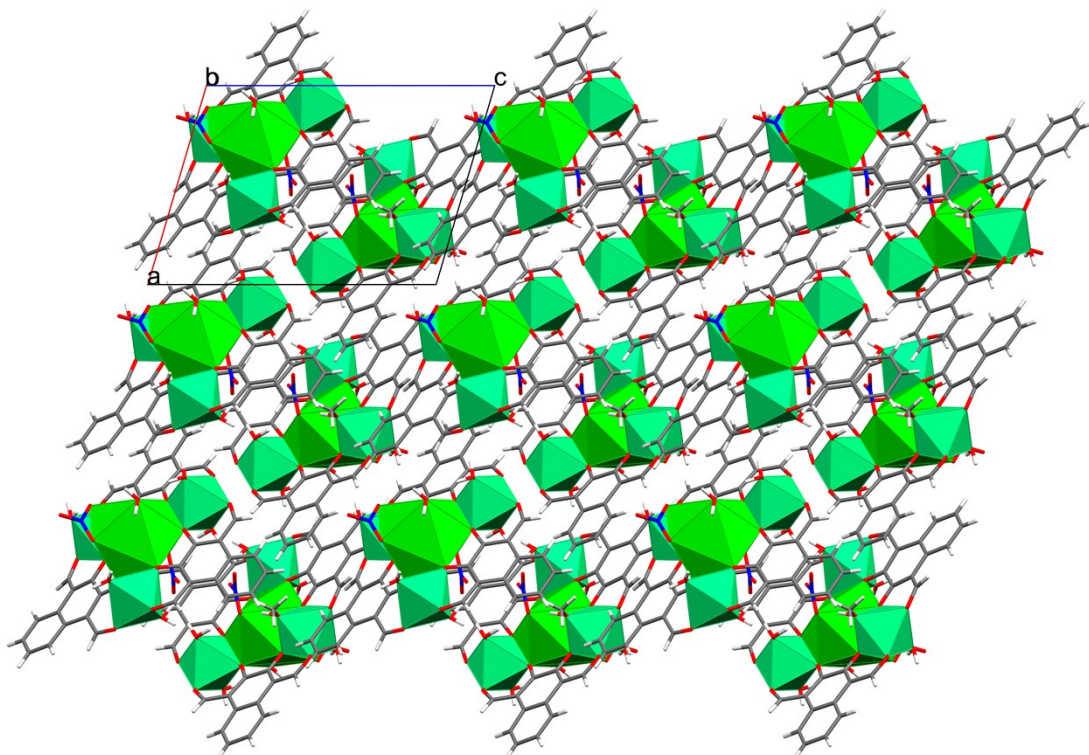


Figure S8. X-ray crystal packing diagram (3x3x3) of C4.

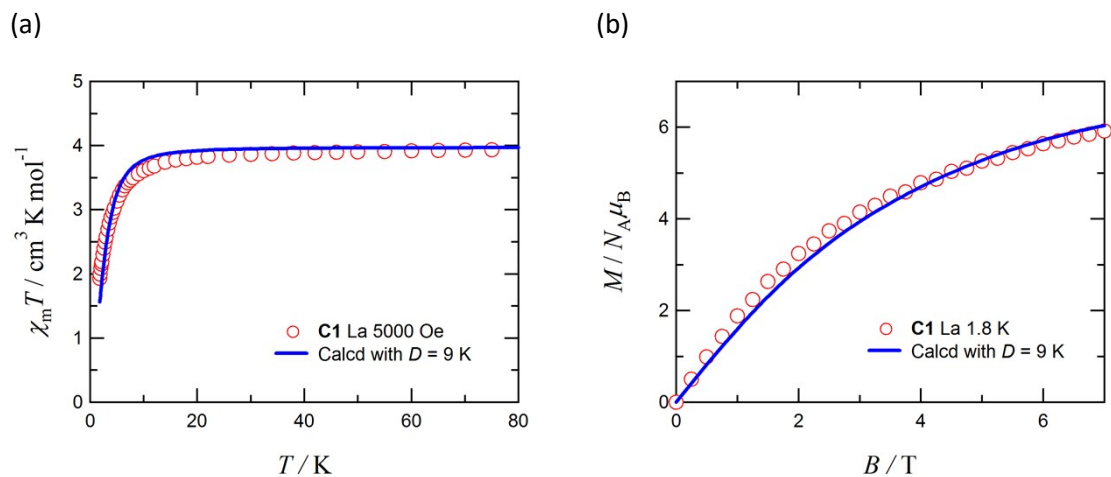


Figure S9. Simulation of the $\chi_m T(T)$ and $M(H)$ data on C1 for optimization of D with $g_{\text{Ni}} = 2.30$.

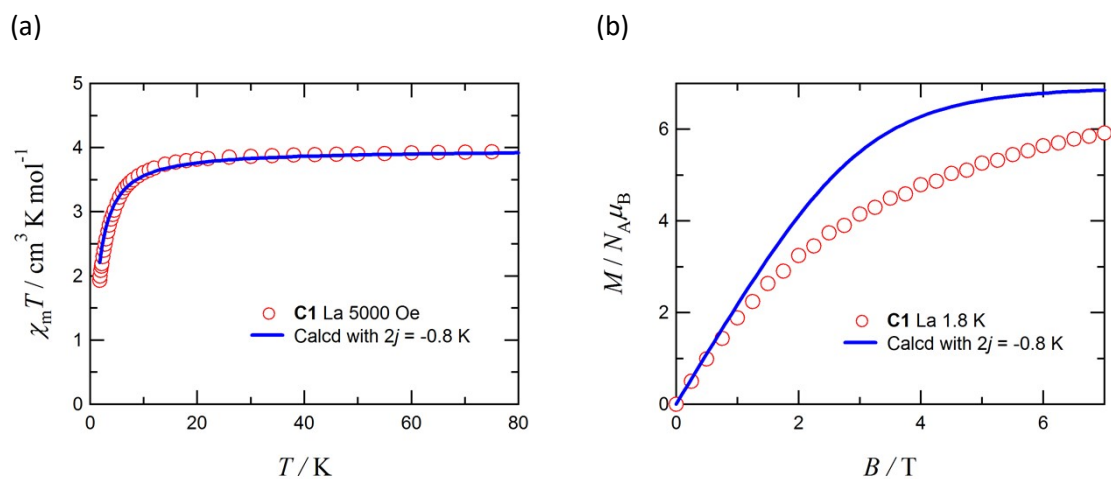


Figure S10. Simulation of the $\chi_m T(T)$ and $M(H)$ data on C1 for optimization of $j_{\text{Ni-Ni}}$ with $g_{\text{Ni}} = 2.30$.

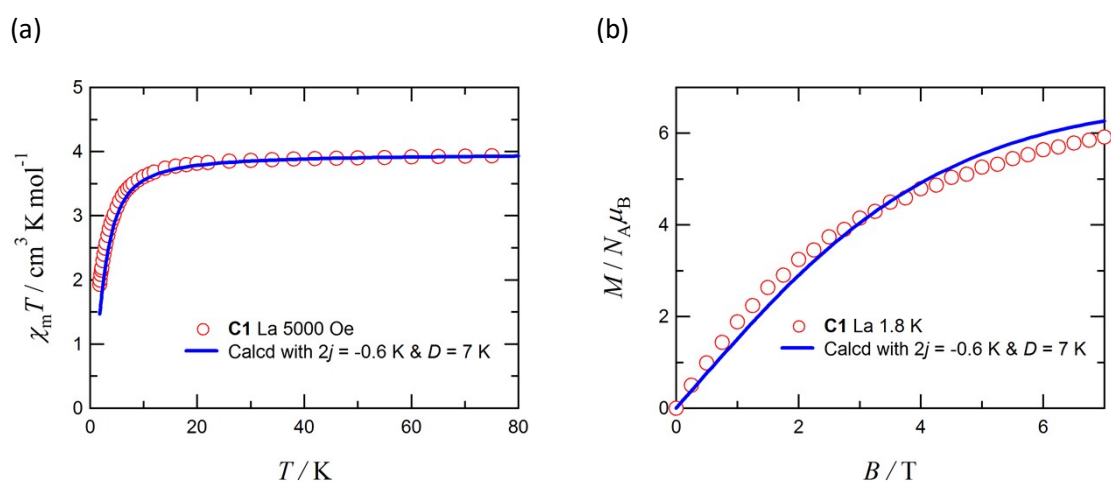


Figure S11. Simulation of the $\chi_m T(T)$ and $M(H)$ data on C1 with $2j_{\text{Ni-Ni}} = -0.6$ K, $D = 7$ K, and $g_{\text{Ni}} = 2.30$.

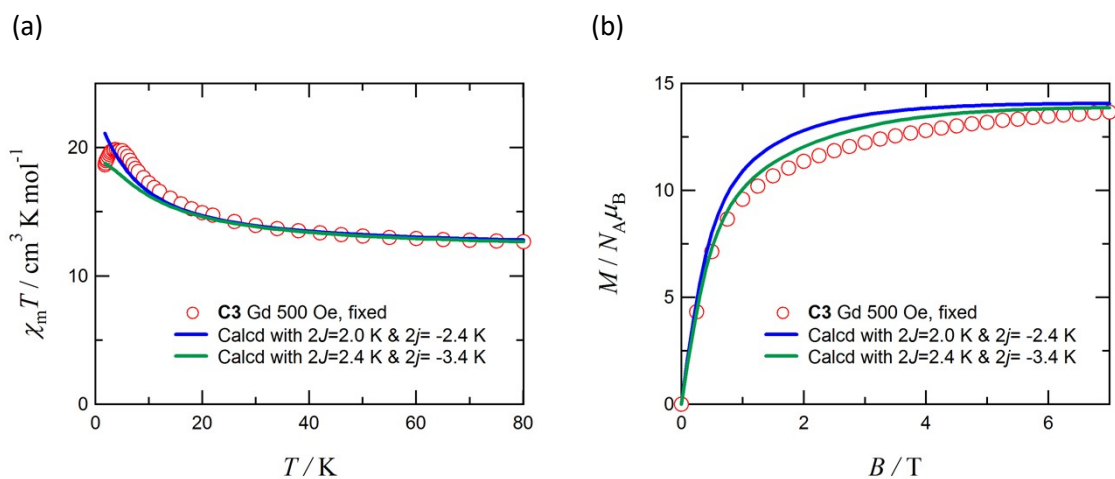


Figure S12. Simulation of the $\chi_m T(T)$ and $M(H)$ data on **C3** for optimization of J and $j_{\text{Ni-Ni}}$ with $g_{\text{Gd}} = 2.00$ and $g_{\text{Ni}} = 2.36 - 2.30$.

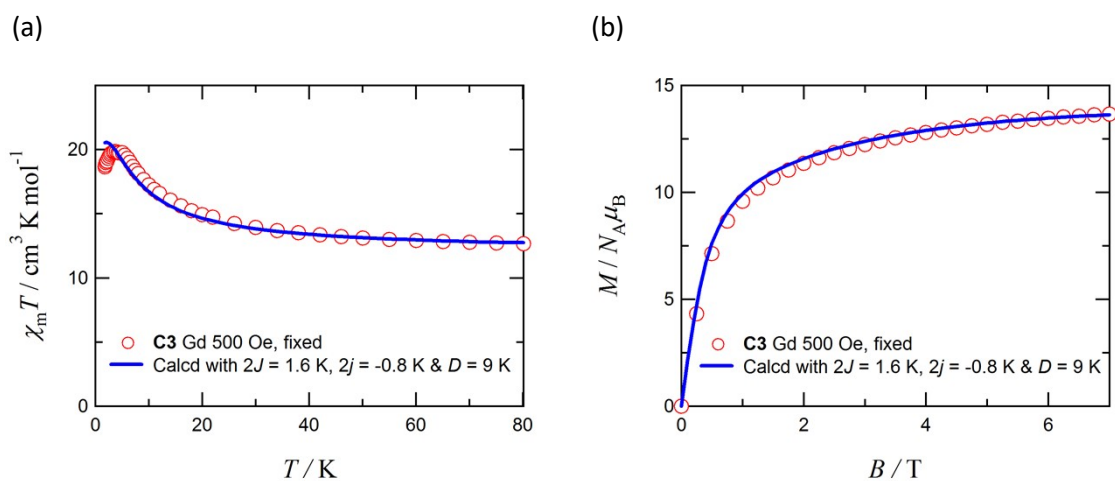


Figure S13. Simulation of the $\chi_m T(T)$ and $M(H)$ data on **C3** with $2J = 1.6 \text{ K}$, $2j_{\text{Ni-Ni}} = -0.8 \text{ K}$, $D = 9 \text{ K}$, $g_{\text{Gd}} = 2.00$, and $g_{\text{Ni}} = 2.36$.

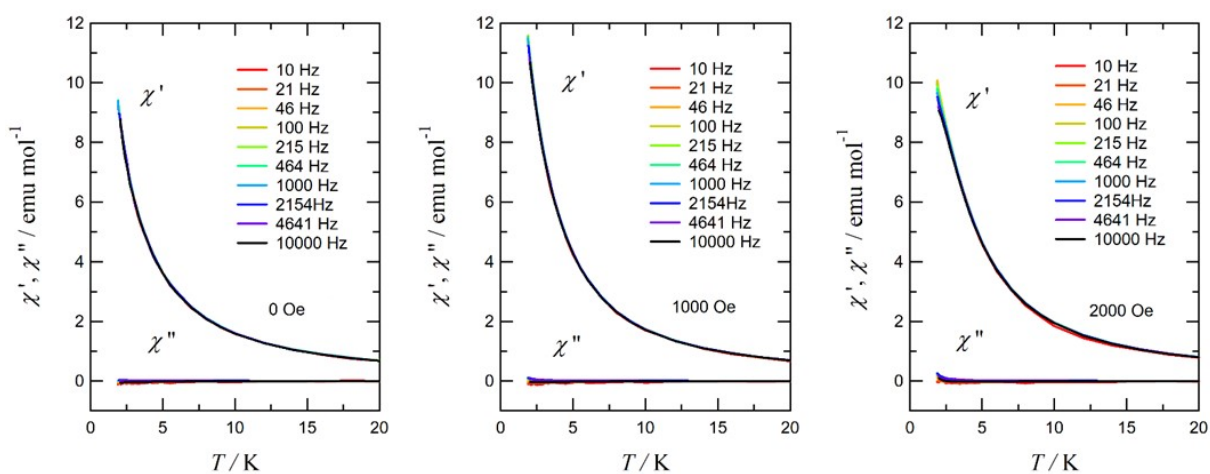


Figure S14. AC magnetic susceptibilities (in-phase χ' and out-of-phase χ'') measured for **C3**. The applied dc bias fields (0, 1000, and 2000 Oe) are indicated. Lines are shown as a guide to the eye.