

## Supporting Information

### Three Isostructural Hexanuclear Lanthanide-Organic Frameworks for Sensitive Luminescence Temperature Sensing Over a Wide Range

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**Table S1** The molar ratio of the starting Eu/Tb salt and that in isomorphous Eu<sup>3+</sup>/Tb<sup>3+</sup> co-doped MOFs calculated by ICP analyses.

Samples	Tested Eu by ICP analysis (ppm)	Tested Tb by ICP analysis (ppm)	The Eu/Tb ratios calculated by ICP analysis
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -0N	0.3233	124.1	0.0026:0.9974
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -1N	0.0673	38.94	0.0017:0.9983
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -2N	0.1002	32.80	0.0030:0.9970

**Table S2** Fitting parameters of isomorphous Eu<sup>3+</sup>/Tb<sup>3+</sup> mixed MOFs with double exponential Mott-Seitz model.

Ln-MOFs	$A_0$	$\alpha_1$	$\Delta E_1$	$\alpha_2$	$\Delta E_2$	$R^2$
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -0N <sup>a</sup>	1.14611	4.93616	121.858	196.053	630.975	0.9986
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -0N <sup>b</sup>	14.9587	4.9435	121.988	196.558	631.520	0.9986
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -1N <sup>a</sup>	1.18613	45.3146	101.299	3032.19	573.568	0.9999
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -1N <sup>b</sup>	299.108	45.1351	101.457	3021.64	573.990	0.9999
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -2N <sup>a</sup>	3.37050	9.48916	48.1217	2677.79	679.935	0.9996
Eu <sub>0.001</sub> Tb <sub>0.999</sub> -2N <sup>b</sup>	136.870	9.1291	48.7506	2589.08	680.871	0.9996

<sup>a</sup> Normalized to 50 K; <sup>b</sup> Normalized to 300 K.

**Table S3** A summary of some reported ratiometric MOF-based thermometers.

Luminescent MOF	Range (K)	$S_m$ (%·K <sup>-1</sup> )	$T_m$ (K)
<b>Eu<sub>0.001</sub>Tb<sub>0.999</sub>(BPDC-0N)</b>	<b>50-300</b>	<b>1.10</b>	<b>188</b>
<b>Eu<sub>0.001</sub>Tb<sub>0.999</sub>(BPDC-1N)</b>	<b>50-300</b>	<b>4.11</b>	<b>50</b>
<b>Eu<sub>0.001</sub>Tb<sub>0.999</sub>(BPDC-2N)</b>	<b>50-300</b>	<b>1.94</b>	<b>50</b>
9.1wt%EuW <sub>10</sub> @Tb-TATB <sup>1</sup>	200-320	2.68	300
19.5wt%EuW <sub>10</sub> @Tb-TATB <sup>1</sup>	200-320	2.37	300
Tb <sub>0.99</sub> Eu <sub>0.01</sub> (bdc) <sub>1.5</sub> (in water) <sup>2</sup>	290-320	0.31	318
Tb <sub>0.99</sub> Eu <sub>0.01</sub> (bdc) <sub>1.5</sub> (solid) <sup>2</sup>	290-320	0.14	318
Dycpia <sup>3</sup>	298-473	0.42	473
ZJU-88⊃perylene <sup>4</sup>	293-353	1.28	293
[Tb <sub>0.970</sub> Eu <sub>0.030</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.19	338
[Tb <sub>0.950</sub> Eu <sub>0.050</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.20	333
[Tb <sub>0.925</sub> Eu <sub>0.075</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.24	314
[Tb <sub>0.900</sub> Eu <sub>0.100</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.31	284
[Tb <sub>0.875</sub> Eu <sub>0.125</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.40	251
[Tb <sub>0.850</sub> Eu <sub>0.150</sub> (CH <sub>3</sub> COO)(1,3-bdc)(H <sub>2</sub> O)] <sup>5</sup>	150-350	0.44	236
Cdots&RB@ZIF-82-MMM-1 <sup>6</sup>	293-353	0.74	293
Eu <sub>0.0025</sub> Tb <sub>0.9975</sub> -BABDC-PBMA <sup>7</sup>	90-240	3.61	240

**Table S4** Crystallographic Data Collection and Refinement Results for **Eu(BPDC-2N)**.

<b>Sample</b>	<b>Eu(BPDC-2N)</b>
Chemical formula	C <sub>72</sub> H <sub>45</sub> Eu <sub>6</sub> F <sub>8</sub> N <sub>12</sub> O <sub>27.50</sub>
Formula weight	2581.96
Temperature (K)	173
Wavelength (Å)	0.71073
Crystal system	Cubic
Space group	Fm-3m
<i>a</i> (Å)	27.498(6)
<i>b</i> (Å)	27.498(6)
<i>c</i> (Å)	27.498(6)
<i>V</i> (Å <sup>3</sup> )	20792(15)
Z	4
Density (calculated g·cm <sup>-3</sup> )	0.825
Absorbance coefficient (mm <sup>-1</sup> )	1.824
<i>F</i> (000)	4924
Crystal size (mm <sup>3</sup> )	0.12 × 0.11 × 0.10
<i>R</i> (int)	0.0464
Goodness of fit on <i>F</i> <sub>2</sub>	1.029
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> [ <i>I</i> > 2σ( <i>I</i> )] <sup>a</sup>	0.0259, 0.0734
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all data) <sup>a</sup>	0.0276, 0.0744
Largest difference peak and hole (e Å <sup>-3</sup> )	0.555, -0.660

<sup>a</sup>

$$R1 = \sum (|F_o| - |F_c|) / \sum |F_o|; wR2 = \left[ \frac{\sum w(|F_o| - |F_c|)^2}{\sum wF_o^2} \right]^{1/2}$$

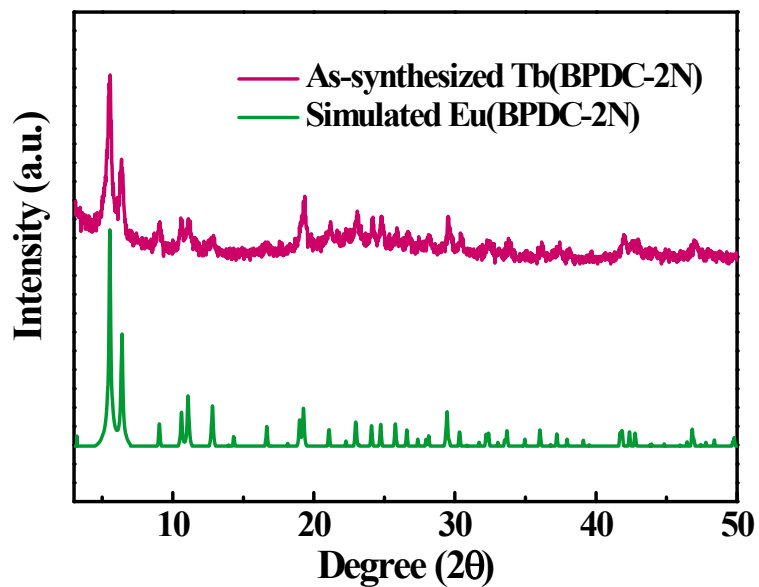


Figure S1. The PXRD patterns of simulated Eu(BPDC-2N) and as-synthesized Tb(BPDC-2N).

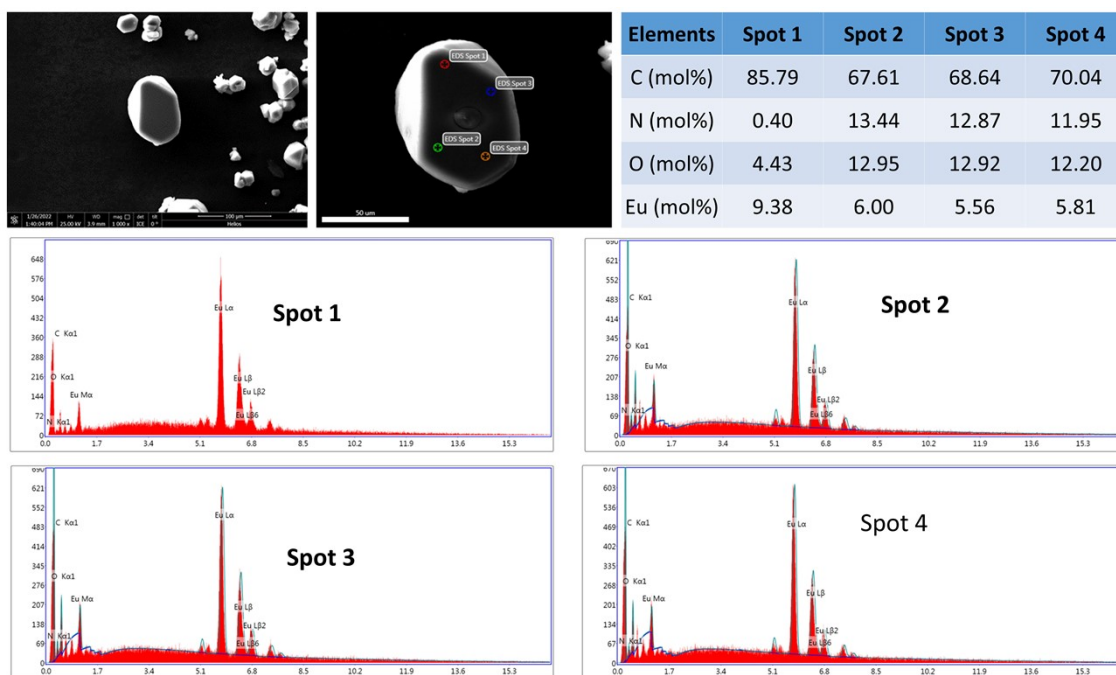


Figure S2. SEM and SEM-EDX of Eu(BPDC-2N).

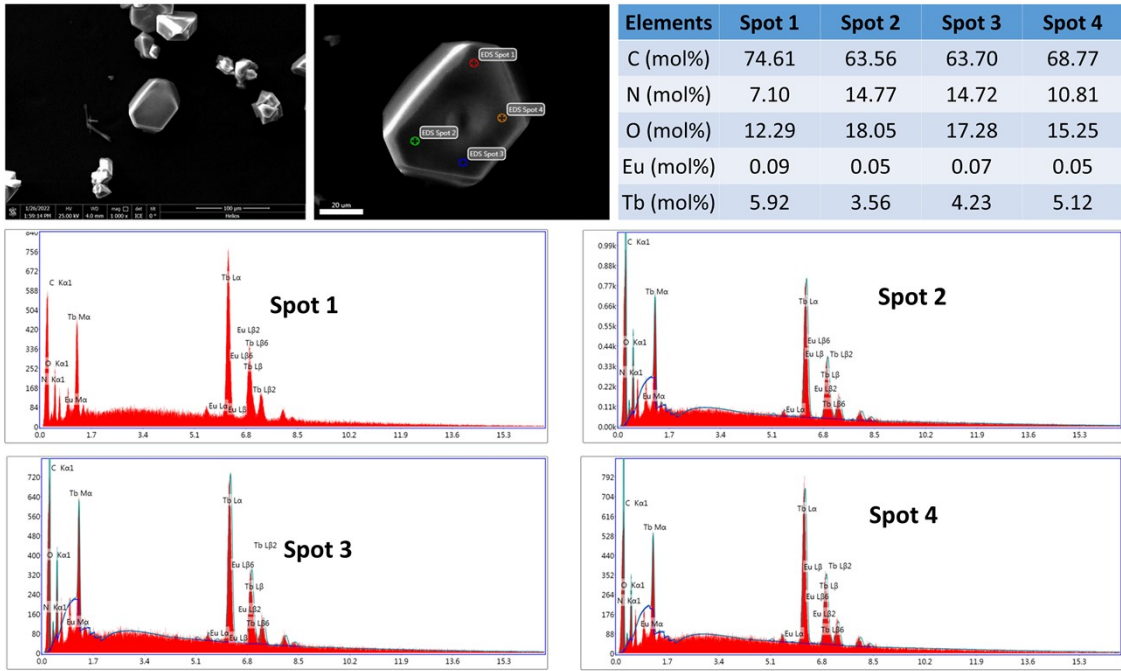


Figure S3. SEM and SEM-EDX of  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-2N})$ .

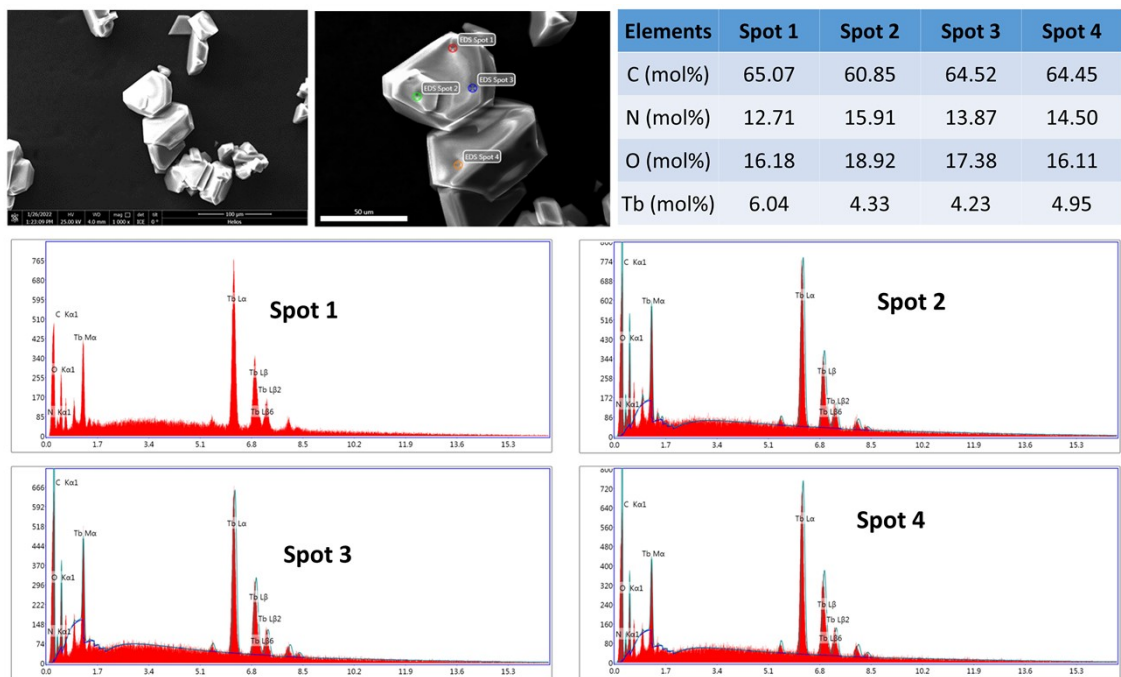
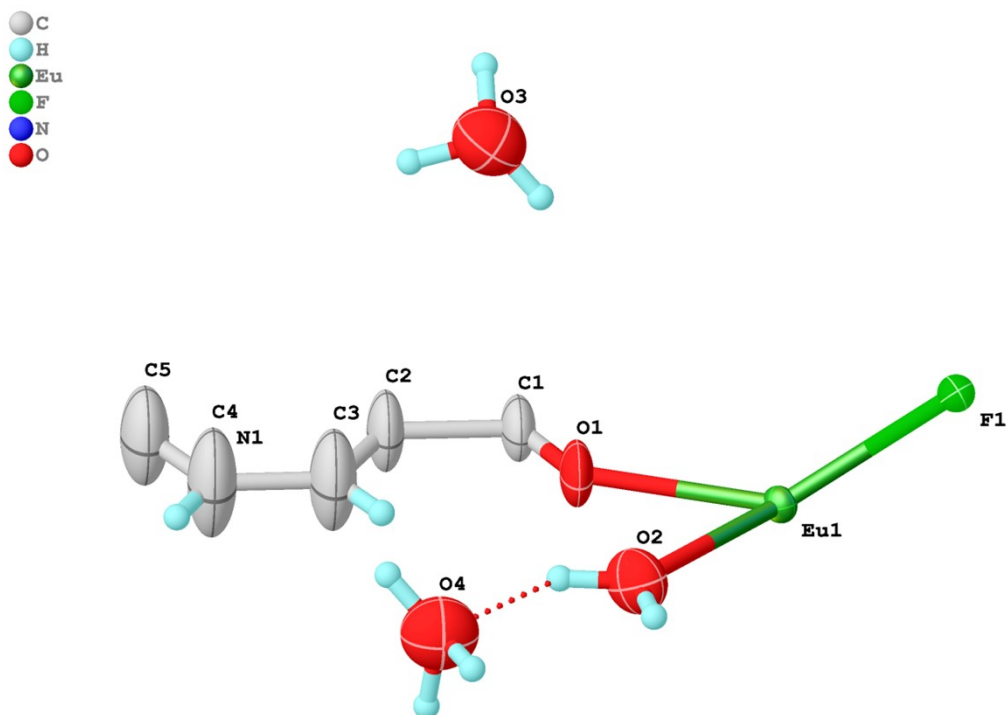
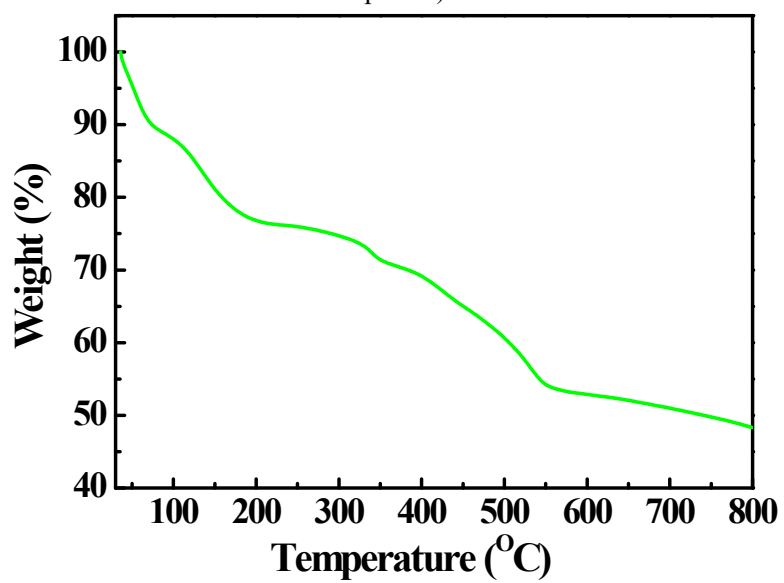


Figure S4. SEM and SEM-EDX of  $\text{Tb}(\text{BPDC-2N})$ .



**Figure S5.** Ortep representation of the asymmetric unit in Eu(BPDC-2N) (50% probability factor for the thermal ellipsoids).



**Figure S6.** TGA curves of Eu(BPDC-2N).

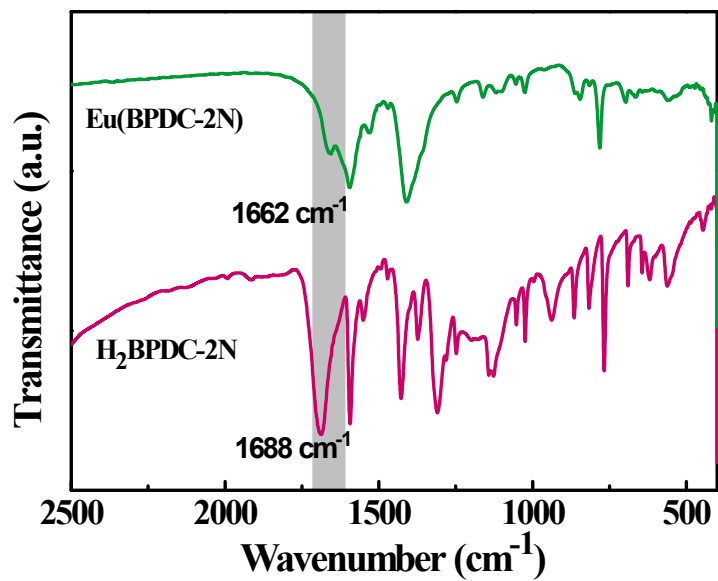


Figure S7. FT-IR spectra of organic ligand H<sub>2</sub>BPDC-2N and Eu(BPDC-2N).

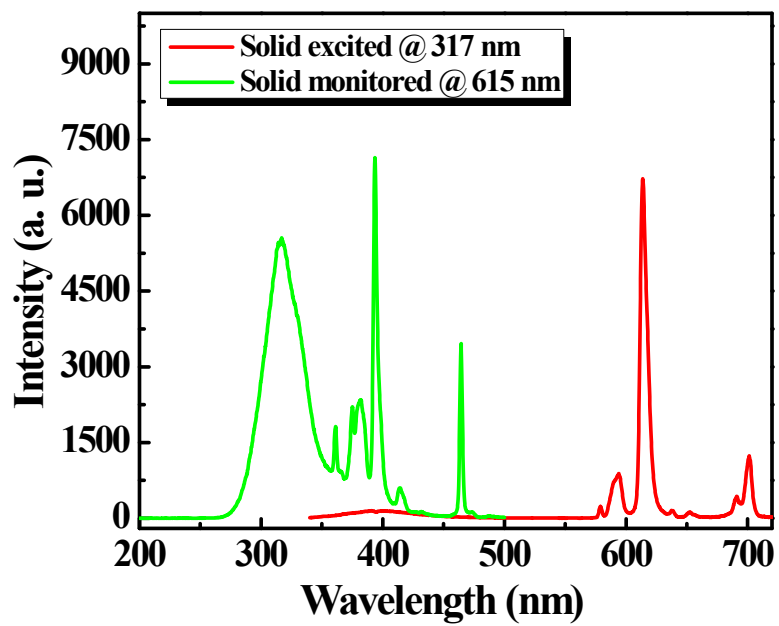


Figure S8. Excitation and emission spectra of Eu(BPDC-0N) at room temperature.

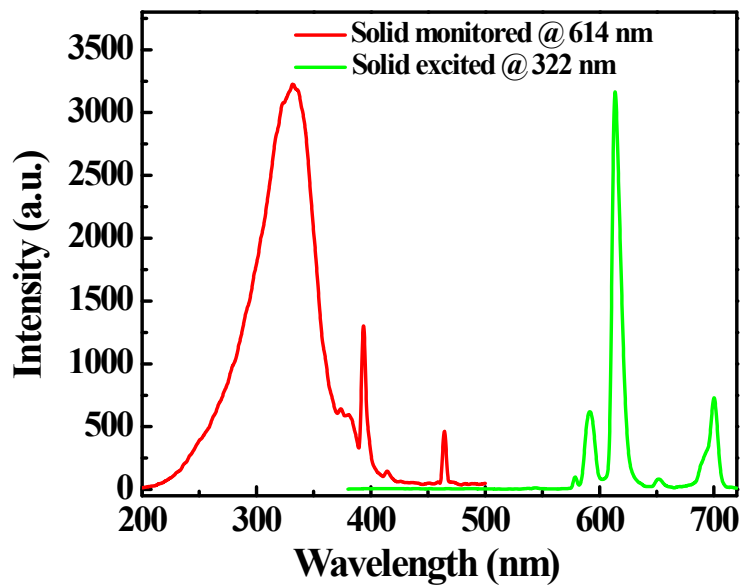


Figure S9. Excitation and emission spectra of Eu(BPDC-1N) at room temperature.

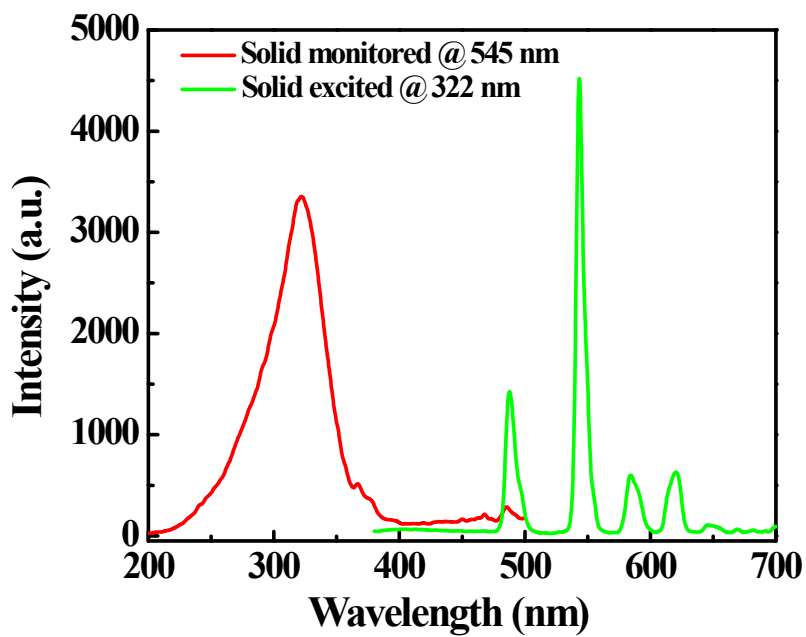


Figure S10. Excitation and emission spectra of Tb(BPDC-1N) at room temperature.



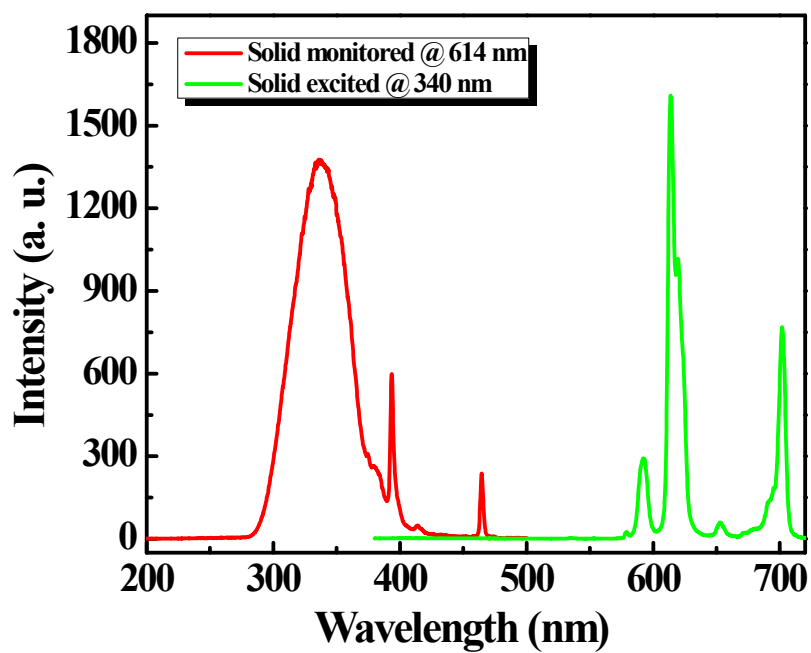


Figure S11. Excitation and emission spectra of Eu(BPDC-2N) at room temperature.

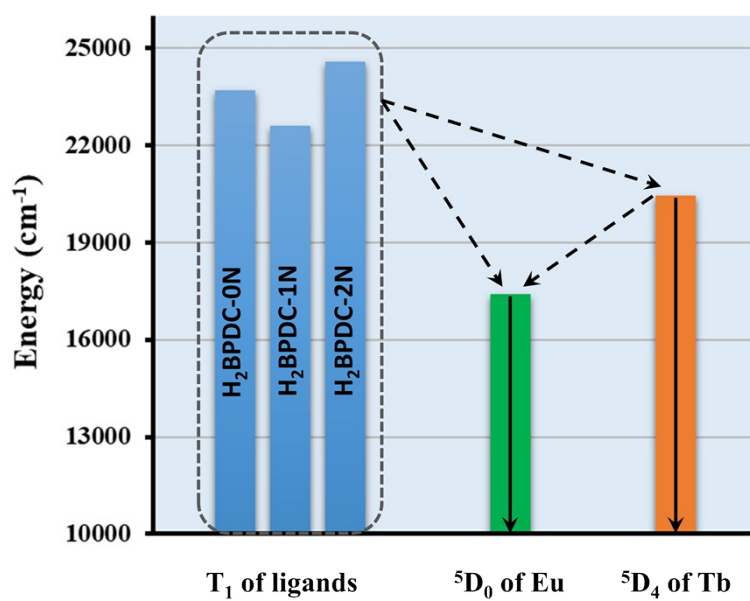
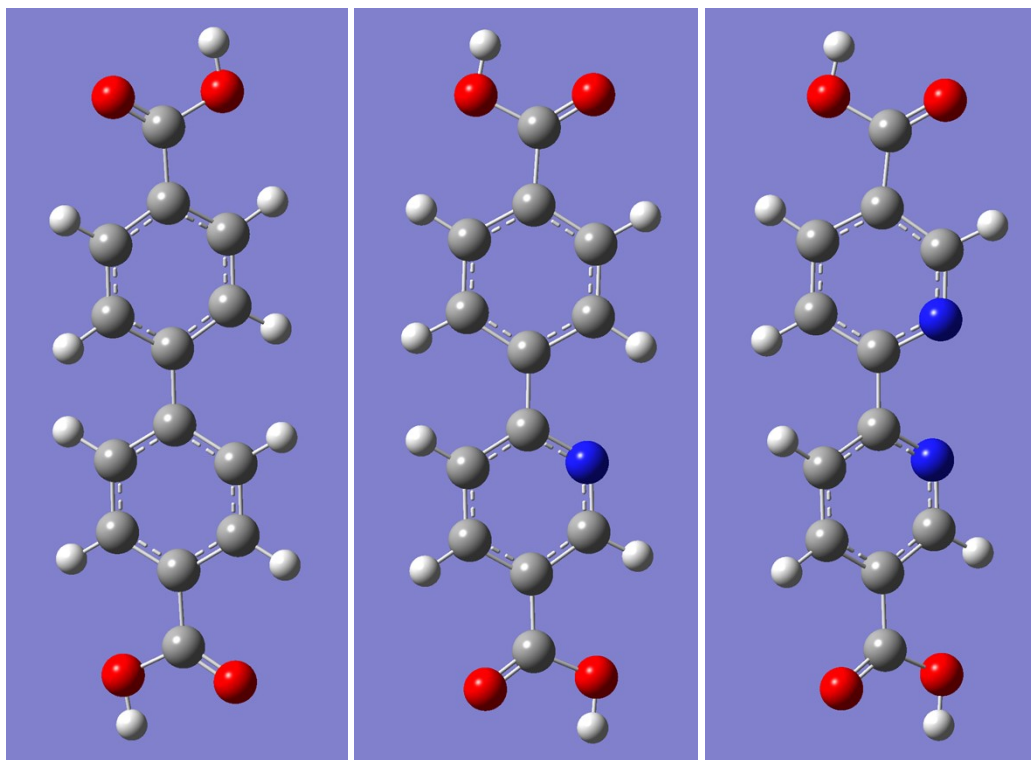
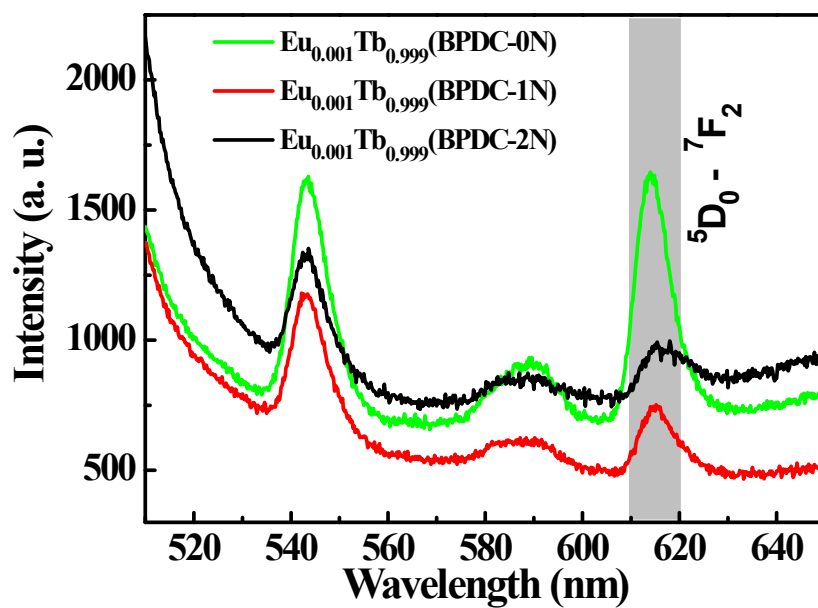


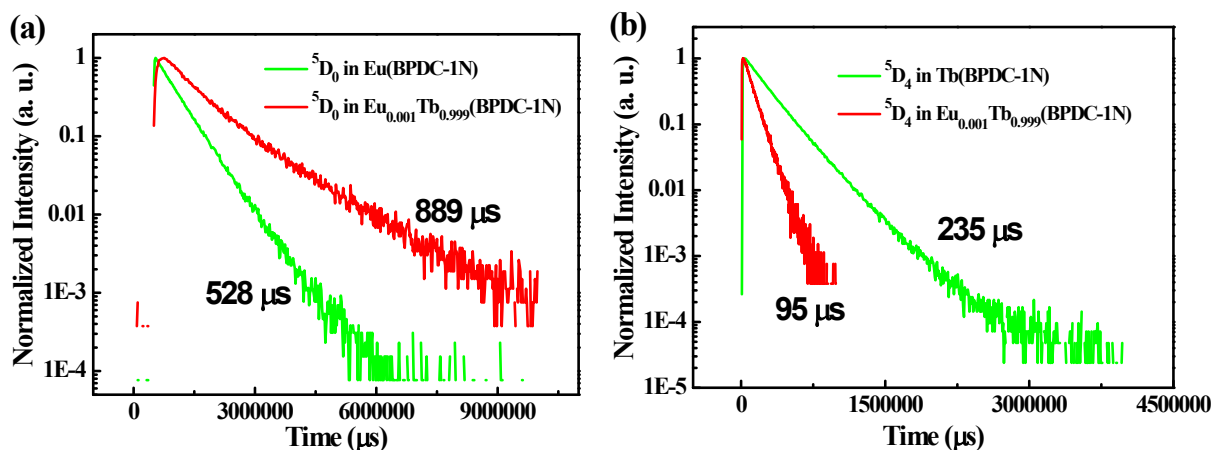
Figure S12. The calculated triplet levels of  $H_2BPDC-xN$  ( $x = 0, 1, 2$ ) and energy levels of Eu ( $^5D_0$ ) and Tb ( $^5D_4$ ).



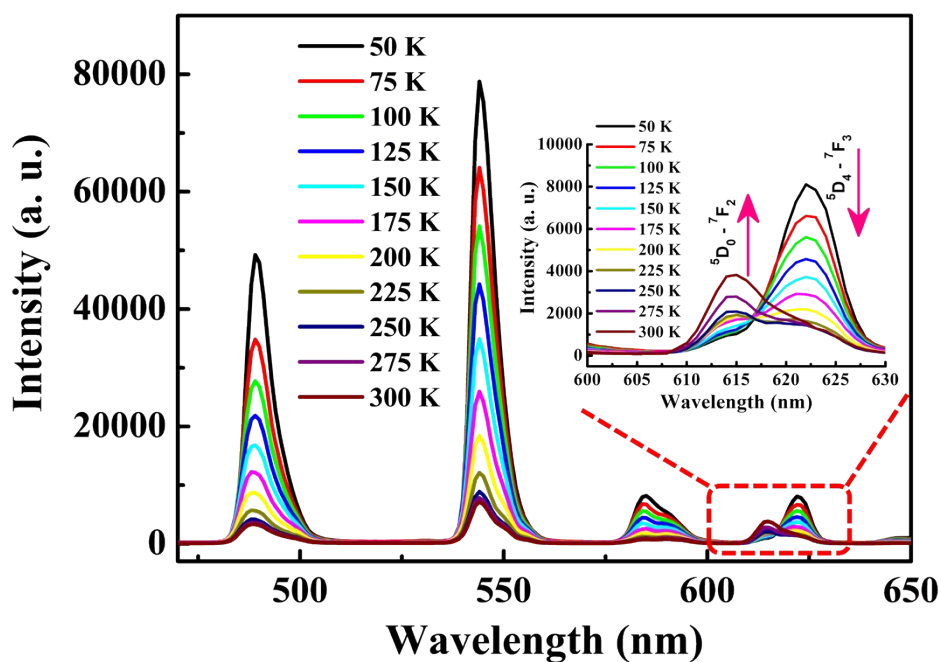
**Figure S13.** The optimized geometry of free ligands H<sub>2</sub>BPDC-*x*N (*x* = 0, 1, 2) at the B3LYP/6-31G\* level.



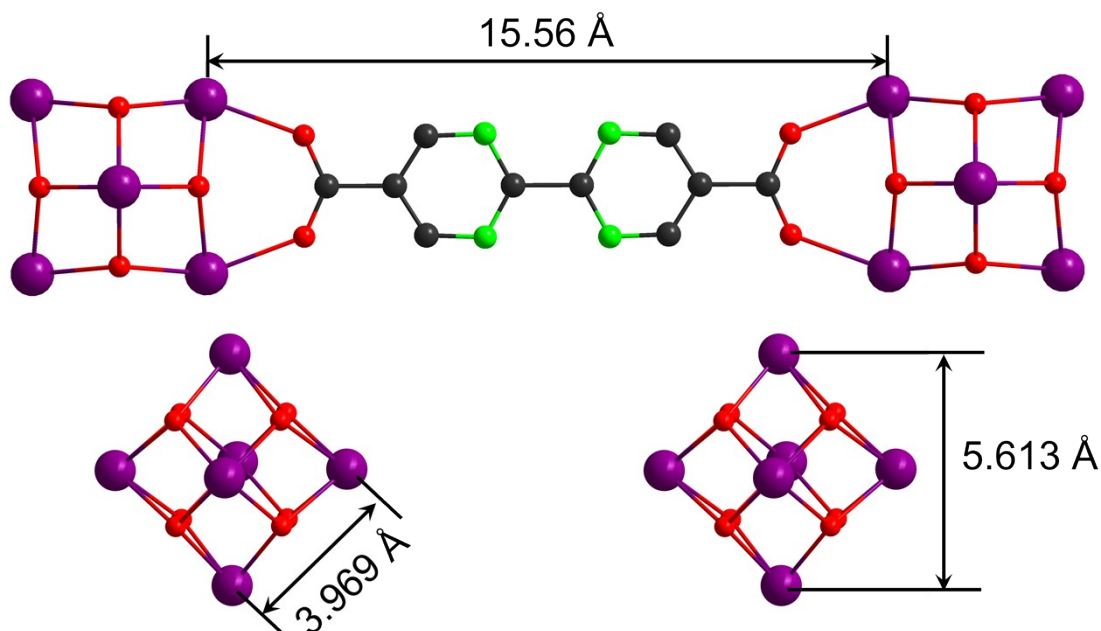
**Figure S14.** Emission spectra of Eu<sub>0.001</sub>Tb<sub>0.999</sub>(BPDC-*x*N) (*x* = 0, 1, 2) excited at 488 nm.



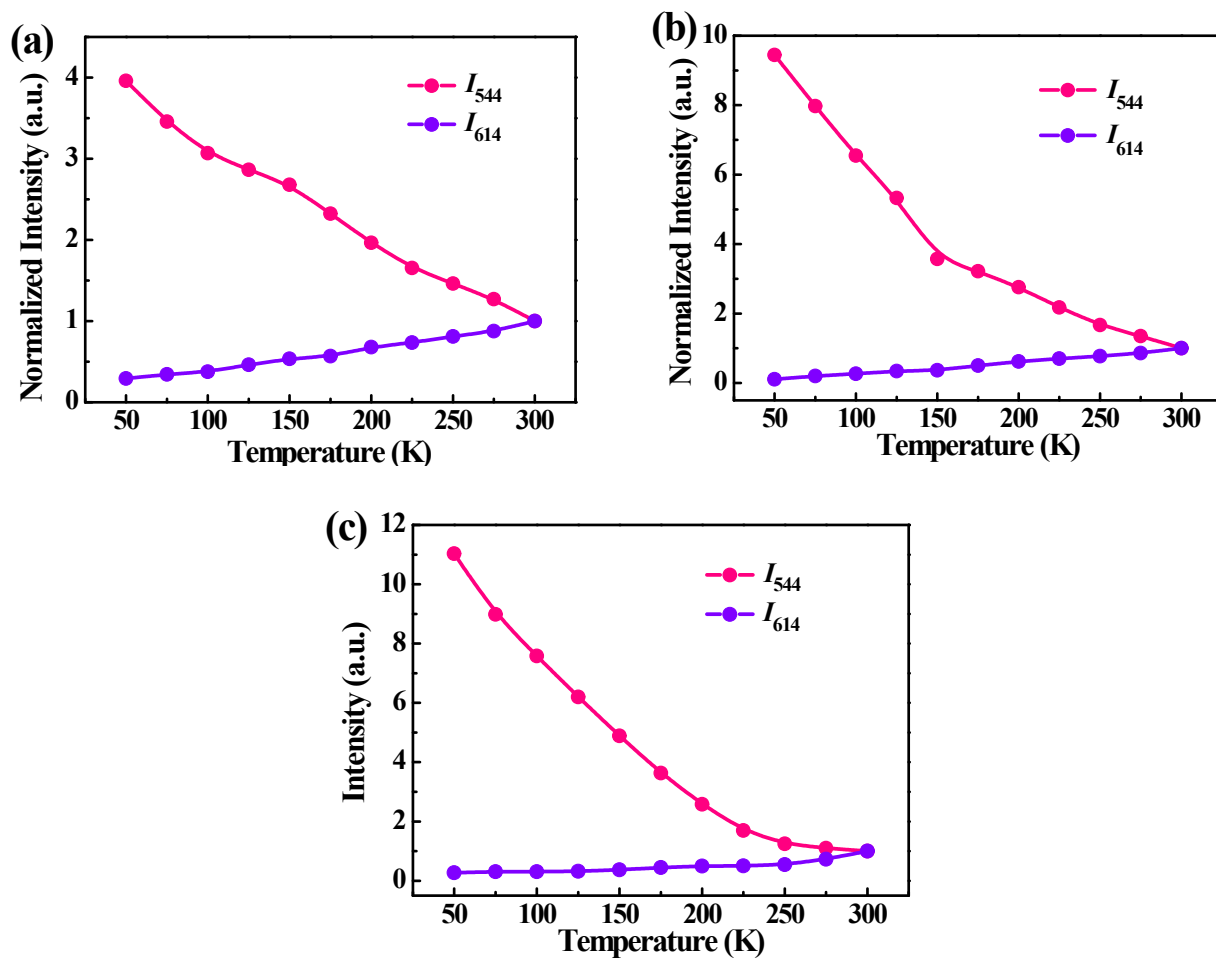
**Figure S15.** Fluorescence decay curves of  $^5D_0$  (a) and  $^5D_4$  (b) in Eu(BPDC-1N), Tb(BPDC-1N), and  $\text{Eu}_{0.001}\text{Tb}_{0.999}$ (BPDC-1N).



**Figure S16.** Temperature-dependent emission spectra of  $\text{Eu}_{0.001}\text{Tb}_{0.999}$ (BPDC-2N) in the temperature range from 50 K to 300 K.



**Figure S17.** The shortest distance of two lanthanide ions in the adjacent SBUs and two possible distances between  $\text{Tb}^{3+}$  and  $\text{Eu}^{3+}$  in the same SBU.



**Figure S18.** Normalized luminescence intensity of  ${}^5\text{D}_4 - {}^7\text{F}_5$  transition and  ${}^5\text{D}_0 - {}^7\text{F}_2$  transition for (a)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-0N})$ , (b)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-1N})$  and (c)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-2N})$ , respectively.

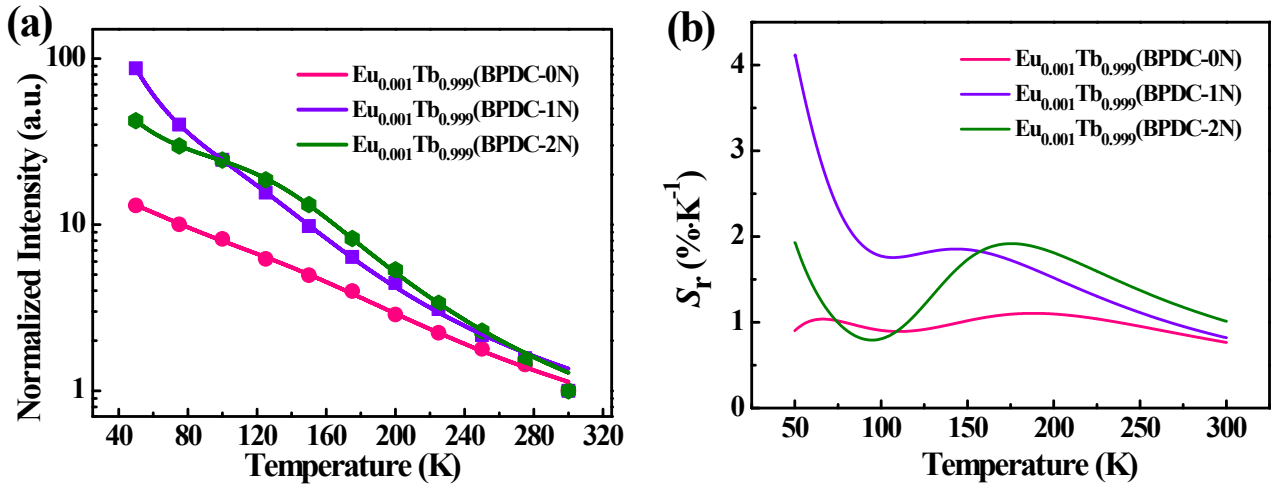


Figure S19. The fitting curves (a) and temperature-dependent relative sensitivity (b) of  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-}x\text{N})$  ( $x = 0, 1, 2$ ) when luminescence intensities were normalized to the intensity at 300 K.

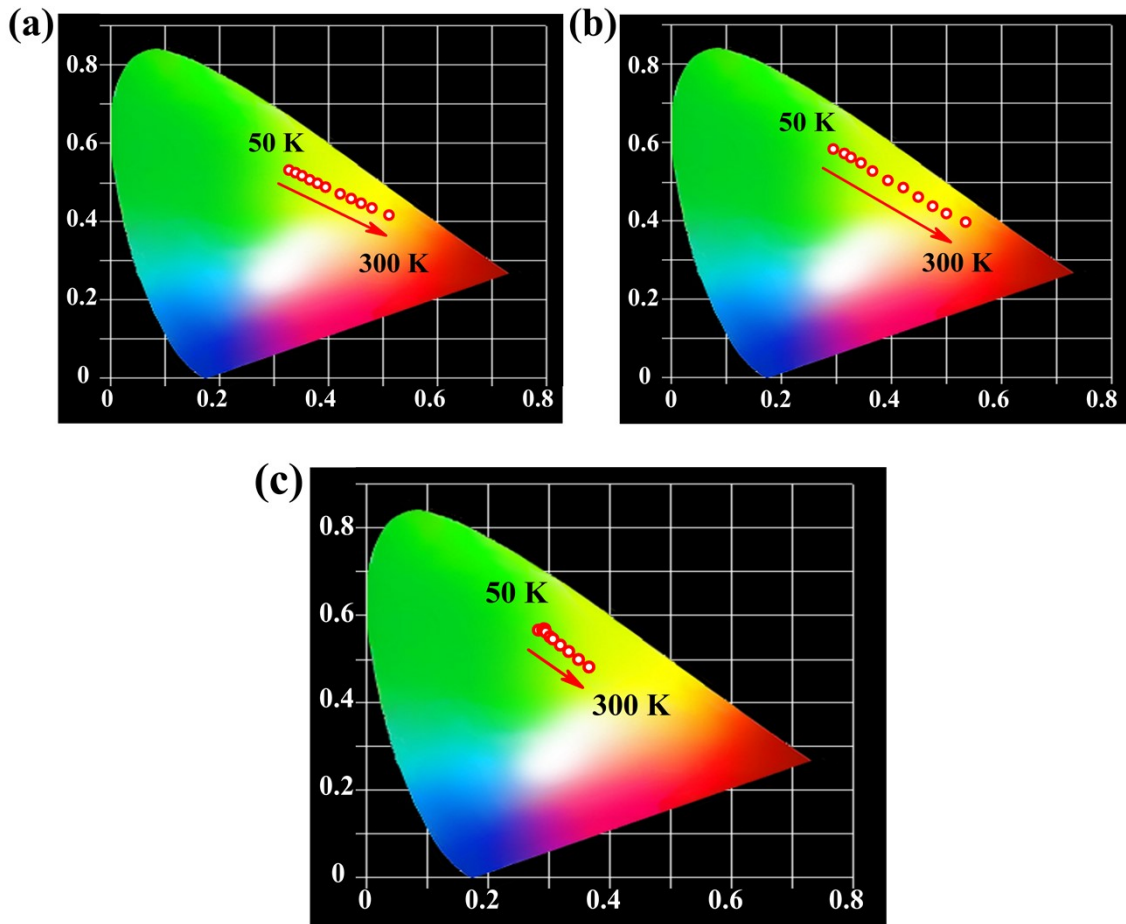
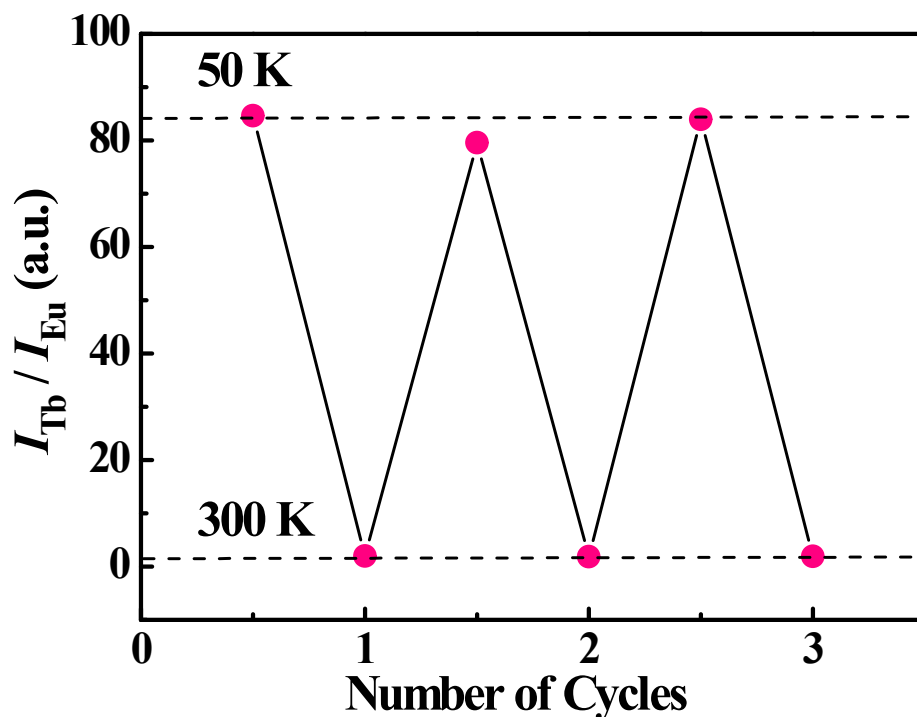


Figure S20. CIE chromaticity diagram presenting the temperature-dependent color of (a)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-}0\text{N})$ , (b)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-}1\text{N})$  and (c)  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-}2\text{N})$  in the temperature range from 50 K to 300 K.



**Figure S21.** The reversible emission intensity ratio of Tb (544 nm) to Eu (614 nm) of  $\text{Eu}_{0.001}\text{Tb}_{0.999}(\text{BPDC-2N})$  in cycles of heating and cooling.

## References

- 1 C. Viravaux, O. Oms, A. Dolbecq, E. Nassar, L. Busson, C. Mellot-Draznieks, R. Dessapt, H. Serier-Brault and P. Mialane, *Journal of Materials Chemistry C*, 2021, **9**, 8323.
- 2 A. Cadiau, C. D. S. Brites, P. M. F. J. Costa, R. A. S. Ferreira, J. Rocha and L. D. Carlos, *ACS Nano*, 2013, **7**, 7213.
- 3 T. Xia, Y. Cui, Y. Yang and G. Qian, *Journal of Materials Chemistry C*, 2017, **5**, 5044.
- 4 Y. Cui, R. Song, J. Yu, M. Liu, Z. Wang, C. Wu, Y. Yang, Z. Wang, B. Chen and G. Qian, *Advanced Materials*, 2015, **27**, 1420.
- 5 V. Trannoy, A. N. Carneiro Neto, C. D. S. Brites, L. D. Carlos and H. Serier-Brault, *Advanced Optical Materials*, 2021, **9**, 2001938.
- 6 Y. Ding, Y. Lu, K. Yu, S. Wang, D. Zhao and B. Chen, *Advanced Optical Materials*, 2021, **9**, 2100945.
- 7 T. Feng, Y. Ye, X. Liu, H. Cui, Z. Li, Y. Zhang, B. Liang, H. Li and B. Chen, *Angewandte Chemie International Edition*, 2020, **59**, 21752.