

## Supplementary Information

### Mesoporous Silica Nanoparticles-Embedded Lanthanide Organic Polyhedra for Enhanced Stability, Luminescence and Cell Imaging†

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## 1. General

Unless otherwise stated, all chemicals and solvents were purchased from commercial companies and used without further purification. Triethanolamine (TEA), Cetyltrimethylammonium bromide (CTAB), tetraethoxysilane (TEOS), 3-aminopropyltriethoxysilane (APS), triethanolamine (TEOA), Eu(OTf)<sub>3</sub>, D-biotin, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide (EDC), N-hydroxysuccinimide (NHS) were purchased from Adamas (Shanghai).

Thermogravimetric analysis (TGA) was investigated on a NETZSCH STA449F3 unit under N<sub>2</sub> atmosphere at a heating rate of 10 K min<sup>-1</sup>. Nitrogen adsorption/desorption isotherms were measured with an ASAP 2020 based on the Brunauer-Emmett-Teller (BET) method. Fourier transform infrared (FT-IR) spectra were recorded on the Bruker VERTEX70 system by mixing samples into KBr to prepare a compressed tablet sample. Inductively coupled plasma (ICP) analyses were performed on a Jobin Yvon Ultima2 spectrometer. The Zeta potentials of the hybrid nanoparticles were analyzed on a Malvern Nanosizer S instrument at room temperature. X-ray photoelectron spectroscopy (XPS) was conducted on a Thermo Fisher ESCALAB 250Xi by using an Al K $\alpha$  ( $\lambda = 8\text{\AA}$ ,  $h\nu = 1486.6\text{ eV}$ ) X-ray source without any etching. The morphology of the hybrid materials was characterized by Field emission scanning electron microscope (FESEM) on a SU-8010 operating and high resolution transmission electron microscope (HRTEM) on a FEI Tecnai F20 operating, respectively. UV-vis spectra are recorded on UV-2700 spectrophotometer from SHIMADZU. The excitation and emission spectra were collected by FS5 spectrofluorometer from Edinburg Photonics. The emission quantum yields in nanoparticles were measured on FS5 with a SC-30 Integrating Sphere.

## 2. Synthesis and Characterization

### 2.1 Synthetic procedures

The ligand (H<sub>4</sub>L) and (Et<sub>4</sub>N)<sub>24</sub>Eu<sub>8</sub>L<sub>12</sub> complex were synthesized according to the

reported method from our previous literature.<sup>[S1]</sup>

**Synthesis of MSNs** MSNs nanoparticles were synthesized by a modified method according to previous reports. Spherical MSNs with uniform diameter were prepared by base catalytic condensation. Briefly, CTAB (1.2 g, 3.3 mmol) and TEOA (3.3 mL, 0.27 mmol) were firstly added into 100 mL of deionized water and heated at 60 °C for 1 hour at a magnetic stirring rate of about 900 rpm. Then TEOS (8.4 mL) was added to the above reaction system drop by drop. Subsequently, the temperature was increased from 60 to 80 °C and stirring continued at 80 °C for 4 h. After centrifugation at 11000 rpm for 10 min, the crude products were further refluxed in the ethanol solution of sodium chloride (5 M) at 80 °C overnight to remove the template of CTAB. Repeat this procedure for three times. After centrifuging and washing with deionized water for three times, white powdery MSNs nanoparticles were obtained by freeze-drying.

**Synthesis of Eu<sub>8</sub>L<sub>12</sub>@MSNs** Mesoporous silica nanoparticles (100 mg) were suspended in cyclohexane (100 mL) and deionized water (20 mL), to which the DMSO solution of Eu(OTf)<sub>3</sub> (60 mg·mL<sup>-1</sup>, 1 mL, 0.10 mmol) was added and vigorously stirred for 24 h at room temperature. Then, the fresh DMSO solution of (Et<sub>4</sub>N)<sub>4</sub>L (mixing H<sub>4</sub>L (20 mg, 0.03 mmol) with (Et<sub>4</sub>N)<sub>4</sub>·OH (25 wt.% in H<sub>2</sub>O, 70 μL, 0.12 mmol) was added and further stirred at 50 °C for 24 h. Afterwards, the suspension was centrifuged and washed with deionized water for three times. After freeze-drying, the hybrid product Eu<sub>8</sub>L<sub>12</sub>@MSNs was obtained as a white powder.

**Synthesis of Eu<sub>8</sub>L<sub>12</sub>@MSNs-NH<sub>2</sub>** TEOS (100 μL) and APS (100 μL) were added into the solution of Eu<sub>8</sub>L<sub>12</sub>@MSNs (50 mg) in ethanol (100 mL) with vigorous stirring at room temperature for 24 hours. Then, the mixture was washed for three times with deionized water and collected by centrifugation. Finally, the white powder of amino-functional hybrid material Eu<sub>8</sub>L<sub>12</sub>@MSNs-NH<sub>2</sub> was obtained by freeze drying.

**Synthesis of Eu<sub>8</sub>L<sub>12</sub>@MSNs-biotin** EDC·HCl (40 mg, 0.21 mmol) and NHS (40 mg, 0.35 mmol) were added into a solution of biotin (20 mg, 0.09 mmol) in

DMSO (4 mL) and the mixture was stirred at room temperature for 2 hours. Then the activated biotin N-hydroxysuccinimide ester was directly added into a dispersion solution of  $\text{Eu}_8\text{L}_{12}@MSNs-NH_2$  nanoparticles (50 mg) in deionized water (100 mL) and the mixture was vigorously stirred at room temperature for 24 hours. The resulting biotin-labeled nanoparticles of  $\text{Eu}_8\text{L}_{12}@MSNs$ -biotin were collected by centrifugating and following washing with deionized water for three times. A white powder was obtained after freeze-drying.

## 2.2 Characterization

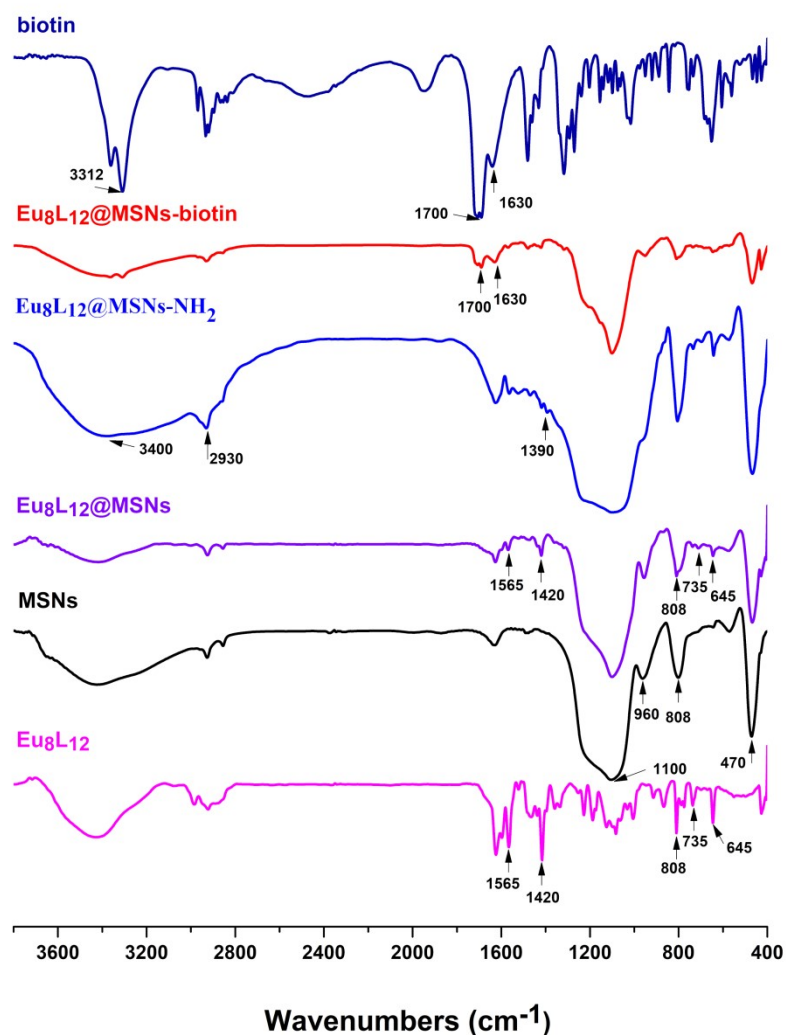


Figure S1. FT-IR spectra of  $\text{Eu}_8\text{L}_{12}$ , MSNs,  $\text{Eu}_8\text{L}_{12}@MSNs$ ,  $\text{Eu}_8\text{L}_{12}@MSNs-NH_2$ ,  $\text{Eu}_8\text{L}_{12}@MSNs$ -biotin, and biotin.

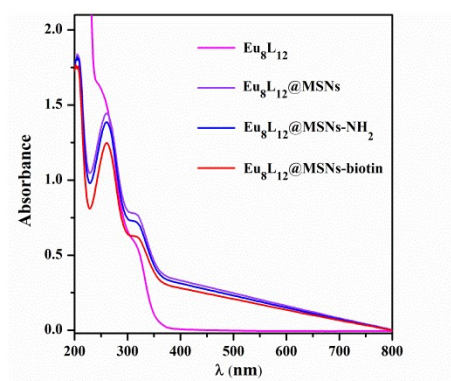


Figure S2. UV-vis absorption spectra of  $\text{Eu}_8\text{L}_{12}$  ( $1.667 \times 10^{-6}$  M),  $\text{Eu}_8\text{L}_{12}@\text{MSNs}$ ,  $\text{Eu}_8\text{L}_{12}@\text{MSNs-NH}_2$  and  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$  (0.125 mg/mL) in water.

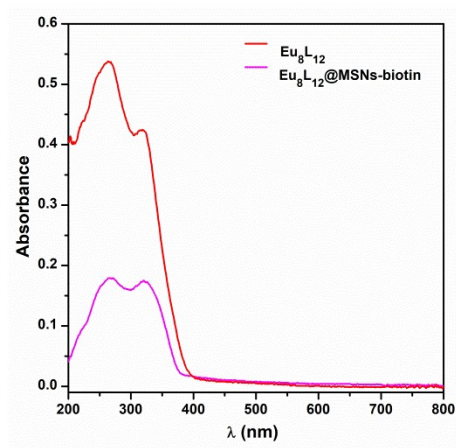


Figure S3. Solid state UV-vis spectra of  $\text{Eu}_8\text{L}_{12}$  and  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$ .

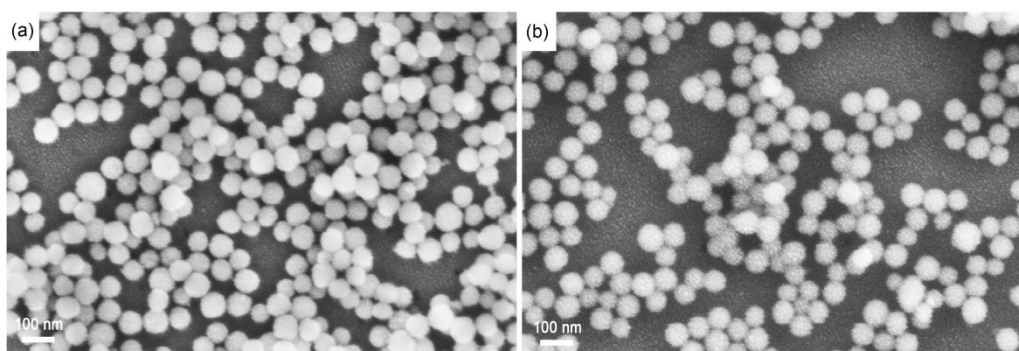


Figure S4. SEM of (a)  $\text{Eu}_8\text{L}_{12}@\text{MSNs}$  and (b)  $\text{Eu}_8\text{L}_{12}@\text{MSNs-NH}_2$ .

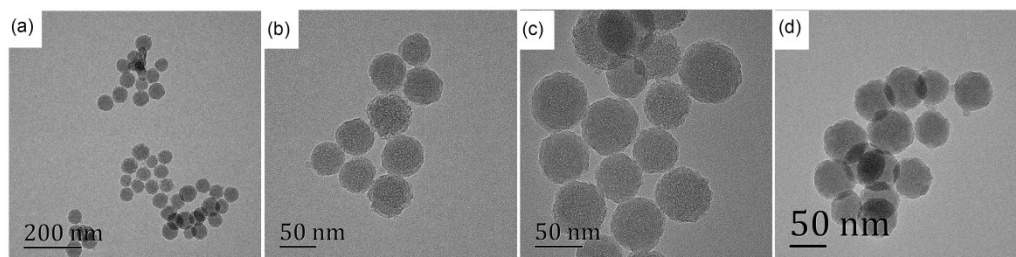


Figure S5. TEM of (a) MSNs, (b)  $\text{Eu}_8\text{L}_{12}@\text{MSNs}$ , (c)  $\text{Eu}_8\text{L}_{12}@\text{MSNs-NH}_2$  and (d)  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$ .

Eu<sub>8</sub>L<sub>12</sub>@MSNs-biotin.

### 2.3 Determination of the loading amount

The exact loading amounts of Eu<sub>8</sub>L<sub>12</sub> in hybrid materials were determined by an method of degradation.

To a fresh DMSO solution of (Et<sub>4</sub>N)<sub>24</sub>Eu<sub>8</sub>L<sub>12</sub> (2 mg/mL referred to H<sub>4</sub>L, 1 mL) was added KOH aqueous solution (1 M, 1 mL) and stirred at 50 °C for 24 h. This solution was diluted into different concentrations for UV-vis absorption testing. A standard calibration curve was plotted based on the UV-vis absorbance at 310 nm.

Similarly, 10 mg Eu<sub>8</sub>L<sub>12</sub>@MSNs was incubated in the mixture solution of DMSO (1 mL) and 1 M KOH aqueous solution (1 M, 1 mL) at 50°C with stirring for 24 h and then diluted to 0.125 mg · mL<sup>-1</sup> for UV-vis testing.

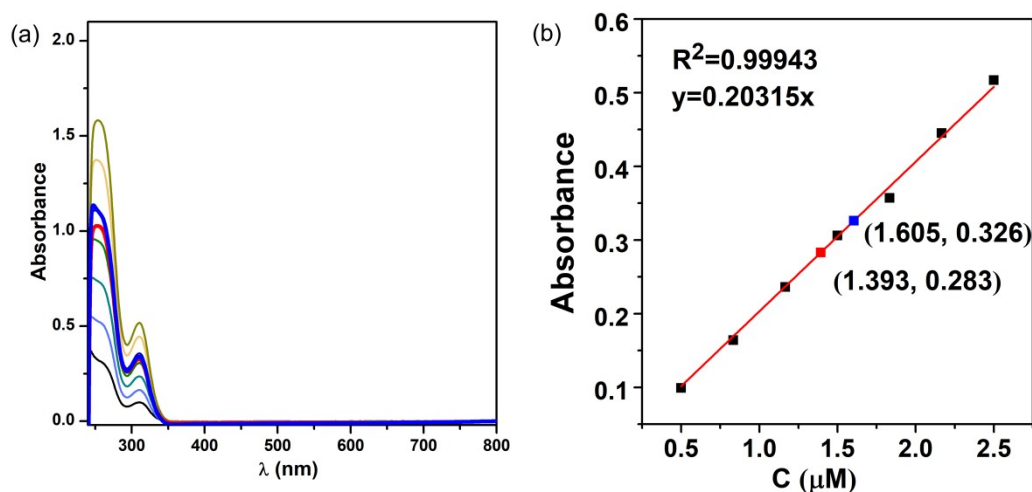


Figure S6. (a) UV-vis absorption spectra and of (Et<sub>4</sub>N)<sub>24</sub>Eu<sub>8</sub>L<sub>12</sub> after degradation at different concentrations; (b) the standard curve based on the absorption at 310 nm (DMSO/KOH aq. (1 M), v/v =1:1, 298 K). The UV-vis absorption spectra of Eu<sub>8</sub>L<sub>12</sub>@MSNs-NH<sub>2</sub> and Eu<sub>8</sub>L<sub>12</sub>@MSNs-biotin after degradation were highlighted as blue curve (square) and red curve (square), respectively.

Table S1. The loading amount of Eu<sub>8</sub>L<sub>12</sub> in different hybrid materials.

	A <sub>310</sub> <sup>a</sup>	C <sup>b</sup> (uM)	Loading (umol · g <sup>-1</sup> )
Eu <sub>8</sub> L <sub>12</sub> @MSNs	0.339	1.669	13.350
Eu <sub>8</sub> L <sub>12</sub> @MSNs-NH <sub>2</sub>	0.326	1.605	12.838
Eu <sub>8</sub> L <sub>12</sub> @MSNs-biotin	0.283	1.393	11.145

<sup>a</sup> The concentrations of the hybrid materials for UV-vis testing were 0.125 mg · mL<sup>-1</sup>.

<sup>b</sup> The concentrations of Eu<sub>8</sub>L<sub>12</sub> in UV-vis testing solution of different hybrid materials.

### 3. Photophysical property

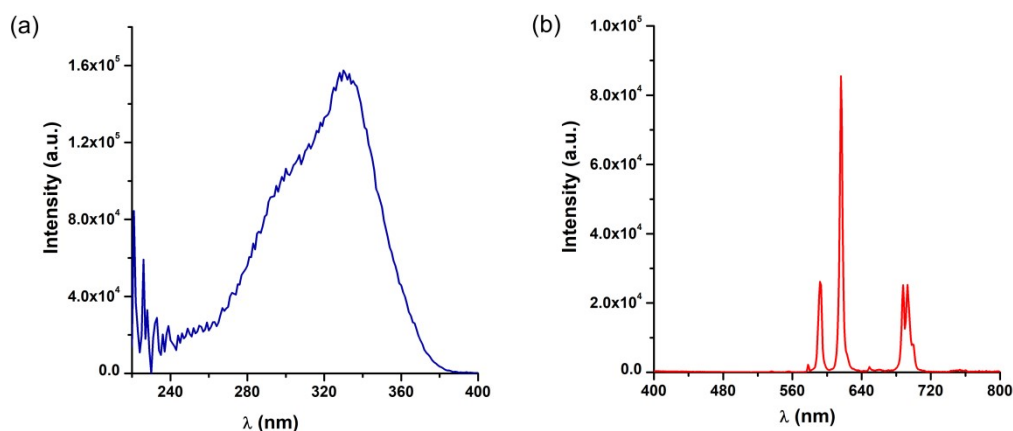


Figure S7. (a) Excitation ( $\lambda_{em} = 616$  nm, slits = 0.5 - 0.5) and (b) emission ( $\lambda_{ex} = 330$  nm, slits = 1.0 - 1.0) spectra of  $\text{Eu}_8\text{L}_{12}$  ( $c = 1.667 \times 10^{-6}$  M) in  $\text{H}_2\text{O}$  at 298 K.

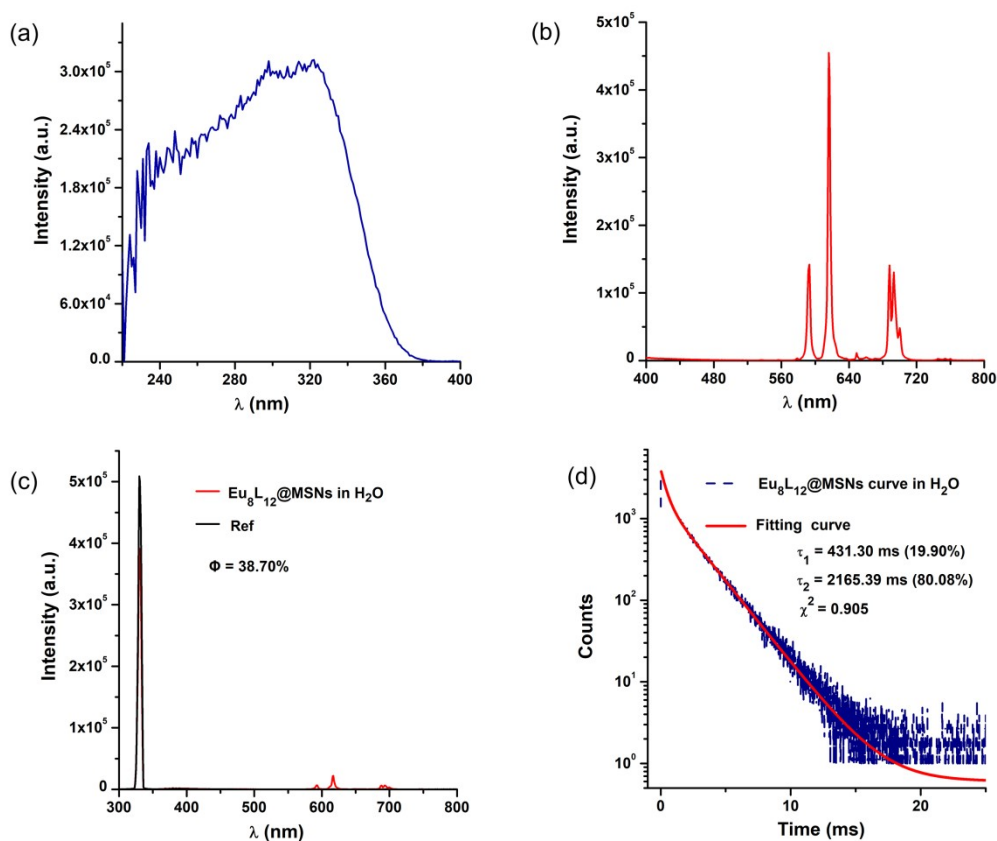


Figure S8. (a) Excitation spectrum ( $\lambda_{em} = 616$  nm, slits = 0.5 - 0.5), (b) emission spectrum ( $\lambda_{ex} = 330$  nm, slits=1.0 - 1.0), (c) quantum yield ( $\Phi_{overall} = 38.70\%$ ,  $\lambda_{ex} = 330$  nm, slits = 6.0 - 0.6), and (d) excited state decay curve ( $\lambda_{ex} = 330$  nm, slits = 2 - 0.5) of  $\text{Eu}_8\text{L}_{12}@MSNs$  ( $c = 0.125$  mg  $\cdot$  mL $^{-1}$ ) in  $\text{H}_2\text{O}$  at 298 K.



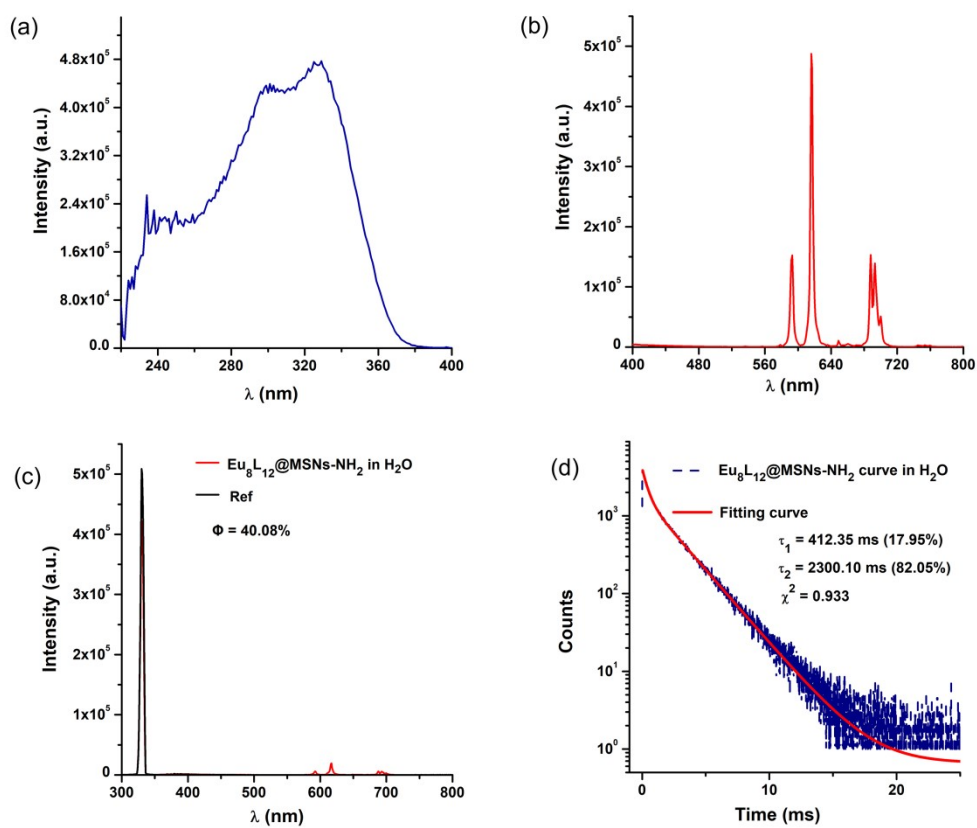


Figure S9. (a) Excitation spectrum ( $\lambda_{\text{em}} = 616 \text{ nm}$ , slits = 0.5 - 0.5), (b) emission spectrum ( $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits=1.0 - 1.0), (c) quantum yield ( $\Phi_{\text{overall}} = 40.08 \%$ ,  $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits = 6.0 - 0.6), and (d) excited state decay curve ( $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits = 2 - 0.5) of  $\text{Eu}_8\text{L}_{12}@MSNs\text{-NH}_2$  ( $c = 0.130 \text{ mg} \cdot \text{mL}^{-1}$ ) in  $\text{H}_2\text{O}$  at 298 K.

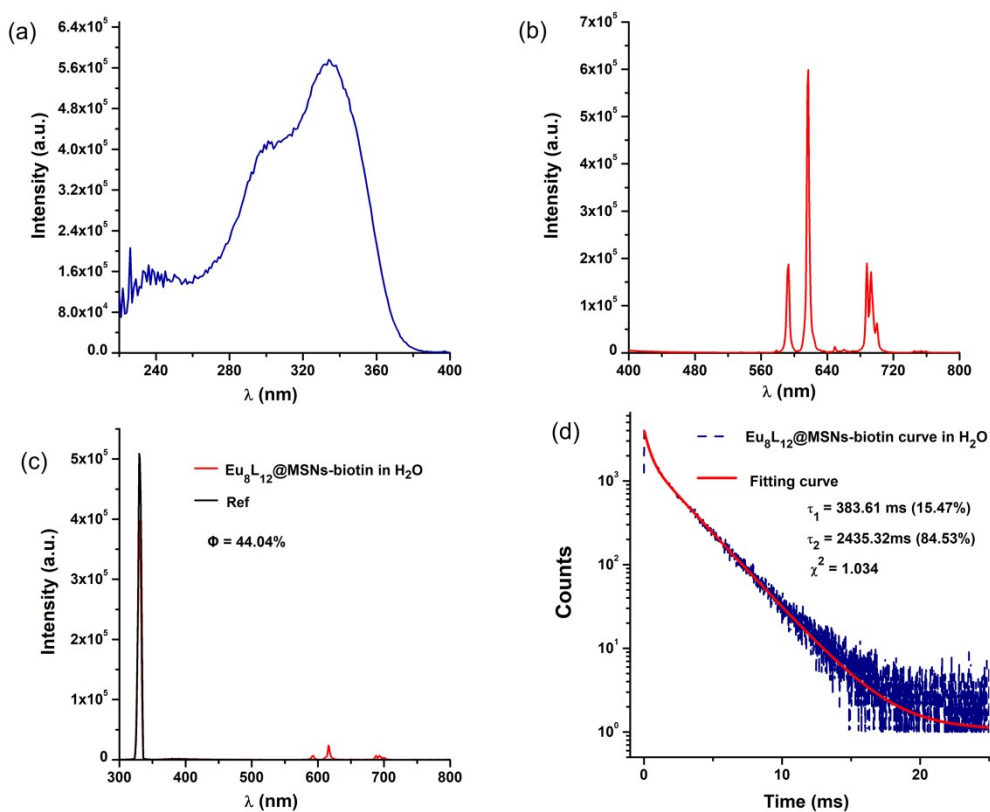


Figure S10. (a) Excitation spectrum ( $\lambda_{\text{em}} = 616 \text{ nm}$ , slits = 0.5 - 0.5), (b) emission spectrum ( $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits=1.0 - 1.0), (c) quantum yield ( $\Phi_{\text{overall}} = 44.04 \%$ ,  $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits = 6.0 - 0.6), and (d) excited state decay curve ( $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits = 2 - 0.5) of  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  ( $c = 0.150 \text{ mg} \cdot \text{mL}^{-1}$ ) in  $\text{H}_2\text{O}$  at 298 K.

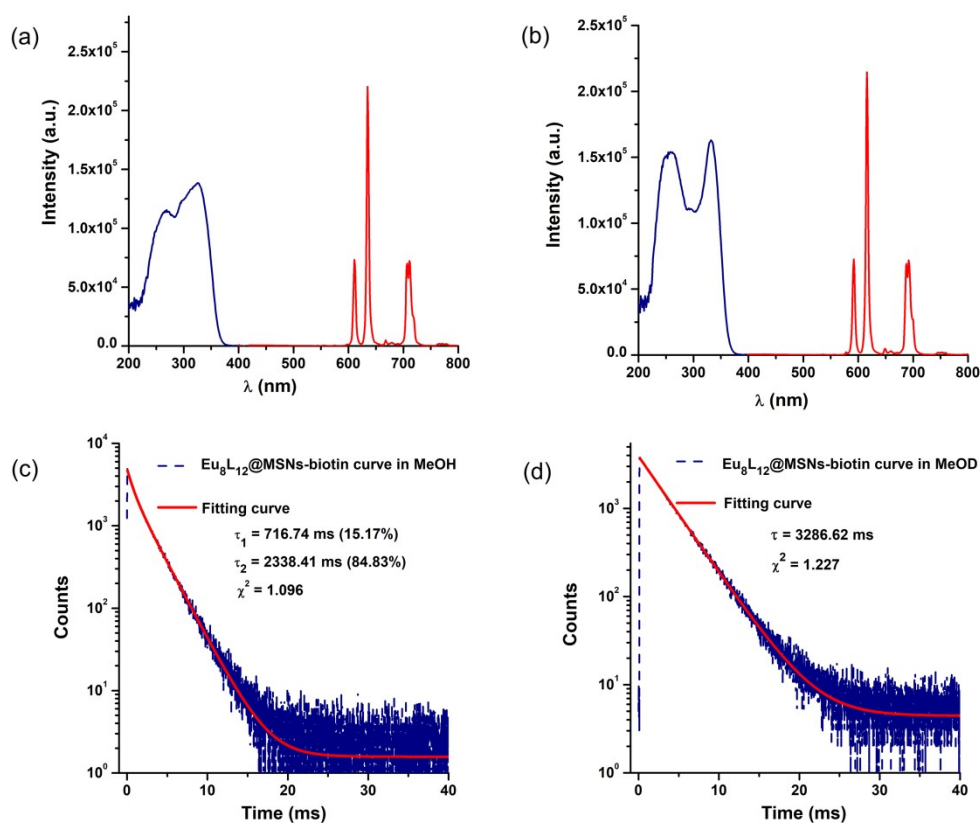


Figure S11. Excitation (blue lines,  $\lambda_{em} = 616$  nm, slits = 2.5 - 3) and emission (red lines,  $\lambda_{ex} = 330$  nm, slits = 3.5 - 4) spectra of  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  ( $c = 0.3$  mg  $\cdot$  mL $^{-1}$ ) in MeOH (a) and in MeOD (b); excited state decay curve for  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  in MeOH (c) and MeOD (d) ( $\lambda_{ex} = 330$  nm, slits = 3 - 3) at 298 K.

#### 4. The stability of hybrid materials

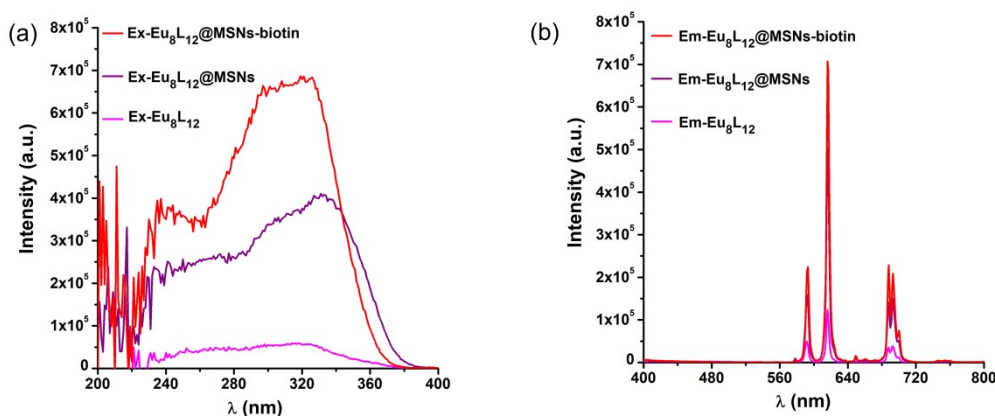


Figure S12. (a) Excitation ( $\lambda_{em} = 616$  nm, slits = 0.5 - 0.8) and (b) emission ( $\lambda_{ex} = 330$  nm, slits = 1.0 - 1.5) spectra of  $\text{Eu}_8\text{L}_{12}$  ( $c = 4.167$   $\mu\text{M}$ ),  $\text{Eu}_8\text{L}_{12}@MSNs$  ( $c = 0.3$  mg  $\cdot$  mL $^{-1}$ ) and  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  ( $c = 0.3$  mg  $\cdot$  mL $^{-1}$ ) in HCl (pH = 4) at 298 K.

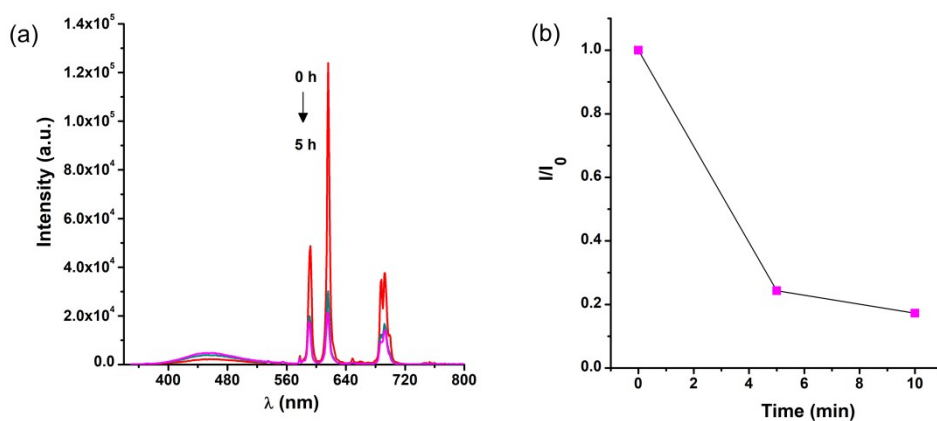


Figure S13. (a) Emission ( $\lambda_{\text{ex}}=330$  nm, slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in HCl (pH = 4) at 298 K.

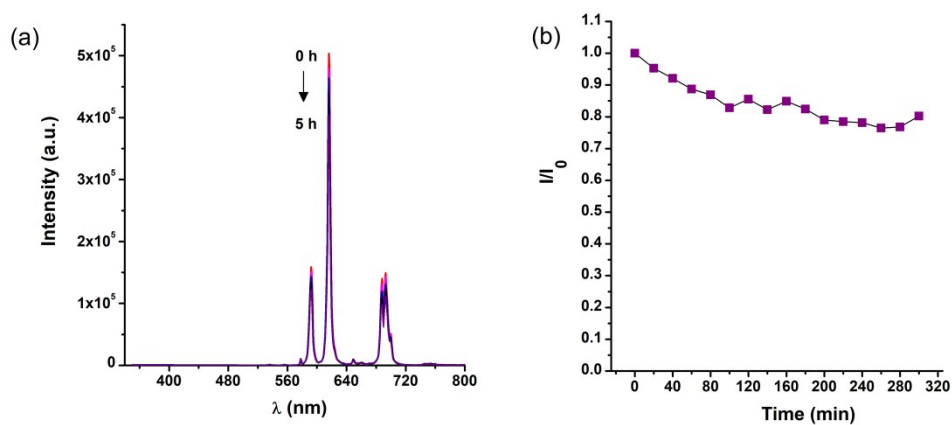


Figure S14. (a) Emission ( $\lambda_{\text{ex}}=330$  nm, slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}@MSNs$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in HCl (pH = 4) at 298 K.

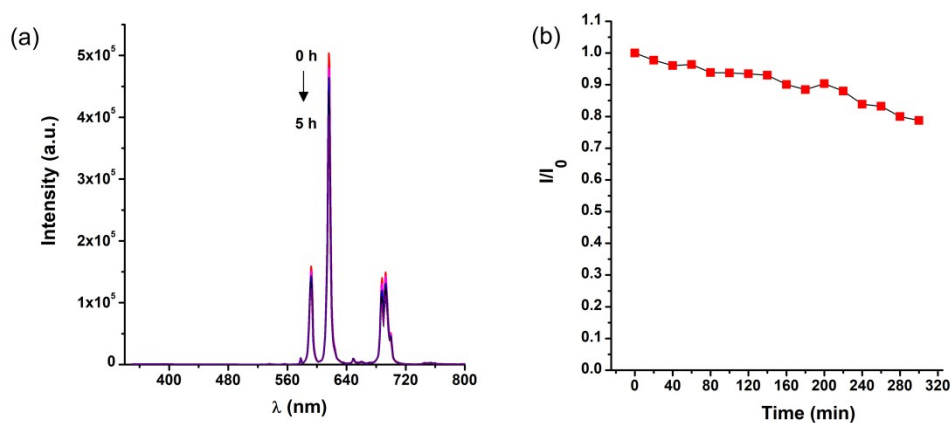


Figure S15. (a) Emission ( $\lambda_{\text{ex}}=330$  nm, slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in HCl (pH = 4) at 298 K.

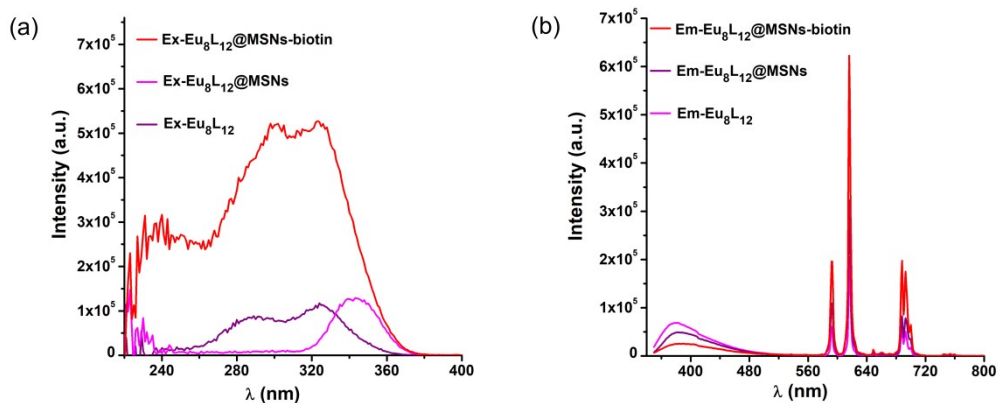


Figure S16. (a) Excitation ( $\lambda_{\text{em}} = 616 \text{ nm}$ , slits = 0.5 - 0.8) and (b) emission ( $\lambda_{\text{ex}} = 330 \text{ nm}$ , slits=1.0 - 1.5) spectra of  $\text{Eu}_8\text{L}_{12}$  ( $c = 4.167 \mu\text{M}$ ),  $\text{Eu}_8\text{L}_{12}@MSNs$  ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) and  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in PBS ( $\text{pH} = 7.4$ , 10 mM  $\text{Na}_2\text{HPO}_4$ , 2 mM  $\text{KH}_2\text{PO}_4$ ) at 298 K.

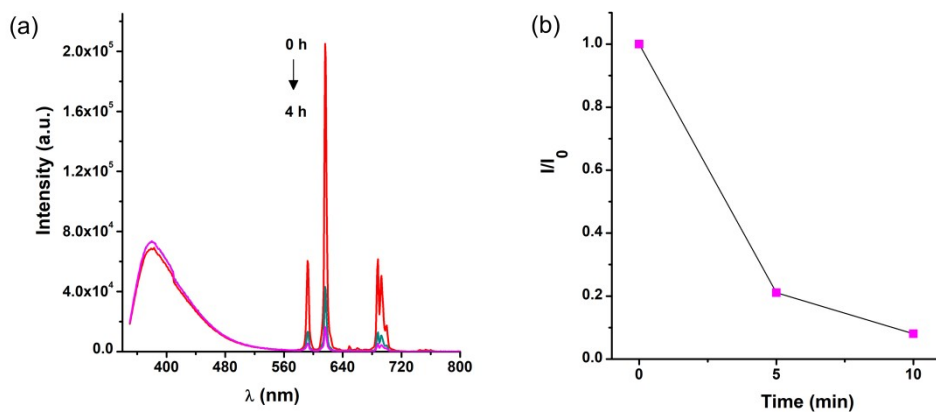


Figure S17. (a) Emission ( $\lambda_{\text{ex}}=330 \text{ nm}$ , slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in PBS ( $\text{pH} = 7.4$ , 10 mM  $\text{Na}_2\text{HPO}_4$ , 2mM  $\text{KH}_2\text{PO}_4$ ) at 298 K.

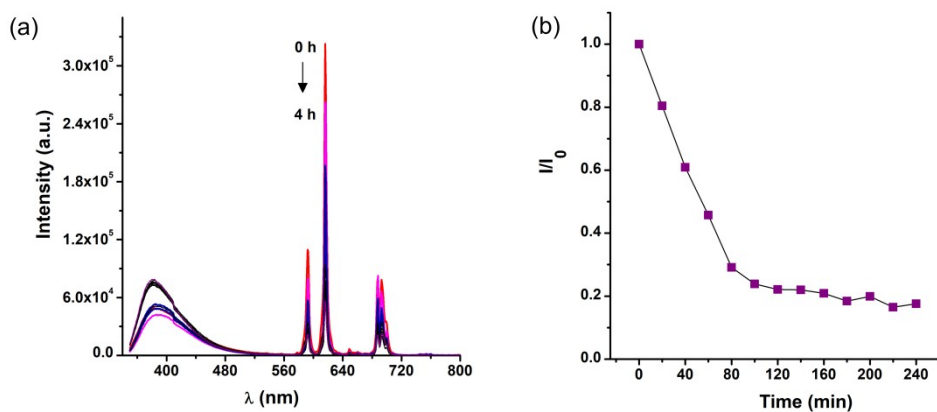


Figure S18. (a) Emission ( $\lambda_{\text{ex}}=330$  nm, slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}@MSNs$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in PBS (pH = 7.4, 10 mM  $\text{Na}_2\text{HPO}_4$ , 2mM  $\text{KH}_2\text{PO}_4$ ) at 298 K.

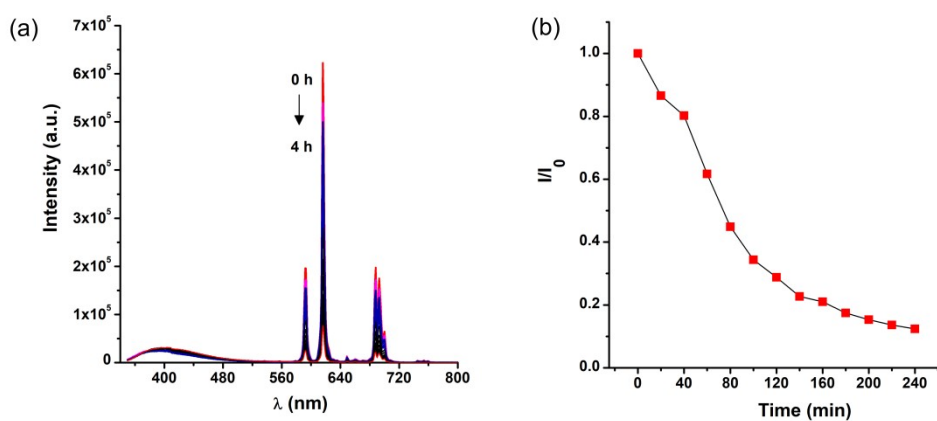


Figure S19. (a) Emission ( $\lambda_{\text{ex}}=330$  nm, slits = 1.0 – 1.5) spectra and (b) emission attenuation of  $\text{Eu}_8\text{L}_{12}@MSNs\text{-biotin}$  at 616 nm ( $c = 0.3 \text{ mg} \cdot \text{mL}^{-1}$ ) in PBS (pH = 7.4, 10 mM  $\text{Na}_2\text{HPO}_4$ , 2mM  $\text{KH}_2\text{PO}_4$ ) at 298 K.

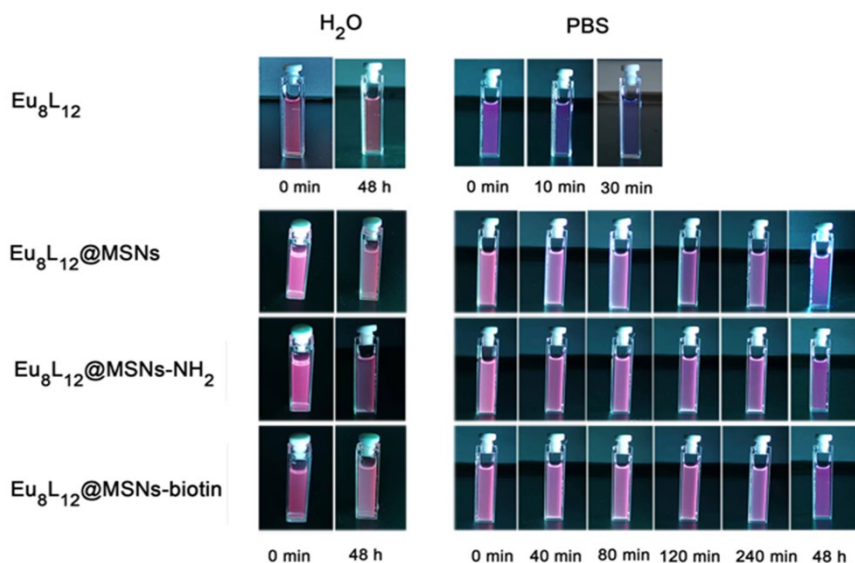


Figure S20. Digital photographs of (1.667  $\mu\text{M}$ ),  $\text{Eu}_8\text{L}_{12}@MSNs$  (0.125  $\text{mg} \cdot \text{mL}^{-1}$ ),  $\text{Eu}_8\text{L}_{12}@MSNs-NH_2$  (0.130  $\text{mg} \cdot \text{mL}^{-1}$ ) and  $\text{Eu}_8\text{L}_{12}@MSNs-biotin$  (0.150  $\text{mg} \cdot \text{mL}^{-1}$ ) dispersed in water and PBS (pH = 7.4, 10 mM  $\text{Na}_2\text{HPO}_4$ , 2 mM  $\text{KH}_2\text{PO}_4$ ) for varied duration time under the irradiation of ultraviolet lamp at 365 nm.

## 5. Tumor-Targeted Imaging

The cell viabilities of MDA-MB-231 human breast cancer cells treated with different concentrations of  $\text{Eu}_8\text{L}_{12}$ , MSNs,  $\text{Eu}_8\text{L}_{12}@MSNs$ ,  $\text{Eu}_8\text{L}_{12}@MSNs-NH_2$ , and  $\text{Eu}_8\text{L}_{12}@MSNs-biotin$  for 72 h have been evaluated by MTT assay.

Table S2.  $\text{IC}_{50}$  value of the hybrid materials towards MDA-MB-231 cancer cells.

	cisplatin	$\text{Eu}_8\text{L}_{12}$	$\text{Eu}_8\text{L}_{12}@SiO_2$	$\text{Eu}_8\text{L}_{12}@SiO_2-APS$	$\text{Eu}_8\text{L}_{12}@SiO_2-Biotin$
$\text{IC}_{50}(\mu\text{g} \cdot \text{mL}^{-1})$	1.4	>80	21.7	45.1	24.9

MDA-MB-231 human breast cancer cells were used to evaluate the living cell imaging of the hybrid materials. Briefly, MDA-MB-231 human breast cancer cells were seeded in 2 cm petri dish and attached for 24 h. Then the cells were incubated with different concentration of hybrid materials for 24 h, and captured with fluorescence microscope (Cytation 5, BioTek Instruments, Inc.).

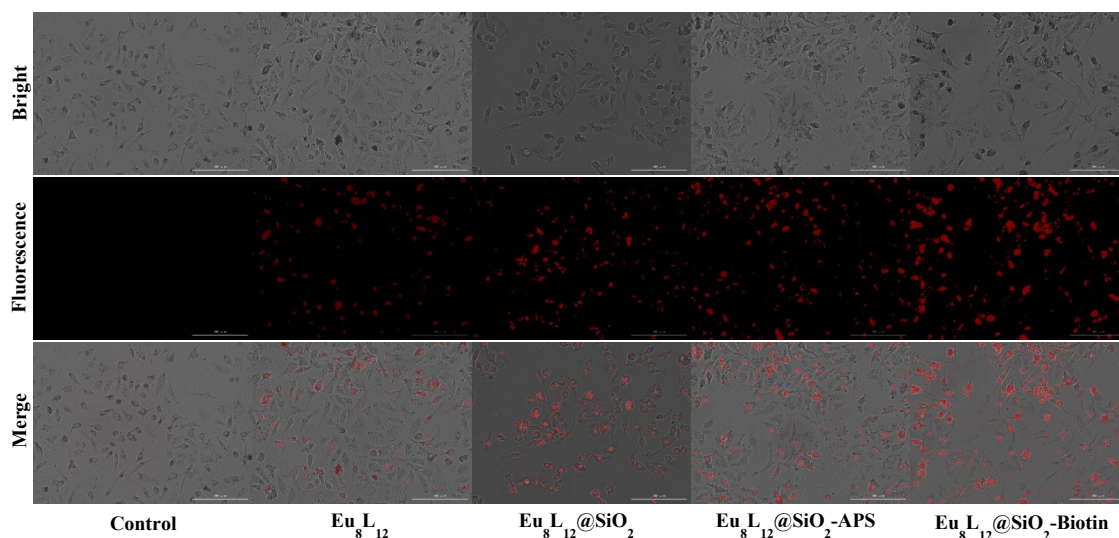


Figure S21. Confocal microscope images of MDA-MB-231 human breast cancer cells co-incubated with  $\text{Eu}_8\text{L}_{12}$ ,  $\text{Eu}_8\text{L}_{12}@\text{MSNs}$ ,  $\text{Eu}_8\text{L}_{12}@\text{MSNs-NH}_2$  and  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$  ( $5 \mu\text{M}$  referred to the complex), respectively.

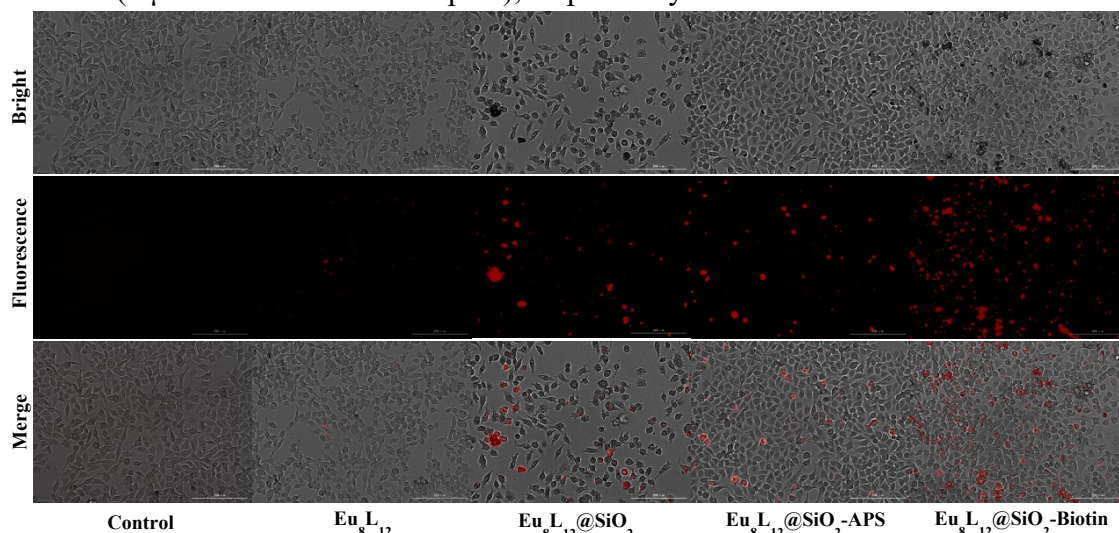


Figure S22. Confocal microscope images of MDA-MB-231 human breast cancer cells co-incubated with  $\text{Eu}_8\text{L}_{12}$ ,  $\text{Eu}_8\text{L}_{12}@\text{MSNs}$ ,  $\text{Eu}_8\text{L}_{12}@\text{MSNs-NH}_2$  and  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$  ( $0.4 \mu\text{M}$  referred to the complex), respectively.

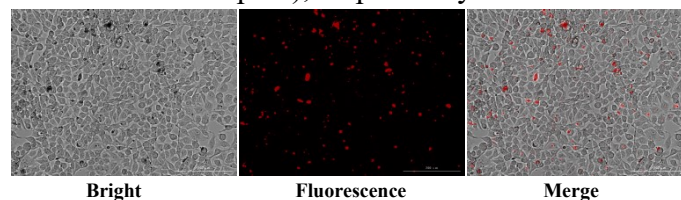


Figure S23. Confocal microscope images of MDA-MB-231 human breast cancer cells co-incubated with  $\text{Eu}_8\text{L}_{12}@\text{MSNs-biotin}$  ( $0.2 \mu\text{M}$  referred to the complex).



Table S3. Intracellular uptake percentages of Eu<sub>8</sub>L<sub>12</sub>@MSNs-biotin in MDA-MB-231 cancer cells and NIH 3T3 mouse fibroblasts cell.<sup>a</sup>

	MDA-MB-231	NIH 3T3
Intracellular uptake (%)	41	32

<sup>a</sup> MDA-MB-231 human breast cancer cells and NIH 3T3 mouse fibroblasts cells were incubated for 4 h in the presence of Eu<sub>8</sub>L<sub>12</sub>@MSNs-biotin (2.0 μM), respectively.

## 6. References

- [S1] Z. Wang, L. Z. He, B. Q. Liu, L.-P. Zhou, L.-X. Cai, S.-J. Hu, X.-Z. Li, Z. Li, T. F. Chen, X. P. Li, and Q.-F. Sun, *J. Am. Chem. Soc.* **2020**, *142*, 16409.