

- 1 **A Computationally Efficient Model to Represent the Chemistry,**
2 **Thermodynamics, and Microphysics of Secondary Organic Aerosol**
3 **(simpleSOM): Model Development and Application to α -pinene SOA**
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16 *Table S.1: SOA mass concentration and O:C observations from several environmental chamber studies*
17 *performed on α -pinene.*

Reference	O:C Range	Oxidant	Max SOA ($\mu\text{g m}^{-3}$)
Aiken et al. ¹	0.28	O_3	~500
Shilling et al. ²	0.29 to 0.45	O_3	0.5 to >140
Chhabra et al. ³	0.30 to 0.43	O_3	57 to 183
Zhang et al. ⁴	0.45 to 0.55	O_3/OH	125 to 250
Järvinen et al. ⁵	0.23 to 0.29	O_3/OH	>600
Nah et al. ⁶	0.45 to 0.52	O_3	62 to 87
Kim et al. ⁷	0.33 to 0.42	O_3/OH	20 to 255
Heaton et al. ⁸	0.31 to 0.37	O_3	~400

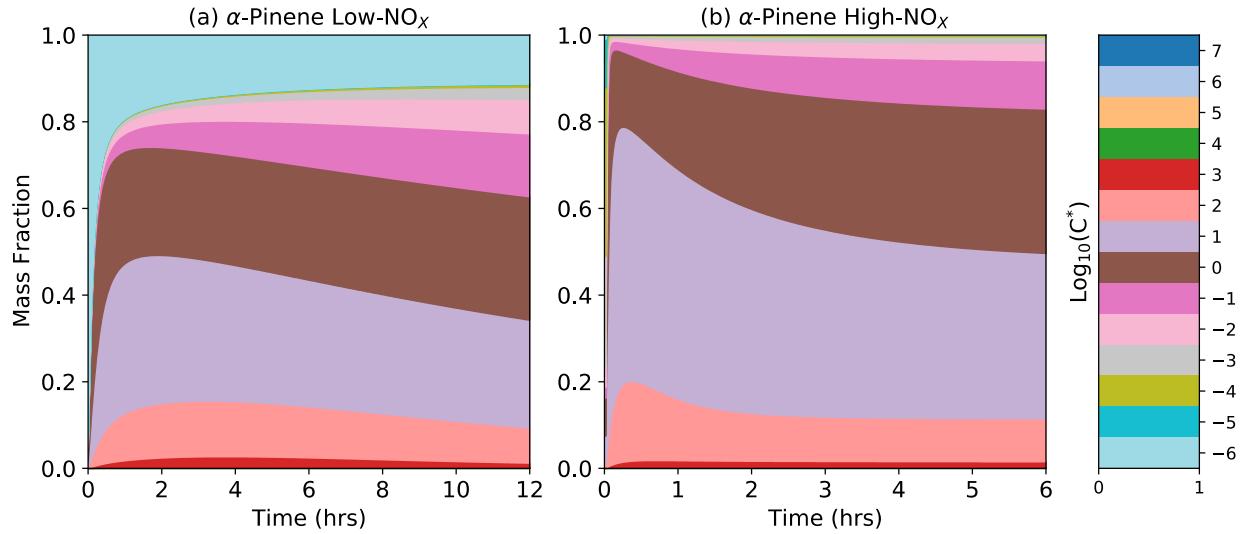
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20 *Table S.2: simpleSOM parameters to model SOA formation from photooxidation of α -pinene for four*
21 *target end-of-experiment oligomer fractions, $f_{\text{olig}}=0, 20\%, 50\%, \text{ and } 80\%$.*

f_{olig}	$k_f (\text{cm}^3 \text{molecules}^{-1} \text{s}^{-1})$	$k_r (\text{s}^{-1})$	m_{frag}	P_{loss}	$\Delta \log c^*$	P_{o1}	P_{o2}	P_{o3}	P_{o4}	P_{ELVOC}
0.0	NA	NA	3.513	0.989	1.630	0.001	0.704	0.260	0.001	0.034
0.2	10^{-24}	1.5×10^{-2}	3.651	0.961	2.198	0.001	0.897	0.067	0.001	0.034
0.5	10^{-24}	2.4×10^{-3}	4.121	0.990	2.785	0.429	0.465	0.071	0.001	0.034
0.8	10^{-24}	2.5×10^{-4}	5.240	0.990	3.140	0.694	0.180	0.091	0.001	0.034

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23 *Table S.3: simpleSOM parameters to model SOA formation from photooxidation of α -pinene for three*
24 *different D_b values: $1 \times 10^{-10}, 3 \times 10^{-19}, \text{ and } 1 \times 10^{-21} \text{ m}^2 \text{s}^{-1}$.*

$D_b (\text{m}^2 \text{s}^{-1})$	m_{frag}	P_{loss}	$\Delta \log c^*$	P_{o1}	P_{o2}	P_{o3}	P_{o4}	P_{ELVOC}
1×10^{-10}	3.513	0.989	1.630	0.001	0.704	0.260	0.001	0.034
3×10^{-19}	3.673	0.976	1.679	0.001	0.719	0.246	0.001	0.034
1×10^{-21}	2.753	0.000	6.936	0.001	0.024	0.941	0.001	0.034

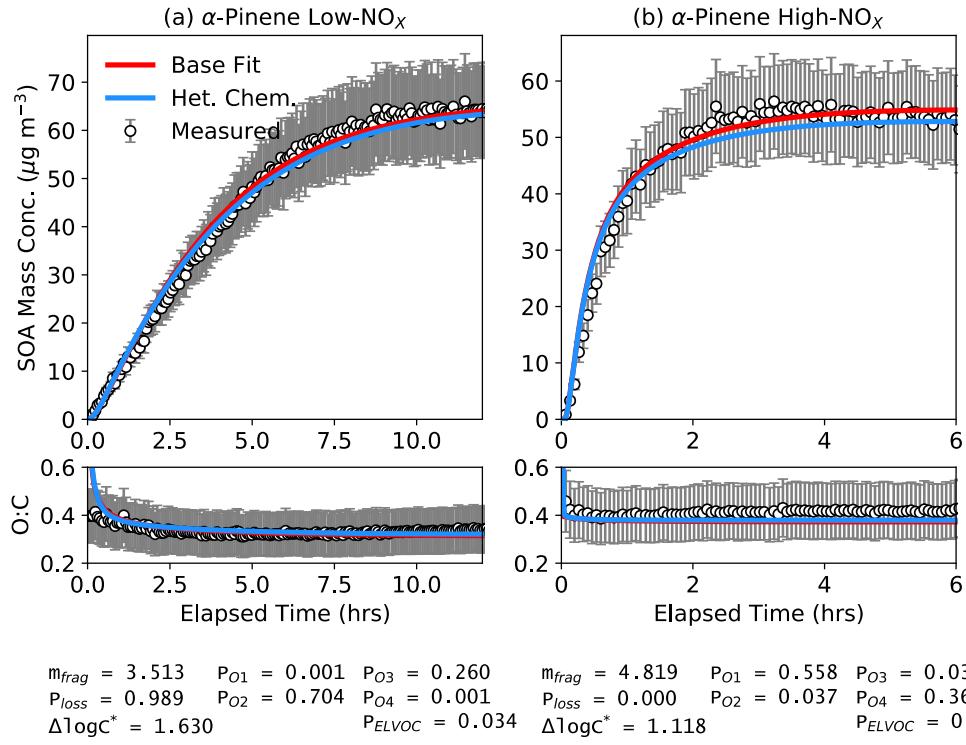
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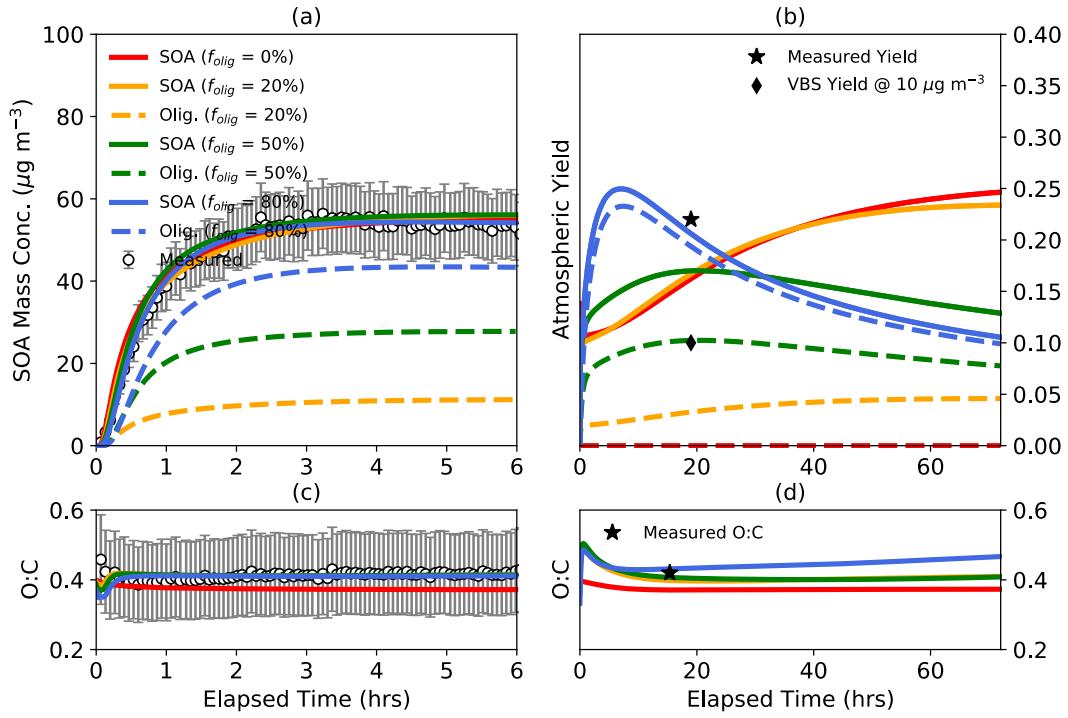
27 *Figure S.1: Normalized, c^* -resolved contributions to SOA over time for simulations performed at (a) low*
 28 *NO_x and (b) high NO_x conditions. These results are from the same simulations shown in Figure 1. Lower-*
 29 *volatility species seem to contribute more strongly to the SOA under the low NO_x case compared to the*
 30 *high NO_x case during the early parts of the experiment.*

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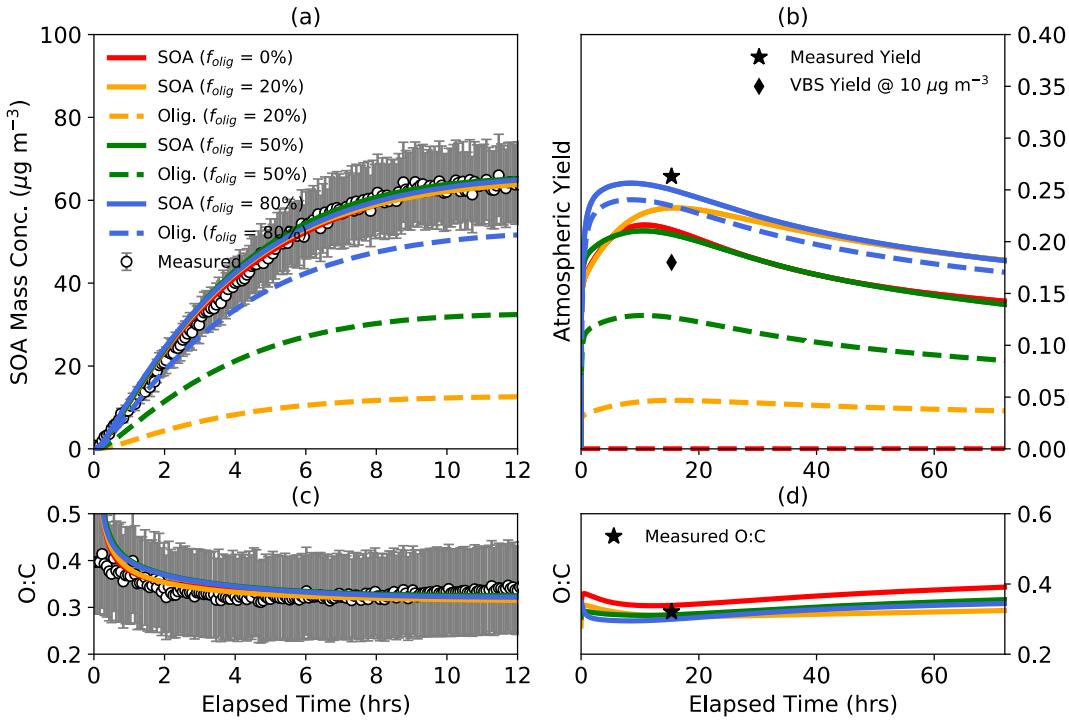


$m_{frag} = 3.513$	$P_{O1} = 0.001$	$P_{O3} = 0.260$	$m_{frag} = 4.819$	$P_{O1} = 0.558$	$P_{O3} = 0.037$
$P_{loss} = 0.989$	$P_{O2} = 0.704$	$P_{O4} = 0.001$	$P_{loss} = 0.000$	$P_{O2} = 0.037$	$P_{O4} = 0.368$
$\Delta \log C^* = 1.630$		$P_{ELVOC} = 0.034$	$\Delta \log C^* = 1.118$		$P_{ELVOC} = 0.000$

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33 *Figure S.2: simpleSOM predictions of SOA mass concentration and SOA O:C ratio compared to*
34 *measurements for (a) low and (b) high NO_x photooxidation experiments performed on α -pinene. Model*
35 *predictions based on fits to the SOA mass concentration and O:C are shown in solid red while those for*
36 *simulations using the base fit parameters but with heterogeneous chemistry included are shown in solid*
37 *blue. The fit parameters for the respective NO_x conditions are listed at the bottom of the figure.*

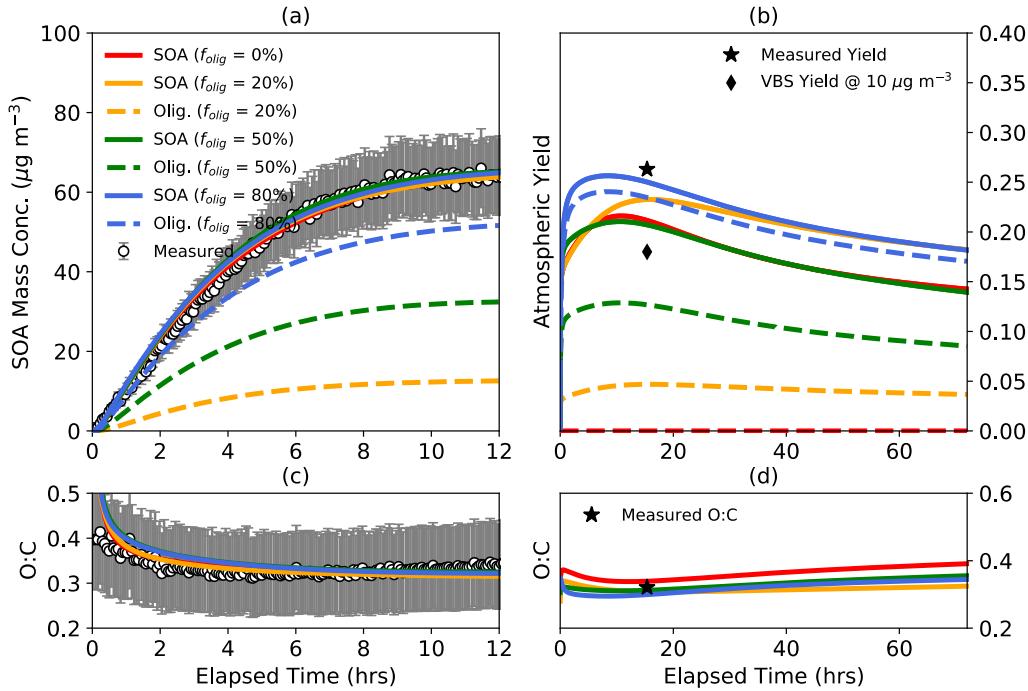


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40 *Figure S.3: Same as Figure 3 but for a high NO_x experiment. simpleSOM predictions of (a) SOA mass*
41 *concentration and (c) SOA O:C ratio based on fits to the observations compared to measurements for a*
42 *low NO_x photooxidation experiment performed on α -pinene for different target end-of-experiment*
43 *oligomer fractions. simpleSOM predictions of (b) SOA mass yields and (d) SOA O:C ratio from*
44 *atmospheric simulations performed under low NO_x conditions. Predictions of total SOA mass are shown*
45 *in solid lines and the oligomer mass are shown in dashed lines.*



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47 *Figure S.4: Same as Figure 3 but for a k_f value of $10^{-25} \text{ cm}^3 \text{ molecules}^{-1} \text{ s}^{-1}$. simpleSOM predictions of (a)*
48 *SOA mass concentration and (c) SOA O:C ratio based on fits to the observations compared to*
49 *measurements for a low NO_x photooxidation experiment performed on α -pinene for different target end-*
50 *of-experiment oligomer fractions. simpleSOM predictions of (b) SOA mass yields and (d) SOA O:C ratio*
51 *from atmospheric simulations performed under low NO_x conditions. Predictions of total SOA mass are*
52 *shown in solid lines and the oligomer mass are shown in dashed lines.*

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56 *Figure S.5: Same as Figure 3 but for a k_f value of $10^{-23} \text{ cm}^3 \text{ molecules}^{-1} \text{ s}^{-1}$. simpleSOM predictions of (a)*
57 *SOA mass concentration and (c) SOA O:C ratio based on fits to the observations compared to*
58 *measurements for a low NO_x photooxidation experiment performed on α-pinene for different target end-*
59 *of-experiment oligomer fractions. simpleSOM predictions of (b) SOA mass yields and (d) SOA O:C ratio*
60 *from atmospheric simulations performed under low NO_x conditions. Predictions of total SOA mass are*
61 *shown in solid lines and the oligomer mass are shown in dashed lines.*

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74 **References**

- 75 (1) Aiken, A. C.; Decarlo, P. F.; Kroll, J. H.; Worsnop, D. R.; Huffman, J. A.; Docherty, K. S.; Ulbrich,
76 I. M.; Mohr, C.; Kimmel, J. R.; Sueper, D.; Sun, Y.; Zhang, Q.; Trimborn, A.; Northway, M.;
77 Ziemann, P. J.; Canagaratna, M. R.; Onasch, T. B.; Alfarra, M. R.; Prevot, A. S. H.; Dommen, J.;
78 Duplissy, J.; Metzger, A.; Baltensperger, U.; Jimenez, J. L. O/C and OM/OC Ratios of Primary,
79 Secondary, and Ambient Organic Aerosols with High-Resolution Time-of-Flight Aerosol Mass
80 Spectrometry. *Environ. Sci. Technol.* **2008**, *42* (12), 4478–4485.
- 81 (2) Shilling, J. E.; Chen, Q.; King, S. M.; Rosenoern, T.; Kroll, J. H.; Worsnop, D. R.; DeCarlo, P. F.;
82 Aiken, A. C.; Sueper, D.; Jimenez, J. L.; Martin, S. T. Loading-Dependent Elemental Composition
83 of α -Pinene SOA Particles. *Atmos. Chem. Phys.* **2008**, *8* (4), 15343–15373.
- 84 (3) Chhabra, P. S.; Flagan, R. C.; Seinfeld, J. H. Elemental Analysis of Chamber Organic Aerosol Using
85 an Aerodyne High-Resolution Aerosol Mass Spectrometer. *Atmos. Chem. Phys.* **2010**, *10* (9), 4111–
86 4131.
- 87 (4) Zhang, X.; McVay, R. C.; Huang, D. D.; Dalleska, N. F.; Aumont, B.; Flagan, R. C.; Seinfeld, J. H.
88 Formation and Evolution of Molecular Products in α -Pinene Secondary Organic Aerosol. *Proc. Natl.
89 Acad. Sci. U. S. A.* **2015**, *112* (46), 14168–14173.
- 90 (5) Järvinen, E.; Ignatius, K.; Nichman, L.; Kristensen, T. B.; Fuchs, C.; Hoyle, C. R.; Höppel, N.;
91 Corbin, J. C.; Craven, J.; Duplissy, J.; Ehrhart, S.; El Haddad, I.; Frege, C.; Gordon, H.; Jokinen, T.;
92 Kallinger, P.; Kirkby, J.; Kiselev, A.; Naumann, K.-H.; Petäjä, T.; Pinterich, T.; Prevot, A. S. H.;
93 Saathoff, H.; Schiebel, T.; Sengupta, K.; Simon, M.; Slowik, J. G.; Tröstl, J.; Virtanen, A.;
94 Vochezer, P.; Vogt, S.; Wagner, A. C.; Wagner, R.; Williamson, C.; Winkler, P. M.; Yan, C.;
95 Baltensperger, U.; Donahue, N. M.; Flagan, R. C.; Gallagher, M.; Hansel, A.; Kulmala, M.;
96 Stratmann, F.; Worsnop, D. R.; Möhler, O.; Leisner, T.; Schnaiter, M. Observation of Viscosity
97 Transition in α -Pinene Secondary Organic Aerosol. *Atmos. Chem. Phys.* **2016**, *16* (7), 4423–4438.
- 98 (6) Nah, T.; McVay, R. C.; Zhang, X.; Boyd, C. M.; Seinfeld, J. H.; Ng, N. L. Influence of Seed Aerosol
99 Surface Area and Oxidation Rate on Vapor Wall Deposition and SOA Mass Yields: A Case Study
100 with α -Pinene Ozonolysis. *Atmos. Chem. Phys.* **2016**, *16* (14), 9361–9379.
- 101 (7) Kim, H.; Liu, S.; Russell, L. M.; Paulson, S. E. Dependence of Real Refractive Indices on O:C, H:C
102 and Mass Fragments of Secondary Organic Aerosol Generated from Ozonolysis and Photooxidation
103 of Limonene and α -Pinene. *Aerosol Sci. Technol.* **2014**, *48* (5), 498–507.
- 104 (8) Heaton, K. J.; Sleighter, R. L.; Hatcher, P. G.; Hall, W. A., 4th; Johnston, M. V. Composition
105 Domains in Monoterpene Secondary Organic Aerosol. *Environ. Sci. Technol.* **2009**, *43* (20), 7797–
106 7802.