1 Tremendously Enhanced Photocurrent Enabled by Triplet-

2 Triplet Annihilation Up-conversion for High-Performance

3 Perovskite Solar Cells

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15 Materials: All chemicals were used directly without any treatment after purchase. Chlorobenzene (CB, anhydrous, 99.8%), dimethyl sulfoxide (DMSO, anhydrous, 16 ≥99.9%), N, N-dimethylformamide (DMF, anhydrous, 99.8%), 4-tert-butylpyridine 17 (tBP), bis(trifluoromethylsulfonyl) amine lithium salt (Li-TFSI), rubrene (Rub, 18 19 99.99%) and dibenzotetraphenylperiflanthene (DBP, 98% HPLC) were obtained from Sigma-Aldrich. Lead iodide (PbI2, 99.999%) and stannic oxide (SnO2) colloid 20 precursor were obtained from Alfa Aesar. Methylamine iodide (CH₃NH₃I, 99.0%) 21 was purchased from TCI. 2,2',7,7'-tetrakis(N, N-di-p-methoxyphenylamine)-9,9'-22 spirobifluorene (Spiro-OMeTAD) was purchased from Xi'an Polymer Light 23 Technology Crop. Acetonitrile (99.9%) was purchased from J&K Scientific Ltd. 24 25

PVSCs Fabrication and Characterization: Perovskite precursor solution was 26 prepared by dissolving PbI₂ and CH₃NH₃I in DMF/DMSO (9:1, v/v) with a molar 27 ratio of 1:1. Indium tin oxide (ITO) glass substrates were cleaned by sequential 28 ultrasonic treatment with detergent solution, acetone, deionized water, and isopropyl 29 alcohol for 20 min, respectively. After drying with a stream of nitrogen, then the 30 substrates were treated with UV-ozone for 20 min in plasma cleaner. The SnO₂ 31 solution (1 mL SnO₂ colloid precursor was diluted by 3 mL of deionized water) was 32 spin-coated on ITO at 3000 rpm for 30 s, followed by annealing at 150 °C for 30 min 33 in air. For one-step anti- solvent method to fabricate the perovskite film, the 34 perovskite precursor was directly spin-coated onto the SnO₂ substrate at 4000 rpm for 35 30 s followed with a dripping of CB (0.15 mL, with and w/o Rub:DBP) after spin 36 coating step for 9 s. The various concentrations Rub:DBP solution were prepared by 37 dissolving different weights of Rub and DBP (mass ratio 10:1) in 1 mL CB. Then the 38 substrate annealed on a hot plate at 100 °C for 10 min. For bilayer structure device 39 preparation, the Rub:DBP solution was spin-coated onto the perovskite layer at 4000 40 rpm for 30 s. After the substrate cooled to room temperature, then Spiro-OMeTAD 41 solution [72.3 mg Spiro-OMeTAD with 17.5 µL LiTFSI solution (520 mg·mL⁻¹ in 42 acetonitrile) and 28.8 µL tBP in 1 mL CB] was spin-coated at 4000 rpm for 30 s on 43

the perovskite films. Finally, 90 nm Ag electrode was evaporated on the films to
complete the preparation of PVSCs with an effective area of 0.04 cm².

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Device Characterization: Current density-voltage (J-V) characteristics were 47 measured using a source meter (Keithley 2400), equipped with a light source (100 48 mW·cm⁻²) under AM 1.5 G filter. The reverse scan range is from 1.2 V to 0 V, with 49 20 mV for each step. The ultraviolet-visible (UV-Vis) spectra was conducted using a 50 SHIMADZU, UV-2600 spectrophotometer. The steady-state photoluminescence (PL) 51 and time-resolved photoluminescence (TRPL) spectra were recorded by an Edinburgh 52 instruments FLS920 spectrometer (Edinburgh Instruments Ltd.). Atomic force 53 microscopy (AFM) images were measured by MultiMode 8-HR (Bruker) atomic force 54 microscope. A Rigaku D/Max-B X-ray diffractometer with Bragg-Brentano 55 parafocusing geometry was employed to test X-ray diffraction (XRD) patterns. The 56 Two-dimensional grazing incidence X-ray diffraction (2D-GIXRD) measurement was 57 performed at the BL14B1 beamline of the Shanghai Synchrotron Radiation Facility 58 (SSRF). External quantum efficiency (EQE) values were measured under 59 monochromatic illumination (Oriel Cornerstone 260 1/4 m monochromator equipped 60 with an Oriel 70613NS QTH lamp), and the calibration of the incident light was 61 performed using a monocrystalline silicon diode. Electrical impedance spectroscopy 62 (EIS) of the PVSCs was performed in a frequency range from 1 MHz to 10 mHz 63 using Zahner electrochemical workstation at an applied bias equivalent to the open-64 circuit voltage of the cell under 1 sun illumination. 65



- 68 Fig. S1. SEM images of (a) the reference and (b-d) Rub:DBP incorporated perovskite
- 69 films with different mass ratio concentration.
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- 72 Fig. S2. AFM images of (a) the reference and (b-d) Rub:DBP incorporated perovskite
- 73 films with different mass ratio concentration.
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Fig. S3. XRD patterns of the reference and Rub:DBP incorporated perovskite films.



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82 Fig. S5. J-V curves of the reference and Rub:DBP incorporated PVSCs under

- 83 standard AM 1.5 G illumination (100 mW \cdot cm⁻²).
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86 **Fig. S6.** (a) V_{oc} , (b) J_{sc} , and (c) FF statistical distribution of the reference and 87 Rub:DBP incorporated PVSCs devices.



Fig. S7. PL spectra of the reference and Rub:DBP incorporated perovskite films onHTL.





92 Fig. S8. XRD patterns of the reference and Rub:DBP incorporated perovskite films93 after aged in air with a RH of 50-70% for 30 days.



96 Fig. S9. The contact angle changes of the reference and Rub:DBP incorporated97 perovskite films.



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Fig. S10. The long-term moisture stability of the unencapsulated reference and
Rub:DBP incorporated PVSCs under air environment with a RH of 50-70%.

	Sample	$\tau_1(ns)$	$\tau_2(ns)$	A ₁ (%)	$A_{2}(\%)$	$\tau_{ave}(ns)$
-	Reference	45.9	430.0	2.45	97.55	356
	with Rub:DBP	2.26	592.1	1.41	98.59	126

Tab. S1. The fitting parameters for time-resolved photoluminescence (TRPL) of thereference and Rub:DBP incorporated perovskite films.

106 Table S2. The charge mobilities of hole-only and electron-only devices based on
107 MAPbI₃ (with and w/o Rub:DBP).

Dovico	Hole mobility (am2.V-1.S-1)	Electron mobility (cm ² ·V ⁻¹ ·S ⁻		
Device	Hole mobility (cm ² V ² S ²)	1)		
Reference	1.69×10 ⁻²	6.60×10 ⁻¹		
with Rub:DBP	1.45×10 ⁻¹	6.94×10 ⁻¹		