

1 **Electronic Supplementary Information (ESI)**

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3 **High-performance Ni-rich Li[Ni_{0.9-x}Co_{0.1}Al_x]O₂ cathodes via multi-stage**
4 **microstructural tailoring from hydroxide precursor to the lithiated oxide**

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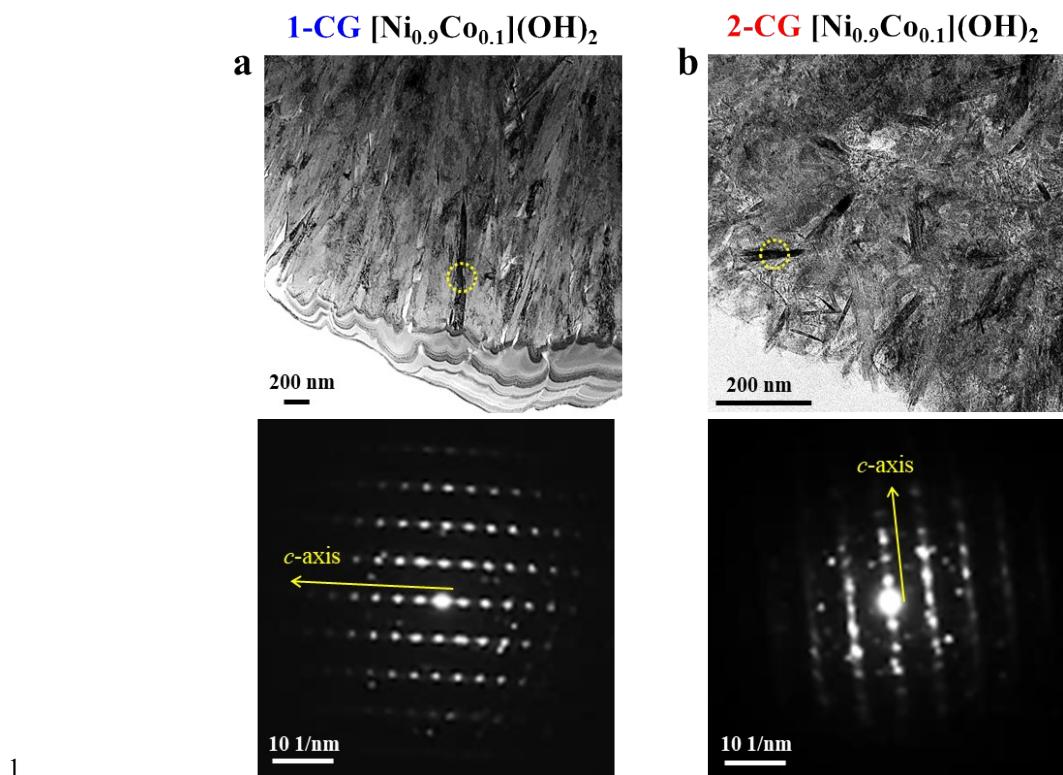
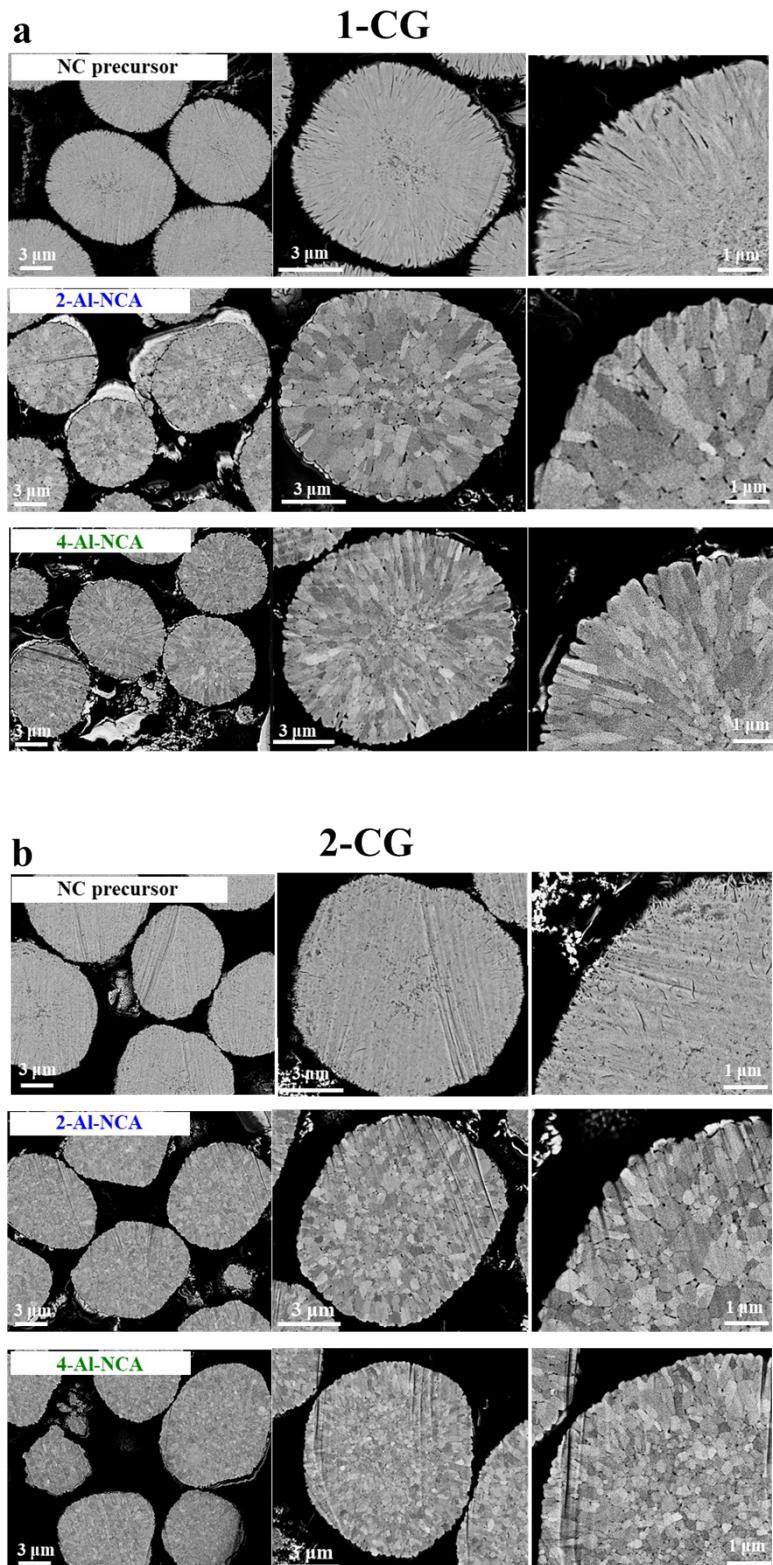
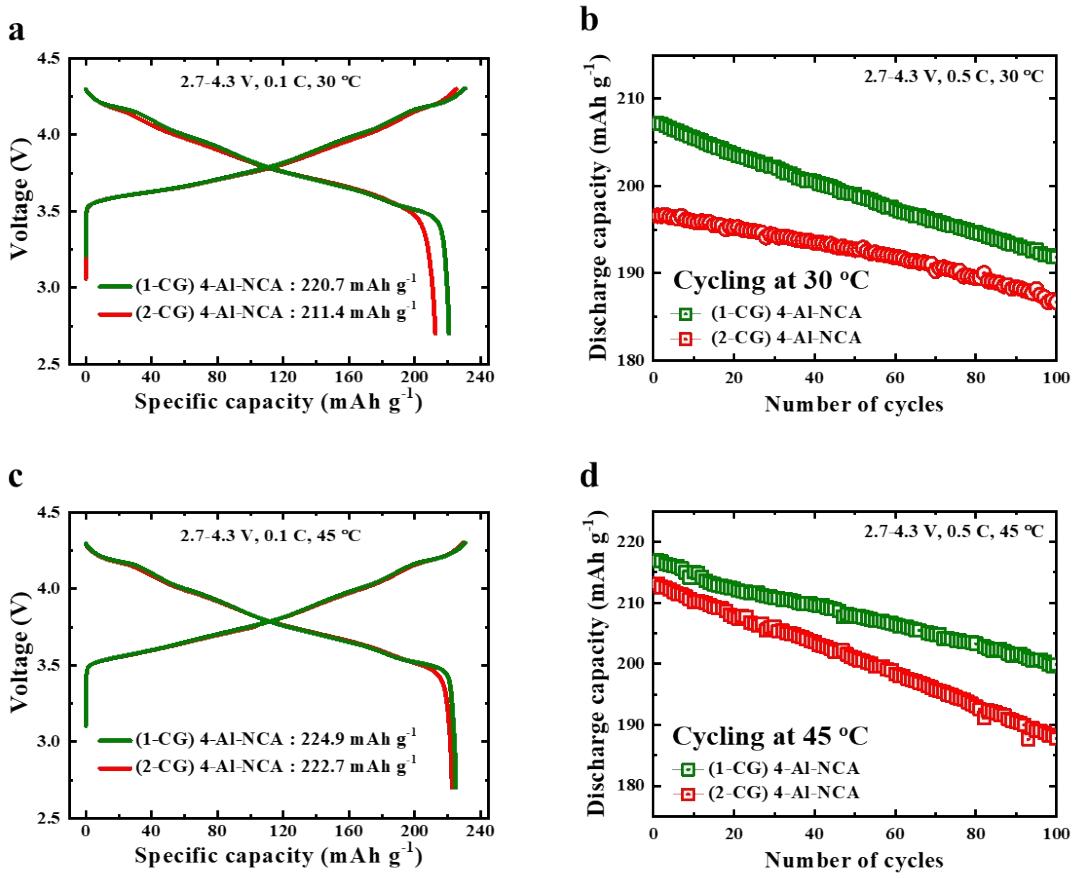


Fig. S1. Bright-field STEM images and selected area electron diffraction (SAED) patterns of marked primary particles of $[\text{Ni}_{0.9}\text{Co}_{0.1}](\text{OH})_2$ precursors; (a) 1-CG and (b) 2-CG.



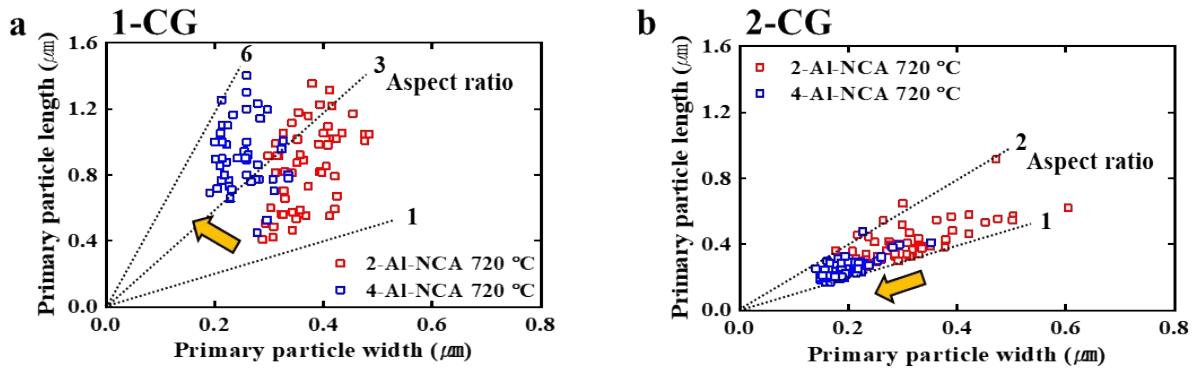
2 **Fig. S2.** Cross-sectional SEM images of $[Ni_{0.9}Co_{0.1}](OH)_2$ precursors, (a) 1-CG and (b) 2-CG, and
3 derived 2-Al-NCA and 4-Al-NCA cathode particles.



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2 **Fig. S3.** Comparison of initial charge–discharge curves (at 0.1 C) at (a) 30 °C and (c) 45 °C of 4-Al-
3 NCA cathodes derived from 1-CG and 2-CG precursors. Comparison of cycling performances (at 0.5
4 C) at (b) 30 °C and (d) 45 °C of 4-Al-NCA cathodes derived from 1-CG and 2-CG precursors.

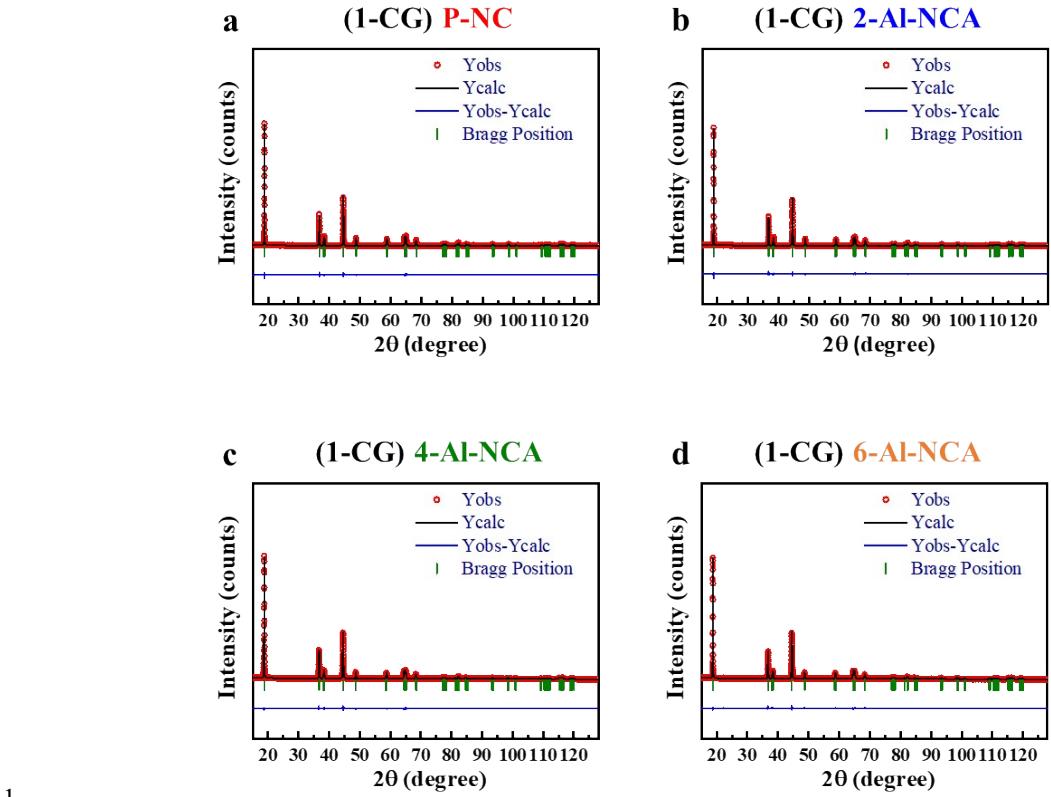
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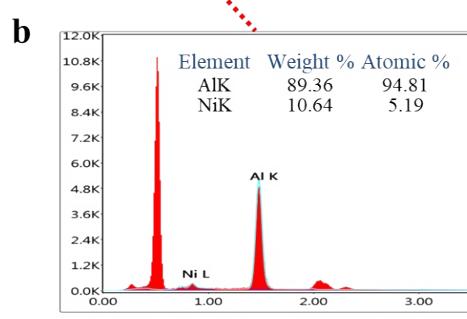
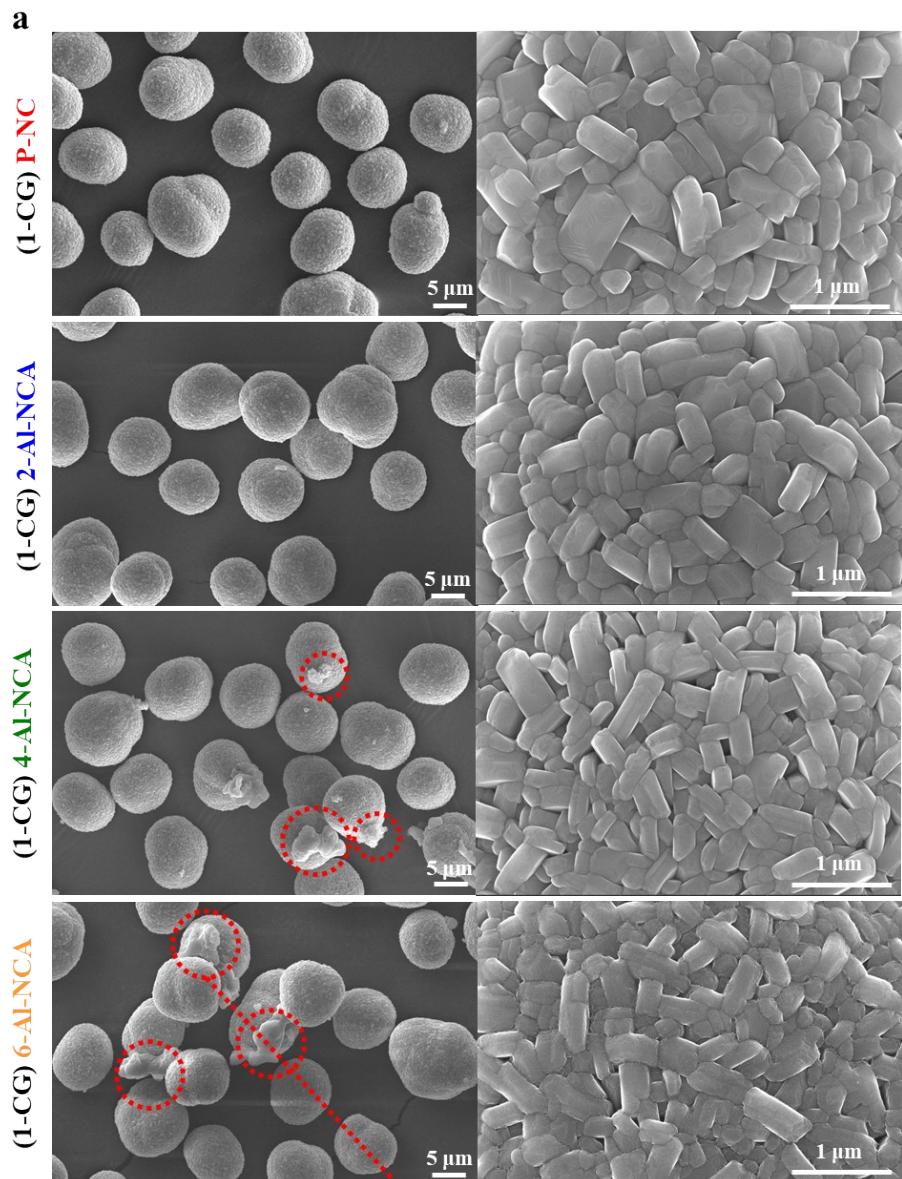


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2 **Fig. S4.** Quantitative comparison of the microstructures (in terms of primary particle length, width, and
3 aspect ratio) of 2-Al-NCA and 4-Al-NCA cathode particles derived from (a) 1-CG and (b) 2-CG
4 precursors. The lines indicate aspect ratio of primary particle (expressed as particle length/width).

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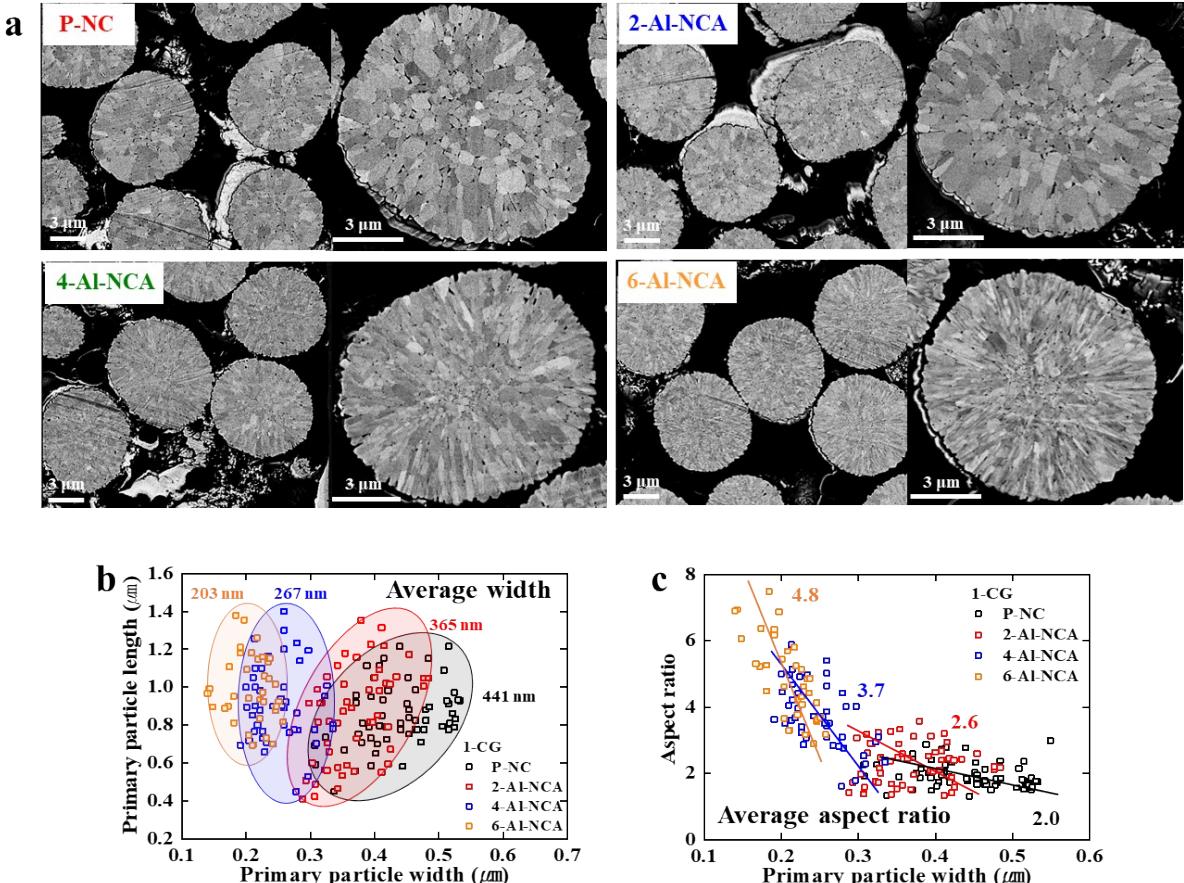




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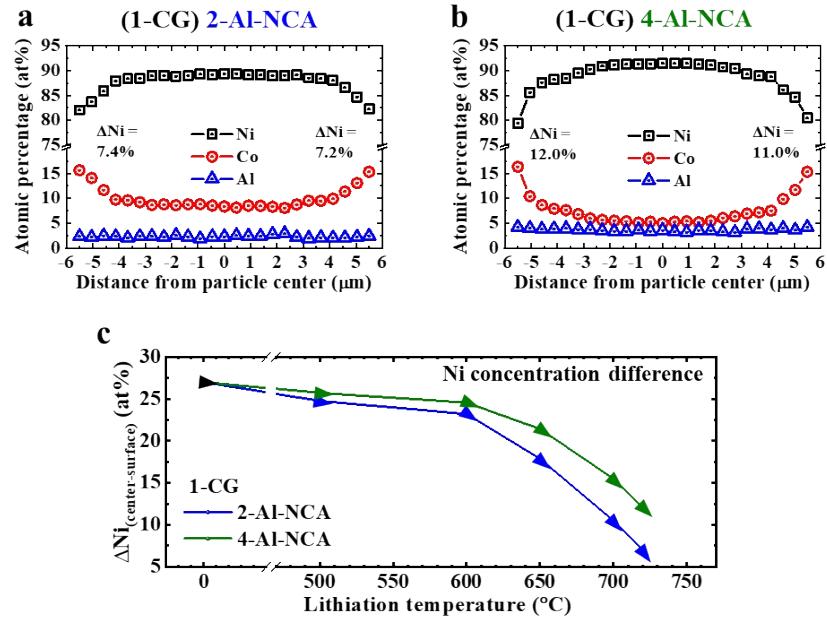
2 **Fig. S6.** (a) SEM images of P-NC, 2-Al-NCA, 4-Al-NCA, and 6-Al-NCA cathodes derived from 1-CG
3 precursors. (b) SEM-EDS result of marked impurity particle.

4



1 **Fig. S7.** (a) Cross-sectional SEM images of P-NC, 2-Al-NCA, 4-Al-NCA, and 6-Al-NCA cathodes
2 derived from 1-CG precursors. Quantitative comparisons of the microstructures of P-NC, 2-Al-NCA,
3 4-Al-NCA, and 6-Al-NCA cathodes in terms of the (b) size (length and width) and (c) aspect ratio of
4 their primary particles.
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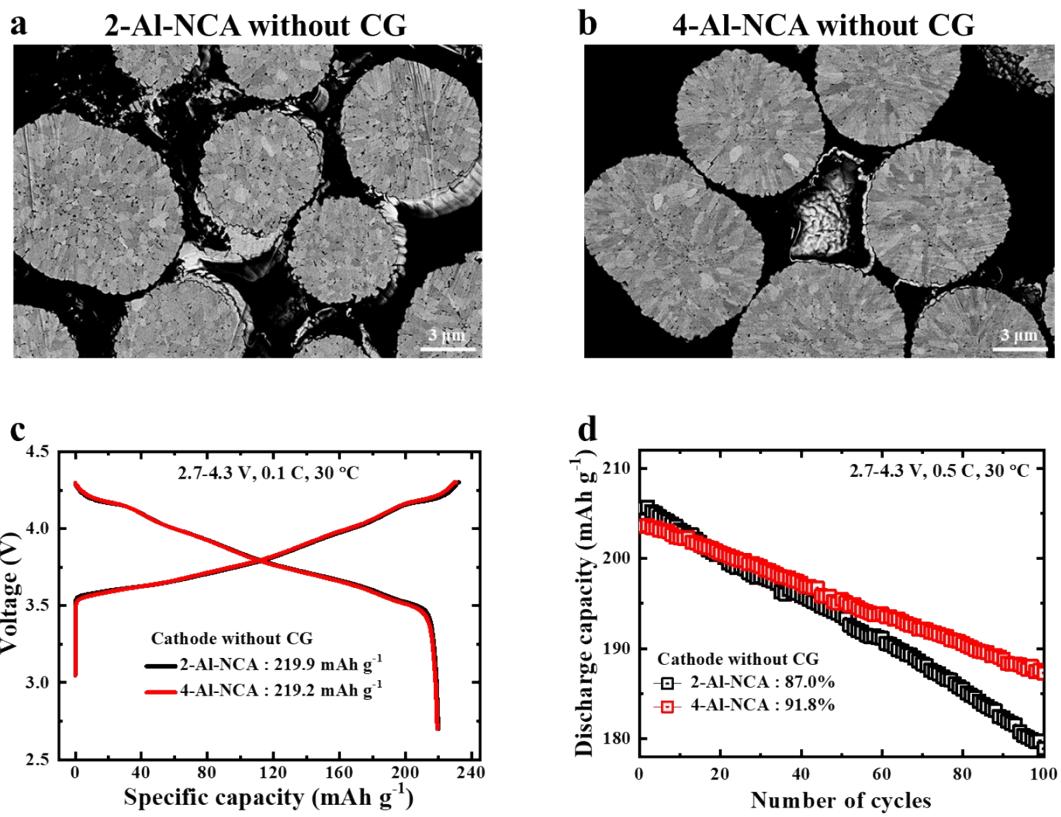
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2 **Fig. S8.** Concentration gradient profiles (Ni, Co, and Al) of (a) 2-Al-NCA and (b) 4-Al-NCA cathode
 3 particles, as determined by electron probe microanalysis. (c) Comparison of Ni concentration difference
 4 between particle center and the surface of 2-Al-NCA and 4-Al-NCA cathodes as a function of lithiation
 5 temperature. The cathodes are derived from 1-CG precursors.

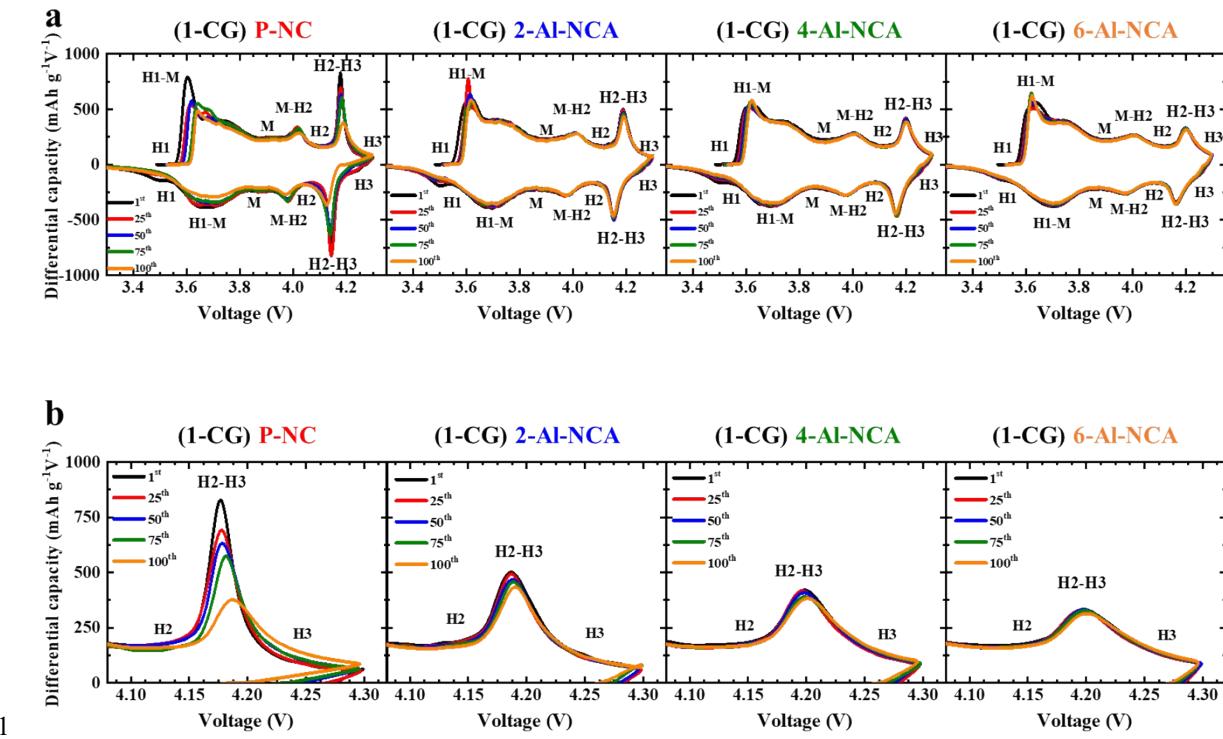
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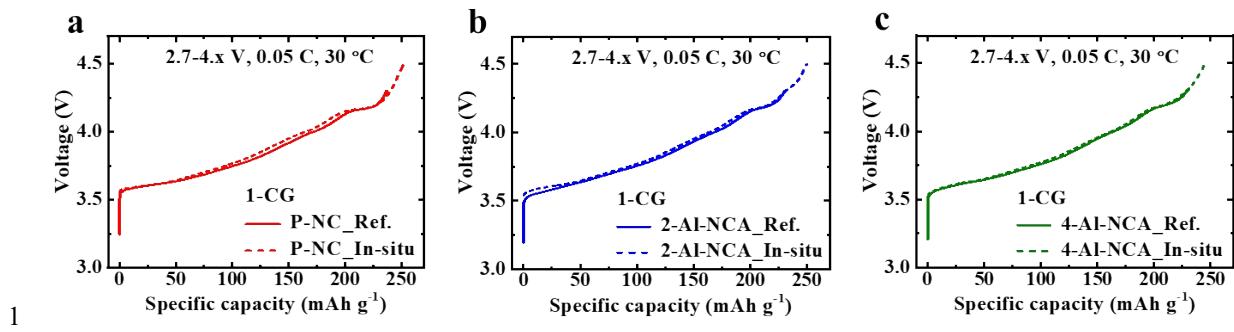
2 **Fig. S9.** Cross-sectional SEM images of (a) 2-Al-NCA, and (b) 4-Al-NCA cathode without
 3 concentration gradients. Comparison of (c) initial charge–discharge curves (at 0.1 C) and (d) cycling
 4 performance (at 0.5 C) of 2-Al-NCA and 4-Al-NCA cathodes derived from conventional precursor
 5 without concentration gradients.

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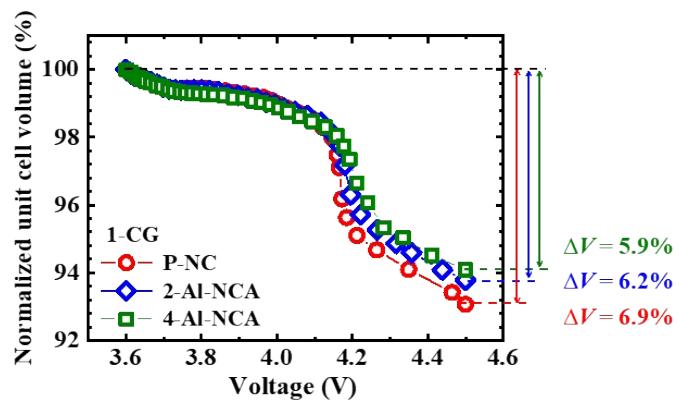
2 **Fig. S10.** (a) Differential capacity profiles of cells featuring P-NC, 2-Al-NCA, 4-Al-NCA, and 6-Al-
3 NCA cathodes. (b) Magnified profiles during charge in the voltage range of 4.08–4.32 V. The cathodes
4 are derived from 1-CG precursors.

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2 **Fig. S11.** Charge profiles of pouch-type half-cells (dashed lines), recorded during in situ XRD, and
3 2032 coin-type half-cells (solid lines) featuring (a) P-NC, (b) 2-Al-NCA, and (c) 4-Al-NCA cathodes.
4 The cathodes are derived from 1-CG precursors.

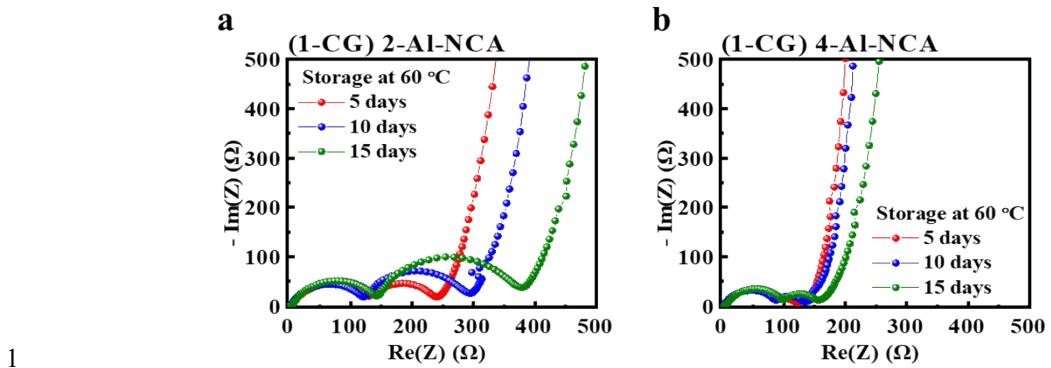
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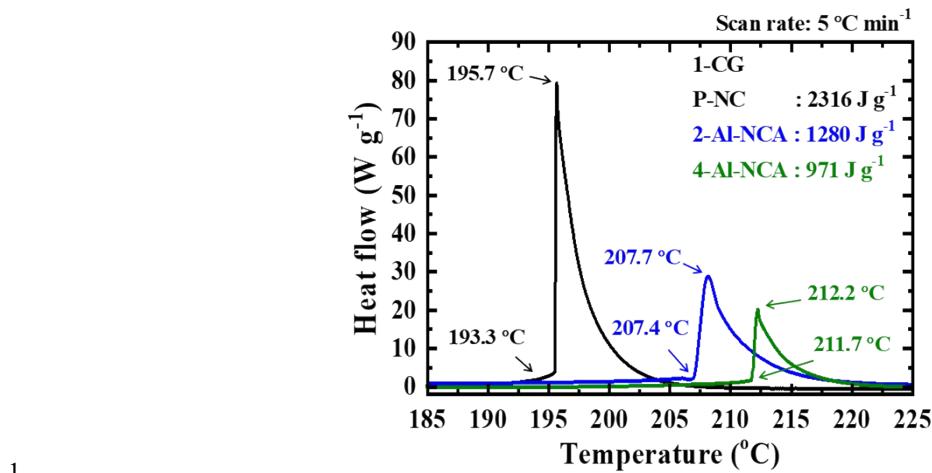
2 **Fig. S12.** Comparison of the unit cell volume changes of P-NC, 2-Al-NCA, and 4-Al-NCA cathodes
 3 during charge. The cathodes are derived from 1-CG precursors.

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2 **Fig. S13. Comparison of the Nyquist plots of the measured electrochemical impedances.** (a) 2-Al-
3 NCA and (b) 4-Al-NCA cathodes, charged to 4.3 V, were immersed in electrolyte (1.2 M LiPF₆ in
4 EC:EMC (3:7 by vol%) with 2 wt% VC) at 60 °C for 15 days, and their electrochemical impedances
5 were measured every 5 days. The cathodes are derived from 1-CG precursors.

6

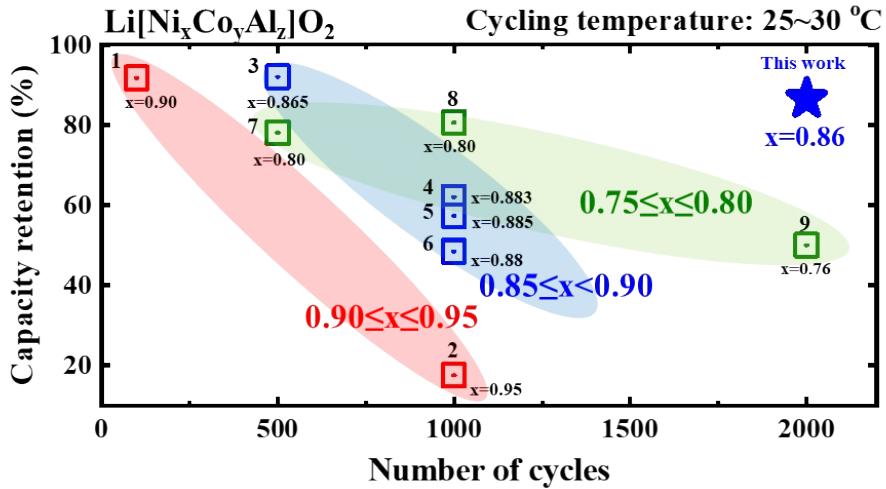


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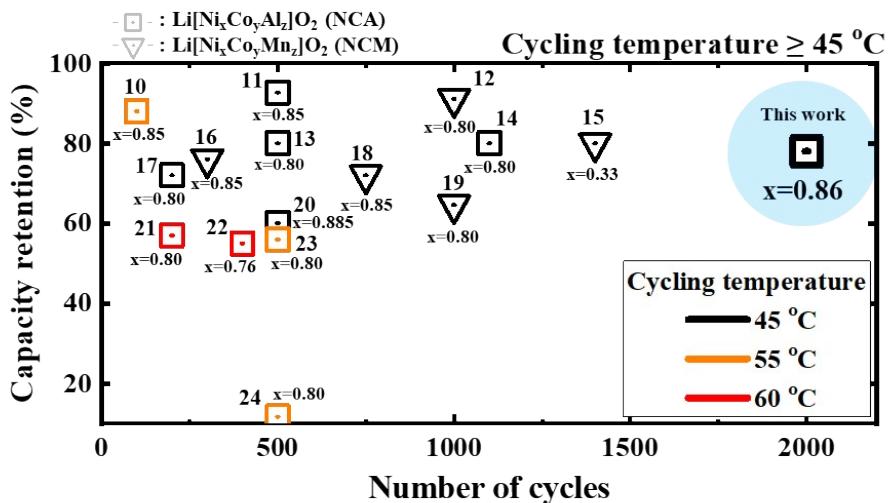
2 **Fig. S14.** Comparison of DSC thermal stability results of P-NC, 2-Al-NCA, and 4-Al-NCA cathodes
3 charged to 4.3 V. The cathodes are derived from 1-CG precursors.

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a Cycling performance at room temperature

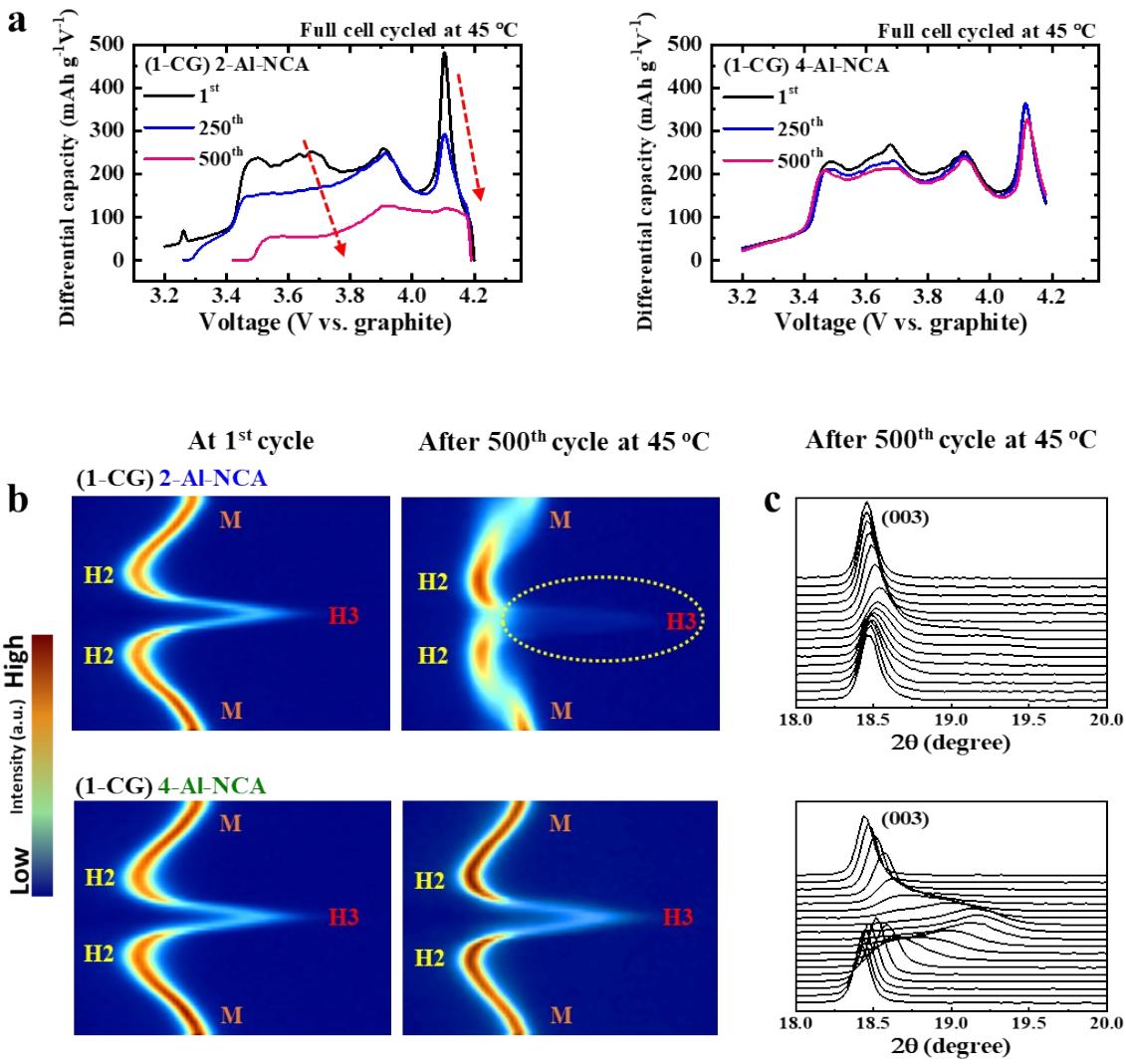


b Cycling performance at elevated temperature



2 **Fig. S15.** The figure illustrates that the proposed Ni-rich 4-Al-NCA cathode demonstrated outstanding
3 long-term cycling performance compared with other Ni-rich NCA and NCM cathodes reported in
4 previous publications. Comparison of the long-term cycling performances at (a) room temperature, and
5

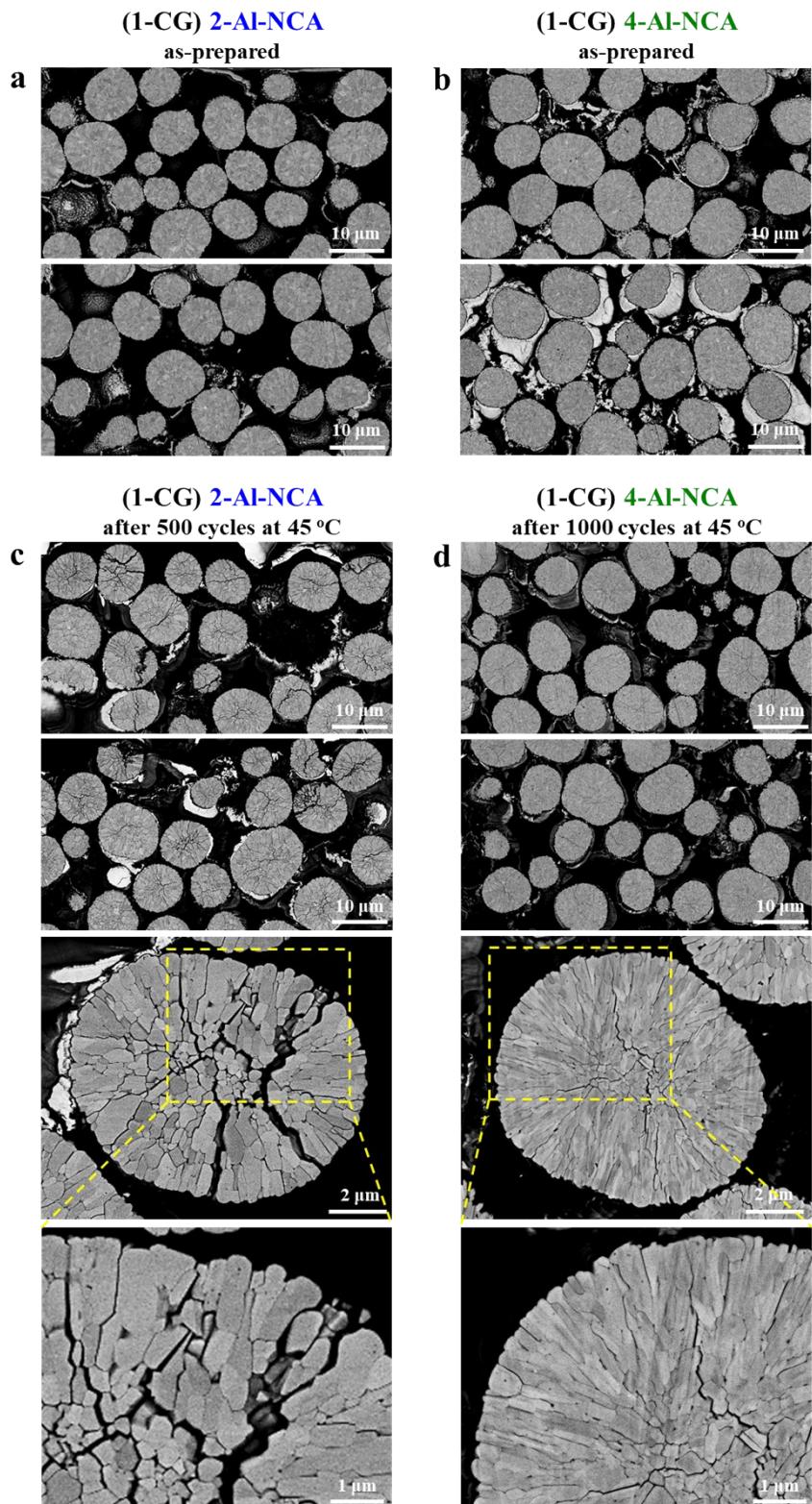
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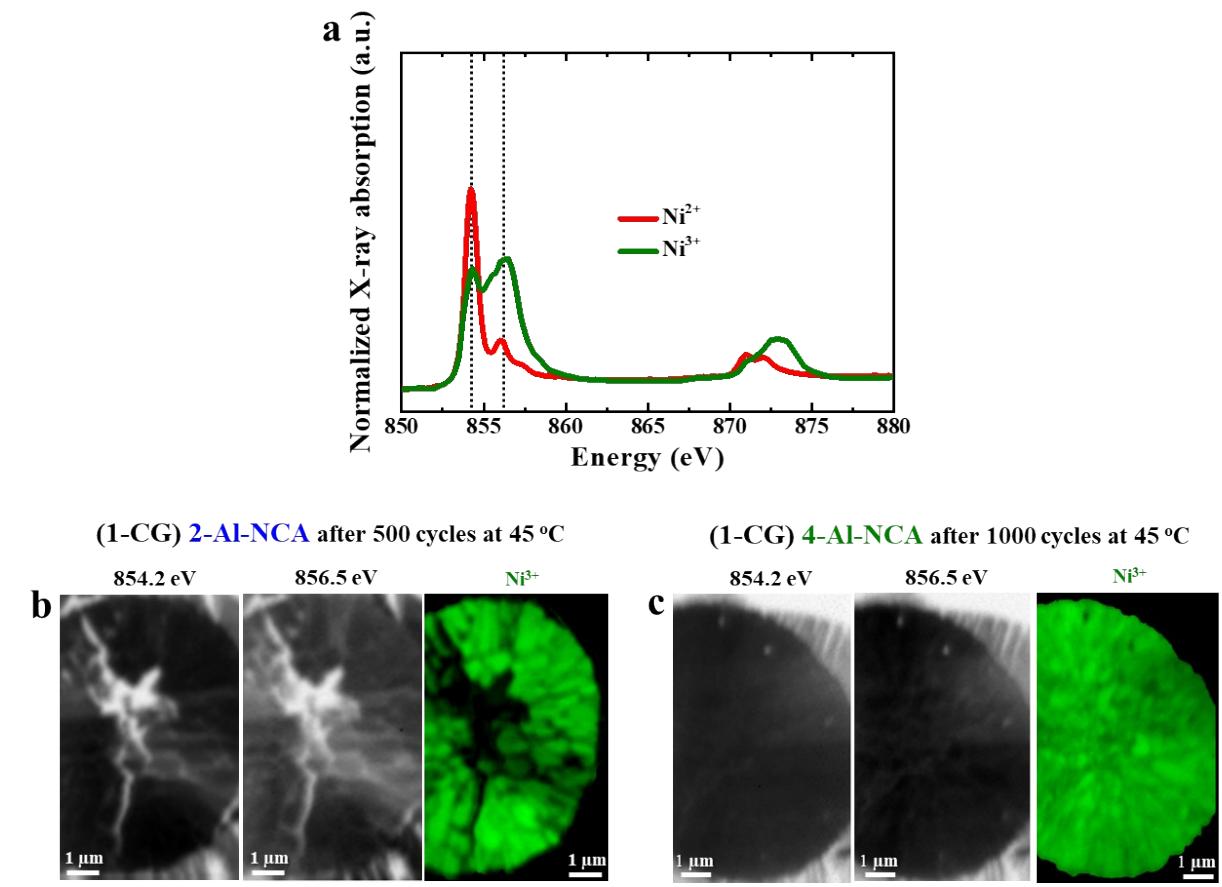
2 **Fig. S16.** (a) Differential capacity profiles of the 1st, 250th, and 500th cycles of cells featuring 2-Al-NCA
3 and 4-Al-NCA cathodes. (b) Contour plots of (003) reflections of 2-Al-NCA and 4-Al-NCA cathodes
4 measured during the 1st and 500th cycles at 45 °C. (c) In situ XRD patterns of 2-Al-NCA and 4-Al-NCA
5 cathodes in the 2θ ranges of 18.0°–20.0° for the (003) reflections after 500 cycles at 45 °C. The cathodes
6 are derived from 1-CG precursors.

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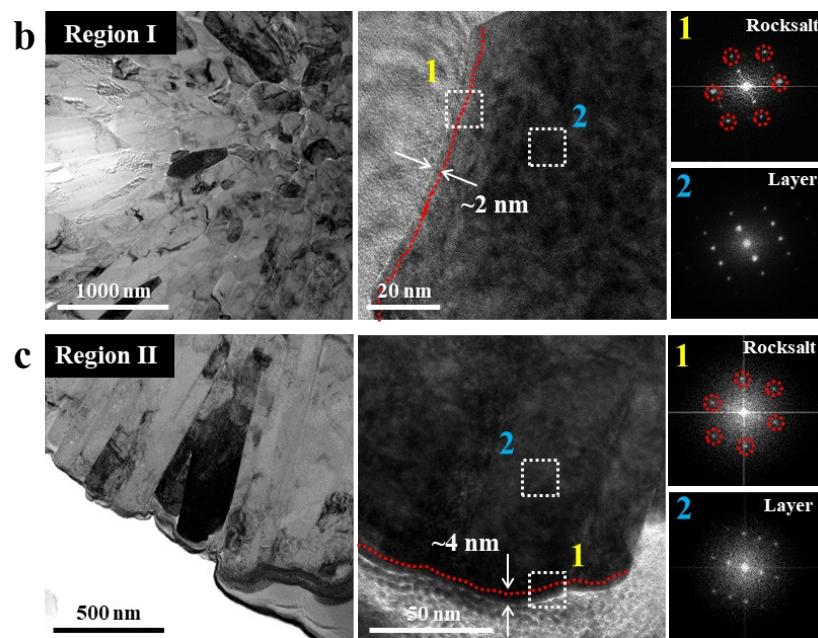
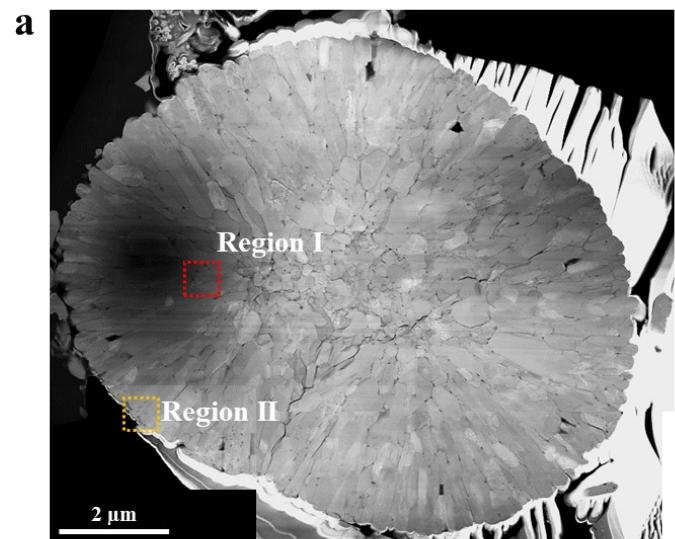
2 **Fig. S17.** Cross-sectional SEM images of as-prepared (a) 2-Al-NCA, and (b) 4-Al-NCA cathodes
3 derived from 1-CG precursors. Comparison of the cross-sectional SEM images of (c) 2-Al-NCA
4 cathode particles after 500 cycles and (d) 4-Al-NCA cathode particles after 1000 cycles at 45 °C. The
5 cathodes are derived from 1-CG precursors.



2 **Fig. S18.** (a) Reference Ni *L*-edge X-ray absorption spectra of $\text{LiMn}_{1.5}\text{Ni}_{0.5}\text{O}_4$ (Ni^{2+} , red solid line) and
3 $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$ (Ni^{3+} , green solid line). Single-energy STXM images obtained at 854.2 and 856.5
4 eV, indicated by black dashed lines in (a), for the cross-sections of a fully discharged (b) 2-Al-NCA
5 cathode particle after 500 cycles at 45 °C and (c) 4-Al-NCA cathode particle after 1000 cycles at 45 °C.
6 The cathodes are derived from 1-CG precursors.

7

(1-CG) 4-Al-NCA
after 1000 cycles at 45 °C



2 **Fig. S19. Structural stability of the 4-Al-NCA cathode after 1000 cycles at 45 °C.** (a) Dark-field
3 cross-sectional STEM image of cycled 4-Al-NCA cathode. Magnified TEM images of the primary
4 particles marked as (b) region I, (c) region II with high-resolution TEM images and corresponding fast
5 Fourier transforms (FFTs). The cathode is derived from a 1-CG precursors.

6

1 **Table S1.** Chemical compositions of cathode materials determined by ICP-OES.

Precursor	Lithiation temperature	Cathode	Ni (at%)	Co (at%)	Al (at%)
1-CG	720 °C	P-NC	89.51	10.49	-
		2-Al-NCA	87.82	10.16	2.02
		4-Al-NCA	85.91	9.98	4.11
		6-Al-NCA	84.39	9.90	5.71
2-CG		2-Al-NCA	87.91	9.98	2.11
		4-Al-NCA	85.94	9.84	4.12

2

1 **Table S2.** Structural parameters of P-NC, 2-Al-NCA, 4-Al-NCA, and 6-Al-NCA cathodes derived from
 2 1-CG precursors, as determined by the Rietveld refinement of the associated XRD data.

Li[Ni _{0.90} Co _{0.10}]O ₂ (P-NC)		Space group: $R\bar{3}m$				
<i>a</i> -axis: 2.8688 (1) Å		<i>c</i> -axis: 14.1793 (2) Å		Volume: 101.061 (3) Å ³		
<i>R</i> _p : 2.00%	<i>R</i> _{wp} : 2.58%	<i>R</i> _{exp} : 2.04%	<i>Chi</i> ² : 1.61			
Atom	Wyckoff position	x	y	z	B _{iso}	*Modified occupancy
Li1	3a	0	0	0	1.000	0.992
Ni2	3a	0	0	0	1.000	0.008
Ni1	3b	0	0	0.5	0.6490	0.892
Co1	3b	0	0	0.5	0.6490	^a 0.100
Li2	3b	0	0	0.5	0.6490	0.008
O1	6c	0	0	0.2420 (7)	0.7260	2

3 *Modified occ. = occupation *12 for Fullprof (occ.*12)

4 ^a) Fixed value

5

Li[Ni _{0.88} Co _{0.10} Al _{0.02}]O ₂ (2-Al-NCA)		Space group: $R\bar{3}m$				
<i>a</i> -axis: 2.8668 (1) Å		<i>c</i> -axis: 14.1850 (1) Å		Volume: 100.961 (2) Å ³		
<i>R</i> _p : 1.95%	<i>R</i> _{wp} : 2.52%	<i>R</i> _{exp} : 2.04%	<i>Chi</i> ² : 1.52			
Atom	Wyckoff position	x	y	z	B _{iso}	*Modified occupancy
Li1	3a	0	0	0	1.000	0.992
Ni2	3a	0	0	0	1.000	0.008
Ni1	3b	0	0	0.5	0.6490	0.872
Co1	3b	0	0	0.5	0.6490	^a 0.100
Al1	3b	0	0	0.5	0.6490	^a 0.020
Li2	3b	0	0	0.5	0.6490	0.008
O1	6c	0	0	0.2420 (1)	0.7260	2

6 *Modified occ. = occupation *12 for Fullprof (occ.*12)

7 ^a) Fixed value

8

Li[Ni _{0.86} Co _{0.10} Al _{0.04}]O ₂ (4-Al-NCA)	Space group: $R\bar{3}m$
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a -axis: 2.8663 (3) Å		c -axis: 14.1871 (1) Å		Volume: 100.941 (3) Å ³		
R_p : 1.09%		R_{wp} : 1.67%		R_{exp} : 0.77% Chi^2 : 4.66		
Atom	Wyckoff position	x	y	z	B _{iso}	*Modified occupancy
Li1	3a	0	0	0	1.000	0.993
Ni2	3a	0	0	0	1.000	0.007
Ni1	3b	0	0	0.5	0.6490	0.853
Co1	3b	0	0	0.5	0.6490	^{a)} 0.100
Al1	3b	0	0	0.5	0.6490	^{a)} 0.040
Li2	3b	0	0	0.5	0.6490	0.007
O1	6c	0	0	0.2415 (3)	0.7260	2

1 *Modified occ. = occupation *12 for Fullprof (occ.*12)

2 ^{a)} Fixed value

3

Li[Ni _{0.84} Co _{0.10} Al _{0.06}]O ₂ (6-Al-NCA)		Space group: $R\bar{3}m$		Volume: 100.920 (2) Å ³		
a -axis: 2.8659 (2) Å		c -axis: 14.1881 (1) Å				
R_p : 1.07%	R_{wp} : 1.59%	R_{exp} : 0.78%	Chi^2 : 4.12			
Atom	Wyckoff position	x	y	z	B _{iso}	*Modified occupancy
Li1	3a	0	0	0	1.000	0.991
Ni2	3a	0	0	0	1.000	0.009
Ni1	3b	0	0	0.5	0.6490	0.831
Co1	3b	0	0	0.5	0.6490	^{a)} 0.100
Al1	3b	0	0	0.5	0.6490	^{a)} 0.060
Li2	3b	0	0	0.5	0.6490	0.009
O1	6c	0	0	0.2417 (1)	0.7260	2

4 *Modified occ. = occupation *12 for Fullprof (occ.*12)

5 ^{a)} Fixed value

6

1 **Table S3.** Structural parameters, I(003)/I(104) XRD peak intensity ratio, and calculated crystallite size
2 of P-NC, 2-Al-NCA, 4-Al-NCA, and 6-Al-NCA cathodes derived from 1-CG precursors.

Precursor	Cathode	<i>a</i> -axis (Å)	<i>c</i> -axis (Å)	V (Å ³)	I (003) / I (104)	*Crystallite size (nm)
1-CG	P-NC	2.8688	14.1793	101.061	2.26	113.2
	2-Al-NCA	2.8668	14.1850	100.961	2.29	104.6
	4-Al-NCA	2.8663	14.1871	100.941	2.23	79.7
	6-Al-NCA	2.8659	14.1881	100.920	2.22	78.5

3 * Scherrer equation (K=0.9) based on (003) reflection $\tau = \frac{K\lambda}{\beta \cos \theta}$

4

1 **Table S4.** Performance comparison of our 4-Al-NCA with those of other layered cathodes reported in
2 previous publications.

Num.	Cathode material	Cathode composition	Electrolyte	Mass loading (mg/cm ²)	Voltage window (V)	Cycling temp. (°C)	C-rate for cycling	1 st discharge capacity (mAh g ⁻¹)	Number of cycles	Cycle retention (%)	Ref.
This work	4-Al-NCA (CG)	Li[Ni _{0.86} Co _{0.10} Al _{0.04}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	—	9~10	3.0~4.2	25	0.8C / 1C	182	2000	86.5
						45	0.8C / 1C	181.8	2000	78.0	—
1	Li[Ni _{0.9} Co _{0.07} Al _{0.03}]O ₂	Li[Ni _{0.9} Co _{0.07} Al _{0.03}]O ₂	1.15 M LiPF ₆ in EC : DMC : EMC, 1 : 2 : 2 by vol%	~7	2.0~4.25	25	1C	181.7	100	91.7	[1]
2	NCA95	Li[Ni _{0.95} Co _{0.05} Al _{0.01}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	~6	3.0~4.2	25	0.8C / 1C	196.7	1000	17.5	[2]
3	CSG NCA	Li[Ni _{0.865} Co _{0.120} Al _{0.015}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	—	3.0~4.2	25	0.8C / 1C	~185	500	92	[3]
4	NCA-89	Li[Ni _{0.883} Co _{0.053} Al _{0.064}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	10	2.5~4.2	25	0.5C / 1C	~185	1000	62	[4]
5	P-NCA85	Li[Ni _{0.885} Co _{0.13} Al _{0.014}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	9	3.0~4.2	25	0.8C / 1C	179.4	1000	57.3	[5]
6	NCA88	Li[Ni _{0.88} Co _{0.10} Al _{0.02}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	~6	3.0~4.2	25	0.8C / 1C	190.1	1000	48.5	[2]
7	NCA (E9)	Li[Ni _{0.80} Co _{0.15} Al _{0.08}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol%	~8	2.5~4.3	30	0.33C / 1C	~160	500	~78	[6]
8	NCA80	Li[Ni _{0.80} Co _{0.16} Al _{0.04}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	~6	3.0~4.2	25	0.8C / 1C	173.6	1000	80.6	[2]
9	NCA	Li[Ni _{0.76} Co _{0.14} Al _{0.10}]O ₂	1.0M LiPF ₆ in EC, EMC, DMC	—	2.5~4.2	25	1C	—	2000	~50	[7]
10	Li[Ni _{0.85} Co _{0.094} Al _{0.056}]O ₂	Li[Ni _{0.85} Co _{0.094} Al _{0.056}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	8	2.5~4.3	55	0.5C	~180	100	88	[8]
11	1-Nb NCA85	Li[Ni _{0.85} Co _{0.125} Al _{0.014} Nb _{0.01}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	9	3.0~4.2	45	0.8C / 1C	180.1	500	92.7	[5]
12	F1-GC80	Li[Ni _{0.80} Co _{0.05} Mn _{0.15}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	~4	3.0~4.2	45	1C	190	1000	91.1	[9]
13	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	1.0M LiPF ₆ in EC, EMC, DMC	—	2.5~4.2	45	0.3C / 1C	—	500	~80	[10]
14	TiP2O7-coated NCA	Li[Ni _{0.80} Co _{0.15} Al _{0.05} Ti _{0.01}]O ₂	1.0 M LiPF ₆ in EC : EMC : DEC, 3 : 3 : 4 by vol%	17.8	2.8~4.2	45	1C	—	1050	80	[11]
15	NCM	Li[Ni _{0.33} Co _{0.33} Mn _{0.33}]O ₂	EC based electrolyte	—	3.0~4.2	45	3C	—	1400	~80%	[12]
16	Li[Ni _{0.85} Co _{0.1} Mn _{0.05}]O ₂	Li[Ni _{0.85} Co _{0.1} Mn _{0.05}]O ₂	1.0 M LiPF ₆ in EC : DEC:EMC, 3 : 2 : 5 by vol% with 0.5% VC, 1% DTD, 1% LiPSI	~33	2.8~4.2	45	0.5C / 1C	—	300	76%	[13]
17	NCA (E9)	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol%	~8	2.5~4.3	45	0.33C / 1C	~175	200	~72	[6]
18	NCM851005	Li[Ni _{0.85} Co _{0.1} Mn _{0.05}]O ₂	1.0 M LiPF ₆ in EC : DEC, 3 : 7 by vol% with 2 wt% VC	~10	2.8~4.2	45	1C	195	750	72%	[14]
19	GC80	Li[Ni _{0.80} Co _{0.05} Mn _{0.15}]O ₂	1.2 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	~4	3.0~4.2	45	1C	192	1000	64.6	[9]
20	P-NCA85	Li[Ni _{0.885} Co _{0.13} Al _{0.014}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol% with 2 wt% VC	9	3.0~4.2	45	0.8C / 1C	180.4	500	60.1	[5]
21	NCA (E9)	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	1.0 M LiPF ₆ in EC : EMC, 3 : 7 by vol%	~8	2.5~4.3	60	0.33C / 1C	~180	200	~57	[6]
22	NCA	Li[Ni _{0.76} Co _{0.14} Al _{0.10}]O ₂	LiPF ₆ in EC, EMC, DMC	—	2.5~4.2	60	1C	—	~400	~55	[7]
23	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	1.0 M LiPF ₆ in EC : EMC : DEC, 1 : 1 by vol%	—	3.0~4.2	55	1C	—	500	55.9	[15]
24	AlF ₃ -coated Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	Li[Ni _{0.80} Co _{0.15} Al _{0.05}]O ₂	1.0 M LiPF ₆ in EC : EMC : DEC, 1 : 1 by vol%	—	3.0~4.2	55	1C	—	500	11.7	[15]

3 *The value was calculated based on figure given in research article.

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