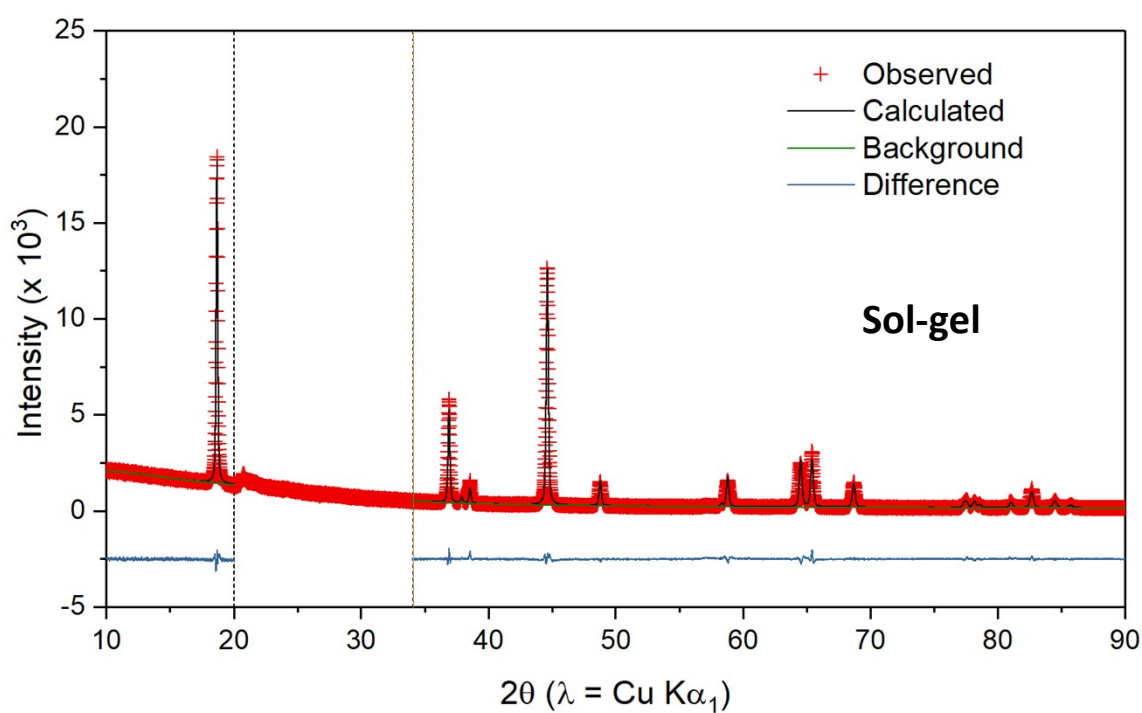
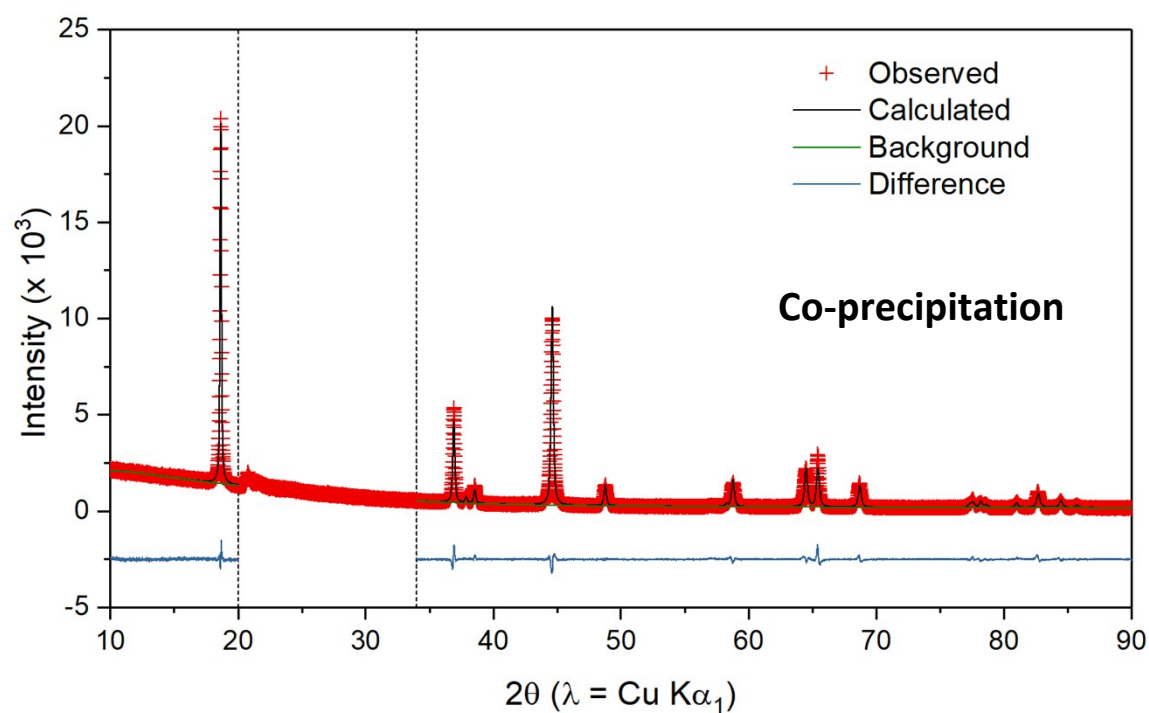


Supplementary Information

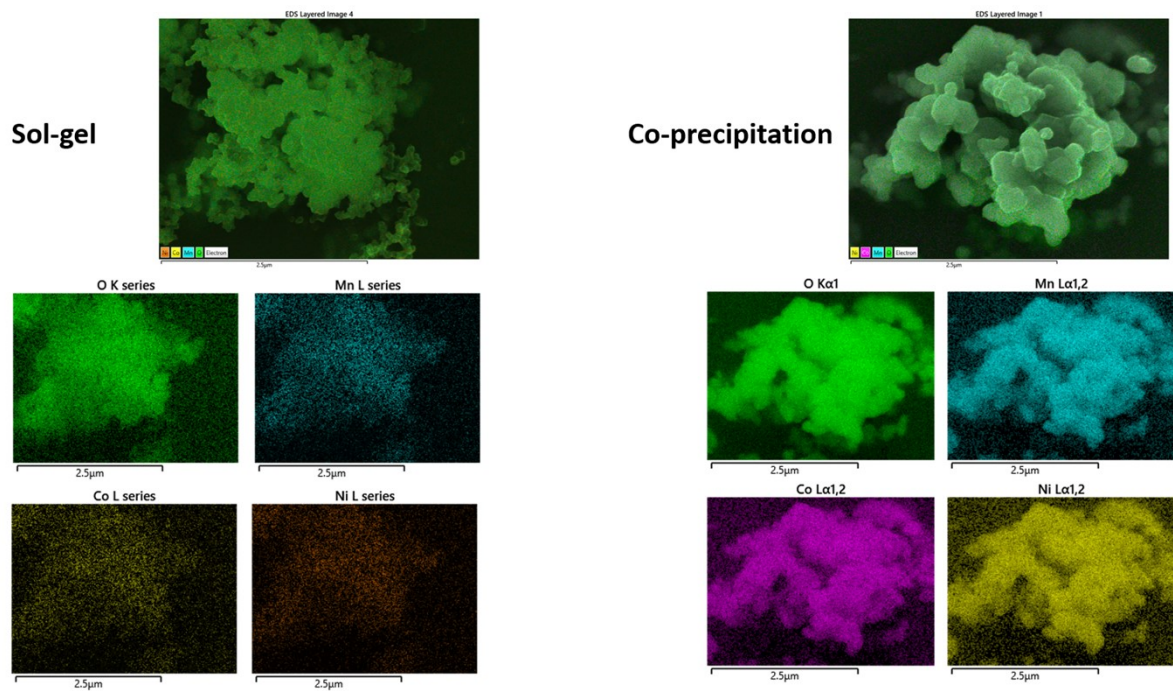
Detection of trapped molecular O₂ in a charged Li-rich cathode by Neutron PDF

Robert A. House, Helen Y. Playford, Ronald I. Smith, Jennifer Holter, Ian Griffiths, Ke-Jin Zhou, Peter
G. Bruce

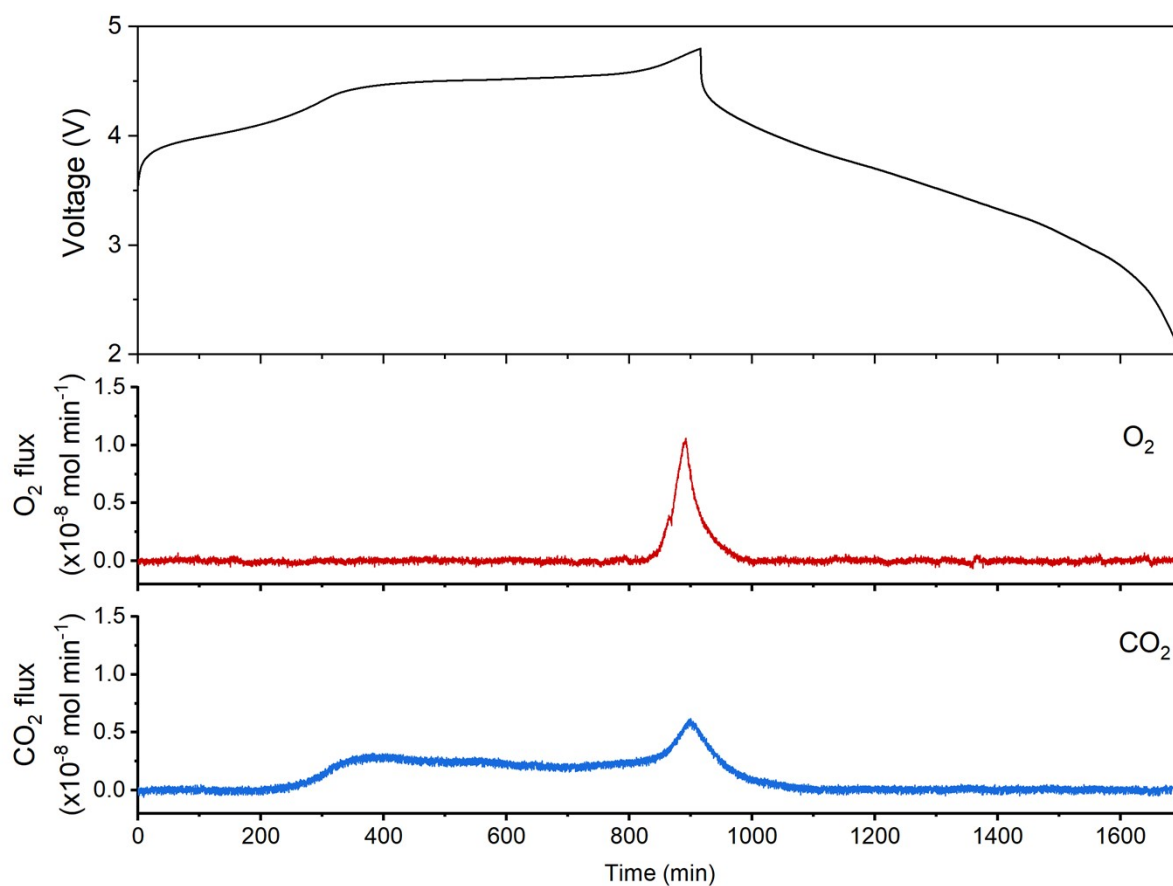


Sup

p. Fig. 1. Fitted PXRD data following Rietveld refinement for pristine $\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ prepared by carbonate co-precipitation (upper panel) and sol-gel (lower panel).

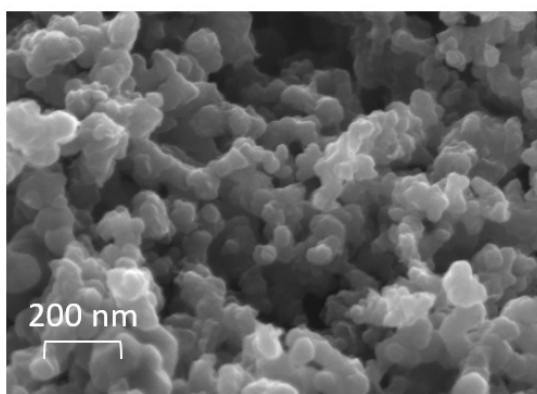
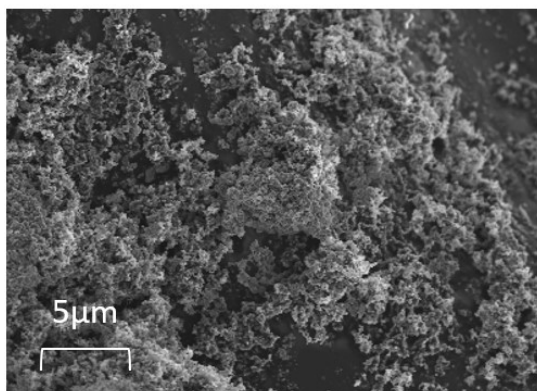


Supp. Fig. 2. EDX images. Homogeneous mixing of transition metals within and between primary particles for pristine $\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ prepared by sol-gel and carbonate co-precipitation.

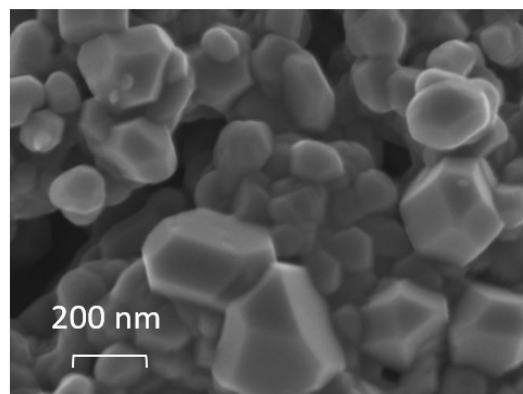
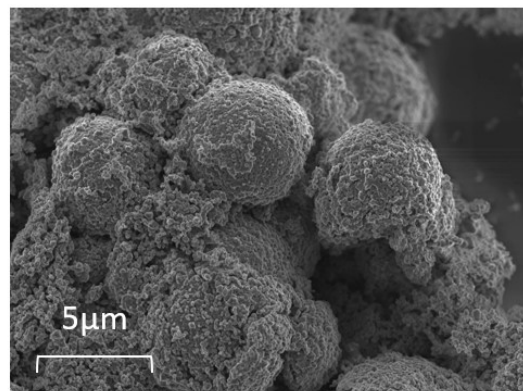


Supp. Fig. 3. OEMS data. Evolved O₂ and CO₂ gases detected by mass spec for co-precipitation derived Li_{1.2}Ni_{0.13}Co_{0.13}Mn_{0.54}O₂ charged at a rate of 20 mA g⁻¹ between 2 and 4.8 V. 0.0041 moles of O₂ and 0.0159 moles of CO₂ were evolved per mole of Li_{1.2}Ni_{0.13}Co_{0.13}Mn_{0.54}O₂ during the first cycle. The electrode active material mass for the experiment was 10.32 mg.

Sol-gel



Co-precipitation



Supp. Fig. 4. SEM images Primary and secondary particle morphology for pristine $\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ prepared by sol-gel and carbonate co-precipitation.

Supp. Tab. 1. X-ray diffraction refinement parameters for a pristine ${}^7\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ prepared by co-precipitation for the neutron study and b $\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ prepared by sol-gel.

a

Atom	Wyckoff	x	y	z	Occupancy	U_{iso}
${}^7\text{Li}$	3a	0	0	0	1	0.0137(2)
Li/Ni/Co/Mn	3b	0	0	0.5	0.2/0.13/0.13/0.54	0.0140(4)
O	6c	0	0	0.2405(1)	1	0.0243(5)
S.G. = R-3m		a = 2.849(1) Å			$\alpha = 90$	
$R_w = 0.0555$		c = 14.226(1) Å			$\beta = 90$	
G.O.F. = 1.6					$\gamma = 120$	

b

Atom	Wyckoff	x	y	z	Occupancy	U_{iso}
${}^7\text{Li}$	3a	0	0	0	1	0.0135(2)
Li/Ni/Co/Mn	3b	0	0	0.5	0.2/0.13/0.13/0.54	0.0126(3)
O	6c	0	0	0.2416(1)	1	0.0201(4)
S.G. = R-3m		a = 2.850(1) Å			$\alpha = 90$	
$R_w = 0.0516$		c = 14.221(1) Å			$\beta = 90$	
G.O.F. = 1.5					$\gamma = 120$	

Supp. Tab. 2. Neutron PDF refinement parameters for pristine ${}^7\text{Li}_{1.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$

Atom	Wyckoff	x	y	z	Occupancy
${}^7\text{Li}$	4h	0	0.3328(7)	0.5	1
${}^7\text{Li}$	2c	0	0	0.5	1
${}^7\text{Li}/\text{Ni}$	2b	0	0.5	0	0.6/0.4
Co/Mn	4g	0	0.1617(8)	0	0.2/0.8
O	8j	0.2511(6)	0.3222(2)	0.2231(3)	1
O	4i	0.2184(7)	0	0.2256(5)	1
S.G. = C2/m, $R_w = 0.092$, $Q_{\text{max}} = 30 \text{ \AA}$, $r = 0.5 - 40 \text{ \AA}$		a = 4.942(1) $\text{\AA}$ b = 8.547(2) $\text{\AA}$ c = 5.031(1) $\text{\AA}$		$\alpha = 90$ $\beta = 109.33(2)$ $\gamma = 90$	

Supp. Tab. 3. X-ray refinement parameters for 4.8V charged ${}^7\text{Li}_{0.1}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$

Atom	Wyckoff	x	y	z	Occupancy	U_{iso}
${}^7\text{Li}/\text{Ni}$	4h	0	0.333	0.5	0.1/0.03	0.0094(7)
${}^7\text{Li}/\text{Ni}$	2c	0	0	0.5	0.1/0.03	0.0094(7)
Ni/Co/Mn	2b	0	0.5	0	0.10/0.13/0.54	0.0094(7)
Ni/Co/Mn	4g	0	0.167	0	0.10/0.13/0.54	0.0094(7)
O	8j	0.251(5)	0.335(2)	0.235(6)	1	0.0094(7)
O	4i	0.235(1)	0	0.246(1)	1	0.0094(7)
S.G. = C2/m		a = 4.912(2) Å			$\alpha = 90$	
$R_w = 0.0252$		b = 8.493(1) Å			$\beta = 108.94(1)$	
G.O.F. = 1.2		c = 4.948(2) Å			$\gamma = 90$	

Supp. Tab. 4. Neutron PDF refinement parameters for 4.8V charged ${}^7\text{Li}_{0.1}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$

Atom	Wyckoff	x	y	z	Occupancy
${}^7\text{Li}$	4h	0	0.333	0.5	0.1
${}^7\text{Li}$	2c	0	0	0.5	0.1
Ni/Co/Mn	2b	0	0.5	0	0.13/0.13/0.54
Ni/Co/Mn	4g	0	0.167	0	0.13/0.13/0.54
O	8j	0.246(2)	0.337(1)	0.227(1)	1
O	4i	0.237(1)	0	0.200(2)	1
S.G. = C2/m, $R_w = 0.193$, $Q_{\text{max}} = 26 \text{ \AA}$, $r = 0.5 - 40 \text{ \AA}$		$a = 4.919(4) \text{ \AA}$ $b = 8.486(8) \text{ \AA}$ $c = 4.975(3) \text{ \AA}$		$\alpha = 90$ $\beta = 109.97(6)$ $\gamma = 90$	

Supp. Tab. 5. Neutron PDF refinement parameters for 4.8V charged ${}^7\text{Li}_{0.1}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ **a without and **b** with Ni migration.**

a

Atom	Wyckoff	x	y	z	Occupancy
${}^7\text{Li}$	4h	0	0.333	0.5	0.1
${}^7\text{Li}$	2c	0	0	0.5	0.1
Ni/Co/Mn	2b	0	0.5	0	0.13/0.13/0.54
Ni/Co/Mn	4g	0	0.167	0	0.13/0.13/0.54
O	8j	0.250(2)	0.330(3)	0.214(5)	1
O	4i	0.256(3)	0	0.201(9)	1
S.G. = C2/m		a = 5.02(3) Å		α = 90	
R _w = 0.193,		b = 8.28 (4) Å		β = 111.9(4)	
Q _{max} = 26 Å		c = 5.04(3) Å		γ = 90	
r = 0.5 – 4 Å					

b

Atom	Wyckoff	x	y	z	Occupancy
⁷ Li/Ni	4h	0	0.333	0.5	0.1/0.031(7)
⁷ Li/Ni	2c	0	0	0.5	0.1/0.031(7)
Ni/Co/Mn	2b	0	0.5	0	0.099(7)/0.13/0.54
Ni/Co/Mn	4g	0	0.167	0	0.099(7)/0.13/0.54
O	8j	0.250(2)	0.329(2)	0.214(4)	1
O	4i	0.256(3)	0	0.200(8)	1
S.G. = C2/m		a = 5.01(5) Å		α = 90	
R _w = 0.166,		b = 8.28 (7) Å		β = 111.7(5)	
Q _{max} = 26 Å		c = 5.04(3) Å		γ = 90	
r = 0.5 – 4 Å					

Supp. Tab. 6. Neutron PDF refinement parameters for 4.8V charged ${}^7\text{Li}_{0.1}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ with Ni migration and solid beta- O_2 phase.

${}^7\text{Li}_{0.1}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}\text{O}_2$ phase fraction = 0.34

Atom	Wyckoff	x	y	z	Occupancy
${}^7\text{Li}/\text{Ni}$	4h	0	0.333	0.5	0.1/0.031
${}^7\text{Li}/\text{Ni}$	2c	0	0	0.5	0.1/0.031
Ni/Co/Mn	2b	0	0.5	0	0.099/0.13/0.54
Ni/Co/Mn	4g	0	0.167	0	0.099/0.13/0.54
O	8j	0.250	0.329	0.214	1
O	4i	0.256	0	0.200	1
S.G. = C2/m $R_w = 0.138$, $Q_{\text{max}} = 26 \text{ \AA}$ $r = 0.5 - 4 \text{ \AA}$					
		$a = 5.01 \text{ \AA}$ $b = 8.28 \text{ \AA}$ $c = 5.04 \text{ \AA}$		$\alpha = 90$ $\beta = 111.7$ $\gamma = 90$	

Beta- O_2 phase fraction = 0.66

Atom	Wyckoff	x	y	z	Occupancy
O	6c	0	0	0.0536	1
S.G. = R-3m $R_w = 0.138$, $Q_{\text{max}} = 26 \text{ \AA}$ $r = 0.5 - 4 \text{ \AA}$					
		$a = 3.307 \text{ \AA}$ $b = 3.307 \text{ \AA}$ $c = 11.26 \text{ \AA}$		$\alpha = 90$ $\beta = 90$ $\gamma = 120$	

Spherical particle diameter (spdiameter = $3.6(8) \text{ \AA}$) was also refined to reduce the contribution of intermolecular correlations to the PDF.