

Table S1 The kinetics equation and particle dispersion model parameters for Se(IV).

Pseudo-first order			Pseudo-second order		
$Q_e$	$k_1$	$R^2$	$Q_e$	$k_2$	$R^2$
105.535	0.059	0.655	106.383	0.015	0.999

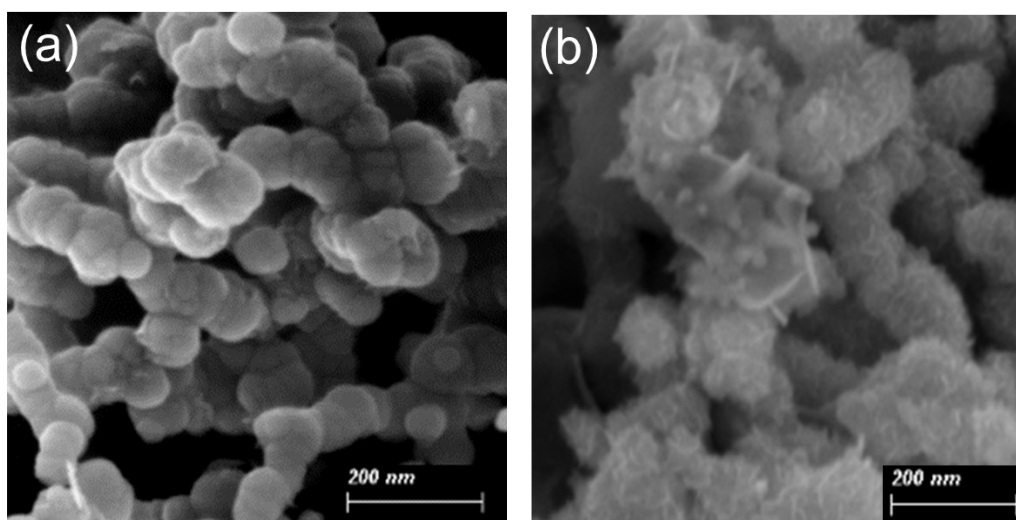


Fig. S1 Scanning electron microscope (SEM) of nZVI. (a) Fresh nZVI; (b) Spent nZVI after the reactions with partial sodium selenate solution (Se(IV)).

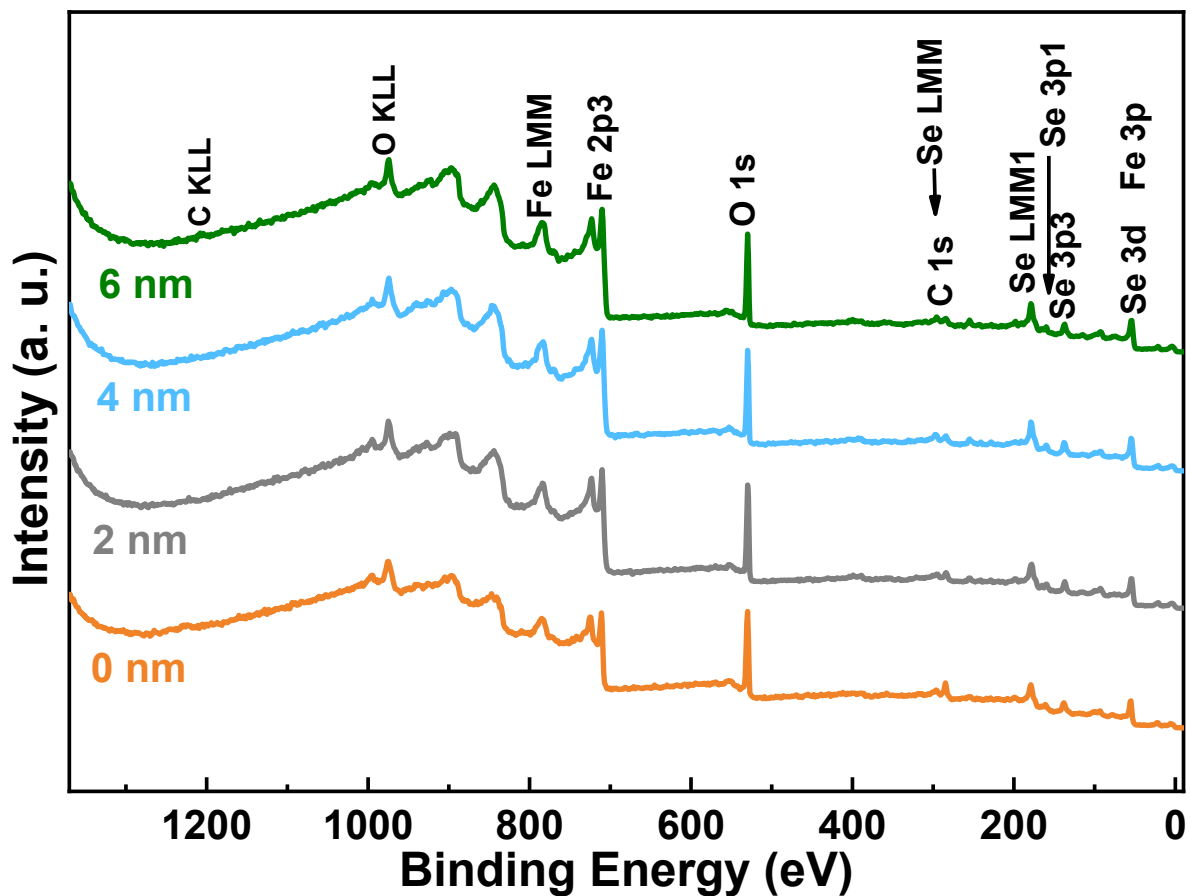


Fig. S2 Survey spectrum of HR-XPS spectra of Se-nZVI.  $[\text{Se(IV)}]_{\text{initial}} = 1.3 \text{ mM}$ ; the nZVI dose is  $1.0 \text{ g L}^{-1}$ , the  $\text{pH}_{\text{initial}}$  is 5.0, and the reaction time is 24 h.

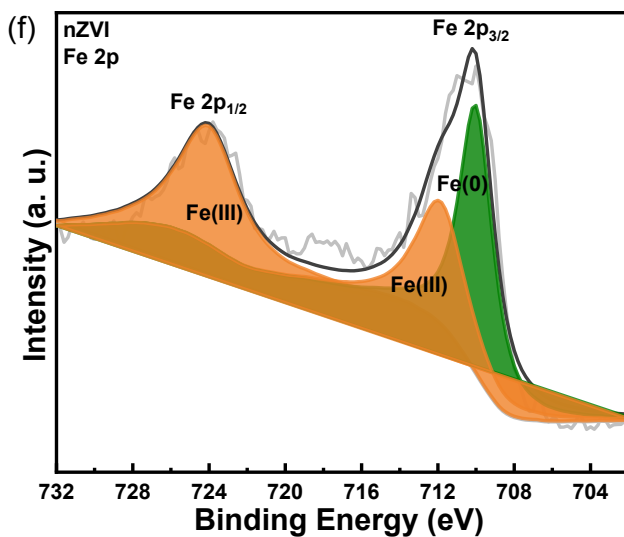
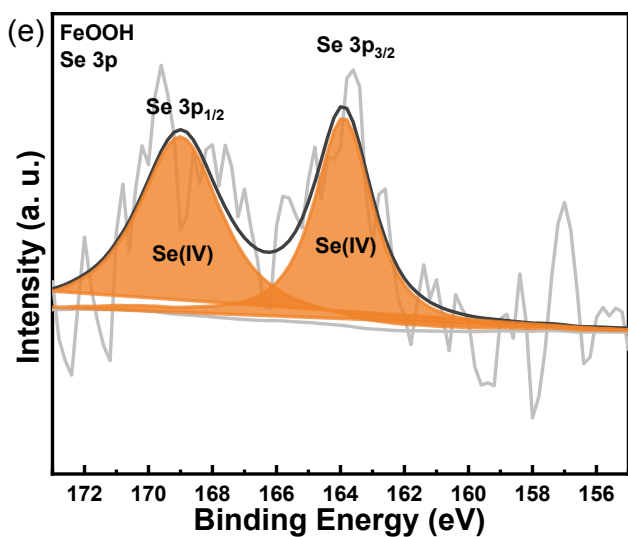
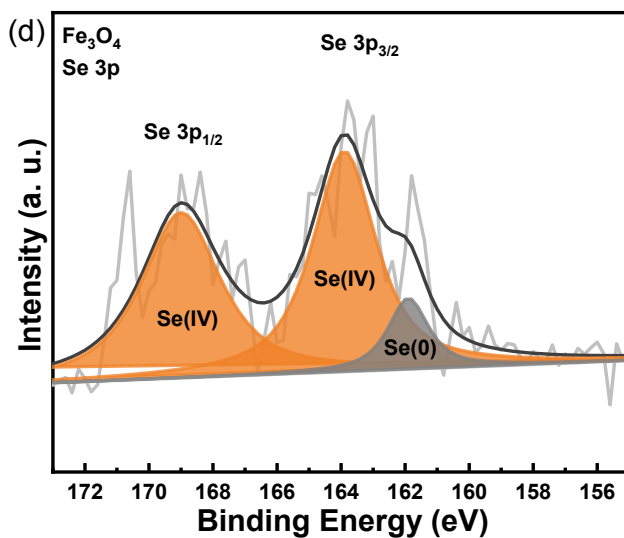
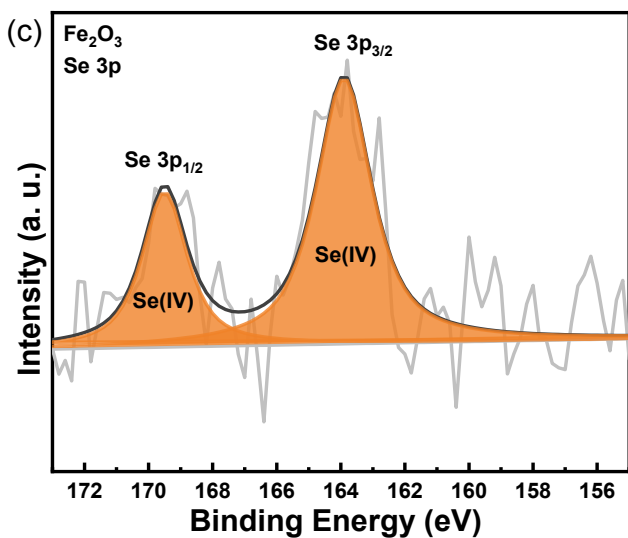
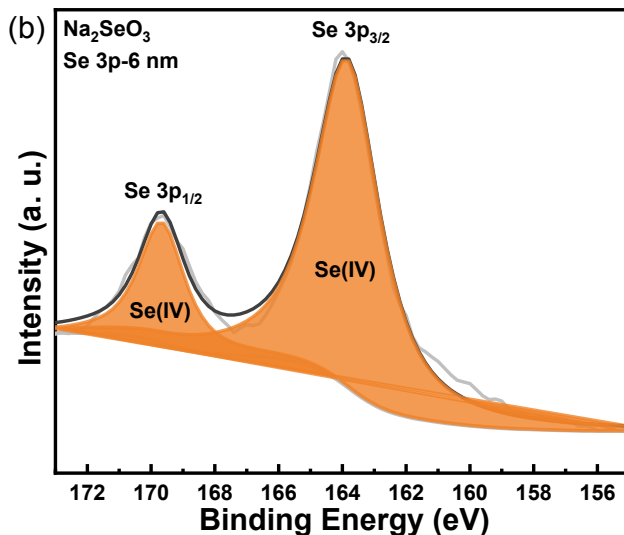
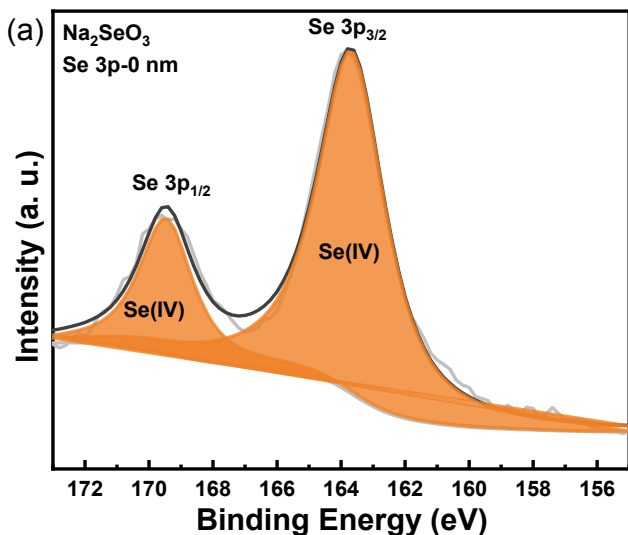


Fig. S3 HR-XPS spectra of Se 3p and Fe 2p species at oxide layer. (a) XPS spectra of Se 3p element in 0 nm of Na<sub>2</sub>SeO<sub>3</sub>; (b) XPS spectra of Se 3p element in 6 nm of Na<sub>2</sub>SeO<sub>3</sub>; (c) XPS spectra of Se 3p element in Fe<sub>2</sub>O<sub>3</sub>-Se(IV); (d) XPS spectra of Se 3p element in Fe<sub>3</sub>O<sub>4</sub>-Se(IV); (e) XPS spectra of Se 3p element in FeOOH-Se(IV); (f) XPS spectra of Fe 2p species in fresh nZVI. [Se(IV)]<sub>initial</sub> = 1.3 mM, the nZVI dose is 1.0 g L<sup>-1</sup>, the pH<sub>initial</sub> is 5.0, and the reaction time is 24 h.

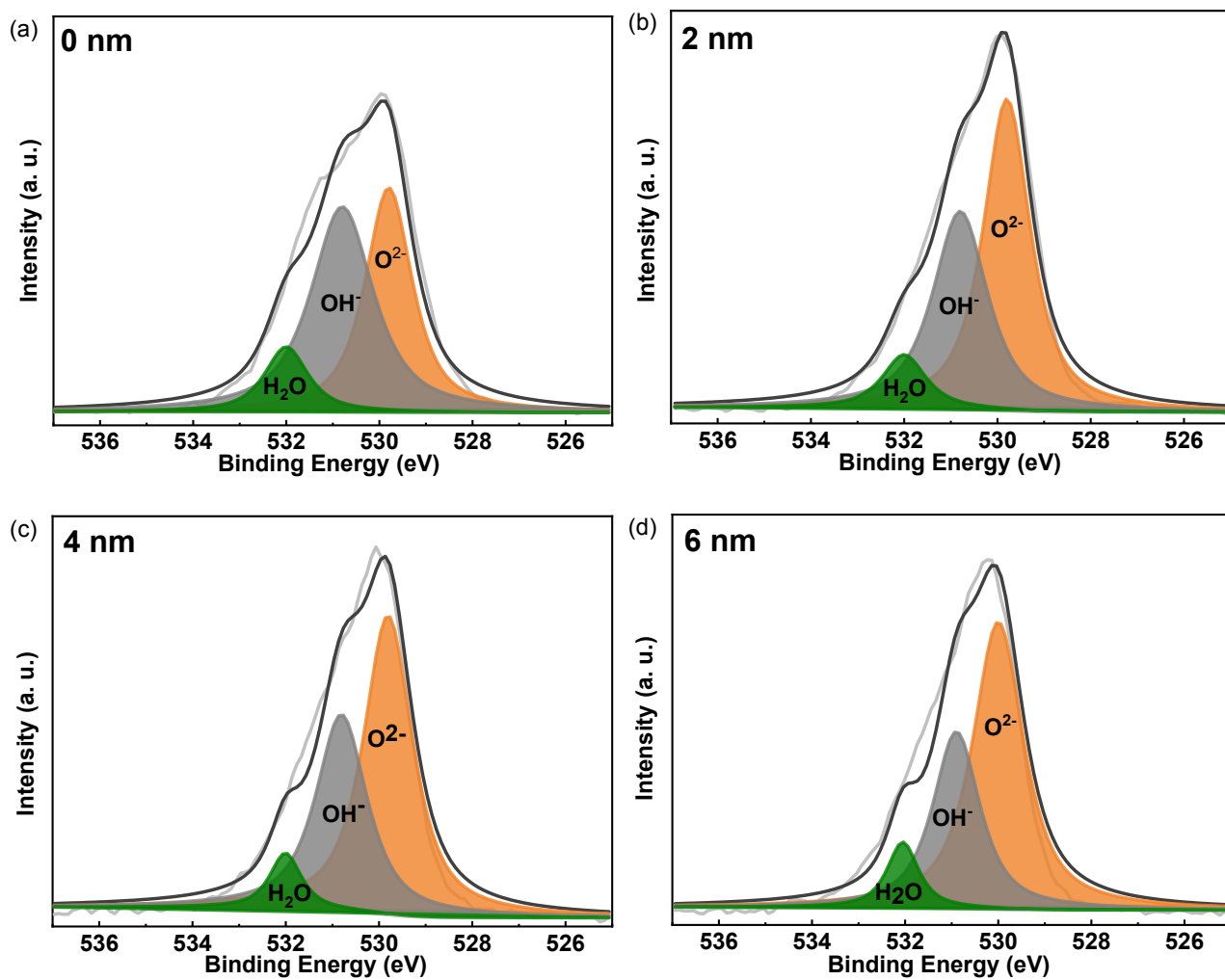


Fig. S4 HR-XPS spectra of O 1s species at selected depth. (a) 0 nm; (b) 2 nm; (c) 4 nm; (d) 6 nm.  $[\text{Se(IV)}]_{\text{initial}} = 1.3 \text{ mM}$ ; the nZVI dose is  $1.0 \text{ g L}^{-1}$ , the  $\text{pH}_{\text{initial}}$  is 5.0, and the reaction time is 24 h.