

Electronic supplementary Information (ESI)

*For*

Boosting Heterogeneous Fenton Reactions for  
Degrading Organic Dyes via Photothermal  
Effect under Neutral Conditions

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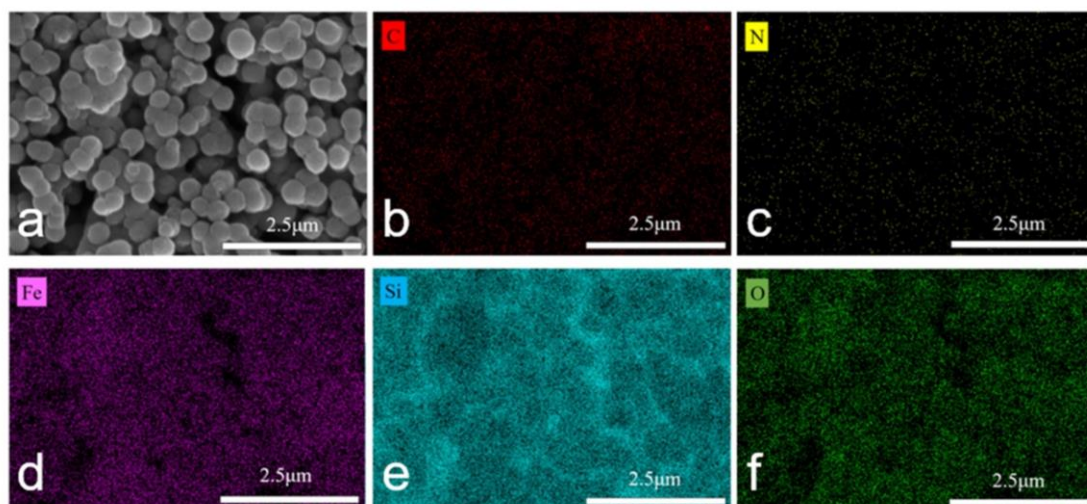
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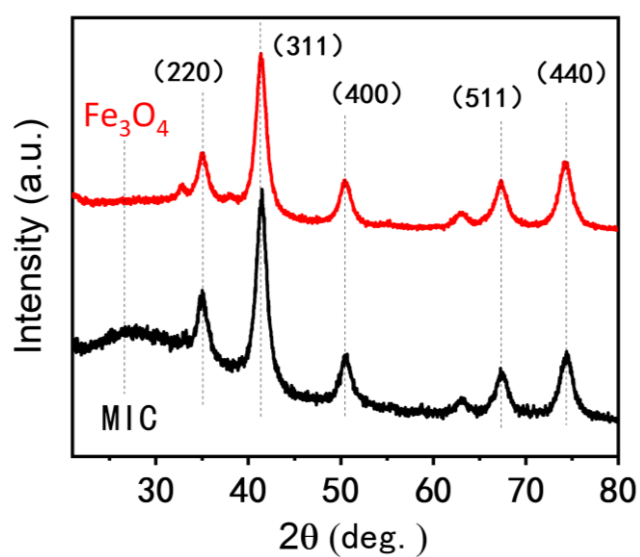
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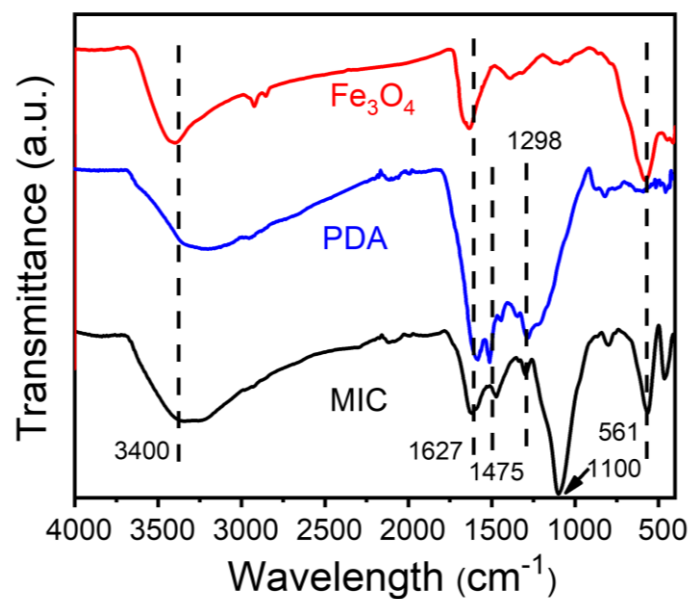
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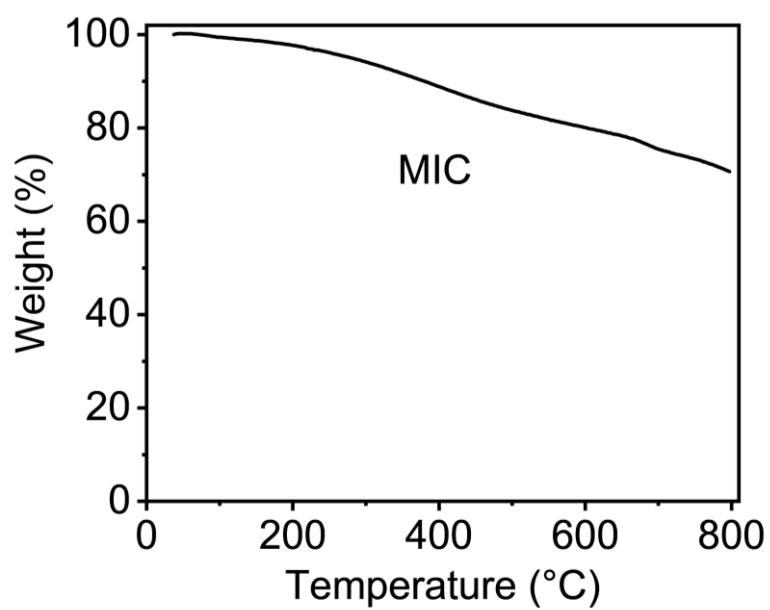
**Figure S1.** Distribution of MIC elements: (a) SEM; (b) C; (c) N; (d) Fe; (e) Si; (f) O.



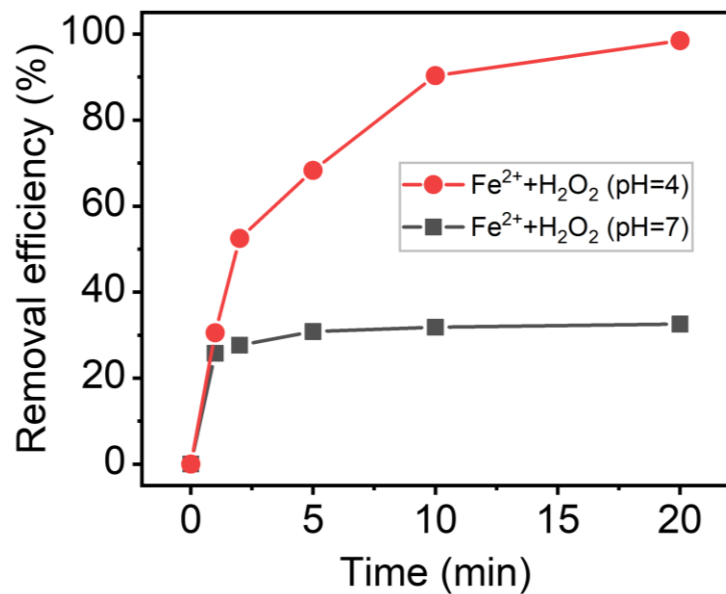
**Figure S2.** XRD pattern of the  $\text{Fe}_3\text{O}_4$  nanoparticles and MIC.



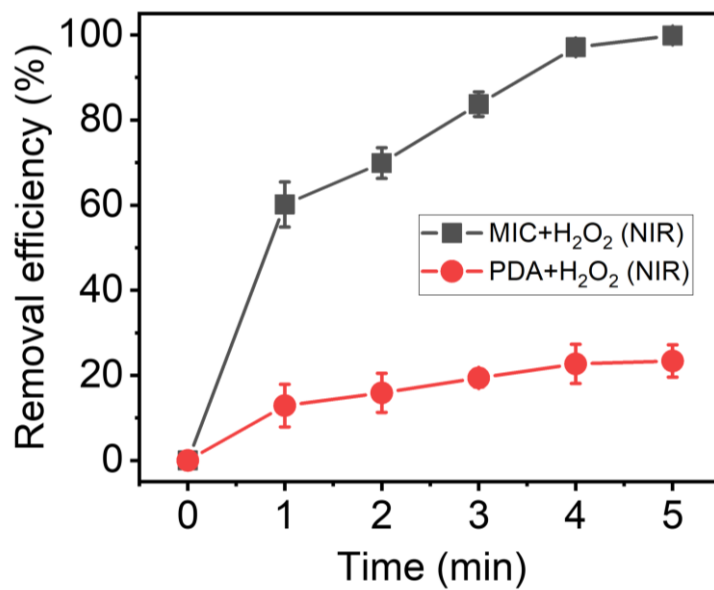
**Figure S3.** FTIR spectra of ultra-small  $\text{Fe}_3\text{O}_4$  nanoparticles, PDA and MIC.



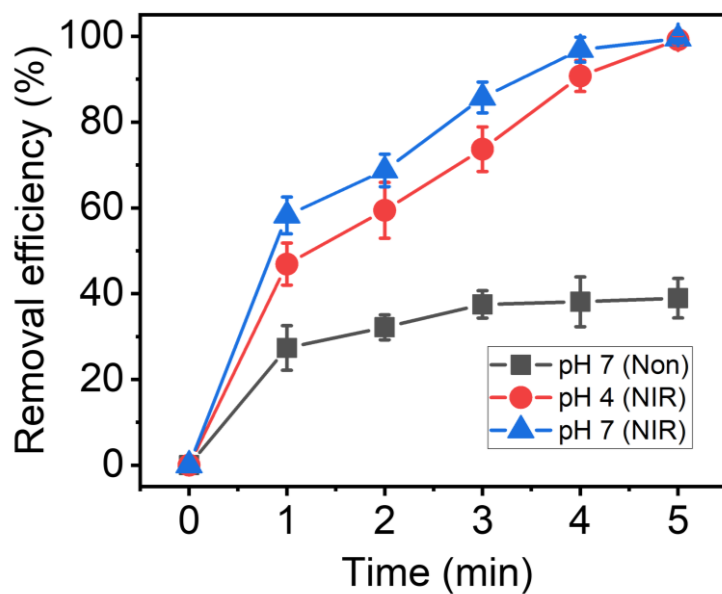
**Figure S4.** Thermogravimetric analysis (TGA) plot curve of MIC under with nitrogen atmosphere.



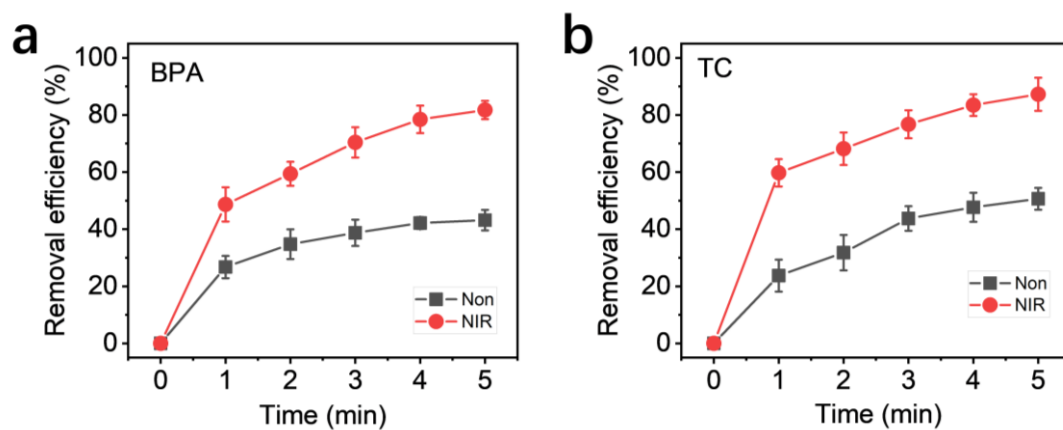
**Figure S5.** Homogeneous Fenton system (Fe<sup>2+</sup>+H<sub>2</sub>O<sub>2</sub>) degraded MB solution at pH=4 and pH=7.



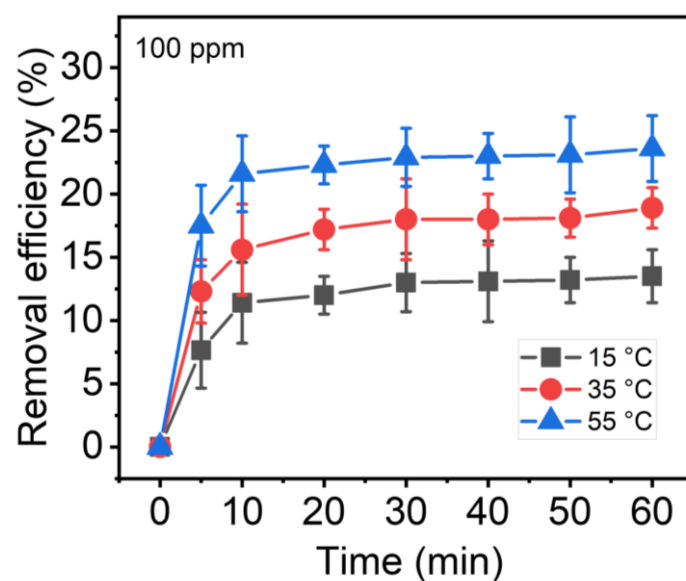
**Figure S6.** Removal efficiency of MIC+H<sub>2</sub>O<sub>2</sub> and PDA+H<sub>2</sub>O<sub>2</sub> to MB at different time.



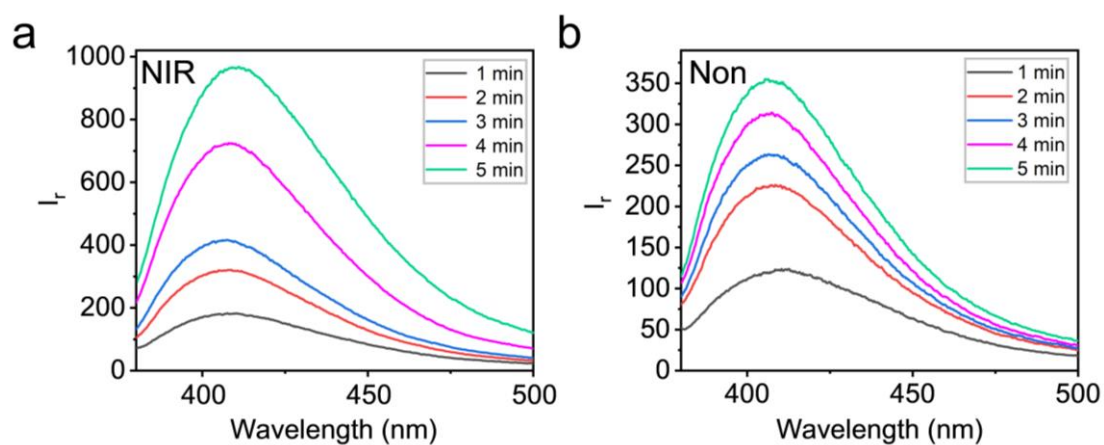
**Figure S7.** Effect of acid (pH 4) and neutral conditions on dye removal efficiency of MIC.



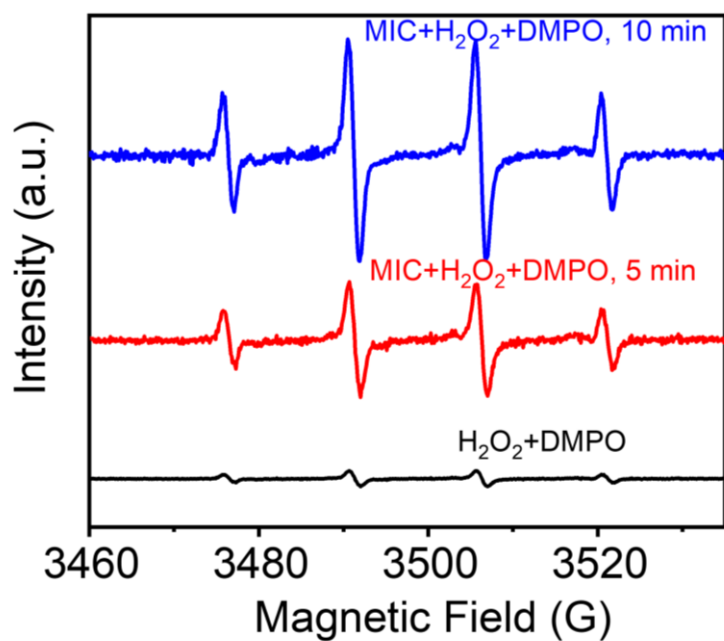
**Figure S8.** Degradation efficiency of MIC to BPA(a) and TC(b), (experimental conditions: MIC:  $250 \mu\text{g mL}^{-1}$ ; BPA:  $22.8 \text{ mg/L}$ , TC:  $100 \text{ mg/L}$ , pH 7).



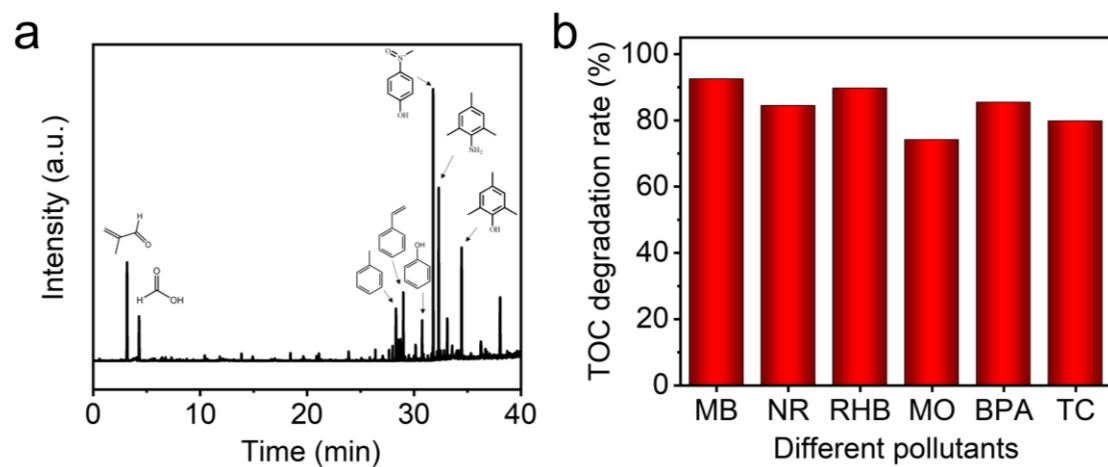
**Figure S9.** Effect of different temperatures on adsorption efficiency of MB by MIC (experimental conditions: MIC: 250  $\mu\text{g mL}^{-1}$ ; MB: 100 mg/L, pH 7).



**Figure S10.** Fluorescence spectra of 2-hydroxyterephthalic acid (a) with or (b) without the NIR laser.



**Figure S11.** The EPR spectra of DMPO-OH adducts in the MIC+H<sub>2</sub>O<sub>2</sub> system.



**Figure S12.** (a) GC-MS spectra in the degradation process of MB and (b) total organic carbon (TOC) degradation rate of different organic pollutants.

**Table S1.** Catalytic performance of multifarious Iron-based materials in different physical fields.

Catalysts/ concentration	Dye concentration (mg/L)	Time (min)	Dye removal rate (%)	pH	Physical field	Refer- ences
PCN-250 (Fe <sub>2</sub> Mn)/327 $\mu\text{g mL}^{-1}$	15	270	100	2.0- 12.0	Full- wavelength halogen lamp	1
Fe <sup>3+</sup> /2.8 $\mu\text{g mL}^{-1}$	6	8	100	3.3	VUV/UV	2
BASF-NPs/250 $\mu\text{g mL}^{-1}$	10	140	100	7.0	UV	3
Fe <sup>2+</sup> /20 $\mu\text{g mL}^{-1}$	50	6	98	3.0	Microwave heating	4
Fe <sup>0</sup> /1000 $\mu\text{g mL}^{-1}$	35	5	99	3.0	Ultrasound	5
Fe <sub>3</sub> O <sub>4</sub> /ZnO/grap- hene	40	60	100	13.0	UV+US	6
SUS/Fe <sub>3</sub> O <sub>4</sub> /200 $\mu\text{g mL}^{-1}$	10	150	100	7.0	E beam	7
<b>MIC (this work)</b> /250 $\mu\text{g mL}^{-1}$	100	5	100	7.0	NIR irradiation	

## References

1. A. Kirchon, P. Zhang, J. Li, E. A. Joseph, W. Chen and H. C. Zhou, Effect of Isomorphic Metal Substitution on the Fenton and Photo-Fenton Degradation of Methylene Blue Using Fe-Based Metal-Organic Frameworks. *ACS Appl. Mater. Interfaces*, 2020, **12**, 9292-9299.
2. M. Li, Z. Qiang, C. Pulgarin and J. Kiwi, Accelerated methylene blue (MB) degradation by Fenton reagent exposed to UV or VUV/UV light in an innovative micro photo-reactor. *Appl. Catal., B*, 2016, **187**, 83-89.
3. G. A. Ashraf, R. T. Rasool, M. Hassan and L. Zhang, Enhanced photo Fenton-like activity by effective and stable Al-Sm M-hexaferrite heterogenous catalyst magnetically detachable for methylene blue degradation. *J. Alloy. Compd.*, 2020, **821**, 153410.



4. S.-T. Liu, J. Huang, Y. Ye, A.-B. Zhang, L. Pan and X.-G. Chen, Microwave enhanced Fenton process for the removal of methylene blue from aqueous solution. *Chem. Eng. J.*, 2013, **215-216**, 586-590.
5. C.-H. Weng and V. Huang, Application of  $\text{Fe}^0$  aggregate in ultrasound enhanced advanced Fenton process for decolorization of methylene blue. *J. Ind. Eng. Chem.*, 2015, **28**, 153-160.
6. R. Saleh and A. Taufik, Degradation of methylene blue and congo-red dyes using Fenton, photo-Fenton, sono-Fenton, and sonophoto-Fenton methods in the presence of iron(II,III) oxide/zinc oxide/graphene ( $\text{Fe}_3\text{O}_4/\text{ZnO}/\text{graphene}$ ) composites. *Sep. Purif. Technol.*, 2019, **210**, 563-573.
7. Y. J. Choe, J. Kim, J. Y. Byun and S. H. Kim, An electro-Fenton system with magnetite coated stainless steel mesh as cathode. *Catal. Today*, 2021, **359**, 16-22.