

## Supplementary Material

Application of Plasma and UV/H<sub>2</sub>O<sub>2</sub> for the Removal of Pharmaceuticals in Synthetic Urine

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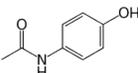
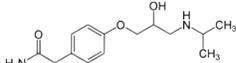
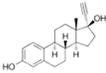
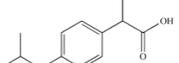
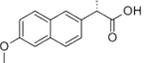
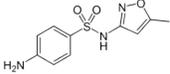
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**Table S1:** Physicochemical characteristics of the six target pharmaceuticals. References for rate constants can be found in figures 2, S1, and S2. Pharmaceutical structures and log  $K_{ow}$  values were found on pubchem. pKas were determined using the SPARC software.

Compound	Structure	pKa	log $K_{ow}$	$k_{OH^-,Pharm}$ [ $M^{-1}s^{-1}$ ]	$k_{O_3^-,Pharm}$ [ $M^{-1}s^{-1}$ ]	$\phi$ [mol Einstein <sup>-1</sup> ]	$\epsilon$ [ $m^2mol^{-1}$ ]	$k_{OH^-,Pharm} / k_{O_3^-,Pharm}$
Acetaminophen		9.38	0.46	2.2E9	1.3E5	4.6E-2	742	1.7E4
Atenolol		9.60	0.16	8.0E9	1.7E3	3.6E-2	52.7	4.7E6
17 $\alpha$ -ethinoestradiol		10.3	3.67	9.8E9	7.4E9	2.6E-2	75	5.4E6
Ibuprofen		5.30	3.97	7.4E9	9.6	8.6E-2	28.2	6.8E8
Naproxen		4.15	3.18	9.6E9	2.0E5	9.3E-3	490	4.8E4
Sulfamethoxazole		1.60 ; 5.70	0.89	6.3E9	2.5E6	2.1E-1	1189	2.5E3

**Table S2:** Recipes for fresh and hydrolyzed synthetic urine.

<b>Chemical Formula</b>	<b>Molecular Weight (g mol<sup>-1</sup>)</b>	<b>Fresh Urine Concentration (mM)</b>	<b>Hydrolyzed Urine Concentration (mM)</b>
CaCl <sub>2</sub> ·H <sub>2</sub> O	129.98	5.0	
MgCl <sub>2</sub> ·6H <sub>2</sub> O	203.21	3.2	
NaCl	58.44	79	60
Na <sub>2</sub> SO <sub>4</sub>	142.04	16	15
Na <sub>3</sub> C <sub>6</sub> H <sub>8</sub> O <sub>7</sub> ·2H <sub>2</sub> O	297.09	2.2	
Na <sub>2</sub> C <sub>2</sub> O <sub>4</sub>	134.00	0.15	
NaH <sub>2</sub> PO <sub>4</sub>	119.98		14
KH <sub>2</sub> PO <sub>4</sub>	139.06	21	
KCl	74.55	21	40
NH <sub>4</sub> Cl	53.49	19	
NH <sub>4</sub> OH	35.04		250
NH <sub>4</sub> HCO <sub>3</sub>	79.06		250
CO(NH <sub>2</sub> ) <sub>2</sub>	60.06	420	
C <sub>4</sub> H <sub>7</sub> N <sub>3</sub> O	113.12	9.7	

**Table S3:** Synthetic urine characteristics.

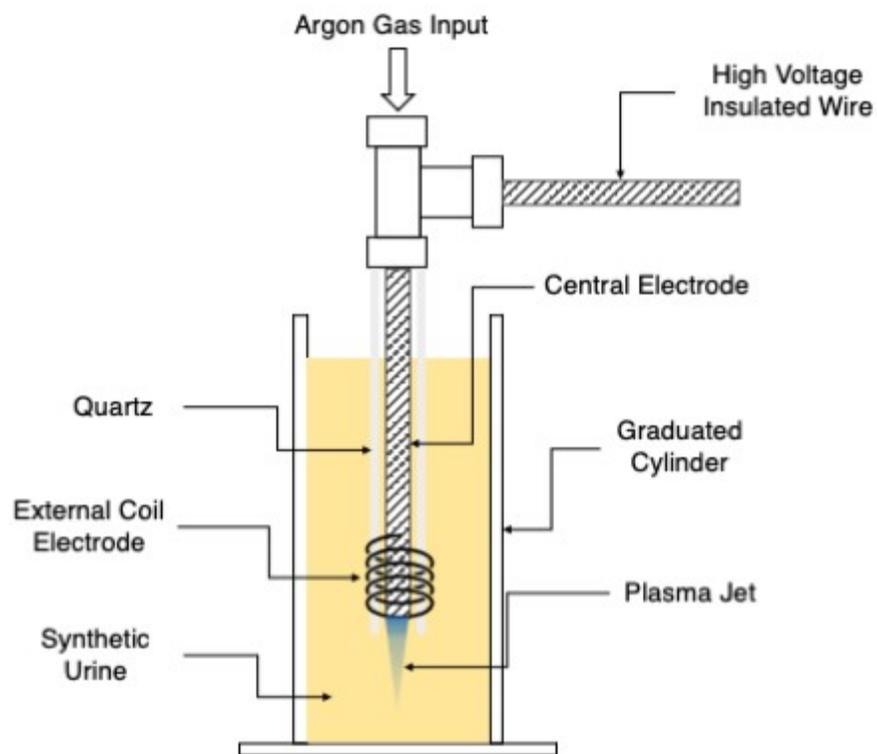
<i>Parameter</i>	<i>Unit</i>	<i>Fresh Urine</i>	<i>Hydrolyzed Urine</i>
<i>pH</i>	-	5.6	9.6
<i>Conductivity</i>	$\text{mS}\cdot\text{cm}^{-1}$	16	36
<i>UV Transmittance</i>	%	0.03	75
<i>Alkalinity</i>	$\text{meq}\cdot\text{L}^{-1}$	22	490

**Table S4:** Target pharmaceuticals. with accurate mass and retention times used for quantification. Acceptable peaks ranged plus or minus 30 seconds from the retention times listed depending on the matrix sampled.

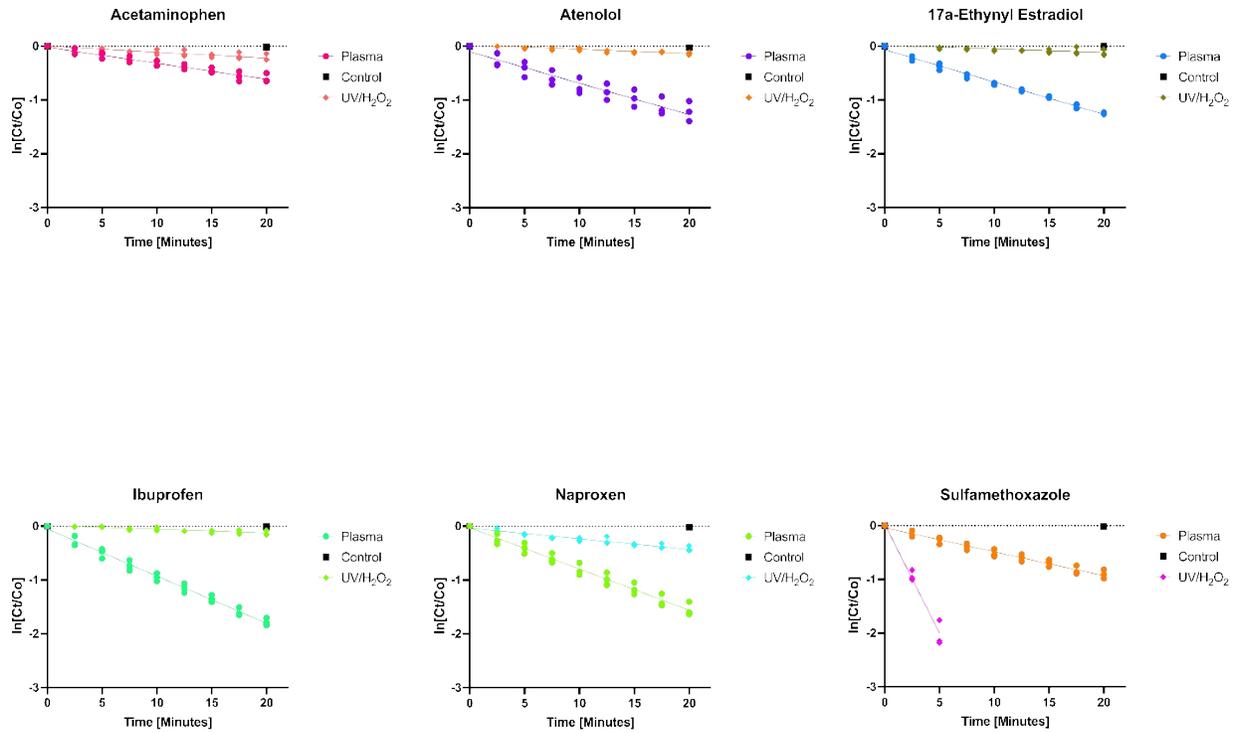
<b>Compound</b>	<b>Formula</b>	<b>Accurate Mass (m/z)</b>	<b>Retention Time (min)</b>	<b>ESI Mode</b>
Acetaminophen	C <sub>8</sub> H <sub>9</sub> NO <sub>2</sub>	152.07060	5.25	Positive
d3-Acetaminophen	C <sub>8</sub> H <sub>6</sub> D <sub>3</sub> NO <sub>2</sub>	155.08943	5.22	Positive
Atenolol	C <sub>14</sub> H <sub>22</sub> N <sub>2</sub> O <sub>3</sub>	267.17031	4.92	Positive
d7-atenolol	C <sub>14</sub> H <sub>15</sub> D <sub>7</sub> N <sub>2</sub> O <sub>3</sub>	274.21425	4.88	Positive
17 $\alpha$ -ethinoestradiol	C <sub>20</sub> H <sub>24</sub> O <sub>2</sub>	279.17434	10.96	Positive
d4-17 $\alpha$ -ethinoestradiol	C <sub>20</sub> H <sub>20</sub> D <sub>4</sub> O <sub>2</sub>	283.19944	10.95	Positive
Ibuprofen	C <sub>13</sub> H <sub>18</sub> O <sub>2</sub>	205.12231	11.61	Positive
d3-ibuprofen	C <sub>13</sub> H <sub>15</sub> D <sub>3</sub> O <sub>2</sub>	208.14241	11.60	Positive
Naproxen	C <sub>14</sub> H <sub>14</sub> O <sub>3</sub>	229.08592	10.64	Positive
d3-Naproxen	C <sub>14</sub> H <sub>11</sub> D <sub>3</sub> O <sub>3</sub>	232.10475	10.62	Positive
Sulfamethoxazole	C <sub>10</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub> S	254.05938	7.23	Positive
d4-sulfamethoxazole	C <sub>10</sub> H <sub>7</sub> D <sub>4</sub> N <sub>3</sub> O <sub>3</sub> S	258.08449	7.22	Positive

**Table S5:** Observed rate constants and for each pharmaceutical across several matrices. \* denotes observed rate constants that were not statistically significantly non-zero.

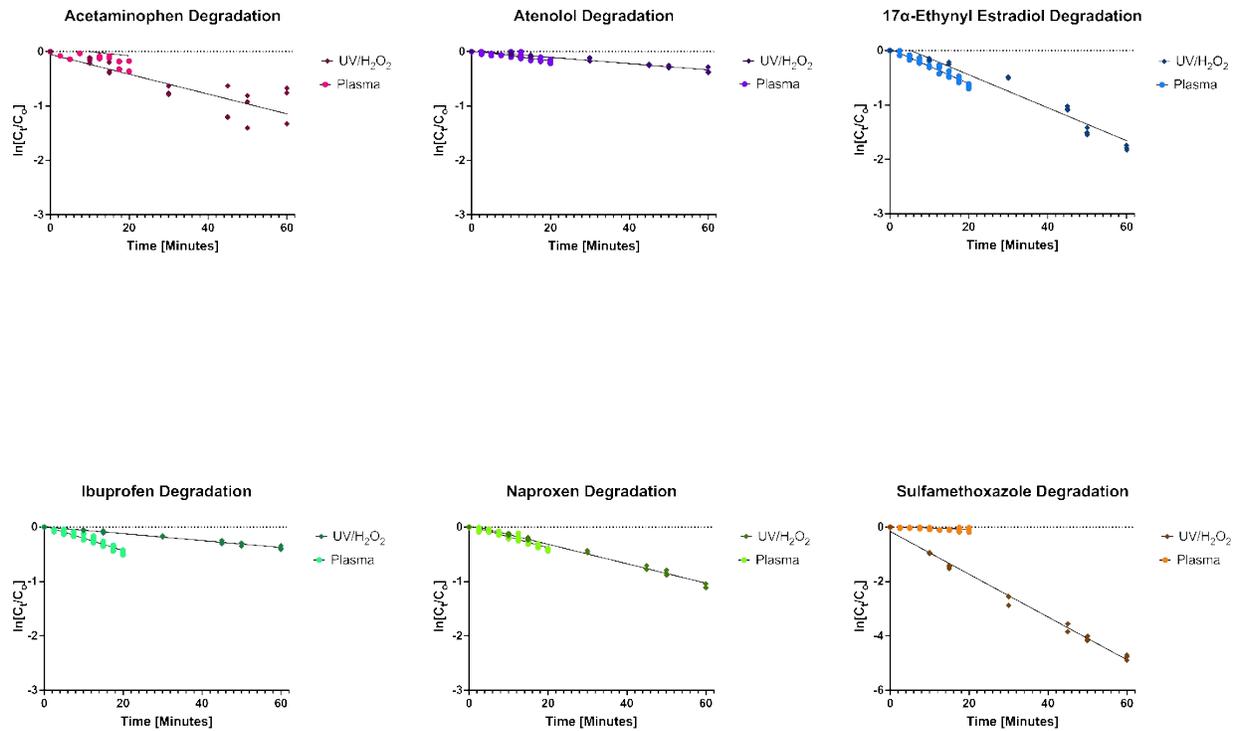
	<b>Observed Rate Constant (min<sup>-1</sup>)</b>					
	<b>Acetaminophen</b>	<b>Atenolol</b>	<b>EE2</b>	<b>Ibuprofen</b>	<b>Naproxen</b>	<b>Sulfamethoxazole</b>
	<i>UV/H<sub>2</sub>O<sub>2</sub></i>					
Nanopure Water	1.84E-04	1.25E-04	1.05E-04	1.17E-04	3.31E-04	6.75E-03
Hydrolyzed Urine	3.04E-04	9.67E-05	5.06E-04	1.06E-04	2.98E-04	1.31E-03
Fresh Urine	6.79E-06*	2.86E-06*	6.20E-06*	3.13E-06*	1.66E-05*	1.36E-04
	<i>Plasma</i>					
Nanopure Water	4.95E-04	9.72E-04	1.00E-03	1.46E-03	1.28E-03	7.44E-04
Hydrolyzed Urine	1.31E-04	1.50E-04	5.31E-04	3.79E-04	3.38E-04	7.68E-05
Fresh Urine	7.84E-05	7.98E-05	3.72E-04	2.64E-04	2.37E-04	1.27E-04



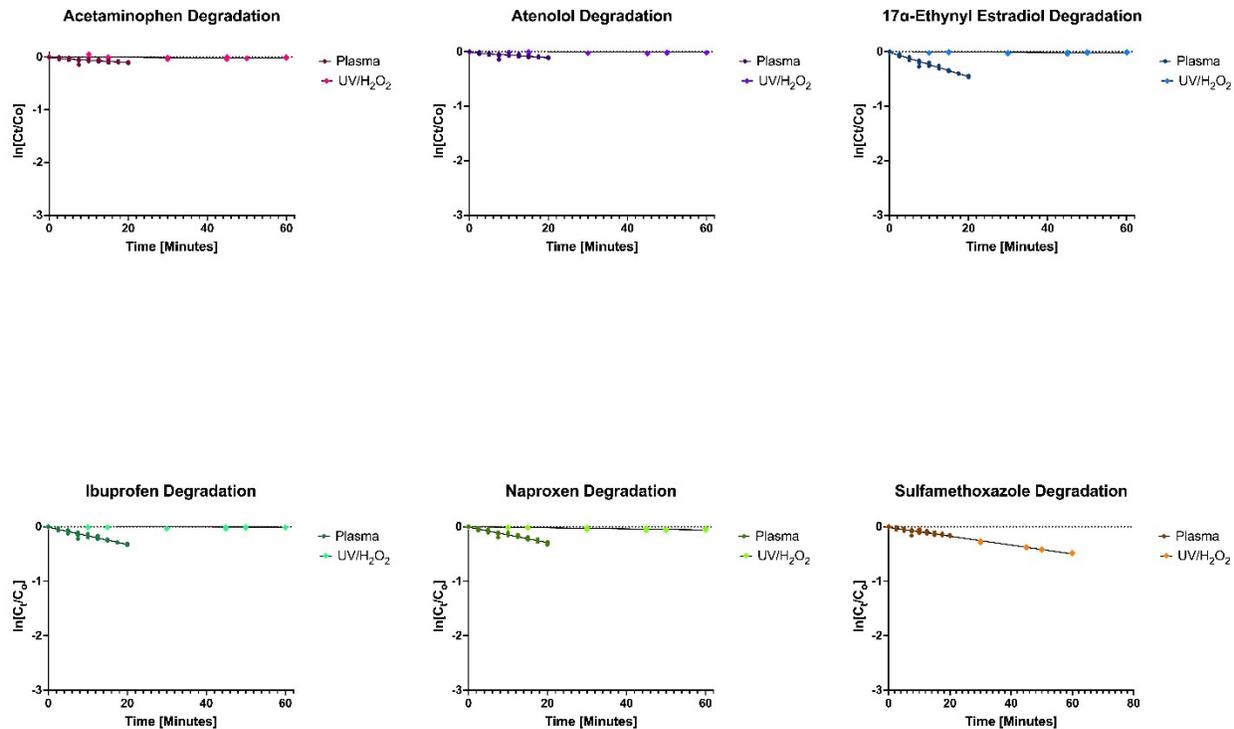
**Figure S1:** Plasma reactor schematic.



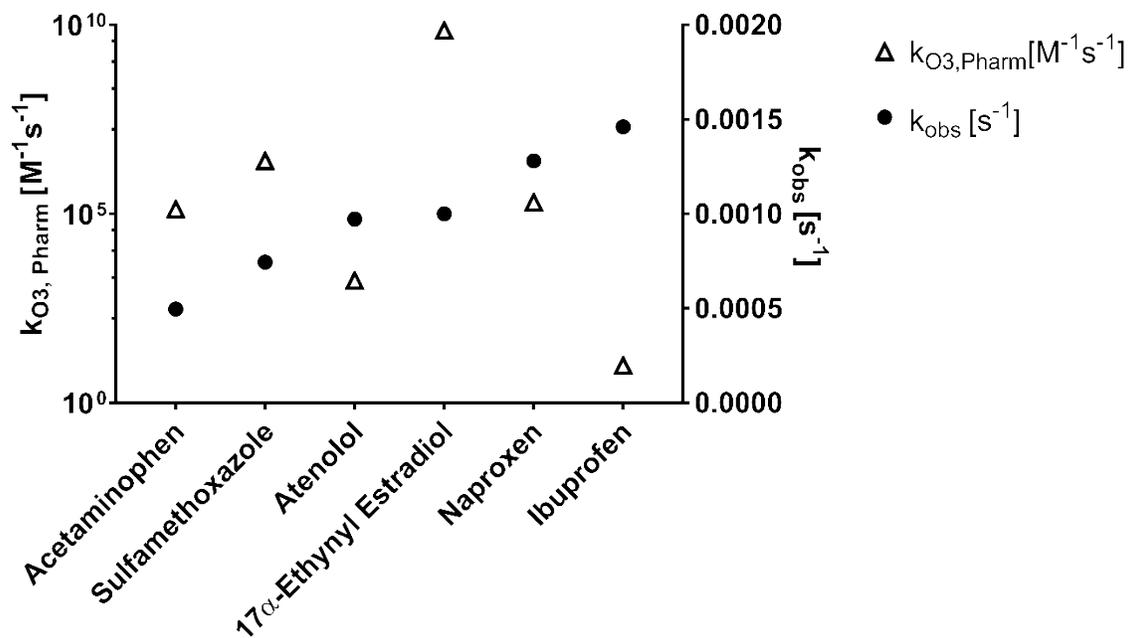
**Figure S2:** Degradation over time plots for each pharmaceutical in nanopure water undergoing treatment by plasma and UV/H<sub>2</sub>O<sub>2</sub>.



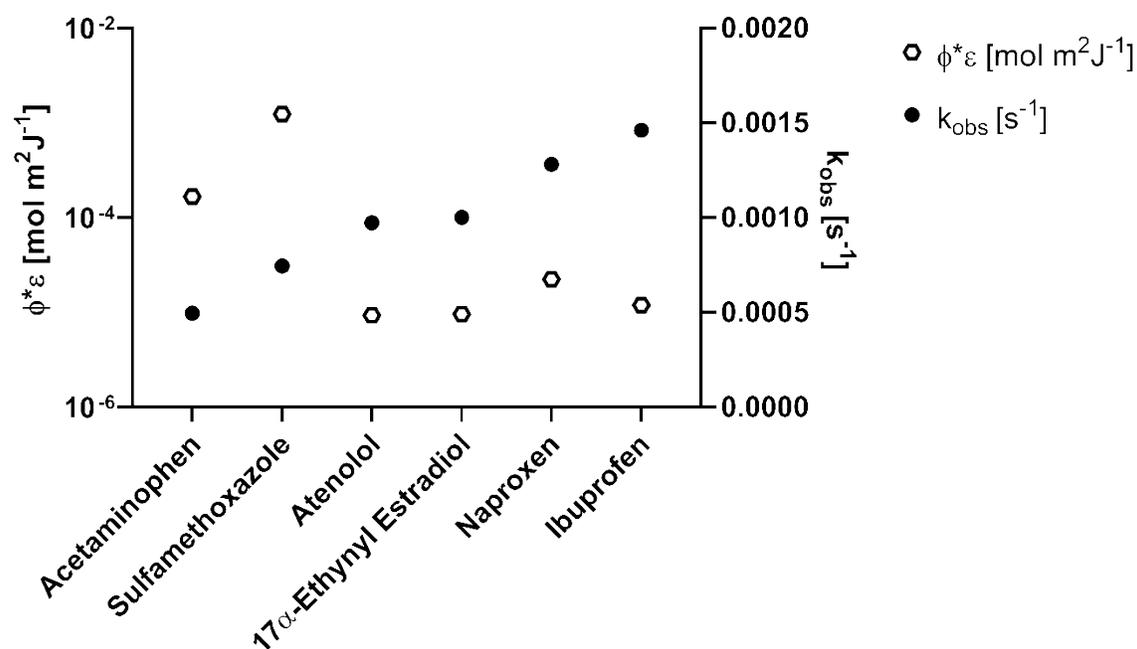
**Figure S3:** Degradation over time plots for each pharmaceutical in synthetic hydrolyzed urine undergoing treatment by plasma and UV/ $H_2O_2$ .



**Figure S4:** Degradation over time plots for each pharmaceutical in synthetic fresh urine undergoing treatment by plasma and UV/H<sub>2</sub>O<sub>2</sub>.



**Figure S5:** Second-order rate constants reported in the literature<sup>1-4</sup> for each pharmaceutical with ozone presented on the left y-axis. Values are presented using a log scale due to the large variation between each constant. Observed first-order rate constants for each pharmaceutical in nanopure water treated by plasma are presented on the right y-axis.



**Figure S6:** Susceptibility to direct photolysis is presented for each pharmaceutical by displaying the quantum yield ( $\phi$ ) multiplied by the molar extinction coefficient ( $\epsilon$ ).<sup>5-7</sup> The left y-axis is presented using a log scale to view the variability among constants. Observed first-order rate constants for each pharmaceutical in nanopure water treated by plasma are presented on the right y-axis.