

## Supporting Information

### Engineering PtCu Nanoparticles for Highly Efficient Methanol Electro-oxidation

#### Reaction

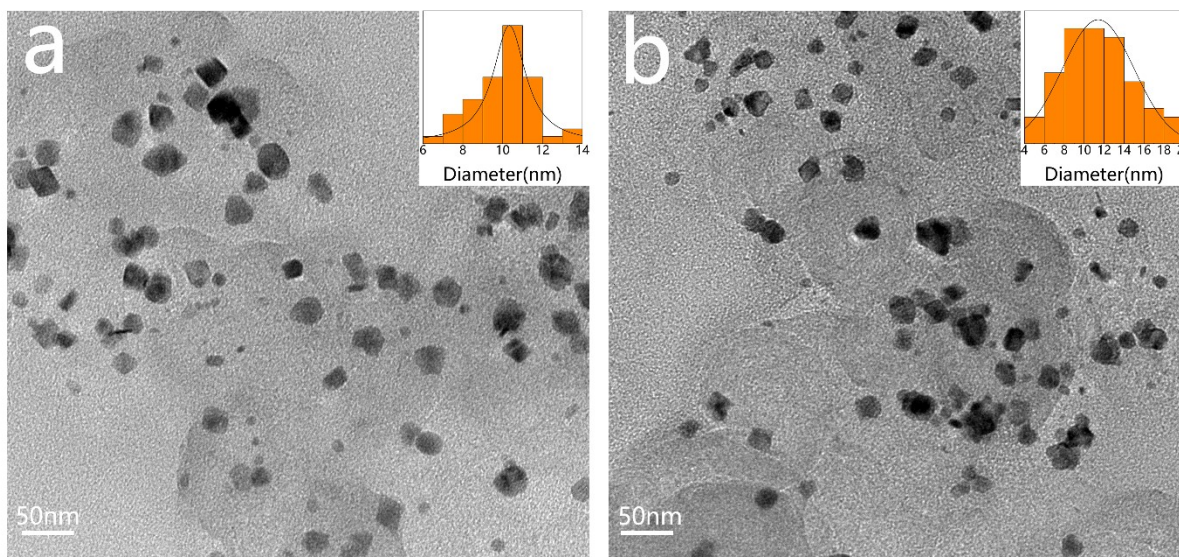
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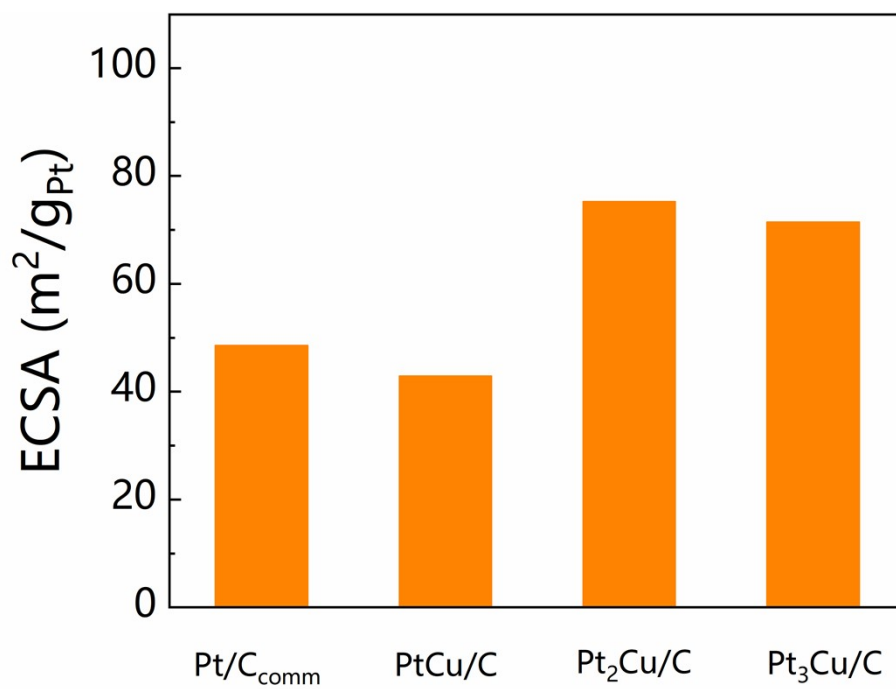
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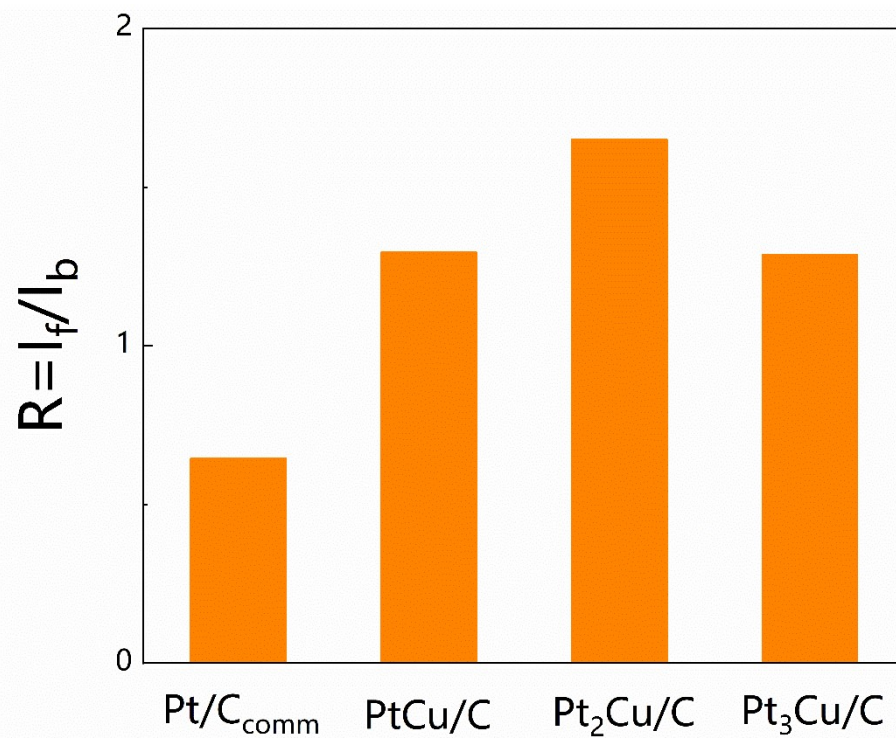
#### Supporting figures and table



**Figure S1.** TEM image of **a.** PtCu/C and **b.** Pt<sub>3</sub>Cu/C.



**Figure S2.** The ECSA of different catalysts calculated based on H<sub>UPD</sub> region.



**Figure. S3** I<sub>f</sub>/I<sub>b</sub> values of different catalysts.

**Table S1.** MOR performance of PtCu bimetallic electrocatalysts published in recent year.

Catalysts	Mass activity (A/mg <sub>Pt</sub> )	Specific activity (mA/cm <sup>2</sup> )	Journal/year
<b>PtCu Alloy</b>	4.1	5.4	in this work
<b>Porous Pt-Cu<sup>1</sup></b>	0.3	1.7	Chem.-Eur. J. 2021
<b>PtCu nanodendrite<sup>2</sup></b>	3.1	2.7	Chem. Comm. 2020
<b>Hollow PtCu nanoring<sup>3</sup></b>	2.2	5.2	J. Mater. Chem. A 2020
<b>PtCu octahedral<sup>4</sup></b>	0.5	2.5	Catal. Sci. Technol. 2020
<b>PtCu alloy nanoclusters<sup>5</sup></b>	1.3	1.8	Electrochim. Acta, 2019
<b>PtCu octahedral<sup>6</sup></b>	1.4	4.1	Nanoscale, 2018
<b>PtCu nanotube<sup>7</sup></b>	2.3	6.1	Energy Environ. Sci.,2017
<b>PtCu Nanoframe<sup>8</sup></b>	0.2	1.7	Adv. Mater.,2016

## Reference

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