## Supplementary Information

## Exploring Hitherto Uninvestigated Reactions of the Fatty Acid Peroxygenase CYP152A1: Catalase Reaction and Compound I Formation

Hiroki Onoda<sup>a,b</sup>, Shota Tanaka<sup>a</sup>, Yoshihito Watanabe<sup>a</sup>, Osami Shoji\*<sup>a</sup>.

<sup>a</sup>Department of Chemistry, Graduate School of Science, Nagoya University, Furo-cho, Chikusa-ku, Nagoya 464-0802, Japan, <sup>b</sup>Department of Medical LifeScience, Graduate School of Medical LifeScience, Yokohama City University, Suehiro-cho, Tsurumi-ku, Yokohama 230-0045, Japan.

Address corresponding to:

Osami Shoji, Department of Chemistry, Graduate School of Science, Nagoya University, Furo-cho, Chikusa-ku, Nagoya 464-0802, Japan. Email: shoji.osami@a.mbox.nagoya-u.ac.jp

## **Supplemental Figures**



Figure S1. Time course of  $O_2$  generation under 4 mM Hydrogen Peroxide with CYP152A1 in the presence (colored) or absence (gray) of n-short-chain fatty acids (A) and n-medium/long chain fatty acids (B). The final concentrations of fatty acid were 20 mM for short-chain fatty acid such as Acetic acid (C<sub>2</sub>), Propionic acid (C<sub>3</sub>), Butanoic acid (C<sub>4</sub>), Pentanoic acid (C<sub>5</sub>), Hexanoic acid (C<sub>6</sub>), and Hexanoic acid(C<sub>7</sub>), 10 mM for medium-chain fatty acid such as Octanoic acid (C<sub>8</sub>) and Nonanoic acid (C<sub>9</sub>), 5 mM for Decanoic acid (C<sub>10</sub>), and 50  $\mu$ M for Myristic acid (C<sub>14</sub>).



Figure S2. Michaelis–Menten kinetics carve of CYP152's catalase reaction rate in the presence (colored) or absence (gray) of n-short-chain fatty acids (A) and n-medium/long chain fatty acids (B).