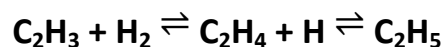


Supplementary Material

Mechanism, Thermochemistry, and Kinetics of the Reversible Reactions:



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E,J-resolved master equation model for a thermally activated reaction

E,J-resolved master equation model for a thermally activated reaction ($C_2H_5 \rightarrow C_2H_4 + H$), which has a well and a product channel, is given by:

$$\begin{aligned} \frac{\partial C_{C_2H_5}(E_i J_i t)}{\partial t} &= \sum_{J_k=0}^{J_{max}} \sum_{E_k=0}^{E_{max}} \omega_{LJ} \cdot P(E_i J_i | E_k J_k) \cdot C_{C_2H_5}(E_k J_k t) \cdot \Delta E - \omega_{LJ} \cdot C_{C_2H_5}(E_i J_i t) - k_{C_2H_5 \rightarrow C_2H_4 + H}(E_i J_i) \cdot C_{C_2H_5}(E_i J_i t) \end{aligned} \quad (S1)$$

All terms have been defined in the main text. Here initial energy distribution of C_2H_5 is assumed to be Boltzmann thermal energy distribution function:

$$F_{C_2H_5}(E_i J_i) = \frac{(2J_i + 1) \cdot \rho_{C_2H_5}(E_i J_i) \cdot \exp\left(\frac{-E_i}{RT}\right)}{\sum_{J_i=0}^{J_{max}} (2J_i + 1) \int_{E_i=0}^{E_{max}} \rho_{C_2H_5}(E_i J_i) \cdot \exp\left(\frac{-E_i}{RT}\right) \cdot dE_i} \quad (S2)$$

Eq. (S1) can be cast in the matrix form:

$$\frac{\partial C}{\partial t} = MC \quad (S3)$$

The reaction rate coefficient, $k_{\text{uni.}}(T,p)$, is equally assigned to the lowest eigenvalue of M .

$$k_{\text{uni.}}(T,p) = |\lambda_1| \quad (\text{S4})$$

Table S1: Calculated thermal rate coefficients (in cm^3/s) for the $\text{C}_2\text{H}_3 + \text{H}_2 \rightarrow \text{C}_2\text{H}_4 + \text{H}$ reaction using the SCTST theory.

T (K)	1000/T (1/K)	$k(T)_1$ (cm^3/s)
300	3.33333333	4.12E-18
325	3.07692308	1.05E-17
350	2.85714286	2.39E-17
375	2.66666667	4.94E-17
400	2.5	9.41E-17
425	2.35294118	1.68E-16
450	2.22222222	2.82E-16
475	2.10526316	4.52E-16
500	2	6.94E-16
525	1.9047619	1.03E-15
550	1.81818182	1.48E-15
575	1.73913043	2.06E-15
600	1.66666667	2.81E-15
625	1.6	3.75E-15
650	1.53846154	4.90E-15
675	1.48148148	6.30E-15
700	1.42857143	7.98E-15
725	1.37931034	9.96E-15
750	1.33333333	1.23E-14
775	1.29032258	1.50E-14
800	1.25	1.81E-14
825	1.21212121	2.16E-14
850	1.17647059	2.56E-14
875	1.14285714	3.01E-14
900	1.11111111	3.52E-14
925	1.08108108	4.08E-14
950	1.05263158	4.70E-14

975	1.02564103	5.39E-14
1000	1	6.14E-14
1025	0.97560976	6.96E-14
1050	0.95238095	7.86E-14
1075	0.93023256	8.83E-14
1100	0.90909091	9.88E-14
1125	0.88888889	1.10E-13
1150	0.86956522	1.22E-13
1175	0.85106383	1.35E-13
1200	0.83333333	1.49E-13
1225	0.81632653	1.64E-13
1250	0.8	1.80E-13
1275	0.78431373	1.97E-13
1300	0.76923077	2.15E-13
1325	0.75471698	2.34E-13
1350	0.74074074	2.54E-13
1375	0.72727273	2.76E-13
1400	0.71428571	2.98E-13
1425	0.70175439	3.21E-13
1450	0.68965517	3.46E-13
1475	0.6779661	3.72E-13
1500	0.66666667	3.99E-13
1525	0.6557377	4.27E-13
1550	0.64516129	4.57E-13
1575	0.63492063	4.88E-13
1600	0.625	5.20E-13
1625	0.61538462	5.53E-13
1650	0.60606061	5.88E-13
1675	0.59701493	6.24E-13
1700	0.58823529	6.61E-13
1725	0.57971014	7.00E-13
1750	0.57142857	7.40E-13
1775	0.56338028	7.81E-13
1800	0.55555556	8.24E-13
1825	0.54794521	8.68E-13
1850	0.54054054	9.13E-13
1875	0.53333333	9.60E-13
1900	0.52631579	1.01E-12
1925	0.51948052	1.06E-12

1950	0.51282051	1.11E-12
1975	0.50632911	1.16E-12
2000	0.5	1.21E-12
2025	0.49382716	1.27E-12
2050	0.48780488	1.33E-12
2075	0.48192771	1.38E-12
2100	0.47619048	1.44E-12
2125	0.47058824	1.50E-12
2150	0.46511628	1.56E-12
2175	0.45977011	1.63E-12
2200	0.45454545	1.69E-12
2225	0.4494382	1.76E-12
2250	0.44444444	1.82E-12
2275	0.43956044	1.89E-12
2300	0.43478261	1.96E-12
2325	0.43010753	2.03E-12
2350	0.42553191	2.10E-12
2375	0.42105263	2.17E-12
2400	0.41666667	2.25E-12
2425	0.41237113	2.32E-12
2450	0.40816327	2.40E-12
2475	0.4040404	2.48E-12
2500	0.4	2.55E-12
2525	0.3960396	2.63E-12
2550	0.39215686	2.71E-12
2575	0.38834951	2.79E-12
2600	0.38461538	2.88E-12
2625	0.38095238	2.96E-12
2650	0.37735849	3.04E-12
2675	0.37383178	3.13E-12
2700	0.37037037	3.21E-12
2725	0.36697248	3.30E-12
2750	0.36363636	3.39E-12
2775	0.36036036	3.47E-12
2800	0.35714286	3.56E-12
2825	0.3539823	3.65E-12
2850	0.35087719	3.74E-12
2875	0.34782609	3.83E-12
2900	0.34482759	3.92E-12

2925	0.34188034	4.01E-12
2950	0.33898305	4.10E-12
2975	0.33613445	4.19E-12
3000	0.33333333	4.28E-12
3025	0.33057851	4.37E-12
3050	0.32786885	4.46E-12
3075	0.32520325	4.56E-12
3100	0.32258065	4.65E-12
3125	0.32	4.74E-12
3150	0.31746032	4.83E-12
3175	0.31496063	4.92E-12
3200	0.3125	5.01E-12
3225	0.31007752	5.10E-12
3250	0.30769231	5.19E-12
3275	0.30534351	5.28E-12
3300	0.3030303	5.37E-12
3325	0.30075188	5.46E-12
3350	0.29850746	5.55E-12
3375	0.2962963	5.64E-12
3400	0.29411765	5.72E-12
3425	0.2919708	5.81E-12
3450	0.28985507	5.89E-12
3475	0.28776978	5.98E-12
3500	0.28571429	6.06E-12

Table S2: Calculated thermal rate coefficients (in cm³/s) for the C₂H₄ + H → C₂H₃ + H₂ reaction using the SCTST theory and the equilibrium constant (K_{eq}).

T (K)	1000/T (1/K)	k(T) ₋₁
1000	1	1.30E-13
1025	0.975609756	1.60E-13
1050	0.952380952	1.96E-13
1075	0.930232558	2.36E-13
1100	0.909090909	2.84E-13
1125	0.888888889	3.38E-13
1150	0.869565217	4.00E-13
1175	0.85106383	4.70E-13
1200	0.833333333	5.50E-13
1225	0.816326531	6.39E-13
1250	0.8	7.39E-13
1275	0.784313725	8.49E-13
1300	0.769230769	9.72E-13
1325	0.754716981	1.11E-12
1350	0.740740741	1.26E-12
1375	0.727272727	1.42E-12
1400	0.714285714	1.60E-12
1425	0.701754386	1.79E-12
1450	0.689655172	2.00E-12
1475	0.677966102	2.23E-12
1500	0.666666667	2.48E-12
1525	0.655737705	2.74E-12
1550	0.64516129	3.02E-12
1575	0.634920635	3.33E-12
1600	0.625	3.65E-12
1625	0.615384615	4.00E-12
1650	0.606060606	4.37E-12
1675	0.597014925	4.76E-12
1700	0.588235294	5.17E-12
1725	0.579710145	5.61E-12
1750	0.571428571	6.07E-12
1775	0.563380282	6.56E-12
1800	0.555555556	7.07E-12

1825	0.547945205	7.61E-12
1850	0.540540541	8.17E-12
1875	0.533333333	8.76E-12
1900	0.526315789	9.38E-12
1925	0.519480519	1.00E-11
1950	0.512820513	1.07E-11
1975	0.506329114	1.14E-11
2000	0.5	1.21E-11
2025	0.49382716	1.29E-11
2050	0.487804878	1.37E-11
2075	0.481927711	1.45E-11
2100	0.476190476	1.53E-11
2125	0.470588235	1.62E-11
2150	0.465116279	1.71E-11
2175	0.459770115	1.80E-11
2200	0.454545455	1.90E-11
2225	0.449438202	2.00E-11
2250	0.444444444	2.10E-11
2275	0.43956044	2.20E-11
2300	0.434782609	2.31E-11
2325	0.430107527	2.42E-11
2350	0.425531915	2.53E-11
2375	0.421052632	2.65E-11
2400	0.416666667	2.77E-11
2425	0.412371134	2.89E-11
2450	0.408163265	3.01E-11
2475	0.404040404	3.14E-11
2500	0.4	3.27E-11
2525	0.396039604	3.40E-11
2550	0.392156863	3.54E-11
2575	0.388349515	3.67E-11
2600	0.384615385	3.81E-11
2625	0.380952381	3.95E-11
2650	0.377358491	4.10E-11
2675	0.373831776	4.25E-11
2700	0.37037037	4.39E-11
2725	0.366972477	4.55E-11
2750	0.363636364	4.70E-11
2775	0.36036036	4.85E-11

2800	0.357142857	5.01E-11
2825	0.353982301	5.17E-11
2850	0.350877193	5.33E-11
2875	0.347826087	5.49E-11
2900	0.344827586	5.66E-11
2925	0.341880342	5.82E-11
2950	0.338983051	5.99E-11
2975	0.336134454	6.16E-11
3000	0.333333333	6.33E-11
3025	0.330578512	6.50E-11
3050	0.327868852	6.68E-11
3075	0.325203252	6.85E-11
3100	0.322580645	7.02E-11
3125	0.32	7.20E-11
3150	0.317460317	7.37E-11
3175	0.31496063	7.55E-11
3200	0.3125	7.73E-11
3225	0.310077519	7.91E-11
3250	0.307692308	8.09E-11
3275	0.305343511	8.26E-11
3300	0.303030303	8.44E-11
3325	0.30075188	8.62E-11
3350	0.298507463	8.80E-11
3375	0.296296296	8.98E-11
3400	0.294117647	9.16E-11
3425	0.291970803	9.34E-11
3450	0.289855072	9.51E-11
3475	0.287769784	9.69E-11
3500	0.285714286	9.87E-11

Table S3: Calculated thermal rate coefficients (in cm³/s) at the high-pressure limit (i.e. the capture rate constants) for the C₂H₄ + H → C₂H₅ reaction using the SCTST theory.

T (K)	k(T) ₂ (cm ³ /s)
200	2.08E-13
225	3.30E-13
250	4.96E-13
275	7.10E-13
300	9.76E-13
325	1.29E-12
350	1.67E-12
375	2.10E-12
400	2.58E-12
425	3.12E-12
450	3.71E-12
475	4.36E-12
500	5.05E-12
525	5.80E-12
550	6.59E-12
575	7.43E-12
600	8.32E-12
625	9.25E-12
650	1.02E-11
675	1.12E-11
700	1.23E-11
725	1.34E-11
750	1.45E-11
775	1.57E-11
800	1.69E-11
825	1.81E-11
850	1.94E-11
875	2.07E-11
900	2.20E-11
925	2.34E-11
950	2.48E-11
975	2.62E-11
1000	2.76E-11
1025	2.91E-11

1050	3.06E-11
1075	3.21E-11
1100	3.37E-11
1125	3.52E-11
1150	3.68E-11
1175	3.84E-11
1200	4.00E-11
1225	4.17E-11
1250	4.33E-11
1275	4.50E-11
1300	4.67E-11
1325	4.84E-11
1350	5.02E-11
1375	5.19E-11
1400	5.37E-11
1425	5.55E-11
1450	5.73E-11
1475	5.91E-11
1500	6.09E-11
1525	6.27E-11
1550	6.46E-11
1575	6.64E-11
1600	6.83E-11
1625	7.02E-11
1650	7.21E-11
1675	7.40E-11
1700	7.59E-11
1725	7.79E-11
1750	7.98E-11
1775	8.17E-11
1800	8.37E-11
1825	8.57E-11
1850	8.76E-11
1875	8.96E-11
1900	9.16E-11
1925	9.36E-11
1950	9.56E-11
1975	9.77E-11

2000	9.97E-11
2025	1.02E-10
2050	1.04E-10
2075	1.06E-10
2100	1.08E-10
2125	1.10E-10
2150	1.12E-10
2175	1.14E-10
2200	1.16E-10
2225	1.18E-10
2250	1.20E-10
2275	1.22E-10
2300	1.25E-10
2325	1.27E-10
2350	1.29E-10
2375	1.31E-10
2400	1.33E-10
2425	1.35E-10
2450	1.37E-10
2475	1.39E-10
2500	1.42E-10
2525	1.44E-10
2550	1.46E-10
2575	1.48E-10

Table S4: Calculated reaction rate coefficients (cm³/s) as functions of temperature and pressure for the C₂H₄ + H → C₂H₅ reaction using the ME/SCTST approach.

T (K) / P(atm)	0.5	1	5	10	50	100
300	8.84E-13	9.23E-13	9.63E-13	9.69E-13	9.74E-13	9.75E-13
400	2.01E-12	2.20E-12	2.47E-12	2.52E-12	2.57E-12	2.57E-12
500	3.10E-12	3.62E-12	4.49E-12	4.72E-12	4.97E-12	5.01E-12
600	3.77E-12	4.66E-12	6.55E-12	7.15E-12	7.96E-12	8.12E-12
700	3.89E-12	5.10E-12	8.13E-12	9.29E-12	1.12E-11	1.16E-11
800	3.66E-12	5.04E-12	9.05E-12	1.09E-11	1.43E-11	1.53E-11
900	3.14E-12	4.52E-12	9.08E-12	1.14E-11	1.67E-11	1.85E-11
1000	2.54E-12	3.80E-12	8.47E-12	1.12E-11	1.81E-11	2.09E-11
1100	1.96E-12	3.04E-12	7.44E-12	1.03E-11	1.85E-11	2.22E-11
1200	1.47E-12	2.36E-12	6.27E-12	9.01E-12	1.79E-11	2.24E-11
1300	1.09E-12	1.80E-12	5.13E-12	7.64E-12	1.66E-11	2.17E-11
1400	8.15E-13	1.37E-12	4.14E-12	6.36E-12	1.50E-11	2.03E-11
1500	6.07E-13	1.04E-12	3.30E-12	5.20E-12	1.32E-11	1.84E-11
1550	5.29E-13	9.10E-13	2.96E-12	4.71E-12	1.23E-11	1.75E-11
1600	4.61E-13	7.98E-13	2.64E-12	4.25E-12	1.14E-11	1.65E-11
1700	3.56E-13	6.23E-13	2.13E-12	3.49E-12	9.87E-12	1.46E-11
1800	2.80E-13	4.94E-13	1.74E-12	2.89E-12	8.51E-12	1.29E-11
1900	2.23E-13	3.98E-13	1.43E-12	2.40E-12	7.34E-12	1.13E-11
2000	1.82E-13	3.27E-13	1.19E-12	2.03E-12	6.38E-12	1.00E-11
2200	1.27E-13	2.30E-13	8.65E-13	1.49E-12	4.91E-12	7.88E-12
2400	9.42E-14	1.71E-13	6.57E-13	1.14E-12	3.89E-12	6.36E-12
2500	8.28E-14	1.52E-13	5.83E-13	1.02E-12	3.51E-12	5.78E-12
2600	7.34E-14	1.35E-13	5.22E-13	9.17E-13	3.19E-12	5.28E-12
2800	5.95E-14	1.10E-13	4.29E-13	7.58E-13	2.68E-12	4.48E-12
3000	4.98E-14	9.17E-14	3.63E-13	6.45E-13	2.31E-12	3.89E-12
3200	4.28E-14	7.91E-14	3.15E-13	5.61E-13	2.03E-12	3.43E-12
3400	3.76E-14	6.94E-14	2.78E-13	4.96E-13	1.81E-12	3.07E-12
3500	3.53E-14	6.52E-14	2.62E-13	4.68E-13	1.71E-12	2.91E-12

Table S4 (Continued): Calculated reaction rate coefficients (cm³/s) as functions of temperature and pressure for the C₂H₄ + H → C₂H₅ reaction using the ME/SCTST approach.

P (Torr) / T(K)	285 K	298 K	400 K	511 K	604 K
0.1	3.62E-14	3.61E-14	3.26E-14	2.46E-14	1.78E-14
0.2	5.43E-14	5.49E-14	5.33E-14	4.21E-14	3.13E-14
0.3	6.83E-14	6.96E-14	7.05E-14	5.74E-14	4.35E-14
0.4	8.02E-14	8.20E-14	8.58E-14	7.14E-14	5.47E-14
0.5	9.05E-14	9.30E-14	9.98E-14	8.44E-14	6.53E-14
0.6	9.98E-14	1.03E-13	1.13E-13	9.67E-14	7.55E-14
0.7	1.08E-13	1.12E-13	1.25E-13	1.08E-13	8.52E-14
0.8	1.16E-13	1.20E-13	1.36E-13	1.20E-13	9.46E-14
0.9	1.23E-13	1.28E-13	1.47E-13	1.30E-13	1.04E-13
1	1.30E-13	1.36E-13	1.57E-13	1.41E-13	1.12E-13
2	1.83E-13	1.94E-13	2.44E-13	2.30E-13	1.91E-13
3	2.21E-13	2.35E-13	3.11E-13	3.05E-13	2.57E-13
4	2.51E-13	2.69E-13	3.68E-13	3.69E-13	3.17E-13
5	2.76E-13	2.97E-13	4.18E-13	4.28E-13	3.71E-13
6	2.97E-13	3.21E-13	4.62E-13	4.81E-13	4.21E-13
7	3.16E-13	3.42E-13	5.02E-13	5.30E-13	4.69E-13
8	3.32E-13	3.61E-13	5.39E-13	5.76E-13	5.13E-13
9	3.47E-13	3.78E-13	5.73E-13	6.19E-13	5.55E-13
10	3.61E-13	3.94E-13	6.04E-13	6.59E-13	5.95E-13
20	4.53E-13	5.02E-13	8.39E-13	9.80E-13	9.24E-13
30	5.07E-13	5.66E-13	9.96E-13	1.21E-12	1.17E-12
40	5.44E-13	6.11E-13	1.11E-12	1.40E-12	1.38E-12
50	5.72E-13	6.44E-13	1.21E-12	1.55E-12	1.56E-12
60	5.93E-13	6.71E-13	1.28E-12	1.68E-12	1.71E-12
70	6.11E-13	6.92E-13	1.35E-12	1.80E-12	1.85E-12
80	6.25E-13	7.10E-13	1.41E-12	1.90E-12	1.98E-12
90	6.37E-13	7.25E-13	1.46E-12	1.99E-12	2.09E-12
100	6.48E-13	7.39E-13	1.50E-12	2.08E-12	2.20E-12
200	7.07E-13	8.14E-13	1.78E-12	2.66E-12	2.98E-12
300	7.33E-13	8.48E-13	1.93E-12	3.00E-12	3.48E-12
400	7.49E-13	8.68E-13	2.02E-12	3.25E-12	3.85E-12
500	7.59E-13	8.81E-13	2.09E-12	3.43E-12	4.14E-12
600	7.66E-13	8.91E-13	2.14E-12	3.58E-12	4.38E-12
700	7.71E-13	8.98E-13	2.18E-12	3.70E-12	4.59E-12
800	7.75E-13	9.04E-13	2.22E-12	3.80E-12	4.76E-12

900	7.79E-13	9.08E-13	2.24E-12	3.89E-12	4.92E-12
1000	7.81E-13	9.12E-13	2.27E-12	3.96E-12	5.06E-12

Table S5: A comparison of the calculated rate coefficients (cm^3/s) at 800 K as a function of pressure for the $\text{C}_2\text{H}_4 + \text{H} \rightarrow \text{C}_2\text{H}_5$ reaction using four different ME approaches

T = 800 K

P (Torr)	TAR ^{a)}	CAR-1 ^{b)}	CAR-2 ^{c)}	SSA ^{d)}
0.1	8.022E-15	8.022E-15	8.020E-15	7.319E-15
0.3	2.091E-14	2.090E-14	2.090E-14	1.918E-14
0.5	3.244E-14	3.241E-14	3.243E-14	2.985E-14
0.7	4.320E-14	4.321E-14	4.320E-14	3.984E-14
1	5.839E-14	5.838E-14	5.839E-14	5.395E-14
3	1.444E-13	1.444E-13	1.444E-13	1.344E-13
5	2.168E-13	2.167E-13	2.168E-13	2.025E-13
7	2.815E-13	2.814E-13	2.815E-13	2.636E-13
10	3.690E-13	3.690E-13	3.691E-13	3.465E-13
30	8.110E-13	8.108E-13	8.111E-13	7.683E-13
50	1.139E-12	1.138E-12	1.139E-12	1.083E-12
70	1.410E-12	1.409E-12	1.410E-12	1.345E-12
100	1.752E-12	1.751E-12	1.752E-12	1.677E-12
300	3.220E-12	3.216E-12	3.221E-12	3.108E-12
500	4.135E-12	4.128E-12	4.135E-12	4.007E-12
700	4.818E-12	4.808E-12	4.819E-12	4.680E-12
1000	5.606E-12	5.605E-12	5.608E-12	5.459E-12

- Thermally activated reaction model ($\text{C}_2\text{H}_5 \rightarrow \text{C}_2\text{H}_4 + \text{H}$).
- Chemically activated reaction model ($\text{C}_2\text{H}_4 + \text{H} \rightleftharpoons \text{C}_2\text{H}_5$), and $k(T,p) = k_\infty \times (1 - \gamma_H)$ as recommended in the Multiwell software package by John R. Barker.
- Chemically activated reaction model ($\text{C}_2\text{H}_4 + \text{H} \rightleftharpoons \text{C}_2\text{H}_5$), and Eq. 20 (see the main text).
- Steady state approach for a chemically activated reaction, and Eq. 15 (see the main text). This SSA model starts to break down at 800 K, but it remains valid when $T < 800$ K.

Table S5 (Continued): A comparison of the calculated rate coefficients (cm³/s) at 1000 K as a function of pressure for the C₂H₄ + H → C₂H₅ reaction using four different ME approaches

T = 1000 K

P (Torr)	TAR ^{a)}	CAR-1 ^{b)}	CAR-2 ^{c)}	SSA ^{d)}
1	2.585E-14	2.559E-14	2.586E-14	1.785E-14
3	6.740E-14		6.746E-14	4.713E-14
5	1.040E-13		1.041E-13	7.324E-14
7	1.377E-13		1.378E-13	9.748E-14
10	1.846E-13	1.834E-13	1.848E-13	1.314E-13
30	4.388E-13		4.393E-13	3.188E-13
50	6.430E-13		6.435E-13	4.723E-13
70	8.206E-13		8.212E-13	6.075E-13
100	1.055E-12	1.024E-12	1.056E-12	7.879E-13
300	2.180E-12		2.182E-12	1.678E-12
500	2.975E-12		2.977E-12	2.325E-12
700	3.614E-12		3.617E-12	2.855E-12
1000	4.401E-12	4.367E-12	4.404E-12	3.520E-12
3000	7.590E-12		7.597E-12	6.319E-12
5000	9.454E-12		9.461E-12	8.024E-12
7000	1.079E-11		1.080E-11	9.277E-12
10000	1.228E-11	1.228E-11	1.229E-11	1.070E-11

d). At 1000 K, the SSA model completely breaks down.

T = 2000 K

P (Torr)	TAR ^{a)}	CAR-1 ^{b)}	CAR-2 ^{c)}
1	8.104E-16	8.107E-16	8.111E-16
3	2.304E-15	2.306E-15	2.306E-15
5	3.726E-15	3.729E-15	3.730E-15
7	5.106E-15	5.108E-15	5.109E-15
10	7.108E-15	7.116E-15	7.116E-15
30	1.946E-14	1.948E-14	1.948E-14
50	3.085E-14	3.088E-14	3.087E-14
70	4.165E-14	4.170E-14	4.169E-14
100	5.711E-14	5.719E-14	5.718E-14
300	1.484E-13	1.486E-13	1.486E-13

500	2.291E-13	2.294E-13	2.293E-13
700	3.038E-13	3.042E-13	3.040E-13
1000	4.083E-13	4.087E-13	4.083E-13
3000	9.884E-13	9.900E-13	9.891E-13
5000	1.470E-12	1.472E-12	1.471E-12
7000	1.899E-12	1.902E-12	1.900E-12
10000	2.477E-12	2.481E-12	2.479E-12
30000	5.414E-12	5.424E-12	5.418E-12
50000	7.620E-12	7.635E-12	7.627E-12
70000	9.466E-12	9.481E-12	9.473E-12
100000	1.182E-11	1.183E-11	1.183E-11

T = 3000 K

P (Torr)	TAR ^{a)}	CAR-2 ^{c)}
1	1.984E-16	1.983E-16
3	5.712E-16	5.707E-16
5	9.299E-16	9.292E-16
7	1.280E-15	1.279E-15
10	1.793E-15	1.791E-15
30	5.006E-15	5.001E-15
50	8.024E-15	8.014E-15
70	1.092E-14	1.091E-14
100	1.512E-14	1.510E-14
300	4.065E-14	4.058E-14
500	6.391E-14	6.380E-14
700	8.587E-14	8.572E-14
1000	1.172E-13	1.169E-13
3000	2.998E-13	2.991E-13
5000	4.595E-13	4.586E-13
7000	6.065E-13	6.056E-13
10000	8.113E-13	8.102E-13
30000	1.939E-12	1.940E-12
50000	2.865E-12	2.872E-12
70000	3.683E-12	3.700E-12
100000	4.778E-12	4.816E-12

Table S6: Calculated tunneling corrections as a function of temperature for the $C_2H_3 + H_2 \rightarrow C_2H_4 + H$ reaction using the SCTST theory.

T(K)	k(T) without tun.	k(T) with tun.	Tun. Correction ($\kappa(T)$)
300	1.07E-18	4.12E-18	3.84E+00
325	3.36E-18	1.05E-17	3.13E+00
350	8.94E-18	2.39E-17	2.67E+00
375	2.09E-17	4.94E-17	2.36E+00
400	4.42E-17	9.41E-17	2.13E+00
425	8.56E-17	1.68E-16	1.96E+00
450	1.54E-16	2.82E-16	1.83E+00
475	2.62E-16	4.52E-16	1.72E+00
500	4.24E-16	6.94E-16	1.64E+00
525	6.56E-16	1.03E-15	1.57E+00
550	9.77E-16	1.48E-15	1.51E+00
575	1.41E-15	2.06E-15	1.46E+00
600	1.98E-15	2.81E-15	1.42E+00
625	2.70E-15	3.75E-15	1.38E+00
650	3.62E-15	4.90E-15	1.35E+00
675	4.75E-15	6.30E-15	1.33E+00
700	6.13E-15	7.98E-15	1.30E+00
725	7.78E-15	9.96E-15	1.28E+00
750	9.74E-15	1.23E-14	1.26E+00
775	1.20E-14	1.50E-14	1.24E+00
800	1.47E-14	1.81E-14	1.23E+00
825	1.78E-14	2.16E-14	1.22E+00
850	2.13E-14	2.56E-14	1.20E+00
875	2.53E-14	3.01E-14	1.19E+00
900	2.98E-14	3.52E-14	1.18E+00
925	3.49E-14	4.08E-14	1.17E+00
950	4.05E-14	4.70E-14	1.16E+00
975	4.67E-14	5.39E-14	1.15E+00
1000	5.36E-14	6.14E-14	1.15E+00
1025	6.12E-14	6.96E-14	1.14E+00
1050	6.95E-14	7.86E-14	1.13E+00
1075	7.85E-14	8.83E-14	1.13E+00
1100	8.83E-14	9.88E-14	1.12E+00
1125	9.89E-14	1.10E-13	1.11E+00

1150	1.10E-13	1.22E-13	1.11E+00
1175	1.23E-13	1.35E-13	1.10E+00
1200	1.36E-13	1.49E-13	1.10E+00
1225	1.50E-13	1.64E-13	1.10E+00
1250	1.65E-13	1.80E-13	1.09E+00
1275	1.81E-13	1.97E-13	1.09E+00
1300	1.98E-13	2.15E-13	1.08E+00
1325	2.17E-13	2.34E-13	1.08E+00
1350	2.36E-13	2.54E-13	1.08E+00
1375	2.56E-13	2.76E-13	1.08E+00
1400	2.78E-13	2.98E-13	1.07E+00
1425	3.00E-13	3.21E-13	1.07E+00
1450	3.24E-13	3.46E-13	1.07E+00
1475	3.49E-13	3.72E-13	1.06E+00
1500	3.75E-13	3.99E-13	1.06E+00
1525	4.03E-13	4.27E-13	1.06E+00
1550	4.32E-13	4.57E-13	1.06E+00
1575	4.62E-13	4.88E-13	1.06E+00
1600	4.93E-13	5.20E-13	1.05E+00
1625	5.25E-13	5.53E-13	1.05E+00
1650	5.59E-13	5.88E-13	1.05E+00
1675	5.94E-13	6.24E-13	1.05E+00
1700	6.31E-13	6.61E-13	1.05E+00
1725	6.69E-13	7.00E-13	1.05E+00
1750	7.08E-13	7.40E-13	1.04E+00
1775	7.48E-13	7.81E-13	1.04E+00
1800	7.90E-13	8.24E-13	1.04E+00
1825	8.34E-13	8.68E-13	1.04E+00
1850	8.78E-13	9.13E-13	1.04E+00
1875	9.24E-13	9.60E-13	1.04E+00
1900	9.72E-13	1.01E-12	1.04E+00
1925	1.02E-12	1.06E-12	1.04E+00
1950	1.07E-12	1.11E-12	1.03E+00
1975	1.12E-12	1.16E-12	1.03E+00
2000	1.18E-12	1.21E-12	1.03E+00

Optimized geometries of various species

C₂H₃ (C_s, ²A'): ae-CCSD(T)/cc-pVQZ // UHF

HEAT-345Q

C

C 1 R1*

H 1 R2* 2 A1*

H 1 R3* 2 A2* 3 T180

H 2 R4* 1 A3* 3 T180

R1 = 1.307413290373744

R2 = 1.081723724184898

A1 = 122.057880654972905

R3 = 1.087181560366102

A2 = 121.279851219171391

T180 = 180.000000000000000

R4 = 1.075581119201201

A3 = 137.363322899459064

C₂H₄ (D_{2h}, ¹A₁): ae-CCSD(T)/cc-pVQZ // RHF

HEAT-345Q

C

C 1 R1*

H 1 R2* 2 A1*

H 1 R2* 2 A1* 3 D180

H 2 R2* 1 A1* 3 D180

H 2 R2* 1 A1* 5 D180

R1 = 1.330902917626653

R2 = 1.079752371421382

A1 = 121.448016067414059

D180 = 180.000000000000000

C₂H₅ (C_s, ²A'): ae-CCSD(T)/cc-pVQZ // UHF

HEAT-345Q

C

C 1 R1

H 1 R2 2 A1

H 1 R3 2 A2 3 T1

H 1 R3 2 A2 3 T2
H 2 R4 1 A3 3 T3
H 2 R4 1 A3 3 T4

R1 = 1.484754348094979
R2 = 1.095430448603470
A1 = 111.646939695568605
R3 = 1.088412112085974
A2 = 111.580217293014798
T1 = 119.360318042895344
T2 = -119.360318042895344
R4 = 1.077652719659850
A3 = 120.813623360314452
T3 = 85.455662573944480
T4 = -85.455662573944480

TS1 (C_s, ²A'): ae-CCSD(T)/cc-pVQZ // UHF

HEAT-345Q

H

C 1 R1*

C 2 R2* 1 A1*

H 2 R3* 3 A2* 1 T1*

H 2 R3* 3 A2* 1 T2*

H 3 R4* 2 A3* 1 T3*

H 3 R4* 2 A3* 1 T4*

R1 = 1.981950449111066
R2 = 1.341149588425816
A1 = 106.655914053263487
R3 = 1.079445246175276
A2 = 121.248975399909156
T1 = 94.655688607126208
T2 = -94.655688607126208
R4 = 1.079247392069564
A3 = 121.328812222024141
T3 = 88.799300980087395
T4 = -88.799300980087395

TS1 (C_s, ²A'): ae-CCSD(T)/cc-pVQZ // ROHF

HEAT-345Q

H

C 1 R1*
C 2 R2* 1 A1*
H 2 R3* 3 A2* 1 T1*
H 2 R3* 3 A2* 1 T2*
H 3 R4* 2 A3* 1 T3*
H 3 R4* 2 A3* 1 T4*

R1 = 1.982646393556012
R2 = 1.343265335057637
A1 = 106.560181074648682
R3 = 1.079480329790683
A2 = 121.229700627428400
T1 = 94.718156988897547
T2 = -94.718156988897547
R4 = 1.079315572788190
A3 = 121.315855945123403
T3 = 88.775569747622228
T4 = -88.775569747622228

TS2 (C_s, ²A'): ae-CCSD(T)/cc-pVQZ // UHF

HEAT-345Q

C
C 1 R1*
H 1 R2* 2 A1*
H 2 R3* 1 A2* 3 T180
H 2 R4* 1 A3* 4 T180
H 1 R5* 2 A4* 3 T180
X 6 R0 1 A5* 3 T180
H 6 R6* 7 A5* 1 T180

R1 = 1.314696068503856
R2 = 1.078080276583277
A1 = 131.559284241466571
R3 = 1.081392379108812
A2 = 121.406021847894522
T180 = 180.000000000000000
R4 = 1.083852697068704
A3 = 121.430010447996494
R5 = 1.446328141095375
A4 = 114.483549666949699
R0 = 1.000000000000000
A5 = 87.390531570229783
R6 = 0.862813820478743

TS3 (C_{2v}, ²B₁): ae-CCSD(T)/cc-pVQZ // UHF

HEAT-345Q

X

H 1 R0

C 2 R1* 1 A1*

C 2 R1* 1 A1* 3 T180

H 3 R2* 2 A2* 1 T90*

H 3 R2* 2 A2* 1 Tm90*

H 4 R2* 2 A2* 1 T90*

H 4 R2* 2 A2* 1 Tm90*

R0 = 1.0000000000000000

R1 = 1.290216448720350

A1 = 144.866813199303550

T180 = 180.0000000000000000

R2 = 1.076159762632992

A2 = 112.708317862970560

T90 = 69.290755272482755

Tm90 = -69.290755272482755

Ro-vibrational parameters and anharmonic constants (all in cm^{-1}) for various species

C_2H_3

	ANO2 a)		ANO1 b)									
ω_1	723.7815	x11	-3.1037									
ω_2	825.7328	x2i	2.5757	1.0783								
ω_3	925.1483	x3i	3.8283	-0.0940	-0.2882							
ω_4	1070.7801	x4i	-34.9720	0.6219	1.7690	-7.3640						
ω_5	1396.2211	x5i	-0.4151	-1.1211	-5.6418	-4.1735	-3.3217					
ω_6	1642.3325	x6i	-2.0176	-7.9209	-6.9465	-11.2744	-6.4847	-7.1344				
ω_7	3076.9052	x7i	-5.4179	-5.8609	-8.2251	-8.7029	-16.8495	-7.8422	-47.7603			
ω_8	3182.3996	x8i	-7.0895	-5.8476	-15.5273	-11.2059	-18.7664	1.9991	-62.4763	-48.9464		
ω_9	3252.9634	x9i	0.1749	-10.1917	-3.5752	1.5972	-1.7544	-0.9093	0.6039	2.0052	-61.8138	

Three rotational constants (cm^{-1})

A	7.8037
B	1.0880
C	0.9549

- a) Obtained at fc-CCSD(T)/ANO2 level of theory
 b) Calculated at fc-CCSD(T)/ANO1 level of theory

C_2H_4

Harmonic vib. frequencies (cm^{-1})

	ANO2 (in cm^{-1})
ω_1	823.9830
ω_2	951.8802
ω_3	965.8709
ω_4	1051.7872
ω_5	1246.7681
ω_6	1367.8359
ω_7	1476.5674
ω_8	1670.8931
ω_9	3140.1665
ω_{10}	3158.7995

ω_{11}	3224.1575
ω_{12}	3250.3880

Three rotational constants (cm^{-1})

A (in cm^{-1})	4.9039
B	1.0038
C	0.8333

Anharmonic constants (cm^{-1})

ANO1	1	2	3	4
x11	4.3442			
x2i	0.7569	2.3874		
x3i	6.1162	0.6447	0.9274	
x4i	1.2604	-5.8769	-8.3740	-1.8225
x5i	-6.1974	5.0296	1.6302	0.1798
x6i	1.8536	-3.0486	-2.3630	-3.8531
x7i	-3.6984	-3.2823	-4.0933	1.7472
x8i	-7.8406	-7.2307	-7.2805	-3.9021
x9i	-3.6327	-5.1113	-4.4890	-6.7800
x10i	-3.2009	-6.4196	-4.7058	-6.6592
x11i	-4.9923	-9.5793	-7.8258	-5.4790
x12i	-2.8880	-8.8687	-8.2962	-5.9247
	5	6	7	8
x55	-0.2895			
x6i	-2.2715	-0.9409		
x7i	-5.6368	-5.0667	-3.0889	
x8i	-15.1488	-9.2395	-1.9102	-0.8700
x9i	-5.4373	-5.2213	-13.0393	-8.1338
x10i	-6.6838	-5.5998	-3.2687	-22.2981
x11i	-1.2917	-7.2734	-11.5576	3.0821
x12i	-6.2928	-7.1946	-10.2969	-1.9772
	9	10	11	12
x99	-13.7476			
x10i	-55.3970	-13.7848		
x11i	-56.5743	-58.6835	-16.1203	
x12i	-57.8048	-56.4728	-64.9344	-16.0017

C₂H₅

Harmonic vib. frequencies (cm⁻¹)

	ANO2 (cm ⁻¹)
ω1 (1DHR)	120.9308
ω2	467.7380
ω3	809.3232
ω4	987.4477
ω5	1071.3188
ω6	1201.1202
ω7	1404.5662
ω8	1477.2985
ω9	1492.0082
ω10	1492.0789
ω11	2986.2513
ω12	3068.4187
ω13	3113.6714
ω14	3159.3045
ω15	3265.1512
Rotational constants	
A (in cm ⁻¹)	3.4689
B	0.7591
C	0.7039

1DHR potential

$$V_{1DHR} = \frac{V_o}{2}(1 - \cos 3\theta), \text{ with } V_o = 30 \text{ cm}^{-1} \text{ and } B_{hr} = 16 \text{ cm}^{-1}$$

Anharmonic constants (cm⁻¹)

	ANO1			
x11	-66.5374			
x2i	-28.3026	16.1158		
x3i	14.3320	15.7641	3.7860	
x4i	-3.1186	1.5941	-0.8608	-1.5916

x5i	-1.2856	-3.7080	-1.6363	-4.2507
x6i	-21.0127	-1.6752	-4.2197	-7.6575
x7i	2.6467	-0.6979	-0.5183	-5.5803
x8i	2.4729	5.2013	-3.4305	-2.6700
x9i	-1.7249	2.5167	-6.7878	-4.2974
x10i	-2.0554	0.2218	-3.2053	-6.1197
x11i	27.0236	8.6438	-0.6061	-10.4427
x12i	-11.0857	0.8991	-2.3042	-0.8768
x13i	-12.8979	-0.5310	-2.3181	-4.1164
x14i	6.3665	-12.4531	-4.2939	-1.5725
x15i	8.0489	-18.9154	-4.9163	0.0350
x55	-5.5251			
x6i	-4.9089	-1.6753		
x7i	-4.8848	-2.6993	-7.9437	
x8i	-2.5616	-5.4478	-2.8990	-4.2150
x9i	-3.2374	-7.3201	-3.6965	-16.3829
x10i	-1.3344	-5.1115	-4.9181	-2.7673
x11i	2.4840	1.4227	2.0284	-5.8662
x12i	0.9516	-5.1087	-2.8412	-1.6679
x13i	0.8205	-3.8459	-6.4572	-8.3597
x14i	-1.1499	-4.0872	-1.0211	2.1353
x15i	-0.0395	-3.6299	-1.2601	-9.7636
x99	-2.1390			
x10i	-2.1252	-1.2562		
x11i	-2.9565	-26.1300	-52.2687	
x12i	-9.0668	-15.9513	-41.9263	-24.5143
x13i	-11.9221	-9.7745	-8.5142	-111.7388
x14i	2.5334	-0.5240	0.0784	0.2501
x15i	-7.6108	-0.5076	0.0427	-0.1926
x1313		-32.9545		
x14i		-0.3972	-28.3601	
x15i		-0.9953	-115.8986	-33.0685

TS1 (H-addition): $C_2H_4 + H = C_2H_5$ Harmonic vib. frequencies (cm^{-1})

	ANO2		ANO1//ROHF		ANO1//UHF
ω_F	767.4551i	xFF	-140.3099	xFF	-135.5767
ω_2	370.1676	x2F	96.0464i	x2F	92.2658i
ω_3	401.3949	x3F	88.5148i	x3F	84.9941i
ω_4	821.8382	x4F	-1.5688i	x4F	-1.4944i
ω_5	931.5782	x5F	-8.3165i	x5F	-9.5865i
ω_6	989.8053	x6F	15.9583i	x6F	12.5083i
ω_7	1044.9995	x7F	-1.5978i	x7F	-3.3770i
ω_8	1245.9548	x8F	-0.1202i	x8F	-0.2021i
ω_9	1344.0524	x9F	-12.6583i	x9F	-15.4547i
ω_{10}	1473.9225	x10F	-0.6742i	x10F	-0.5546i
ω_{11}	1630.0208	x11F	-16.9948i	x11F	-22.4470i
ω_{12}	3145.5794	x12F	2.0521i	x12F	2.4312i
ω_{13}	3160.1214	x13F	0.3138i	x13F	0.3076i
ω_{14}	3230.7760	x14F	2.4960i	x14F	2.8817i
ω_{15}	3256.2492	x15F	2.2070i	x15F	2.5923i
A	2.5384				
B	0.7655				
C	0.7577				

Anharmonic constants (cm^{-1}) based on ROHF reference

	ANO1			
x2i	16.0345			
x3i	25.7955	10.6079		
x4i	-0.0539	-1.3565	4.0408	
x5i	-4.4511	-5.4092	2.2118	3.0111
x6i	9.6620	9.2680	2.8359	0.7933
x7i	-1.5530	4.4124	1.1411	-5.4453
x8i	0.7593	-0.9412	-5.9654	2.5701
x9i	-5.7409	-4.6132	1.8176	-2.6713
x10i	-0.2677	-1.5669	-3.6771	-3.8530
x11i	-7.1966	-5.2222	-6.6580	-7.5557
x12i	0.1074	-0.3111	-3.6370	-4.9339

x13i	-0.5003	-0.9020	-3.2029	-5.9527
x14i	0.3101	-0.1679	-4.9788	-8.4569
x15i	0.3100	-0.3189	-3.1020	-9.4817
x66	1.4572			
x7i	-7.6258	-1.9319		
x8i	1.4040	-0.1624	-0.4064	
x9i	-3.7231	-4.7518	-2.2220	-1.3253
x10i	-2.3949	0.0813	-5.6425	-3.9907
x11i	-7.1637	-2.8043	-13.8367	-7.2183
x12i	-4.9088	-6.3084	-5.4191	-4.5649
x13i	-4.9268	-6.1624	-6.5206	-4.8564
x14i	-8.6007	-5.2777	-2.4295	-5.6315
x15i	-7.3367	-5.5725	-6.2087	-5.4657
x1010	-3.0954			
x11i	-1.5867	-2.8165		
x12i	-14.5355	-10.4016	-13.8484	
x13i	-3.4441	-9.6859	-55.4792	-13.9026
x14i	-11.5307	0.0240	-57.0367	-58.4871
x15i	-10.3450	-4.0409	-57.4978	-56.9514
X1414			-16.3626	
x15i			-64.3319	-16.2069

Anharmonic constants (cm^{-1}) based on UHF reference

	ANO1			
x2i	15.7947			
x3i	25.3090	10.4867		
x4i	-0.0179	-1.3292	4.0906	
x5i	-4.9511	-5.8622	2.2737	2.8688
x6i	8.9323	8.5877	2.8418	0.9463
x7i	-2.1268	3.7839	1.2487	-4.8542
x8i	0.7381	-0.9331	-5.9797	2.6278
x9i	-5.9965	-4.7933	1.9586	-2.3536
x10i	-0.2259	-1.5438	-3.6674	-3.5257
x11i	-8.1947	-6.1348	-6.7831	-6.7705
x12i	0.1965	-0.2079	-3.6163	-4.7627

x13i	-0.4402	-0.8487	-3.1841	-5.9200
x14i	0.4204	-0.0581	-4.9643	-8.3049
x15i	0.4257	-0.2062	-3.0509	-9.4727
x66	1.2173			
x7i	-7.3940	-1.5914		
x8i	1.4484	0.0098	-0.3843	
x9i	-3.6355	-4.1085	-2.1284	-1.0242
x10i	-2.2048	0.1716	-5.6281	-4.1168
x11i	-7.0087	-2.5882	-13.9444	-6.6733
x12i	-4.9094	-6.2810	-5.3969	-4.6749
x13i	-4.7666	-6.0465	-6.5106	-5.0027
x14i	-8.5971	-5.2354	-2.2469	-5.9405
x15i	-7.1889	-5.5096	-6.1819	-5.7616
x1010	-3.0876			
x11i	-1.5486	-2.7527		
x12i	-14.2314	-9.8997	-13.8472	
x13i	-3.4076	-9.0524	-55.2007	-13.9043
x14i	-11.5034	0.6058	-57.0672	-58.1850
x15i	-10.3234	-3.5844	-57.1863	-56.9863
X1414			-16.3540	
x15i			-64.0215	-16.1983

TS2 (H-abstraction): $C_2H_3 + H_2 = H + C_2H_4$

Harmonic vib. frequencies (cm^{-1})

	ANO2		ANO1
ω_F	1324.6845i	xFF	-188.4614
ω_2	270.2214	x2F	42.8014i
ω_3	369.9700	x3F	47.6905i
ω_4	820.5762	x4F	34.1140i
ω_5	907.2815	x5F	22.2190i
ω_6	945.1807	x6F	97.1985i
ω_7	961.1345	x7F	55.9111i

ω_8	1090.1186	x8F	126.0746i
ω_9	1150.9694	x9F	55.0554i
ω_{10}	1402.9928	x10F	0.6030i
ω_{11}	1634.4538	x11F	-18.7725i
ω_{12}	2020.2071	x12F	-719.5663i
ω_{13}	3113.3153	x13F	8.5032i
ω_{14}	3201.9921	x14F	4.9874i
ω_{15}	3222.0012	x15F	-6.6256i
Rot. Constants (cm^{-1})			
A	2.3160		
B	0.8262		
C	0.6090		

Anharmonic constants (cm^{-1})

	ANO1			
x2i	5.4362			
x3i	8.5864	-0.1263		
x4i	3.9678	6.0822	1.5207	
x5i	2.2262	2.9678	1.2832	0.0693
x6i	4.9963	4.7434	6.2374	1.2240
x7i	3.2705	3.2502	3.9399	2.0405
x8i	10.0911	3.3053	19.5157	5.6041
x9i	8.7286	7.3529	-1.0648	2.6045
x10i	-0.3183	-0.5538	-0.5606	-2.1553
x11i	-2.3541	-2.1965	-8.0171	-7.7348
x12i	-27.0073	-33.3051	-33.1005	-23.7729
x13i	0.9181	-1.0165	-3.4023	-6.8178
x14i	0.7903	-0.1331	-4.0496	-6.6477
x15i	-1.4582	-3.7108	-6.0940	-9.1372
x66	21.4011			
x7i	38.7716	4.3712		
x8i	18.8874	11.7577	86.0359	
x9i	5.3233	3.8397	68.3730	1.9112
x10i	-1.5365	-3.3421	-0.1255	-1.3339
x11i	-15.2214	-11.3350	-15.2662	-10.4470
x12i	-142.4893	-76.1627	-477.6116	-117.7658
x13i	-1.6311	-4.1144	-2.1672	-2.7199

x14i	-4.1193	-9.7395	-1.9704	-3.7469
x15i	-4.0757	-4.3671	-3.9257	-8.9607
x1010	-2.8839			
x11i	-6.1315	-4.1072		
x12i	1.3269	56.7401	309.5159	
x13i	-15.5842	-16.7884	-10.3559	-32.8144
x14i	-17.7090	0.0980	-4.6527	-92.1822
x15i	-2.2859	-0.0883	8.5886	-11.3268
X1414			-35.1459	
x15i			-17.5354	-53.6398

TS3 (self-isomerization): $\text{CH}_3\text{-CH}_2 = \text{CH}_2\text{-CH}_3$

Rot-vibrational parameters

	ANO2		ANO1
ω_F	1939.2060i	xFF	-77.6585
ω_2	562.6250	x2F	11.4998i
ω_3	730.5467	x3F	-22.0354i
ω_4	733.6306	x4F	-15.2601i
ω_5	781.1570	x5F	-0.7756i
ω_6	1133.0625	x6F	6.7879i
ω_7	1196.1857	x7F	-1.3776i
ω_8	1278.2507	x8F	16.1550i
ω_9	1421.9647	x9F	-0.8977i
ω_{10}	1449.9166	x10F	-4.6339i
ω_{11}	2231.5941	x11F	-43.9037i
ω_{12}	3163.9074	x12F	4.4052i
ω_{13}	3171.3154	x13F	4.5035i
ω_{14}	3275.7843	x14F	6.8416i
ω_{15}	3295.9487	x15F	6.7478i
A	3.5921		
B	0.8016		
C	0.7228		

Anharmonic constants (cm^{-1})

	ANO1			
x2i	10.2984			
x3i	19.2086	4.9853		
x4i	11.5131	11.9534	-10.7225	
x5i	4.1715	-1.7772	5.3367	4.3069
x6i	2.4531	-6.6765	-5.3144	-3.2882
x7i	-3.3910	1.1336	-4.3800	-5.5812
x8i	17.0752	-9.7449	-6.3407	-5.7236
x9i	3.7719	-1.6424	-0.2042	-3.2586
x10i	10.7972	0.1842	-0.6640	-6.4331
x11i	-10.5250	-8.4715	-7.8468	-0.0560
x12i	-9.1498	-3.8933	-0.7521	-3.0234
x13i	-10.8910	-5.8204	0.6180	-2.7102
x14i	-11.1958	-8.4254	-4.0015	-4.1864
x15i	-12.6925	-8.7113	-3.8720	-2.8160
x66	-6.4991			
x7i	-5.9018	-1.1844		
x8i	0.2075	-0.8720	-1.0515	
x9i	-4.3152	-5.0396	-1.4878	-2.8963
x10i	-5.7843	-8.0789	0.4570	-15.4395
x11i	-9.9074	-2.0710	-36.4598	-1.3269
x12i	-0.0064	-4.5916	-3.5463	-3.0247
x13i	0.2410	-5.0511	-3.9016	-5.3718
x14i	1.4979	-3.8456	-4.5693	-11.1819
x15i	1.0874	-5.3435	-5.1944	-10.7363
x1010	-4.4882			
x11i	-2.8893	-29.0936		
x12i	-2.7807	0.4836	-14.5841	
x13i	-1.0670	-0.3919	-58.1918	-14.7212
x14i	-9.0928	0.9906	-58.3092	-59.9562
x15i	-10.2561	-0.0835	-59.1982	-57.9707

X1414			-17.1329	
x15i			-67.6014	-16.8619