Supporting Information

Heterogeneous Ru/TiO₂ for hydroaminomethylation of

olefins: multicomponent synthesis of amines

Jinghua An,^{a, b} Zhuyan Gao,^{b, c} Yehong Wang,^b Zhixin Zhang,^b Jian Zhang,^b and Feng Wang^{b, *}

^a College of Chemistry, Chemical Engineering and Materials Science, Key Laboratory of Molecular and Nano Probes, Ministry of Education, Collaborative Innovation Center of Functionalized Probes for Chemical Imaging in Universities of Shandong, Institute of Molecular and Nano Science, Shandong Normal University, Jinan 250014, P. R. China

^b State Key Laboratory of Catalysis, Dalian National Laboratory for Clean Energy, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023, China

^c University of Chinese Academy of Sciences, Beijing 100049, China

*Correspondence author: Feng Wang. E-mail: wangfeng@dicp.ac.cn

Materials

All chemicals and reagents are of analytical grade and used as purchased without further purification. The detail list of the used chemicals and reagents are shown in Table S1.

Chemicals	Purity	Corporation			
	36.0-38.0% metal	The Non Ferrous Metal			
	basis	Institute of Shenyang, China			
	> 200/	The Non Ferrous Metal			
	- 39%	Institute of Shenyang, China			
DACI	> 00%	The Non Ferrous Metal			
F UCI2	> 9970	Institute of Shenyang, China			
TiO ₂ (P25)	99.8%	Aladdin			
TiO ₂ -A (Anatase)	99.8%	Aladdin			
TiO ₂ -R (Rutile)	99.8%	Aladdin			
ZnO	AR	Aladdin			
WO ₃	99.9%	Aladdin			
MgO	99.9%	Aladdin			
Tetrahydrofuran	AR	Macklin			
Ethyl acetate	۸P	TianDa Chemical Reagent			
		Co., Ltd, China			
Petroleum ether	AR	Tianjin Fuyu Fine			
	7.41.	Chemical Co., Ltd, China			
Aniline	99.5%	Sinopharm Chemical			
Cyclohexylamine	98.5%	Reagent Co., Ltd			
<i>p</i> -Methylaniline	99.0%	ACROS ORGANICS			
p-Chloroaniline	98.0%	Energy chemical			
<i>p</i> -Tertbutylaniline	98.0%				
<i>p</i> -Methoxyaniline	97.0%				
<i>p</i> -Fluoroaniline	99.0%				
o-Toluidine	99.0%				
o-Chloroaniline	CP				
o-Methoxyaniline	98.0%	Aladdin			
2-ethylaniline	99.0%	Alauulii			
2, 6-diethylaniline	99.0%				
<i>m</i> -Toluidine	99.0%				
2, 6-Dimethylaniline	99.0%				
2,6-Dimethoxyaniline	99.0%				

Table S1. The detail list of the used chemicals and reagents.

Furfurylamine	99.0%			
1-Naphthylamine	99.0%			
Benzylamine	99.0%			
n-Hexylamine	99.0%			
N-Methylaniline	98.0%			
N-Ethylaniline	99.0%			
Diphenylamine	99.0%			
Dibenzylamine	98.0%			
Di-n-propylamine	99.0%			
Piperidine	99.0%			
<i>p</i> -Methoxystyrene	95.0%			
<i>p</i> -Fluorostyrene	97.0%			
<i>p</i> -Chlorostyrene	97.0%			
Alpha-methylstyrene	AR			
Cyclopentene	96%	Aladdin		
Indoline	99.0%	Macklin		
Morpholine	98.0%	Macklin		
Styrene	98.0%			
<i>p</i> -Methylstyrene	98.0%	TCI		
1-Penten	98%			
Cyclohexene	99.0%			
<i>p</i> -Methylaniline	99.0%	ACIOS ORGANICS		
DMSO-D ₆	99.8 atom % D,			
CDCL	contains 0.03% (v/v)	Sigma-Aldrich, America		
	TMS			
Ethylene	99 9%	Dalian Guangming Special		
Propylene	00.070	Gas Products Co., Ltd, China		
CO	99.9999%	Dalian Institute of Chemical		
Millipore-purified	18 MO∙cm	Physics, Chinese Academy of		
water		Sciences, China		

General procedure for calculation of the initial rates

Hydroformylation reaction of cyclohexene

Catalyst (Ru/TiO₂, 0.1 g), cyclohexene (0.5 mmol), solvent (THF, 2.0 mL) and a magnetic stir bar were added into the autoclave reactor of 15 mL, respectively. The reactor was purged with CO for three times. After that, CO and H₂ were charged into the reactor to 1.0 MPa. Finally, the reactor was sealed and placed in a preheated mantle at 160 °C for 30 min. After reaction, the autoclave reactor was cooled with ice water. The crude reaction mixture was analyzed by gas chromatography with *n*-dodecane as internal standard to give cyclohexane formaldehyde in 16% yields.

Condensation reaction of cyclohexane formaldehyde and aniline

The reaction occurred spontaneously and rapidly when adding cyclohexane formaldehyde (0.4 mmol) to aniline (0.5 mmol) in THF (2.0 mL) even in the absence of Ru/TiO₂.

Hydrogenation reaction of imine

Catalyst (Ru/TiO₂, 0.1 g), imine (0.4 mmol), solvent (THF, 2.0 mL), and a magnetic stir bar were added into the autoclave reactor of 15 mL, respectively. The reactor was purged with H₂ for three times. After that, H₂ were charged into the reactor to 1.0 MPa. Finally, the reactor was sealed and placed in a preheated mantle at 160 °C for 5 min. After reaction, the autoclave reactor was cooled with ice water. The crude reaction mixture was analyzed by gas chromatography with *n*-dodecane as internal standard to give **1**. The reaction occurred rapidly when exposing imine to H₂ in THF at 160 °C in the presence of Ru/TiO₂.

General procedure for the reusability of Ru/TiO₂

To examine the reusability of the Ru/TiO₂, the Ru/TiO₂ was recovered by centrifugation and washed with ethanol (3×6 mL). After that, the recovered catalyst was dried under 80 °C overnight and reused for the next run.

General procedure for the effect of removal of catalyst in Figure S1(b)

Catalyst (Ru/TiO₂, 0.1 g), aniline (0.4 mmol), cyclohexene (1.0 mmol), solvent (THF, 2.0 mL), and magnetic stir bar were added into the autoclave reactor of 15 mL, respectively. The reactor was purged with CO for three times. After that, CO and H₂ were charged into the reactor to a certain pressure. Finally, the reactor was sealed and placed in a preheated mantle at 160 °C. After reaction for 2 h, the filtrate was obtained after separating catalyst through

centrifugation. And the obtained filtrate was collected and analyzed by GC. After analyzation, the resulting filtrate was placed into the autoclave reactor to continue react for another 1 h under the same conditions with the original reaction. Then, repeating the above steps.



Figure S1. (a) The reused experiment of Ru/TiO₂. Reaction conditions: Ru/TiO₂ (0.1 g), aniline (0.4 mmol), cyclohexene (1.0 mmol), CO (0.5 MPa), H₂ (0.5 MPa), THF (2.0 mL), 160 °C, 4 h. (b) Effect of removal of catalyst Ru/TiO₂. Reaction conditions: Ru/TiO₂ (0.1 g), aniline (0.4 mmol), cyclohexene (1.0 mmol), CO (0.5 MPa), H₂ (0.5 MPa), THF (2.0 mL), 160 °C.

In order to further investigate the reason for the deactivation of the catalyst, TEM, XRD, and IR characterization of the catalyst after reaction was conducted and shown in Figure S2. The TEM characterization of the catalyst after reaction was given in Figure S2A. The resulting data indicate that the Ru particle size increases slightly from 1.8 nm to 2.0 nm after reaction. In addition, XRD characterization was conducted for the catalyst after reaction and no peaks related to Ru species were observed (Figure S2B), indicating that Ru species are highly dispersed on TiO₂, which is in accordance with the TEM characterization result. Hereinafter, Ru/TiO₂ catalyst was quite stable after reaction.

The IR characterization of the catalyst after reaction was conducted and shown in Figure S2B. Three remarkable peaks at 2074, 2002 and 1884 cm⁻¹ were observed on Ru/TiO₂ after reaction. According to the references,¹ the peak at 2074 cm⁻¹ was assigned to the multicarbonyl species on the partially oxidized Ru sites [Ru^{$\delta+1$}(CO)_x], although there is no consensus regarding the oxidation state of Ru (δ) and the number of carbonyl groups (*x*). The peak at 2002 and 1884 cm⁻¹ were usually assigned to the metallic Ru particles. The

appearance of the peak at 1884 cm⁻¹ was due to the slight increase of the Ru particle size after reaction.² The IR result indicates that the excess amount of CO can absorbed on Ru/TiO₂ surface, which might have negative effect on the absorption of other substrates in the next reaction recycle.



Figure S2. TEM (A), XRD (B) and IR (C) of Ru/TiO₂ after reaction.



Figure S3. The hydrogenation of *p*-nitrotoluene. Reaction conditions: Ru/TiO_2 (0.1 g), *p*-nitrotoluene (0.2 mmol), H₂ (0.5 MPa), CO (0.5 MPa), 160 °C, 8 h.



Figure S4. The product distribution of the reaction to synthesize amine (1) from aniline, cyclohexene, CO, and H₂. Reaction conditions: catalyst (0.1 g), aniline (0.4 mmol), cyclohexene (1.0 mmol), 160 °C, 3 h.



Figure S5. (A) STEM image of Ru/TiO₂. (B) The HRTEM image of Ru/TiO₂. (B) and (C) HAADF-STEM image of Ru/TiO₂. (E) Elemental mapping image of Ru. (F) Elemental mapping image of Ti.

Entry	Catalysts	Reaction systems	<i>T</i> /°C	P _{CO} / MPa	P _{H2} / MPa	Yield of products /%	Ref. ^[a]	Published year
1	Ru/TiO ₂	heterogen eous	165	0.5	0.5	91	This work	-
2	Rhodium exchanged titanosilicates (ETS-10 and ETS-4)	heterogen eous	100	1.35	5.4	>95	10	2014
3	[Rh(cod) ₂]BF 4 (0.1 mol%); IPHOS (0.4 mol%)	homogene ous	120	1.0	2.0	>99	6a	2002
4	[Rh(cod) ₂]BF	homogene ous	125	0.7	3.3	98	6b	2003
5	Ru ₃ (CO) ₁₂	homogene ous	130	1.0	5.0	93	7a	2013
6	Ru ₃ (CO) ₁₂ (0. 5 mol%); K ₂ CO ₃ (20%)	homogene ous	130	4.0	-	81	7b	2014
7	Rh(acac)(CO)	homogene ous	50	0.2	0.2	89	8	2017
8	$[Rh(COD)_2]B$ F4; pH = 4.8 HCO_2Na buffer	homogene ous	80	H ₂ :C0 0.34	D (1:1, MPa)	91	9	2018

Table S2. Summary of various catalytic systems for the synthesis of amines via hydroaminomethylation of olefins.

[a] The number were in accordance with the reference numbers in the manuscript.

Characterization of products

Product 1.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **1** as a light yellow liquid in 84% yield.³ ¹H NMR (400 MHz, DMSO) δ = 7.03 (t, *J* = 7.9, 2H), 6.58 – 6.43 (m, 3H), 5.53 (s, 1H), 2.02 (t, *J* = 5.9, 2H), 1.69 (ddd, *J* = 24.0, 19.3, 8.5, 5H), 1.31 – 0.73 (m, 6H). ¹³C NMR (101 MHz, DMSO) δ = 149.66, 129.26, 115.50, 112.22, 49.94, 37.28, 31.29, 26.66, 26.03.

Product 1a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **1a** as a light yellow liquid in 85% yield. Spectral data were in accordance with those previously reported.⁴ ¹H NMR (400 MHz, DMSO) δ = 6.85 (d, *J* = 8.2, 2H), 6.45 (d, *J* = 8.4, 2H), 5.30 (t, *J* = 5.8, 1H), 2.79 (t, *J* = 6.3, 2H), 2.12 (s, 3H), 1.69 (ddd, *J* = 30.1, 24.2, 11.5, 5H), 1.32 – 0.79 (m, 6H). ¹³C NMR (101 MHz, DMSO) δ = 147.45, 129.71, 129.71, 123.80, 112.39, 112.39, 50.27, 37.30, 31.31, 31.31, 26.68, 26.04, 26.04, 20.52.

Product 2a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **2a** as a

light yellow liquid in 72% yield.⁵ Spectral data were in accordance with those previously reported. ¹H NMR (400 MHz, DMSO) $\delta = 6.75 - 6.62$ (m, 2H), 6.56 - 6.42 (m, 2H), 5.09 (s, 1H), 3.62 (s, 3H), 2.77 (d, J = 6.2, 2H), 1.73 - 1.43 (m, 4H), 1.26 - 0.77 (m, 7H). ¹³C NMR (101 MHz, DMSO) $\delta = 150.78, 144.03, 115.04, 115.04, 113.25, 113.25, 50.88, 50.88, 37.35, 31.35, 31.35, 26.69, 26.06, 26.06.$

Product 3a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **3a** as a light yellow liquid in 97% yield.⁶ Following was the ¹H NMR spectrum of **3a**. ¹H NMR (400 MHz, DMSO) δ = 7.05 (t, *J* = 6.0, 2H), 6.54 (t, *J* = 6.0, 2H), 5.79 (t, *J* = 5.6, 1H), 2.81 (t, *J* = 6.2, 2H), 1.82 – 1.56 (m, 5H), 1.02 (dddd, *J* = 26.2, 18.7, 17.0, 8.8, 6H). ¹³C NMR (101 MHz, DMSO) δ = 148.56, 128.94, 128.94, 118.65, 113.50, 113.50, 49.92, 37.18, 31.20, 31.20, 26.61, 25.98, 25.98.

Product 4a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **4a** as a light yellow liquid in 97% yield.⁷ Following was the ¹H NMR and ¹³C NMR spectrum of **4a**. ¹H NMR (400 MHz, CD3CN) δ = 6.92 – 6.80 (m, 12H), 6.66 – 6.47 (m, 2H), 4.32 (s, 1H), 2.86 (d, *J* = 6.7, 2H), 1.85 – 1.12 (m, 11H). ¹³C NMR (101 MHz, CD₃CN) δ = 156.64, 146.51, 115.86, 115.64, 113.49, 113.42, 51.14, 37.81, 31.53, 31.53, 26.96, 26.36, 26.36. Product 5a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **5a** as a light yellow liquid in 96% yield. Following was the ¹H NMR spectrum of **5a**. ¹H NMR (400 MHz, DMSO) δ = 7.04 – 6.88 (m, 2H), 6.45 (t, *J* = 7.4, 2H), 4.69 (s, *J* = 5.2, 1H), 2.90 (t, *J* = 6.2, 2H), 2.06 (s, 3H), 1.83 – 1.50 (m, 6H), 1.19 (dd, J = 21.9, 8.5, 5H). ¹³C NMR (101 MHz, DMSO) δ = 147.06, 130.14, 127.16, 121.85, 115.63, 109.32, 50.03, 36.82, 31.36, 31.36, 26.73, 26.05, 26.05, 18.13.

Product 6a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **6a** as a light yellow liquid in 60% yield.⁸ Following were the ¹H NMR and ¹³C NMR spectra of **6a**. ¹H NMR (400 MHz, DMSO) $\overline{\delta}$ = 6.80 (tt, *J* = 13.7, 6.7, 2H), 6.60 – 6.35 (m, 2H), 3.76 (s, 3H), 2.89 (t, *J* = 6.4, 2H), 1.84 – 1.44 (m, 5H), 1.36 – 0.70 (m, 6H). ¹³C NMR (101 MHz, DMSO) $\overline{\delta}$ = 146.67, 138.75, 121.52, 115.55, 110.12, 109.40, 55.73, 49.72, 37.10, 31.22, 31.22, 26.66, 25.97, 25.97.

Product 7a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **7a** as a light yellow liquid in 97% yield. Following was the ¹H NMR spectrum of **7a**. ¹H NMR (400 MHz, DMSO) δ = 7.29 – 6.95 (m, 2H), 6.76 – 6.35 (m, 2H), 5.19 (s,

1H), 2.97 (t, J = 6.2, 2H), 1.65 (ddd, J = 22.9, 14.5, 10.0, 5H), 1.06 (dt, J = 22.6, 18.4, 6H). ¹³C NMR (101 MHz, DMSO) $\delta = 144.69$, 129.37, 128.43, 118.04, 116.44, 111.58, 49.46, 36.79, 31.06, 26.63, 25.93.

Product 8a.

Ba H

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **8a** as a light yellow liquid in 82% yield. Following was the ¹H NMR spectrum of **8a**. ¹H NMR (400 MHz, CD₃CN) δ = 7.01 (dd, *J* = 14.9, 7.5 Hz, 2H), 6.56 (dd, *J* = 7.6, 5.1 Hz, 2H), 4.05 (s, 1H), 2.97 (d, *J* = 6.7 Hz, 2H), 2.46 (q, *J* = 7.5 Hz, 2H), 1.96 – 1.57 (m, 7H), 1.32 – 1.19 (m, 4H), 1.17 (t, *J* = 7.5 Hz, 3H). ¹³C NMR (101 MHz, CD₃CN) δ = 146.64, 128.52, 127.99, 127.35, 117.87, 116.59, 110.19, 50.52, 37.55, 31.66, 31.66, 27.01, 26.40, 26.40, 24.23, 13.18

Product 9a.

ya

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **9a** as a light yellow liquid in 96% yield. Following were the ¹H NMR and ¹³C NMR of **9a**. ¹H NMR (400 MHz, DMSO) δ = 6.90 (d, *J* = 7.4, 2H), 6.69 (t, *J* = 7.4, 1H), 3.58 (s, 1H), 2.71 (d, *J* = 6.5, 2H), 2.20 (s, 6H), 1.81 (d, *J* = 12.9, 2H), 1.72 – 1.57 (m, 3H), 1.55 – 1.32 (m, 2H), 1.28 – 1.06 (m, 4H). ¹³C NMR (101 MHz, DMSO) δ = 147.01, 129.43, 129.43, 128.92, 128.92, 121.34, 54.90, 38.93, 38.93, 31.26, 26.70, 26.09, 26.09, 18.80, 18.80.

Product 10a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **10a** as a light yellow liquid in 75% yield. Following were the ¹H NMR and ¹³C NMR of **10a**. ¹H NMR (400 MHz, CD₃CN) δ = 7.00 (d, *J* = 7.5, 2H), 6.87 (t, *J* = 7.5, 1H), 3.11 (s, 1H), 2.72 (d, *J* = 6.5, 2H), 2.63 (q, *J* = 7.5, 4H), 1.99 – 1.62 (m, 7H), 1.29 (dd, *J* = 25.0, 12.0, 4H), 1.18 (t, *J* = 7.5, 6H). ¹³C NMR (101 MHz, CD₃CN) δ = 146.29, 137.10, 127.24, 127.24, 122.91, 117.87, 57.37, 39.63, 31.65, 31.65, 27.01, 26.46, 26.46, 24.59, 24.59, 15.02, 15.02.

Product 11a



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **11a** as a light yellow liquid in 41% yield. Following was the ¹H NMR and ¹³C NMR spectrum of **11a**. ¹H NMR (400 MHz, DMSO) δ = 7.69 (d, *J* = 7.4, 1H), 6.67 (dd, *J* = 8.9, 7.4, 1H), 6.59 – 6.53 (m, 2H), 3.74 (s, 6H), 3.01 (d, *J* = 6.6, 2H), 1.73 – 1.57 (m, 6H), 1.28 (dd, *J* = 17.7, 11.3, 5H). ¹³C NMR (101 MHz, DMSO) δ = 150.04, 150.04, 128.00, 118.62, 105.91, 105.91, 56.40, 56.40, 53.08, 38.45, 30.98, 30.98, 26.71, 25.99, 25.99.

Product 13a.

13a

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1: 100 as eluent to get pure product 1**3a** as a red liquid in 99% yield. Following were the ¹H NMR and ¹³C NMR spectra of

13a. ¹H NMR (400 MHz, DMSO) δ = 8.18 (d, *J* = 8.2, 1H), 7.73 (d, *J* = 8.1, 1H), 7.47 – 7.35 (m, 2H), 7.26 (dd, *J* = 14.5, 6.7, 1H), 7.05 (d, *J* = 8.1, 1H), 6.45 (d, *J* = 7.6, 1H), 6.12 (t, *J* = 5.4, 1H), 3.05 (t, *J* = 6.1, 2H), 1.90 – 1.55 (m, 7H), 1.29 – 1.06 (m, 4H). ¹³C NMR (101 MHz, DMSO) δ = 144.75, 134.56, 128.35, 127.33, 125.98, 124.22, 123.39, 122.13, 115.26, 103.12, 50.22, 36.50, 31.51, 31.51, 26.70, 26.06, 26.06.

Product 17a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **17a** as a light yellow liquid in 99% yield.⁹ Following were the ¹H NMR and ¹³C NMR spectra of **17a**. ¹H NMR (400 MHz, DMSO) δ = 7.13 (dd, *J* = 8.6, 7.3, 2H), 6.71 – 6.28 (m, 3H), 3.12 (d, *J* = 6.8, 2H), 2.89 (s, 3H), 1.67 (t, *J* = 18.4, 6H), 1.20 – 0.84 (m, 5H). ¹³C NMR (101 MHz, DMSO) δ = 149.67, 129.37, 115.47, 111.89, 55.89, 42.50, 36.71, 31.03, 26.57, 25.98.

Product 18a

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **18a** as a light yellow liquid in 85% yield. Following were the ¹H NMR and ¹³C NMR spectra of **18a**. ¹H NMR (400 MHz, DMSO) δ = 7.12 (t, *J* = 7.9, 2H), 6.62 (d, *J* = 8.5, 2H), 6.52 (t, *J* = 7.2, 1H), 3.41 – 3.27 (m, 2H), 3.04 (d, J=7.0, 2H), 1.84 – 1.09 (m, 11H), 1.03 (t, *J* = 6.9, 4H). ¹³C NMR (101 MHz, DMSO) δ = 148.39, 129.48, 129.48, 115.17, 112.07, 112.07, 56.67, 45.54, 36.77, 31.06, 31.06. 26.62, 26.05, 26.05, 11.93. Product 19a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:10 as eluent to get pure product **19a** as a light yellow liquid in 45% yield.¹⁰ Following were the ¹H NMR and ¹³C NMR spectra of **19a**. ¹H NMR (400 MHz, DMSO) δ = 7.25 (t, *J* = 7.8, 4H), 7.00 – 6.84 (m, 6H), 3.53 (d, *J* = 7.1, 2H), 1.81 – 1.57 (m, 6H), 1.38 – 0.98 (m, 5H). ¹³C NMR (101 MHz, DMSO) δ = 148.79, 148.79, 129.70, 129.70, 129.70, 129.70, 121.37, 121.37, 121.15, 121.15, 121.15, 121.15, 58.54, 36.73, 31.06, 31.06, 26.55, 26.00, 26.00.

Product 20a



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **20a** as a light yellow liquid in 96% yield. Following were the ¹H NMR and ¹³C NMR spectra of **20a**. ¹H NMR (400 MHz, DMSO) δ = 7.45 – 7.16 (m, 10H), 3.47 (s, 4H), 2.14 (d, *J* = 7.2, 2H), 1.79 (d, *J* = 11.4, 2H), 1.68 – 1.36 (m, 4H), 1.36 – 0.85 (m, 7H). ¹³C NMR (101 MHz, DMSO) δ = 140.08, 140.08, 128.92, 128.92, 128.92, 128.92, 128.61, 128.61, 128.61, 128.61, 127.28, 127.28, 60.57, 58.62, 58.62, 35.35, 31.64, 31.64, 26.79, 25.96, 25.96.

Product 25a.

25a(L)

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **25a** as a

light yellow liquid in 80% yield. Following were the ¹H NMR and ¹³C NMR spectra of **25a(L)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.35 – 7.03 (m, 7H), 6.63 – 6.53 (m, 3H), 4.33 (s, 1H), 3.08 (dd, *J* = 12.4, 6.7, 2H), 2.76 – 2.67 (m, 2H), 1.87 (dd, *J* = 14.7, 7.6, 2H). ¹³C NMR (101 MHz, CD₃CN) δ = 142.89, 141.18, 129.61, 129.61, 128.96, 128.93, 126.34, 126.34, 116.87, 112.84, 112.84, 43.40, 33.48, 31.46.

Product 26a.

26a(L)

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **26a** as a light yellow liquid in 78% yield. Following were the ¹H NMR and ¹³C NMR spectra of **26a(L)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.23 – 6.99 (m, 6H), 6.64 – 6.51 (m, 3H), 4.33 (s, 1H), 3.06 (t, *J* = 7.0, 2H), 2.71 – 2.62 (m, 2H), 2.29 (s, 3H), 1.90 – 1.80 (m, 2H). ¹³C NMR (101 MHz, CD₃CN) δ = 149.68, 139.70, 135.78, 129.61, 129.61, 129.52, 129.52, 128.86, 128.86, 116.85, 112.83, 112.83, 43.40, 33.04, 31.52, 20.59.

Product 27a.

27a(L)

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **27a** as a light yellow liquid in 80% yield. Following were the ¹H NMR and ¹³C NMR spectra of **27a(L)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.35 – 7.17 (m, 2H), 7.17 – 6.95 (m, 4H), 6.63 – 6.34 (m, 3H), 4.33 (s, 1H), 3.06 (t, *J* = 6.9, 2H), 2.74 – 2.65 (m, 2H), 1.86 (td, *J* = 14.2, 7.1, 2H). ¹³C NMR (101 MHz, CD₃CN) δ = 160.54, 149.64, 138.86, 130.60, 130.53, 129.62, 117.88, 117.88, 116.90, 115.51, 115.30, 112.85, 43.28, 32.60, 31.50.

Product 28a.

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **28a** as a light yellow liquid in 30% yield. Following were the ¹H NMR and ¹³C NMR spectra of **28a(L)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.38 – 7.16 (m, 5H), 7.12 – 7.03 (m, 2H), 6.53 (ddd, *J* = 9.6, 7.9, 1.0, 3H), 4.23 (s, 1H), 2.91 (ddd, *J* = 21.6, 12.9, 7.2, 3H), 1.86 (q, *J* = 7.3, 2H), 1.26 (d, *J* = 7.0, 3H).

Product 29a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **29a** as a light yellow liquid in 76% yield. Following were the ¹H NMR and ¹³C NMR spectra of **29a(L+B)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.30 – 6.84 (m, 4H), 6.75 – 6.33 (m, 6H), 4.27 (s, 2H), 3.06 (dt, *J* = 14.1, 7.2, 3H), 3.01 – 2.95 (m, 0.5H), 2.89 – 2.80 (m, 0.5H), 1.62 – 1.52 (m, 3H), 1.35 (dddd, *J* = 18.4, 14.6, 9.9, 5.3, 14H), 0.95 – 0.75 (m, 9H). ¹³C NMR (101 MHz, CD₃CN) δ = 149.85, 149.78, 149.78, 129.59, 129.59, 116.74, 116.62, 112.81, 112.81, 112.77, 112.77, 50.31, 43.93, 37.33, 32.90, 32.01, 29.60, 27.16, 22.97, 20.40, 17.81, 14.26, 13.94.

Product 30a.

30a(L)

The crude product was purified by silicagel column chromatography using

ethyl acetate: petroleum ether = 1:200 as eluent to get pure product **30a** as a light yellow liquid in 97% yield. Following were the ¹H NMR and ¹³C NMR spectra of **30a(L)**. ¹H NMR (400 MHz, CD₃CN) δ = 7.19 – 7.06 (m, 2H), 6.66 (t, J = 6.9, 2H), 6.55 (t, J = 7.2, 1H), 3.34 – 3.15 (m, 4H), 1.64 – 1.43 (m, 4H), 1.39 – 1.26 (m, 4H), 0.92 (dt, J = 6.1, 4.6, 6H). ¹³C NMR (101 MHz, CD₃CN) δ = 148.88, 129.64, 129.64, 115.49, 112.32, 112.32, 50.82, 50.82, 29.69, 29.69, 20.54, 20.54, 13.86, 13.86.

Product 31a.

31a

The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **31a** as a light yellow liquid in 98% yield.¹¹ Following were the ¹H NMR and ¹³C NMR spectra of **31a**. ¹H NMR (400 MHz, DMSO) δ = 7.12 (t, *J* = 8.0, 2H), 6.61 (d, *J* = 8.2, 2H), 6.52 (t, *J* = 7.2, 1H), 3.26 – 3.14 (m, 4H), 1.62 – 1.34 (m, 4H), 0.87 (t, J=7.4, 6H). ¹³C NMR (101 MHz, DMSO) δ = 148.28, 129.51, 129.51, 115.19, 111.86, 111.86, 52.39, 52.39, 20.41, 20.41, 11.68, 11.68.

Product 32a.



The crude product was purified by silicagel column chromatography using ethyl acetate: petroleum ether = 1:100 as eluent to get pure product **32a** as a light yellow liquid in 96% yield.¹² Following were the ¹H NMR and ¹³C NMR spectra of **32a**. ¹H NMR (400 MHz, DMSO) δ = 7.12 (t, *J* = 7.9, 2H), 6.72 – 6.48 (m, 3H), 3.23 (d, *J* = 7.3, 4H), 1.77 – 1.00 (m, 18H). ¹³C NMR (101 MHz, DMSO) δ = 148.70, 129.38, 129.38, 115.52, 112.93, 112.93, 56.25, 56.25, 38.28, 38.28, 30.57, 30.57, 30.57, 30.57, 24.96, 24.96, 24.96, 24.96.

¹H NMR and ¹³C NMR spectra of products



















































References:

1. Abdel-Mageed, A. M.; Widmann, D.; Olesen, S. E.; Chorkendorff, I.; Behm, R. J., Selective CO Methanation on Highly Active Ru/TiO2 Catalysts: Identifying the Physical Origin of the Observed Activation/Deactivation and Loss in Selectivity. *ACS Catal.* **2018**, *8* (6), 5399-5414.

2. Ftouni, J.; Muñoz-Murillo, A.; Goryachev, A.; Hofmann, J. P.; Hensen, E. J. M.; Lu, L.; Kiely, C. J.; Bruijnincx, P. C. A.; Weckhuysen, B. M., ZrO2 Is Preferred over TiO2 as Support for the Ru-Catalyzed Hydrogenation of Levulinic Acid to γ-Valerolactone. *ACS Catal.* **2016**, *6* (8), 5462-5472.

3. Guo, H.; Wang, B.; Qiu, P.; Gao, R.; Sun, M.; Chen, L., N,S-Codoped Carbon Shells Embedded with Ultrafine Co NPs for Reductive Amination with Formic Acid. *ACS Sustainable Chem. Eng.* **2019**, *7* (9), 8876-8884.

4. Cui, X.; Deng, Y.; Shi, F., Reductive N-Alkylation of Nitro Compounds to N-Alkyl and N,N-Dialkyl Amines with Glycerol as the Hydrogen Source. *ACS Catal.* **2013**, *3* (5), 808-811.

5. Kharitonov, V. B.; Ostrovskii, V. S.; Nelyubina, Y. V.; Muratov, D. V.; Chusov, D.; Loginov, D. A., Tris(pyrazolyl)borate rhodium complexes. Application for reductive amination and esterification of aldehydes in the presence of carbon monoxide. *J Organomet Chem* **2020**, *925*.

6. Pedrajas, E.; Sorribes, I.; Junge, K.; Beller, M.; Llusar, R., Selective reductive amination of aldehydes from nitro compounds catalyzed by molybdenum sulfide clusters. *Green Chem.* **2017**, *19* (16), 3764-3768.

7. Li, Z.-L.; Sun, K.-K.; Wu, P.-Y.; Cai, C., Iron-Catalyzed Regioselective α -C–H Alkylation of N-Methylanilines: Cross-Dehydrogenative Coupling between Unactivated C(sp3)–H and C(sp3)–H Bonds via a Radical Process. *The Journal of organic chemistry* **2019**, *84* (11), 6830-6839.

8. Sang, C. S.; Young, G. K.; Chil, H. D.; Hong, S. K.; Tae, J. K., Selective synthesis of N-(cyclohexymethyl)-N-alkylamines from primary amines and pimelaldehyde using tetracarbonylhydridoferrate, HFe(CO)4–, as a reducing reagent. *Tetrahedron Lett.* **1990**, *31* (1), 105-106.

9. Cabrero-Antonino, J. R.; Adam, R.; Junge, K.; Beller, M., A general protocol for the reductive Nmethylation of amines using dimethyl carbonate and molecular hydrogen: mechanistic insights and kinetic studies. *Catal. Sci. Technol.* **2016**, *6* (22), 7956-7966.

10. Matsumura, T.; Nakada, M., Direct reductive amination using triethylsilane and catalytic bismuth(III) chloride. *Tetrahedron Lett.* **2014**, *55* (10), 1829-1834.

11. Sorribes, I.; Junge, K.; Beller, M., Direct catalytic N-alkylation of amines with carboxylic acids. *J. Am. Chem. Soc.* **2014**, *136* (40), 14314-9.

12. Srivastava, V. K.; Eilbracht, P., Ruthenium carbonyl-complex catalyzed hydroaminomethylation of olefins with carbon dioxide and amines. *Catal. Commun.* **2009**, *10* (14), 1791-1795.