Electronic Supplementary Information

Sterically Controlling 2-Carboxylated Imidazolium Salts for One-Step Efficient Hydration of Epoxides into 1,2-Diols

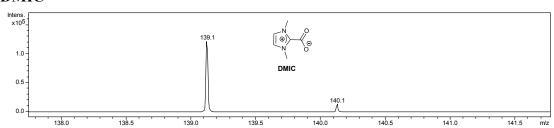
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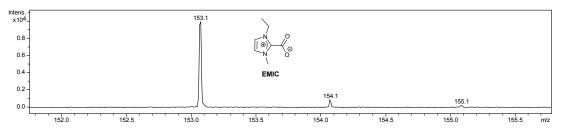
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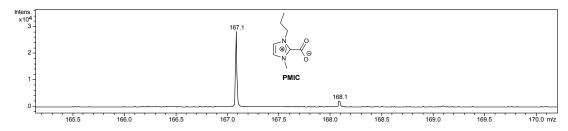
DMIC



EMIC



PMIC



BMIC

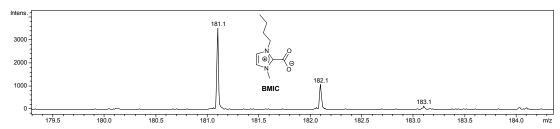


Figure S1. ESI-MS spectra of 2-carboxylated imidazolium salts with different substituent chain lengths from C1 to C4.

Table S1. C/N/H elemental analysis of 2-carboxylated imidazolium salts.

Samples	C [wt.%]	N [wt.%]	C/N molar ratio	C/N theoretical molar ratio
DMIC	19.55	50.47	3.01	3.0
EMIC	16.58	50.22	3.53	3.5
PMIC	13.12	45.08	4.01	4.0
BMIC	12.18	47.57	4.56	4.5

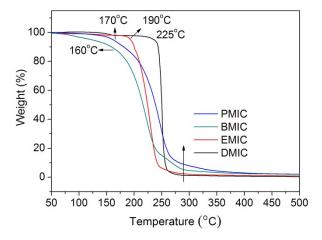
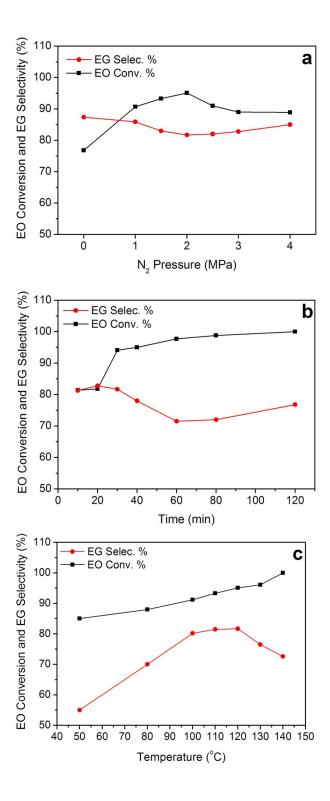


Figure S2. TGA curve of 2-carboxylated imidazolium salts with different chain lengths from C1 to C4.

Table S2. Saturated vapor pressure (PL) of EO under different temperatures calculated from the Aspen Plus TM software

TEMP/°C	50	100	120	130	140
PL/kPa	392.5	1380.1	2073.1	2502.7	2994.2



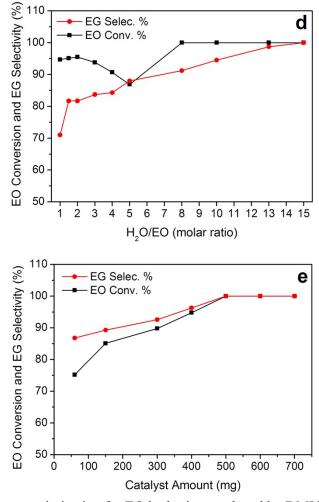


Figure S3. N₂ pressure optimization for EO hydration catalyzed by DMIC (a); Reaction time optimization for EO hydration catalyzed by DMIC (b); Reaction temperature optimization for EO hydration catalyzed by DMIC (c); H₂O/EO molar ratio optimization for EO hydration catalyzed by DMIC (d); Catalyst amount optimization for EO hydration catalyzed by DMIC (e).

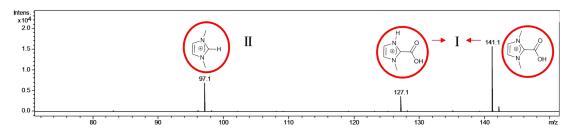


Figure S4. ESI-MS spectra for DMIC in H₂O with positive ion mode