

Supplementary information

An assessment of electrified methanol production from an environmental perspective

Evangelos Delikonstantis^{1,2}, Elorri Igos^{3}, Stavros-Alexandros Theofanidis^{2,4}, Enrico Benetto³, Guy B. Marin¹, Kevin Van Geem¹ and Georgios D. Stefanidis^{1,5*}*

¹ Laboratory for Chemical Technology, Ghent University, Technologiepark 125, B-9052 Ghent, Belgium

² AristEng S.à r.l., 77, Rue de Merl, L-2146, Luxembourg City, Luxembourg

³ Luxembourg Institute of Science and Technology (LIST), 5, Avenue des Hauts-Fourneaux, L-4362, Esch-sur-Alzette, Luxembourg

⁴ Department of Chemical Engineering, Aristotle University of Thessaloniki, University Campus, 54124, Thessaloniki, Greece

⁵ School of Chemical Engineering, National Technical University of Athens, Iroon Polytechniou 9, Athens, 15780, Greece

Correspondence to:

Georgios D. Stefanidis

Tel: + 32 (0) 9 331 17 41

e-mail: georgios.stefanidis@ugent.be,
gstefani@mail.ntua.gr

Elorri Igos

Tel: +352 275 888 5015

e-mail: elorri.igos@list.lu

Table S-1. Process conditions and flows of the major streams in plasma-assisted methanol production process.

		DMR in	DMR out	SCMR in	SCMR out	Syngas	MeOH-R in	MeOH
Mass Flows	kg/hr	30026	30026	53229	53229	19826	68791	16353
CO ₂	kg/hr	22005	12595	12595	14171	1051	12726	31
CO	kg/hr		8095	8095	16445	16273	45030	
H ₂	kg/hr		647	647	2503	2502	9530	
Natural gas	kg/hr	8021	3979	3979			5	
C ₂ H ₆	kg/hr		75	75				
C ₂ H ₄	kg/hr		175	175				
C ₂ H ₂	kg/hr		879	879				
C ₃ H ₈	kg/hr		14	14				
Water	kg/hr		2450	26771	20110		5	1
MeOH	kg/hr						1495	16321
Coke	kg/hr		1117					
Pressure	bar	1	1	1	1	30	30	1
Temperature	°C	25	200	1000	1000	250	250	30

Table S-2. Process conditions and flows of the major streams in thermocatalytic methanol production process.

		DMR in	DMR out	WGS in	WGS out	SMR in	SMR out	Syngas	MeOH-R in	MeOH
Mass Flows	kg/hr	18766	18766	25979	25979	22351	22351	44052	218976	3557
CO ₂	kg/hr	13753	1315	1315	7739		71	7806	85723	87
CO	kg/hr		15246	15246	12281		18068	30348	90041	
H ₂	kg/hr		1039	1039	1576		3914	5490	28607	
Natural gas	kg/hr	5013	750	750	107	10528	154	260	8451	
C ₂ H ₆	kg/hr									
C ₂ H ₄	kg/hr									
C ₂ H ₂	kg/hr									
C ₃ H ₈	kg/hr									
Water	kg/hr		377	7629	4276	11823	144	148	253	2
MeOH	kg/hr								5901	35468
Coke	kg/hr		39							
Pressure	bar	1	1	1	1	1	1	30	30	1
Temperature	°C	750	750	750	750	1000	1000	250	250	30

Table S-3. Detailed material and energy flows of the plasma-assisted DMR-to-MeOH process utilizing natural gas as carbon source.

Plasma-assisted DMR-to-MeOH ; feed: natural gas + CO₂						
		kg/h	m³/h (25°C; 1bar)	kg/kg_{MeOH}	m³/kg_{MeOH}	
Raw material	Natural gas (CH ₄)	8021	12376	0.49	0.76	
	CO ₂	8701	(CO ₂ from 2nd flash drum is recycled)	0.53		
	Water	4221	(water from 1st flash drum is recycled)	0.26		
Utility		MW		MJ/kg_{MeOH}		
	Hot utility (oil)	30.7		6.75		
	Cold utility (water)	-39.6		393	kg/kg_{MeOH}	
	Cold utility (refrig.)	-23.9	47	1	kWh/Kg_{MeOH}	
			kW		kWh/kg_{MeOH}	
	Electricity	103594		6.3		
		4088		0.25	CW recirculation (20 bar loss)	
	Make-up water			7.9	kgwater/kg_{MeOH} (5% loss)	
Catalyst Bi		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	m³/kg_{MeOH}	
	Mg(NO ₃) ₂ ·6H ₂ O	1.3 · 10 ⁻³	1.68	7.8 · 10 ⁻⁸		
	Al(NO ₃) ₃ ·9H ₂ O	3.7 · 10 ⁻³	4.87	2.3 · 10 ⁻⁷		
	Ni(NO ₃) ₂ ·6H ₂ O	3.8 · 10 ⁻⁴	0.50	2.3 · 10 ⁻⁸		
	Fe(NO ₃) ₃ ·9H ₂ O	3.1 · 10 ⁻⁵	4 · 10 ⁻²	1.9 · 10 ⁻⁹		
	NH ₄ OH (L/kg _{cat} ; density = 0.88 kg/L)	1.3 · 10 ⁻³	1.76		8.2 · 10 ⁻⁸	
	Catalyst demand Kg (4 mol _{CO} /s/Kg _{cat})	20				
Catalyst lifetime Kg _{cat} /h (3 years)	8.0 · 10 ⁻⁴					
Catalyst MeOH		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	m³/kg_{MeOH}	
	Cu(NO ₃) ₂ ·3H ₂ O	7.9 · 10 ⁻²	2.41	4.8 · 10 ⁻⁶		
	Al(NO ₃) ₃ ·9H ₂ O	4.4 · 10 ⁻²	1.33	2.7 · 10 ⁻⁶		
	Zn(NO ₃) ₂ ·6H ₂ O	3.4 · 10 ⁻²	1.03	2.1 · 10 ⁻⁶		

	Catalyst demand Kg ($2.7 \text{ mol}_{\text{MeOH}}/\text{s}/\text{Kg}_{\text{cat}}$) x 10 (catalyst guard)	575		
	Catalyst lifetime $\text{Kg}_{\text{cat}}/\text{h}$ (2 years)	$3.3 \cdot 10^{-2}$		
		kg/h	Molecular weights (kg/koml)	kg/kgMeOH
	CO ₂ (purge flaring+Waste 1/vapor distillate)	2144	44.1 CO ₂	0.13
	H ₂ O (purge flaring+vapor distillate)	2144	18.0 H ₂ O	0.13
Wastes	CO ₂ biogenic	-	2498 demand for flaring	-
	Carbon	1117		0.07
	Water (Waste 2/Bottom liquid)	1685		0.10
	Catalysts	$3.0 \cdot 10^{-2}$		$2 \cdot 10^{-6}$
		kg/h		
Product	MeOH	16353		

Table S-4. Detailed material and energy flows of the plasma-assisted DMR-to-MeOH process utilizing biogas as carbon source.

Plasma-assisted DMR-to-MeOH ; feed: biogas					
		kg/h	m³/h (25°C; 1bar)	kg/kg_{MeOH}	m³/kg_{MeOH}
Raw material	Natural gas (CH ₄)	-	-	-	-
	Biogas	30026		1.84	
	Water	4221	(water from 1st flash drum is recycled)	0.26	
Utility		MW		MJ/kg_{MeOH}	
	Hot utility (oil)	30.7		6.75	
	Cold utility (water)	-39.6		393	kg/kg_{MeOH}
	Cold utility (refrig.)	-23.9	47	1	kWh/Kg_{MeOH}
		kW		kWh/kg_{MeOH}	
	Electricity	103594		6.3	
		4088		0.25	CW recirculation (20 bar loss)
	Make-up water			7.9	kgwater/kg_{MeOH} (5% loss)
Catalyst Bi		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	Mg(NO ₃) ₂ ·6H ₂ O	1.3 · 10 ⁻³	1.68	7.8 · 10 ⁻⁸	
	Al(NO ₃) ₃ ·9H ₂ O	3.7 · 10 ⁻³	4.87	2.3 · 10 ⁻⁷	
	Ni(NO ₃) ₂ ·6H ₂ O	3.8 · 10 ⁻⁴	0.50	2.3 · 10 ⁻⁸	
	Fe(NO ₃) ₃ ·9H ₂ O	3.1 · 10 ⁻⁵	4 · 10 ⁻²	1.9 · 10 ⁻⁹	
	NH ₄ OH (L/kg _{cat} ; density = 0.88 kg/L)	1.3 · 10 ⁻³	1.76		8.2 · 10 ⁻⁸
	Catalyst demand Kg (4 mol _{CO} /s/Kg _{cat}) Catalyst lifetime Kg _{cat} /h (3 years)	20 8.0 · 10 ⁻⁴			
H₂S removal		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	ZnO Material demand Kg/h (13.9 kgbiogas/ZnO)	2160		0.13	

		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	m³/kg_{MeOH}
Catalyst MeOH	Cu(NO ₃) ₂ ·3H ₂ O	7.9 ·10 ⁻²	2.41	4.8 ·10 ⁻⁶	
	Al(NO ₃) ₃ ·9H ₂ O	4.4 ·10 ⁻²	1.33	2.7 ·10 ⁻⁶	
	Zn(NO ₃) ₂ ·6H ₂ O	3.4 ·10 ⁻²	1.03	2.1 ·10 ⁻⁶	
	Catalyst demand Kg (2.7 mol _{MeOH} /s/K _{gcat}) x 10 (catalyst guard)	575			
	Catalyst lifetime Kg _{cat} /h (2 years)	3.3 ·10 ⁻²			
		kg/h	Molecular weights (kg/koml)	kg/kg_{MeOH}	
Wastes	CO ₂ (purge flaring+Waste 1/vapor distillate)	2144	44.1 CO ₂	0.13	
	H ₂ O (purge flaring+vapor distillate)	2144	18.0 H ₂ O	0.13	
	CO ₂ biogenic	13304	2498 demand for flaring	0.81	
	Carbon	1117		7.0 ·10 ⁻²	
	Water (Waste 2/Bottom liquid)	1685		0.10	
	Catalysts	3.0 ·10 ⁻²		2 ·10 ⁻⁶	
		kg/h			
Product	MeOH	16353			

Table S-5. Detailed material and energy flows of the thermocatalytic DMR-to-MeOH process utilizing natural gas as carbon source.

Thermocatalytic DMR-to-MeOH ; feed: natural gas + CO₂					
		kg/h	m³/h (25°C; 1bar)	kg/kg_{MeOH}	m³/kg_{MeOH}
Raw material	Natural gas (CH ₄)	15541	23978	0.44	0.67
	CO ₂	13753		0.39	
	Water	14797	(water from 1st, 2nd & 3rd flash drum is recycled)	0.42	
Utility		MW		MJ/kg_{MeOH}	
	Hot utility (oil)	100.3		10	
	Cold utility (water)	-103.7		475	kg_{water}/kg_{MeOH}
	Cold utility (refrig.)	-6.6	13	0.13	kWh_{cooling}/Kg_{MeOH}
		kW		kWh/kg_{MeOH}	
	Electricity	38225		1.1	
		10667		0.3	CW recirculation (20 bar loss)
	Make-up water			9.5	kg_{water}/kg_{MeOH} (5% loss)
Catalyst DMR		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	Mg(NO ₃) ₂ ·6H ₂ O	2.5 · 10 ⁻³	1.68	6.9 · 10 ⁻⁸	
	Al(NO ₃) ₃ ·9H ₂ O	7.1 · 10 ⁻³	4.87	2.0 · 10 ⁻⁷	
	Ni(NO ₃) ₂ ·6H ₂ O	7.3 · 10 ⁻⁴	0.50	2.1 · 10 ⁻⁸	
	Fe(NO ₃) ₃ ·9H ₂ O	5.9 · 10 ⁻⁵	0.04	1.6 · 10 ⁻⁹	
	NH ₄ OH (= 2 L/kg _{cat}) (considering density = 0.88 kg/L)	2.6 · 10 ⁻³	1.76		7.0 · 10 ⁻⁵
	Catalyst demand Kg (5.9 mol_{CO}/s/Kg_{cat})	26			
Catalyst lifetime Kg_{cat}/h (2 years)	1.5 · 10 ⁻³				
Catalyst MeOH		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	Cu(NO ₃) ₂ ·3H ₂ O	1.6 · 10 ⁻¹	2.41	4.5 · 10 ⁻⁶	
	Al(NO ₃) ₃ ·9H ₂ O	8.8 · 10 ⁻²	1.33	2.5 · 10 ⁻⁶	
	Zn(NO ₃) ₂ ·6H ₂ O	6.8 · 10 ⁻²	1.03	1.9 · 10 ⁻⁶	
	Catalyst demand Kg (2.7 mol _{MeOH} /s/Kg _{cat}) x 10 (catalyst guard)	1154			

	Catalyst lifetime Kg_{cat}/h (2 years)	6.6 · 10 ⁻²		
Catalyst SMR		kg/h	kg/kg_{cat}	kg/kg_{MeOH}
	Mg(NO ₃) ₂ ·6H ₂ O	1.8 · 10 ⁻³	1.68	5.0 · 10 ⁻⁸
	Al(NO ₃) ₃ ·9H ₂ O	5.1 · 10 ⁻³	4.87	1.4 · 10 ⁻⁷
	Ni(NO ₃) ₂ ·6H ₂ O	5.2 · 10 ⁻⁴	0.50	1.5 · 10 ⁻⁸
	Fe(NO ₃) ₃ ·9H ₂ O	4.2 · 10 ⁻⁵	0.04	1.2 · 10 ⁻⁹
	NH ₄ OH (= 2 L/kg _{cat} ; density = 0.88 kg/L)	1.8 · 10 ⁻³	1.76	
	Catalyst demand Kg (3.9 mol_{CO}/s/Kg_{cat})	46		5.0 · 10 ⁻⁵
	Catalyst lifetime Kg_{cat}/h (5 years)	1.0 · 10 ⁻³		
Catalyst WGS		kg/h	kg/kg_{cat}	kg/kg_{MeOH}
	La(NO ₃) ₃ ·6H ₂ O	4.7 · 10 ⁻²	2.20	1.3 · 10 ⁻⁶
	(NH ₄) ₂ Ce(NO ₃) ₆	1.7 · 10 ⁻²	0.80	4.9 · 10 ⁻⁷
	Fe(NO ₃) ₃ ·9H ₂ O	1.6 · 10 ⁻²	0.72	4.4 · 10 ⁻⁷
	Glycerol	4.1 · 10 ⁻²	1.90	1.2 · 10 ⁻⁶
	Catalyst demand Kg (0.4 mol_{CO}/s/Kg_{cat})	378		
	Catalyst lifetime Kg_{cat}/h (2 years)	2.2 · 10 ⁻²		
Wastes		kg/h	Molecular weights (kg/koml)	kg/kg_{MeOH}
	CO ₂ (purge flaring+Waste 1/vapor distillate)	7103	44.1 CO ₂	0.20
	H ₂ O (purge flaring+vapor distillate)	7515	18.0 H ₂ O	0.21
	CO ₂ biogenic	-	*8473 kg/h O ₂ demand for flaring	-
	Carbon	39		1.1 · 10 ⁻³
	Water (Waste 2/Bottom liquid)	2363		7.0 · 10 ⁻²
	Catalysts	9.0 · 10 ⁻²		2.5 · 10 ⁻⁶
Product		kg/h		
	MeOH	35557		

Table S-6. Detailed material and energy flows of the thermocatalytic DMR-to-MeOH process utilizing biogas as carbon source.

Thermocatalytic DMR-to-MeOH ; feed: biogas					
		kg/h	m³/h (25°C; 1bar)	kg/kg_{MeOH}	m³/kg_{MeOH}
Raw material	Natural gas (CH ₄)	10528	16243	0.30	0.46
	Biogas	18766		0.53	
	Water	14797	(water from 1st, 2nd & 3rd flash drum is recycled)	0.42	
Utility		MW		MJ/kg_{MeOH}	
	Hot utility (oil)	100.3		10	
	Cold utility (water)	-103.7		475	kg_{water}/kg_{MeOH}
	Cold utility (refrig.)	-6.6	13	0.13	kWh_{cooling}/Kg_{MeOH}
		kW		kWh/kg_{MeOH}	
	Electricity	38225		1.1	
		10667		0.3	CW recirculation (20 bar loss)
	Make-up water			9.5	kg_{water}/kg_{MeOH} (5% loss)
Catalyst DMR		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	Mg(NO ₃) ₂ ·6H ₂ O	2.5 · 10 ⁻³	1.68	6.9 · 10 ⁻⁸	
	Al(NO ₃) ₃ ·9H ₂ O	7.1 · 10 ⁻³	4.87	2.0 · 10 ⁻⁷	
	Ni(NO ₃) ₂ ·6H ₂ O	7.3 · 10 ⁻⁴	0.50	2.1 · 10 ⁻⁸	
	Fe(NO ₃) ₃ ·9H ₂ O	5.9 · 10 ⁻⁵	0.04	1.6 · 10 ⁻⁹	
	NH ₄ OH (= 2 L/kg _{cat}) (considering density = 0.88 kg/L)	2.6 · 10 ⁻³	1.76		7.0 · 10 ⁻⁵
	Catalyst demand Kg (5.9 mol_{CO}/s/Kg_{cat})	26			
Catalyst lifetime Kgcat/h (2 years)	1.5 · 10 ⁻³				
Catalyst MeOH		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	m³/kg_{MeOH}
	Cu(NO ₃) ₂ ·3H ₂ O	1.6 · 10 ⁻¹	2.41	4.5 · 10 ⁻⁶	
	Al(NO ₃) ₃ ·9H ₂ O	8.8 · 10 ⁻²	1.33	2.5 · 10 ⁻⁶	
	Zn(NO ₃) ₂ ·6H ₂ O	6.8 · 10 ⁻²	1.03	1.9 · 10 ⁻⁶	
	Catalyst demand Kg (2.7 mol _{MeOH} /s/Kg _{cat}) x 10 (catalyst guard)	1154			

	Catalyst lifetime Kg_{cat}/h (2 years)	$6.6 \cdot 10^{-2}$			
		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	
Catalyst SMR	Mg(NO ₃) ₂ ·6H ₂ O	$1.8 \cdot 10^{-3}$	1.68	$5.0 \cdot 10^{-8}$	
	Al(NO ₃) ₃ ·9H ₂ O	$5.1 \cdot 10^{-3}$	4.87	$1.4 \cdot 10^{-7}$	
	Ni(NO ₃) ₂ ·6H ₂ O	$5.2 \cdot 10^{-4}$	0.50	$1.5 \cdot 10^{-8}$	
	Fe(NO ₃) ₃ ·9H ₂ O	$4.2 \cdot 10^{-5}$	0.04	$1.2 \cdot 10^{-9}$	
	NH ₄ OH (= 2 L/kg _{cat} ; density = 0.88 kg/L)	$1.8 \cdot 10^{-3}$	1.76		$5.0 \cdot 10^{-5}$
	Catalyst demand Kg (3.9 mol_{CO}/s/Kg_{cat})	46			
	Catalyst lifetime Kg_{cat}/h (5 years)	$1.0 \cdot 10^{-3}$			
		kg/h	kg/kg_{cat}	kg/kg_{MeOH}	
Catalyst WGS	La(NO ₃) ₃ ·6H ₂ O	$4.7 \cdot 10^{-2}$	2.20	$1.3 \cdot 10^{-6}$	
	(NH ₄) ₂ Ce(NO ₃) ₆	$1.7 \cdot 10^{-2}$	0.80	$4.9 \cdot 10^{-7}$	
	Fe(NO ₃) ₃ ·9H ₂ O	$1.6 \cdot 10^{-2}$	0.72	$4.4 \cdot 10^{-7}$	
	Glycerol	$4.1 \cdot 10^{-2}$	1.90	$1.2 \cdot 10^{-6}$	
	Catalyst demand Kg (0.4 mol_{CO}/s/Kg_{cat})	378			
	Catalyst lifetime Kg_{cat}/h (2 years)	$2.2 \cdot 10^{-2}$			
		kg/h	Molecular weights (kg/koml)	kg/kg_{MeOH}	
Wastes	CO ₂ (purge flaring+Waste 1/vapor distillate)	7103	44.1 CO ₂	0.20	
	H ₂ O (purge flaring+vapor distillate)	7515	18.0 H ₂ O	0.21	
	CO ₂ biogenic	-	*8473 kg/h O ₂ demand for flaring	-	
	Carbon	39		$1.1 \cdot 10^{-3}$	
	Water (Waste 2/Bottom liquid)	2363		$7.0 \cdot 10^{-2}$	
	Catalysts	$9.0 \cdot 10^{-2}$		$2.5 \cdot 10^{-6}$	
		kg/h	kg/k_{gcat}	kg/kg_{MeOH}	
H₂S removal	ZnO				
	Material demand Kg/h (13.9 kgbiogas/ZnO)	1350		0.13	
		kg/h			
Product	MeOH	35557			

Table S-7. Ecoinvent dataset used for each mass and energy flow in the LCI modelling.

<i>Inputs from Technosphere</i>	
Natural gas	Natural gas, high pressure {Europe without Switzerland} market group for Cut-off, U
Biogas	Biogas {CH} anaerobic digestion of manure Cut-off, U
Water	Water, deionised {Europe without Switzerland} market for water, deionised Cut-off, U
Hot utility	Heat, central or small-scale, natural gas {RER} market group for Cut-off, U
Electricity, low voltage	Electricity, low voltage {RER} market group for Cut-off, U
Electricity, high voltage	Electricity, high voltage {RER} market group for Cut-off, U
Cold utility, makeup water	Water, decarbonised {RER} water production, decarbonised adapted
Catalyst DMR/BiMR/SMR	Catalyst Fe-Ni/MgAl ₂ O ₄
Catalyst MeOH	Catalyst Cu-Zn-Al
Catalyst WGS	Catalyst La-Ce-Fe
ZnO for H ₂ S removal	Zinc oxide {GLO} market for Cut-off, U
<i>Emissions to air</i>	
CO ₂ from flaring	Carbon dioxide
H ₂ O from flaring	Water
<i>Outputs to Technosphere</i>	
Waste water	Wastewater, unpolluted {RoW} treatment of, capacity 5E9l/year Cut-off, U
Carbon waste	Inert waste, for final disposal {CH} treatment of inert waste, inert material landfill Cut-off, U
Catalyst waste	Spent catalyst base from ethylene oxide production {GLO} market for Cut-off, U

Table S-8. Key performance indicators (KPI) as function of CH₄ conversion (x) in the DMR reactor. The KPI's were normalized to functional unit: Electricity, HV and LV (y₁ and y₂) = kWh/kg_{MeOH}; Heat, fossil-based (y₃) = MJ/kg_{MeOH}; Heat, electrically conductive (y₇) = kWh/kg_{MeOH}; Natural gas (y₄) = m³/kg_{MeOH}; Biogas (y₅) = m³/kg_{MeOH}; Direct CO₂ (y₆) = kg_{CO2}/kg_{MeOH}. X ranges from 0.38-0.63 and 0.76-0.92 for the plasma-assisted and thermocatalytic DMR-to-MeOH process, respectively.

KPI	Proposed DMR-to-MeOH processes			
	Scenario PNGG, PNGHVPV and PNGHVWM		Scenario PBGG	
HV electricity ¹	(y ₁)	$y_1 = 0.0469x^2 - 0.1554x + 6.3384; R^2=1$	$y_1 = 0.0469x^2 - 0.1554x + 6.3384; R^2=1$	
LV electricity ²	(y ₂)	$y_2 = 0.0172x^2 - 0.1296x + 1.3498; R^2=1$	$y_2 = 0.0172x^2 - 0.1296x + 1.3498; R^2=1$	
Heat	(y ₃)	$y_3 = 0.5481x^2 - 6.616x + 9.981; R^2=1$	$y_3 = 0.5481x^2 - 6.616x + 9.981; R^2=1$	
Natural gas	(y ₄)	$y_4 = -0.0659x + 0.7906; R^2=0.9992$		
Biogas	(y ₅)		$y_5 = -0.139x + 1.6687; R^2=0.9992$	
Direct CO ₂	(y ₆)	$y_6 = 0.0027x^2 + 0.0264x + 0.1156; R^2=1$	$y_6 = 0.0027x^2 + 0.0264x + 0.1156; R^2=1$	
		Scenario TNG	Scenario TBG	
LV electricity	(y ₂)	$y_2 = -0.0084x^2 + 0.0092x + 1.4295; R^2=0.9874$	$y_2 = -0.0084x^2 + 0.0092x + 1.4295; R^2=0.9874$	
Heat	(y ₃)	$y_3 = -1.8869x^2 + 2.4965x + 9.3912; R^2=0.9991$	$y_3 = -1.8869x^2 + 2.4965x + 9.3912; R^2=0.9991$	
Natural gas	(y ₄)	$y_4 = -0.0123x + 0.6847; R^2=0.9995$	$y_4 = -0.0083x + 0.4638; R^2=0.9995$	
Biogas	(y ₅)		$y_5 = -0.0083x + 0.4638; R^2=0.9995$	
Direct CO ₂	(y ₆)	$y_6 = -0.002\ln(x) + 0.1991; R^2=0.9340$	$y_6 = -0.002\ln(x) + 0.1991; R^2=0.9340$	
		Scenario TNGG	Scenario TNGPV and TNGWM	
LV electricity	(y ₂)	$y_2 = -0.5325x^2 + 0.7027x + 4.0382; R^2=0.9993$	$y_2 = -0.0084x^2 + 0.0092x + 1.4295; R^2=0.9874$	
LV-H electricity ³	(y ₇)		$y_7 = -1.8869x^2 + 2.4965x + 9.3912; R^2=0.9991$	
Natural gas	(y ₄)	$y_4 = -0.0123x + 0.6847; R^2=0.9995$	$y_4 = -0.0123x + 0.6847; R^2=0.9995$	
Direct CO ₂	(y ₆)	$y_6 = -0.002\ln(x) + 0.1991; R^2=0.9340$	$y_6 = -0.002\ln(x) + 0.1991; R^2=0.9340$	
		Scenario PNGWM	Scenario TNGWM	
HV electricity	(y ₁)	$y_1 = 0.0469x^2 - 0.1554x + 6.3384; R^2=1$		
LV electricity	(y ₂)	$y_2 = 0.1694x^2 - 1.9674x + 4.1223; R^2=1$	$y_2 = -0.5325x^2 + 0.7027x + 4.0382; R^2=0.9993$	
Natural gas	(y ₄)	$y_4 = -0.0659x + 0.7906; R^2=0.9992$	$y_4 = -0.0123x + 0.6847; R^2=0.9995$	
Direct CO ₂	(y ₆)	$y_6 = 0.0027x^2 + 0.0264x + 0.1156; R^2=1$	$y_6 = -0.002\ln(x) + 0.1991; R^2=0.9340$	

¹ High voltage electricity only utilized in plasma reactor

² Low voltage electricity utilized for electrically conductive heating and fluids transfer

³ Low voltage electricity utilized for electrically conductive heating in thermocatalytic reactor

Table S-9. Environmental footprint results for all the scenarios

Indicator	Unit	PNGG	PBGG	TNG	TBG
Climate change	kg CO ₂ eq	4.12E+00	7.46E+00	1.86E+00	2.82E+00
Ozone depletion	kg CFC11 eq	5.96E-07	4.28E-07	3.76E-07	3.28E-07
Ionising radiation	kBq U-235 eq	1.77E+00	1.86E+00	3.58E-01	3.85E-01
Photochemical ozone formation	kg NMVOC eq	8.69E-03	9.80E-03	3.14E-03	3.47E-03
Particulate matter	disease inc.	5.43E-08	5.80E-07	1.67E-08	1.68E-07
Human toxicity, non-cancer	CTUh	3.23E-08	4.53E-08	1.08E-08	1.46E-08
Human toxicity, cancer	CTUh	9.82E-10	1.21E-09	4.30E-10	4.98E-10
Acidification	mol H ⁺ eq	1.93E-02	9.10E-02	5.43E-03	2.60E-02
Eutrophication, freshwater	kg P eq	3.14E-03	3.22E-03	6.49E-04	6.74E-04
Eutrophication, marine	kg N eq	3.33E-03	5.68E-03	9.62E-04	1.64E-03
Eutrophication, terrestrial	mol N eq	3.19E-02	3.49E-01	9.55E-03	1.01E-01
Ecotoxicity, freshwater	CTUe	4.14E+01	5.30E+01	1.48E+01	1.82E+01
Land use	Pt	1.51E+01	2.49E+01	3.83E+00	6.65E+00
Water use	m ³ depriv.	1.07E+00	1.10E+00	5.83E-01	5.92E-01
Resource use, fossils	MJ	1.04E+02	8.17E+01	5.15E+01	4.51E+01
Resource use, minerals and metals	kg Sb eq	1.13E-05	1.68E-05	6.38E-06	8.01E-06
Indicator	Unit	PNGHVPV	PNGHVWM	TNGG	TNGPV
Climate change	kg CO ₂ eq	1.82E+00	1.71E+00	2.29E+00	1.79E+00
Ozone depletion	kg CFC11 eq	4.05E-07	3.91E-07	4.00E-07	3.60E-07
Ionising radiation	kBq U-235 eq	3.30E-01	3.35E-01	1.01E+00	6.81E-01
Photochemical ozone formation	kg NMVOC eq	4.04E-03	4.04E-03	5.14E-03	4.19E-03
Particulate matter	disease inc.	2.80E-08	3.57E-08	3.34E-08	3.06E-08
Human toxicity, non-cancer	CTUh	1.49E-08	3.77E-08	2.36E-08	2.11E-08
Human toxicity, cancer	CTUh	8.03E-10	1.28E-09	7.21E-10	6.55E-10
Acidification	mol H ⁺ eq	6.45E-03	7.64E-03	1.15E-02	8.75E-03
Eutrophication, freshwater	kg P eq	6.39E-04	8.71E-04	1.82E-03	1.28E-03
Eutrophication, marine	kg N eq	1.25E-03	1.20E-03	1.95E-03	1.50E-03
Eutrophication, terrestrial	mol N eq	1.26E-02	1.23E-02	1.88E-02	1.46E-02
Ecotoxicity, freshwater	CTUe	2.08E+01	3.52E+01	2.77E+01	2.35E+01
Land use	Pt	1.72E+01	5.69E+00	9.45E+00	2.35E+01
Water use	m ³ depriv.	5.43E-01	5.70E-01	8.51E-01	8.14E-01
Resource use, fossils	MJ	5.42E+01	5.24E+01	6.53E+01	5.41E+01
Resource use, minerals and metals	kg Sb eq	9.58E-06	2.75E-05	1.36E-05	2.77E-05
Indicator	Unit	TNGWM	PNGWM	TNGWM	
Climate change	kg CO ₂ eq	1.73E+00	1.21E+00	6.15E-01	
Ozone depletion	kg CFC11 eq	3.52E-07	3.47E-07	2.57E-07	
Ionising radiation	kBq U-235 eq	6.74E-01	3.58E-02	1.87E-02	
Photochemical ozone formation	kg NMVOC eq	3.98E-03	3.00E-03	1.70E-03	
Particulate matter	disease inc.	2.77E-08	3.06E-08	1.64E-08	
Human toxicity, non-cancer	CTUh	2.30E-08	3.71E-08	2.17E-08	
Human toxicity, cancer	CTUh	7.32E-10	1.28E-09	7.53E-10	
Acidification	mol H ⁺ eq	8.65E-03	5.04E-03	2.96E-03	
Eutrophication, freshwater	kg P eq	1.28E-03	3.87E-04	2.23E-04	
Eutrophication, marine	kg N eq	1.43E-03	7.37E-04	4.18E-04	
Eutrophication, terrestrial	mol N eq	1.40E-02	8.01E-03	4.58E-03	

Ecotoxicity, freshwater	CTUe	2.47E+01	3.25E+01	1.89E+01
Land use	Pt	6.86E+00	3.35E+00	1.74E+00
Water use	m ³ depriv.	7.27E-01	4.58E-01	4.82E-01
Resource use, fossils	MJ	5.32E+01	4.15E+01	2.94E+01
Resource use, minerals and metals	kg Sb eq	1.53E-05	2.91E-05	1.87E-05
