Supplementary material

The need to change the approach to the safe use of herbicides by developing chiral and environmentally friendly formulations: a series of enantioselective (R)- and (S)-phenylethylammonium chloroacetates

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1. Synthesis of QASs

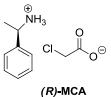
General: (*R*)-(+)- and (*S*)-(-)-1-phenylethylammonium chloro-, dichloro- and trichloroacetates were synthesized in the reaction of the corresponding chloro-, dichloro- and trichloroacetic acids with 10% excess of (R)-(+)- or (S)-(-)-1-phenylethylamine without solvent. In case of chloroacetates, the reactions were performed at room temperature, whereas in case of di- and trichloroacetates, the reactions were carried out in flasks immersed in a cooling ice-water bath. After the reaction was completed, the excess of amine was evaporated with a rotary vacuum evaporator. (*R*)-(+)- and (*S*)-(-)-1-Phenylethylammonium sulfates [(*R*)-SO₄, (*S*)-SO₄] were synthesized in the reaction of H_2SO_4 (aq. 25%) with (R)-(+)- or (S)-(-)-1-phenylethylamine in a 1:2 molar ratio without solvent.¹ The reactions were carried out at low temperature using a cooling ice-water bath, collected solids were dried and subjected to further analysis. Diisopropylammonium sulfate (i-Pr₂NH⁺)₂SO₄ (DIPA-SO₄) was prepared in the reaction performed in chloroform according to the procedure used in the literature for tris(diisopropylammonium) sulfate-hydrogensulfate $[(i-Pr_2NH^+)_3(SO_4^{2-})(HSO_4^{-})]$ except that the amine and concentrated (95%) sulfuric acid were used in a 2:1 molar ratio instead of 3:2 as in the original procedure.² A small amount of the sulfate-hydrogensulfate was detected in the crude reaction mixture (MS: m/z = 500.3039). Crystallization of the crude diisopropylammonium sulfate (DIPA-SO₄) was carried out in methanol instead of water² to avoid hydrolysis of the product to the sulfate-hydrogensulfate. The structure of DIPA-SO₄ was clearly confirmed by X-ray analysis (see below). Similar protocol for the DIPA-SO₄ synthesis, however, without use of any solvent, was performed by Reiß and Engel.³

Optical rotation values were measured in 100 mm cell on Perkin Elmer 241 MC under Na lamp radiation in MeOH. The ¹H NMR and ¹³C NMR spectra were recorded on a Bruker AV 200, AV 500 spectrometers in CDCl₃, MeOD or D₂O with chemical shifts (δ) given in ppm relative to TMS as an internal standard. The low resolution FAB(+) and FAB(-) mass spectra (LSI, Cs⁺, 13 keV) of pure compounds were obtained using a Finnigan Mat 95 spectrometer. High-resolution mass spectrometry (HRMS) measurements were performed using Finnigan Mat 95 and Synapt G2-Si mass spectrometer (Waters) equipped with an ESI source and quadrupole-Time-of-Flight mass analyser. The mass spectrometer was operated in the positive and negative ion detection modes. The measurement was performed with capillary voltage set to 2.7 kV and sampling cone to 20 V. The source temperature was 110 °C. The results of the measurements were processed using the MassLynx 4.1 software (Waters) incorporated with the instrument. Melting points were measured using Boetius apparatus. Elemental analysis was determined with an EuroVector analizator model 301.

¹ J. Cihelka, D. Havlíček, R. Gyepes, I. Němec, Z. Koleva, S-(-)-1-phenyl ethyl ammonium(1+) sulphate and S-(-)-1-phenyl ethyl ammonium(1+) hydrogen phosphate 2.5 hydrate, preparation and characterization of crystallographic, optical and dielectric properties, J. Mol. Struct., 2010, 980 (1-3), 31-38. DOI: 10.1016/j.molstruc.2010.06.033

² G. Sh. Mohammadnezhad, M. M. Amini, H. R. Khavasi, S. W. Ng, Tris(diiso-propyl-ammonium) hydrogensulfate sulfate, *Acta Cryst E*, 2008, 64 (8), 1564. DOI: 10.1107/S160053680802237X

³ G. J. Reiß, J. S. Engel, Bis(diisopropylammonium) sulfate, Acta Cryst E, 2004, 60 (6), 985-987. DOI: 10.1107/S1600536804010645



(*R*)-(+)-1-Phenylethylammonium chloroacetate ((*R*)-MCA) m.p. 78-80 °C; $[\alpha]_{25}^{D}$ = + 6.9 (1, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.55 (d, 3H, J = 6.0 Hz, CH₃), 3.57 (d, 2H, J = 2.0 Hz, CH₂Cl), 4.24 (q, 1H, J = 8.0 Hz, CH), 7.29-7.41 (m, 5H, C₆H₅), 8.04 (br s, 3H, NH₃⁺); ¹³C NMR (126 MHz, CDCl₃) δ (ppm) = 21.08 (s, CH₃), 44.30 (s, CH₂), 51.18 (s, CH), 126.62 (s, 2 x C_{Ar}), 128.67 (s, C_{Ar}), 129.00 (s, 2 x C_{Ar}), 138.86 (s, C_{Ar}), 173.20 (s, COO⁻); HRMS [TOF ES-] for ClCH₂COO⁻, Calcd for C₂H₂ClO₂: 92.9743, Found: 92.9743; MS-FAB(-): m/z (%) = 93, 95 (ClCH₂COO⁻, ³⁵Cl/³⁷Cl, 30, 10); HRMS [TOF ES+] for C₆H₅CH(CH₃)NH₃⁺, Calcd for C₈H₁₂N: 122.0970, Found:

122.0969; MS-FAB (+): m/z (%) = 122 [C₆H₅CH(CH₃)NH₃⁺, 100]; Anal. Calcd for C₁₀H₁₄ClNO₂: C, 55.69; H, 6.54; N, 6.49; Cl, 16.44 Found: C, 55.72; H, 6.60; N, 6.51; Cl, 16.21.

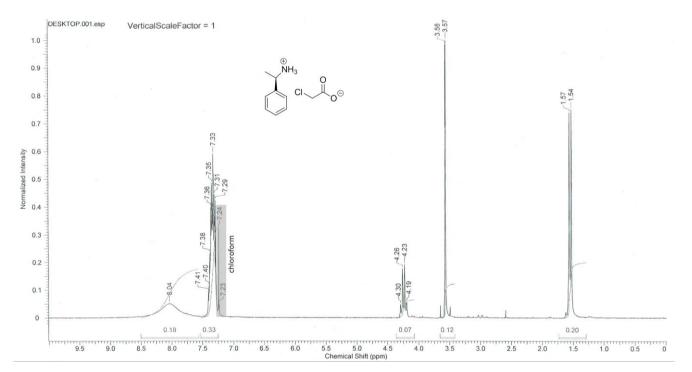


Figure S1 ¹H NMR spectrum of (R)-MCA.

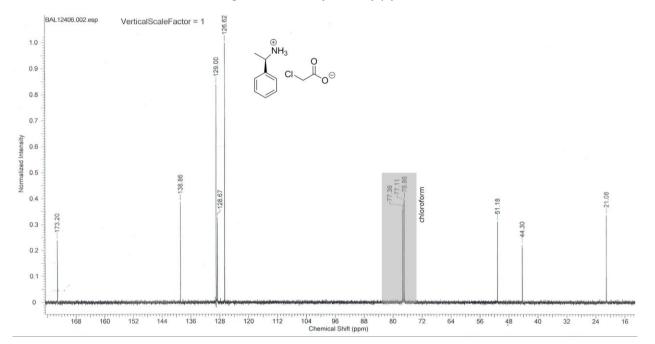
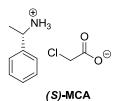
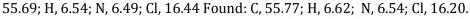
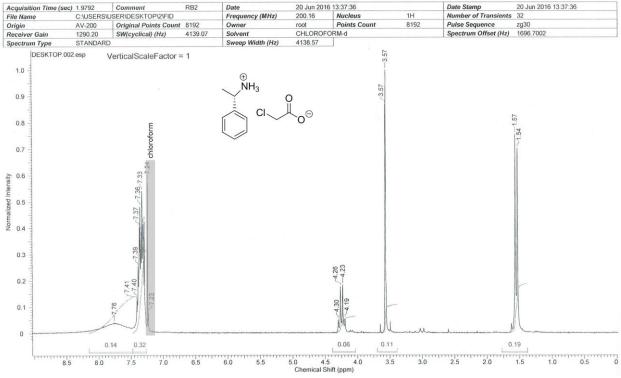


Figure S2 ¹³C NMR spectrum of (R)-MCA.



(*S*)-(-)-1-Phenylethylammonium chloroacetate (*(S*)-MCA) oil; ; $[\alpha]_{25}^D = -6.6$ (1.7, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.55 (d, 3H, J = 6.0 Hz, CH₃), 3.57 (s, 2H, CH₂Cl), 4.24 (q, 1H, J = 8.0 Hz, CH), 7.25-7.41 (m, 5H, C₆H₅), 7.76 (br s, 3H, NH₃⁺); ¹³C NMR (126 MHz, CDCl₃) δ (ppm) = 21.07 (s, CH₃), 44.31 (s, CH₂), 51.18 (s, CH), 126.65 (s, 2 x C_{Ar}), 128.67 (s, C_{Ar}), 129.00 (s, 2 x C_{Ar}), 138.86 (s, C_{Ar}), 173.19 (s, COO⁻); MS-FAB (-): m/z (%) = 93, 95 (ClCH₂COO⁻, ³⁵Cl/³⁷Cl, 20, 7); MS-FAB (+): m/z (%) = 122 [C₆H₅CH(CH₃)NH₃⁺, 100]; Anal. Calcd for C₁₀H₁₄ClNO₂: C, 49: Cl 16 44 Found: C 55 77: H 6.62: N 6.54: Cl 16 20







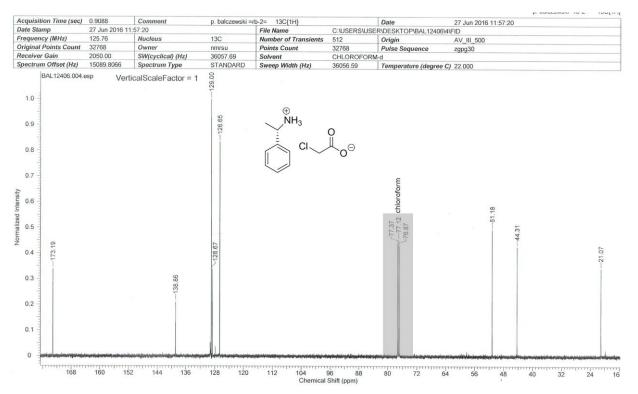
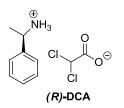


Figure S4 ¹³C NMR spectrum of (S)-MCA.



(*R*)-(+)-1-Phenylethylammonium dichloroacetate ((*R*)-DCA) m.p. 105-106 °C; ; $[\alpha]_{25}^{D} = +3.3$ (1, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.54 (d, 3H, J = 6.0 Hz, CH₃), 4.24 (q, 1H, J = 6.0 Hz, CH), 5.48 (s, 1H, Cl₂CH), 7.25-7.37 (m, 5H, C₆H₅); ¹³C NMR (126 MHz, D₂O) δ (ppm) = 19.57 (s, CH₃), 50.89 (s, CH), 68.40 (s, Cl₂CH), 126.40 (s, 2 x C_{Ar}), 128.97 (s, C_{Ar}), 129.17 (s, C_{Ar}), 129.39 (s, CAr), 138.26 (s, CAr), 170.78 (s, COO⁻); MS-FAB (-): m/z (%) = 93, 95 (ClCH₂COO⁻, ${}^{35}Cl/{}^{37}Cl, 20, 7$; MS-FAB (+): m/z (%) = 122 [C₆H₅CH(CH₃)NH₃+, 100]; Anal. Calcd for C10H13Cl2NO2: C, 48.02; H, 5.24; N, 5.60; Cl, 28.35 Found: C, 47.91; H, 5.47; N, 5.42; Cl, 28.34.

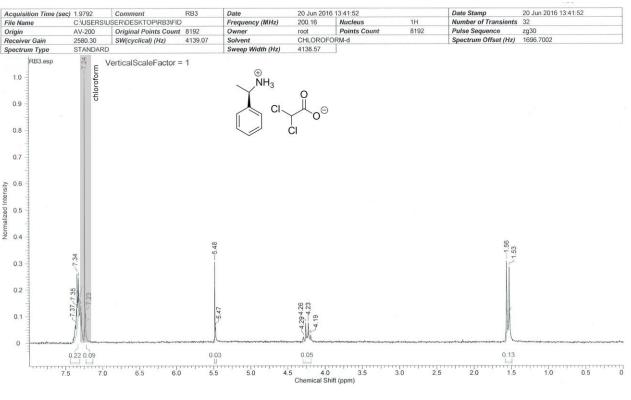
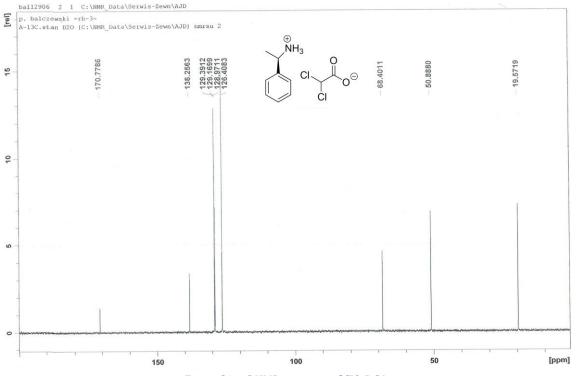
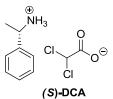


Figure S5 1H NMR spectrum of (R)-DCA.







(*S*)-(-)-1-Phenylethylammonium dichloroacetate ((*S*)-DCA) m.p. 104-105 °C; $[\alpha]_{25}^{D} = -3.2$ (1, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.57 (d, 3H, J = 8.0 Hz, CH₃), 4.26 (q, 1H, J = 6.0 Hz, CH), 5.49 (s, 1H, Cl₂CH), 7.25-7.36 (m, 5H, C₆H₅); ¹³C NMR (126 MHz, CDCl₃) δ (ppm) = 21.38 (s, CH₃), 51.53 (s, CH), 68.84 (s, Cl₂CH), 126.56 (s, 2 x C_{Ar}), 128.65 (s, C_{Ar}), 129.04 (s, 2 x C_{Ar}), 139.28 (s, C_{Ar}), 169.74 (s, COO⁻); MS-FAB (-): m/z (%) = 127, 129, 131 (Cl₂CHCOO⁻, ³⁵Cl/³⁷Cl, 100, 68, 10); 255, 257, 259, 261 [2(Cl₂CH₂COO⁻) + 1, ³⁵Cl/³⁷Cl, 38, 50, 29, 6]; MS-FAB

(+): m/z (%) = 122 [C₆H₅CH(CH₃)NH₃⁺, 100]; Anal. Calcd for C₁₀H₁₃Cl₂NO₂: C, 48.02; H, 5.24; N, 5.60; Cl, 28.35 Found: C, 48.17; H, 5.31; N, 5.63; Cl, 28.38.

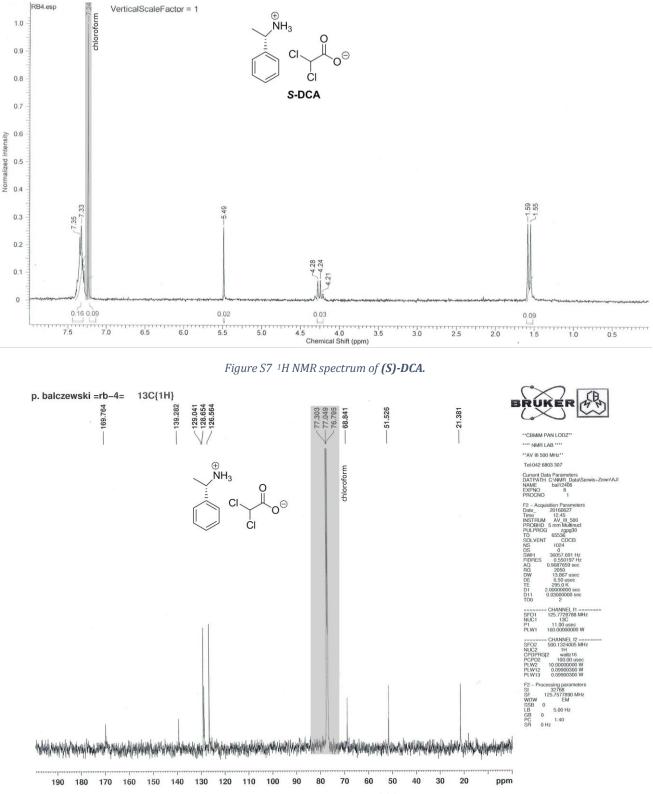
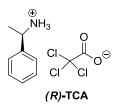


Figure S8 ¹³C NMR spectrum of (S)-DCA.



(*R*)-(+)-1-Phenylethylammonium trichloroacetate ((*R*)-TCA) m.p. 82-84 °C; ; $[\alpha]_{25}^{D}$ = +11.4 (1, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.55 (d, 3H, J = 6.0 Hz, CH₃), 4.24 (q, 1H, J = 6.0 Hz, CH), 5.59 (br s, 3H, NH₃⁺), 7.25-7.39 (m, 5H, C₆H₅); ¹³C NMR (126 MHz, CDCl₃) δ (ppm) = 20.97 (s, CH₃), 51.78 (s, CH), 95.74 (s, CCl₃), 126.88 (s, 2 x C_{Ar}), 128.77 (s, C_{Ar}), 129.03 (s, 2 x C_{Ar}), 138.83 (s, C_{Ar}), 165.78 (s, COO^{-}); MS-FAB (-): m/z (%) = 161, 163, 165, 167 (Cl₃CCOO⁻, ³⁵Cl/³⁷Cl, 100, 95, 30, 7); 323, 325, 327, 329, 331, 333 [2(Cl₃CCOO⁻) + 1, ³⁵Cl/³⁷Cl, 20, 42, 30, 17, 6, 3]; MS-FAB (+): m/z (%) = 122 [C₆H₅CH(CH₃)NH₃⁺, 100]; Anal. Calcd for C₁₀H₁₂Cl₃NO₂: C, 42.21; H, 4.25; N,

4.92; Found: C, 42.35; H, 4.31; N, 4.96.

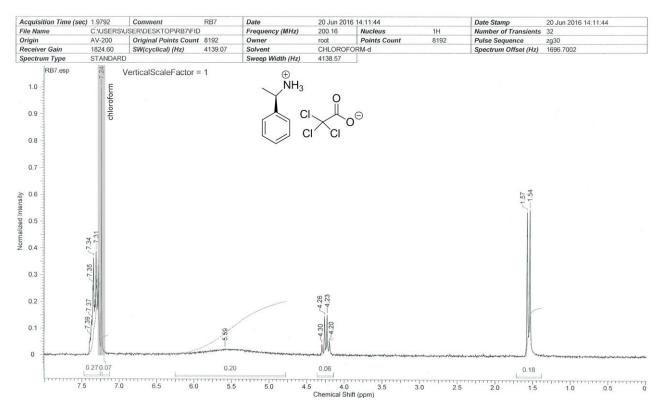
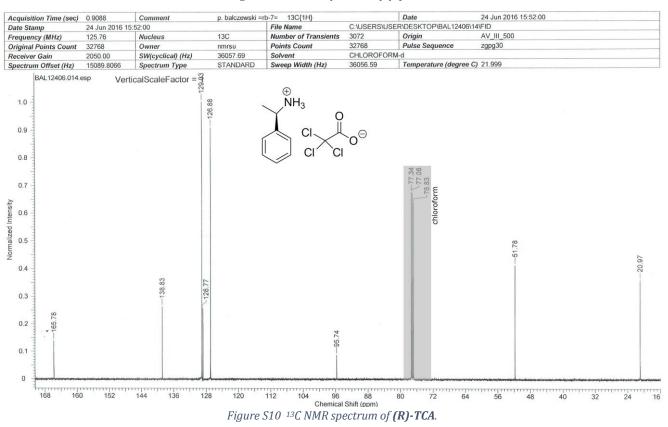
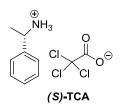


Figure S9 ¹H NMR spectrum of (R)-TCA.



S7



(*S*)-(-)-1-Phenylethylammonium trichloroacetate ((*S*)-TCA) m.p. 81-83 °C; ; $[\alpha]_{25}^{D}$ = - 11.9 (1, MeOH); ¹H NMR (200 MHz, CDCl₃) δ (ppm) = 1.55 (d, 3H, J = 6.0 Hz, CH₃), 4.26 (q, 1H, J = 8.0 Hz, CH), 7.26-7.37 (m, 5H, C₆H₅); ¹³C NMR (126 MHz, CDCl₃) δ (ppm) = 21.14 (s, CH₃), 51.20 (s, CH), 95.73 (s, CCl₃), 126.80 (s, 2 x C_{Ar}), 128.73 (s, C_{Ar}), 129.03 (s, 2 x C_{Ar}), 139.05 (s, C_{Ar}), 165.85 (s, COO⁻); MS-FAB (-): m/z (%) = 161, 163, 165, 167 (Cl₃CCOO⁻, ³⁵Cl/³⁷Cl, 100, 96, 31, 4); 323, 325, 327, 329, 331, 333 [2(Cl₃CCOO⁻) + 1, ³⁵Cl/³⁷Cl, 24, 48, 35, 17, 4, 1]; MS-FAB (+):

m/z (%) = 122 [C₆H₅CH(CH₃)NH₃⁺, 100]; Anal. Calcd for C₁₀H₁₂Cl₃NO₂: C, 42.21; H, 4.25; N, 4.92; Found: C, 42.39; H, 4.42; N, 5.01.

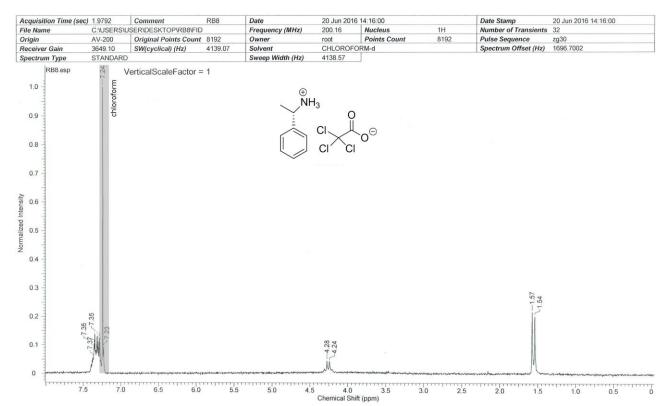


Figure S11 ¹H NMR spectrum of (S)-TCA.

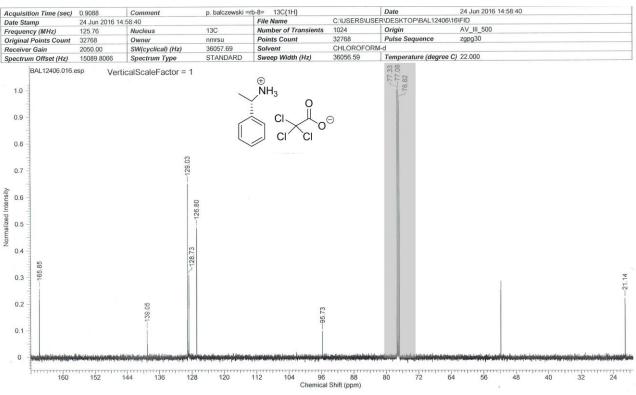
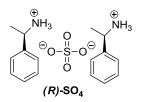
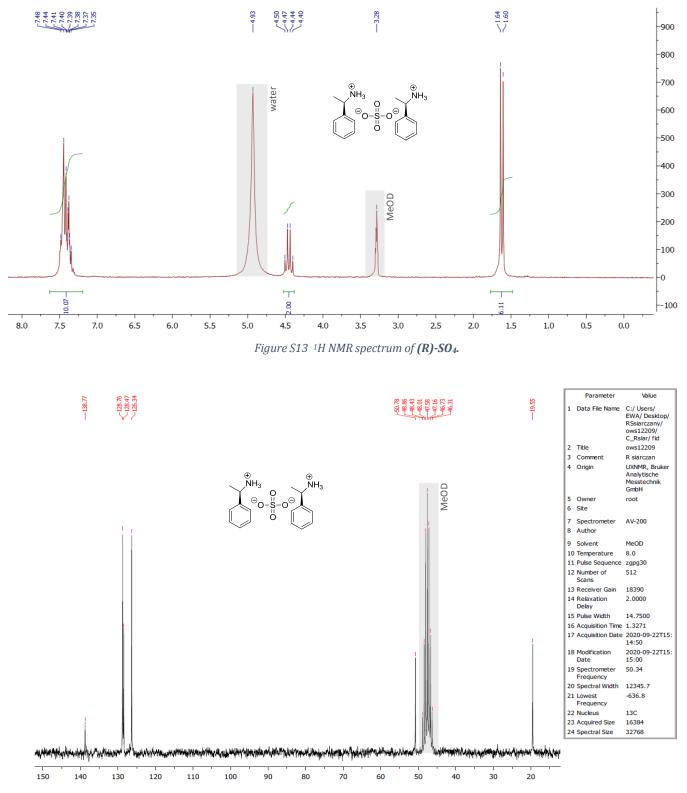


Figure S12 ¹³C NMR spectrum of (S)-TCA.

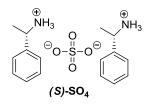


(*R*)-(+)-1-Phenylethylammonium sulfate ((*R*)-SO₄) m.p. 242-248 °C; ; $[\alpha]_{25}^{D}$ = + 4.5 (1, MeOH); ¹H NMR (200 MHz, MeOD) δ (ppm) = 1.62 (d, 6H, J = 8.0 Hz, 2 x CH₃), 4.46 (q, 2H, J = 6.0 Hz, 2 x CH), 7.35-7.48 (m, 10H, 2 x C₆H₅); ¹³C NMR (50 MHz, MeOD) δ (ppm) = 19.55 (s, 2 x CH₃), 50.78 (s, 2 x CH), 126.34 (s, 2 x C_Ar), 128.47 (s, 4 x C_Ar), 128.76 (s, 4 x C_Ar), 138.77 (s, 2 x C_Ar); HRMS [TOF ES+] for [M+1, 100], Calcd for C₁₆H₂₅N₂O₄S: 341.1535, Found: 341.1542; Anal. Calcd for C₁₆H₂₄N₂O₄S: C, 56.45; H, 7.11; N, 8.23; S, 9.42; Found: C, 56.72;

H, 7.12; N, 8.21; S, 9.48.

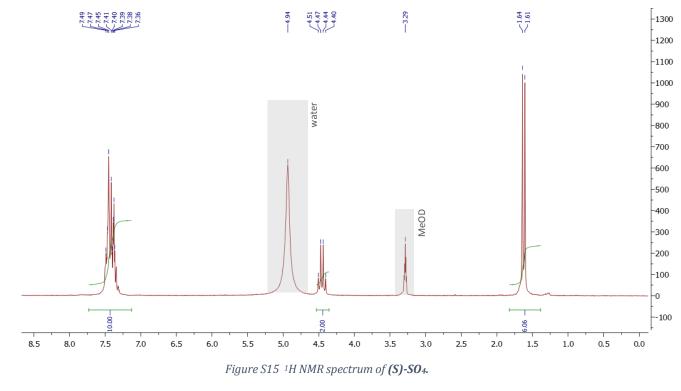


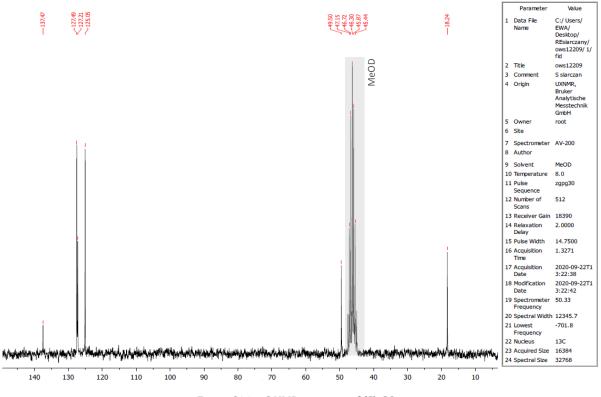




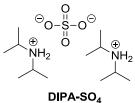
(*S*)-(-)-1-Phenylethylammonium sulfate ((*S*)-SO₄) m.p. 241-248 °C; ; $[\alpha]_{25}^{D}$ = - 4.8 (1, MeOH); ¹H NMR (200 MHz, MeOD) δ (ppm) = 1.62 (d, 6H, J = 6.0 Hz, 2 x CH₃), 4.45 (q, 2H, J = 8.0 Hz, 2 x CH), 7.36-7.49 (m, 10H, 2 x C₆H₅); ¹³C NMR (50 MHz, MeOD) δ (ppm) = 18.24 (s, 2 x CH₃), 49.50 (s, 2 x CH), 125.05 (s, 2 x C_{Ar}), 127.21 (s, 4 x C_{Ar}), 127.49 (s, 4 x C_{Ar}), 137.47 (s, 2 x C_{Ar}); HRMS [TOF ES+] for [M+1, 100], Calcd for C₁₆H₂₅N₂O₄S: 341.1535, Found: 341.1543; Anal. Calcd for C₁₆H₂₄N₂O₄S: C, 56.45; H, 7.11; N, 8.23; S, 9.42, Found: C,

56.69; H, 7.10; N, 8.24; S, 9.50.









Diisopropylammonium sulfate (DIPA-SO₄) m.p. 182-185 °C; ¹H NMR (200 MHz, MeOD) δ (ppm) = 1.32 (d, 24H, J = 8.0 Hz, 4 x CH₃), 4.96 (s, 4H, 4 x CH); ¹³C NMR (50 MHz, MeOD) δ (ppm) = 18.00 (s, 4 x CH₃), 46.79 (s, 4 x CH); HRMS [TOF ES+] for [M+1, 100]: Calcd: 301.2161; Found: 301.2161; HRMS [TOF ES+] for 102 [C₆H₁₆N⁺, 100], Calcd: 102.1283, Found: 102.1279; Anal. Calcd for C₁₂H₃₂N₂O₄S: C, 47.97; H, 10.73; N, 9.32; S, 10.67; Found: C, 47.91; H, 10.53; N, 9.35; S, 11.02.

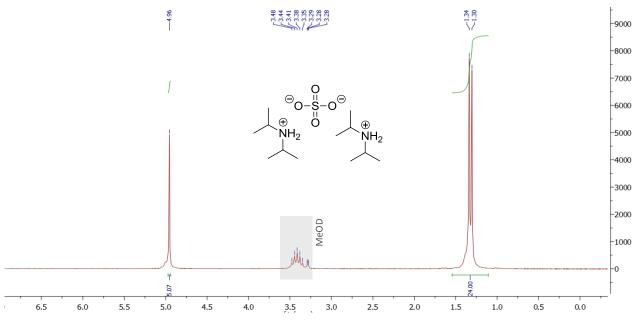
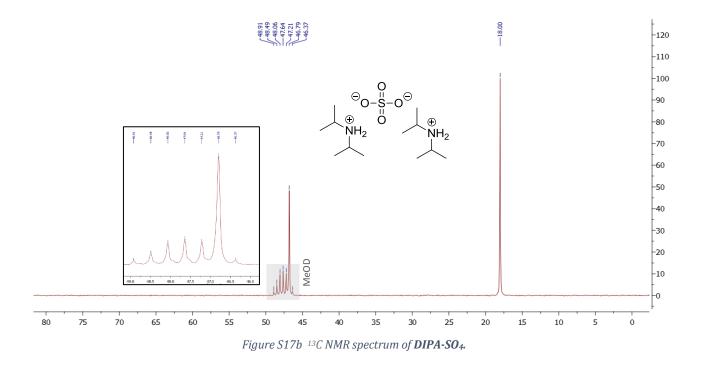


Figure S17a ¹H NMR spectrum of **DIPA-SO**₄.



2. X-ray crystal structure determination of diisopropylammonium sulfate (DIPA-SO₄)

Crystallographic data for DIPA-SO4: $2C_6H_{16}N^+ \cdot SO4^{2-}$, M=300.46, monoclinic, P2₁/n, a = 8.7824(2) Å, b = 8.82350(10) Å, c = 22.0837(4) Å, β = 95.202(2) °, V = 1704.25(5) Å³, Z = 4, Dc = 1.171 g/cm³, F(000)=664, crystal size 0.55x0.48x0.36 mm. Diffraction data were collected at 293(2) K, using an Oxford Diffraction Xcalibur^{TM3} diffractometer (graphite-monochromated CuKα radiation, CCD detector). The structure was solved by direct methods and refined by full-matrix least-squares on F² with SHELXS-97 and SHELXL-97, respectively. The disorder on oxygen atoms of SO4²⁻ anion was refined over two positions with the occupancy factor for the major component of 0.854(8). All non-hydrogen atoms were refined anisotropically. H atoms of methyl groups were positioned geometrically and constrained to ride on their parent atoms, with C-H distances of 0.96 Å and with U_{iso} values of 1.5Ueq(C). All other hydrogen atoms were located in a difference Fourier maps and refined isotropically with N-H and C-H distances in the ranges of 0.79(2)-0.92(2) Å and 0.91(2)-0.98(2) Å, respectively. R1=0.0349 and wR2=0.0960 for 2879 reflections with F² > 2σ(F²) and 226 parameters; R1=0.0366 and wR2=0.0979 for 3063 unique reflections, S=1.035. Further details on the crystal structure investigation have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC: 2073361.

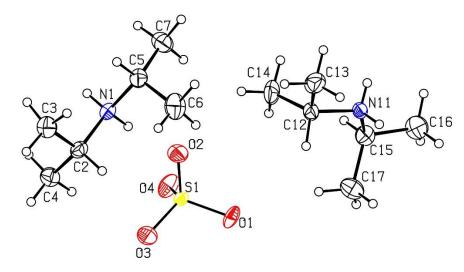


Figure. S18. A view of the asymmetric part of unit cell of DIPA-SO₄ with the atom numbering scheme and displacement ellipsoids drawn at the 30 % probability level. The minor component on oxygen atoms of SO₄²⁻ anion has been omitted for clarity.

3. Phytotoxicity studies

Table S1. Variations in the yield of fresh weight (FW, g/pot), dry weight (DW, g/g f.m.) and germination for spring barley as the effect of increasing concentration of quaternary ammonium salts (QASs) in soil (in mg/kg of soil d.w.). Data are means \pm SD from 3 independent experiments. Values denoted with the same letters do not differ statistically at the level of p<0.05 (Tukey's HSD test).

IL		Concentration of QAS [mg/kg of soil d.w.]								
		0	1	10	100	200	400	800	1000	LSD0.95
Spring barley										
(R)-MCA	FW	4.495±0.047 ^a	4.258±0.135 ab	4.368±0.461 ab	4.781±0.177 ^a	4.743±0.151 ^a	4.811±0.097 ^a	3.800±0.139 ^b	2.327±0.144 °	0.212
	DW	0.0790±0.0039 ^{bc}	0.0789±0.0033 ^{bc}	0.0780±0.0020°	0.0773±0.0031°	0.0759±0.0013 °	0.0755±0.0029 °	0.0842±0.0032 ^b	0.1041±0.0035 ^a	0.0030
	sprouts	20 ^a	20 ^a	18±1 ^b	20 ^a	20 ^a	20 ^a	19±1 ^{ab}	19±1 ^{ab}	1
	FW	4.495±0.47 ^a	4.398±0.105 ^a	4.364±0.260 ^a	4.064±0.237 ^a	4.570±0.420 ^a	4.842±0.213 ^a	4.534±0.477 ^a	3.210±0.296 ^b	0.230
(S)-MCA	DW	0.790±0.0039 ^b	0.0771±0.0020 ^b	0.0786±0.0021 ^b	0.0766±0.0020 b	0.0735±0.0090 ^b	0.0752±0.0024 ^b	0.0783±0.0036 ^b	0.0895±0.0026 ^a	0.0041
	sprouts	20 ^a	20±1 a	19±2 ^{ab}	18±1 ^b	19±1 ^{ab}	20 ^a	20 ^a	19±1 ^{ab}	1
	FW	5.083±0.086 ^a	4.941±0.169 ^a	4.956±0.256 ^a	5.064±0.390 ^a	5.011±0.127 ^a	4.499±0.092 ^a	3.406±0.218 ^b	3.277±0.378 ^b	0.250
(R)-DCA	DW	0.757±0.0041 °	0.0735±0.0030 °	0.0715±0.0028 °	0.0756±0.0010 °	0.0711±0.0028 °	0.0751±0.0030 °	0.0863±0.0019 ^b	0.1069±0.0018 ^a	0.0027
	sprouts	19±1 ^a	19±1 ^a	19±1 ^a	20 ^a	19±1 ^a	19±1 ^a	19±1 ^a	20 ^a	1
	FW	5.083±0.089 ^a	5.327±0.245 ^a	4.973±0.481 ^a	5.620±0.310 ^a	-	-	-	5.539±0.477 ^a	0.476
(S)-DCA	DW	0.0757±0.0041 a	0.0739±0.0026 ^a	0.0746±0.0019 ^a	0.0733±0.0033 ^a	-	-	-	0.0736±0.0039 ^a	0.0042
	sprouts	19±1 ^{ab}	20 ^a	19±1 ^b	20 ^a	-	-	-	19±1 ^{ab}	1
	FW	3.815±0.140 ^{ab}	3.514±0.235 abc	3.957±0.047 ^a	3.686±0.091 ab	3.763±0.052 ^{ab}	3.067±0.059 ^{abc}	2.384±0.182 bc	2.193±0.233 °	0.154
(R)-SO ₄	DW	0.0746±0.0024 °	0.0764±0.0014 °	0.0744±0.0043 °	0.0809±0.0028 °	0.0765±0.0020 °	0.0783±0.0020 °	0.0891±0.0010 °	0.1029±0.0026 ^a	0.0025
	sprouts	19±1 ^a	19±1 ^a	20 ^a	18 ^a	18±2 ^a	19±2 ^a	18±1 ^a	19±2 ^a	1
	FW	3.815±0.140 ^a	3.555±0.015 ^a	3.770±0.057 ^a	3.532±0.091 ^a	-	-	-	3.573±0.131 ^a	0.133
(S)-SO ₄	DW	0.746±0.0024 °	0.0769±0.0016 ^{bc}	0.0780±0.0010 ^{abc}	0.0810±0.0021 ^a	-	-	-	0.0808 ± 0.0015^{ab}	0.0024
	sprouts	19±1 ^a	17±1 ^a	19 ^a	19±1 ^a	-	-	-	18 ^a	1
	FW	3.645±0.050 ^a	3.855±0.168 ^a	4.055±0.079 ^a	4.177±0.100 ^a	-	-	-	3.724±0.024 ^a	0.132
DIPA-SO ₄	DW	0.0753±0.0016 ^a	0.0712±0.0013 ^a	0.0730±0.0014 ^a	0.0770±0.0047 ^a	-	-	-	0.0738±0.0010 ^a	0.0032
	sprouts	19±1 ^a	18 ^a	19 ^a	19 ^a	-	-	-	19±1 ^a	1
				Co	ncentration of QA	S [mg/kg of soil d.v	w.]			
		0	1	10	20	40	80	100	1000	
	FW	5.155±0.039 ^a	4.968±0.088 ^a	4.697±0.070 ^a	3.973±0.089 ^b	1.724±0.467 °	0.348±0.002 ^d	0.175±0.042 ^d	-	0.207
(R)-TCA	DW	0.0724±0.0024 ^d	0.0743±0.0024 ^d	0.0771±0.0022 ^d	0.0842±0.0021 °	0.1047±0.0059 ^b	0.1285±0.0055 ^a	0.1357±0.0008 ^a	-	0.0037
	sprouts	19±1 ^a	20±1 a	20±1 a	19±1 ^a	20 ^a	20 ^a	20 ^a	19±1 ^a	0
	FW	5.155±0.039 ^a	5.191±0.180 ^a	5.321±0.231 ^a	3.633±0.197 ^b	2.595±0.523 °	1.495±0.083 ^d	0.645±0.044 ^d	-	0.269
(<i>S</i>)-TCA	DW	0.0724±0.0024 ^d	0.0725±0.0026 ^d	0.0781±0.0031 ^d	0.0870±0.0024 °	0.0981±0.0041 b	0.1165±0.0037 ^a	0.1192±0.0037 ^a	-	0.0036
	sprouts	19±1 ^a	20±1 a	20±1 a	20±1 a	20±1 ^a	20±1 a	20 a	19±1 ^a	1

Table S2. Variations in the yield of fresh weight (FW, g/pot), dry weight (DW, g/g f.m.) and germination for common radish as the effect of increasing concentration of quaternary ammonium salts (QASs) in soil (in mg/kg of soil d.w.). Data are means \pm SD from 3 independent experiments. Values denoted with the same letters do not differ statistically at the level of p<0.05 (Tukey's HSD test).

IL		Concentration of QAS [mg/kg of soil d.w.]								
		0	1	10	100	200	400	800	1000	LSD0.95
Common radish										
(R)-MCA	FW	5.134±0.857 ^a	5.065±0.131 ^a	4.716±0.141 ^a	3.736±0.189 ^a	5.107±0.697 ^a	4.216±0.353 a	0.490±0.514 ^b	-	0.510
	DW	0.0530±0.0022 °	0.0522±0.0011 °	0.0530±0.0026 °	0.0534±0.0011 °	0.0547±0.0022 °	0.0642±0.0011 ^b	0.1200±0.0011 ^a	-	0.0019
	sprouts	18 ^a	17±1 a	17±1 ^a	18±1 ^a	19±1 ^a	19±1 ^a	6±4 ^b	0±1 °	2
	FW	5.052±0.042 ^a	5.447±0.395 ^a	4.618±0.188 ^{ab}	4.806±0.138 ab	5.104±0.385 ^a	4.198±0.335 ab	3.506±0.295 b	-	0.314
(S)-MCA	DW	0.0530±0.0022 ^b	0.0506±0.0029 ^b	0.0538±0.0022 ^b	0.0547±0.0014 ^b	0.0554±0.0035 ^b	0.0588±0.0039 ^b	0.0791±0.0121 ^a	-	0.0057
	sprouts	18 ^a	19±2 a	17±1 ^a	18±2 a	18±1 ^a	18 ^a	17±2 a	0±1 b	1
	FW	5.824±0.642 ^a	6.109±0.612 ^a	5.542±0.455 ^a	5.457±0.310 ^a	5.927±0.402 ^a	4.456±0.392 ab	3.275±0.193 bc	1.808±0.166 °	0.442
(R)-DCA	DW	0.0540±0.0022 °	0.0535±0.0026 °	0.0512±0.0025 °	0.0560±0.0009bc	0.0526±0.0031 °	0.0581±0.0013 bc	0.0646±0.0050 ^b	0.1172±0.0165 ^a	0.0064
	sprouts	17±2 ^a	18±2 ^a	16±1 ^a	17±1 ^a	19±1 ^a	17±1 ^a	16±1 ^a	17±2 ^a	1
	FW	5.824±0.642 ^a	5.371±0.263 ^{ab}	5.264±0.001 ^{ab}	5.412±0.0053 ^{ab}	5.542±0.111 ^{ab}	5.883±0.182 ^a	5.244±0.238 ^{ab}	4.523±0.186 b	0.091
(S)-DCA	DW	0.0540±0.0022 b	0.0569±0.0015 ^{ab}	0.0537±0.0020 b	0.0550±0.0017 ^a	0.0552±0.0044 ^a	0.0544±0.0024 ^b	0.0569±0.0035 ^{ab}	0.0599±0.0022 ^a	0.0027
	sprouts	18±1 ^{ab}	17±1 ^{ab}	16±1 ^b	17±1 ^{ab}	19±1 ^a	19±1 ab	18±1 ^{ab}	18±1 ^{ab}	1
	FW	6.354±0.334 ^{ab}	6.008±0.121 ^{ab}	6.569±0.111 ^{ab}	6.643±0.116 ^a	5.731±0.138 abc	5.449±0.177 bc	4.757±0.347 ^{cd}	4.213±0.122 ^d	0.212
(<i>R</i>)-SO ₄	DW	0.0523±0.0018 b	0.0559±0.0034 ^b	0.0557±0.0009 ^b	0.0567±0.0014 ^b	0.0560±0.0020 b	0.0588±0.0014 b	0.0613±0.0037 ^b	0.0922±0.0086 ^a	0.0037
	sprouts	19±2 ^a	19±1 ^a	19±1 ^a	20 ^a	19±2 ^a	19±1 ^a	17±1 ^a	20 ^a	1
	FW	6.354±0.334 ^{ab}	7.042±0.137 ^a	6.727±0.212 ^{ab}	5.808±0.039 ^{ab}	5.187±0.015 ^{ab}	4.947±0.025 ab	4.767±0.0003 ^{ab}	4.336±0.105 b	0.158
(S)-SO ₄	DW	0.0523±0.0018 °	0.0549±0.0005 °	0.0532±0.0030 °	0.0573±0.0009bc	0.0586 ± 0.0009^{abc}	0.0587±0.0021 ^{abc}	0.0634 ± 0.004 ab	0.655±0.0039 ^a	0.0025
	sprouts	19±1 ^a	19±1 ^a	19±1 ^a	19±1 ^a	18 ^a	18±1 ^a	18±1 ^a	19±1 ^a	1
	FW	6.507±0.155 ^{abc}	6.702±0.144 ^a	6.595±0.195 ^{ab}	6.096±0.085 abcd	5.551±0.059 bcde	5.483±0.107 ^{cde}	5.369±0.029 de	4.598±0.022 °	0.150
DIPA-SO ₄	DW	0.0537±0.0012 ^{ab}	0.0524±0.0017 ^b	0.0500±0.0021 b	0.0541±0.0013 ^{ab}	0.0533±0.0017 ^b	0.0542±0.0016 ^{ab}	0.0583±0.0024 ^a	0.583±0.0017 ^a	0.0018
	sprouts	19±1 ^a	19±1 ^a	19±1 ^a	19 ^a	18±1 ^a	18±1 ^a	18±1 ^a	18±1 ^a	1
	Concentration of QAS [mg/kg of soil d.w.]									
		0	1	10	20	40	80	100	1000	
	FW	8.430±0.441 ^{ab}	8.512±0.640 ^a	8.481±0.281 ^{ab}	7.991±0.578 ^{abc}	7.403±0.253 bc	7.081±0.284 °	7.031±0.356 °	3.770±0.180 ^d	0.420
(<i>R</i>)-TCA	DW	0.0503±0.0010 ^{cd}	0.0491±0.0015 ^d	0.0524±0.0017 ^{cd}	0.0530±0.0018 °	0.0575±0.0024 ^b	0.0569±0.0023 ^b	0.0593±0.0025 b	0.0810±0.0028 ^a	0.0021
	sprouts	20 ^a	20 ^a	20 ^a	20±1 ^a	20 ^a	20 ^a	20 ^a	20±1 a	0
(<i>S</i>)-TCA	FW	8.430±0.441 a	7.800±0.222 ^{ab}	8.387±0.435 ^a	8.359±0.100 ^a	7.725±0.520 ^{ab}	6.832±0.15 ^b	6.830±0.397 ^b	2.581±0.532 °	0.369
	DW	0.0503±0.0010 ^b	0.0516±0.0013 ^b	0.0498±0.0016 ^b	0.0523±0.0014 ^b	0.0538±0.0015 ^b	0.0557±0.0015 ^b	0.0584±0.0005 ^b	01298±0.0281 ^a	0.0101
	sprouts	20 ^a	20 ^a	20±1 ^a	20 ^a	20±1 ^a	19±1 ^a	20 ^a	20±1 ^a	1

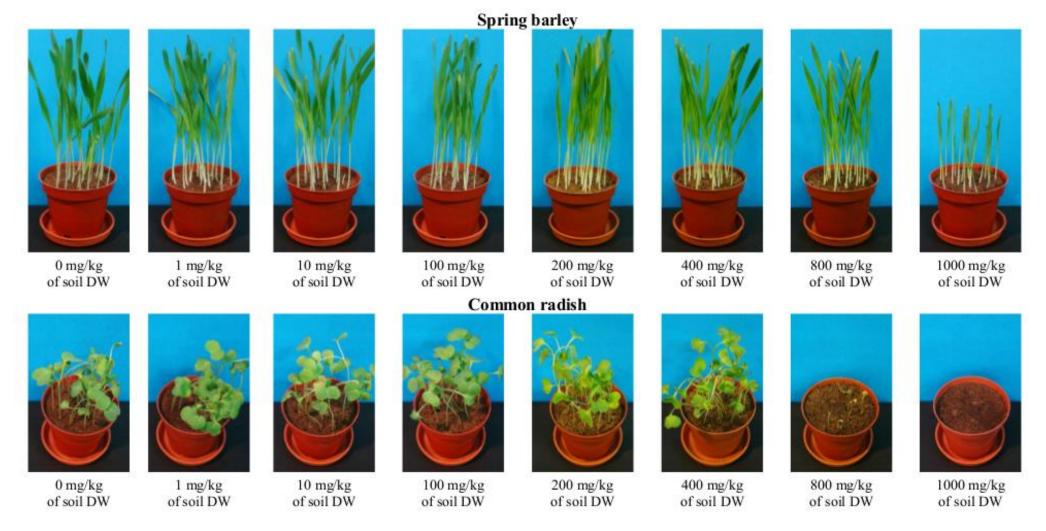


Figure S19. Digital photographs of plants seedlings in soil treated with aqueous solutions of **(R)-MCA**.

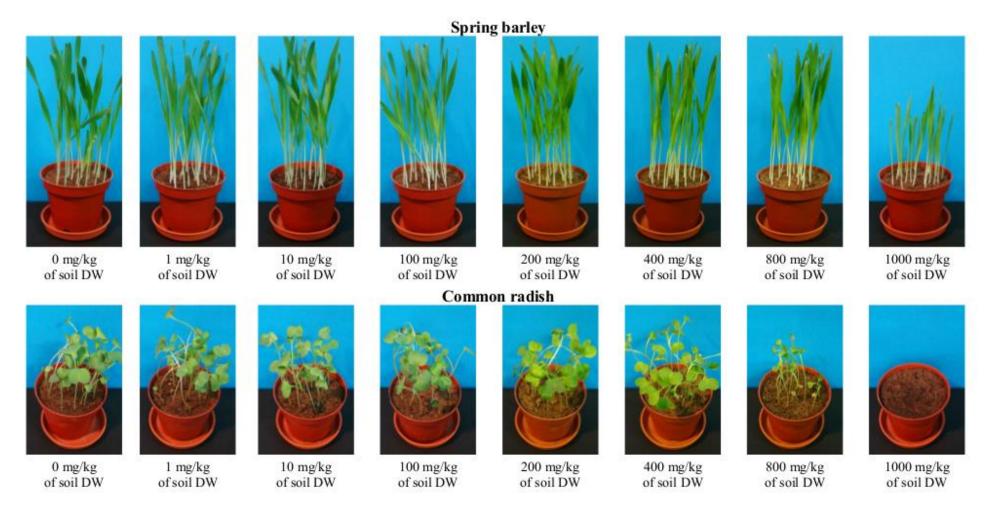


Figure S20. Digital photographs of plants seedlings in soil treated with aqueous solutions of (S)-MCA.

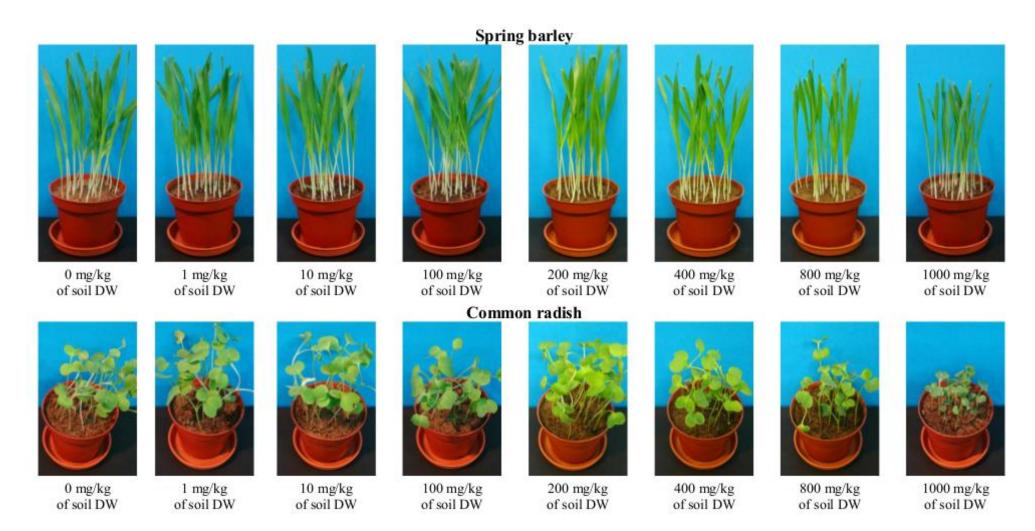
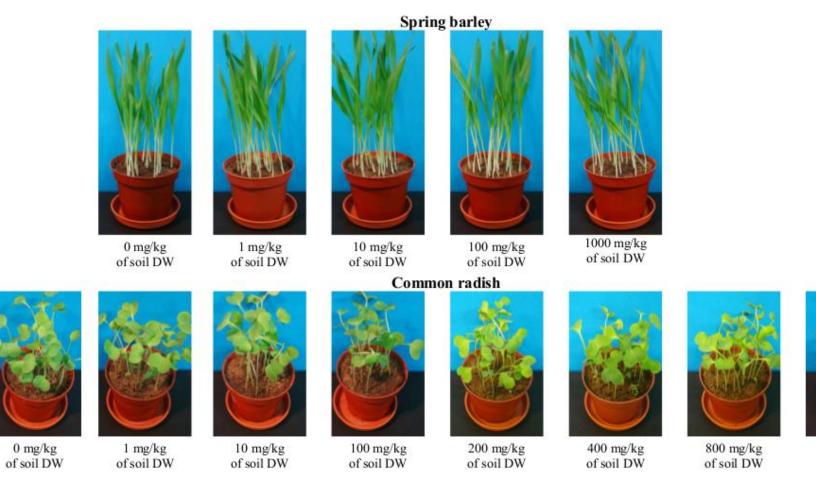


Figure S21. Digital photographs of plants seedlings in soil treated with aqueous solutions of **(R)-DCA**.



1000 mg/kg of soil DW

Figure S22. Digital photographs of plants seedlings in soil treated with aqueous solutions of (S)-DCA.

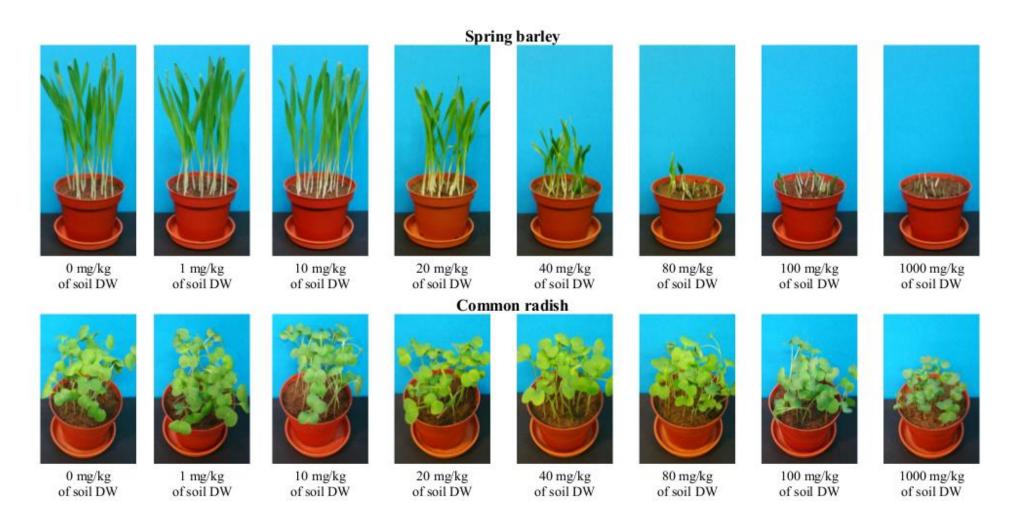


Figure S23. Digital photographs of plants seedlings in soil treated with aqueous solutions of **(R)-TCA**.

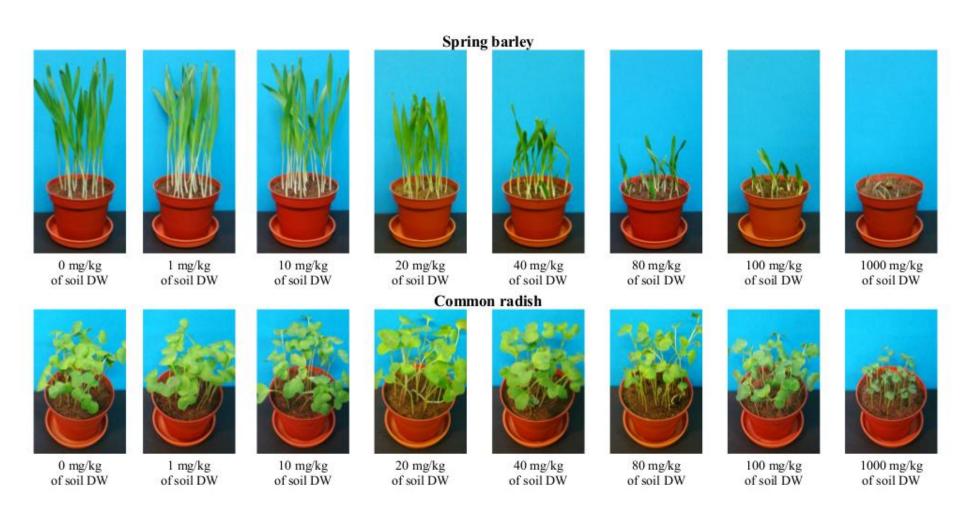
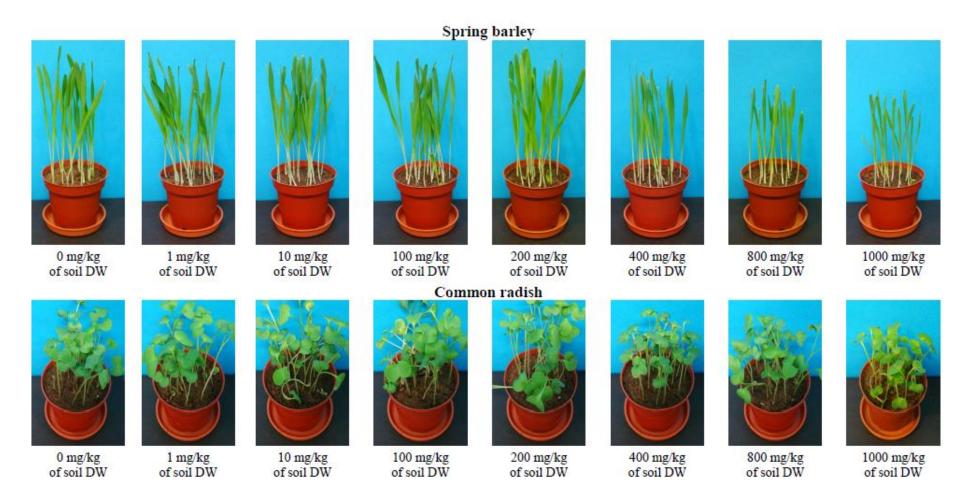


Figure S24. Digital photographs of plants seedlings in soil treated with aqueous solutions of **(S)-TCA**.



*Figure S25. Digital photographs of plants seedlings in soil treated with aqueous solutions of (R)-SO*₄*.*

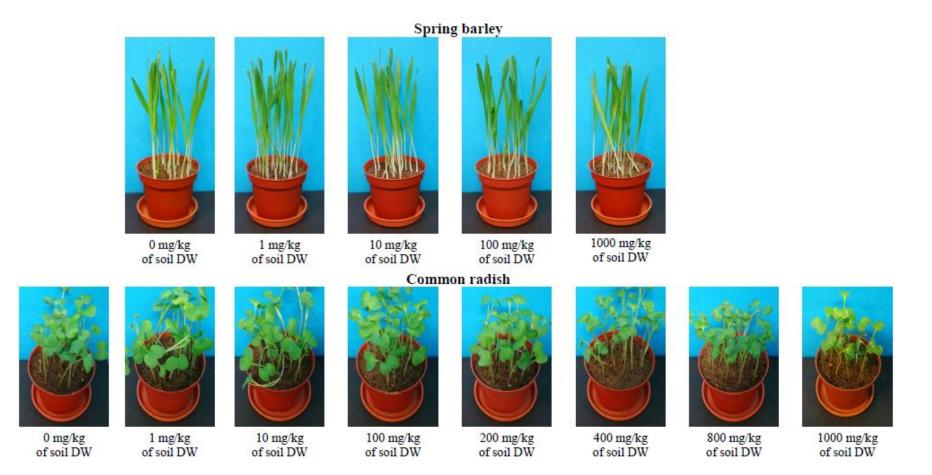


Figure S26. Digital photographs of plants seedlings in soil treated with aqueous solutions of (S)-SO₄.

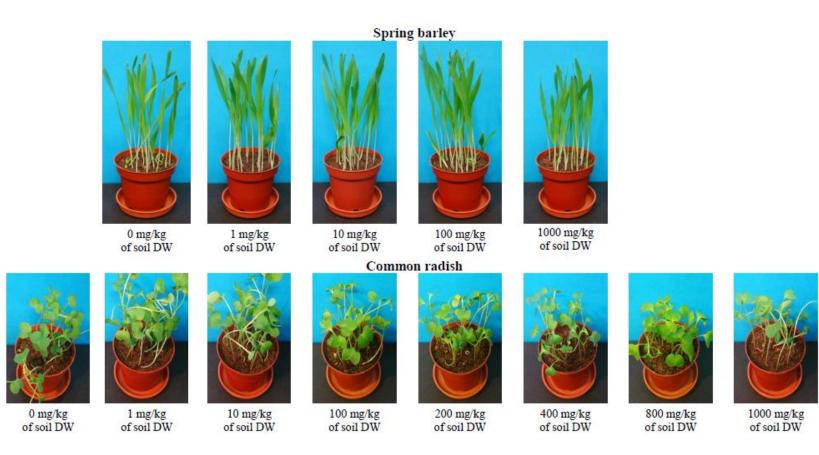


Figure S27. Digital photographs of plants seedlings in soil treated with aqueous solutions of **DIPA-SO**₄.

4. Herbicidal activity

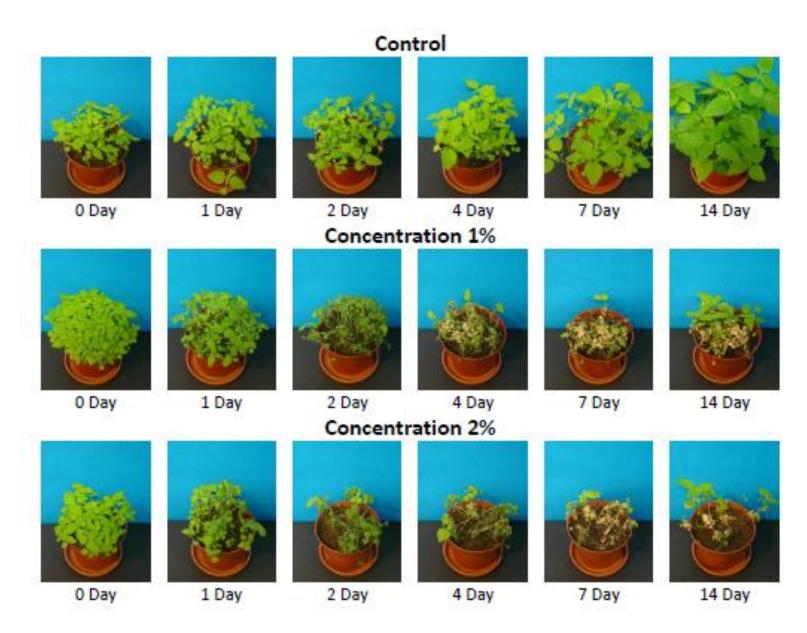


Figure S28. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **(R)-MCA**.

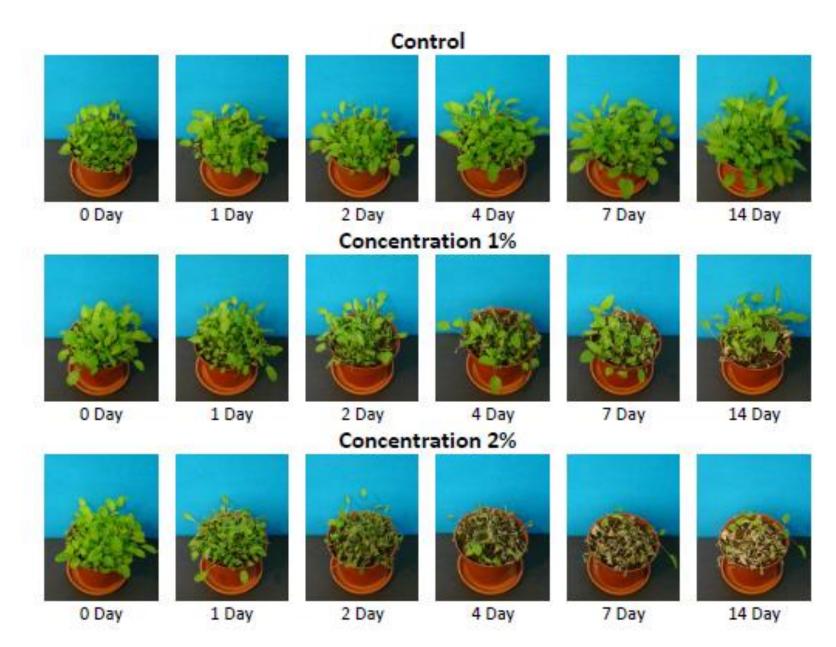


Figure S29. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of **(R)-MCA**.

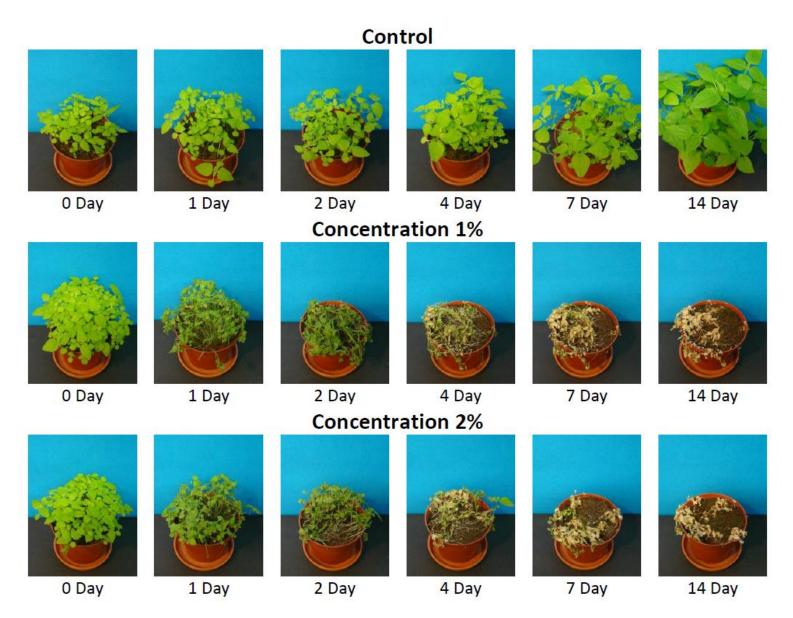


Figure S30. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of (S)-MCA.

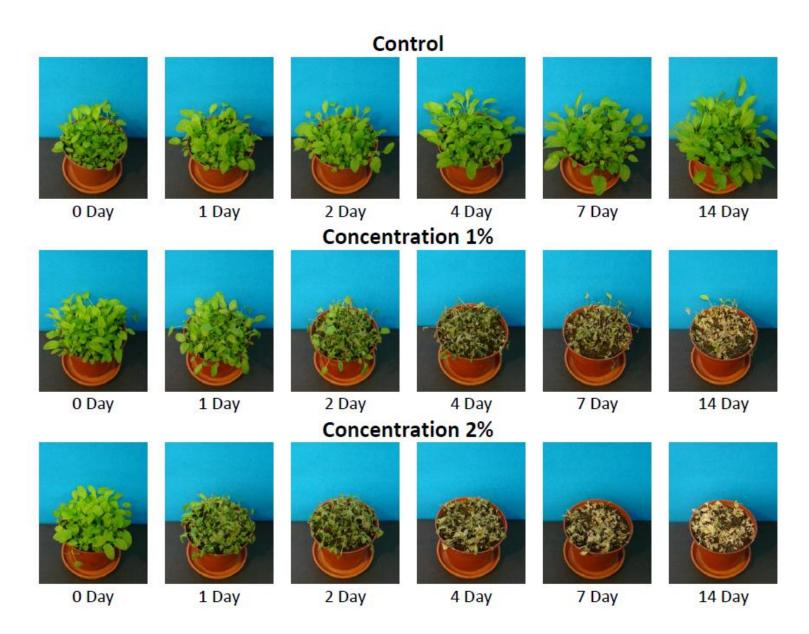


Figure S31. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of (S)-MCA.

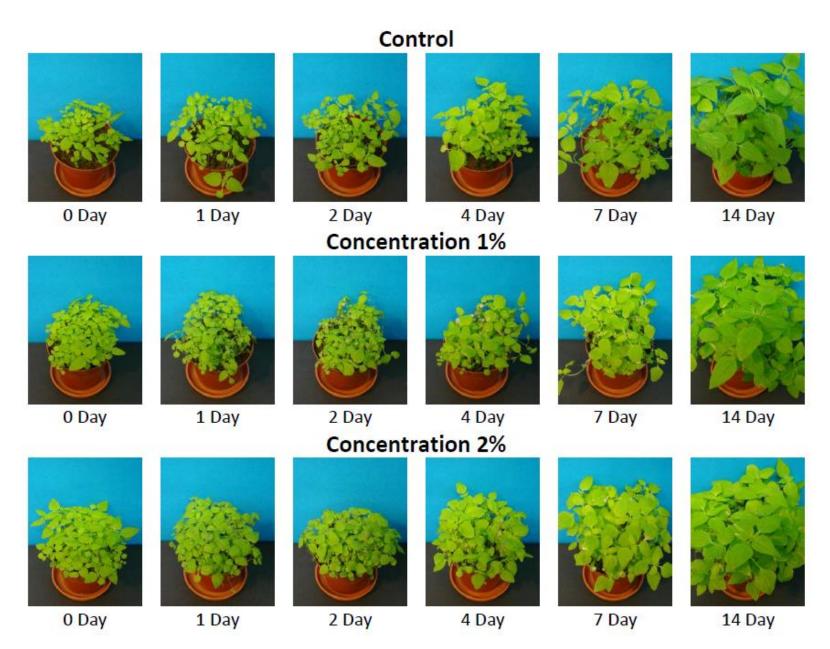


Figure S32. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **(R)-DCA**.

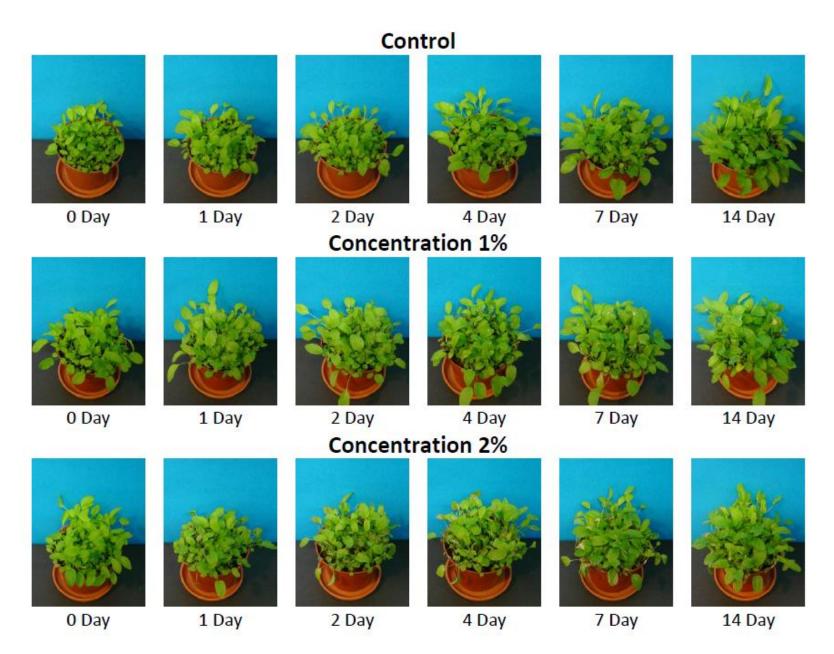


Figure S33. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of (R)-DCA.

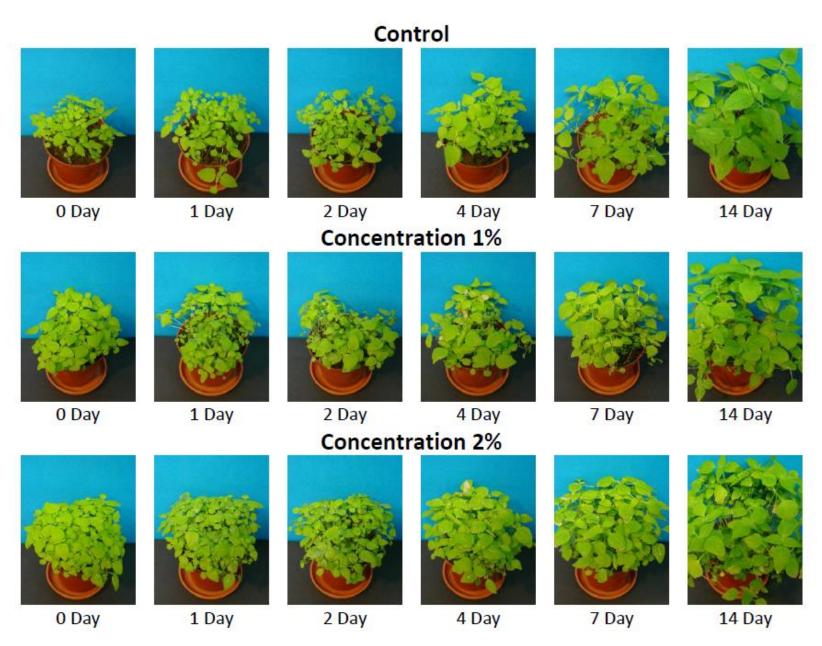


Figure S34. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **(S)-DCA**.

Control 7 Day 0 Day 1 Day 2 Day 4 Day 14 Day **Concentration 1%** 1 Day 2 Day 4 Day 7 Day 0 Day 14 Day **Concentration 2%** 1 Day 2 Day 4 Day 0 Day 7 Day 14 Day

Figure S35. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of **(S)-DCA**.

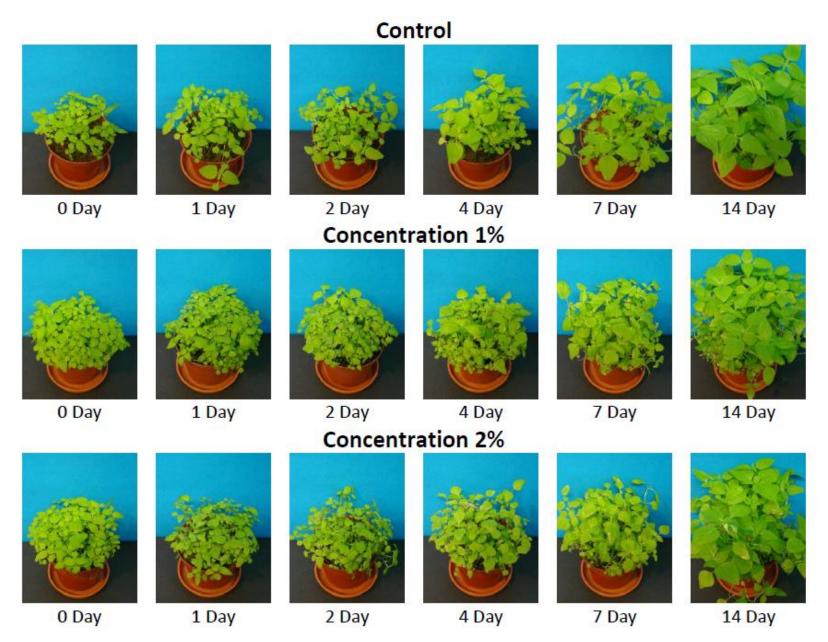


Figure S36. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **(R)-TCA**.

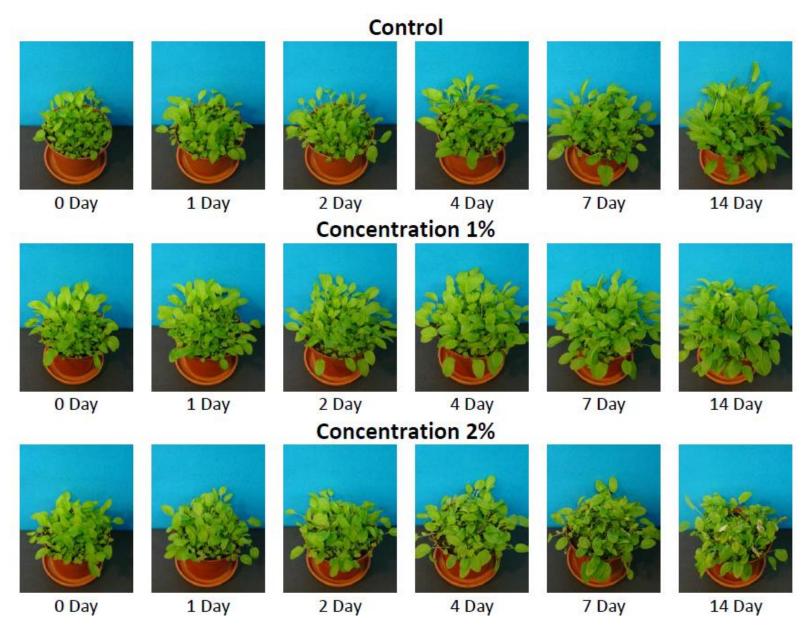


Figure S37. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of **(R)-TCA**.

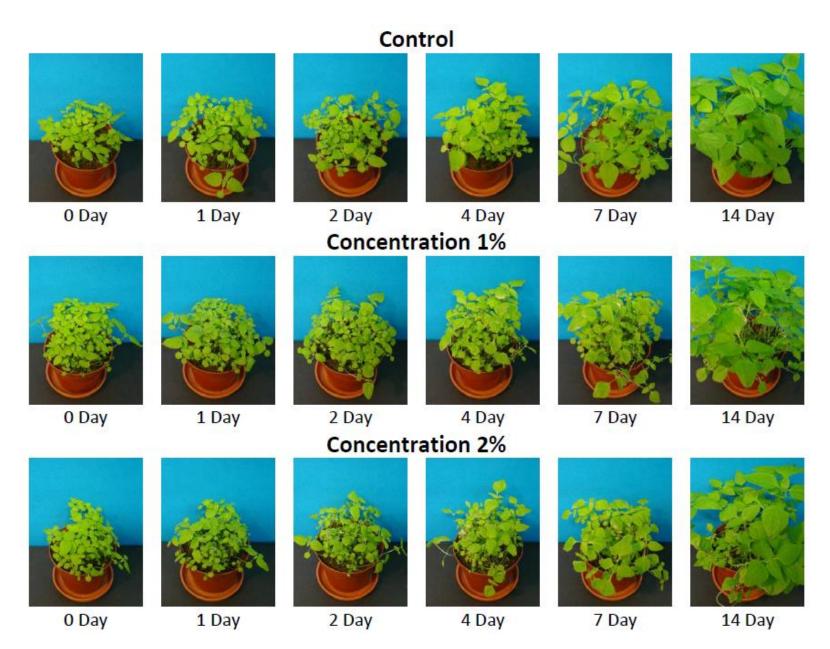


Figure S38. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **(S)-TCA**.

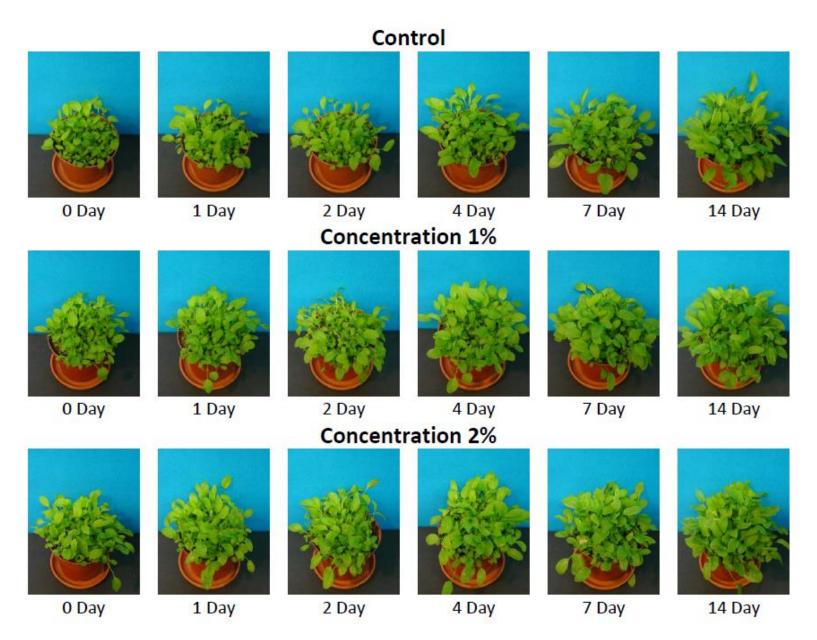


Figure S39. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of **(S)-TCA**.

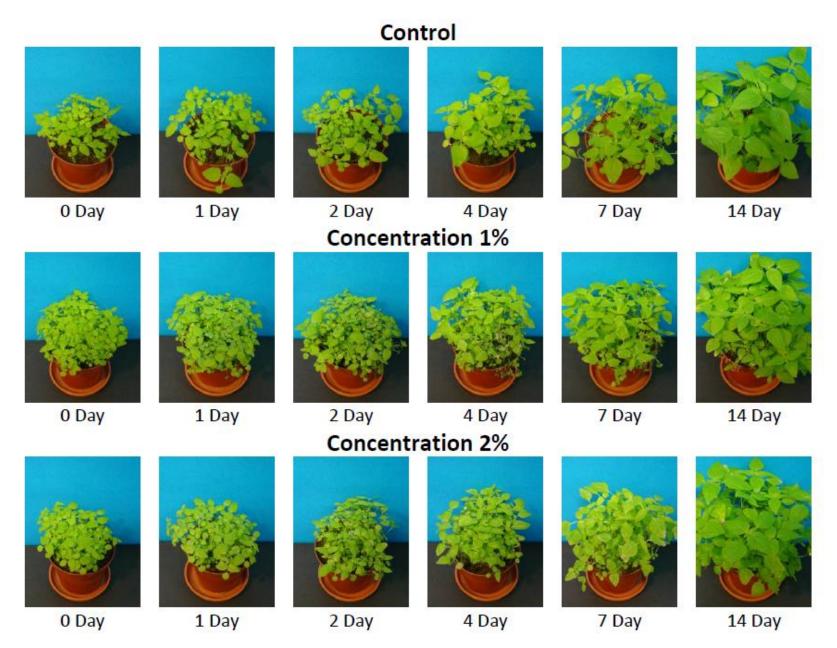


Figure S40. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of (R)-SO₄.

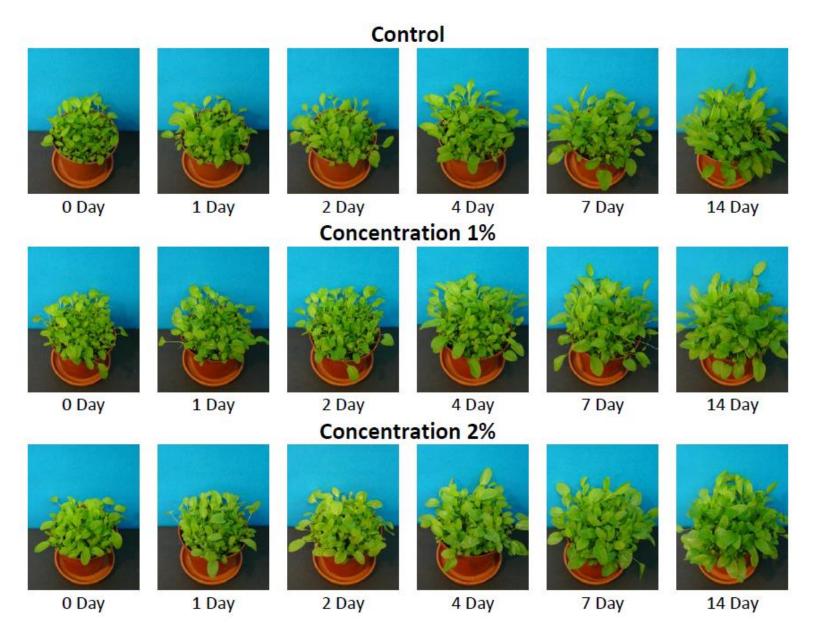


Figure S41. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of (R)-SO₄.

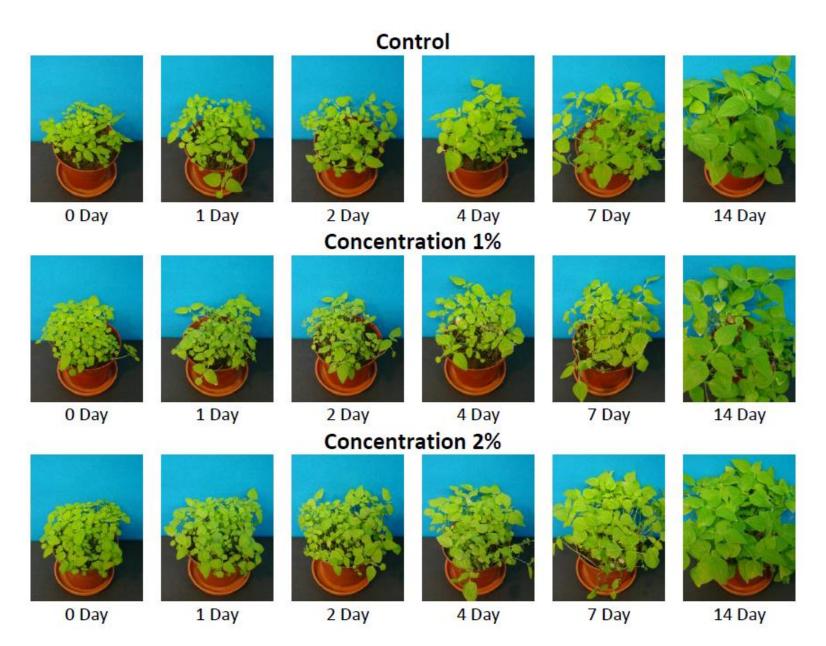


Figure S42. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of (S)-SO4.

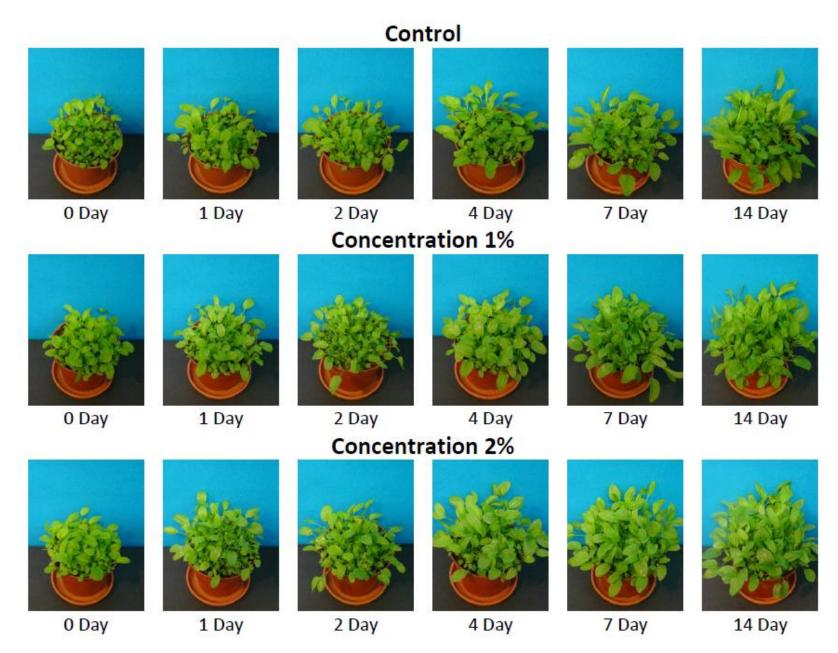


Figure S43. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of (S)-SO₄.

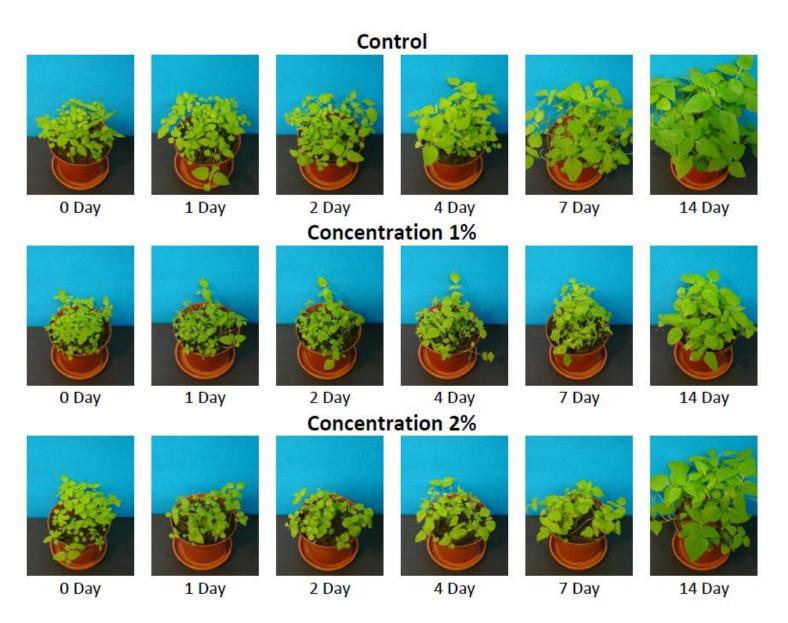


Figure S44. Digital photographs of gallant soldier after the spraying of 1% and 2% aqueous solutions of **DIPA-SO**₄.

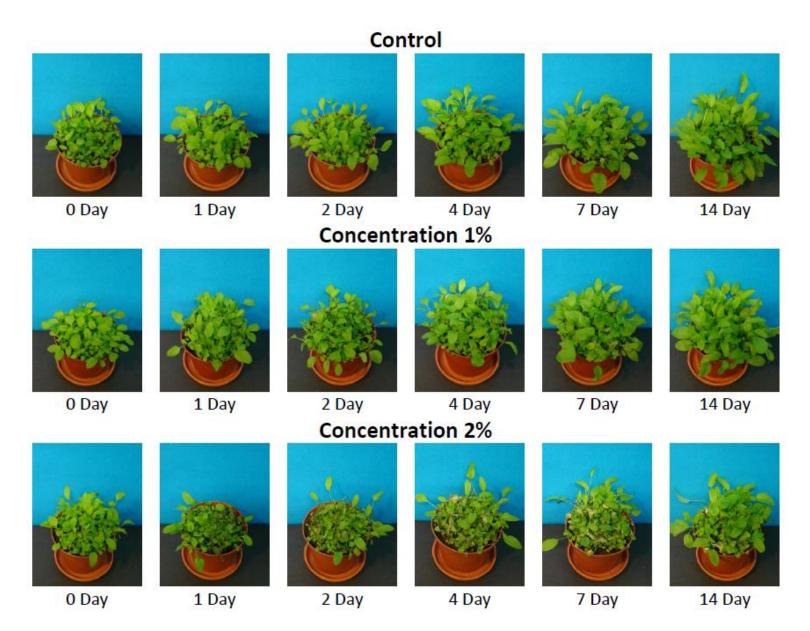


Figure S45. Digital photographs of common sorrel after the spraying of 1% and 2% aqueous solutions of **DIPA-SO**₄.