

Supplementary Material: Isotope analysis of iron on structural materials of nuclear power plant using double-pulse laser ablation molecular isotopic spectrometry

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S-1. Chemical composition of structural materials

Table S-1. Chemical Composition of Utilized Specimens

| Stainless steel 304 (wt%) | | Inconel 690 (wt%) | | Portland cement (wt%) | |
|---------------------------|------|-------------------|------|--------------------------------|------|
| Fe | 70.9 | Ni | 60.4 | CaO | 64.6 |
| Cr | 18.2 | Cr | 31.0 | SiO ₂ | 21.3 |
| Ni | 8.5 | Fe | 7.0 | Al ₂ O ₃ | 5.6 |
| Mn | 1.8 | Cu | 0.5 | Fe ₂ O ₃ | 3.4 |
| Si | 0.6 | Mn | 0.5 | MgO | 2.1 |
| N | 0.05 | Cu | 0.5 | SO ₃ | 2.1 |
| C | 0.05 | C | 0.05 | N ₂ O | 0.05 |
| S | 0.01 | Si | 0.02 | | |
| P | 0.03 | | | | |

S-2. Optimized measurement condition according to matrix

The optimized measurement conditions (see Table S-2) were determined by considering the result of Figure S-1.

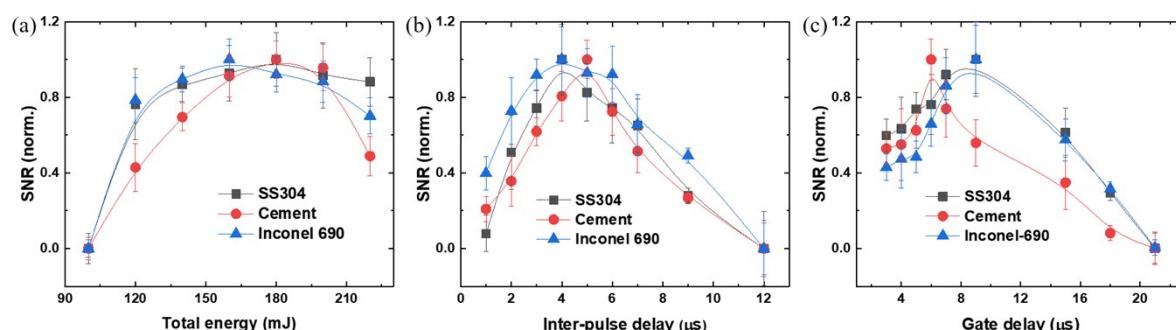


Figure S-1. Signal to noise ratio (SNR) of iron oxide emission as a function of (a) total energy, (b) inter-pulse delay, and (c) gate delay according to the matrix

Table S-2. Optimized Measurement Conditions for Surface Detection of Each matrix

| | Total energy (mJ) | Inter-pulse delay (μs) | Gate delay (μs) | Gate width (μs) |
|---------|-------------------|------------------------|-----------------|-----------------|
| SS304 | 180 | 4 | 4 | 100 |
| Inconel | 165 | 4 | 4 | 100 |
| Cement | 180 | 5 | 4 | 100 |

S-3. Methods of plasma characterization

Under the local thermal equilibrium (LTE) condition, plasma temperature (T) could be determined by Boltzmann plot method with below equation (1)^{S-1}.

$$\ln \left(\frac{I_{mn} \lambda_{mn}}{A_{ki} g_m} \right) = -\frac{E_m}{k_B T} + \ln \left(\frac{hc n_0(T)}{4\pi Z(T)} \right) \quad (1)$$

I_{mn} : Line intensity of transition from upper level m to lower lever n

λ_{mn} : Wavelength of the transition

A_{mn} : Spontaneous emission probability

g_m : Statistic weight of the upper level m

E_m : Energy of the upper level m

k_B : Boltzmann constant

h : Planck constants

$n_0(T)$: Number density

$Z(T)$: Partition function

Table S-3. Utilized Spectral Parameters of Iron Atomic Emission^{S-1}

| λ_{mn} (nm) | E_m (cm ⁻¹) | g_m | A_{mn} (x10 ⁸ s ⁻¹) |
|---------------------|---------------------------|-------|--|
| 371.9 | 26875 | 11 | 0.163 |
| 373.5 | 33695 | 11 | 0.902 |
| 374.6 | 27395 | 7 | 0.115 |
| 375.8 | 34329 | 7 | 0.634 |
| 376.4 | 34547 | 5 | 0.540 |

Plasma electron density could be obtained using Stark broadening effect of Fe I (541.4 nm) with the below equation (2)^{S-2}.

$$\Delta\lambda_S^{Fe} = 0.3612 \frac{N_e}{10^{23}} \left(\frac{T}{13000} \right)^{1.6551} \quad (2)$$

$\Delta\lambda_S^{Fe}$: Linewidth of iron atomic line (541.4 nm)

N_e : electron density [m⁻³]

References

(S-1) Rifai, K.; Laville, S.; Vidal, F.; Sabsabi, M.; Chaker, M. Quantitative analysis of metallic traces in water-based liquids by UV-IR double-pulse laser-induced breakdown spectroscopy. *J. Anal. At. Spectrom.* **2012**, 27(2), 276-283.

(S-2) Zielinska, S.; Pellerin, S.; Dzierzega, K.; Valensi, F.; Musiol, K.; Briand, F. Measurement of atomic Stark parameters of many Mn I and Fe I spectral lines using GMAW process. *J. Phys. D: Appl. Phys.* **2010**, 43(43), 434005.