

The supporting information for

**A synergetic effect between photogenerated carriers and photothermally enhanced electrochemical urea-assisted hydrogen generation on Ni-NiO/Nickel Foam catalyst**

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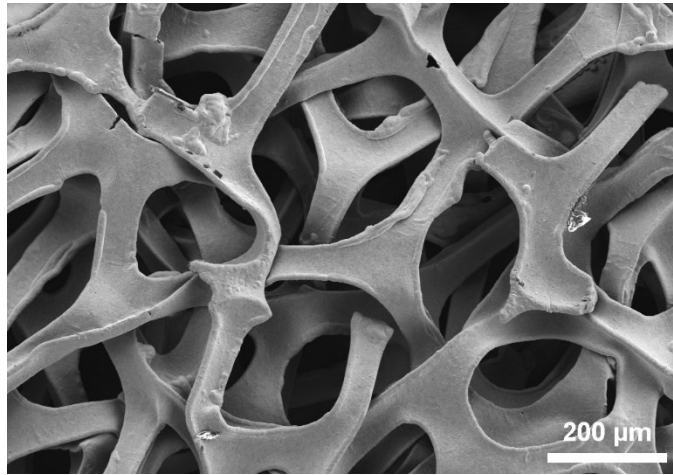
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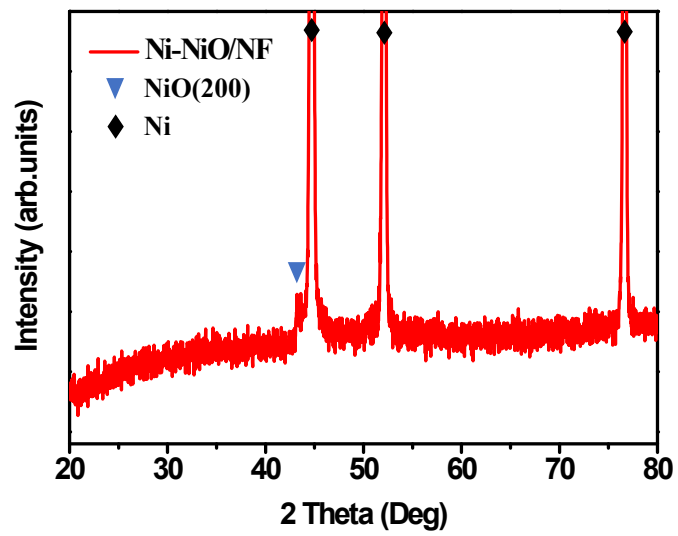
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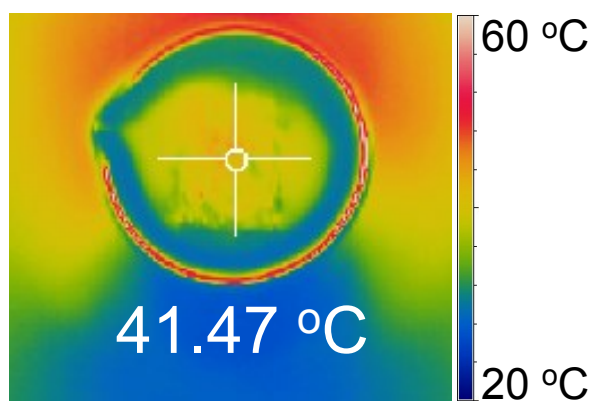
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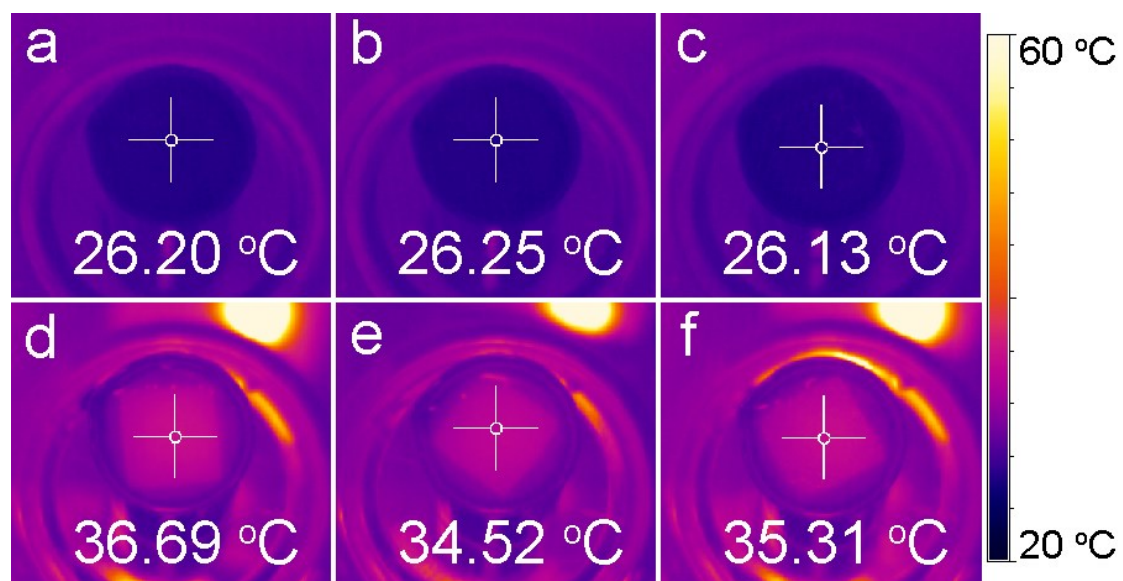
**Fig. S1.** SEM image of the Ni foam.



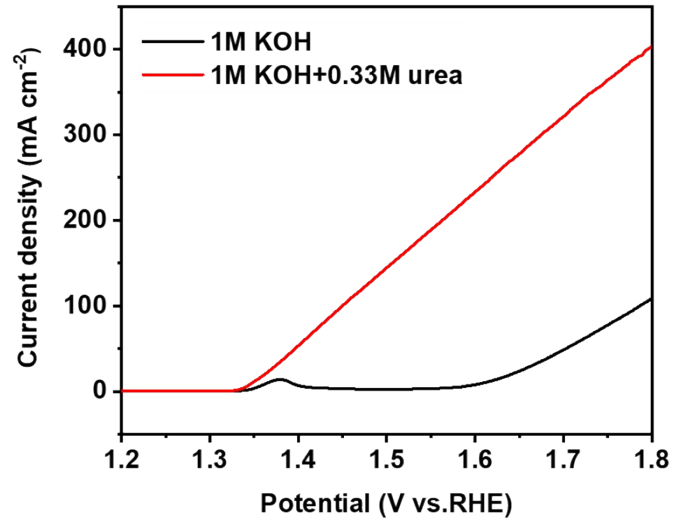
**Fig. S2.** X-ray diffraction (XRD) pattern of Ni-NiO/NF.



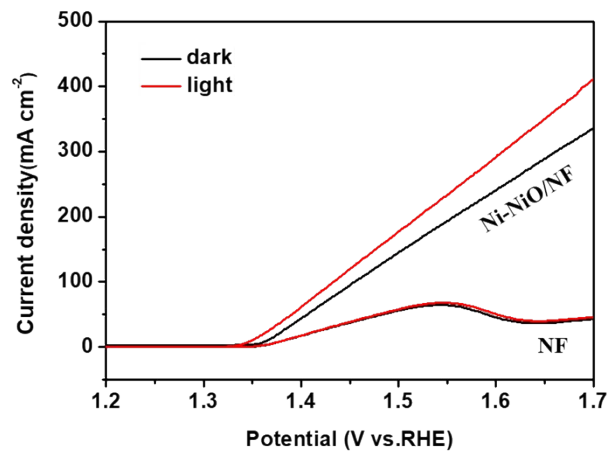
**Fig. S3.** Infrared photograph of Ni-NiO/NF at full spectrum irradiation of  $150 \text{ mW cm}^{-2}$  for 60 seconds.



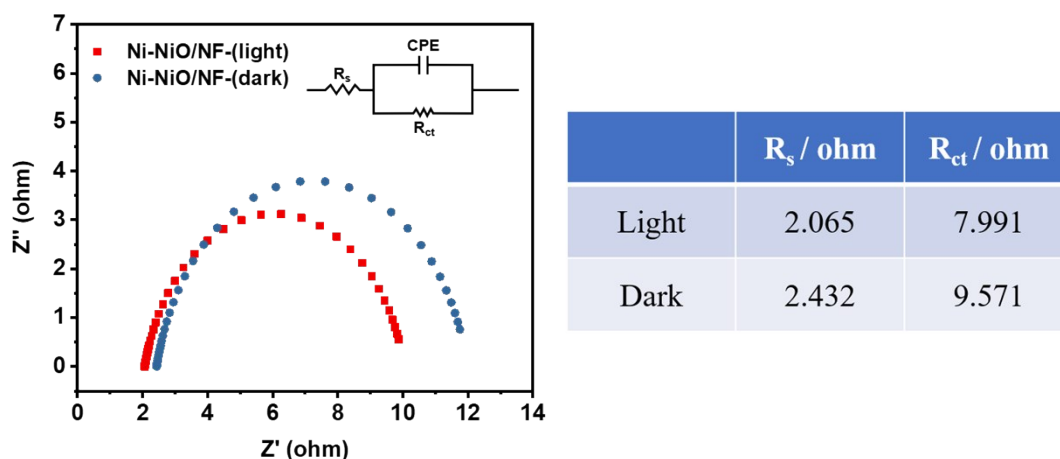
**Fig. S4.** Infrared photographs without irradiation of (a) Ni-NiO/NF, (b) Ni(OH)<sub>2</sub>/NF and (c) NF. Bottom row: same infrared photographs under full spectrum irradiation of the highest temperature of (d) Ni-NiO/NF, (e) Ni(OH)<sub>2</sub>/NF and (f) NF.



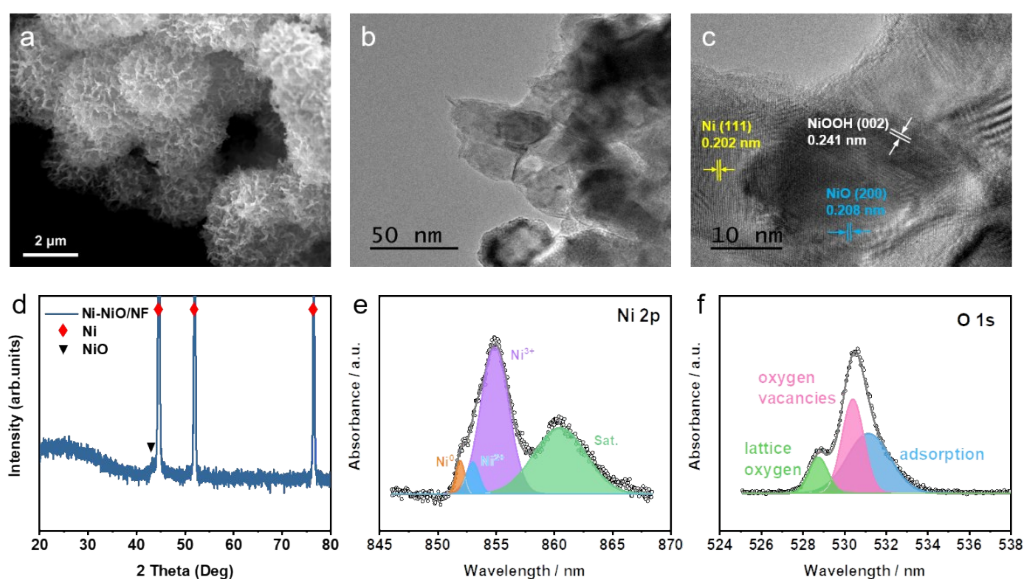
**Fig. S5.** Linear sweep voltammetry curves for the UOR and OER of Ni-NiO/NF.



**Fig. S6.** Polarization curves for the UOR of Ni-NiO/NF and NF with and without irradiation.



**Fig. S7.** Electrochemical impedance spectroscopy (EIS): Nyquist plots of Ni-NiO/NF electrode for UOR. The solution resistance and charge transfer resistance for Ni-NiO/NF electrode are also listed.



**Fig. S8.** (a) The SEM image of Ni-NiO after the durability test. (b) TEM image, (c) HRTEM image of Ni-NiO/NF after the durability test. (d) X-ray diffraction (XRD) pattern of Ni-NiO/NF after the durability test. XPS spectra of (e) Ni 2p and (f) O 1s of Ni-NiO/NF after the durability test.

**Table S1** The atomic concentration of Ni, O and C species for Ni-NiO/NF and Ni(OH)<sub>2</sub>/NF from full XPS spectrum.

Catalysts	% At Conc		
	Ni 2p	O 1s	C 1s
Ni-NiO/NF	36	40	24
Ni(OH) <sub>2</sub> /NF	27	45	28

**Table S2** Comparison of the activity of Ni-NiO/NF||Ni-NiO/NF and other reported materials in the urea-assisted electrochemical hydrogen production.

Catalysts	Electrolyte	$\eta$ (mV) at $j = 10 \text{ mA cm}^{-2}$	Cell voltage (V @ $j \text{ mA cm}^{-2}$ )	Stability	Reference
<b>Ni-NiO/NF  Ni-NiO/NF</b>	<b>1 M KOH+0.33M Urea</b>	<b>1.48</b>	<b>1.7@60</b>	<b>5 h</b>	<b>This work</b>
NF@Acid-H <sub>2</sub>   NF@Acid-H <sub>2</sub>	1 M KOH+0.5 M Urea	1.49	1.62@50	20 h	1
1% Cu: $\alpha$ -Ni(OH) <sub>2</sub> /NF  1% Cu: $\alpha$ -Ni(OH) <sub>2</sub> /NF	1 M KOH+0.33 M Urea	1.49	1.74@60	40 h	2
Ni <sub>4</sub> N/Cu <sub>3</sub> N/CF  Ni <sub>4</sub> N/Cu <sub>3</sub> N/CF	1 M KOH+0.5 M Urea	1.48	1.7@50	10 h	3
Ni-Co <sub>9</sub> S <sub>8</sub> /CC  Ni-Co <sub>9</sub> S <sub>8</sub> /CC	1 M KOH+0.33 M Urea	1.52	1.76@60	20 h	4
CoMn/CoMn <sub>2</sub> O <sub>4</sub>   CoMn/CoMn <sub>2</sub> O <sub>4</sub>	1 M KOH+0.5 M Urea	1.51	1.68@100	60000s	5
Pt/C  RuO <sub>2</sub>	1 M KOH+0.33 M Urea	1.53	--	--	6

MnO <sub>2</sub> /MnCo <sub>2</sub> O <sub>4</sub>    MnO <sub>2</sub> /MnCo <sub>2</sub> O <sub>4</sub>	1 M KOH+0.5 M Urea	1.58	1.85@60	15 h	7
NiMoS  NiMoS	1 M KOH+0.5 M Urea	1.59	2@120	10 h	8
NF-Pt/C  NF-Pt/C	1 M KOH+0.5 M Urea	1.68	--	--	9
Pt/C-IrO <sub>2</sub>    Pt/C-IrO <sub>2</sub>	1 M KOH+0.5 M Urea	1.72	--	--	9

## References

- 1 B. Zhang, S. Wang, Z. Ma and Y. Qiu, *Applied Surface Science*, 2019, **496**, 143710.
- 2 J. Xie, L. Gao, S. Cao, W. Liu, F. Lei, P. Hao, X. Xia and B. Tang, *Journal of Materials Chemistry A*, 2019, **7**, 13577-13584.
- 3 J. Li, C. Yao, X. Kong, Z. Li, M. Jiang, F. Zhang and X. Lei, *ACS Sustainable Chemistry & Engineering*, 2019, **7**, 13278-13285.
- 4 P. Hao, W. Zhu, L. Li, J. Tian, J. Xie, F. Lei, G. Cui, Y. Zhang and B. Tang, *Electrochimica Acta*, 2020, **338**, 135883.
- 5 C. Wang, H. Lu, Z. Mao, C. Yan, G. Shen and X. Wang, *Advanced Functional Materials*, 2020, **30**, 2000556.
- 6 Q. Liu, L. Xie, F. Qu, Z. Liu, G. Du, A. M. Asiri and X. Sun, *Inorganic Chemistry Frontiers*, 2017, **4**, 1120-1124.
- 7 C. Xiao, S. Li, X. Zhang and D. R. MacFarlane, *Journal of Materials Chemistry A*, 2017, **5**, 7825-7832.
- 8 X. Wang, J. Wang, X. Sun, S. Wei, L. Cui, W. Yang and J. Liu, *Nano Research*, 2018, **11**, 988-996.
- 9 S. Chen, J. Duan, A. Vasileff and S. Z. Qiao, *Angewandte Chemie International Edition*, 2016, **55**, 3804-3808.