

The application of amine-based materials for carbon capture and utilisation: an overarching view

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Table S1. Performance of various selected polyamine-based CO₂ adsorbents (under flue gas or air –relevant conditions).

Adsorbent (class)	Amine (wt% / mmol N/g)	Temp. / °C (des. gas)	CO ₂ conc. ^a (humidity)	Total system pressure ^b	CO ₂ capacity ^c / mmol/g	Method	No. cycles	Comments	Ref.
INORGANIC OXIDE-SUPPORTED									
48% PEI/SBA-15	PEI (M _w 1800) 48 wt% (11.14 mmol N/g)	Ads: 30 Des: 120	400 ppm	1 atm	1.10	TGA			(1)
41% PEI/SBA-15	PEI (M _w 1800) 48 wt% (9.52 mmol N/g)	Ads: 30 Des: 120	10%	1 atm	1.23	TGA			(1)
50% PPG/SBA-15 (1)	PPG (50 wt%)	Ads: 30 Des: 120	400 ppm	1 atm	0.63	TGA			(1)
	PPG (50 wt%)	Ads: 30 Des: 120	10%	1 atm	1.12 – 0.49	TGA	5		(1)
EtSNTs-50 PEI encapsulated in nanotubes	PEI (Mw 600) (11.2 mmol N/g)	Ads: 30 Des: 100	400 ppm	1 atm	~ 1.00	Fix bed Reactor (breakthrough)	8		(2)

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40% TPTA/SBA-15	TPTA 41 wt% organic (8.71 mmol N/g)	Ads: 35 Des: 70	400 ppm	1 atm	3.15 mmol CO ₂ /g SiO ₂ (1.86 mmol/g sorbent)	TGA			(3)
SBA-15/PEI/PEG200	PEI (Mw 800) (8.38 mmol N/g)	Ads: 30 Des: 110	400 ppm	1 atm	1.64	TGA			(4)
Templated Alumina/PEI/PEG200	PEI (Mw 800) (5.98 mmol N/g)	Ads: 30 Des: 110	400 ppm	1 atm	1.29	TGA			(4)
TRI-SBA-15-130-0.4 (2)	TRI (~30 wt% organic)	Ads: 25 Des: 120 (N ₂)	5%	1 atm	1.88	TGA			(5)
		Ads: 25 Des: 120 (N ₂)	5%	1 atm	1.78 – 1.82	TGA	12		(5)
(PEI-25-APTES-25)-PQCS2129 (1/2)	PEI, APTES (7.5 mmol N/g)	Ads: 40 Des: 120 (N ₂)	0.04 bar	0.81 bar	2.70	Perkin–Elmer Pyris 1 TGA		N ₂ to balance pressure during adsorption. Isosteric heat of adsorption at CO ₂ uptake of 2.6 mmol/g calculated as 56 kJ/mol CO ₂ .	(6)
		Ads: 60 Des: 105 (He) Regen:105 (He, 90 vol % H ₂ O)	10% (8 vol %)	1.01 bar	2.43 ± 0.26	Fixed-bed flow system (breakthrough)	8	Regeneration under highly humidified He stream.	(6)
PEI / TMPED / silica (1/2)	PEI, TMPED	Ads: 50 Des: 175 (He/ N ₂ , 4% v/v H ₂ O)	13.4% (4% v/v)	1 atm	1.80 – 2.00	Fixed bed reactor	>100		(7)
FS-PEI-50 (1)	PEI (M _w 25,000) (50 wt %)	Ads: 25 Des: 85 (0.9 x10 ⁻³ bar) Regen:(air)	Ambient air (410 – 420 ppm)	10.3 bar	1.65 – 1.71	All-glass, grease-free flow system (Breakthrough)	4	“Air from the laboratory atmosphere was dried filtered and compressed to 150 psi and used directly for the	(8)

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								adsorption measurements”	
		Ads: 25 Des: 85 (0.9 x10 ⁻³ bar)	Ambient air (410 – 420 ppm) (RH 67%)	10.3 bar	1.41				(8)
FS-PEI-33 (1)	PEI (M_w 25000) (33 wt %)	Ads: 25 Des: 85 (0.9 x10 ⁻³ bar)	Ambient air (410 – 420 ppm) (RH 67%)	10.3 bar	1.74				(8)
PEHAS13 – MCF support (3)	PEI/aziridine (4.49 mmol N/g)	Ads: 75 Des: 110 (Ar)	10% (100% water saturation)	1 atm	0.85	packed bed flow reactor (Breakthrough)		Adsorption experiments were ended when the uptake rates became less than 0.2 μmol CO ₂ /min.	(9)
S-70C-2-24h (3)	PEI/aziridine (7.23 mmol N/g)	Ads: 25 Des: 110	10%	1 atm	0.93	TGA			(10)
SynA50 (1)	PEI (M_w 800) (11.20 mmol N/g) γ-alumina	Ads: 25 Des: 110 (in Ar)	400 ppm	1 atm	1.74	TGA			(11)
		Ads: 25 Des: 110 (in Ar)	10%	1 atm	1.95	TGA			
SynA40 (1)	PEI (M_w 800) (8.66 mmol n/g) γ-alumina	Ads: 25 Des: 110 (in Ar)	400 ppm	1 atm	1.33	TGA			(11)
		Ads: 25 Des: 110 (in Ar)	10%	1 atm	1.73	TGA			
MS40 (1)	PEI (M_w 800) (9.23 mmol N/g) Mesoporous silica SBA-15	Ads: 25 Des: 110 (in Ar)	400 ppm	1 atm	1.05	TGA			(11)

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		Ads: 25 Des: 110 (in Ar)	10%	1 atm	1.61	TGA			
PEI_H-SiO ₂ (1)	PEI (M_w 800) 2.62 g PEI/g sorbent	Ads: 30 Des: 110 (He)	400 ppm (19% RH)	1 atm	3.36 – ~ 2.8	Fixed bed system (breakthrough)	5	The breakthrough measurement continued until the effluent CO ₂ concentration reached 99 % of that of the feed stream.	(12)
		Ads: 30 Des: 110 (He)	400 ppm	1 atm	~ 2.34	Fixed bed system (breakthrough)	5		
		Ads: 50 Des: 110 (He)	400 ppm	1 atm	2.60	TGA			
		Ads: 80 Des: 110 (He)	10 %	1 atm	4.10	TGA			
PEI/Silica (1)	PEI (M_w 800) (10.5 mmol N/g)	Ads: 25 Des: 110	400 ppm	1 atm	2.36 – 1.65	TGA	4		(13)
A-PEI/Silica (1/2)	PEI (M_w 800), APTES (10.7 mmol N/g)	Ads: 25 Des: 110	400 ppm	1 atm	2.26 – 2.05	TGA	4		(13)
T-PEI/Silica (1)	PEI (M_w 800), tetrapropyl orthotitanate (10.5 mmol N/g)	Ads: 25 Des: 110	400 ppm	1 atm	2.19 – 2.16	TGA	4		(13)
PAA_MCM_41 (1)	PAA (M_n 1130 Da) (7.24 mmol N/g)	Ads: 25 Des: 120 (Ar)	400 ppm	1 atm	0.86	TGA			(14)
		Ads: 25 Des: 120 (Ar)	10%	1 atm	~ 1.50	TGA	3		

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PEHA-PO-1-2/50S (1)	PEHA (59.5 wt% organic) 1:2 molar ratio of PEHA/propylene oxide	Ads: 25 Des: 60 (N ₂)	400 ppm	1 atm	~ 1.25	TGA	15		(15)
TEPA-PO-1-2/50S (1)	TEPA (61.0 wt% organic) 1:2 molar ratio TEPA/propylene oxide	Ads: 25 Des: 60 (N ₂)	400 ppm	1 atm	~ 1.34	TGA	15		(15)
OTHER SUPPORTED									
F-C-PSI, 10% PEI	PEI (M _w 800 Da) (5.2 mmol N/g fibre)	Ads: 37 Des: 110 (N ₂)	10 mol% (100% RH)	1 atm	1.25 mmol/g fibre	Simulated flue gas flow system (breakthrough)		Capacity /g fibre	(16)
PAI/silica/PEI fibres	PEI (M _w 800) (5.9 mmol N/g fibre)	Ads: 35 Des: 90 (N ₂)	14 mol% (100% RH)	1 atm	0.85 mmol/g fibre (b) 1.19 mmol/g fibre (q)	Simulated flue gas flow system (breakthrough)		Capacity /g fibre	(17)
		Ads: 35 Des: 120 (He)	10%	1 atm	1.02 – 0.83 mmol/g fibre	TGA	5	Capacity /g fibre	
PAI/silica/PEI fibres + glycerol	PEI (M _w 800) (5.9 mmol N/g fibre)	Ads: 35 Des: 90 (N ₂)	14 mol% (100% RH)	1 atm	1.3 mmol/g fibre (b) 2.0 mmol/g fibre (q)	Simulated flue gas flow system (breakthrough)		Capacity /g fibre	(17)
PEI-CA-SiO ₂	PEI (M _w 800 Da) 0.7 g PEI/g SiO ₂	Ads: 35 Des: 90 (N ₂)	380 ppm	1 atm	0.62 mmol/g fibre (b) @ 95% of C0	Column breakthrough		Demonstrated vacuum- assisted desorption in TSA	(18)
		Ads: 35 Des: 110 (He)	395 ppm	1 atm	~0.62 mmol/g fibre	TGA	11	~95% of the CO ₂ capacity was retained at the end of cycles	

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PG-R20:1 PEI/glass fiber matrix + epichlorohydrin	PEI	Ads: 30 Des: 80 (vacuum)	16%	1 atm	0.26	Agilent 6820 gas chromatogram adsorption/desorption process (Breakthrough)			(19)
		Ads: 30 Des: 80 (vacuum)	16% (80% RH)	1 atm	3.98	(Breakthrough)			
PEI/G-silica sheets	PEI (M_n 600)	Ads: 75 Des: 120	15%	1 bar	3.89	magnetic suspension microbalance operated in a closed system	20		(20)
exfGO-D x 7.0TETA	TETA (7 g TETA/g sample = 88 wt%)	Ads: 75 Des: 70	100%	0.15 bar	7.5	Quantachrome Autosorb-iQC.			(21)
exfGO-D x 6.0TEPA	TETA (6 g/g sample = 86 wt%)	Ads: 75 Des: 100 (N ₂)	15%	1 bar	6.4 – 5.0	TGA	50		(21)
HP20/PEI-50	PEI (50 wt %)	Ads: 25 Des: 100 (N ₂)	400 ppm	1 atm	2.26 – 2.22	U-shaped glass tube	5		(22)
		Ads: 75 Des: 100 (N ₂)	15 %	1 atm	2.95 – 2.91	TGA	5		
PEI/PIM-1 composite	PEI (M_w 800) (21 wt %)	Ads: 35 Des: 110 (vacuum < 5 mTorr)	100%	0.0004 bar	0.23	Micromeritics ASAP 2020			(23)
		Ads: 35 Des: 110 (vacuum < 5 mTorr)	100%	0.1 bar	1.15	Micromeritics ASAP 2020			
		Ads: 35 Des: 110	10%	1 atm	~0.9	TGA	10	Lost ~ 10% uptake capacity over 10 cycles	

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		(N ₂)							
A-PEI-300	PEI (M _w 300)	Ads: 25 Des: 110 (vacuum)	100%	0.15 bar	3.60	Volumetric (PCTpro-E&E)		Selectivity measurement showed no N ₂ adsorption. IAST measured: from 12 - 45 kJmol/g between ~ 0.5 - 3.0 mmol CO ₂ /g	(24)
35 wt% PEI/3 wt% (Zn)ZIF-8/SiO ₂	PEI (M _w 800) 35 wt%	Ads: 50 Des: 120 (5.6 v/v% H ₂ O/N ₂)	15% (5.6 v/v %)	1 atm	2.84 (50th cycle) (deactivation of 8% between 10 - 250th cycle)	Packed Bed Reactor	250	Adsorption gas included 4.5% O ₂ Stability to SO ₂ , NO _x and H ₂ S measured	(25)
AEAPDMS-NFC	AEAPDMS-NFC (5.9 mmol N/g)	Ads: 30 Des: 90 (30 mbar)	400–530 ppm (60% RH)	1 atm	(average ads capacity) 0.90 reduction in adsorption capacity by up to 5%	Packed bed	100	TVS cycles, reduction of adsorption capacity by up to 5%.	(26)
PP-AM-HBP-NH2	PEHA (6.39 mmol N/g)	Ads: 25 Des: 90 (N ₂)	10% volume	1 atm	5.60 ± 0.1	Breakthrough	15	Sample pre-swollen in water at 25 °C. Swelling degree was 81.69%.	(27)
UNSUPPORTED									
PEI-C60	PEI (M _w 25,000) (75 wt%)	Ads: 90 Des: 90 (vacuum)	100%	0.15 bar	3.41	Setaram PCTPro volumetric apparatus, (single component)		High CO ₂ /N ₂ and CO ₂ /CH ₄ selectivity demonstrated. Cycles showed 60% starting capacity after 100 cycles	(28)
PEI HB-4.0% EPC	PEI	Ads: ambient Des: 160 (air)	15%	~ 2 bar	~1.37	CO ₂ filled balloon equipped with needle inserted into screw-cap septum seal of vial containing sample	10	Water weight ratio of sample 70% Sample also regenerated in microwave	(29)
1-G0/600PEI	PEI (M _w 600)	Ads: 25 Des: 120 (He)	60 mL/min CO ₂ /30 mL/min He	1 atm	0.93	TGA		Heat of adsorption 103 kJ mol/g at low adsorption values.	(30)

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								Measured on Micromeritics ASAP- 2020 apparatus	
		Ads: 65 Des: 120 (humid N ₂)	15% (5.7 vol% H ₂ O)	1 atm	2.98 (average)	Packed bed reactor (breakthrough)	350	Simulated flue gas contained 4.5 vol% O ₂	
1-G0-TEPA	TEPA (14 wt% N)	Ads: 25 Des: 120 (He)	60 mL/min CO ₂ /30 mL/min He	1 atm	2.02	TGA			(31)
		Ads: 65 Des: 120 (humid N ₂)	15% (5.7 vol% H ₂ O)	1 atm	1.68 (average)	Packed bed reactor (breakthrough)	25	Simulated flue gas contained 4.5 vol% O ₂	
BC40	PEI (M _w 25,000) (5.79 mmol N/g)	Ads: 90 Des: 155 (100% CO ₂)	10%	1 atm	2.16 – 2.14	TGA	29		(32)
		Ads: 90 Des: 110 (vacuum)	100%	0.1 bar	2.29	Isorb (Gas sorption analyzer)			
E200-FZ	PEI (M _w 750,000) (8.2 mmol N/g)	Ads: 25 Des: 110 (He)	10%	1 atm	2.01	TGA			(33)
		Ads: 25 Des: 110 (N ₂)	10% (65% RH)	1 atm	3.36	Breakthrough			
E200-LN2	PEI (M _w 750,000) (8.8 mmol N/g)	Ads: 25 Des: 110 (He)	10%	1 atm	2.81	TGA			(33)
		Ads: 25 Des: 110 (N ₂)	10% (65% RH)	1 atm	5.50	Breakthrough			

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^a If total pressure 1 atm, balance is inert gas Ar, He or N₂

^b Total pressure assumed 1 atm unless otherwise stated for TGA and breakthrough

^c Maximum adsorption capacity reported/equilibrium capacity

(b) breakthrough capacity

(q) pseudo equilibrium capacity

PEI: Polyethylenimine

PPG: Poly(propylene guanidine)

PAA: Poly(allylamine)

TEPA: Tetraethylenepentamine

TETA: Triethylenetetramine

PEHA: Pentaethylenehexamine

TRI: 3-[2-(2-Aminoethylamino)ethylamino]propyl trimethoxysilane

APTES: 3-(aminopropyl)triethoxysilane

TMPED: N¹-(3-trimethoxysilylpropyl)diethylenediamine

AEAPDMS-NFC: N-(2-aminoethyl)-3-aminopropylmethyldimethoxysilane nanofibrillated cellulose

References

1. Park SJ, Lee JJ, Hoyt CB, Kumar DR, Jones CW. Silica supported poly(propylene guanidine) as a CO₂ sorbent in simulated flue gas and direct air capture. *Adsorption*. 2020;26(1):89-101.
2. Liu L, Chen J, Tao L, Li H, Yang Q. Aminopolymer Confined in Ethane-Silica Nanotubes for CO₂ Capture from Ambient Air. *ChemNanoMat*. 2020;6(7):1096-103.
3. Pang SH, Lee L-C, Sakwa-Novak MA, Lively RP, Jones CW. Design of Aminopolymer Structure to Enhance Performance and Stability of CO₂ Sorbents: Poly(propylenimine) vs Poly(ethylenimine). *Journal of the American Chemical Society*. 2017;139(10):3627-30.
4. Sakwa-Novak MA, Tan S, Jones CW. Role of Additives in Composite PEI/Oxide CO₂ Adsorbents: Enhancement in the Amine Efficiency of Supported PEI by PEG in CO₂ Capture from Simulated Ambient Air. *ACS Applied Materials & Interfaces*. 2015;7(44):24748-59.
5. Jahandar Lashaki M, Sayari A. CO₂ capture using triamine-grafted SBA-15: The impact of the support pore structure. *Chemical Engineering Journal*. 2018;334:1260-9.
6. Fauth DJ, Gray ML, Pennline HW, Krutka HM, Sjostrom S, Ault AM. Investigation of Porous Silica Supported Mixed-Amine Sorbents for Post-Combustion CO₂ Capture. *Energy & Fuels*. 2012;26(4):2483-96.
7. Fisher II JC, Gray M. Cyclic Stability Testing of Aminated-Silica Solid Sorbent for Post-Combustion CO₂ Capture. *ChemSusChem*. 2015;8(3):452-5.
8. Goeppert A, Czaun M, May RB, Prakash GKS, Olah GA, Narayanan SR. Carbon Dioxide Capture from the Air Using a Polyamine Based Regenerable Solid Adsorbent. *Journal of the American Chemical Society*. 2011;133(50):20164-7.
9. Drese JH, Choi S, Didas SA, Bollini P, Gray ML, Jones CW. Effect of support structure on CO₂ adsorption properties of pore-expanded hyperbranched aminosilicas. *Microporous and Mesoporous Materials*. 2012;151:231-40.
10. Chaikittisilp W, Didas SA, Kim H-J, Jones CW. Vapor-Phase Transport as A Novel Route to Hyperbranched Polyamine-Oxide Hybrid Materials. *Chemistry of Materials*. 2013;25(4):613-22.
11. Chaikittisilp W, Kim H-J, Jones CW. Mesoporous Alumina-Supported Amines as Potential Steam-Stable Adsorbents for Capturing CO₂ from Simulated Flue Gas and Ambient Air. *Energy & Fuels*. 2011;25(11):5528-37.
12. Kwon HT, Sakwa-Novak MA, Pang SH, Sujan AR, Ping EW, Jones CW. Aminopolymer-Impregnated Hierarchical Silica Structures: Unexpected Equivalent CO₂ Uptake under Simulated Air Capture and Flue Gas Capture Conditions. *Chemistry of Materials*. 2019;31(14):5229-37.
13. Choi S, Gray M, Jones C. Amine-Tethered Solid Adsorbents Coupling High Adsorption Capacity and Regenerability for CO₂ Capture From Ambient Air. *ChemSusChem*. 2011;4:628-35.

Electronic Supplementary Information

14. Chaikittisilp W, Khunsupat R, Chen TT, Jones CW. Poly(allylamine)-Mesoporous Silica Composite Materials for CO₂ Capture from Simulated Flue Gas or Ambient Air. *Industrial & Engineering Chemistry Research*. 2011;50(24):14203-10.
15. Goeppert A, Zhang H, Sen R, Dang H, Prakash GKS. Oxidation-Resistant, Cost-Effective Epoxide-Modified Polyamine Adsorbents for CO₂ Capture from Various Sources Including Air. *ChemSusChem*. 2019;12(8):1712-23.
16. Labreche Y, Lively RP, Rezaei F, Chen G, Jones CW, Koros WJ. Post-spinning infusion of poly(ethyleneimine) into polymer/silica hollow fiber sorbents for carbon dioxide capture. *Chemical Engineering Journal*. 2013;221:166-75.
17. Labreche Y, Fan Y, Rezaei F, Lively RP, Jones CW, Koros WJ. Poly(amide-imide)/Silica Supported PEI Hollow Fiber Sorbents for Postcombustion CO₂ Capture by RTSA. *ACS Applied Materials & Interfaces*. 2014;6(21):19336-46.
18. Sujan AR, Pang SH, Zhu G, Jones CW, Lively RP. Direct CO₂ Capture from Air using Poly(ethyleneimine)-Loaded Polymer/Silica Fiber Sorbents. *ACS Sustainable Chemistry & Engineering*. 2019;7(5):5264-73.
19. Li P, Ge B, Zhang S, Chen S, Zhang Q, Zhao Y. CO₂ Capture by Polyethylenimine-Modified Fibrous Adsorbent. *Langmuir*. 2008;24(13):6567-74.
20. Yang S, Zhan L, Xu X, Wang Y, Ling L, Feng X. Graphene-Based Porous Silica Sheets Impregnated with Polyethyleneimine for Superior CO₂ Capture. *Advanced Materials*. 2013;25(15):2130-4.
21. Gadipelli S, Lu Y, Skipper NT, Yildirim T, Guo Z. Design of hyperporous graphene networks and their application in solid-amine based carbon capture systems. *Journal of Materials Chemistry A*. 2017;5(34):17833-40.
22. Chen Z, Deng S, Wei H, Wang B, Huang J, Yu G. Polyethylenimine-Impregnated Resin for High CO₂ Adsorption: An Efficient Adsorbent for CO₂ Capture from Simulated Flue Gas and Ambient Air. *ACS Applied Materials & Interfaces*. 2013;5(15):6937-45.
23. Pang SH, Jue ML, Leisen J, Jones CW, Lively RP. PIM-1 as a Solution-Processable “Molecular Basket” for CO₂ Capture from Dilute Sources. *ACS Macro Letters*. 2015;4(12):1415-9.
24. Lin Y, Lin H, Wang H, Suo Y, Li B, Kong C, et al. Enhanced selective CO₂ adsorption on polyamine/MIL-101(Cr) composites. *Journal of Materials Chemistry A*. 2014;2(35):14658-65.
25. Luz I, Soukri M, Lail M. Flying MOFs: polyamine-containing fluidized MOF/SiO₂ hybrid materials for CO₂ capture from post-combustion flue gas. *Chemical Science*. 2018;9(20):4589-99.
26. Gebald C, Wurzbacher JA, Tingaut P, Steinfeld A. Stability of Amine-Functionalized Cellulose during Temperature-Vacuum-Swing Cycling for CO₂ Capture from Air. *Environmental Science & Technology*. 2013;47(17):10063-70.
27. He H, Hu Y, Chen S, Zhuang L, Ma B, Wu Q. Preparation and Properties of A Hyperbranch-Structured Polyamine adsorbent for Carbon Dioxide Capture. *Scientific Reports*. 2017;7(1):3913.
28. Andreoli E, Dillon EP, Cullum L, Alemany LB, Barron AR. Cross-Linking Amine-Rich Compounds into High Performing Selective CO₂ Absorbents. 2014;4:7304.
29. Xu X, Pejcic B, Heath C, Wood CD. Carbon capture with polyethylenimine hydrogel beads (PEI HBs). *Journal of Materials Chemistry A*. 2018;6(43):21468-74.
30. Thompson SJ, Soukri M, Lail M. Phosphorous dendrimer bound polyethyleneimine as solid sorbents for post-combustion CO₂ capture. *Chemical Engineering Journal*. 2018;350:1056-65.
31. Thompson SJ, Soukri M, Lail M. Phosphorus Dendrimer Derived Solid Sorbents for CO₂ Capture from Post-Combustion Gas Streams. *Energy & Fuels*. 2018;32(8):8658-67.
32. Hamdy LB, Wakeham RJ, Taddei M, Barron AR, Andreoli E. Epoxy Cross-Linked Polyamine CO₂ Sorbents Enhanced via Hydrophobic Functionalization. *Chemistry of Materials*. 2019;31(13):4673-84.
33. Yoo C-J, Narayanan P, Jones CW. Self-supported branched poly(ethyleneimine) materials for CO₂ adsorption from simulated flue gas. *Journal of Materials Chemistry A*. 2019;7(33):19513-21.