

## **Supporting Information**

### **Highly efficient non-noble metallic NiCu nanoalloy catalysts for hydrogenation of nitroarenes**

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**Table S1** Textural properties of the prepared catalysts

Sample	Ni (wt%)	Ni dispersion <sup>a</sup> (%)	Cu (wt%)	C (wt%)	S <sub>BET</sub> (m <sup>2</sup> g <sup>-1</sup> )	V <sub>p</sub> (cm <sup>3</sup> g <sup>-1</sup> )	D <sub>a</sub> (nm)	d <sub>x</sub> <sup>b</sup> (nm)	d <sub>t</sub> <sup>c</sup> (nm)
SiO <sub>2</sub>	/	/	/	/	324	1.03	13.0	/	/
C@SiO <sub>2</sub> -800	/	/	/	/	246	0.69	11.6	/	/
NiOCuO@SiO <sub>2</sub> -800	13.9	/	4.2	/	211	0.69	12.8	12.7	11.8
NiCu@SiO <sub>2</sub> -H <sub>2</sub> -800	14.9	5.0	4.6	/	223	0.69	12.4	21.4	20.2
Ni/C@SiO <sub>2</sub> -800	14.3	5.9	/	11.4	275	0.75	10.9	14.0	13.9
Cu/C@SiO <sub>2</sub> -800	/	/	4.0	13.3	213	0.62	11.7	19.9	20.0
NiCu/C@SiO <sub>2</sub> -750	13.8	8.2	3.7	8.1	243	0.68	10.7	4.0	3.9
NiCu/C@SiO <sub>2</sub> -800	13.7	7.8	3.6	9.7	241	0.67	11.2	4.4	4.6
NiCu/C@SiO <sub>2</sub> -850	13.8	7.5	3.7	10.3	226	0.65	11.7	5.0	5.1
NiCu/C@SiO <sub>2</sub> -900	13.9	6.7	3.6	9.1	200	0.65	12.8	6.5	6.4

<sup>a</sup>Ni dispersion was calculated by the CO chemisorption method.

<sup>b</sup>Average crystal size from XRD.

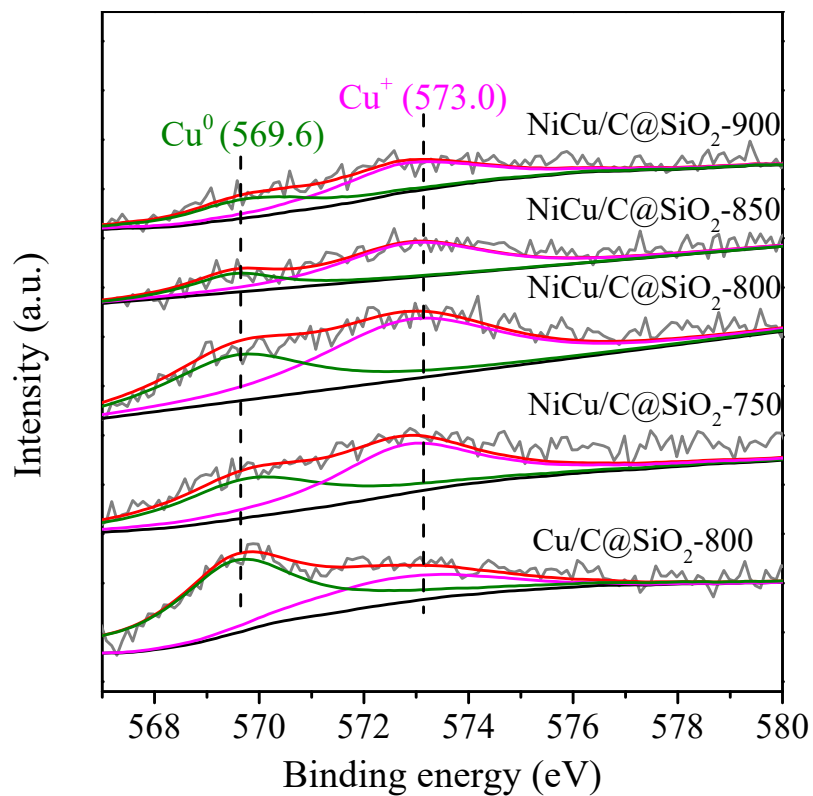
<sup>c</sup>Average particle size from TEM.

**Table S2** Total Ni contents and Ni compositions in individual Ni species on the catalyst surface by XPS

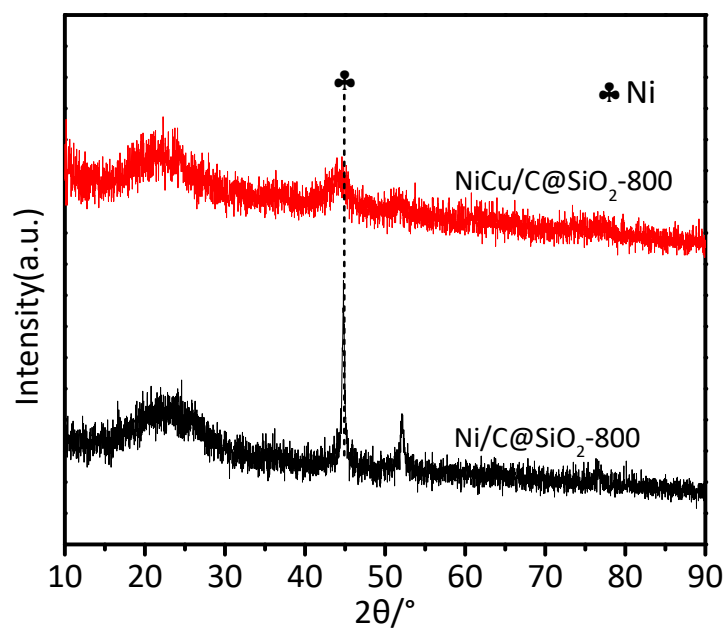
Catalyst	Total surface Ni (wt%)	Metallic Ni (wt%)	NiO (wt%)
Ni/C@SiO <sub>2</sub> -800	4.7	1.3	3.4
NiCu/C@SiO <sub>2</sub> -750	4.3	1.5	2.8
NiCu/C@SiO <sub>2</sub> -800	4.0	2.4	1.6
NiCu/C@SiO <sub>2</sub> -850	4.0	2.5	1.5
NiCu/C@SiO <sub>2</sub> -900	4.1	2.7	1.4

**Table S3** Activity comparison between NiCu/C@SiO<sub>2</sub>-800 and metal catalysts ever reported for selective reduction of nitrobenzene

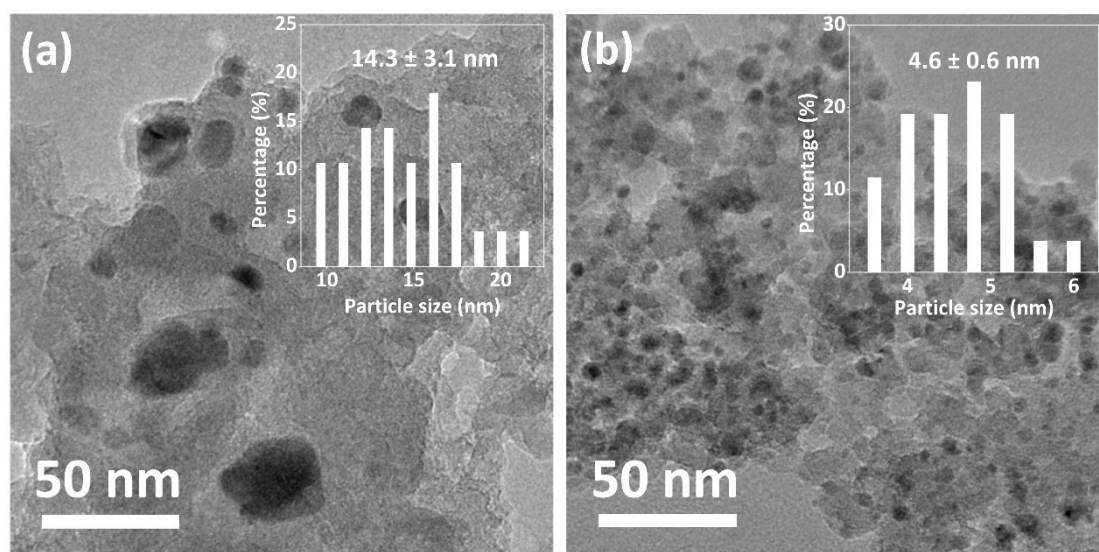
Catalyst	Nitrobenzene (mmol)	Hydrogen source	Temperature (°C)	Time	Con. (%)	TOF	Ref.
NiCu/C@SiO <sub>2</sub> -800	50	H <sub>2</sub>	120	0.5 h	91	46.5 s <sup>-1</sup>	This work
Ni-Co/CeO <sub>2</sub> -CEC	0.81	H <sub>2</sub>	150	1 h	88	563 h <sup>-1</sup>	1
NiFe@NC-3	0.64	H <sub>2</sub>	80	2.5 h	98	65 h <sup>-1</sup>	2
Ni/CeO <sub>2</sub> -CAS	2.4	H <sub>2</sub>	210	7 h	84	2 h <sup>-1</sup>	3
Co/mCN-900	1	H <sub>2</sub>	120	4 h	100	12 h <sup>-1</sup>	4
CoN <sub>x</sub> -OMC	1	H <sub>2</sub>	110	1.5 h	>99	56 h <sup>-1</sup>	5
Co@mesoNC	1	H <sub>2</sub>	110	2 h	55	42 h <sup>-1</sup>	6
acid MOF@C-800	0.5	H <sub>2</sub>	120	20 h	98	1 h <sup>-1</sup>	7
CoO <sub>x</sub> @NC-800	0.5	H <sub>2</sub>	110	5 h	100	2 h <sup>-1</sup>	8
Co@NC-800	0.5	H <sub>2</sub>	100	14 h	>99	12 h <sup>-1</sup>	9
0.09%Pt/γ-Al <sub>2</sub> O <sub>3</sub>	1.27	H <sub>2</sub>	40	50 min	2	216 h <sup>-1</sup>	10
0.2%Pt/Fe <sub>2</sub> O <sub>3</sub>	1	H <sub>2</sub>	30	2 h	>99	3170 h <sup>-1</sup>	11
Ni-N-C-700	0.25	NaBH <sub>4</sub>	r.t.	15 min	2	668 h <sup>-1</sup>	12
Fe-CN@550	1	NaBH <sub>4</sub>	50	10 h	98	1968 h <sup>-1</sup>	13
PdCo/CCF	1	NaBH <sub>4</sub>	r.t.	0.5 h	91	142 h <sup>-1</sup>	14
Co@NC-800	1	N <sub>2</sub> H <sub>4</sub> ·H <sub>2</sub> O	80	30 min	100	33 h <sup>-1</sup>	15



**Fig. S1 Cu LMM Auger spectra of  $\text{Cu/C@SiO}_2$  and  $\text{NiCu/C@SiO}_2$ -*T* catalysts.**



**Fig. S2** XRD patterns of spent Ni/C@SiO<sub>2</sub>-800 and NiCu/C@SiO<sub>2</sub>-800 catalysts for the hydrogenation of nitrobenzene after the 10th run.



**Fig. S3** TEM images of spent catalysts after the 10th run, (a) Ni/C@SiO<sub>2</sub>-800 and (b) NiCu/C@SiO<sub>2</sub>-800.

## References

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