Rationalization of Excited State Energy Transfer in D- π -A Porphyrin Sensitizers Enhancing Efficiency in Dye-Sensitized Solar Cells

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Figure S1. ¹H NMR spectrum (500 MHz, CDCl3) of 1.



Figure S2. HRMS of 1 (APPI-FTMS).



Figure S3. ¹H NMR spectrum (500 MHz, CDCl3) of 3.



Figure S4. MALDI-TOF of 3.



Figure S5. ¹H NMR spectrum (500 MHz, CDCl3) of LG8.



Figure S6. MALDI-TOF of LG8.



Figure S7. ¹H NMR spectrum (500 MHz, CDCl3) of LG9.



Figure S8. MALDI-TOF of LG9.



Figure S9. ¹H NMR spectrum (500 MHz, CDCl3) of LG10.



Figure S10. MALDI-TOF of LG10.



Figure S11. UV-visible absorption spectra of LG8-LG10 in THF solvent.



Figure S12. Theoretical absorption spectra of *LG9*, *LG10* Dyes by using *B3LYP* method *PCM* model in tetrahydrofuran solvent with M06-2X function.



Figure S13. Fluorescence spectra of LG8, LG9 and LG10 in THF solution



Figure S14. Reduction spectra of dyes LG8, LG9 and LG10 in THF.



Figure. S15. Optimized structure of LG8, LG9 and LG10 dyes.



Figure S16. Oxidative OTTLE studies of *LG 9* and *LG10* sensitizers in 0.3M TBAP/THF with an applied potential of +0.90V (vs. SCE/KCl).



Figure S17. a) Emission spectra of equi-absorbing solutions of Donor, Zn Por, LG8-LG10 (OD lex = 0.08) in CH₂Cl₂ solvent at 357 nm. b) Emission spectra of equi-absorbing solutions of Zn Por, LG8-LG10 (OD lex = 0.08) in CH₂Cl₂ solvent at 410 nm.



Figure S18. a) Emission spectra of equi-absorbing solutions of Donor, Zn Por, LG8-LG10 (OD lex = 0.08) in o-DCB solvent at 357 nm. b) Emission spectra of equi-absorbing solutions of Zn Por, LG8-LG10 (OD lex = 0.08) in o-DCB solvent at 410 nm.



Figure S19. Streak images for LG8, LG9 and LG10 dyes with an electrolyte (a) containing I_3 /I redox couple, (b) without redox mediator. The device was pump at 430 nm (10Hz) with an energy of 700 μ J/cm².



Figure S20. Comparison of transient absorption spectrum of LG8 (red curve), LG9 (green curve) and LG10 (blue curve)-based DSSC after pump pulse at 430nm with an energy of 700μ J/cm².



Figure S21. TG/DTG curves of *LG8*,*LG9*, and *LG10* porphyrins with heating rate 10 °C.min⁻¹ under nitrogen using 10 mg of sample.

Table S1: Optimized energies, HOMO-LUMO energies and ground state dipolemoment by DFT studies by using B3LYP/6-31G (d, p) in vacuum.

Dye	<i>ªE</i> , K.cal./mol	[▶] HOMO (H),	^b LUMO (L)	^b H-L gap	°µ
LG 8	-4703479	-4.666	-2.535	2.13	2.340
LG 9	-4300925	-4.664	-2.501	2.16	2.131
LG 10	-4212222	-4.732	-2.737	2.00	3.814

^aTotal minimum energy of LG8-LG10, ^bvalues in eV, ^cvalues in Debye units.

Dye	${}^a\lambda_{max}$	^b f	¢Е (eV)	% of Molecular Orbital Contribution	
	470	1 092	2.764	H-1→L+2 (30%), HOMO→L+1 (44%)H-6→LUMO (3%), H-2→LUMO	
	470	1.085		(4%), H-2→L+1 (4%), HOMO→LUMO (8%)	
	(2)(1,202 2,092	626	5 1 2 02	2 082	H-2→LUMO (16%), H-1→L+2 (33%), HOMO→L+1 (25%),
	030	1.295	2.982	HOMO→L+3 (11%), H-2→L+3 (5%), HOMO→LUMO (3%)	
	667	1.330	1.998	H-1->L+2 (11%), HOMO->LUMO (83%) HOMO->L+1 (2%)	
	LC 0 459	0.020	2.767	H-2→LUMO (45%), H-1→L+1 (28%), HOMO→L+2 (12%)	
LG9	438	0.929		H-5→LUMO (7%)	
	567	1 5 1 9	48 2.599	H-2→LUMO (28%), H-1→L+1 (46%), HOMO→LUMO (16%)	
	502	1.340		H-2→L+2 (2%), HOMO→L+3 (2%)	
	646	1.074		H-1→L+1 (13%), HOMO→LUMO (82%)	
LG10	467	1.156	3.062	H-1→L+1 (57%), HOMO->LUMO (11%), HOMO→L+2 (13%)	
	619	1 472	1.473 2.604	H-2→LUMO (30%), H-1→L+1 (44%), HOMO→LUMO (16%)	
		9 1.4/3		HOMO→L+3 (2%)	
	663	1.074	2.037	H-1→L+1 (13%), HOMO→LUMO (82%)	

Table S2: Singlet excited state properties of dyes by B3LYP method and M06-2X function intetrahydrofuran solvent in PCM model.

^aTheoretical absorbance in nm, ^bOscillator strength, and ^cExcited state energy in eV.