Self-Standing Porous Carbon Electrodes for Lithium-Oxygen Batteries with Lean Electrolyte and High Areal Capacity Condition

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Figure S1. TEM images of (a) CMK powder and (b) CMK-based membrane.



Figure S2. Photographic images of (a) SuperP-, (b) VXC-, (c) Black pearl-, (d) KB-, and (e) CMK-based self-standing membranes



Figure S3. Pore volume and BET surface area of series of carbon powders



Figure S4. Schematic illustration of cell configuration



Figure S5. XRD patterns of the Li₂O₂, KB-, and CMK-based membranes after discharge process. The *symbols, and •symbols denote Li₂O₂ crystal, and SUS-based sample folder, respectively.



Figure S6. SEM images of (a,b) KB-, and (c,d) CMK-based membranes (a,c) before and (b,d) after discharge process.



Figure S7. On-line electrochemical mass spectroscopy analysis of first cycle: (a,b) discharge/charge profiles; corresponding (c,d) O₂, CO₂, and H₂O exhaust rates during first charging of KB-, and CMK-based membranes. Dash gray lines indicate the theoretical O₂ evolution rate

carbon electrode		amount of				
details	mass loading (mg/cm ²)	electrolyte (mg/cm ²)	areal capacity (mAh/cm ²)	E/C (g/Ah)	cycle number	ref number
KB	5.5	25.80	4.00	6.5	30	This work
KB	5.5	20.50	4.00	5.1	15	This work
KB	5.5	15.20	4.00	3.8	1	This work
СМК	5.8	28.30	4.00	7.1	30	This work
CMK	5.8	22.60	4.00	5.7	20	This work
CMK	5.8	17.00	4.00	4.3	11	This work
KB	1.00	8.30	3.00	2.8	6	23
KB	5.40	28.20	4.00	7.1	37	23
Graphene	2.00	60.00	2.00	30.0	18	27
CNT	0.20	59.80	12.60	4.7	11	40
CNT	0.20	59.80	1.20	49.8	170	40
SuperP	30.00	40.82	13.45	3.0	2	41
SuperP	30.00	40.82	9.74	4.2	3	41
SuperP	30.00	40.82	6.70	6.1	4	41
SuperP	30.00	40.82	3.38	12.1	8	41
CNT	7.50	87.50	10.00	8.8	7	42
CNT	7.50	87.50	4.50	19.4	40	42
CNT	7.50	87.50	3.00	29.2	45	42
MoS2	0.40	52.00	1.00	52.0	190	43
CNT	6.66	83.30	2.00	41.7	23	44

Table S1. The performance of LOBs with low E/C conditions (E/C < 20 g/Ah) investigated in the present study and reported in the literature.