

# Supporting Information for: Boron-based non-fullerene small molecule acceptors via nitrogen substitutions: a theoretical study

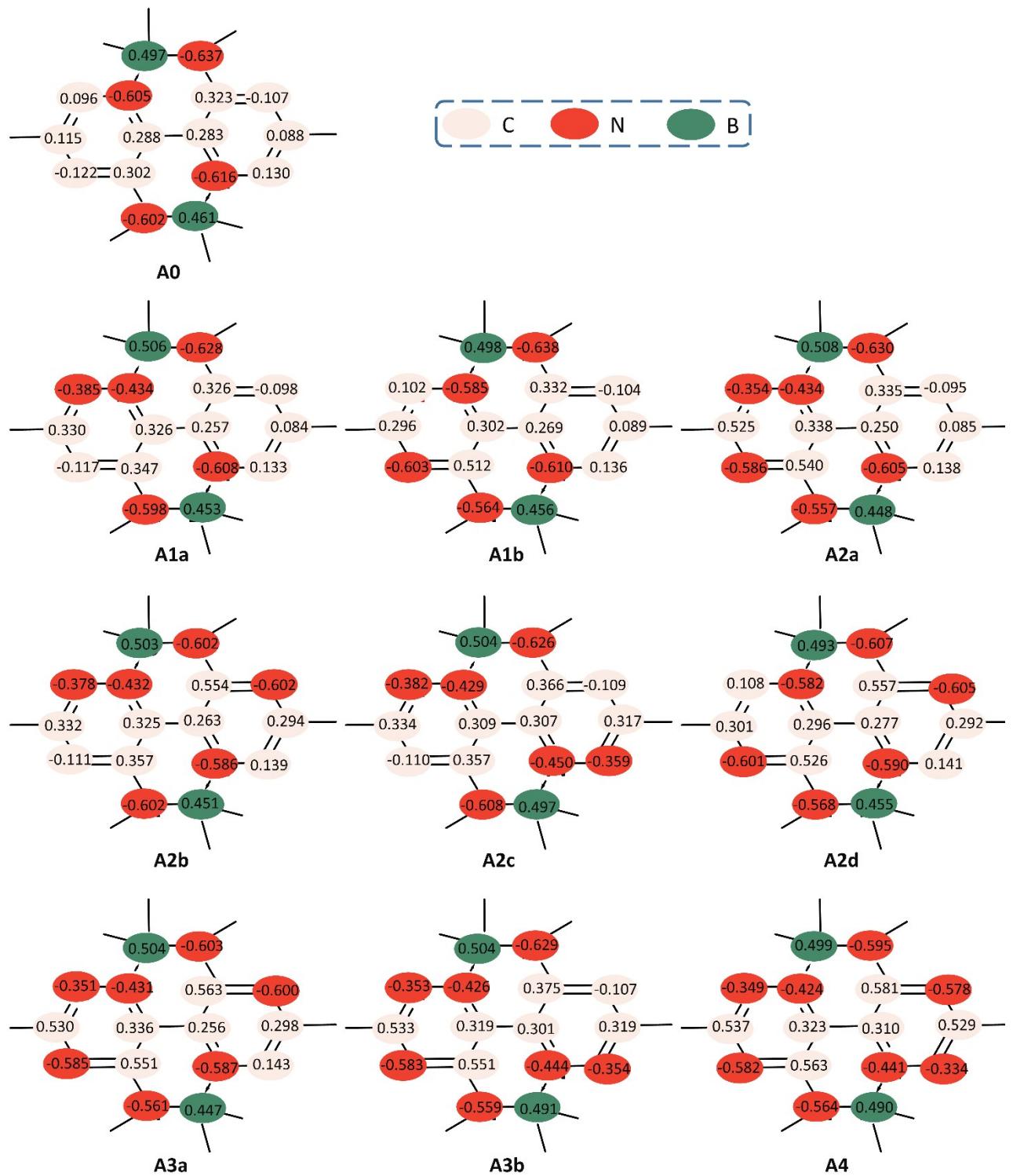
Jie Yang, Xun Wu, Quan-Song Li\*, and Ze-Sheng Li\*

Key Laboratory of Cluster Science of Ministry of Education, Beijing Key Laboratory of Photoelectronic/Electrophotonic Conversion Materials, School of Chemistry and chemical engineering, Beijing Institute of Technology, Beijing 100081, China.

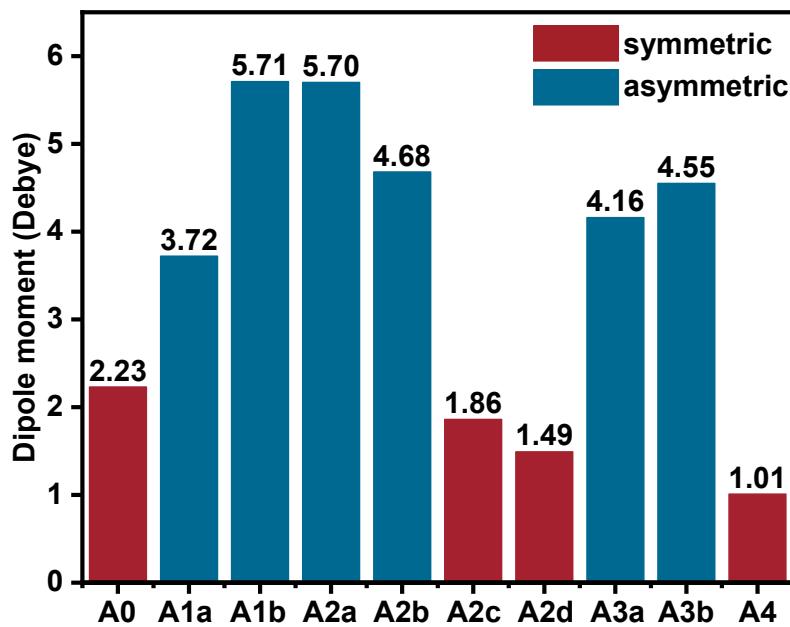
E-mail: liquansong@bit.edu.cn, zeshengli@bit.edu.cn

## Contents

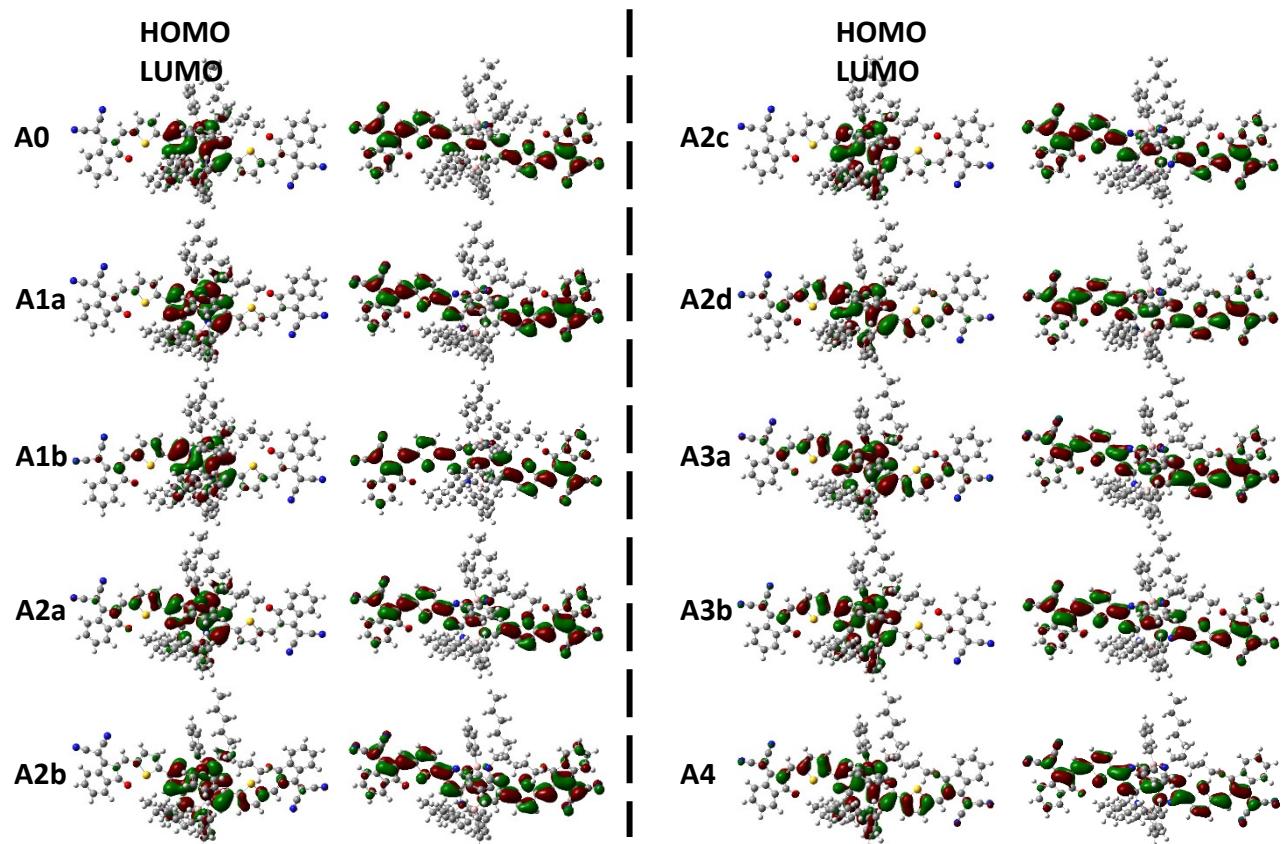
<b>Fig. S1</b> The Mulliken atom charge of core in studied acceptors.....	SI1
<b>Fig. S2</b> The dipole moments of the studied molecules.....	SI2
<b>Fig. S3</b> The contour plots of the frontier molecular orbitals.....	SI3
<b>Fig. S4</b> The comparison between A0 and its carbon-substituted A0-C.....	SI4
<b>Fig. S5</b> The detailed information for the primary transfer pathways.....	SI5
<b>Fig. S6</b> The detailed packing information for the most favoured hopping pathway.....	SI6
<b>Fig. S7</b> The optimized structures of the PTB7-Th/acceptors.....	SI7
<b>Fig. S8</b> The calculated absorption spectra of PTB7-Th/acceptors.....	SI8
<b>Fig. S9</b> The types of the excited states of the studied two interfaces.....	SI9
<b>Fig. S10</b> The electrostatic potential of the A0 and A2d.....	SI10
<b>Table S1</b> The structural information of the studied acceptors.....	SI11
<b>Table S2</b> The binding energy of the studied acceptors.....	SI12
<b>Table S3</b> The excitation information of the studied interfaces.....	SI13



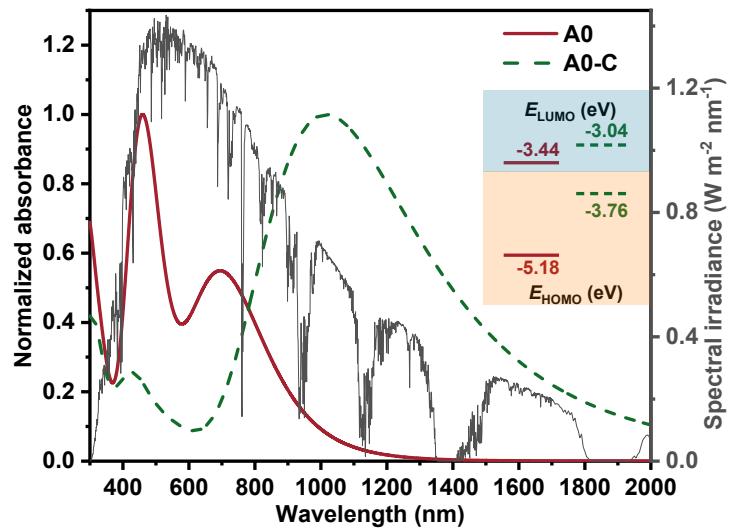
**Fig. S1** The Mulliken atom charge of core in studied acceptors.



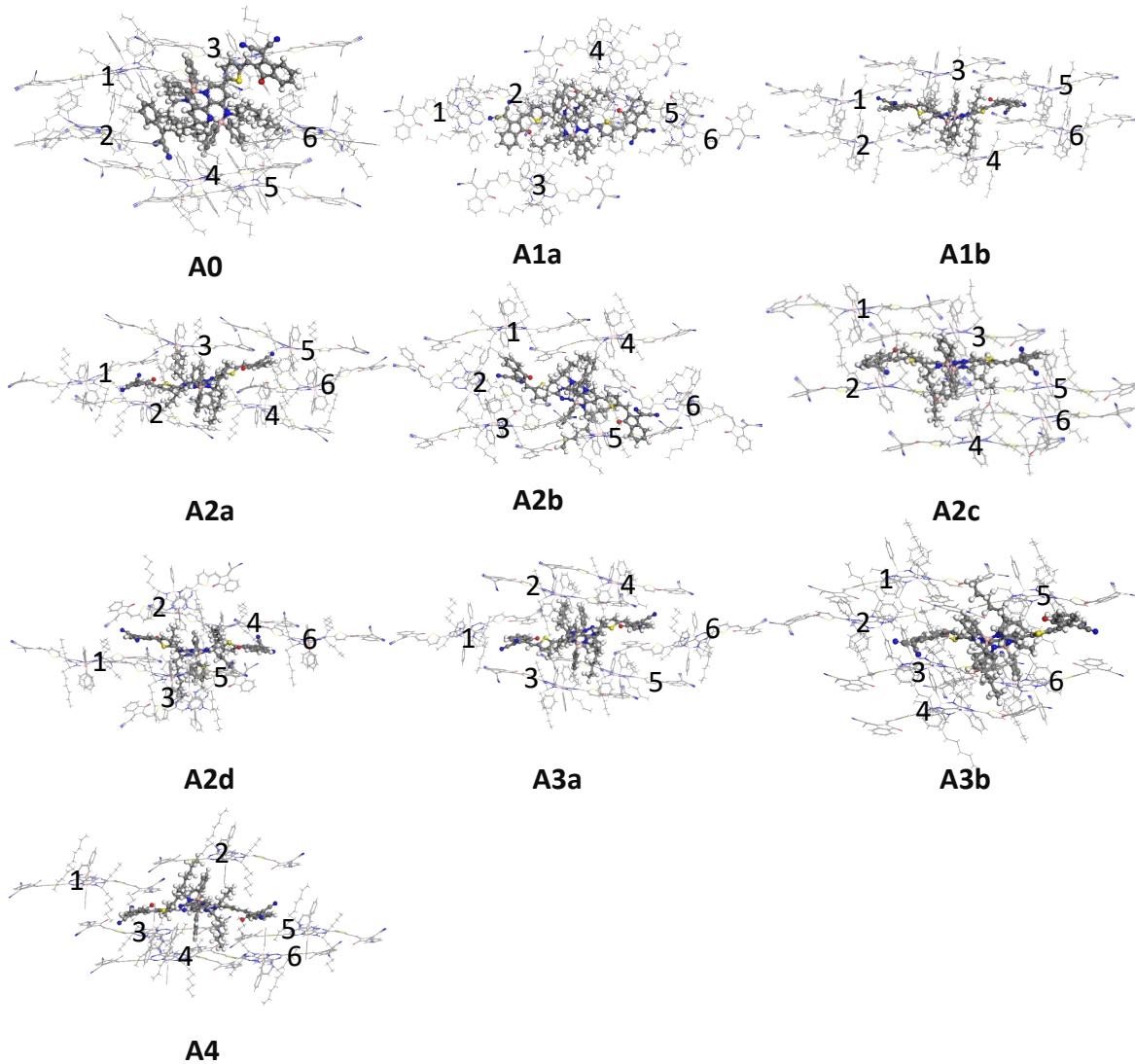
**Fig. S2** The dipole moments of studied molecules (the red bars stand for the symmetric molecules and the blue ones mean the asymmetric molecules).



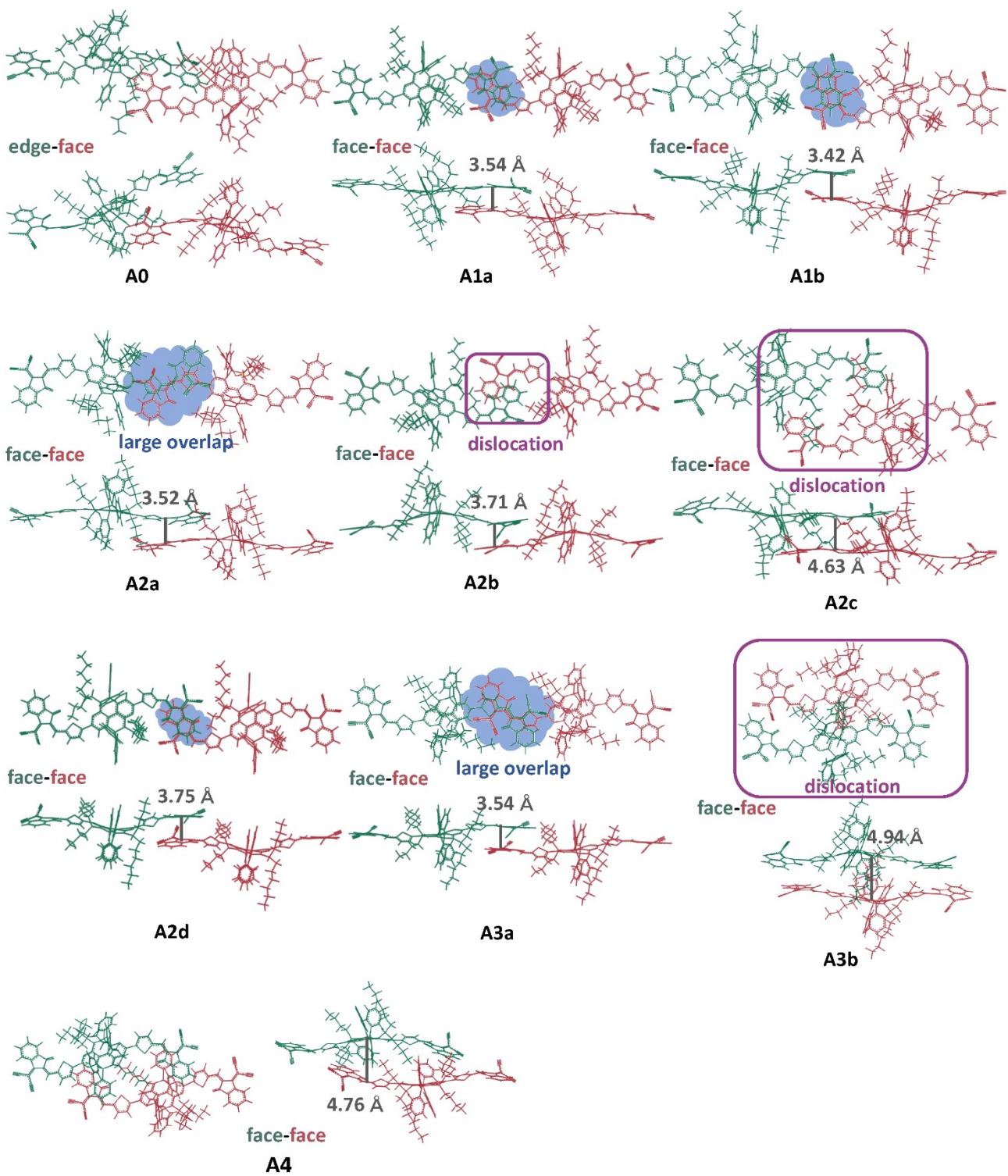
**Fig. S3** The contour plots of the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) for studied acceptors.



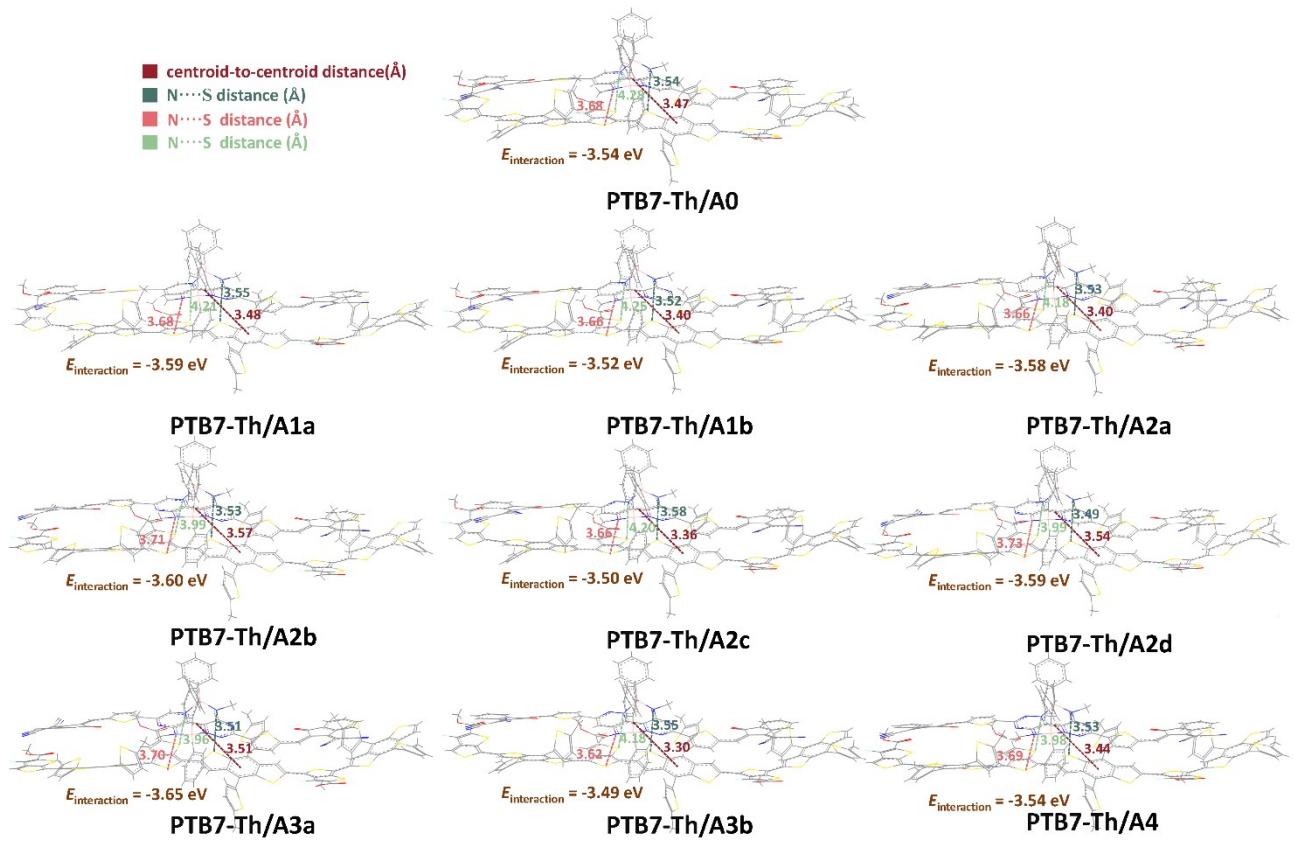
**Fig. S4** The absorption spectra and frontier molecular orbitals of A0 (in red solid line) and its carbon counterpart A0-C (in green dash line) together with the sunlight irradiation spectrum (in grey solid line).



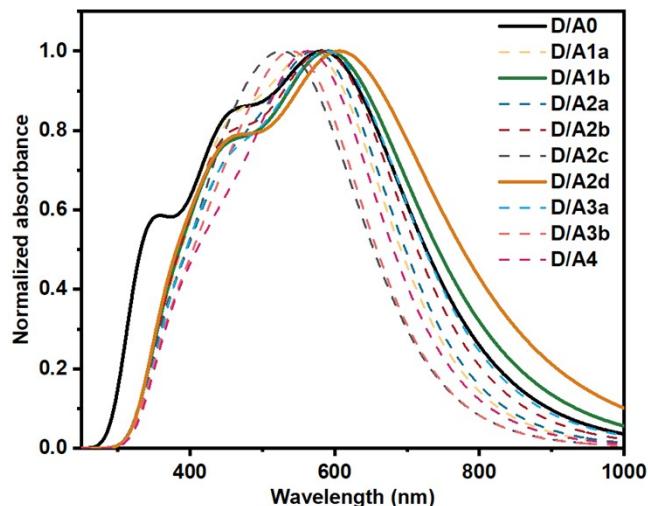
**Fig. S5** The detailed information for the primary transfer pathways (marked by numbers) in the predicted crystal structures of the studied molecules.



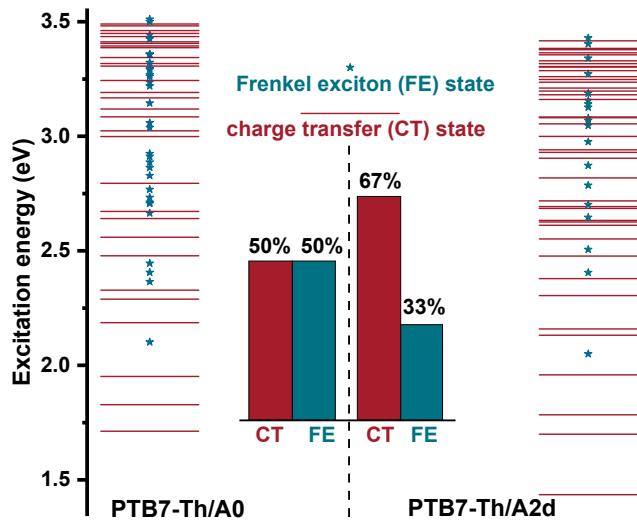
**Fig. S6** The detailed packing information for the most favoured hopping pathway of acceptors.



**Fig. S7** The optimized structures of the PTB7-Th/acceptors with the centroid-to-centroid distances, N···S distances, and the weak interaction energy ( $E_{\text{interaction}}$ ).

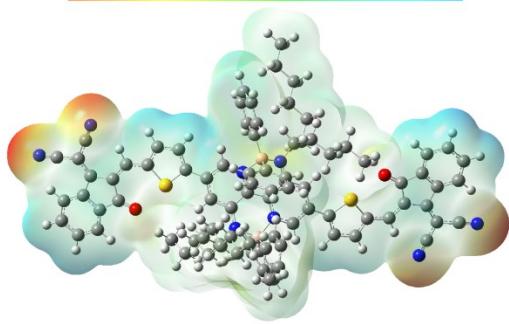


**Fig. S8** The calculated absorption spectra of the studied donor/acceptor (D/A) interfaces, composed by the donor PTB7-Th and the studied acceptors.

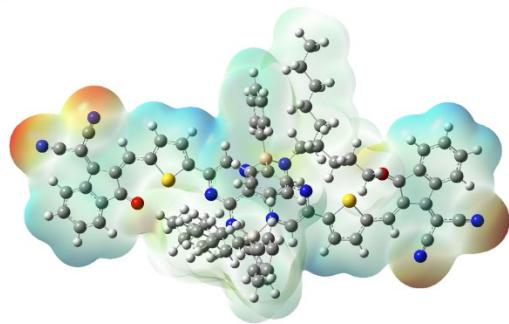


**Fig. S9** The types (charge transfer state: CT, Frenkerl exciton state: FE) of the lowest sixty excited states and the corresponding percentage in the PTB7-Th/A0 and PTB7-Th/A2d.

-40 kcal/mol  40 kcal/mol



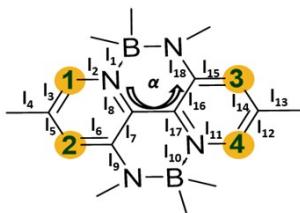
A0: average ESP = 3 kcal/mol



A2d: average ESP = 3 kcal/mol

**Fig. S10** The electrostatic potential (ESP) maps and average ESP of the A0 and A2d.

**Table S1** The calculated bond length  $l_n$  ( $\text{\AA}$ ) and dihedral angle  $\alpha$  ( $^\circ$ ) in molecules.



acceptor	A0	A1a	A1b	A2a	A2b	A2c	A2d	A3a	A3b	A4
$l_1$	1.63	1.62	1.63	1.62	1.63	1.63	1.64	1.63	1.63	1.63
$l_2$	1.34	1.32	1.34	1.33	1.32	1.32	1.34	1.33	1.33	1.33
$l_3$	1.40	1.35	1.40	1.34	1.35	1.35	1.40	1.34	1.34	1.35
$l_4$	1.46	1.46	1.45	1.46	1.46	1.46	1.45	1.46	1.46	1.45
$l_5$	1.39	1.39	1.34	1.35	1.39	1.38	1.34	1.33	1.33	1.33
$l_6$	1.41	1.41	1.35	1.35	1.41	1.41	1.35	1.35	1.35	1.35
$l_7$	1.44	1.44	1.45	1.45	1.44	1.44	1.45	1.45	1.45	1.45
$l_8$	1.36	1.35	1.35	1.34	1.35	1.35	1.35	1.34	1.33	1.33
$l_9$	1.35	1.34	1.33	1.32	1.33	1.33	1.33	1.32	1.32	1.32
$l_{10}$	1.63	1.64	1.64	1.64	1.63	1.63	1.64	1.64	1.63	1.62
$l_{11}$	1.34	1.33	1.33	1.33	1.34	1.32	1.34	1.34	1.31	1.33
$l_{12}$	1.40	1.40	1.40	1.40	1.40	1.34	1.40	1.40	1.35	1.34
$l_{13}$	1.46	1.46	1.46	1.46	1.45	1.46	1.45	1.45	1.46	1.46
$l_{14}$	1.39	1.39	1.39	1.39	1.34	1.39	1.34	1.33	1.39	1.33
$l_{15}$	1.39	1.41	1.41	1.41	1.35	1.41	1.34	1.35	1.41	1.35
$l_{16}$	1.45	1.44	1.44	1.44	1.45	1.45	1.45	1.45	1.45	1.45
$l_{17}$	1.36	1.36	1.36	1.36	1.35	1.34	1.35	1.35	1.34	1.33
$l_{18}$	1.35	1.35	1.35	1.35	1.34	1.34	1.34	1.33	1.34	1.32
$ \alpha $	10.82	12.71	13.07	14.24	11.86	12.10	12.49	12.92	13.81	11.87

**Table S2** The calculated vertical ionization potential energy (VIP), vertical electron affinity energy (VEA), first vertical excitation energy ( $E_1$ ), and exciton binding energy ( $E_b$ ).

acceptor	VIP (eV)	VEA (eV)	$E_1$ (eV)	$E_b$ (eV)
A0	5.21	3.42	1.41	0.38
A1a	5.53	3.43	1.68	0.42
A1b	5.36	3.50	1.47	0.40
A2a	5.69	3.51	1.74	0.44
A2b	5.66	3.54	1.66	0.45
A2c	5.83	3.46	1.90	0.47
A2d	5.47	3.62	1.45	0.40
A3a	5.79	3.65	1.68	0.46
A3b	6.00	3.56	1.93	0.51
A4	6.11	3.72	1.88	0.51

**Table S3** The calculated the type of exciton, excited energy ( $E$ ), oscillator strength  $f$ , transferred charge  $|\Delta q|$ , and D index of the PTB7-Th/A0 and PTB7-Th/A2d interfaces.

PTB7-Th/A0						PTB7-Th/A2d					
$S_n$	nature	$E$ (eV)	$f$	$ \Delta q $ (e)	D (Å)	$S_n$	nature	$E$ (eV)	$f$	$ \Delta q $ (e)	D (Å)
$S_1$	$CT_1$	1.71	0.04	0.78	2.37	$S_1$	$CT_1$	1.44	0.05	0.89	2.88
$S_2$	$CT_2$	1.83	0.74	0.31	0.75	$S_2$	$CT_2$	1.70	0.82	0.40	1.18
$S_3$	$CT_3$	1.95	0.06	0.81	2.58	$S_3$	$CT_3$	1.78	0.17	0.65	1.85
$S_4$	$FE_1$	2.10	2.45	0.01	0.58	$S_4$	$CT_4$	1.96	0.05	0.91	3.41
$S_5$	$CT_4$	2.19	0.01	0.93	3.07	$S_5$	$FE_1$	2.05	2.10	0.19	1.25
$S_6$	$CT_5$	2.29	0.02	0.48	2.39	$S_6$	$CT_5$	2.13	0.09	0.91	2.92
$S_7$	$CT_6$	2.33	0.06	0.46	2.95	$S_7$	$CT_6$	2.16	0.24	0.84	2.38
$S_8$	$FE_2$	2.37	0.02	0.35	1.70	$S_8$	$CT_7$	2.30	0.06	0.59	2.10
$S_9$	$FE_3$	2.41	0.06	0.48	3.16	$S_9$	$FE_2$	2.38	0.06	0.35	1.05
$S_{10}$	$FE_4$	2.45	0.17	0.09	0.69	$S_{10}$	$CT_8$	2.40	0.02	0.86	4.12
$S_{11}$	$CT_7$	2.48	0.05	0.87	2.83	$S_{11}$	$FE_3$	2.48	0.02	0.09	0.51
$S_{12}$	$CT_8$	2.56	0.05	0.73	2.32	$S_{12}$	$CT_9$	2.51	0.02	0.87	3.64
$S_{13}$	$CT_9$	2.64	0.04	0.74	5.48	$S_{13}$	$CT_{10}$	2.55	0.02	0.80	3.70
$S_{14}$	$FE_5$	2.66	0.16	0.21	1.98	$S_{14}$	$CT_{11}$	2.61	0.34	0.24	1.11
$S_{15}$	$CT_{10}$	2.67	0.02	0.78	2.02	$S_{15}$	$CT_{12}$	2.63	0.06	0.81	2.40
$S_{16}$	$FE_6$	2.71	0.62	0.06	1.14	$S_{16}$	$FE_4$	2.63	0.18	0.26	0.59
$S_{17}$	$FE_7$	2.71	0.59	0.33	1.03	$S_{17}$	$CT_{13}$	2.65	0.15	0.56	2.06
$S_{18}$	$FE_8$	2.73	0.26	0.11	0.67	$S_{18}$	$CT_{14}$	2.69	0.36	0.38	1.81
$S_{19}$	$FE_9$	2.77	0.08	0.09	0.33	$S_{19}$	$CT_{15}$	2.69	0.30	0.64	1.83
$S_{20}$	$CT_{11}$	2.79	0.05	0.76	5.04	$S_{20}$	$FE_5$	2.70	0.07	0.28	0.98
$S_{21}$	$FE_{10}$	2.83	0.12	0.61	2.10	$S_{21}$	$CT_{16}$	2.72	0.09	0.26	1.59
$S_{22}$	$FE_{11}$	2.86	0.01	0.24	3.54	$S_{22}$	$FE_6$	2.79	0.01	0.14	0.44
$S_{23}$	$FE_{12}$	2.89	0.01	0.18	0.64	$S_{23}$	$CT_{17}$	2.82	0.20	0.43	2.03
$S_{24}$	$FE_{13}$	2.91	0.05	0.43	1.26	$S_{24}$	$FE_7$	2.87	0.01	0.08	0.52
$S_{25}$	$FE_{14}$	2.92	0.15	0.17	0.31	$S_{25}$	$CT_{18}$	2.90	0.13	0.51	1.54
$S_{26}$	$CT_{12}$	3.00	0.03	0.58	2.01	$S_{26}$	$CT_{19}$	2.93	0.04	0.73	4.24
$S_{27}$	$CT_{13}$	3.02	0.00	0.62	1.63	$S_{27}$	$CT_{20}$	2.94	0.05	0.69	3.88
$S_{28}$	$FE_{15}$	3.04	0.27	0.04	0.93	$S_{28}$	$FE_8$	2.98	0.01	0.00	3.53
$S_{29}$	$FE_{16}$	3.06	0.11	0.14	1.77	$S_{29}$	$CT_{21}$	3.00	0.14	0.51	1.78
$S_{30}$	$CT_{14}$	3.09	0.01	0.49	1.30	$S_{30}$	$FE_9$	3.05	0.11	0.39	1.05
$S_{31}$	$CT_{15}$	3.12	0.01	0.62	8.28	$S_{31}$	$FE_{10}$	3.05	0.05	0.06	0.86

$S_{32}$	$FE_{17}$	3.14	0.01	0.14	2.07	$S_{32}$	$CT_{22}$	3.07	0.02	0.26	2.81
$S_{33}$	$CT_{16}$	3.17	0.01	0.32	5.39	$S_{33}$	$FE_{11}$	3.08	0.02	0.17	0.73
$S_{34}$	$CT_{17}$	3.19	0.00	0.57	2.40	$S_{34}$	$FE_{12}$	3.08	0.09	0.48	1.65
$S_{35}$	$FE_{18}$	3.22	0.02	0.04	3.04	$S_{35}$	$CT_{23}$	3.08	0.01	0.69	2.13
$S_{36}$	$FE_{19}$	3.24	0.02	0.17	1.30	$S_{36}$	$FE_{13}$	3.13	0.02	0.29	1.32
$S_{37}$	$CT_{18}$	3.24	0.01	0.42	2.70	$S_{37}$	$FE_{14}$	3.14	0.01	0.53	1.44
$S_{38}$	$FE_{20}$	3.26	0.01	0.03	1.22	$S_{38}$	$FE_{15}$	3.16	0.01	0.08	0.79
$S_{39}$	$FE_{21}$	3.27	0.00	0.03	2.36	$S_{39}$	$CT_{24}$	3.16	0.00	0.16	2.47
$S_{40}$	$FE_{22}$	3.29	0.01	0.37	0.86	$S_{40}$	$CT_{25}$	3.18	0.01	0.66	2.80
$S_{41}$	$FE_{23}$	3.30	0.03	0.31	2.56	$S_{41}$	$FE_{16}$	3.19	0.03	0.31	4.67
$S_{42}$	$CT_{19}$	3.31	0.00	0.25	1.39	$S_{42}$	$CT_{26}$	3.20	0.00	0.22	2.41
$S_{43}$	$CT_{20}$	3.32	0.01	0.49	1.22	$S_{43}$	$CT_{27}$	3.21	0.00	0.41	2.79
$S_{44}$	$FE_{24}$	3.32	0.01	0.27	2.68	$S_{44}$	$CT_{28}$	3.24	0.00	0.37	2.30
$S_{45}$	$CT_{21}$	3.34	0.00	0.76	2.61	$S_{45}$	$CT_{29}$	3.25	0.01	0.51	1.65
$S_{46}$	$FE_{25}$	3.36	0.35	0.18	0.61	$S_{46}$	$CT_{30}$	3.26	0.03	0.31	1.78
$S_{47}$	$FE_{26}$	3.36	0.06	0.47	4.04	$S_{47}$	$FE_{17}$	3.27	0.05	0.46	1.00
$S_{48}$	$CT_{22}$	3.39	0.01	0.35	1.12	$S_{48}$	$CT_{31}$	3.29	0.01	0.65	4.42
$S_{49}$	$CT_{23}$	3.39	0.08	0.76	1.98	$S_{49}$	$CT_{32}$	3.30	0.02	0.66	2.05
$S_{50}$	$CT_{24}$	3.41	0.04	0.41	1.44	$S_{50}$	$CT_{33}$	3.30	0.03	0.73	2.28
$S_{51}$	$CT_{25}$	3.41	0.06	0.60	1.50	$S_{51}$	$CT_{34}$	3.32	0.00	0.49	2.84
$S_{52}$	$FE_{27}$	3.42	0.02	0.12	0.94	$S_{52}$	$CT_{35}$	3.33	0.10	0.61	2.10
$S_{53}$	$CT_{26}$	3.44	0.02	0.49	0.43	$S_{53}$	$FE_{18}$	3.34	0.16	0.30	1.61
$S_{54}$	$FE_{28}$	3.44	0.02	0.32	0.93	$S_{54}$	$CT_{36}$	3.36	0.19	0.50	3.17
$S_{55}$	$CT_{27}$	3.45	0.01	0.78	5.29	$S_{55}$	$CT_{37}$	3.36	0.03	0.38	2.03
$S_{56}$	$CT_{28}$	3.46	0.10	0.42	1.23	$S_{56}$	$CT_{38}$	3.38	0.05	0.34	1.00
$S_{57}$	$CT_{29}$	3.48	0.01	0.50	1.67	$S_{57}$	$CT_{39}$	3.38	0.07	0.14	3.10
$S_{58}$	$CT_{30}$	3.49	0.01	0.16	2.39	$S_{58}$	$FE_{19}$	3.40	0.07	0.29	1.73
$S_{59}$	$FE_{29}$	3.50	0.03	0.30	1.91	$S_{59}$	$CT_{40}$	3.42	0.02	0.31	1.19
$S_{60}$	$FE_{30}$	3.51	0.02	0.41	1.70	$S_{60}$	$FE_{20}$	3.43	0.04	0.09	0.78