Electronic Supplementary Information for:

The characterisation of commercial 2D carbons: graphene, graphene oxide and reduced graphene oxide.

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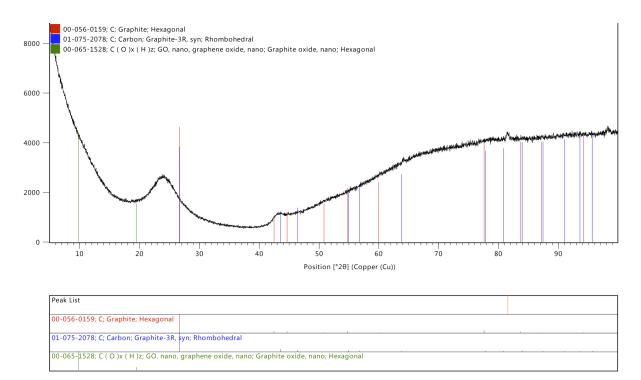


Fig. S1. XRD fit for 1_GO

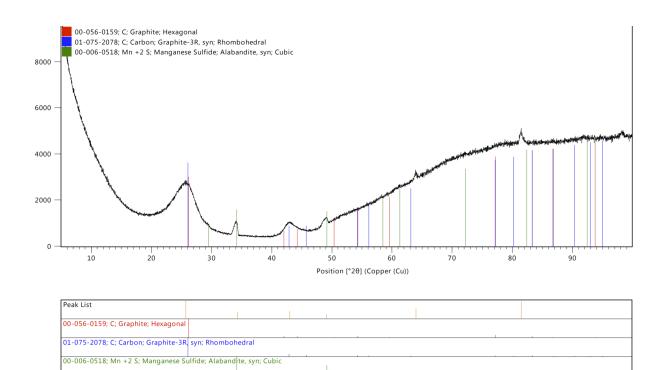


Fig. S2. XRD fit for 2_rGO

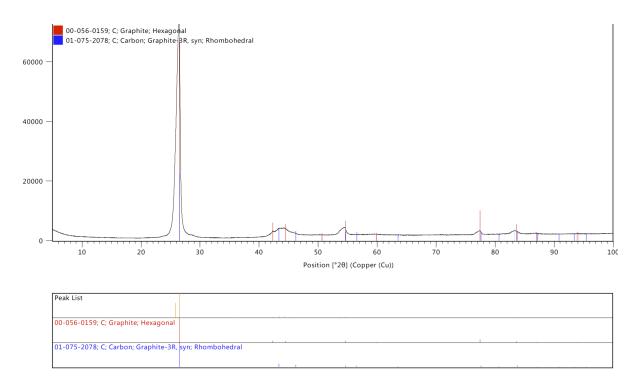


Fig. S3. XRD fit for 3_GNP

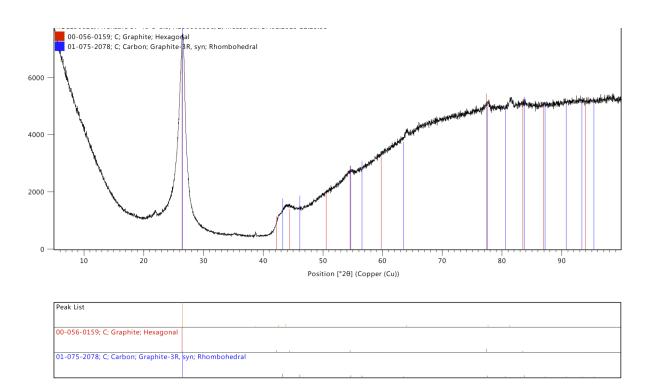


Fig. S4. XRD fit for 4_rGO

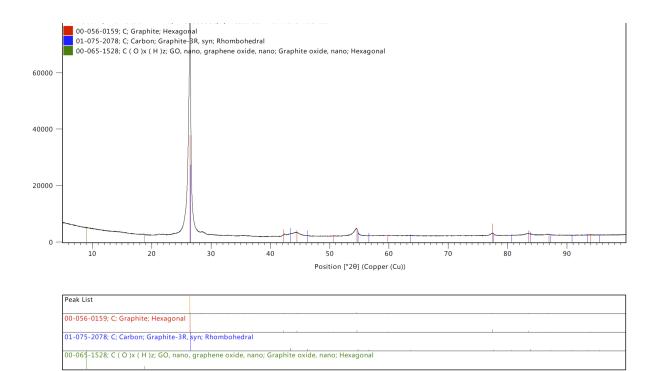


Fig. S5. XRD fit for 5_GNP

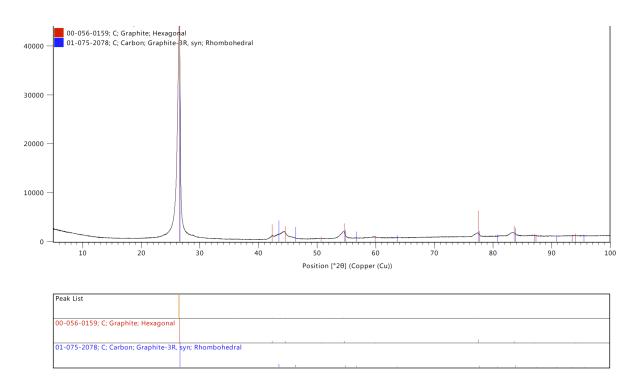


Fig. S6. XRD fit for 6_GNP

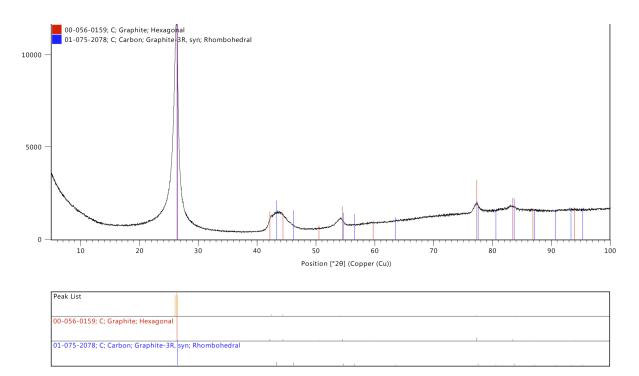


Fig. S7. XRD fit for 7_GO

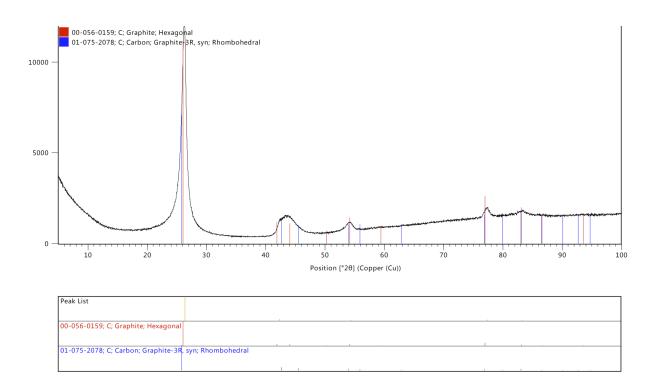


Fig. S8. XRD fit for 8_rGO

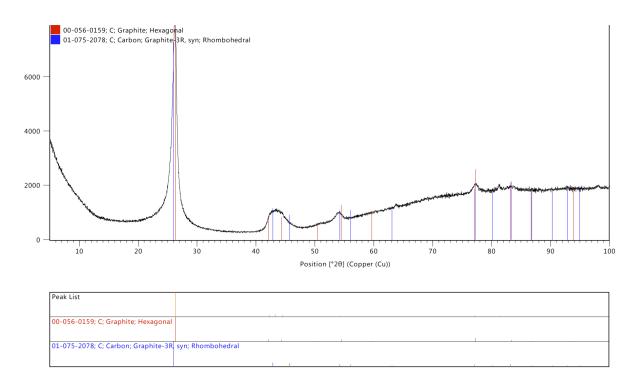


Fig. S9. XRD fit for 9_GNP

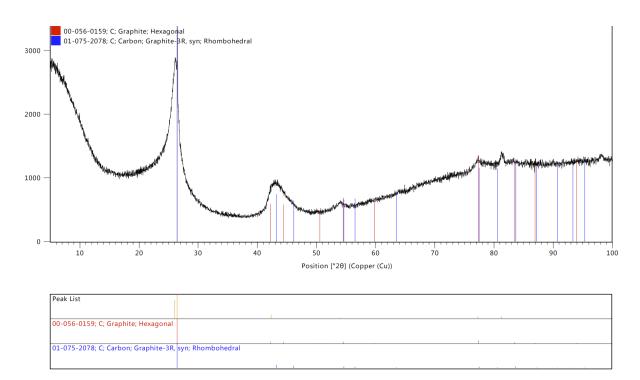


Fig. S10. XRD fit for 10_GO

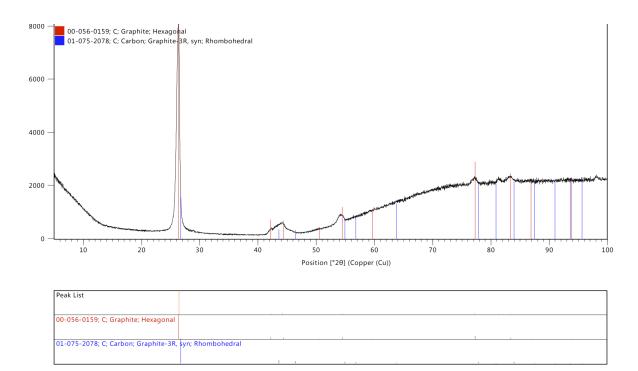


Fig. S11. XRD fit for 11_GNP

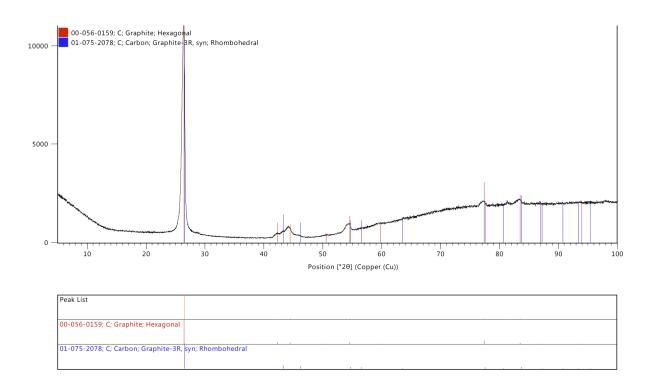


Fig. S12. XRD fit for 12_GNP

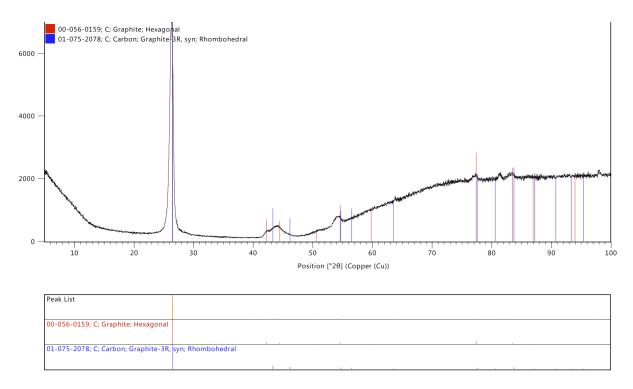
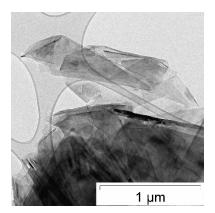
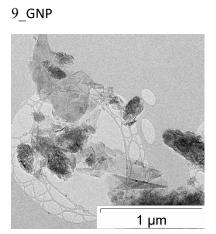


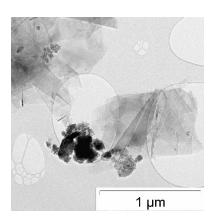
Fig. S13. XRD fit for 13_GNP



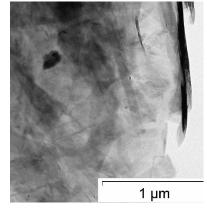


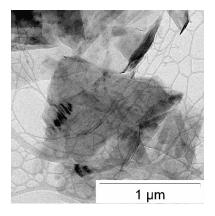
13_GNP

1 µm

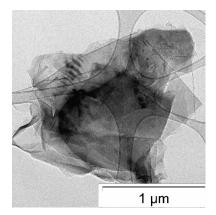


11_GNP





12_GNP



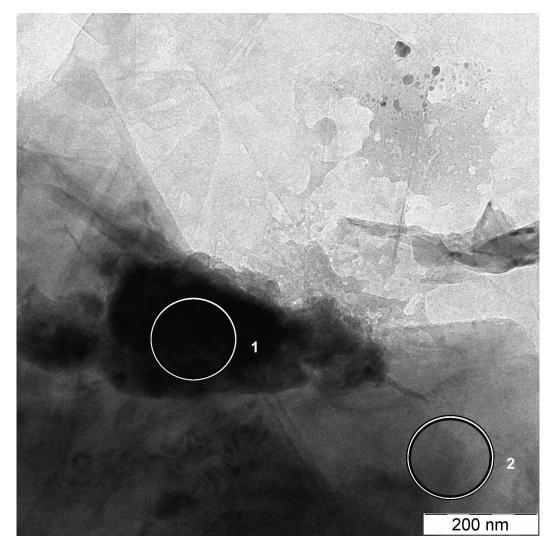
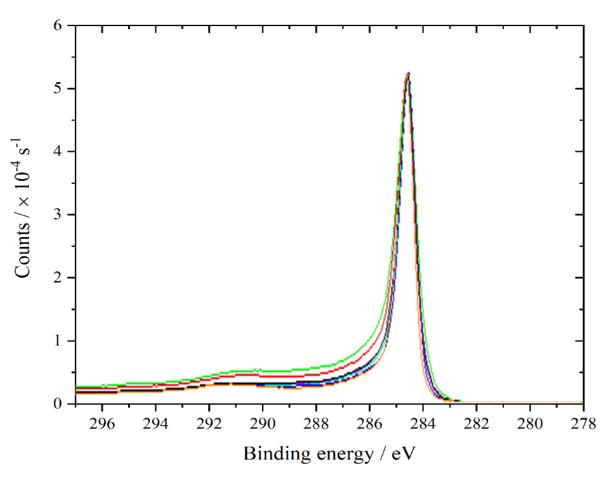


Fig. S15. 200 keV TEM image of a 2D carbon.

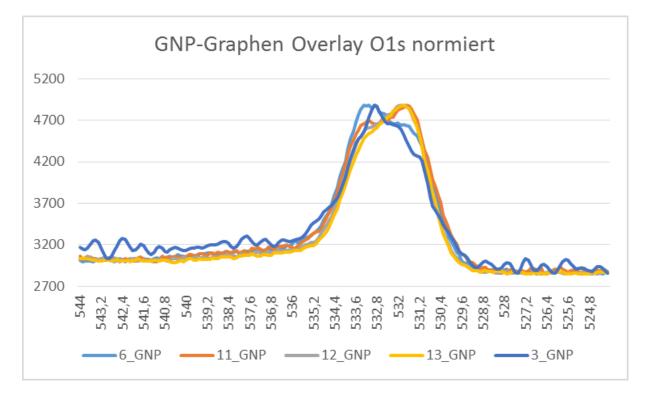
Table S1

EDX spot analyses in the TEM Jeol 2010F a	areas analysed marked by circles in Fig. S2.

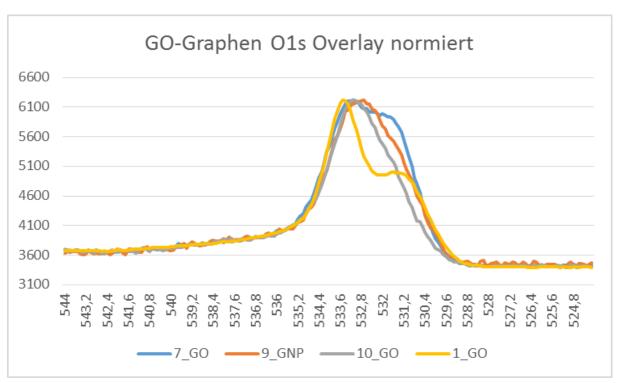
1 •		J	5	8	
	А	rea 1	Area	n 2	
Element	Wt.%	At.%	Wt.%	At.%	
С	62.18	79.23	88.75	92.90	
0	10.83	10.36	6.83	5.36	
Na	0.86	0.57	0.33	0.18	
Mg	0.38	0.24			
Al	1.72	0.98			
Si	5.56	3.03	0.14	0.06	
S	0.50	0.24	3.11	1.22	
Cl	0.46	0.20	0.48	0.17	
Κ	0.32	0.12	0.14	0.04	
Ca	2.83	1.08			
Cr	1.63	0.48			
Mn	0.27	0.08			
Fe	11.13	3.05	0.21	0.05	
Ni	1.33	0.35			



В







D

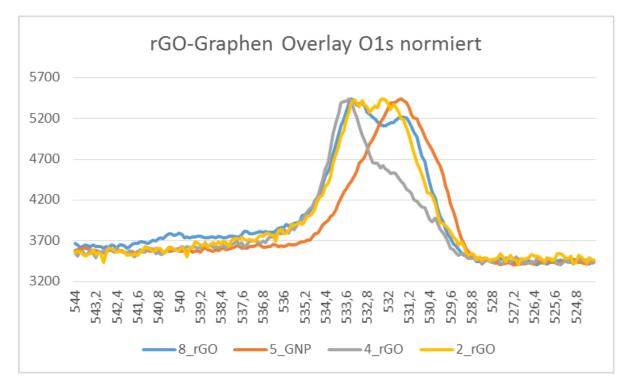


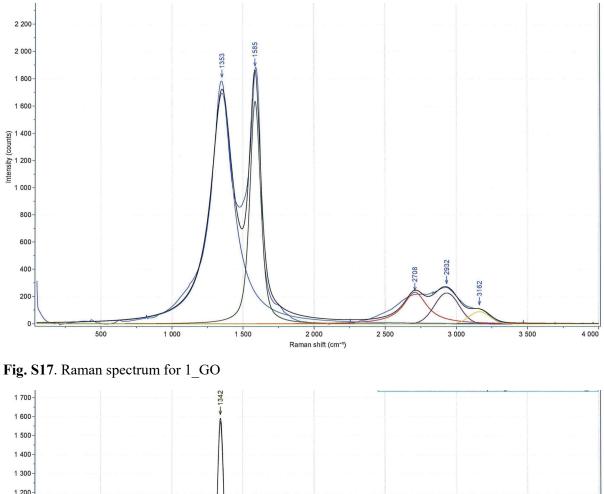
Fig. S16. A: Representative O1s XPS spectra of the GNP samples. B-D: Normalised to the largest peak, O1s XPS spectra of the 2D carbon samples.

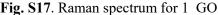
Table S2

Previous data [1] on 3D carbons, spherical aggregates/agglomerates of commercial carbon blacks before and after oxidizing aftertreatment. In a separate oxidation experiment on gb original, after moderate post-oxidation with O₃, XPS revealed an increase in surface-oxygen from 1.45 to 13.5%.

Carbon black, oxidation treatment ^a	H / wt.%	BET / m ² g ⁻¹	XPS O / at.%	XPS O1s C–OH/C=O
ac original	0.018	40	0.04	?
lb original	0.265	20	0.3	2.2
fb original	0.33	120	0.7	6.6
fb HNO ₃	0.26	ca. 140	9.8	2.2
gb original	0.37	360	1.45	1.4
gb NO ₂	0.35	ca. 460	8.5	2.7
gb NO ₂	0.38	"	9.1	2.2
gb NO ₂	0.44	"	11.8	2.2
gb HNO ₃	0.32	"	10.5	4.3
gb O ₃	0.53	"	19.3	1.6

^{*a*}fb: furnace black, gb: gas black, lb: lamp black (all Degussa-Hüls), ac: acetylene black (Denka Black).





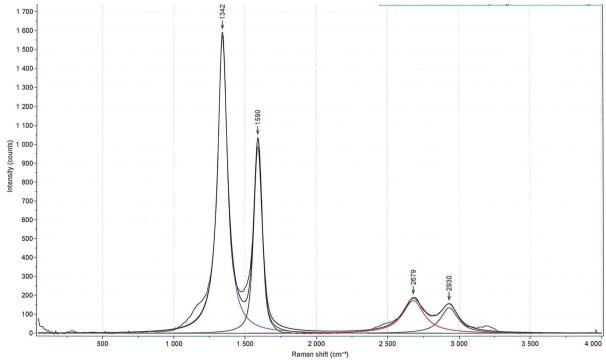


Fig. S18. Raman spectrum for 2_rGO

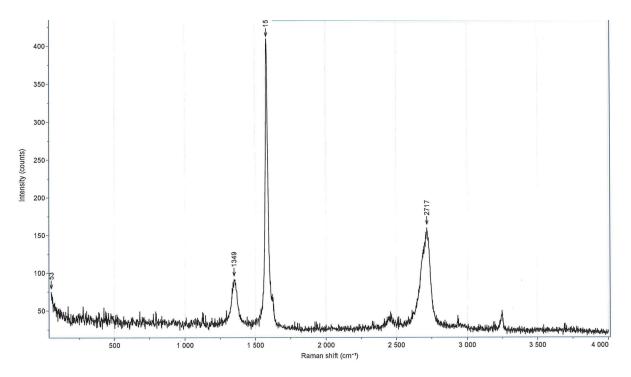


Fig. S19. Raman spectrum for 3_GNP

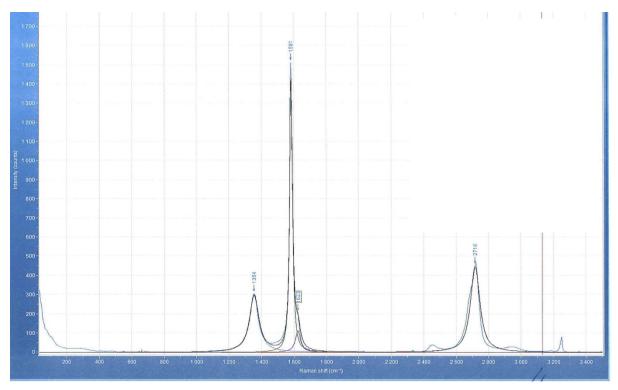


Fig. S20. Raman spectrum for 4_rGO. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

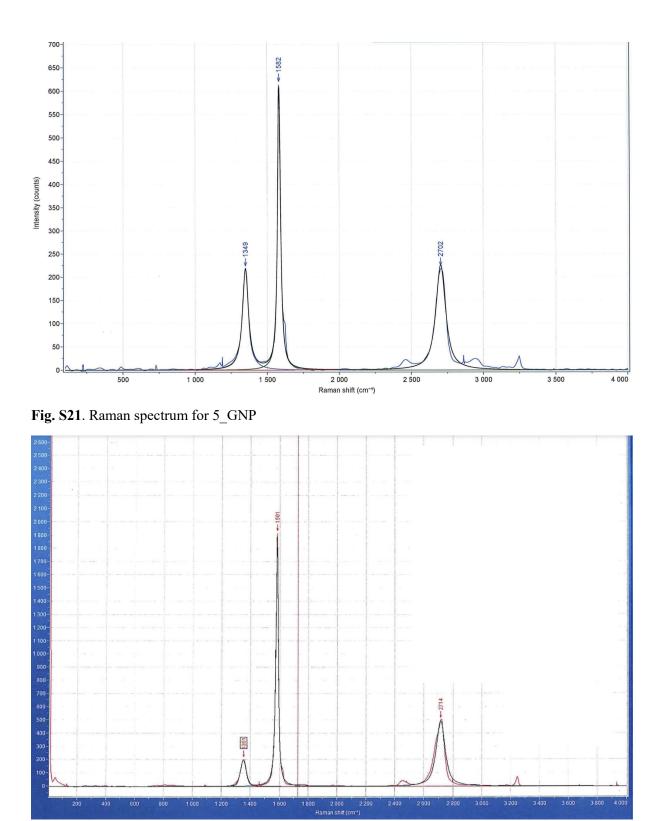


Fig. S22. Raman spectrum for 6_GNP. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

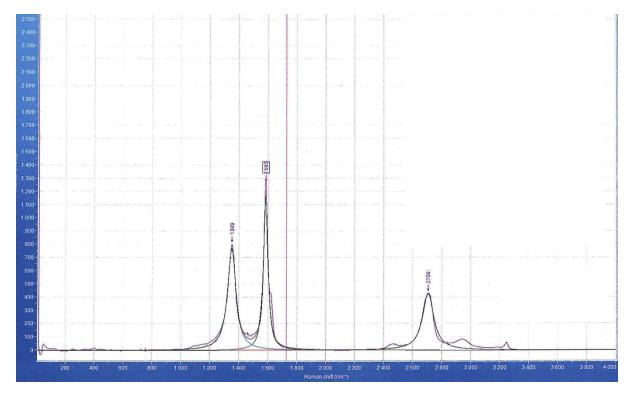


Fig. S23. Raman spectrum for 7_GO. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

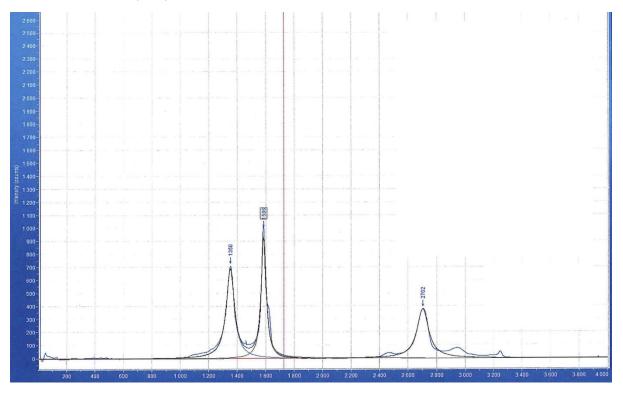


Fig. S24. Raman spectrum for 8_rGO. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

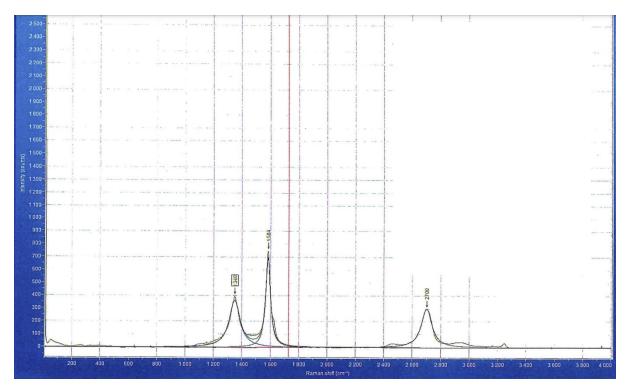


Fig. S25. Raman spectrum for 9_GNP. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

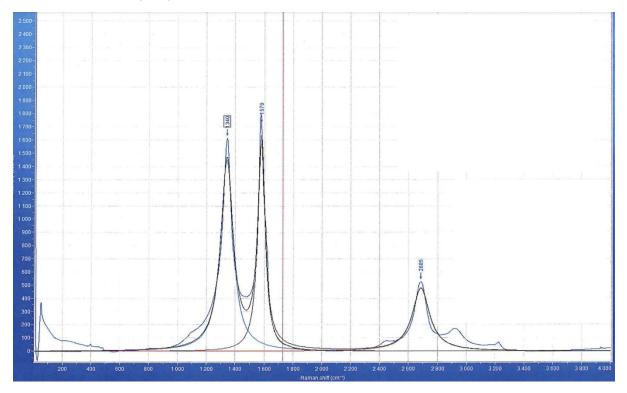


Fig. S26. Raman spectrum for 10_GO. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

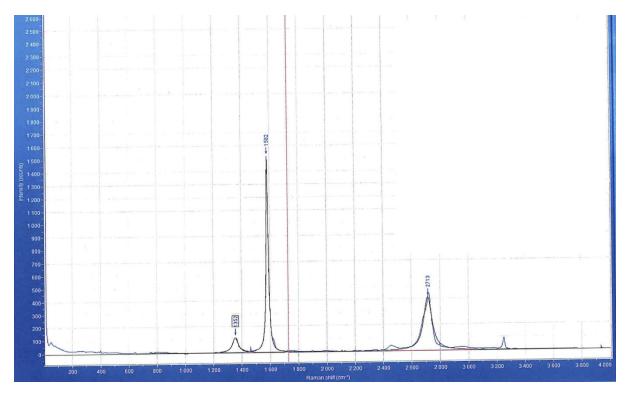


Fig. S27. Raman spectrum for 11_GNP. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

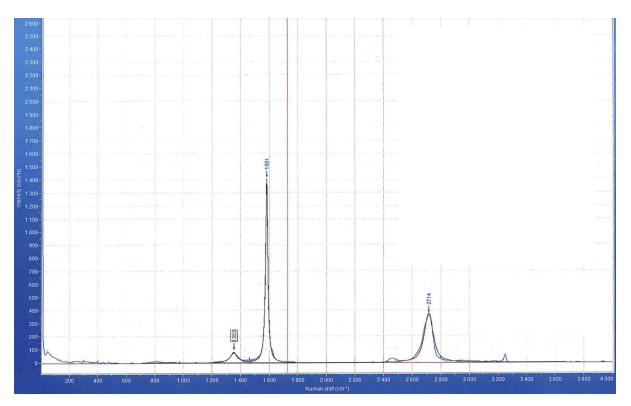


Fig. S28. Raman spectrum for 12_GNP. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

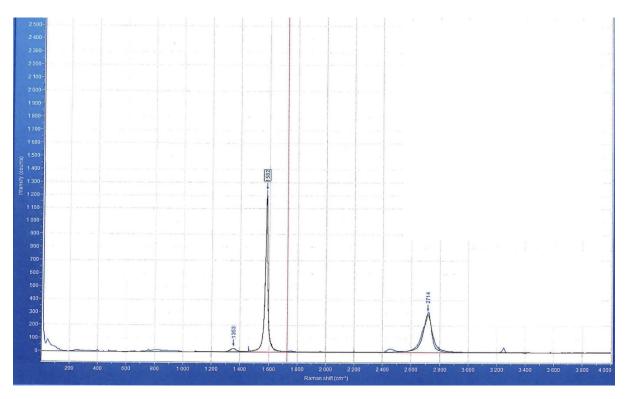


Fig. S29. Raman spectrum for 13_GNP. Note: annotations intended for in-house use have been removed. We emphasise that it is only the annotations that were removed, the spectra themselves have not been altered in any way.

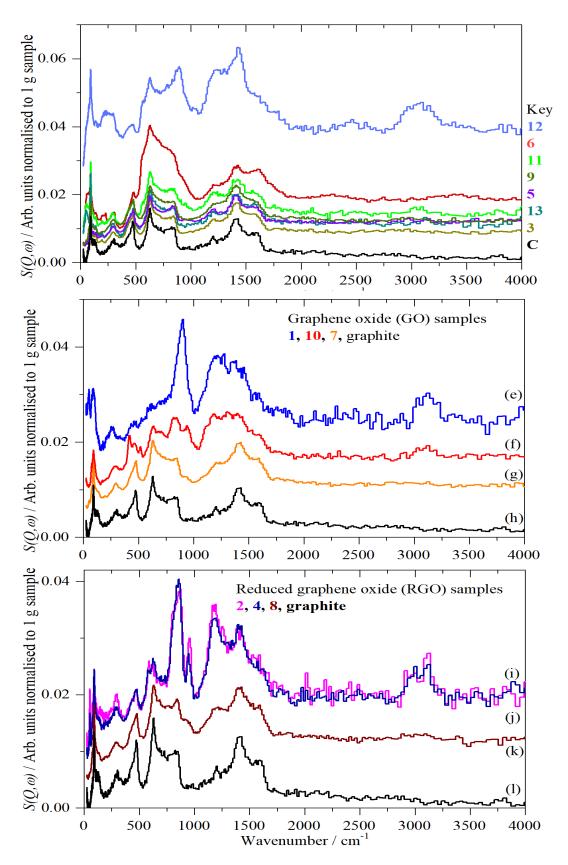


Fig. S30. Full range (0-4000 cm-1) TOSCA INS spectra of the 2D carbon samples. In each panel the bottom-most spectrum is a reference spectrum of graphite. Top panel: graphene samples. Middle panel: graphene oxide samples. Bottom panel: reduced graphene oxide samples.

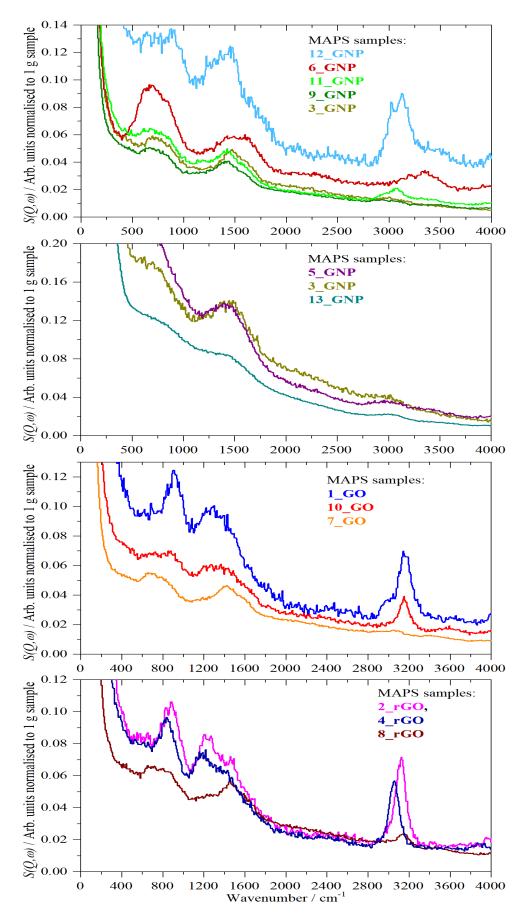


Fig. S31. MAPS INS spectra $(0 - 4000 \text{ cm}^{-1})$ of the 2D carbon samples.

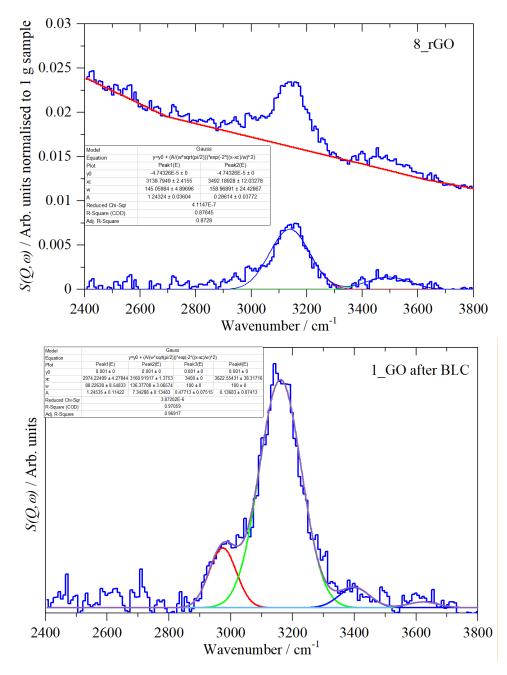


Fig. S32. Examples of curve fitting of the MAPS spectra in the C–H and O–H stretch region. The process first involves baseline correction (BLC) and then fitting to the minimum number of Gaussian functions. Top: 8_rGO, bottom: 1_GO.

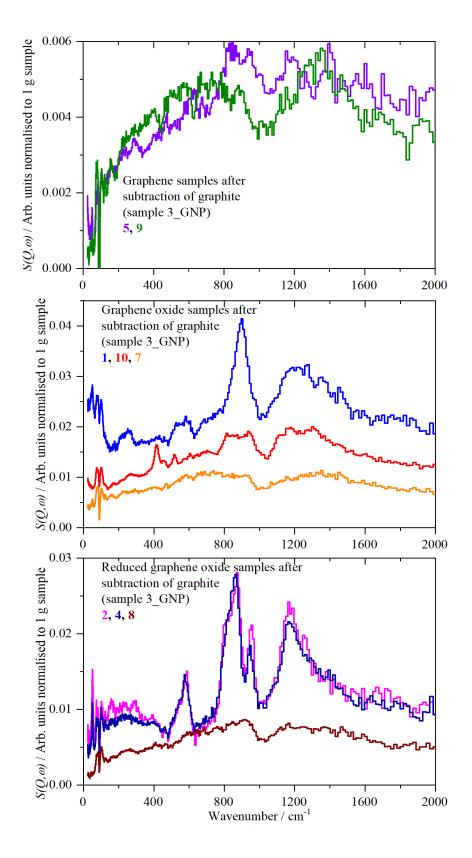


Fig. S33. TOSCA spectra of the 2D carbon samples after subtraction of the graphite contribution (3_GNP), grouped by sample type. Top panel graphenes: a) 5_GNP, b) 9_GNP. Middle panel graphene oxides: c) 1_GO, d) 10_GO, e) 7_GO (×2 ordinate expanded). Bottom panel reduced graphene oxides: f) 2_rGO, g) 4_rGO, h) 8_rGO. Note the different ordinate scales in the three panels.

Sample	Half-width of 2θ signal (ca. 26.5°) / Degrees	90 cm ⁻¹ peak width / cm ⁻¹
1_GO	4.85	29.5
2 rGO	4.25	20.5
3 GNP	0.749	9.1
4 rGO	1.306	10
5 GNP	0.381	9.1
6 GNP	0.506	8.9
7 GO	0.904	7.6
8 rGO	0.925	7.6
9 GNP	1.795	10.7
$1\overline{0}$ GO	1.613	29.9
11 ^{GNP}	0.582	8.1
12 GNP	0.536	11.9
13_GNP	0.658	8

Table S3 Correlation of 90 cm⁻¹ feature in the INS spectra with the width of the XRD peak at 26.5° (001).

Explanation of the 12 cm⁻¹ feature in Fig. 12a.

The intense peak at 12 cm⁻¹ observed in the simulated INS spectrum of graphene (Fig. 12a) does not originate in interlayer motion, unlike the 90 cm⁻¹ peak in graphite. The conversion from the phonon density of states ($G(\omega)$ which is what is calculated) to the experimentally observable $S(Q,\omega)$ involves multiplication by $1/\omega$ [2]. For a regular 3D material, the density of states of acoustic phonons at low energy roughly scales with frequency in a parabolic shape. This has the important implication that because the phonon DOS dies out faster than the $1/\omega$ term can boost the INS intensity as $\omega \rightarrow 0$, the overall INS intensity converges to zero. For 2D graphene, however, its phonon DOS near zero frequency converges to a finite value due to the quadratic flexural mode (and the fact that we now have a 2D reciprocal space). As a result, we lose an important counter-balance to the $1/\omega$ term and hence see the apparent divergence of intensity at low energy.

References

[1] P. Albers, A. Karl, J. Mathias, D. K. Ross, S. F. Parker, INS-, XPS- and SIMS-investigations on the controlled post-oxidation of pigment blacks – Detection of different species of strongly adsorbed water, *Carbon* 39 (2001) 1663-1676, https://doi.org/10.1016/S0008-6223(00)00294-3.

[2] G. L. Squires, *Introduction to Thermal Neutron Scattering*, Dover, 1996, pp54-56.