

Aluminum sheet induced zinc oxide nanosheets decorated with silver nanoparticles for ultrasensitive SERS sensing of crystal violet

Xuejuan Chen, Qiuli Wang, Lixia Qin*, Xiaoxia Liu, Shi-Zhao Kang, Taiyang Zhang,
Xiangqing Li*

*School of Chemical and Environmental Engineering, Center of Graphene Research,
Shanghai Institute of Technology, 100 Haizhu Road, Shanghai 201418, China*

*Xiangqing Li. Tel.: +86 21 60873061; Fax: +86 21 64253317

xqli@sit.edu.cn (Xiangqing Li); lxqin@sit.edu.cn (Lixia Qin)

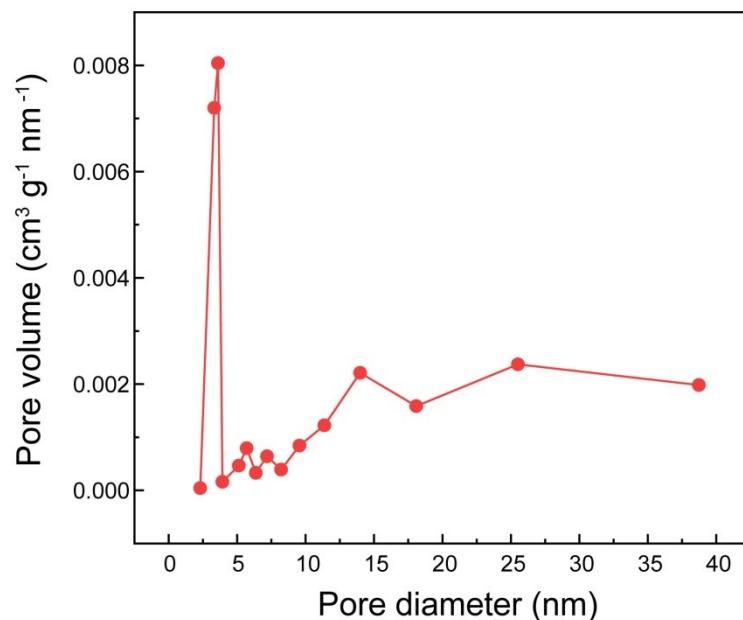


Fig. S1 Pore diameter distribution of Al/ZnO.

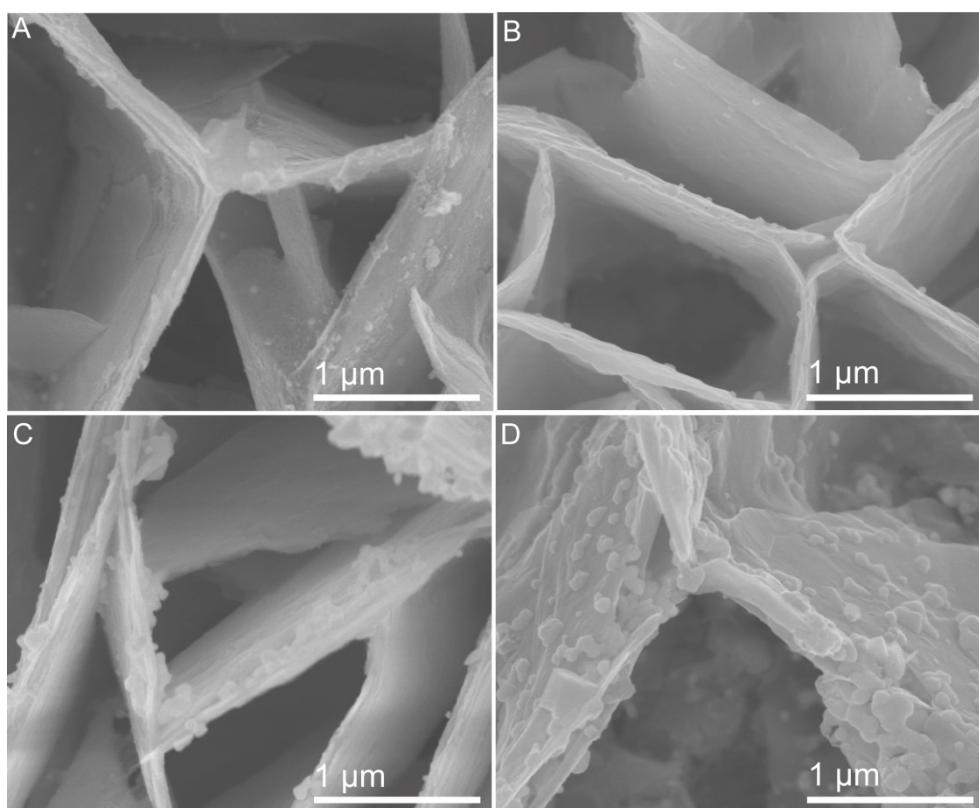


Fig. S2 (A ~ D): SEM images of Al/ZnO/Ag at other reduction times (5, 10, 20 and 30 min).

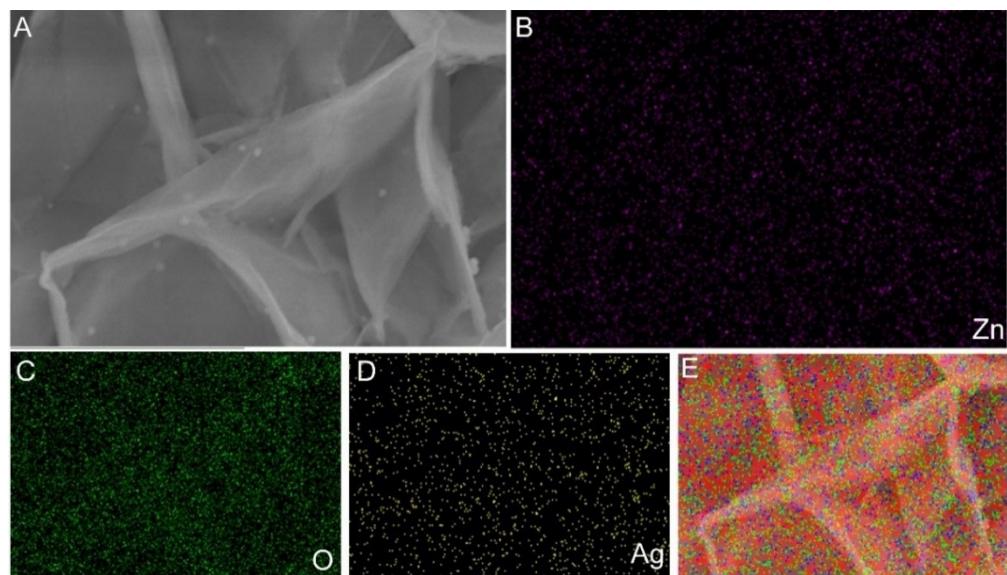


Fig. S3 (a) Energy-filtered SEM image of the Al/ZnO/Ag; Elemental distribution of Zn (B), O (C) and Ag (D); and the overlay distribution of elements (E).

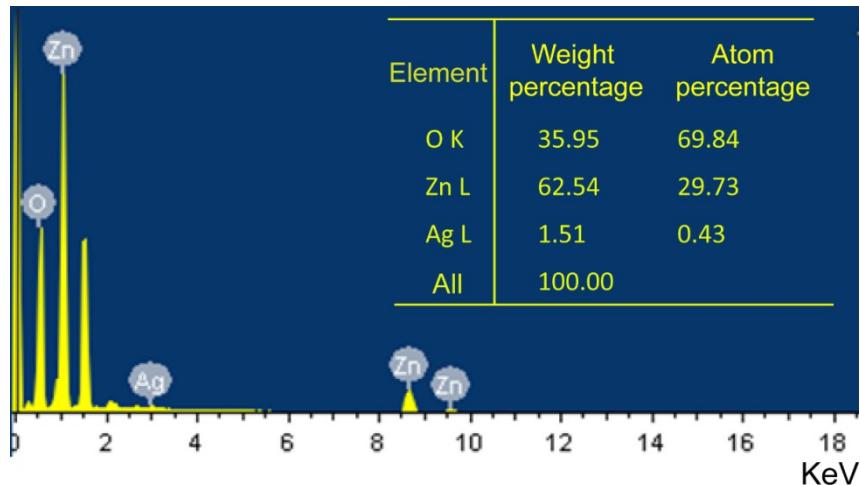


Fig. S4 EDX analysis of the Al/ZnO/Ag.

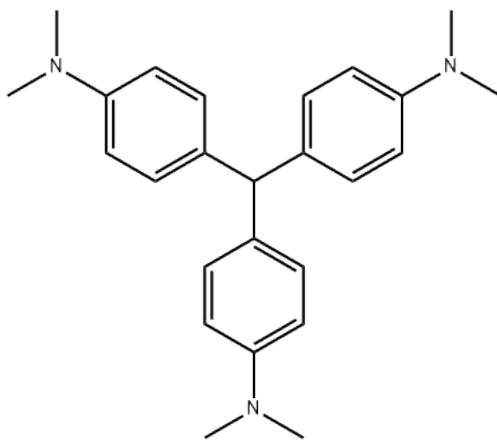


Fig. S5 The structure of CV.

In addition, the analytical enhancement factor (AEF) of the substrate can be estimated according to the peak at 1609 cm^{-1} by the following formula:¹

$$\text{AEF} = (\text{I}_{\text{SERS}}/\text{C}_{\text{SERS}})/(\text{I}_{\text{RS}}/\text{C}_{\text{RS}})$$

Where I_{SERS} is the SERS intensity of CV at the concentration of $\text{C}_{\text{SERS}} (1 \times 10^{-7}\text{ mol}\cdot\text{L}^{-1})$ on the Al/ZnO/Ag substrate. I_{RS} is the Raman intensity of the CV at the concentration of $\text{C}_{\text{RS}} (1 \times 10^{-3}\text{ mol}\cdot\text{L}^{-1})$ on Al sheet. Thus, the AEF is estimated as 2.45×10^8 for the Al/ZnO/Ag.

Table S1 The comparison of SERS performance toward CV on various substrates.

Materials	Analyte	LOD	EF
Cellophane/Ag ²	CV	$1 \times 10^{-9} \text{ mol} \cdot \text{L}^{-1}$	2.0×10^5
Paper-based Au/AgNP ³	CV	$8.1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1}$	-
Ag/CdS ⁴	CV	$1 \times 10^{-12} \text{ mol} \cdot \text{L}^{-1}$	1.0×10^7
Au@Ag-frustum arrays ⁵	CV	$1 \times 10^{-10} \text{ mol} \cdot \text{L}^{-1}$	2.67×10^7
This work	CV	$3.6 \times 10^{-14} \text{ mol} \cdot \text{L}^{-1}$	2.45×10^8

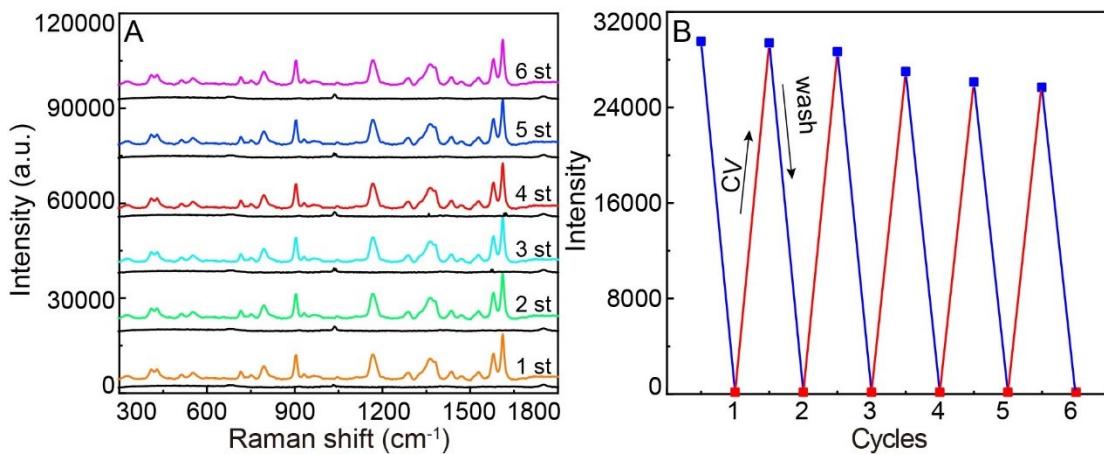


Fig. S6 (A) SERS spectra of the CV ($1 \times 10^{-7} \text{ mol} \cdot \text{L}^{-1}$) on the same Al/ZnO/Ag substrate for 6 cycles and (B) the line chart of SERS intensity at 1609 cm^{-1} with cycle.

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