Supplementary Information

LED-driven controlled deposition of Ni onto TiO_2 for the visible-light expanded conversion of carbon dioxide into C_1 - C_2 alkanes.

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EPR Operando protocol

The following protocol was performed on both P25 and Ni/TiO₂ to compare the response of the materials to UV light and to different gas mixtures.

The sample was first activated in 30% H_2 at 250°C. Then the influence of UV light and of the H_2/CO_2 gas (separately and together) was investigated at room temperature. Then the sample was reactivated and the procedure was repeated at 250 °C.

In general the spectra had lower resolution at 250 °C and changes in the spectra were more difficult to distinguish from the noise level. Overall the same evolution was observed, but less clear and only room temperature spectra are shown in the text. The spectrum assigned to Ni species (Type 1, see main text) changed with temperature.

The protocol performed on both P25 and on Ni/TiO_2 are given below. Small changes in time intervals have been ignored when making the table.

Event	Flow [ml/min]	T [°C]	
30% H ₂ on	15/35 H ₂ /He	Room temperature	
Heat (15 min)	15/35 H ₂ /He	Ramp to 250°C by 15°/min	
Activation in $H_2(40 \text{ min})$	15/35 H ₂ /He	250°C	
Cool to RT(10 min)	15/35 H ₂ /He	Cool to RT	
Light applied(100 s)	15/35 H ₂ /He	Room temperature	
H_2+CO_2 on(10 min)	4/16/30 CO ₂ /H ₂ /He	Room temperature	
Light applied(100 s)	4/16/30 CO ₂ /H ₂ /He	Room temperature	
H ₂ +CO ₂ off	4/16/30 CO ₂ /H ₂ /He	Room temperature	
30% H ₂ on	15/35 H ₂ /He	Room temperature	
Heat (15 min)	15/35 H ₂ /He	Ramp to 250°C by 15°/min	
Activation in $H_2(5 \text{ min})$	15/35 H ₂ /He	256°C	
Light applied(100 s)	15/35 H ₂ /He	256°C	
H_2+CO_2 on(10 min)	4/16/30 CO ₂ /H ₂ /He	256°C	
Light applied(100s)	4/16/30 CO ₂ /H ₂ /He	256°C	
H_2 +CO ₂ off, H_2 on	15/35 H ₂ /He	256°C	
Cool to RT(10 min)	15/35 H ₂ /He	Cool to RT	
H ₂ off	50 He	Room temperature	

Table S1. EPR protocol for the Operando characterization of P25 and Ni/TiO₂ catalyst.



Figure S1: Detail of the XPS fitting parameters employed to calculate Ni(0)/Ni(II) ratios in the Ni2p3/2 regions registered in the in situ XPS measurements.

Table S2. Main parameters of Gaussian/Lorentzian peaks used for Ni2p fitting corresponding to Ni^0 and Ni^{2+} shapes.

Component for Ni2p Fittings		B. Energy / eV	FWMH	Areas / Ratio
Ni(II)	Ni Oxidized-1	856.2	4.01	1.0
component	Ni Oxidized-2	862.0	5.13	0.5
	Ni Oxidized-3	855.4	2.04	0.15
Ni ⁰	Ni Metallic-1	852.3	2.55	1.0
component	Ni Metallic-2	856.9	3.31	0.18

Table S3. Percentages of Ni and Ti observed by XPS as determined by Ni2p/Ti2p corrected areas and ratio of Ni⁰ and Ni²⁺ as determined by fitting using parameters of Table 1.

Spectrum	Ni2p %	Ti2p %	Ni ⁰ %	Ni ²⁺ %
		00.0	2	07
Original	7.1	92.9	3	97
H ₂ 400 °C	1.6	98.4	57	43
CO ₂ +H ₂ 225 °C	4.3	95.7	16	84
H ₂ 450 °C	2.2	97.8	45	54
CO ₂ +H ₂ 225 °C	2.4	97.6	49	51

Catalyst before reaction



Figure S2. Additional HAADF-STEM representative images of the Ni/TiO2: (a-c) before

reaction; (d-f): after reaction.



Figure S3. Additional TEM images of the Ni/TiO_2-A catalyst after photocatalytic hydrogenation

under LED irradiation at 365 nm



Figure S4. Productivity *vs.* time plots for P25 (a) under LED light irradiation at 365nm and for Ni/TiO₂ under LED light irradiation at 365nm and H₂:Ar (1:4) gas mixture.