

## Supporting Information

# PEDOT:PSS-Glued MoO<sub>3</sub> Nanowires Network for All-Solid-State Flexible Transparent Supercapacitors

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## Calculations

The areal capacitance of AgNFs/MoO<sub>3</sub>/PEDOT:PSS (AMP) film electrodes and the assembled flexible transparent supercapacitors (FTSCs) was calculated from CV and GCD curves based on the following equations, respectively:

$$C = \frac{1}{2v\Delta V} \int I(V) dV \quad (1)$$

$$C = \frac{I(A)\Delta t}{\Delta V} \quad (2)$$

$$C_{area} = \frac{C}{S} \quad (3)$$

where  $C$  is a total capacitance in F,  $v$  is a scan rate in V s<sup>-1</sup>,  $I(V)$  is an instant current, and  $V$  is an applied voltage in V.  $I(A)$  refers to the discharge current;  $\Delta V(V)$  represents the changed potential during the discharge time  $\Delta t(s)$ , and  $S$  corresponds to the effective area of AMP electrodes.

The energy density ( $E$ ) and power density ( $P$ ) of the FTSCs were calculated using the following equations:

$$E = \frac{1}{2} \times C_{area} \times \frac{(\Delta V)^2}{3600} \quad (4)$$

$$P = \frac{E}{\Delta t} \times 3600 \quad (5)$$

Where  $E$  and  $P$  correspond to the areal energy density and power density, respectively.

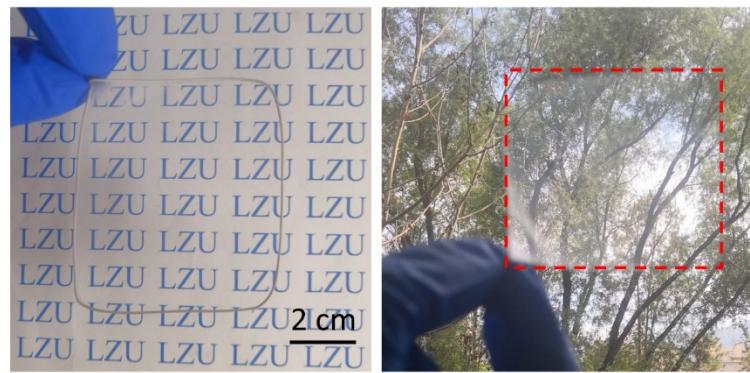


Fig. S1. Optical photographs of a freestanding AgNFs network.

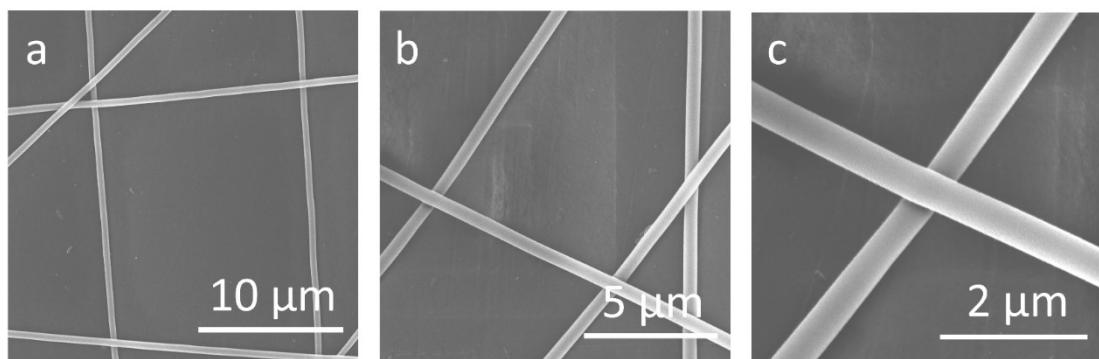


Fig. S2. SEM images of AgNFs network.

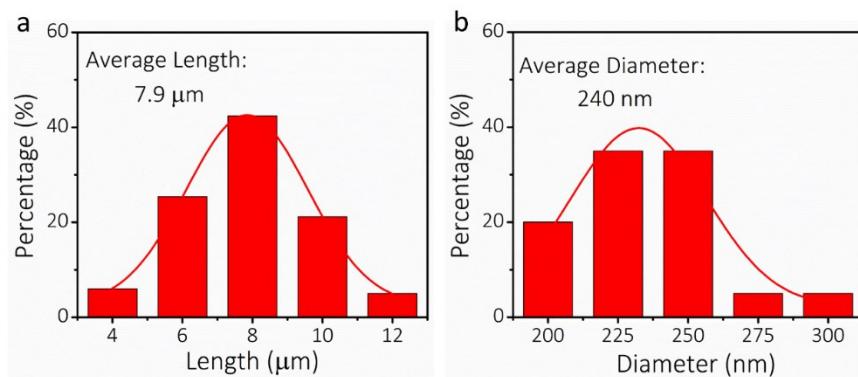


Fig. S3. (a) Length histogram and (b) diameter histogram of the as-synthesized  $\text{MoO}_3$  nanowires.

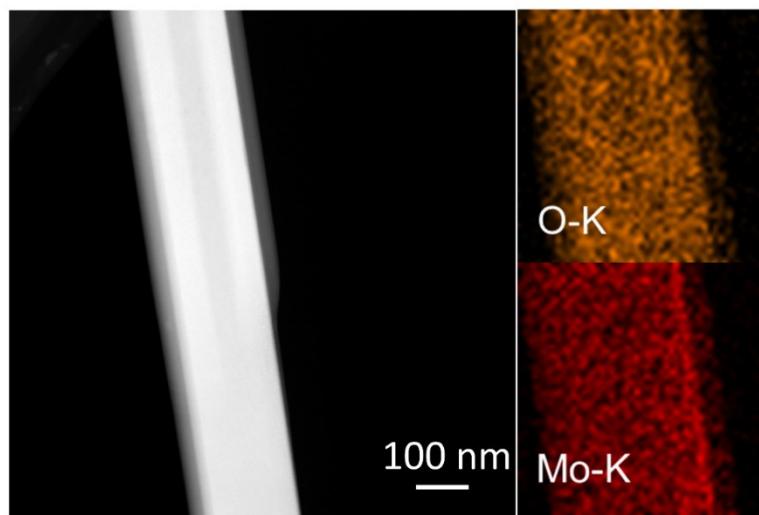


Fig. S4. Element mapping images of the as-synthesized MoO<sub>3</sub> nanowires.

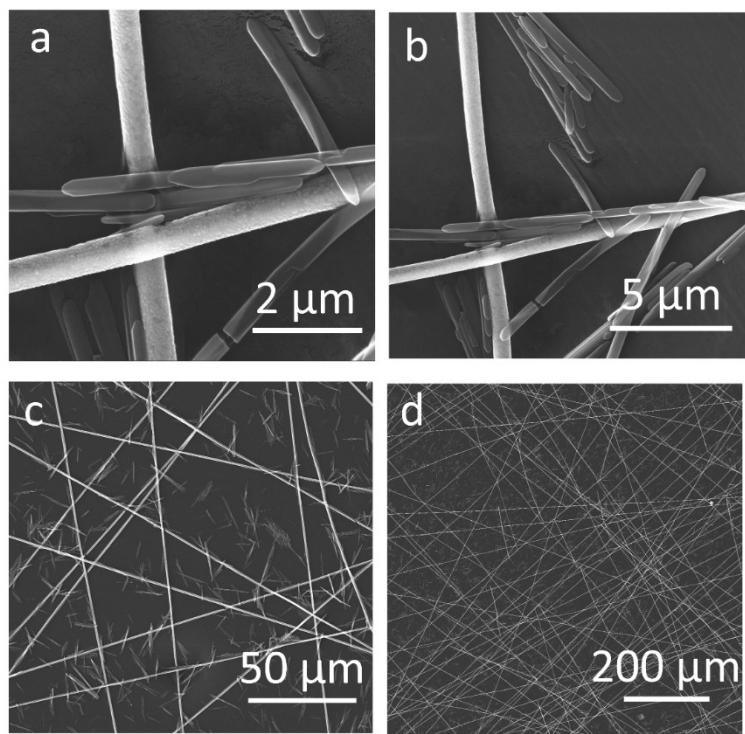


Fig. S5. SEM images of AMP electrode (the precursor ratio of 6:1).

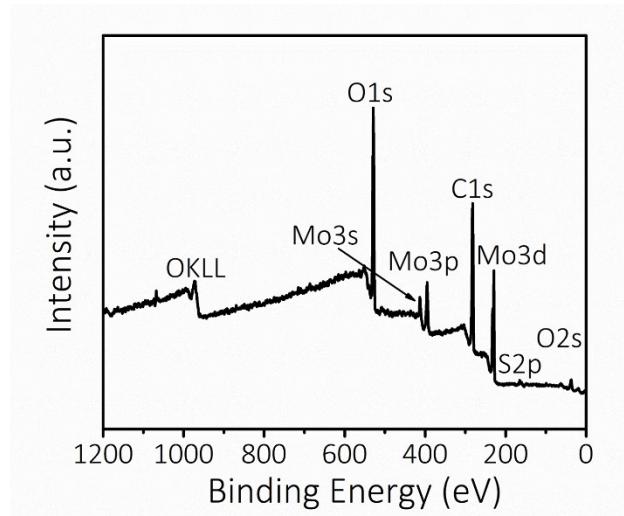


Fig. S6. XPS spectrum of PEDOT:PSS-glued  $\text{MoO}_3$  nanowires (the precursor ratio of 6:1).

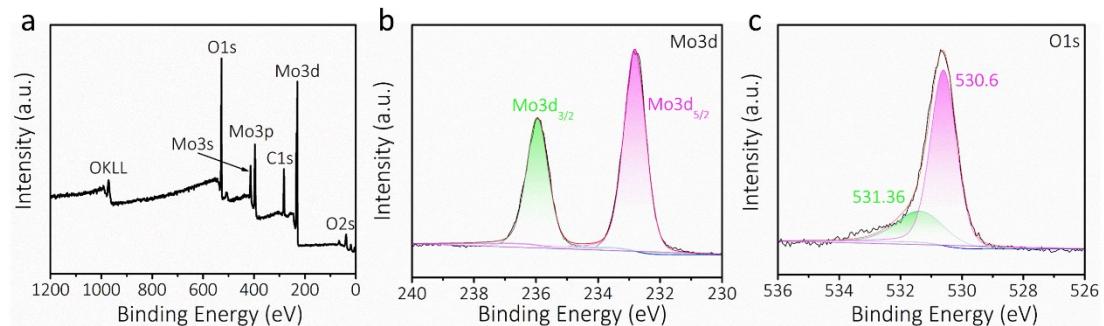


Fig. S7. (a) XPS survey spectrum of  $\text{MoO}_3$  nanowires. High-resolution XPS spectra of (b) Mo 3d, and (c) O 1s for  $\text{MoO}_3$  nanowires.

The survey spectrum (Fig. S7a) indicates the presence of Mo and O elements. The Mo 3d<sub>5/2</sub> and Mo 3d<sub>3/2</sub> peaks located at 232.8 and 235.9 eV reveal a +6 oxidation state of  $\text{MoO}_3$  phase (Fig. S7b). The O1s peak at 530.6 eV (Fig. S7c) corresponds to the oxygen atoms of  $\text{MoO}_3$ , and the one centered at 531.4 eV is related to the adsorbed oxygen.

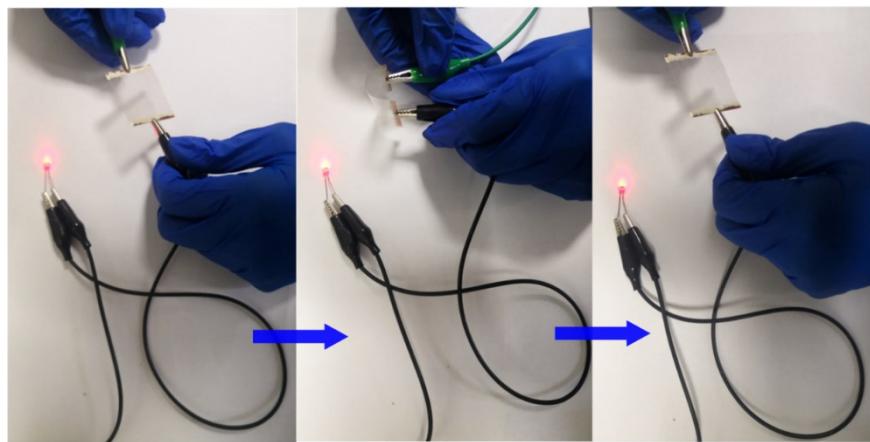


Fig. S8. Optical photographs of AgNFs network connected in a LED circuit.

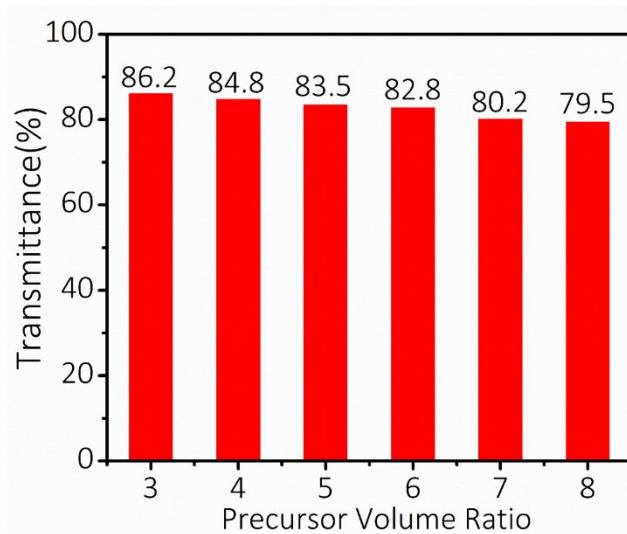


Fig. S9. The optical transmittances at 550 nm for the electrodes with different precursor ratios.

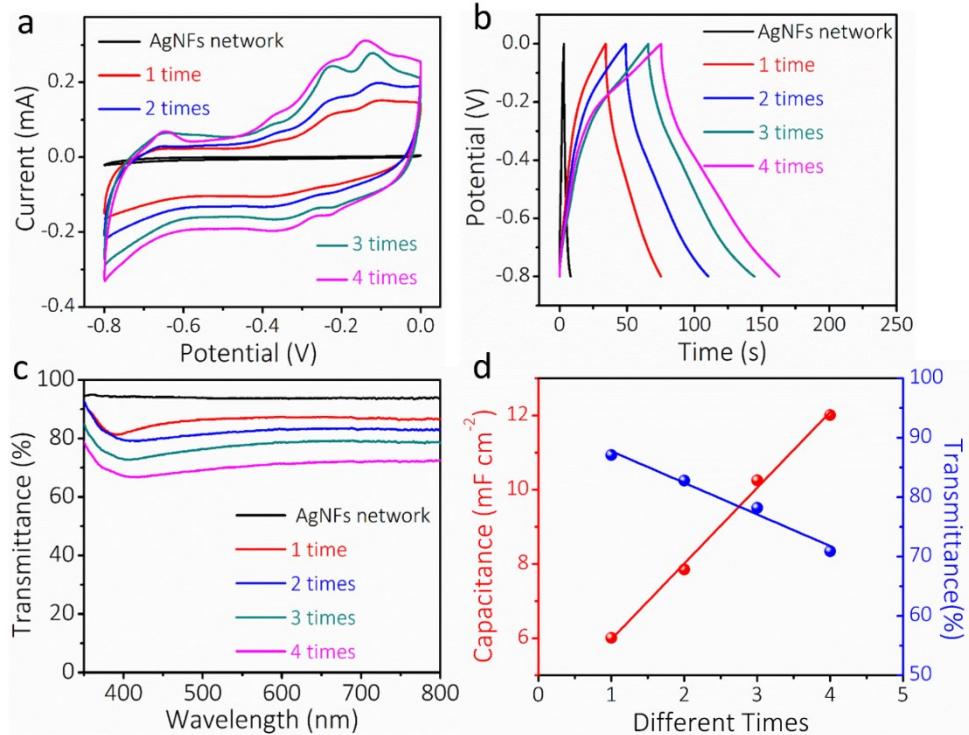


Fig. S10. Electrochemical performance of AMP electrodes (precursor ratio of 6:1) that were spin-coated the precursor hybrid dispersion for 1 to 4 times, AgNFs network was as a control experiment; (a) CV patterns at  $10 \text{ mV s}^{-1}$ , (b) GCD patterns at  $0.10 \text{ mA cm}^{-2}$ ; (c) Transmittance spectra of AMP electrodes; (d) Optical transmittance and areal capacitance calculated by GCD curves.

Fig. S10a presents CV curves measured at  $10 \text{ mV s}^{-1}$  for AMP electrodes with different spin-coating times. The enclosed area of CV curves enlarges with increasing spin-coating times, indicating an enhanced areal capacitance. Fig. S10b shows GCD curves at a current density of  $0.10 \text{ mA cm}^{-2}$ , the charge-discharge time of AMP electrodes is increased sequentially. The CV and GCD curves indicate that the electrochemical performance becomes better with the increase of spin-coating times. Meanwhile, the optical transparency of film electrodes is gradually degradable due to the more  $\text{MoO}_3$  nanowires in the film electrodes (Fig. S10c). The calculated areal capacitance and optical transmittance at 550 nm are collected in Fig. S10d. The linear relationship is found with the spin-coating times of AMP precursor hybrid dispersion. For each spin-coating time, the areal capacitance of  $1.56 \text{ mF cm}^{-2}$  is enhanced, and the

optical transmittance of 5.32 % is degraded.

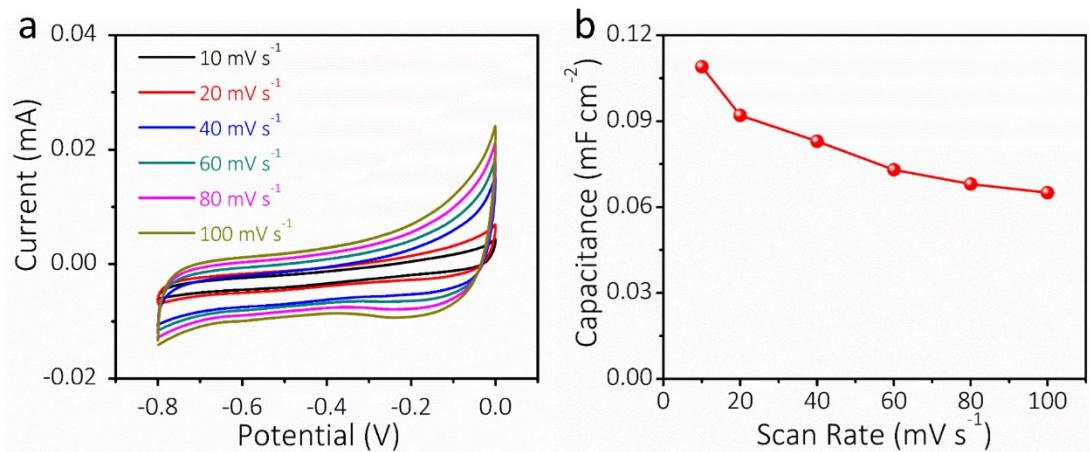


Fig. S11. (a) CV curves at different scan rates for the AgNFs network, and (b) the areal capacitance as a function of scan rate determined from the CV curves.

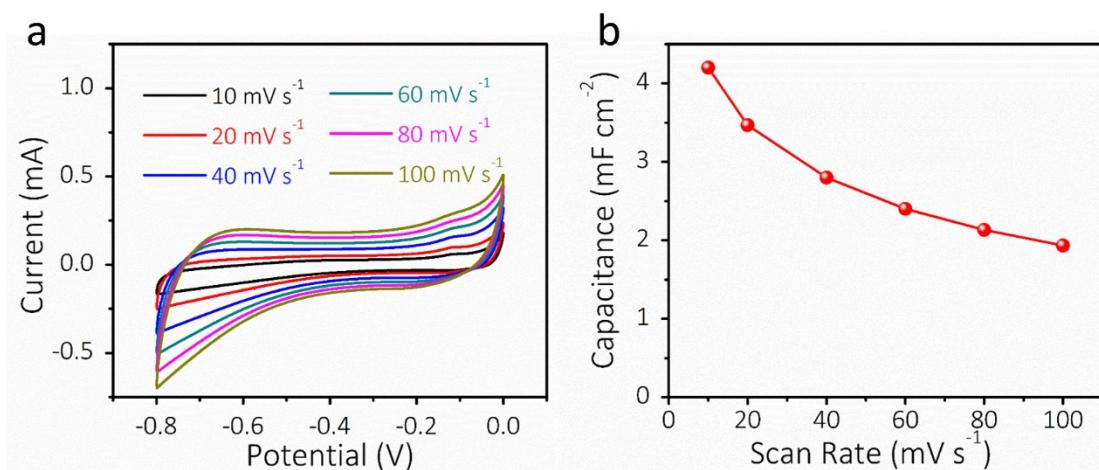


Fig. S12. (a) CV curves at different scan rates for the AgNFs/MoO<sub>3</sub> (AM) electrodes, and (b) the areal capacitance as a function of scan rate determined from the CV curves.

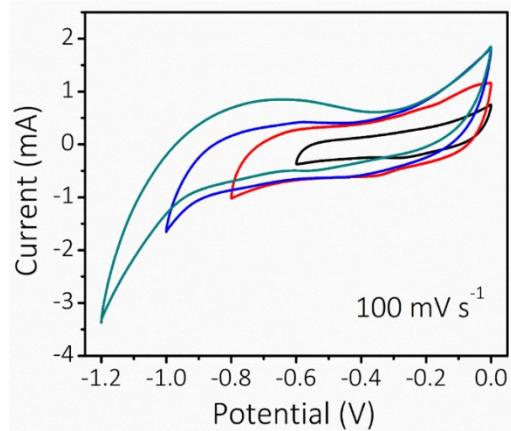


Fig. S13. CV curves measured at  $100 \text{ mV s}^{-1}$  for AMP electrode (precursor ratio of 6:1) with different voltage windows.

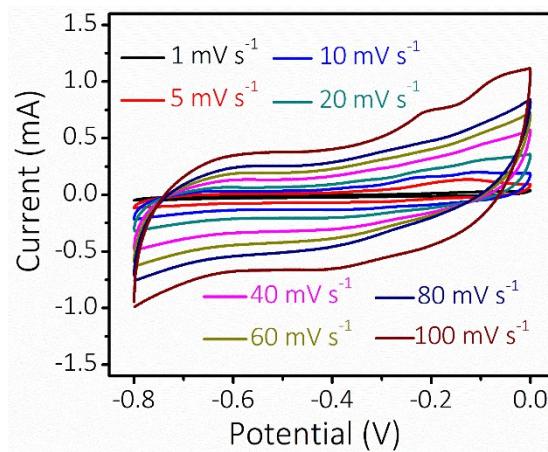


Fig. S14. CV curves at different scan rates for AMP electrodes (precursor ratio of 6:1).

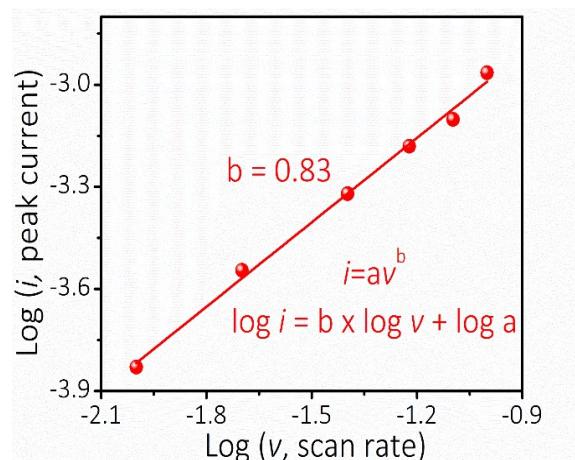


Fig. S15. The current response against scan rate and the extracted b-value for AMP electrodes (precursor ratio of 6:1).

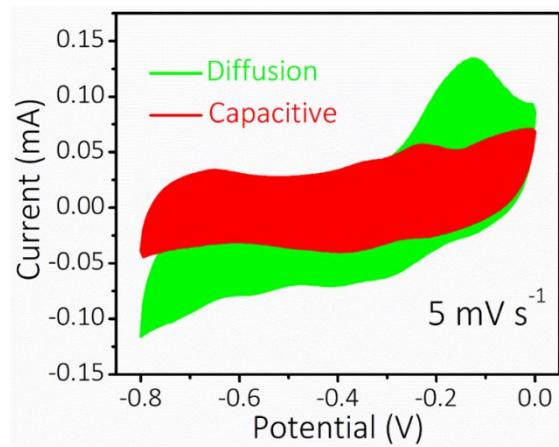


Fig. S16. The capacitive contribution in CV results of AMP electrode (precursor ratio of 6:1) at  $5 \text{ mV s}^{-1}$ .

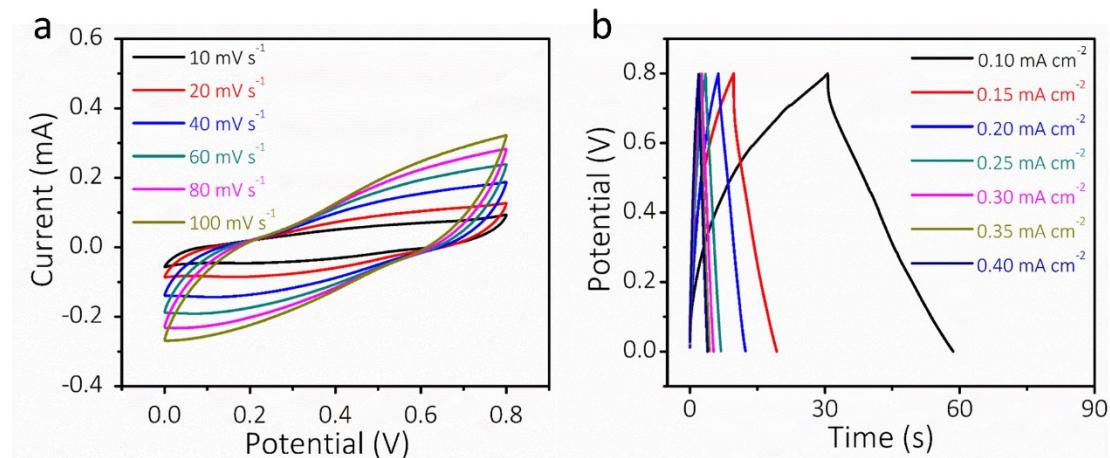


Fig. S17. CV curves at different scan rates and GCD curves at different current densities for the AM-based FTSC devices.

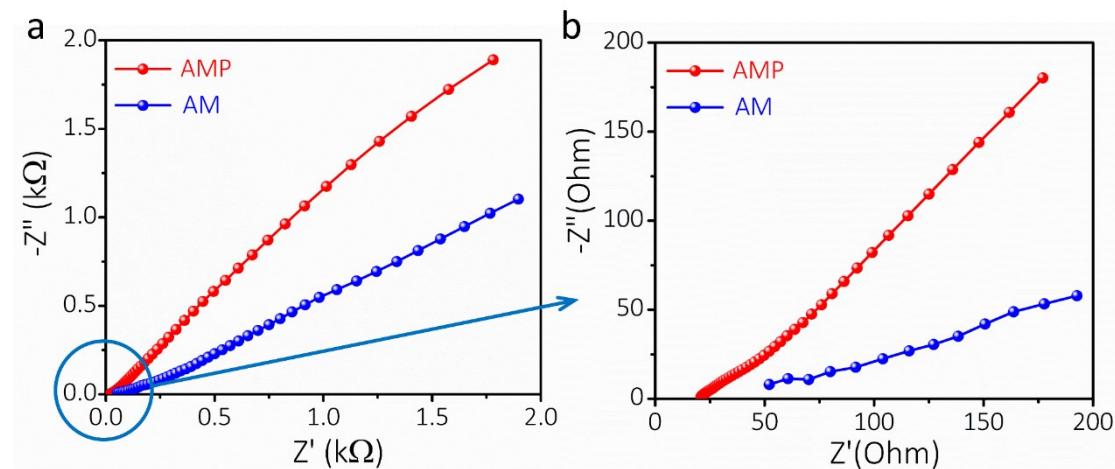


Fig. S18. EIS curves of (a) the AMP-based and (b) AM-based FTSC devices.

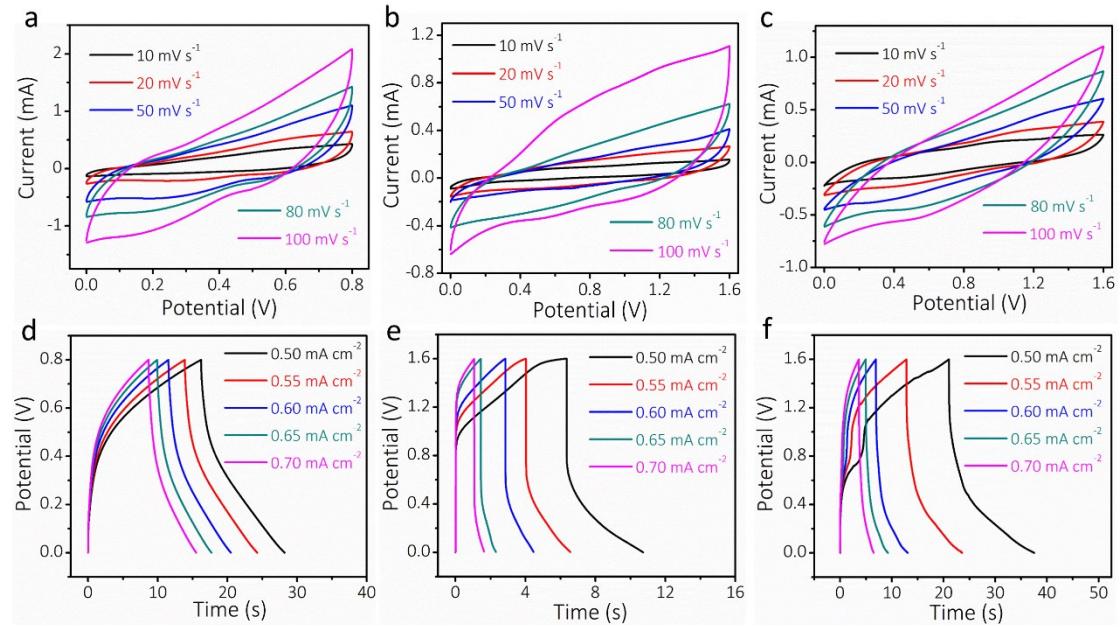


Fig. S19. CV curves of two FTSCs connected in parallel (a), in series (b), and two ones in series with the one in parallel (c). GCD curves of two FTSCs connected in parallel (d), in series (e), and two ones in series with the one in parallel (f).

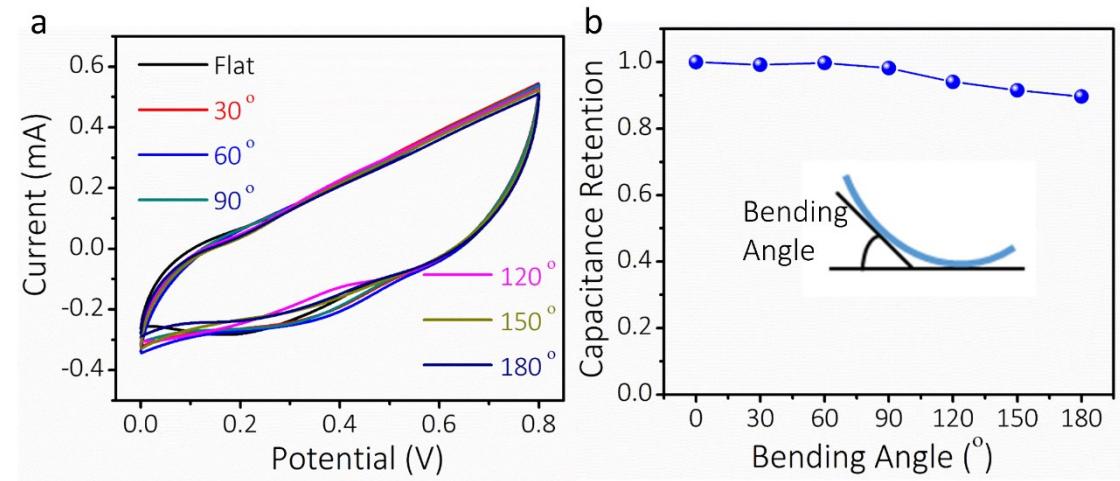


Fig. S20. The electrochemical performance of AMP-based FTSCs under different bending angles. (a) CV curves at  $50 \text{ mV s}^{-1}$  and (b) the calculated capacitance retention.

Table S1. Comparison of the sheet resistance, optical transparency, and mechanical flexibility of transparent electrodes after 1000 bending cycles.

Material	Rs ( $\Omega \text{ sq}^{-1}$ )	Transmittance (T) (%)	R/R <sub>0</sub>	Bending radius (mm)	Reference
Cu nanofibers	50	90	1.1	--	1
AgNWs/PEDOT:PSS	45	92	1.2	--	2
Ag grid	30	85	1.03	15	3
Ag grid	6.8	82	1.3	10	4
Ag grid	5.2	91	1.6	15	5
AgNFs/MoO <sub>3</sub> /PEDOT:PSS	9.7	82.8	1.05	10	This work

Table S2. Comparison of the areal capacitance and optical transparency for transparent supercapacitors.

Material	Capacitance ( $\text{mF cm}^{-2}$ )	Transmittance (T %)	Reference
PEDOT:PSS	1.18	65	6
Graphene	3.3	47	7
RuO <sub>2</sub> /PEDOT	0.84	80	8
PEDOT:PSS/AgNWs	0.6	51	9
MnO <sub>2</sub> /ITO/PET	4.73	44	10
MnO <sub>2</sub> @AuNFs	2.07	79	11
Co(OH) <sub>2</sub> /AgNWs	0.54	54	12
Au@MnO <sub>2</sub> nanomesh	0.795	36	13
HTSE film	3.64	85	14
GNHC-GF	5.48	44	15
AgNFs/MoO <sub>3</sub> /PEDOT:PSS	6.33	65	This work

Table S3. Energy densities and power densities of our FTSCs device in comparison with the reported transparent supercapacitors.

Active material	Power density ( $\mu\text{W cm}^{-2}$ )	Energy density ( $\mu\text{Wh cm}^{-2}$ )	Reference
Ti <sub>3</sub> C <sub>2</sub> Tx (symmetric)	0.62	0.049	16
	1.24	0.034	
	2.51	0.025	
	5.14	0.02	
	10.91	0.018	
PEDOT:PSS	1.16	0.009	17
	1.82	0.008	
	3.7	0.008	
	11.76	0.0079	
	20.92	0.006	
RuO <sub>2</sub> /PEDOT:PSS	0.28	0.015	8
	1.13	0.014	
	2.08	0.014	
	4.83	0.014	
	19.47	0.011	
Graphene QDs	0.094	0.00079	18
	0.083	0.00082	
	0.072	0.00084	
	0.062	0.00087	
RMGO	9	0.014	19
CVD-Graphene	2	0.0028	19
AgNFs/MoO <sub>3</sub> /PEDOT:PSS	40	0.623	This work
	60	0.538	
	80	0.459	
	100	0.400	

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120	0.342
140	0.302
160	0.282

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## Notes and references

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