

## Supplementary Information

# Promoting electrocatalytic overall water splitting by sulfur incorporation into CoFe(oxy)hydroxide.

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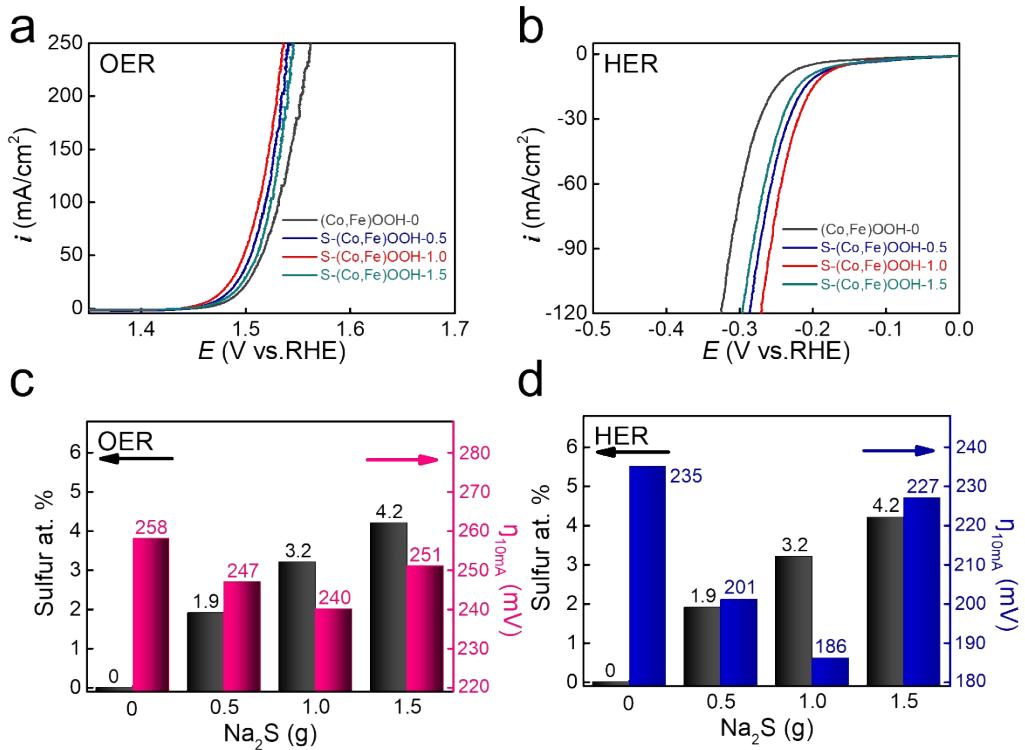
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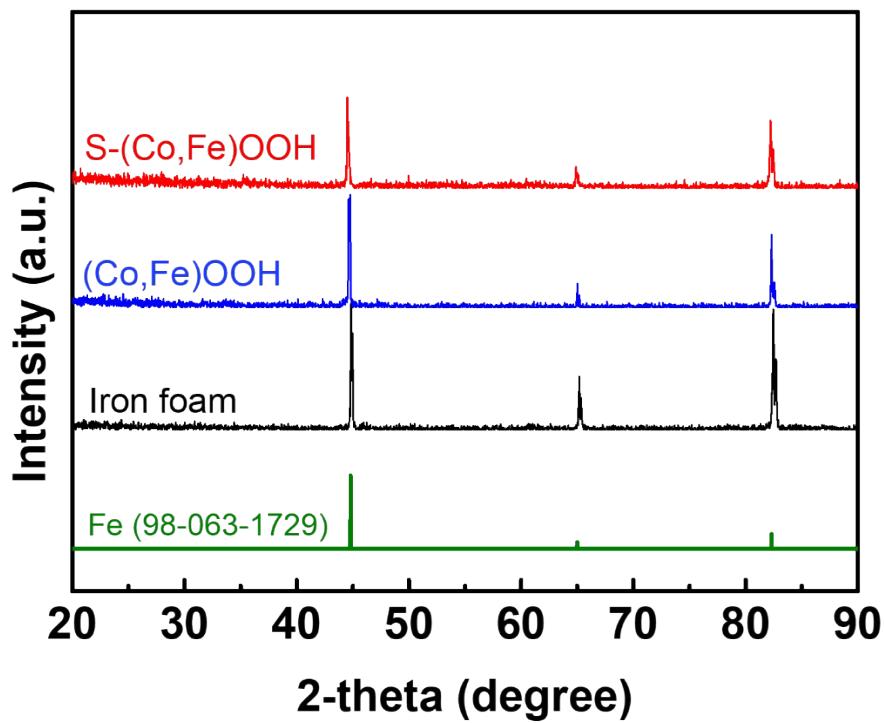
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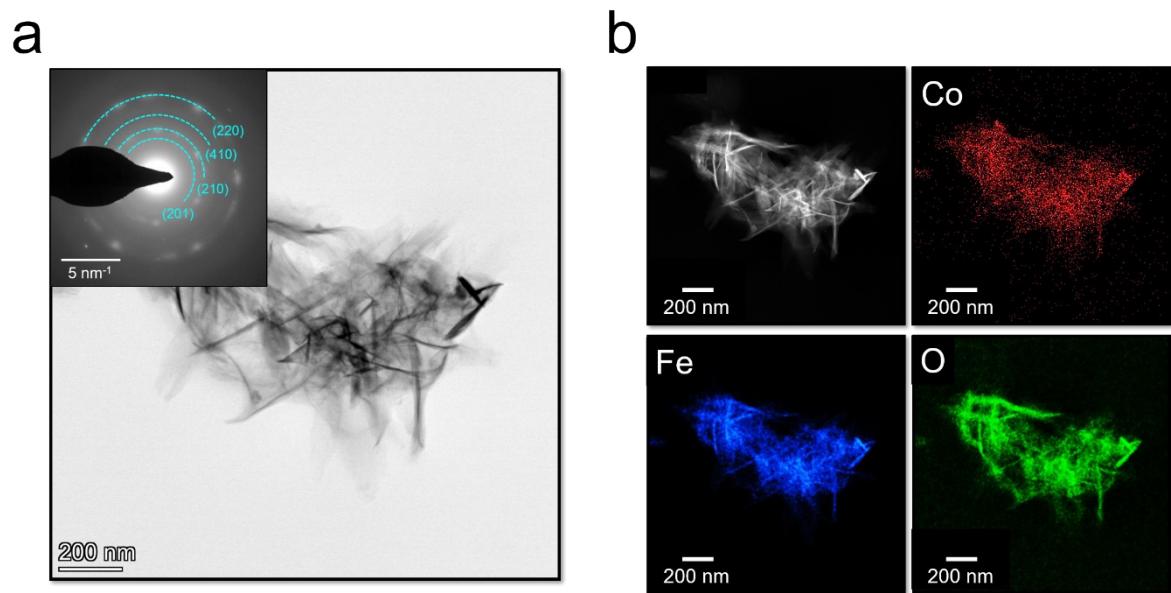
KEYWORDS : Overall water splitting reaction; Hydrogen production; Sulfur incorporation; Cobalt–iron (oxy)hydroxide; Bifunctional electrocatalysts.



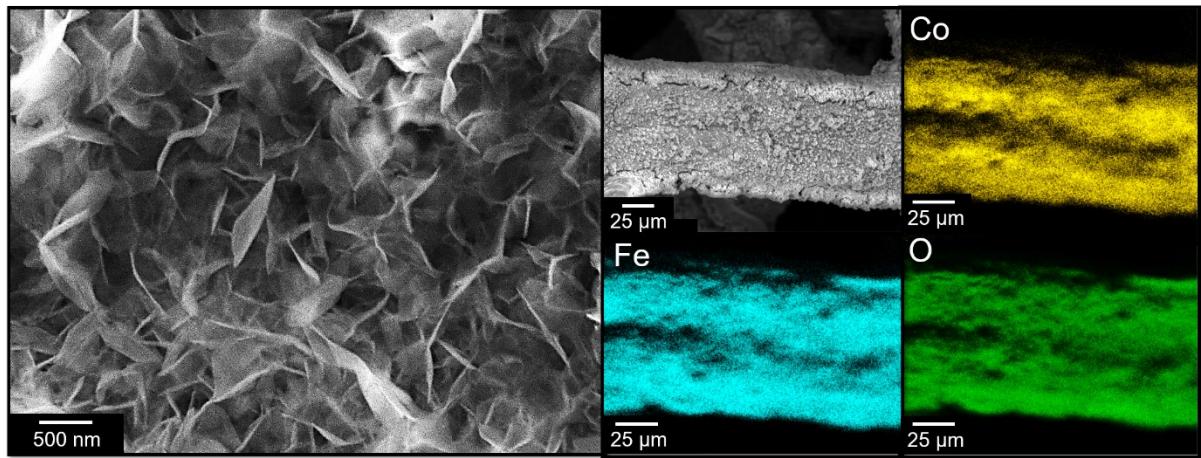
**Figure S1.** Electrochemical analysis of  $S-(\text{Co},\text{Fe})\text{OOH}$  prepared with different amounts of  $\text{Na}_2\text{S}$  (0, 0.5, 1.0, and 1.5 g). (a) Polarization curves for OER. (b) Polarization curves for HER. (c) Comparison of sulfur content and overpotential for OER at  $+10 \text{ mA}/\text{cm}^2$ . (d) Comparison of sulfur content and overpotential for HER at  $-10 \text{ mA}/\text{cm}^2$ .



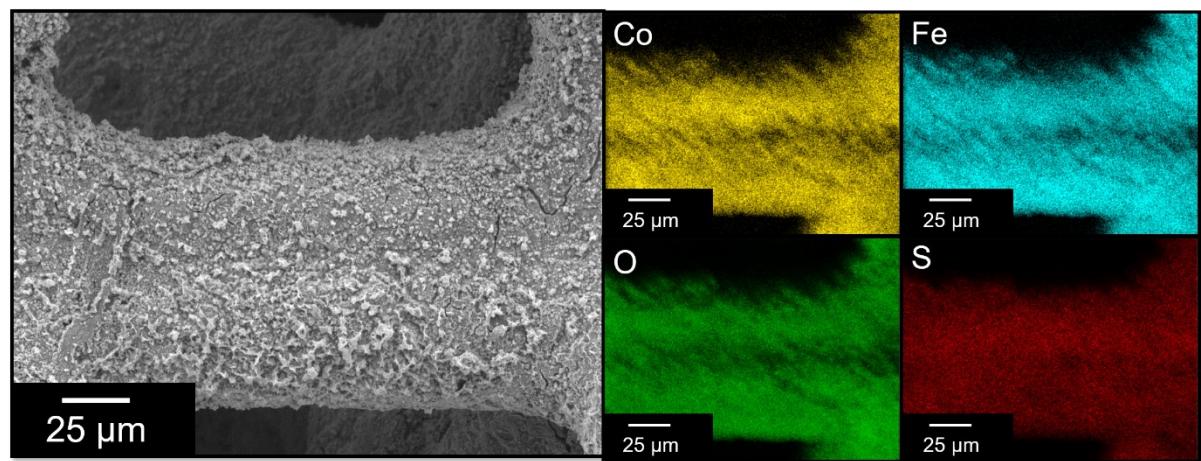
**Figure S2.** XRD patterns of Iron foam,  $(\text{Co},\text{Fe})\text{OOH}$  and  $\text{S}-(\text{Co},\text{Fe})\text{OOH}$ .



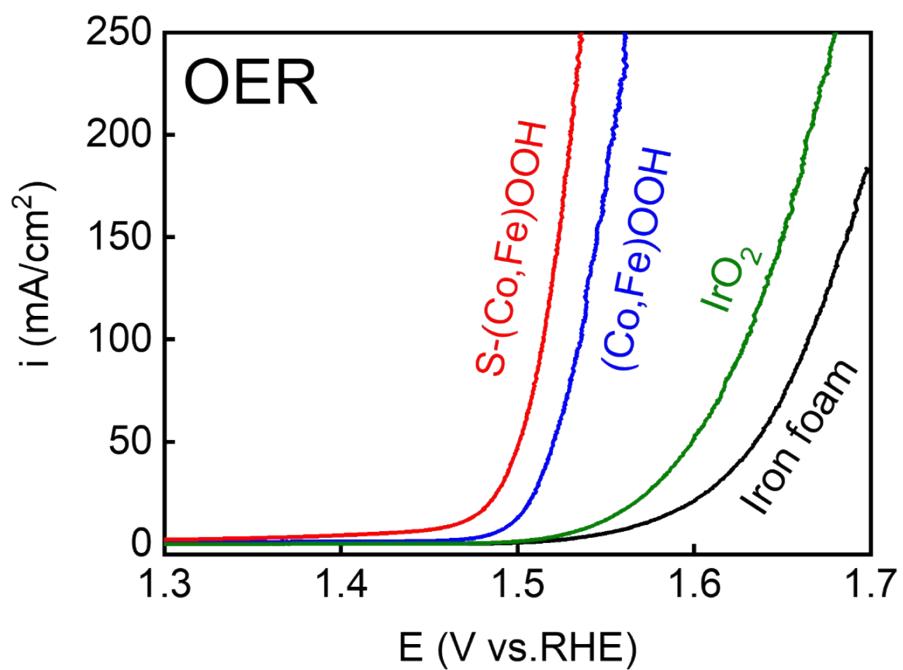
**Figure S3.** (a) High resolution transmission electron microscopy (HR-TEM) image of (Co,Fe)OOH with selected area electron diffraction (SAED) ring patterns. (b) TEM-EDS mapping images of (Co,Fe)OOH.



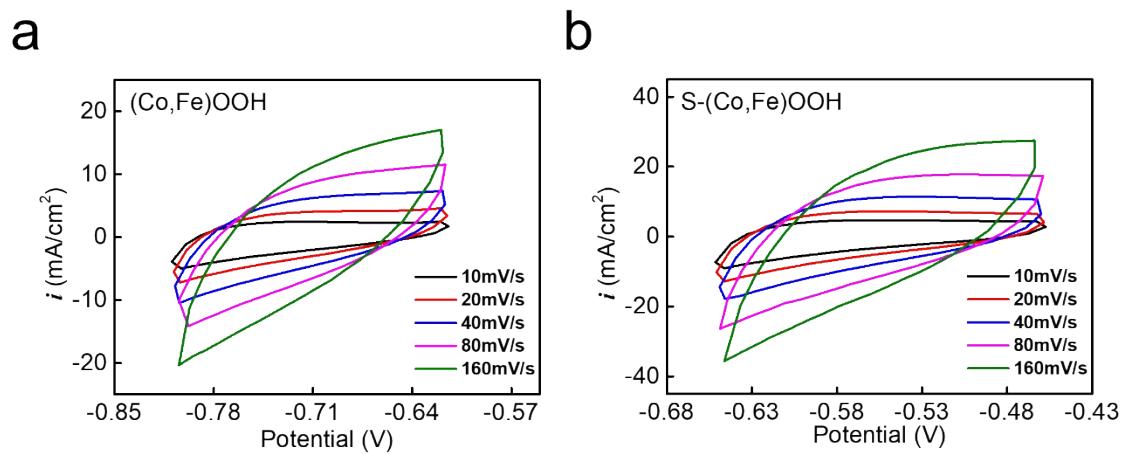
**Figure S4.** Scanning electron microscopy (SEM) image and Energy Dispersive Spectroscopy (EDS) mapping images of  $(\text{Co},\text{Fe})\text{OOH}$ .



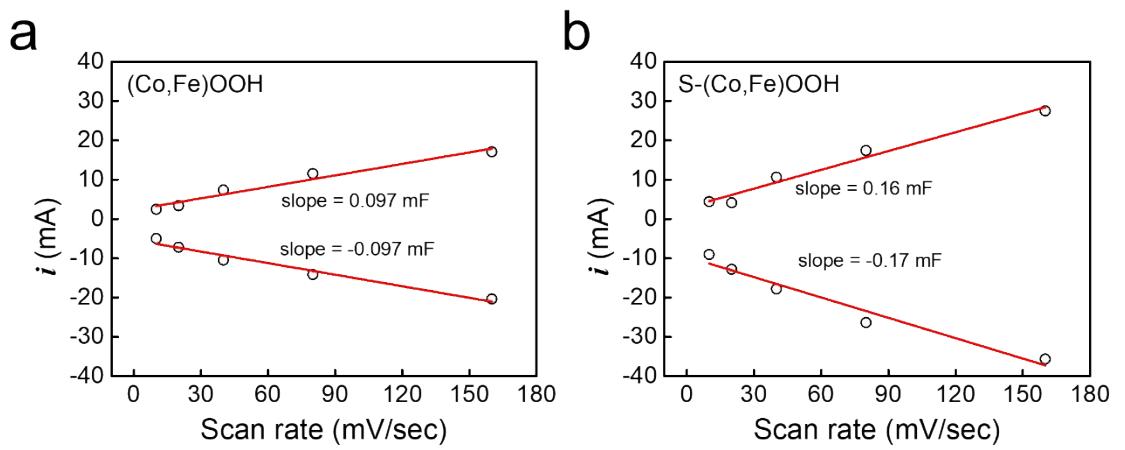
**Figure S5.** Energy Dispersive Spectroscopy (EDS) mapping images of S-(Co,Fe)OOH.



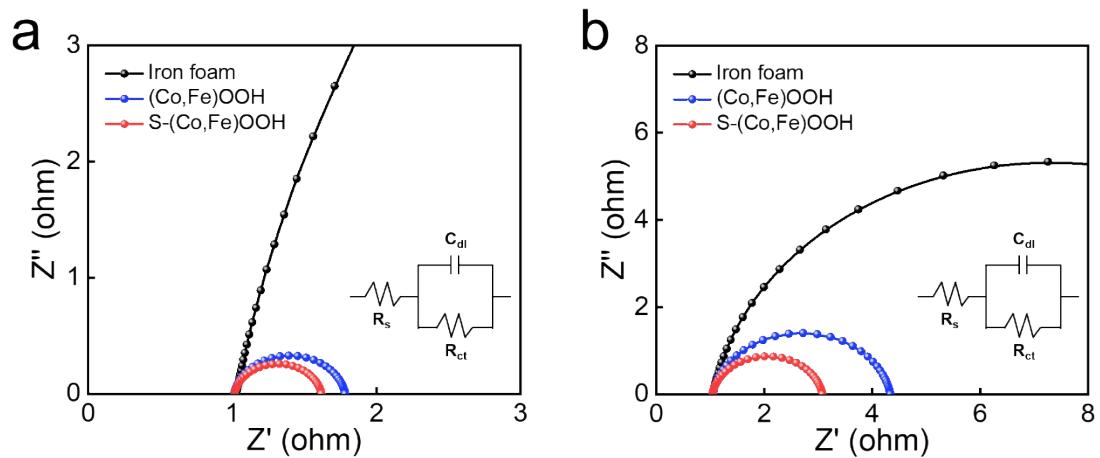
**Figure S6.** Forward swept polarization curves of Iron foam,  $\text{IrO}_2$ ,  $(\text{Co,Fe})\text{OOH}$ , and  $S\text{-(Co,Fe)OOH}$  for OER.



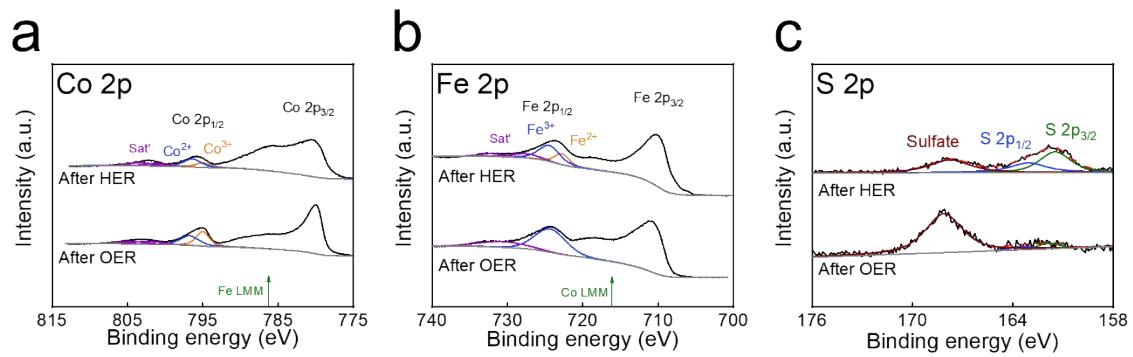
**Figure S7.** Cyclic voltammetry of (a) (Co,Fe)OOH and (b) S-(Co,Fe)OOH in non-faradaic region with different scan rates.



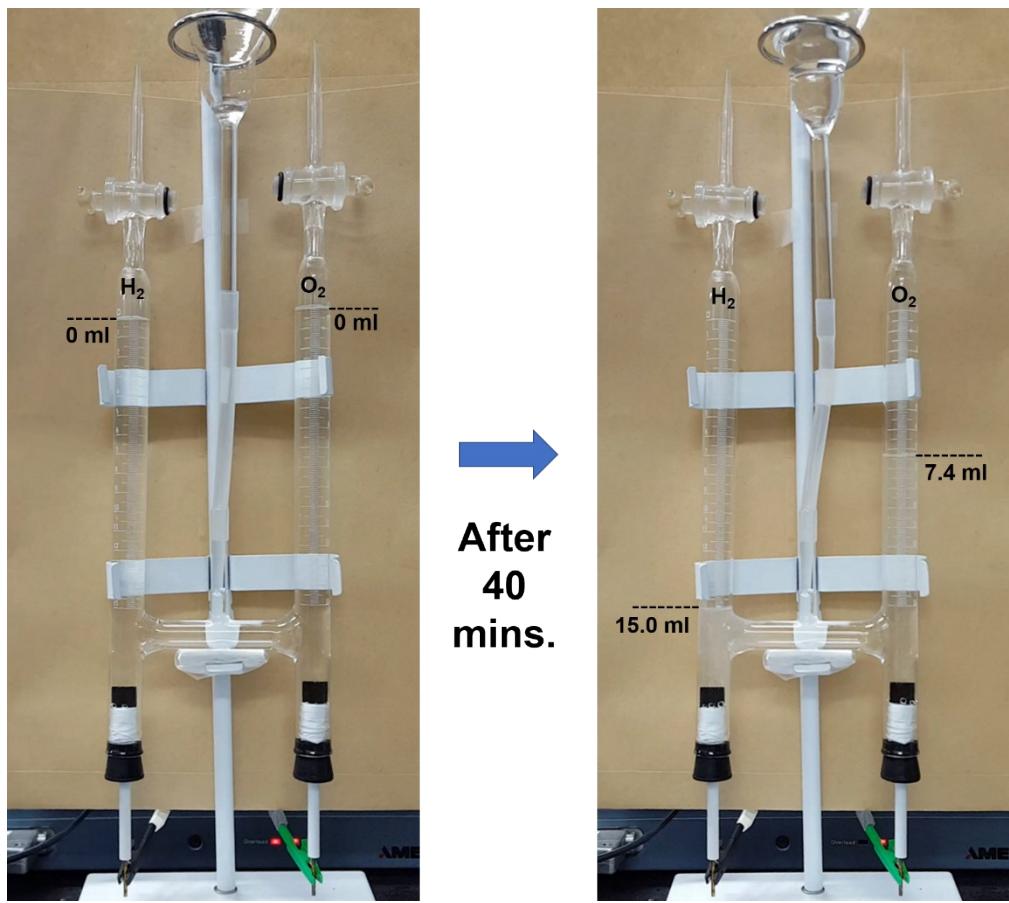
**Figure S8.** Double layer capacitance ( $C_{dl}$ ) of (a) (Co,Fe)OOH and (b) S-(Co,Fe)OOH



**Figure S9.** Electrochemical impedance spectroscopy (EIS) of Iron foam, (Co,Fe)OOH, and S-(Co,Fe)OOH for (a) OER at 1.53 V<sub>RHE</sub> and (b) HER at -0.25 V<sub>RHE</sub>.



**Figure S10.** High-resolution XPS spectra of S-(Co,Fe)OOH after durability test. (a) Co 2p, (b) Fe 2p and (c) S 2p.



**Figure S11.** Faradaic efficiency measurement of S-(Co,Fe)OOH at 50 mA/cm<sup>2</sup> for 40 mins.

**Table S1.** Comparison of the electrocatalytic activity with recently reported transition metal-based catalyst for OER in 1 M KOH electrolyte.

Catalysts	$\eta_j$ (mV)	j (mA/cm <sup>2</sup> )	Tafel slope (mV/dec)	Electrolytes	Reference
S-(Co,Fe)OOH	240	10	39	1 M KOH	This work
	268	50			
	282	100			
Cr-doped FeNi-P/NCN	240	10	72.36	1 M KOH	1
	290	50			
NiCo <sub>2</sub> S <sub>4</sub> /NF	243	10	54.9	1 M KOH	2
	320	50			
Ni <sub>2</sub> Fe <sub>1</sub> O	244	10	39	1 M KOH	3
	273	50			
Ni–Fe-LDH–MoS <sub>2</sub>	250	10	45	1 M KOH	4
Mo <sub>51</sub> Ni <sub>40</sub> Fe <sub>9</sub> nanobelts	257	10	51	1 M KOH	5
NiCo <sub>2</sub> S <sub>4</sub>	260	10	40	1 M KOH	6
Fe <sub>0.5</sub> Co <sub>0.5</sub> P	261	10	NA	1 M KOH	7
	281	50			
CeO <sub>2</sub> /Co <sub>3</sub> O <sub>4</sub> interface nanotubes	265	10	68.1	1 M KOH	8
CeO <sub>x</sub> /CoS	269	10	50	1 M KOH	9
Ni <sub>x</sub> S <sub>y</sub> -N,S-doped carbon	270	10	68.9	1 M KOH	10
Ni <sub>3</sub> FeN	280	10	46	1 M KOH	11
Fe-Ni <sub>3</sub> S <sub>2</sub> /FeNi	283	10	54	1 M KOH	12
	320	20			
CoFe(OH) <sub>x-2</sub> /Glassy carbon	293	10	67.4	1 M KOH	13
FeCoNi-ATNs/NF	295	10	52.7	1 M KOH	14
Co <sub>9</sub> S <sub>8</sub> @N-doped Carbon	302	10	67	1 M KOH	15
CuCo <sub>2</sub> S <sub>4</sub>	310	10	86	1 M KOH	16
HG-NiFe	313	10	39	1 M KOH	17
	350	50			
CoMoS <sub>3</sub> nanotube	320	10	NA	1 M KOH	18
	370	20			
NiFe-OH/NiFeP	323	10	77	1 M KOH	19
FeCo-P/C/Glassy carbon	360	10	58.4	1 M KOH	20

**Table S2.** Comparison of the electrocatalytic activity with recently reported transition metal-based catalyst for HER in 1 M KOH electrolyte.

Catalysts	$\eta_j$ (mV)	j (mA/cm <sup>2</sup> )	Tafel slope (mV/dec)	Electrolytes	Reference
S-(Co,Fe)OOH	186	-10	78	1 M KOH	This work
	236	-50			
	262	-100			
Ni–Fe–LDH–MoS <sub>2</sub>	180	-10	77	1 M KOH	4
Ni-Fe-P nanocubes	180	-10	85.5	1 M KOH	21
Ni <sub>3</sub> S <sub>2</sub> /NF	189	-10	89.3	1 M KOH	22
Cr-doped FeNi– P/NCN	190	-10	68.51	1 M KOH	23
Co <sub>3</sub> O <sub>4</sub> - MTA	190	-20	98	1 M KOH	24
CoS <sub>2</sub> HNSs	193	-10	100	1 M KOH	25
NF@NiFe LDH	198	-10	130	1 M KOH	26
FeNi@NC/CNT	202	-10	113.7	1 M KOH	27
Co <sub>0.75</sub> Fe <sub>0.25</sub> @NC	202	-10	68	1 M KOH	28
Exfoliated NiFe LDH/defective	210	-10	110	1 M KOH	29
CoP@BCN-1	215	-10	52	1 M KOH	30
Cu <sub>0.3</sub> Co <sub>2.7</sub> P/NC	220	-10	122	1 M KOH	31
Co <sub>2</sub> P	247	-10	103	1 M KOH	32
NiFeOF	253	-10	96	1 M KOH	33
Co/CoP-5	253	-10	73.8	1 M KOH	34
FeNi foam	299	-10	76.8	1 M KOH	35
NiFeP	355	-10	58.8	1 M KOH	36

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