

Synthesis and coordination of a hybrid phosphinoferroocene sulfonamide ligand

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SUPPORTING INFORMATION

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Additional structural diagrams

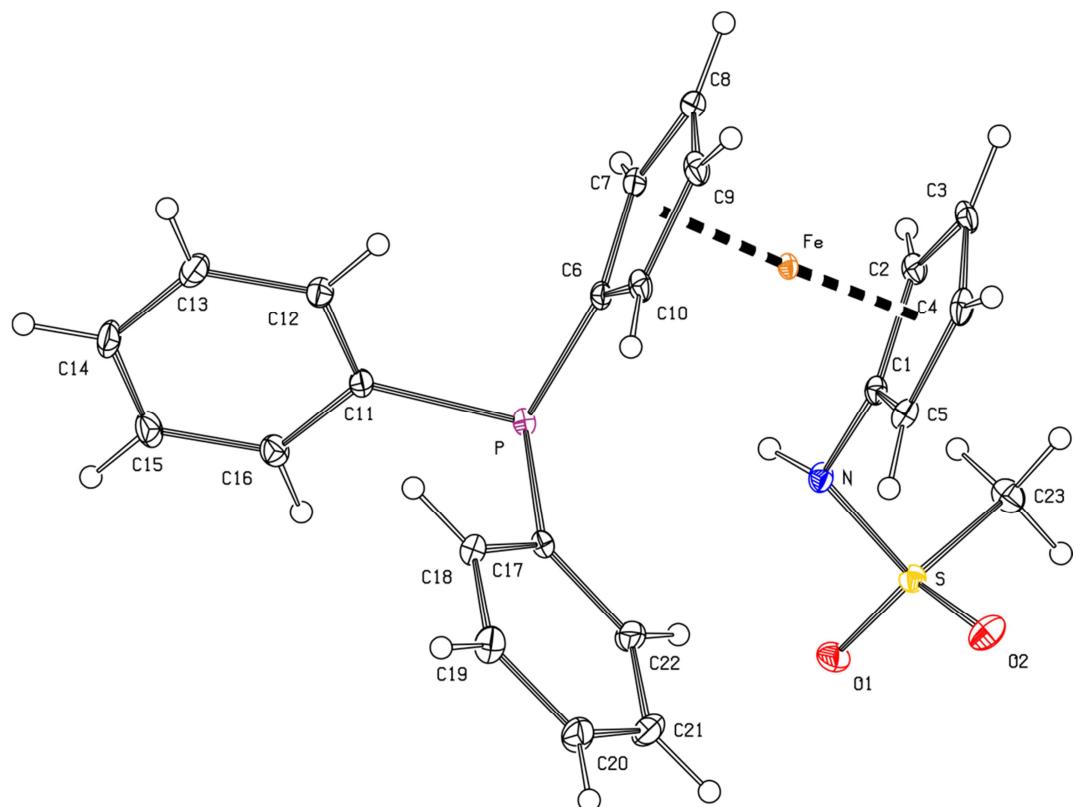


Figure S1. PLATON plot of the molecular structure of **1** showing displacement ellipsoids at the 30% probability level.

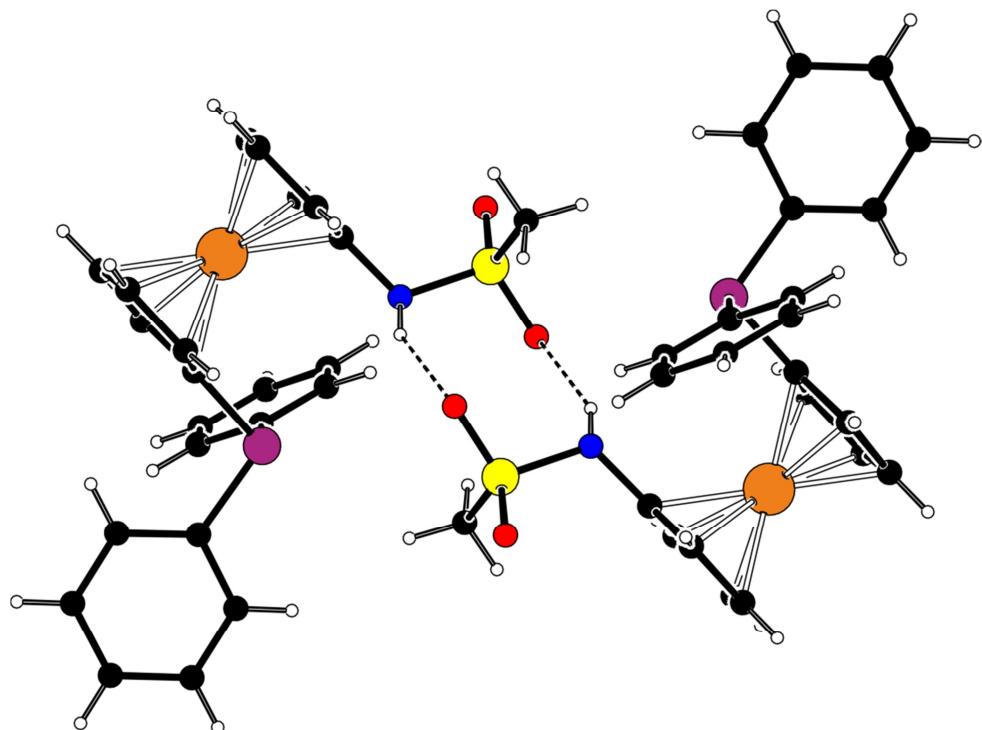


Figure S2. View of the hydrogen-bonded dimeric motif in the structure of **1**.

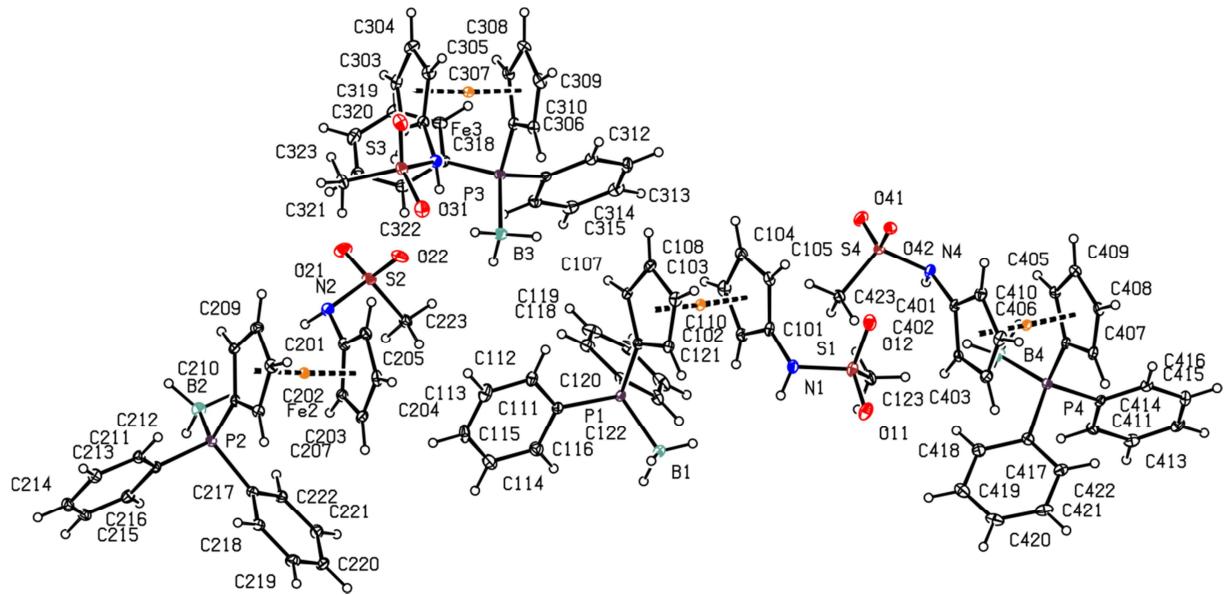


Figure S3. PLATON plot of the four independent molecules in the crystal structure of **2** at the 30% probability level.

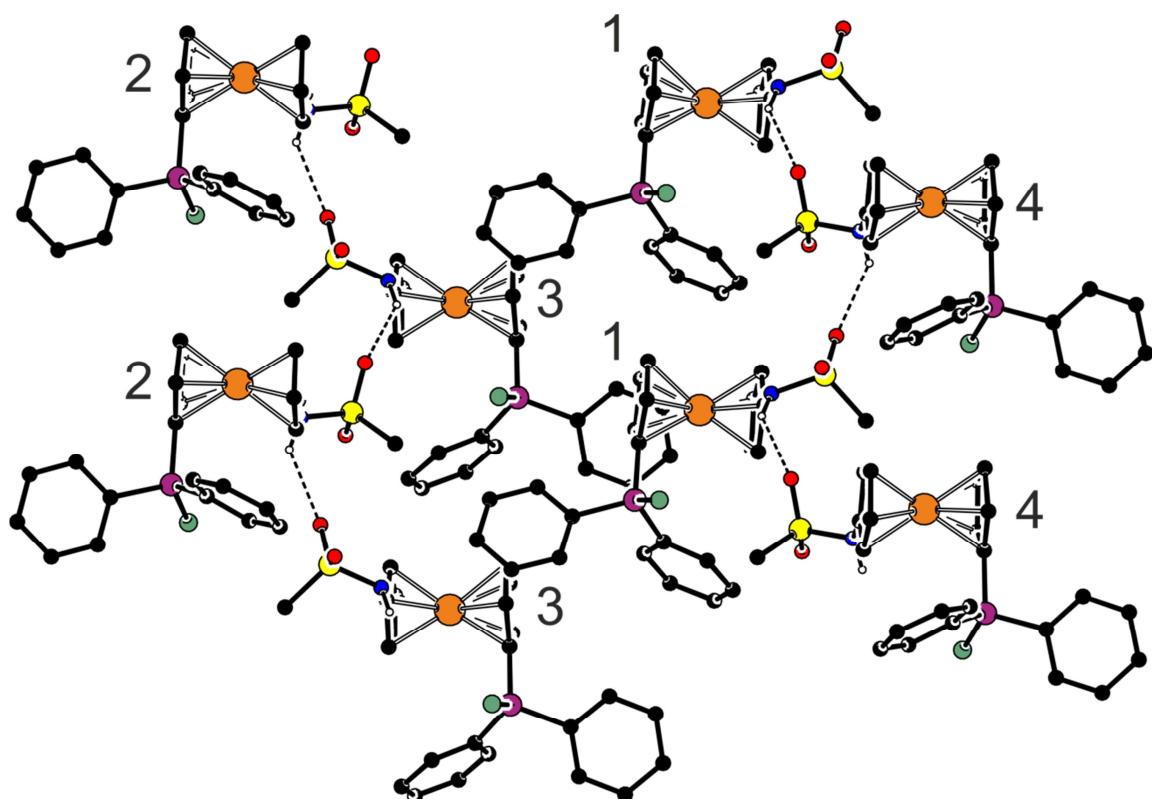


Figure S4. Section of the hydrogen-bonded chains in the structure of **2** (only the NH hydrogens are shown for clarity and hydrogen bonds are indicated by dashed lines). The independent molecules are distinguished by numbers.

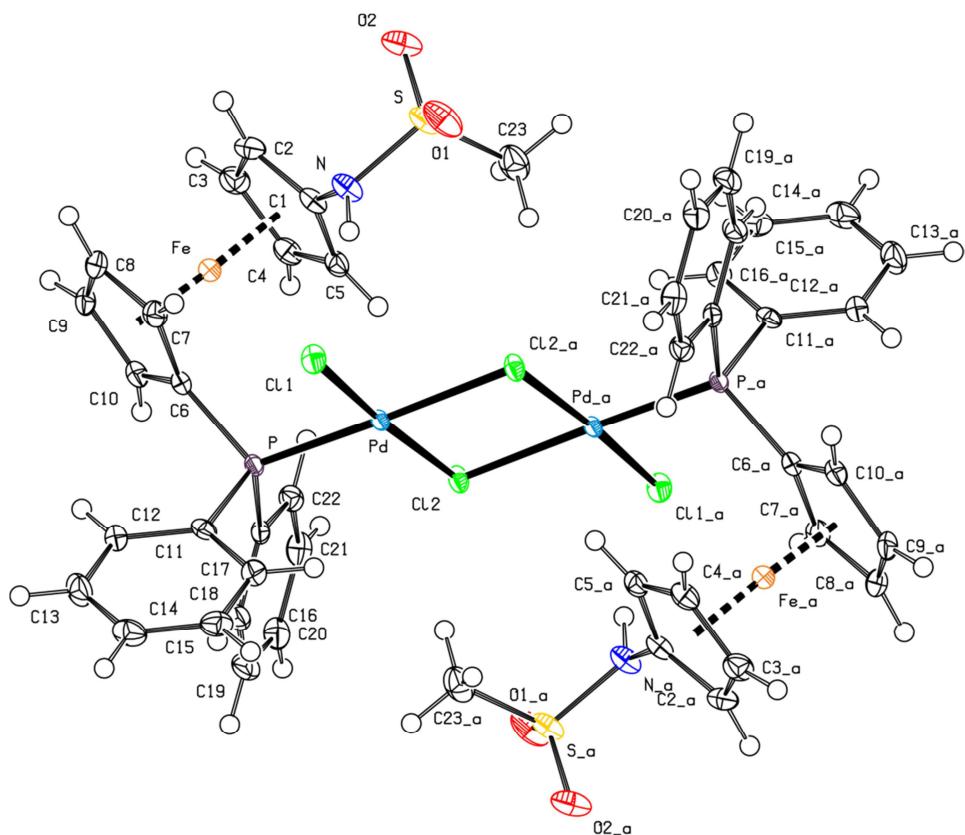


Figure S5. PLATON plot of the complex molecule in the structure of **4·2CHCl₃** with 30% probability ellipsoids (Note: the half of the molecule is generated by crystallographic inversion).

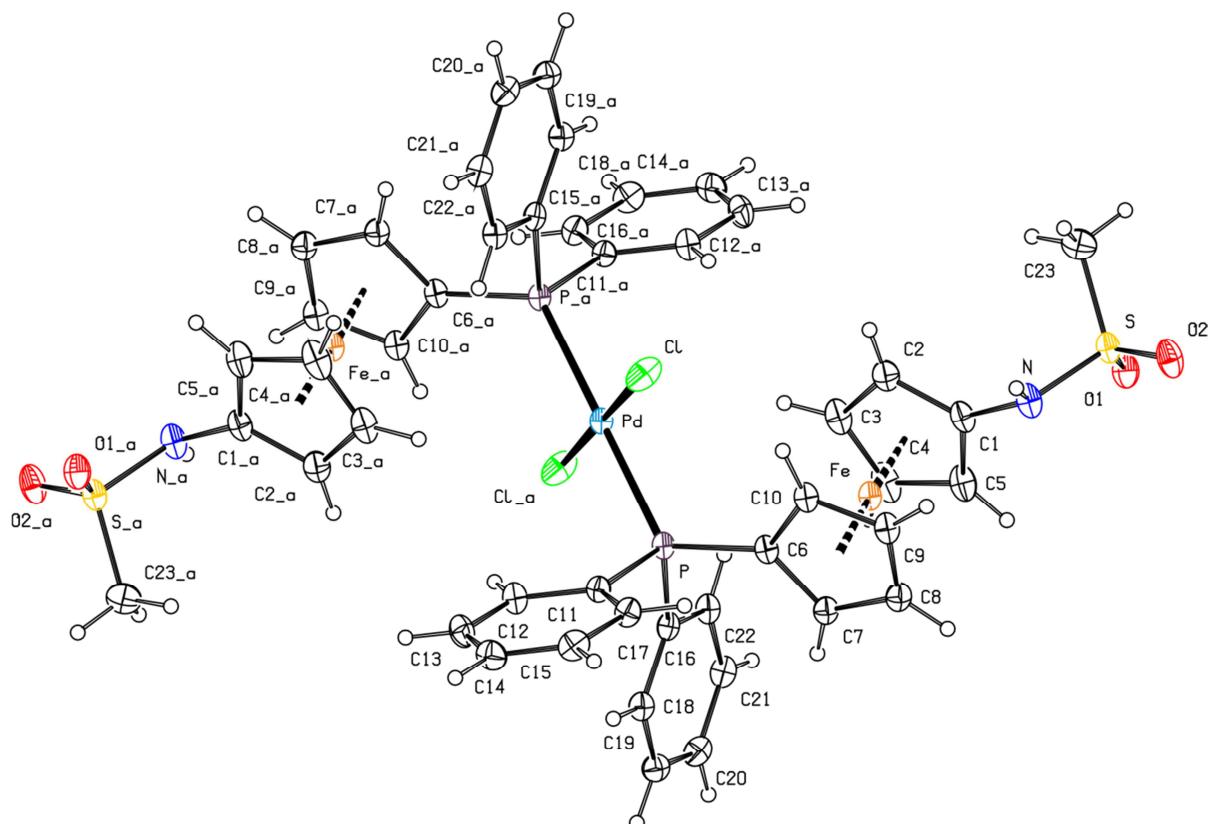


Figure S6. PLATON plot of the molecular structure of **5** showing 30% probability ellipsoids (Note: the half of the molecule is generated by crystallographic inversion).

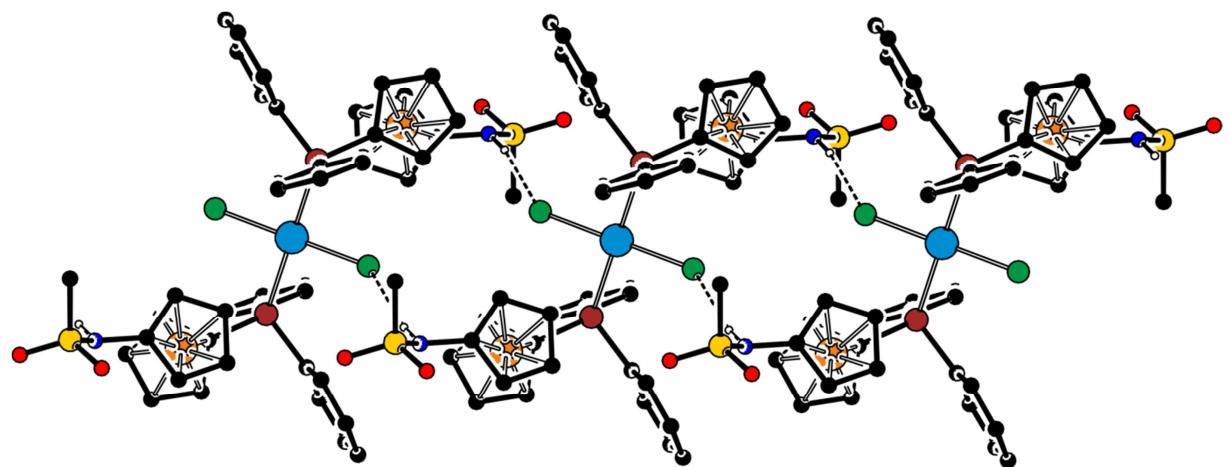


Figure S7. Section of the infinite, hydrogen-bonded chain in the structure of **5**.

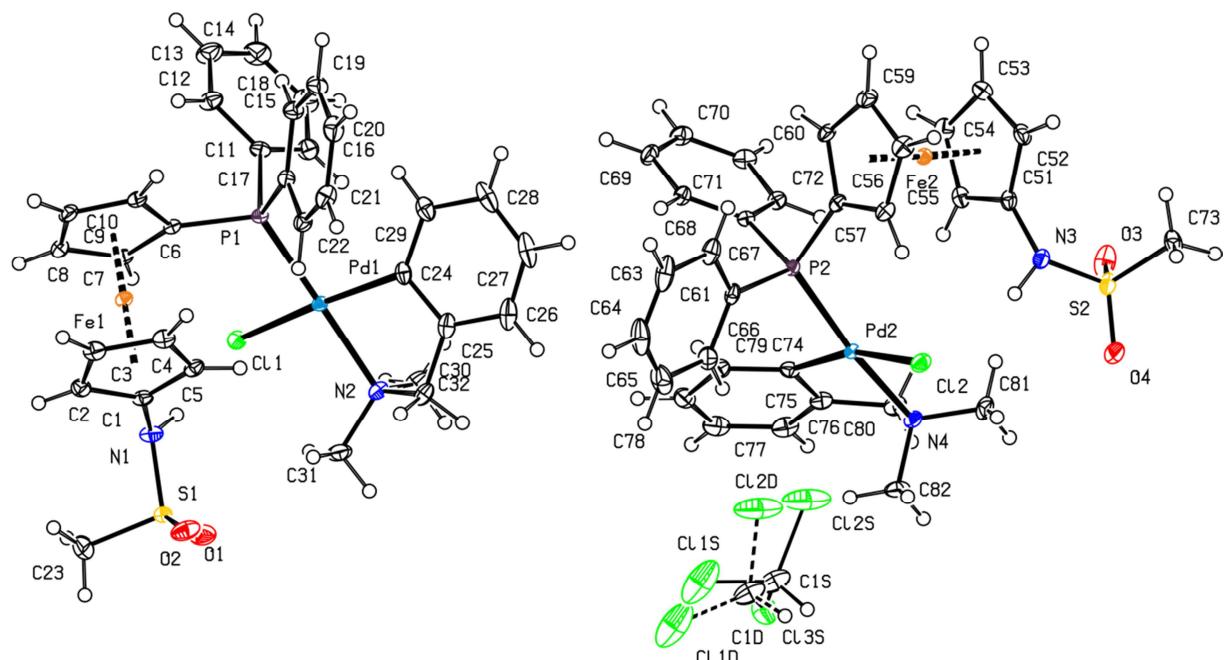


Figure S8. PLATON plot of the molecular structure of **6·½CHCl₃** (30% probability ellipsoids).

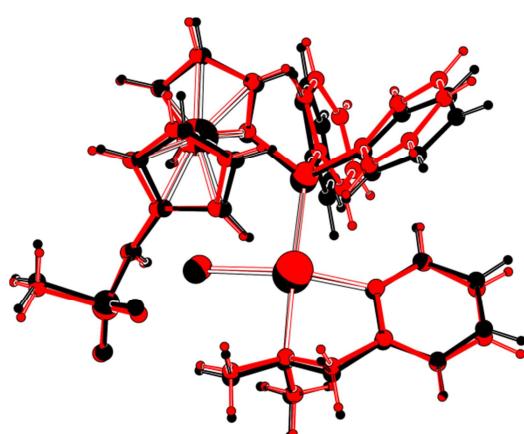


Figure S9. Least-squares overlap of the independent molecules in the structure of **6·½CHCl₃**.

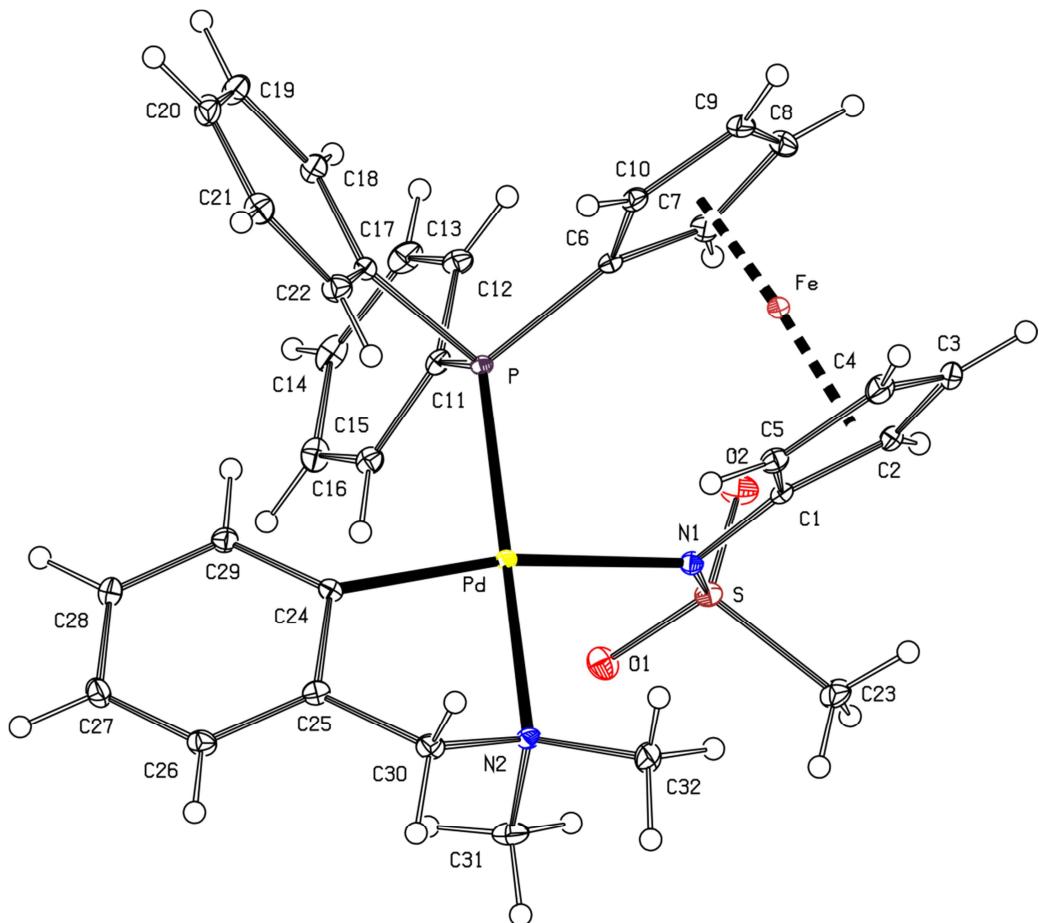


Figure S10. PLATON plot of the molecular structure of **7** (displacement ellipsoids enclose the 30% probability level).

Table S1 Selected crystallographic data and structure refinement parameters^a

Compound	1	2	4·2CHCl₃
Formula	C ₂₃ H ₂₂ FeNO ₂ PS	C ₂₃ H ₂₅ BFeNO ₂ PS	C ₄₈ H ₄₆ Cl ₁₀ Fe ₂ N ₂ O ₄ P ₂ Pd ₂ S ₂
<i>M</i>	463.29	477.13	1519.93
Crystal system	triclinic	orthorhombic	triclinic
Space group	<i>P</i> -1 (no. 2)	<i>Pca</i> 2 ₁ (no. 29) ^d	<i>P</i> -1 (no. 2)
<i>T</i> (K)	120(2)	120(2)	120(2)
<i>a</i> [Å]	8.8310(5)	18.5588(9)	10.8367(6)
<i>b</i> [Å]	9.6737(5)	9.0762(4)	11.5827(7)
<i>c</i> [Å]	12.2527(6)	51.788(3)	11.5985(7)
α [°]	99.502(2)	90	84.816(2)
β [°]	95.461(2)	90	75.910(2)
γ [°]	91.567(2)	90	79.007(2)
<i>V</i> [Å ³]	1026.69(9)	8723.3(7)	1384.6(1)
<i>Z</i>	2	16	1
μ (Mo K α) [mm ⁻¹]	0.934	7.292	1.814
Diffrns collected	20233	67871	27631
Independent diffrns	4736	16364	6372
Observed ^a diffrns	4343	15405	5698
<i>R</i> _{int} ^b [%]	2.20	3.68	2.41
No. of parameters	264	1085	326
<i>R</i> ^c obsd diffrns [%]	2.49	3.19	2.49
<i>R</i> , <i>wR</i> ^c all data [%]	2.89, 6.34	3.45, 7.98	3.03, 5.73
$\Delta\rho$ [e Å ⁻³]	0.61, -0.50	0.45, -0.43	0.80, -0.64

^a Diffractions with $I > 2\sigma(I)$. ^b Definitions: $R_{\text{int}} = \sum |F_o^2 - F_o^2(\text{mean})| / \sum F_o^2$, where $F_o^2(\text{mean})$ denoted the average intensity of symmetry-equivalent diffractions. ^c $R = \sum | |F_o| - |F_c| | / \sum |F_o|$, $wR = [\sum \{w(F_o^2 - F_c^2)^2\} / \sum w(F_o^2)^2]^{1/2}$. ^d Flack's enantiomorph parameter: 0.008(2).

Table S1 continued

Compound	5	6·½CHCl₃	7
Formula	C ₄₆ H ₄₄ Cl ₂ Fe ₂ N ₂ O ₄ P ₂ PdS ₂	C _{32.5} H _{34.5} Cl _{2.5} FeN ₂ O ₂ PPdS	C ₃₂ H ₃₃ FeN ₂ O ₂ PPdS
<i>M</i>	1103.89	799.02	702.88
Crystal system	<i>P</i> -1 (no. 2)	<i>P</i> -1 (no. 2)	<i>P</i> 2 ₁ /n (no. 14)
Space group	triclinic	triclinic	monoclinic
<i>T</i> (K)	120(2)	120(2)	120(2)
<i>a</i> [Å]	8.8684(8)	10.6520(5)	13.4617(5)
<i>b</i> [Å]	11.294(1)	16.9584(8)	13.7207(5)
<i>c</i> [Å]	12.122(1)	20.2403(9)	15.3668(7)
α [°]	101.103(2)	109.208(1)	90
β [°]	93.464(2)	101.163(2)	94.062(1)
γ [°]	105.757(2)	100.816(2)	90
<i>V</i> [Å ³]	1138.4(2)	3260.3(3)	2831.2(2)
<i>Z</i>	1	4	4
μ (Mo K α) [mm ⁻¹]	11.166	1.348	1.312
Diffrns collected	22389	63690	43435
Independent diffrns	4447	14991	6491
Observed ^a diffrns	4157	13669	5955
<i>R</i> _{int} ^b [%]	4.99	2.06	2.64
No. of parameters	278	791	365
<i>R</i> ^c obsd diffrns [%]	3.84	2.67	1.99
<i>R</i> , <i>wR</i> ^c all data [%]	4.08, 10.4	3.09, 6.37	2.31, 5.12
$\Delta\rho$ [e Å ⁻³]	1.56, -0.90	1.24, -1.47	0.56, -0.56

Copies of the NMR spectra

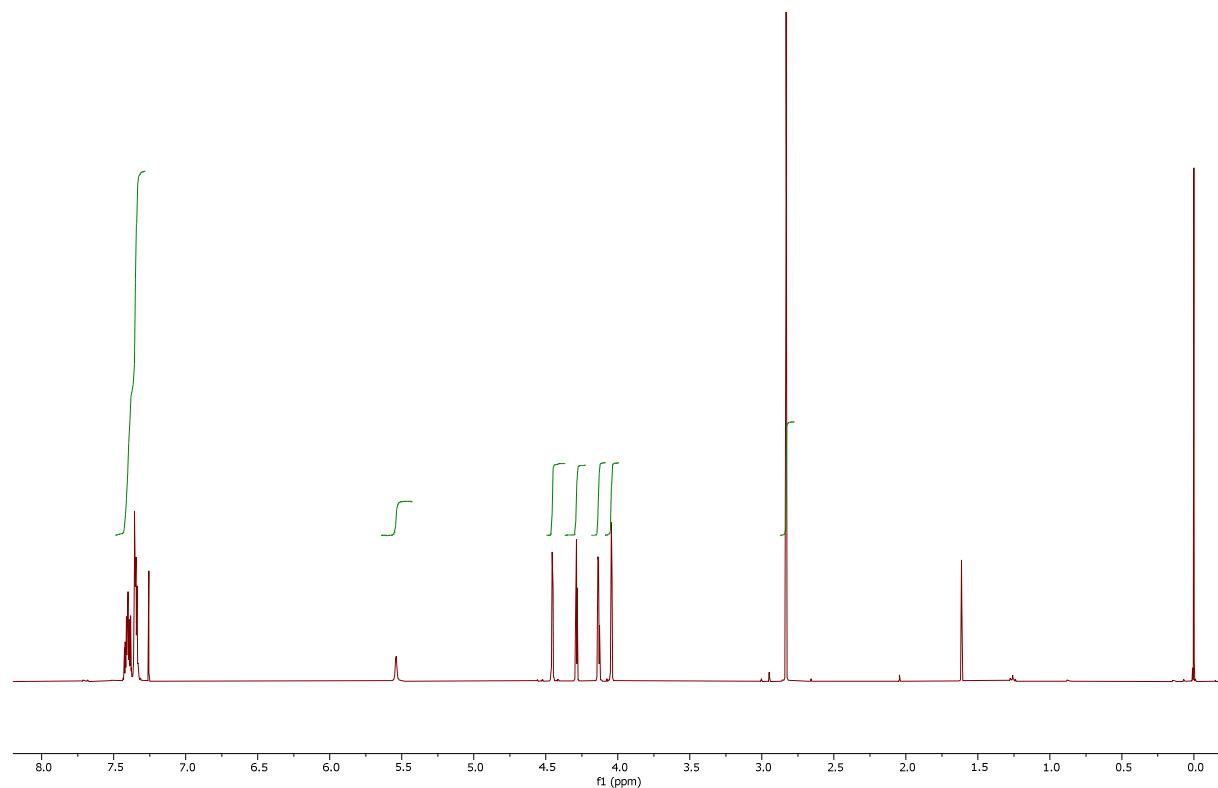


Figure S11. ^1H NMR (400 MHz, CDCl_3) spectrum of **1**.

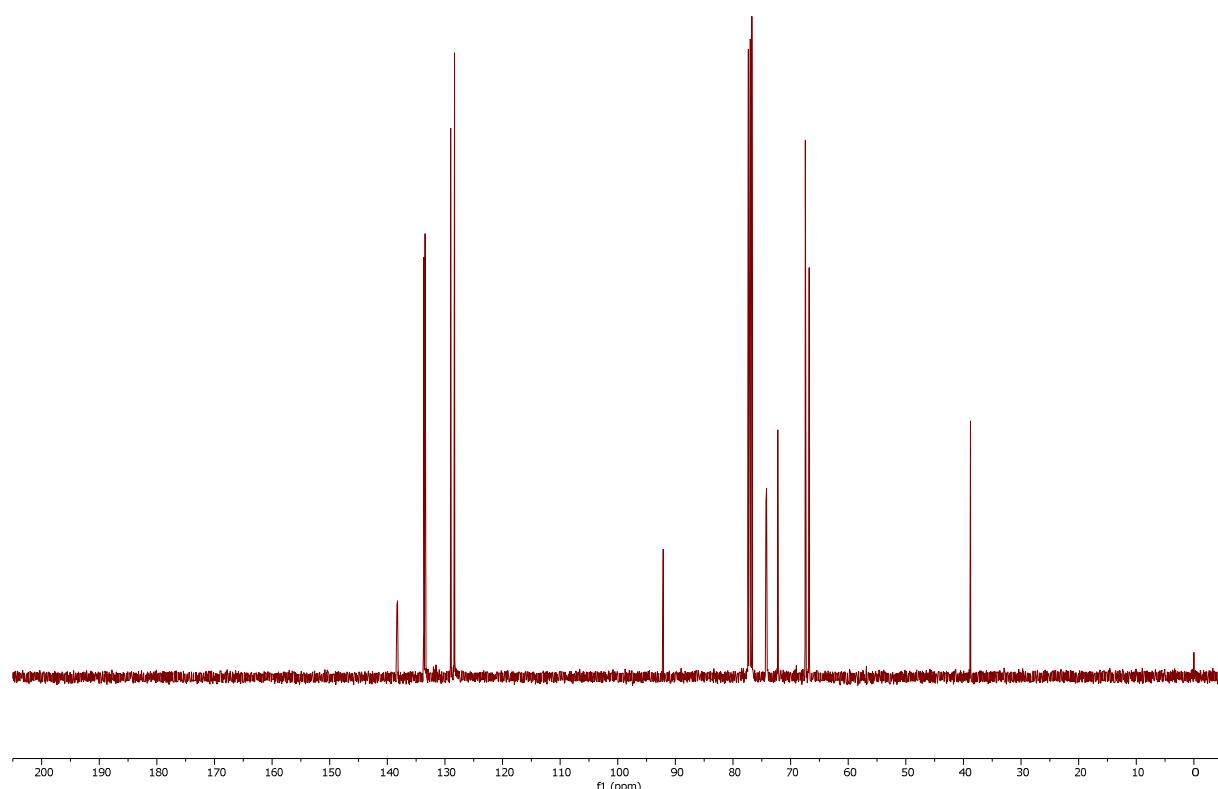


Figure S12. $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, CDCl_3) spectrum of **1**.

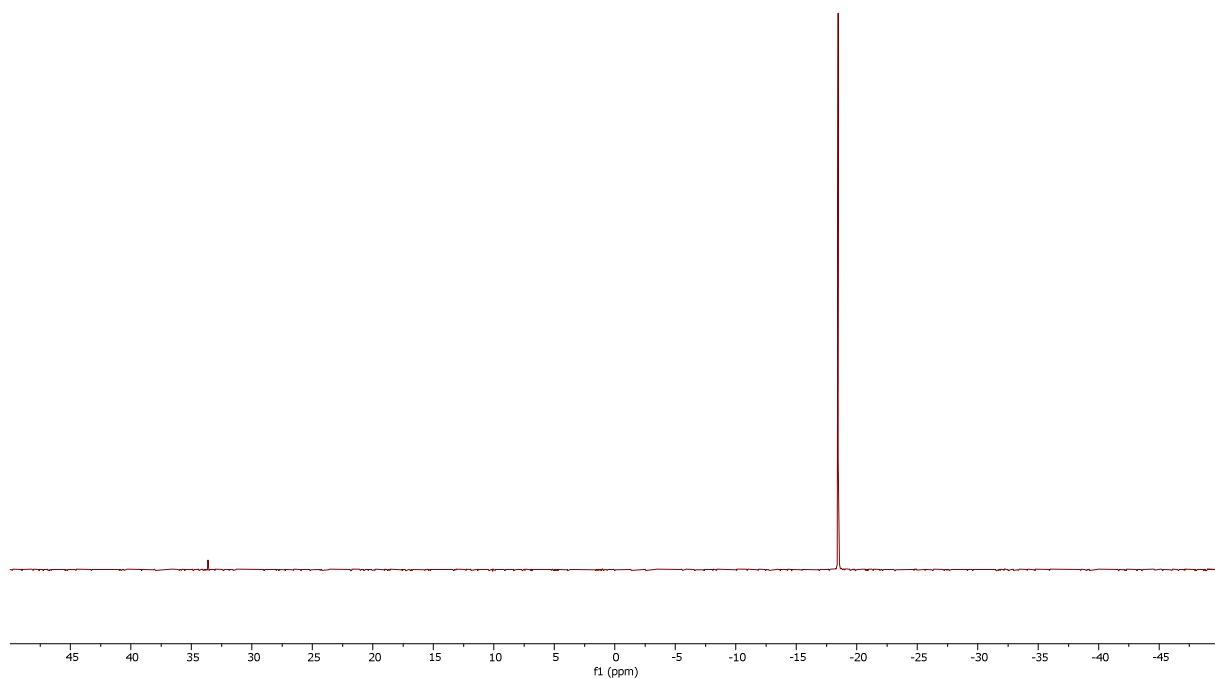


Figure S13. $^{31}\text{P}\{\text{H}\}$ NMR (162 MHz, CDCl_3) spectrum of **1**.

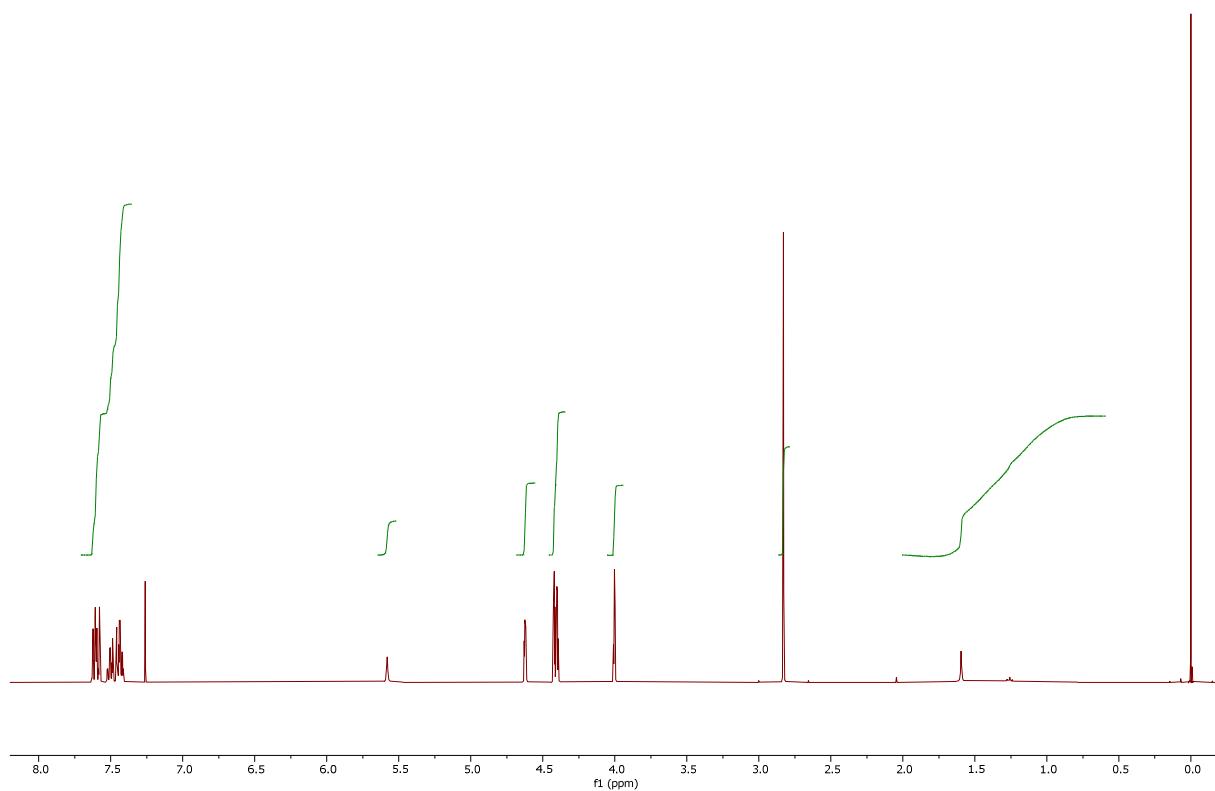


Figure S14. ^1H NMR (400 MHz, CDCl_3) spectrum of 2.

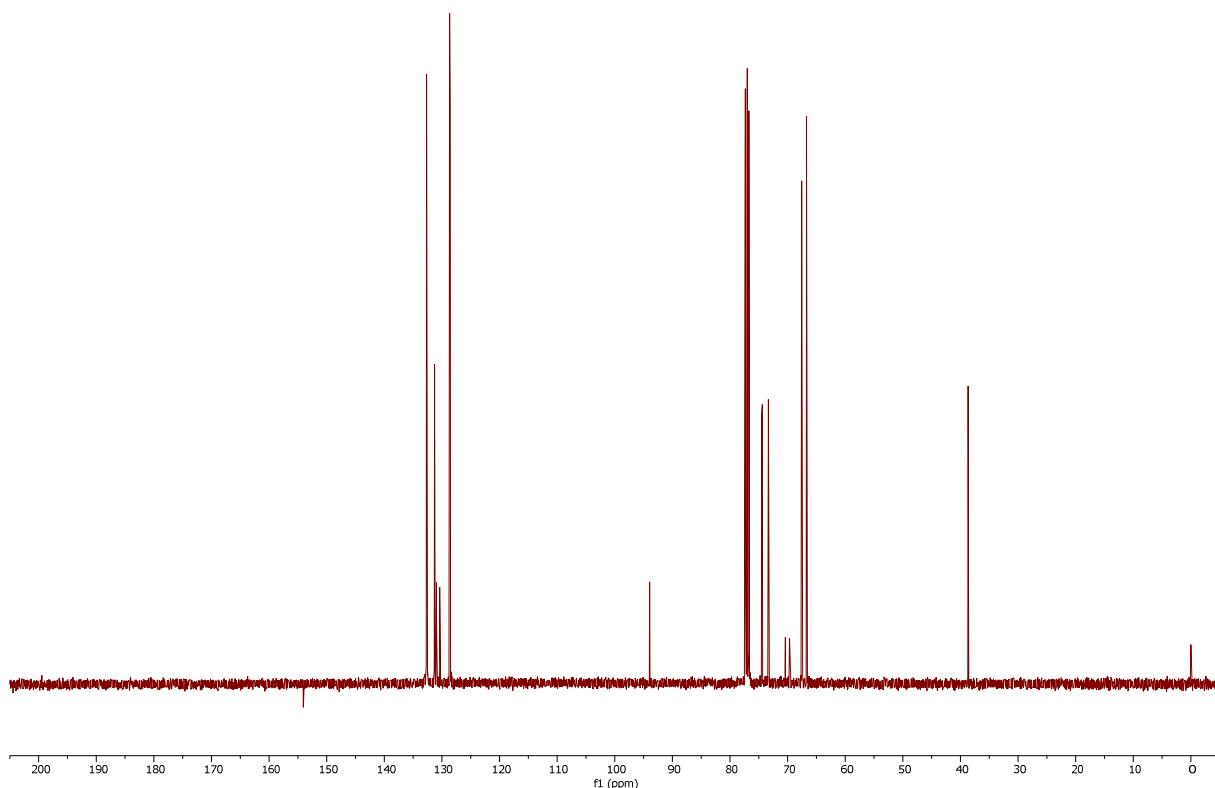


Figure S15. $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, CDCl_3) spectrum of 2.

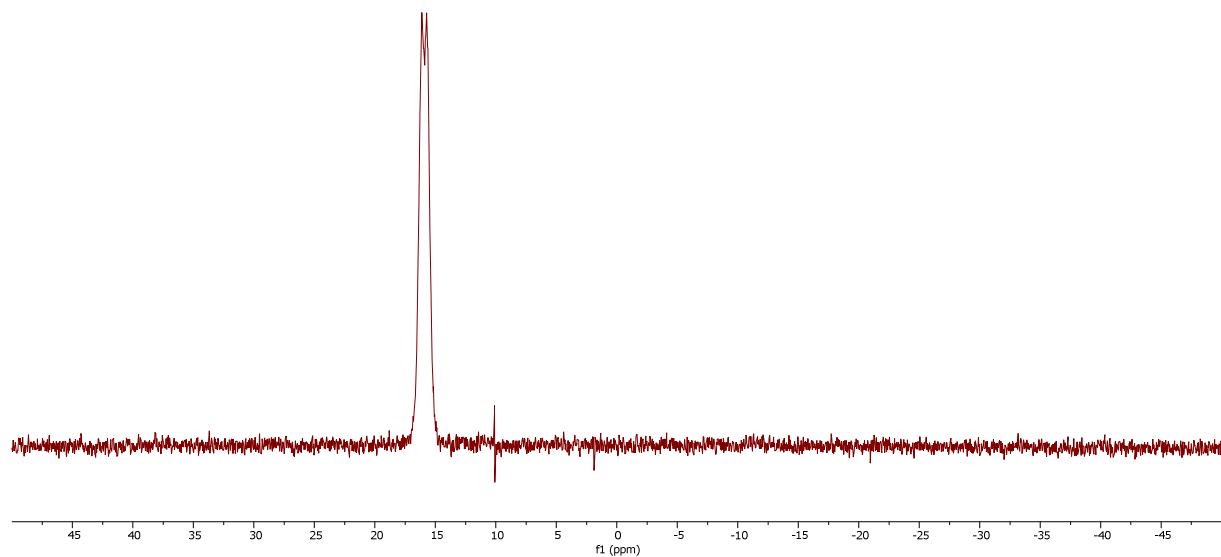


Figure S16. $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) spectrum of **2**.

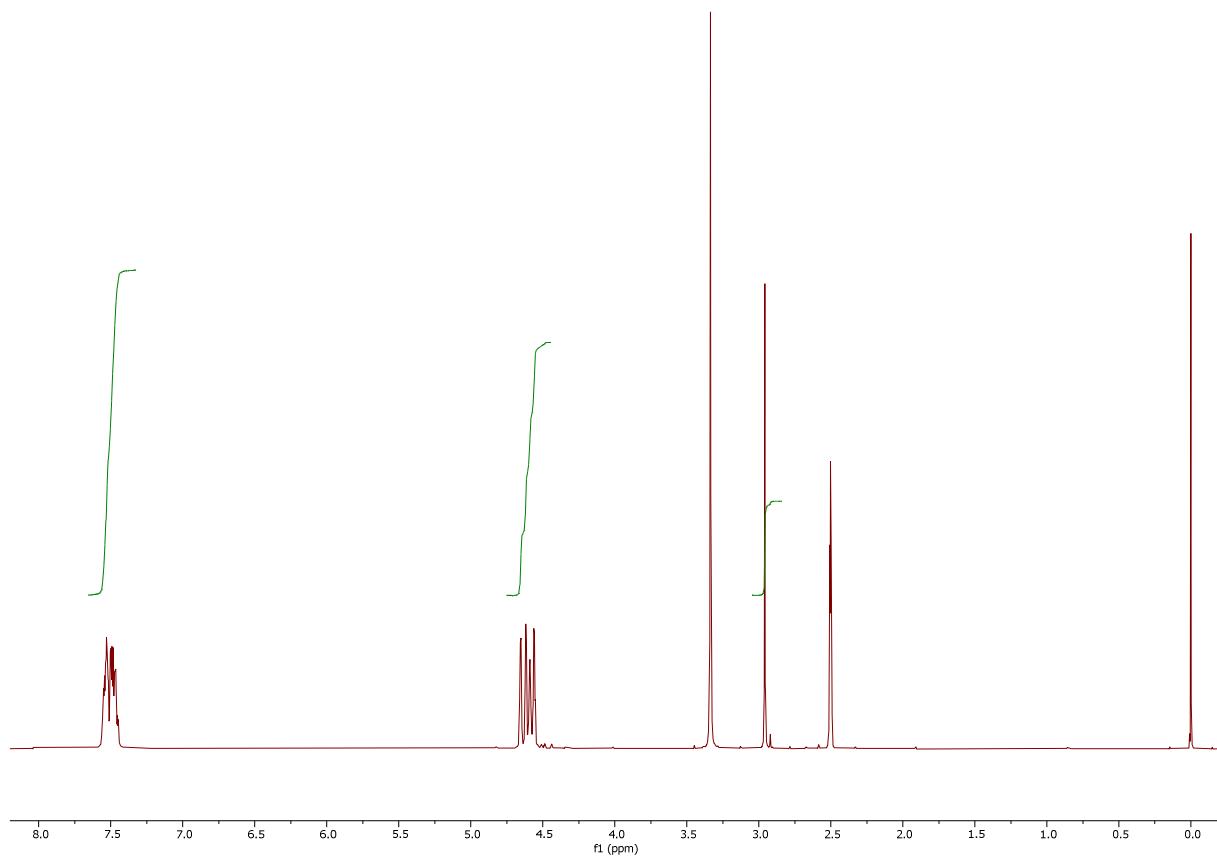


Figure S17. ^1H NMR (400 MHz, dmso- d_6) spectrum of **4**.

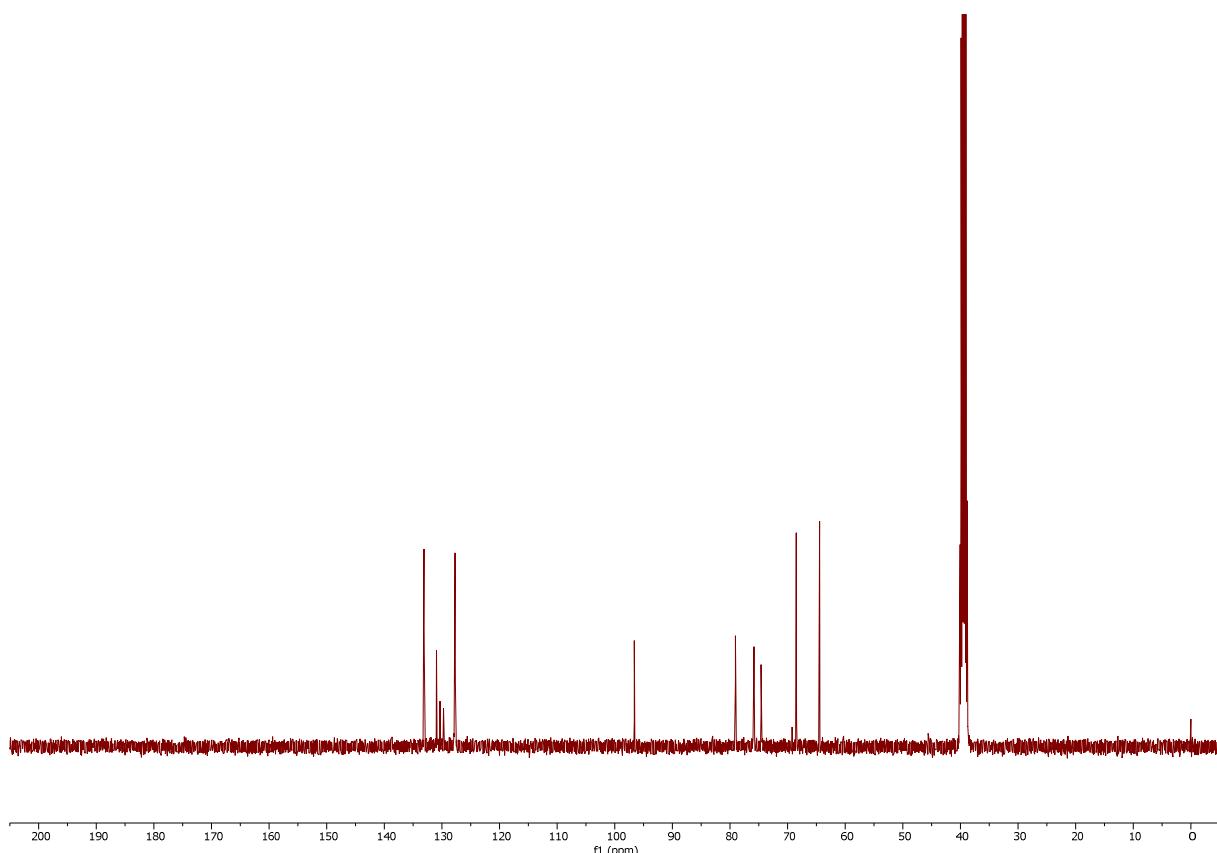


Figure S18. $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, dmso- d_6) spectrum of **4**.

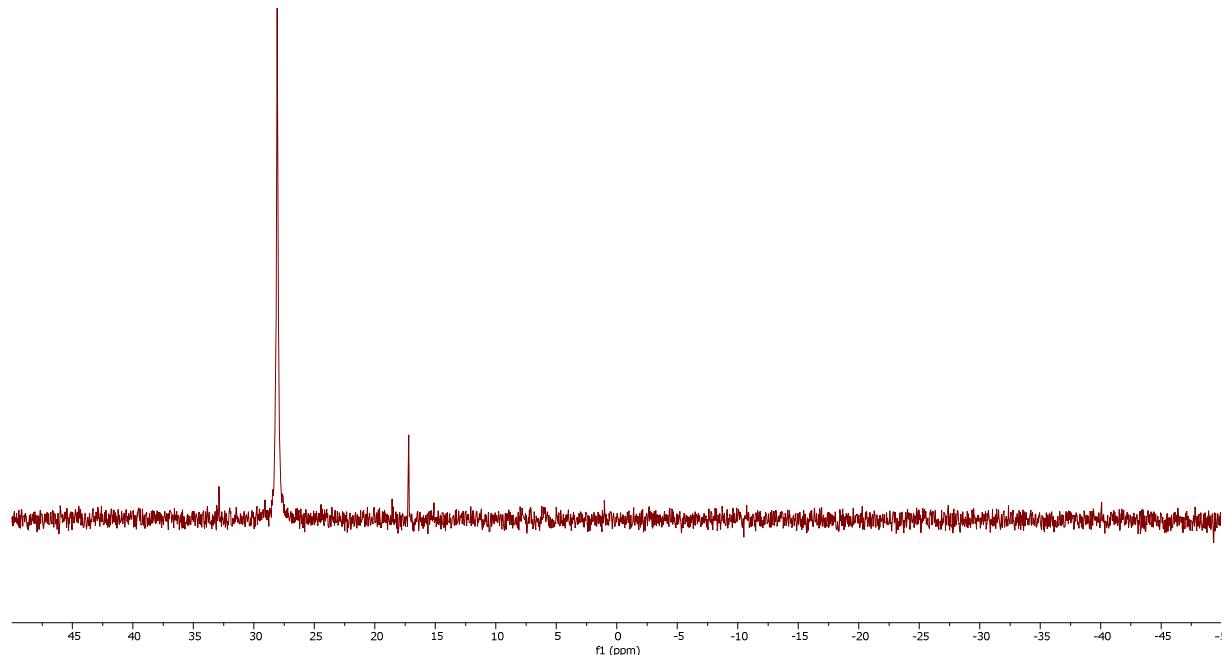


Figure S19. $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, dmso-d₆) spectrum of **4**.

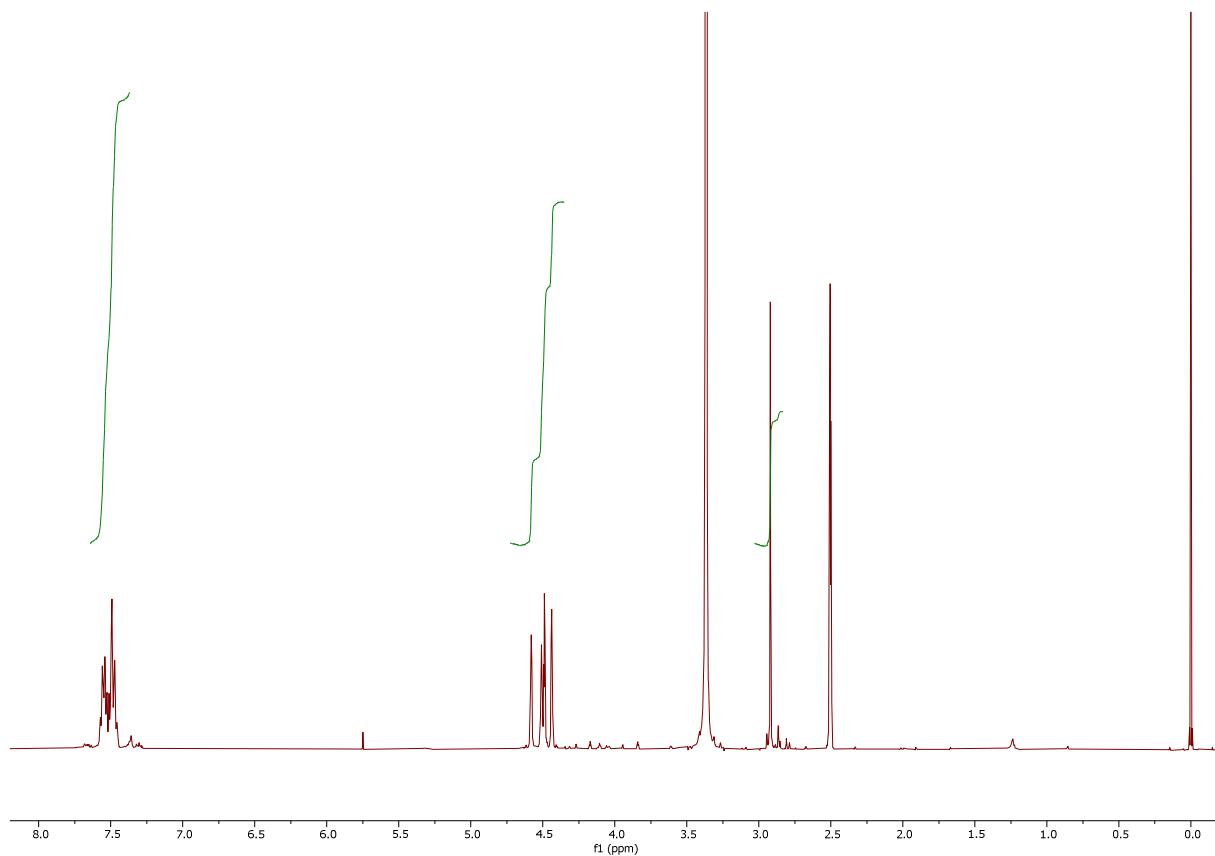


Figure S20. ^1H NMR (400 MHz, dmso-d_6) spectrum of 5.

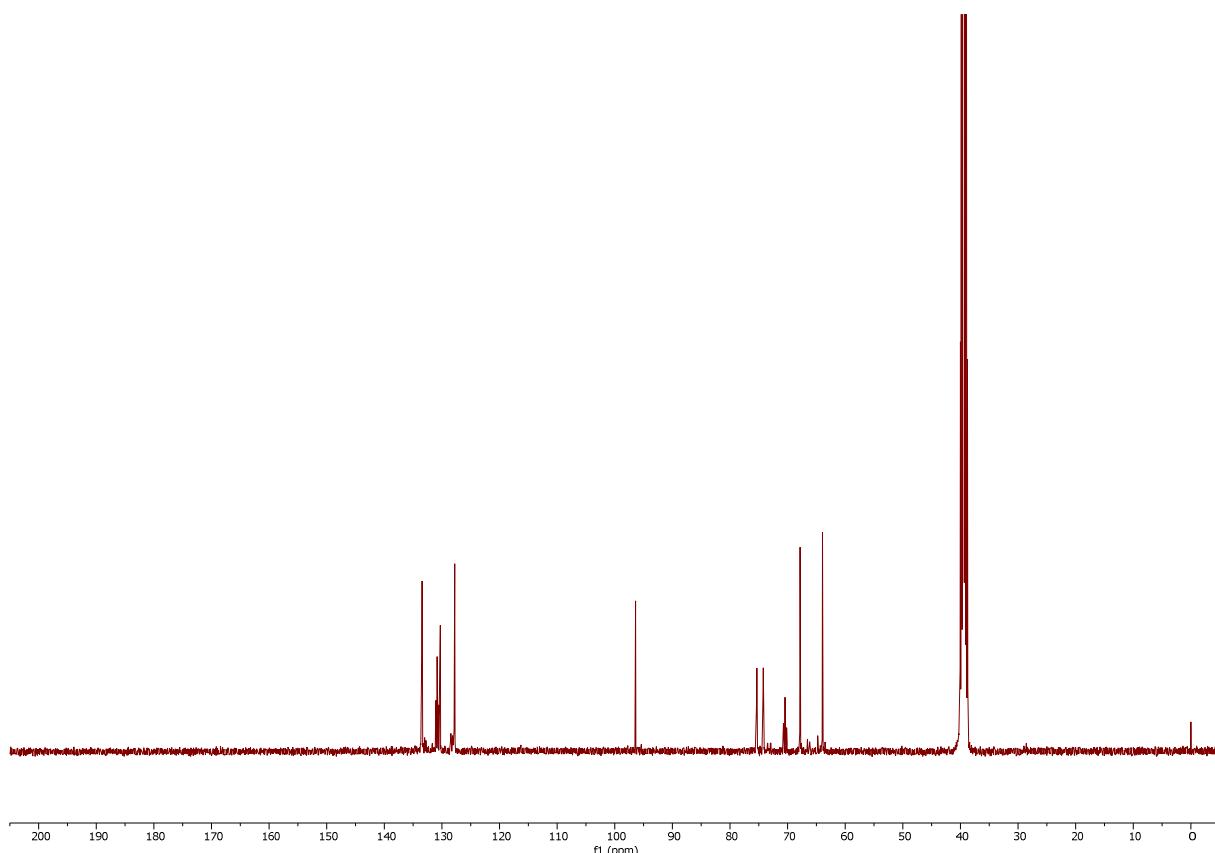


Figure S21. $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, dmso-d_6) spectrum of 5.

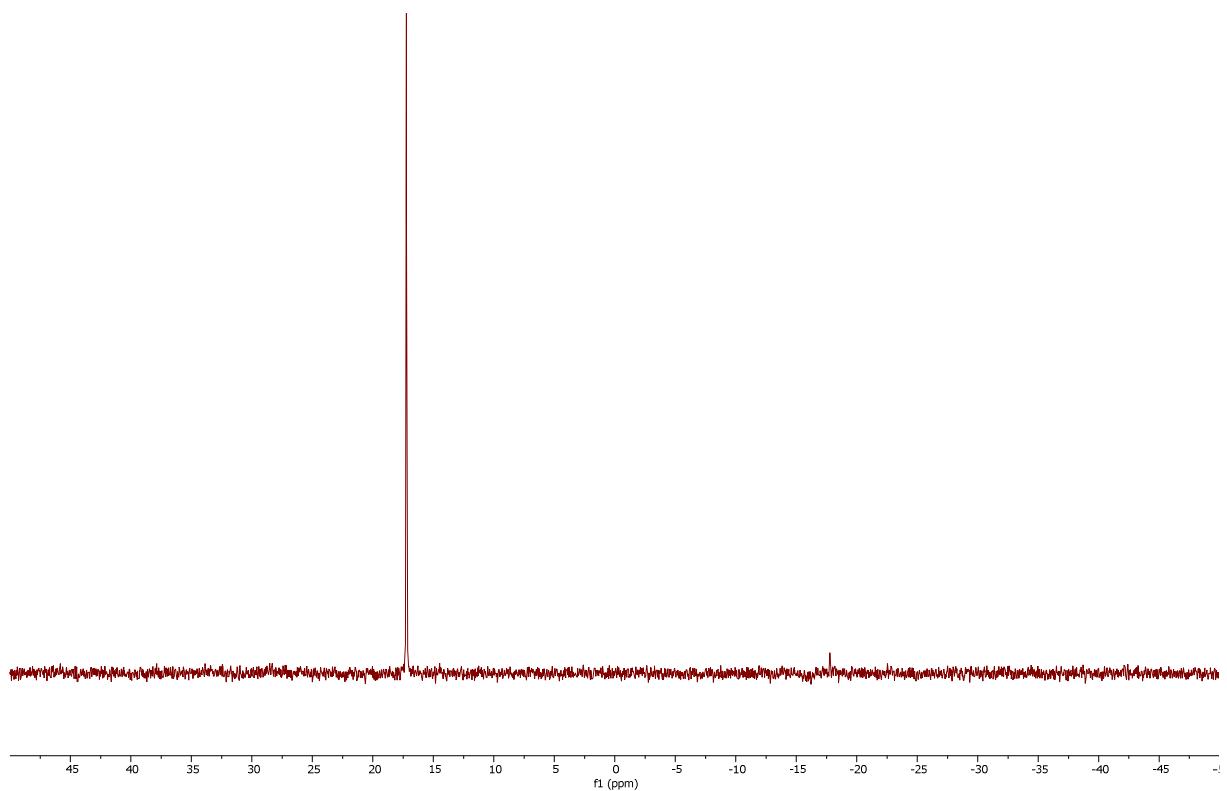


Figure S22. $^{31}\text{P}\{\text{H}\}$ NMR (162 MHz, dmso-d₆) spectrum of **5**.

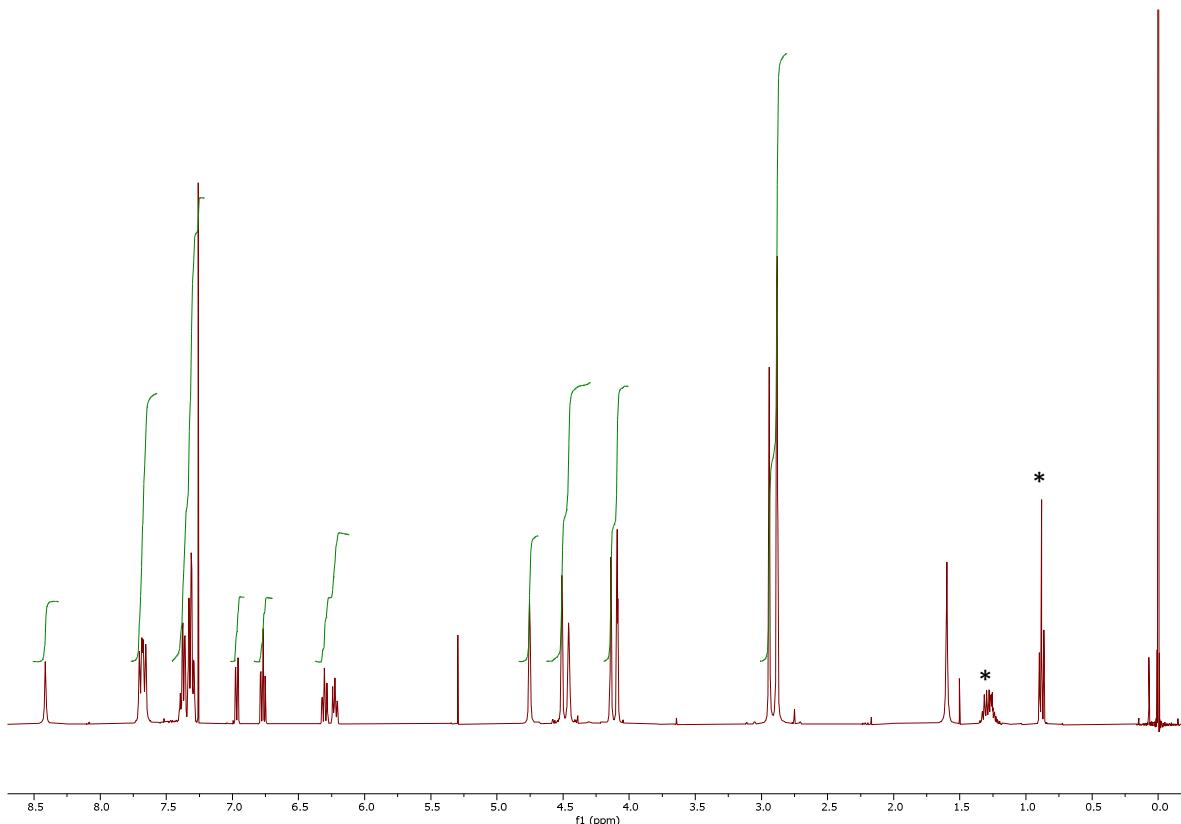


Figure S23. ^1H NMR (400 MHz, CDCl_3) spectrum of **6** (signals marked with * are due to residual pentane used to precipitate the product).

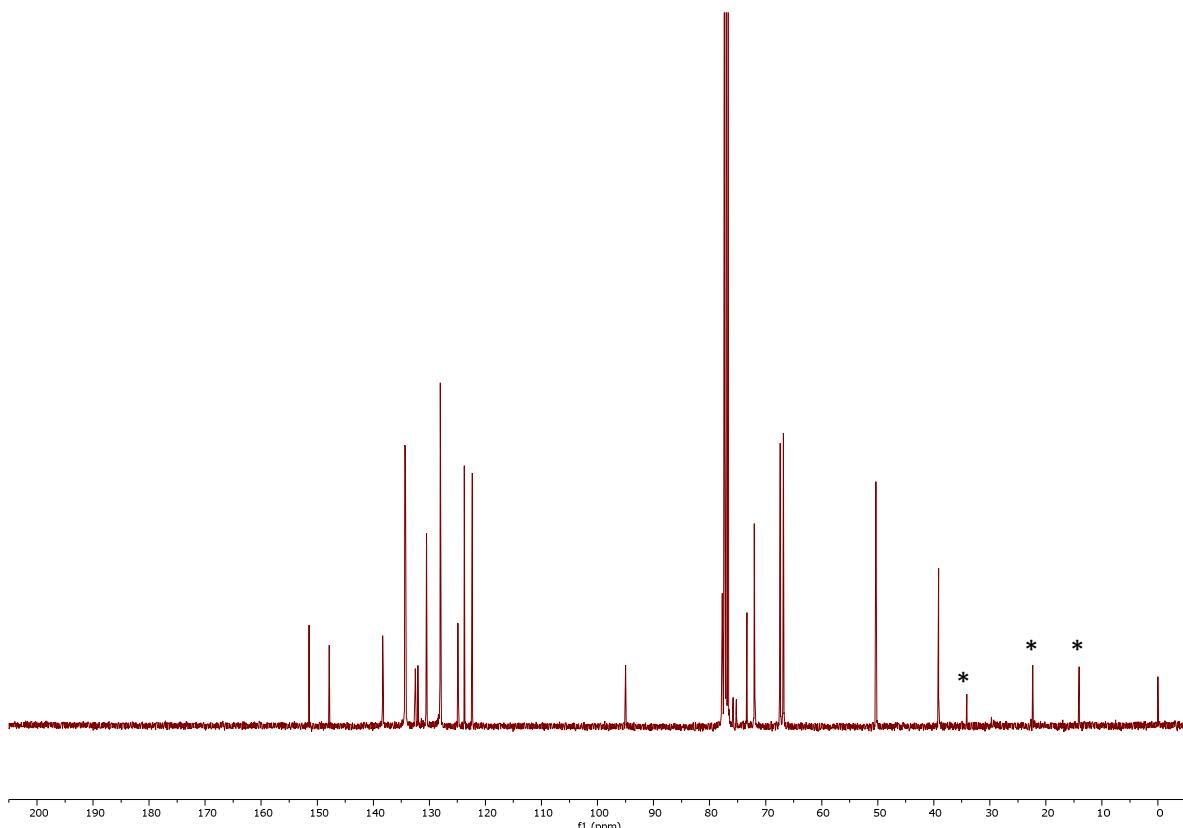


Figure S24. $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, CDCl_3) spectrum of **6** (signals marked with * are due to residual pentane used to precipitate the product).

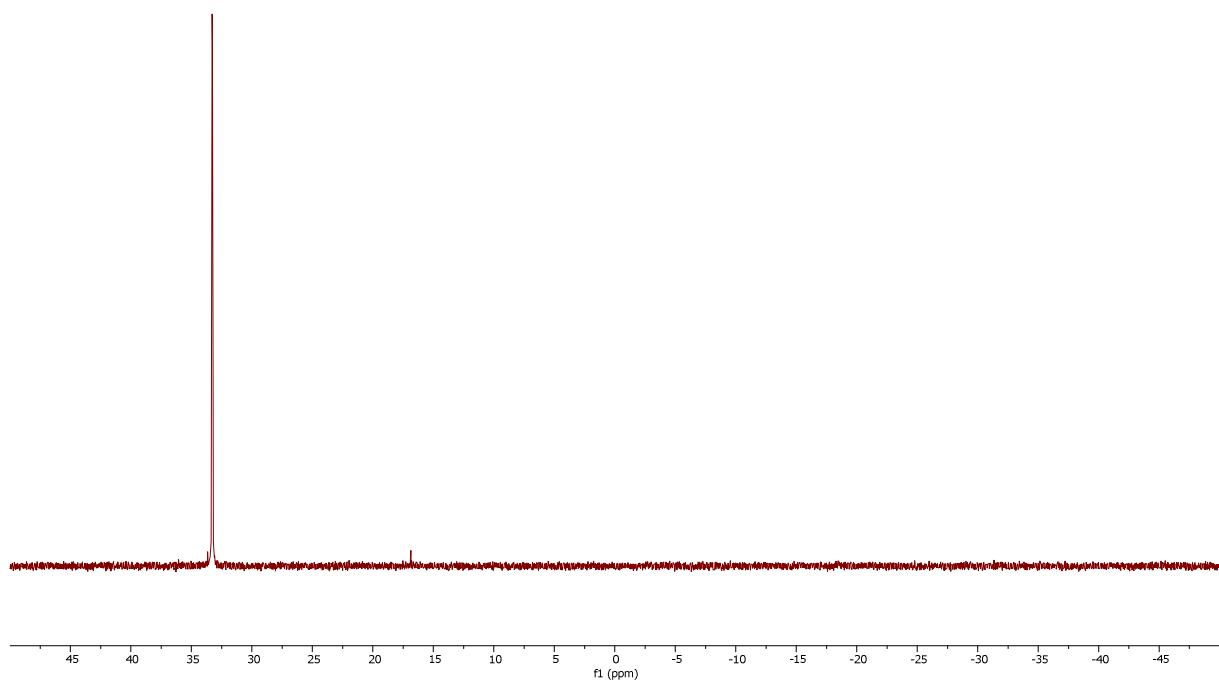


Figure S25. $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) spectrum of **6**.

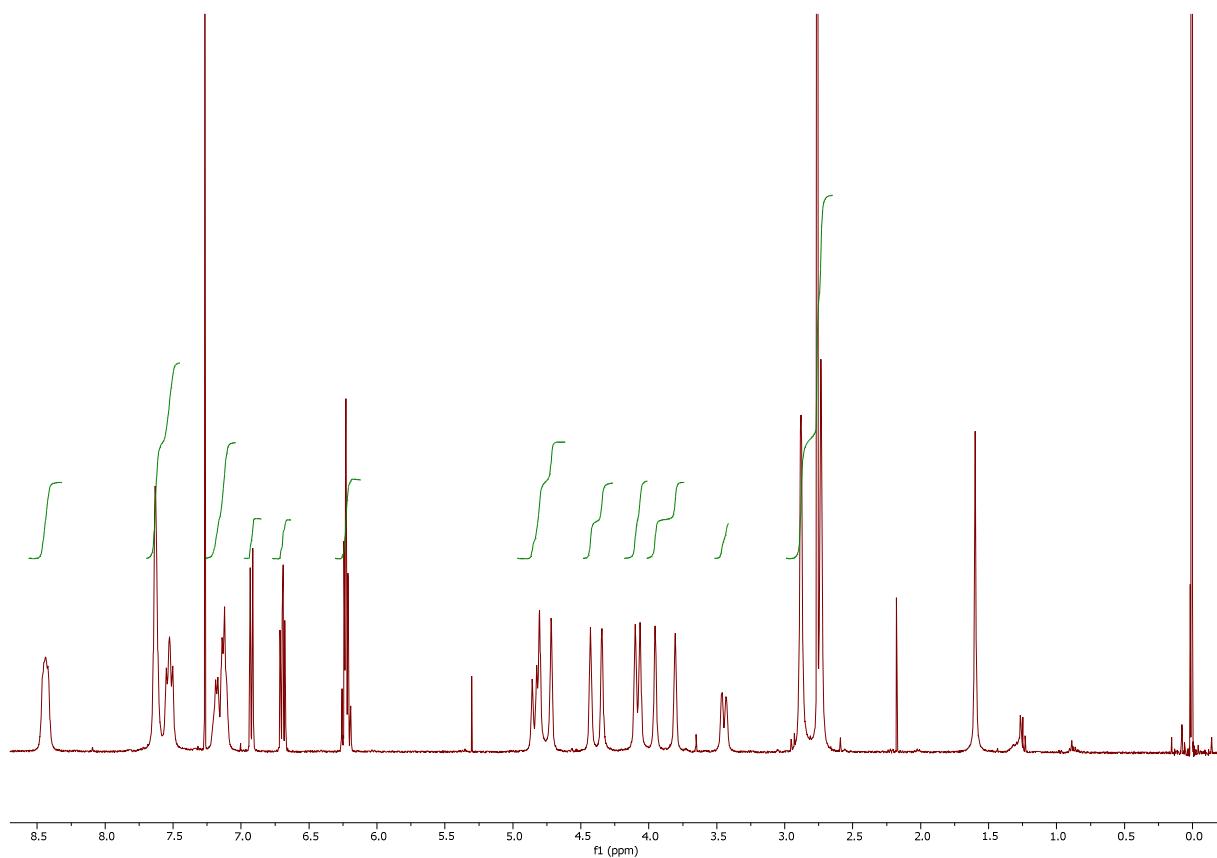


Figure S26. ^1H NMR (400 MHz, CDCl_3) spectrum of 7.

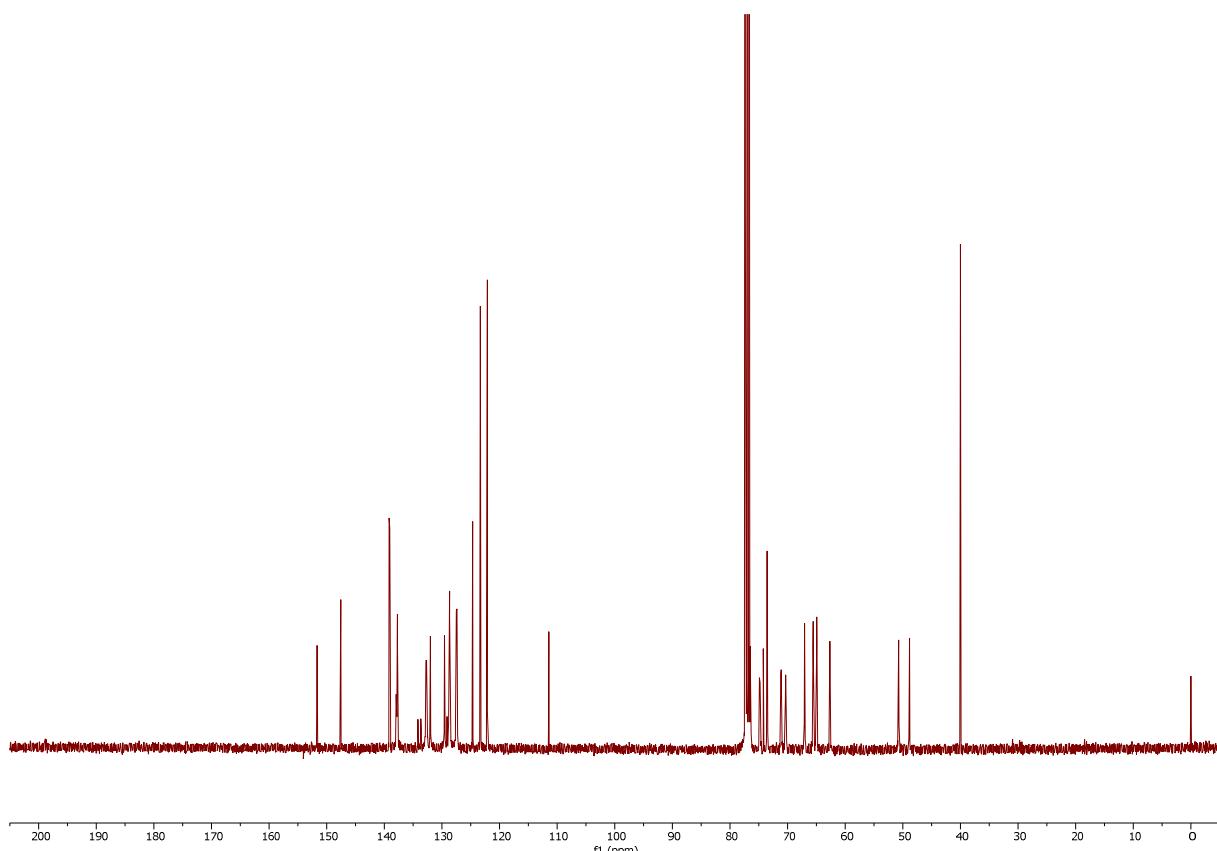


Figure S27. $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) spectrum of 7.

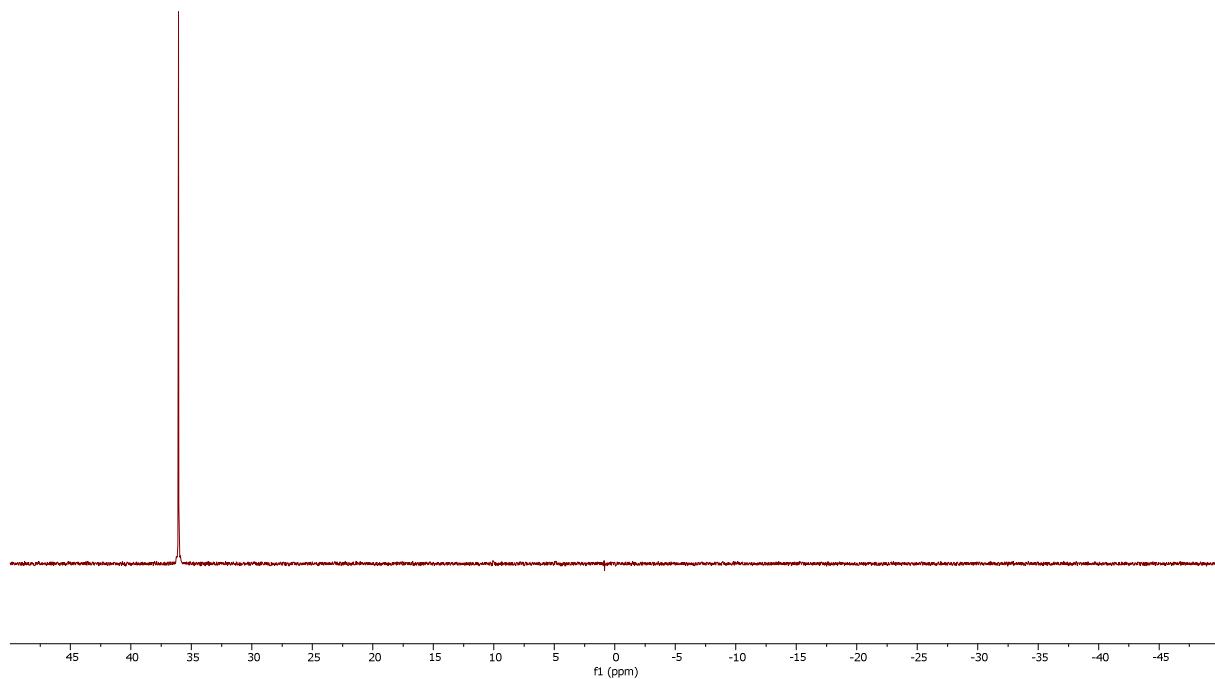


Figure S28. $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) spectrum of **7**.