

## Supplementary Information

### Designing neurotransmitter dopamine functionalized naphthalene diimide molecular architectures for high-performance organic supercapacitor electrode materials

Madan R. Biradar,<sup>§,†,¶</sup> Akshay V. Salkar,<sup>‡,¶</sup> Pranay P. Morajkar,<sup>\*,‡</sup>, Sheshanath V. Bhosale<sup>\*,‡</sup>,  
Sidhanath V. Bhosale,<sup>§,†\*</sup>

<sup>§</sup>Polymers and Functional Materials Division, CSIR-Indian Institute of Chemical Technology, Hyderabad-500 007, Telangana, India; <sup>†</sup>Academy of Scientific and Innovative Research (AcSIR), Ghaziabad- 201002, Uttar Pradesh, India; <sup>‡</sup>School of Chemical Sciences, Goa University, Taleigao Plateau- 403206, Goa, India.

<sup>¶</sup>= equal contribution from the first two authors

\*Corresponding Authors: pranay@unigoa.ac.in; svbhosale@unigoa.ac.in; bhosale@iict.res.in

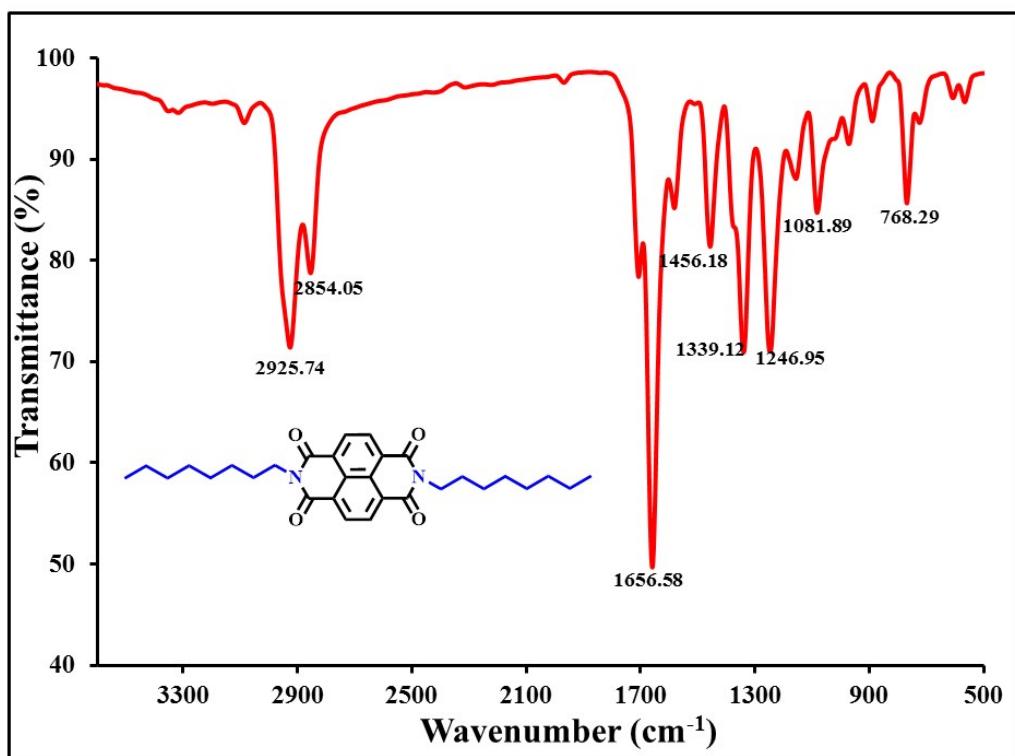


Fig. S1 FT-IR spectra of NDI.

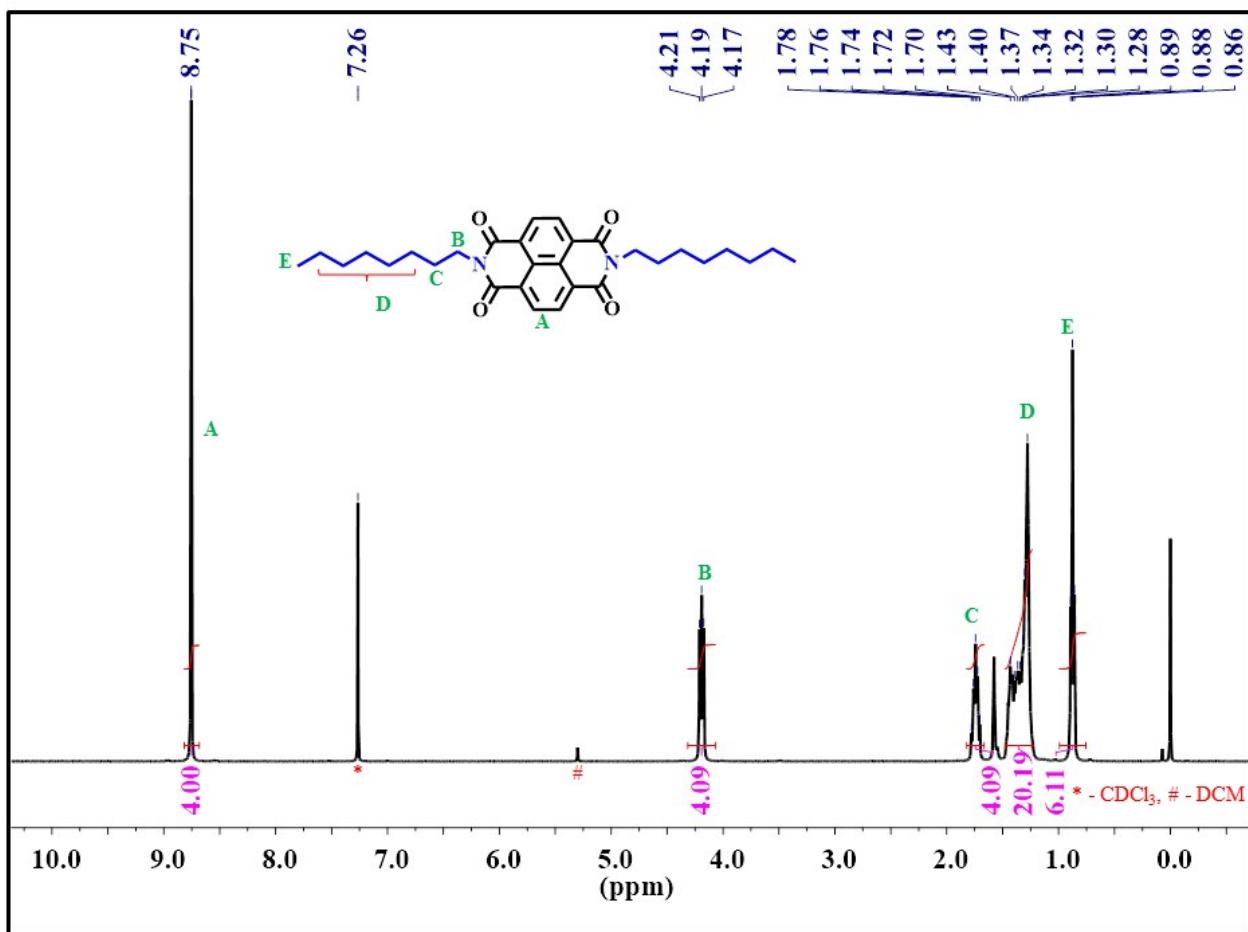
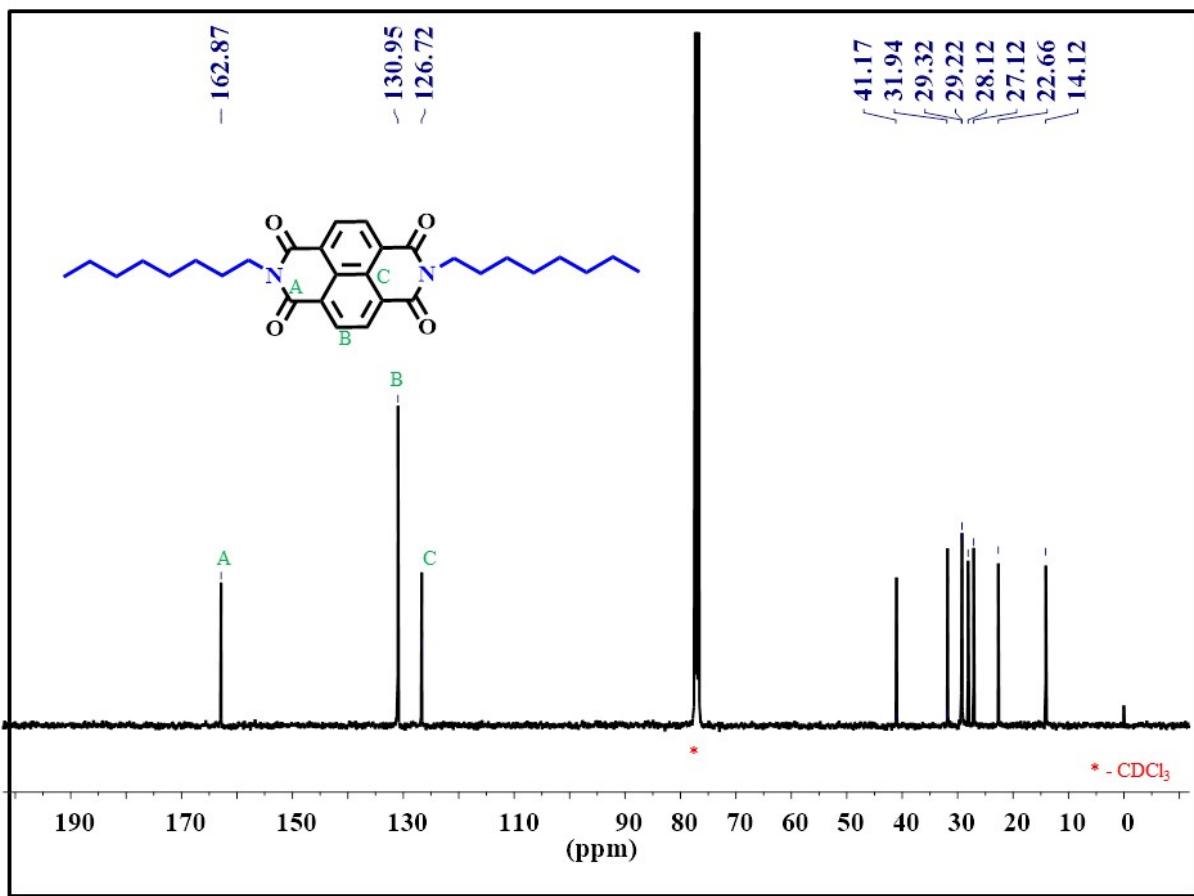
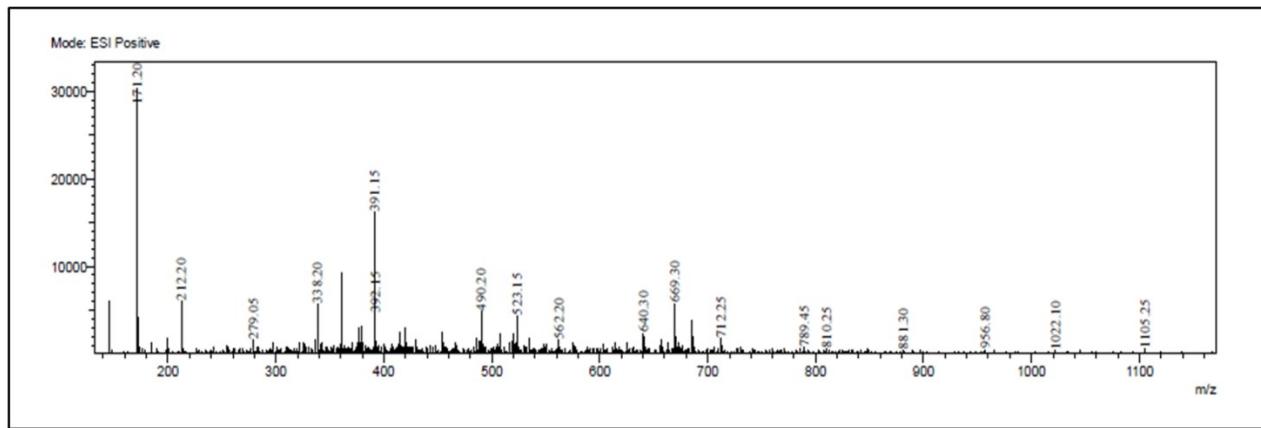


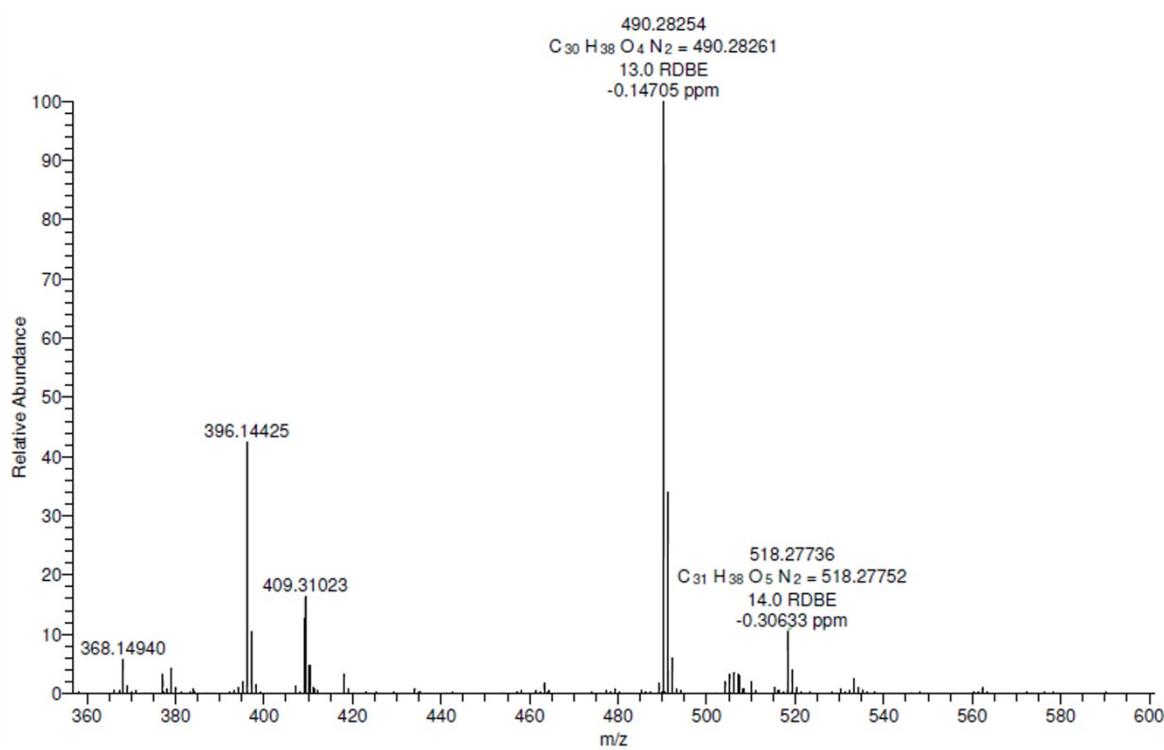
Fig. S2 <sup>1</sup>H NMR spectra of NDI.



**Fig. S3**  $^{13}\text{C}$  NMR spectra of NDI.



**Fig. S4** LR-MS spectrum of NDI.



**Fig. S5** HRMS spectrum of NDI

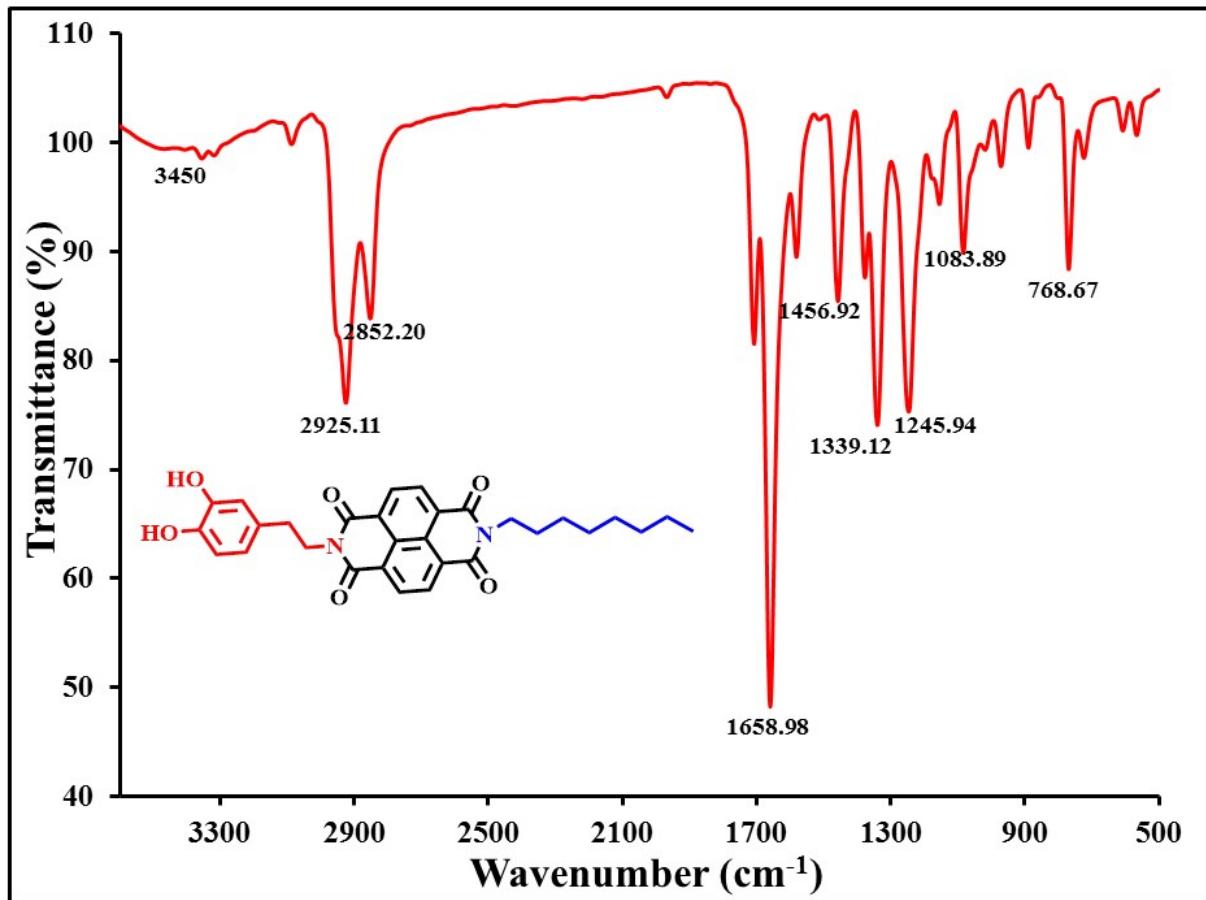
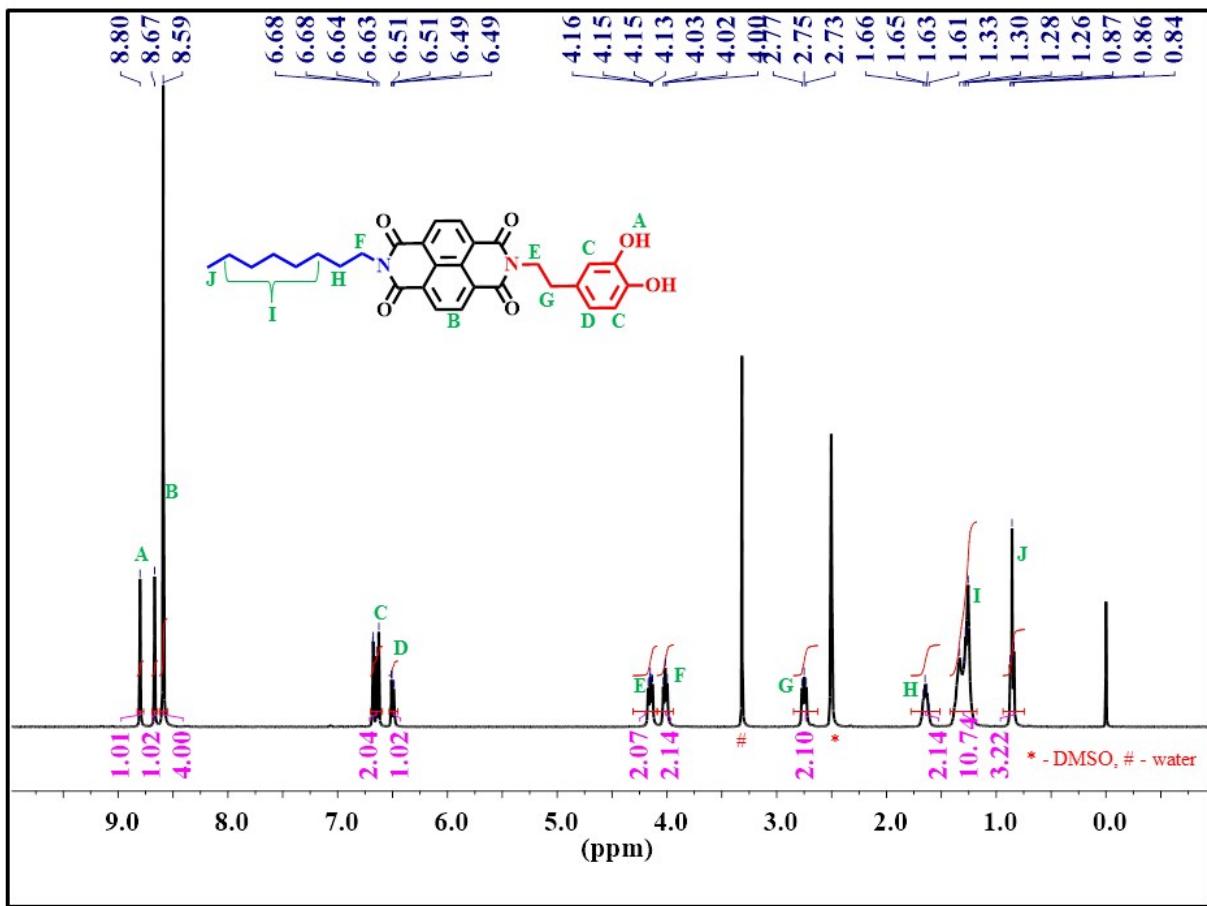
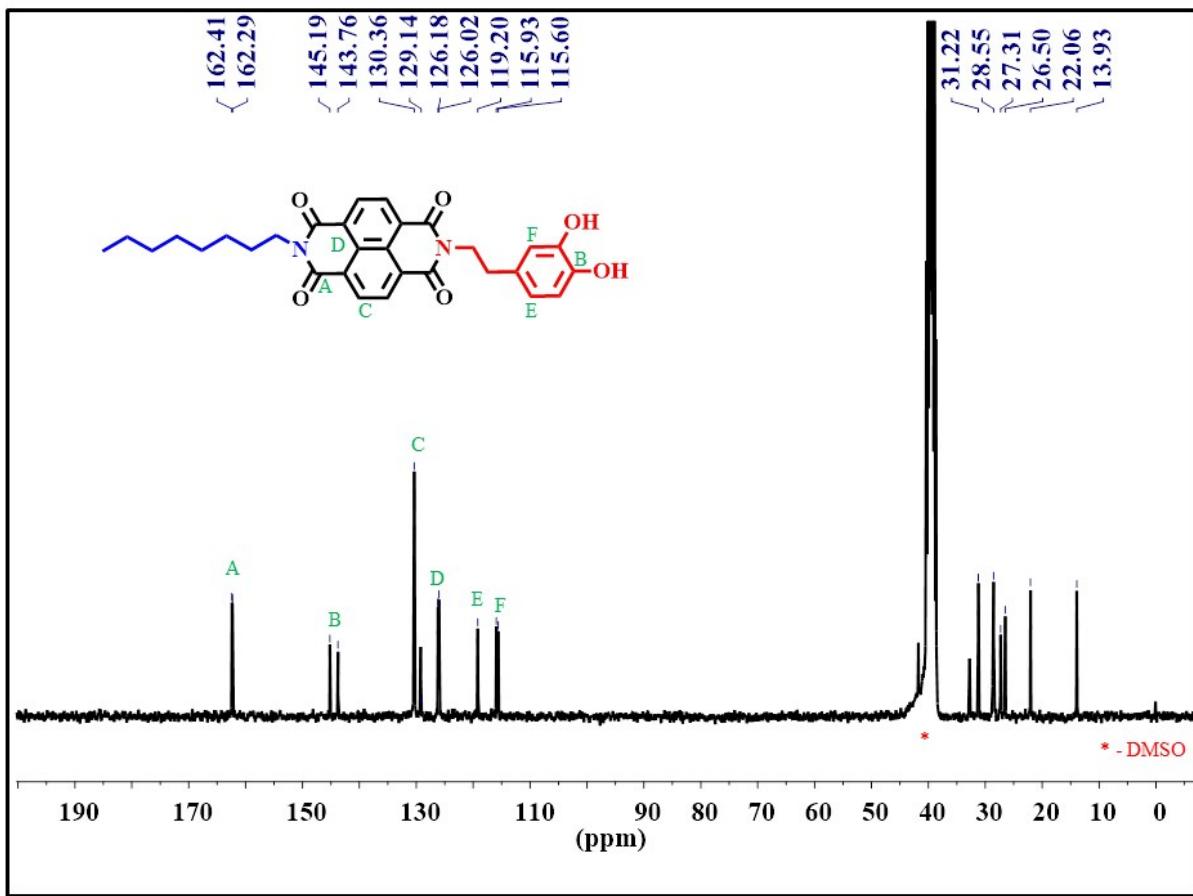


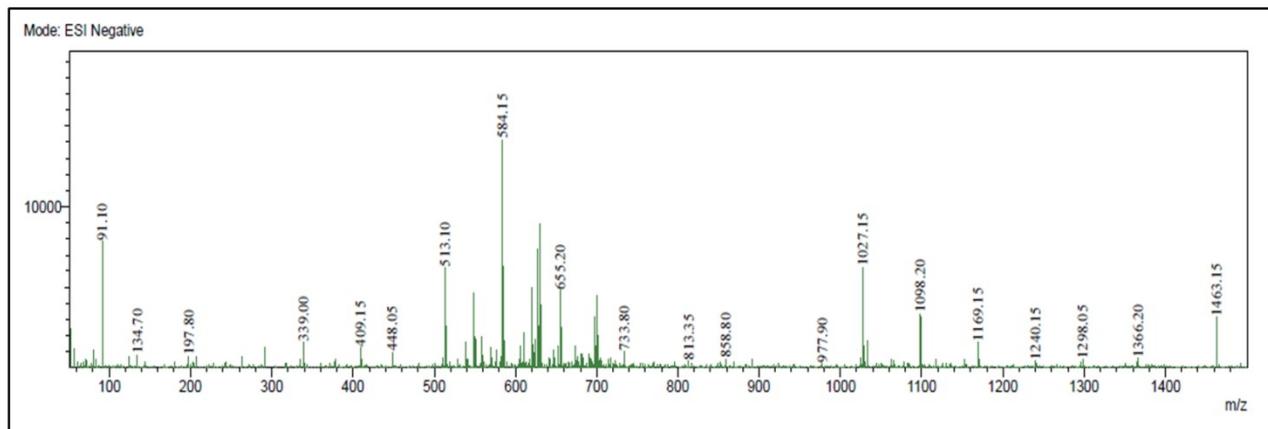
Fig. S6 FT-IR spectra of NDI-1DP.



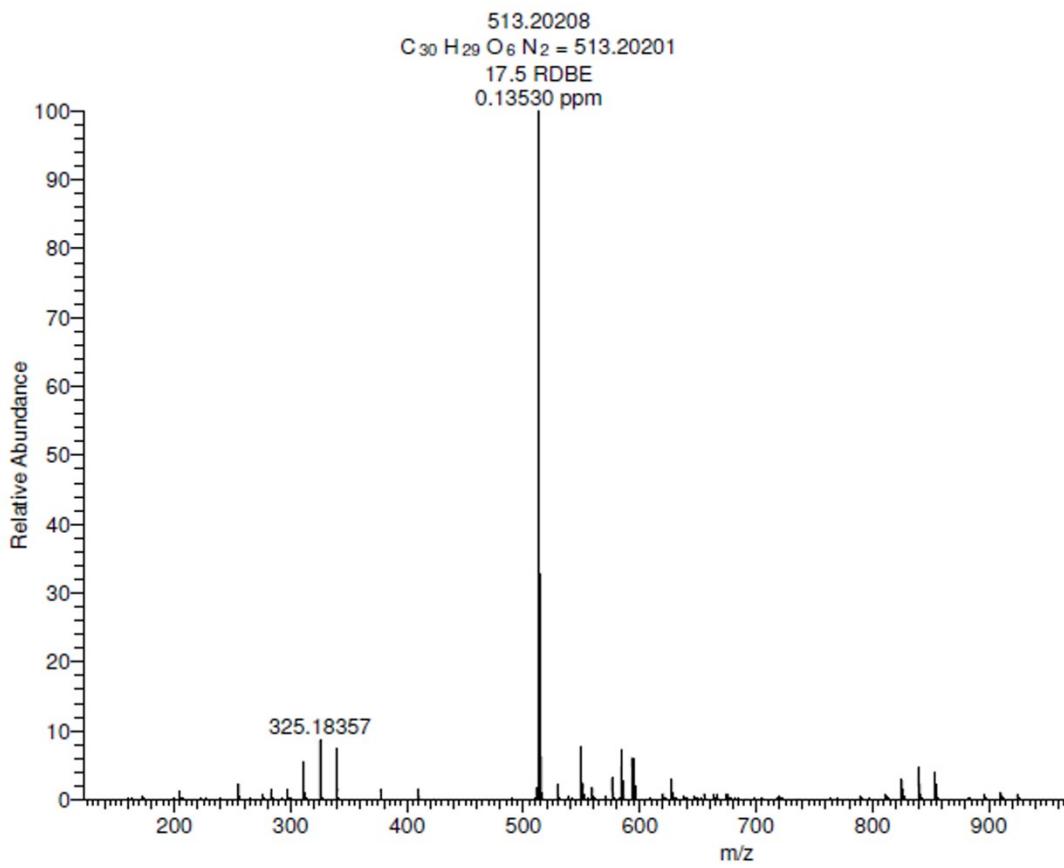
**Fig. S7**  $^1\text{H}$  NMR spectra of NDI-1DP.



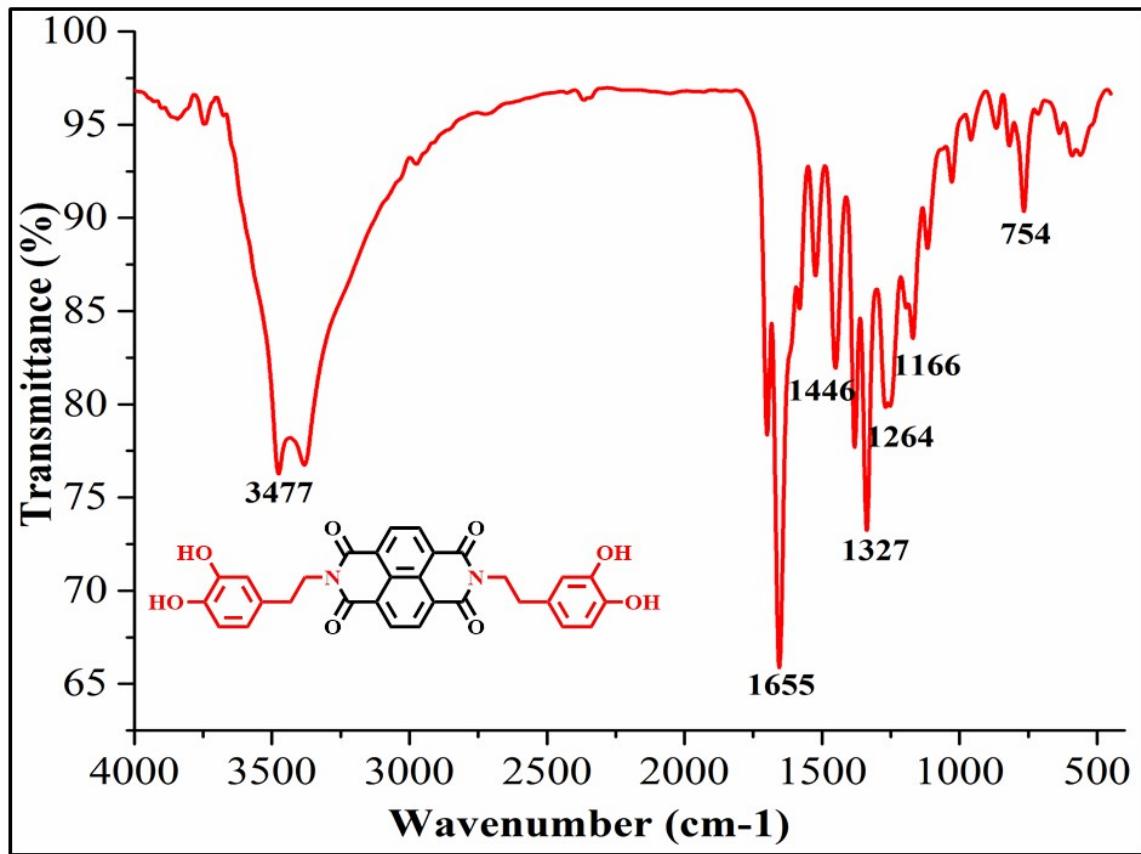
**Fig. S8**  $^{13}\text{C}$  NMR spectra of NDI-1DP.



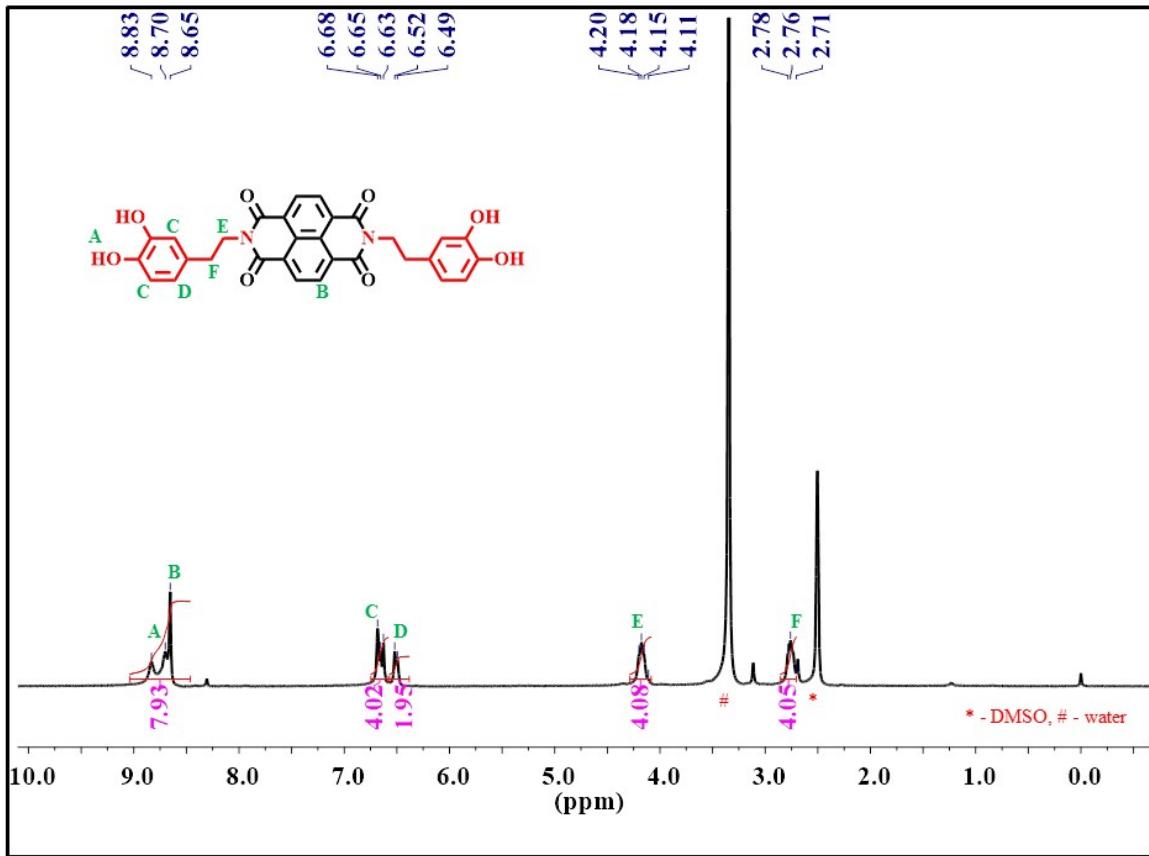
**Fig. S9** LR-MS (negative mode) spectrum of NDI-1DP.



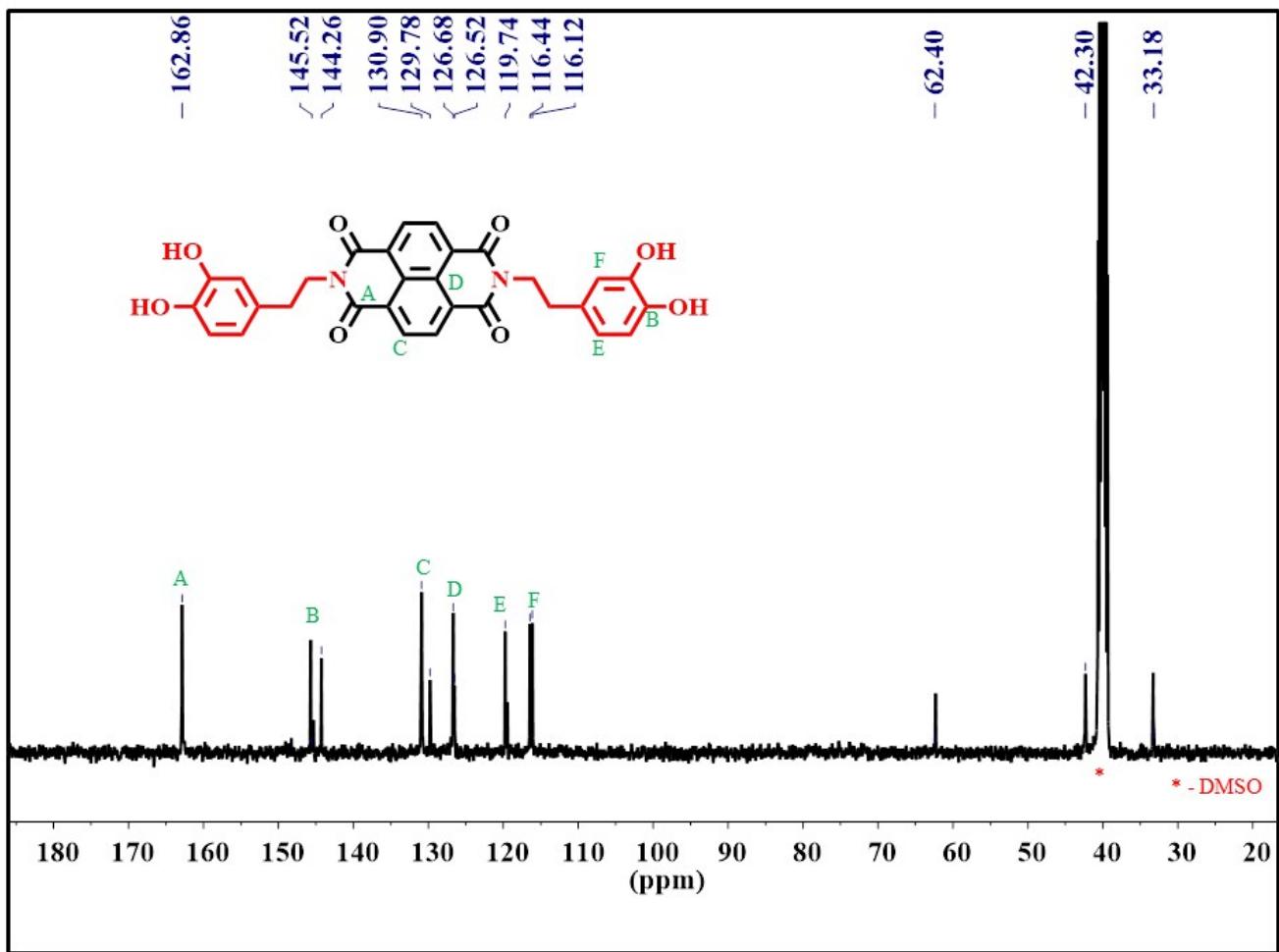
**Fig. S10** HRMS spectrum of NDI-1DP



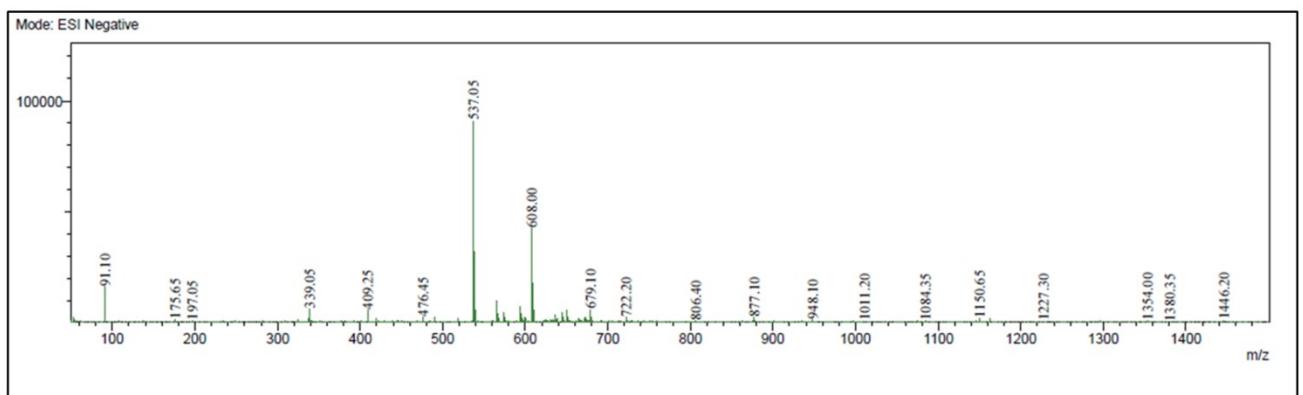
**Fig. S11** FT-IR spectra of NDI-2DP.



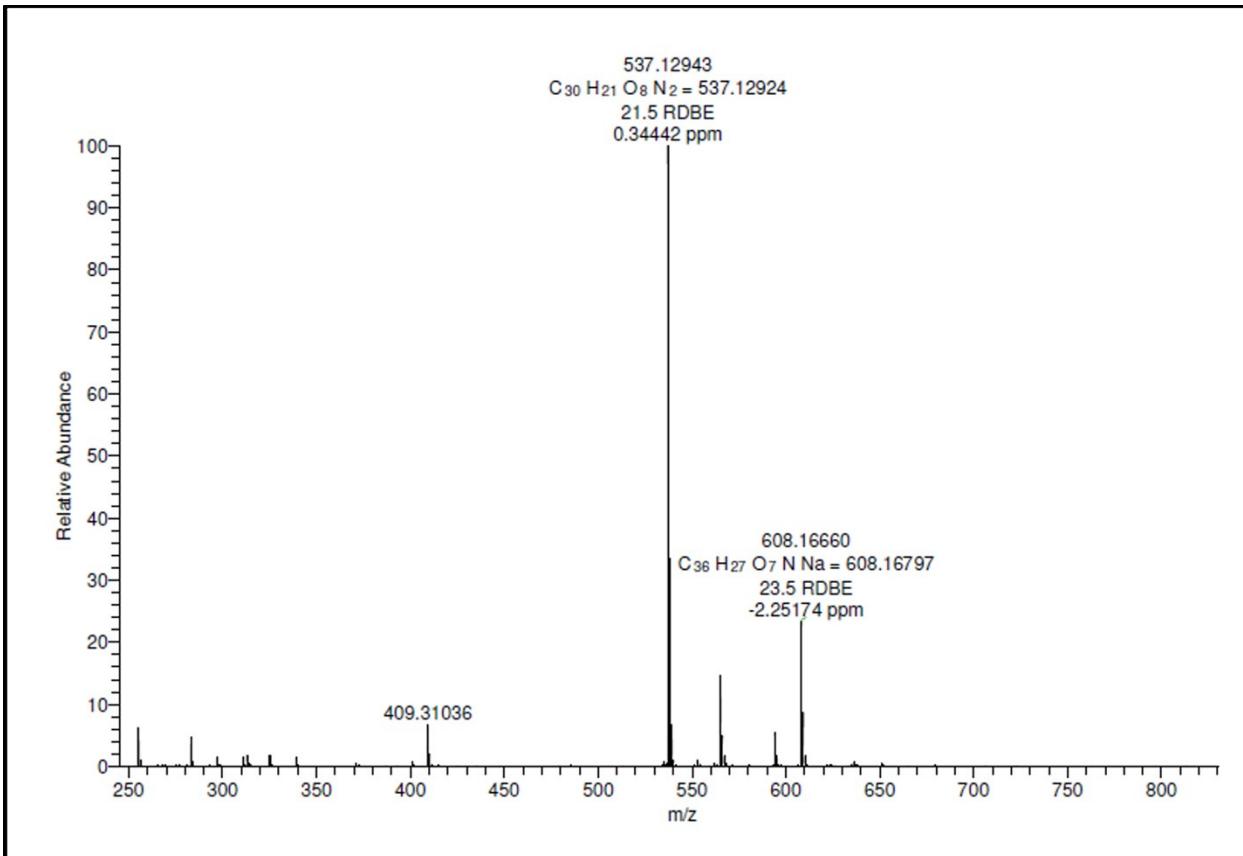
**Fig. S12**  $^1\text{H}$  NMR spectra of NDI-2DP.



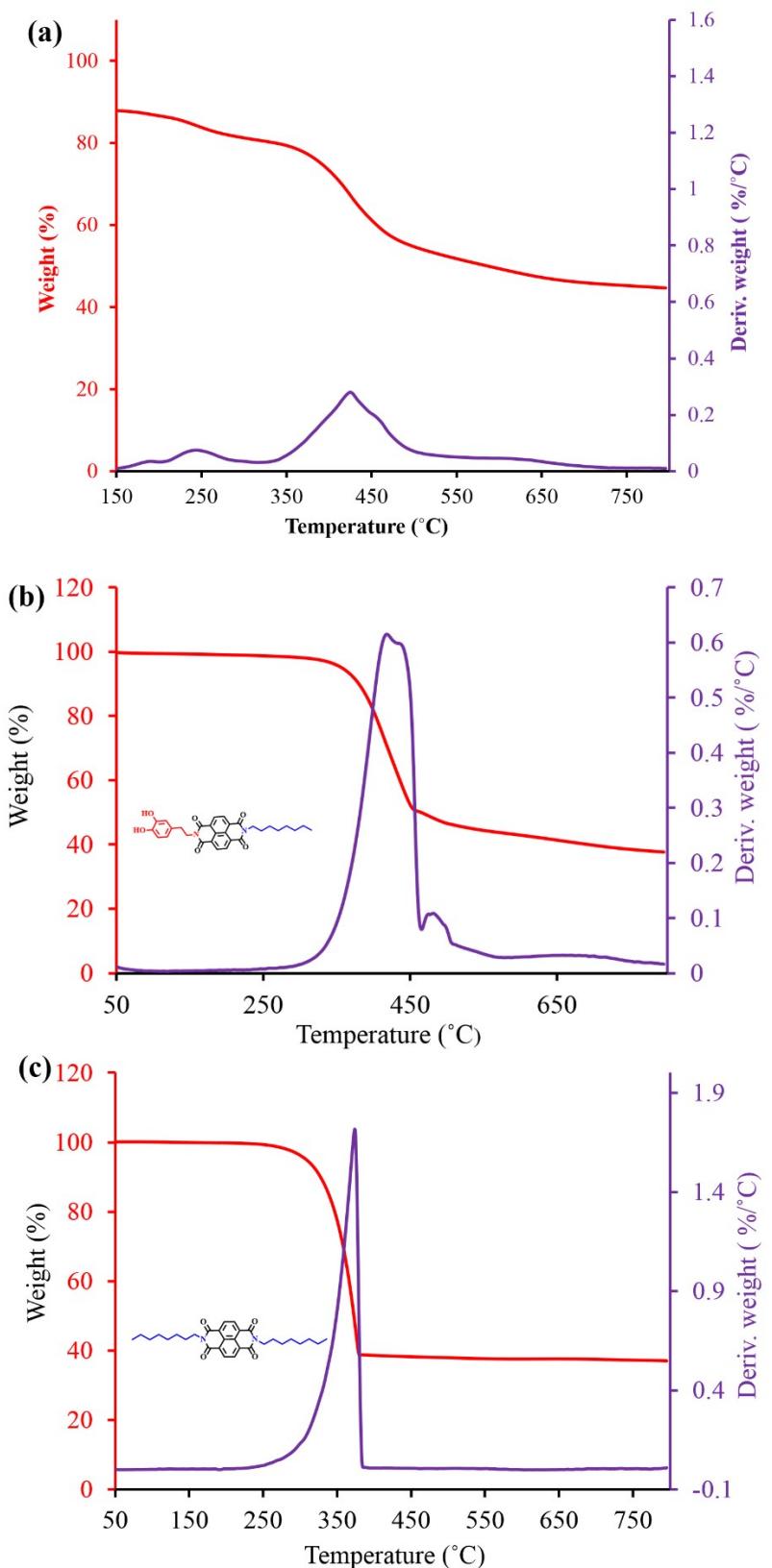
**Fig. S13**  $^{13}\text{C}$  NMR spectra of NDI-2DP.



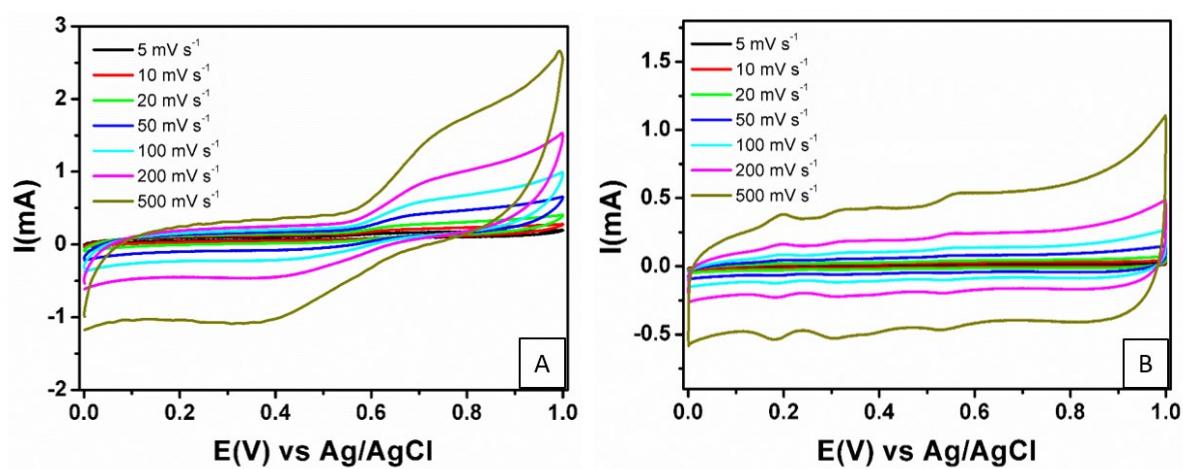
**Fig. S14** LR-MS (negative mode) spectrum of NDI-2DP.



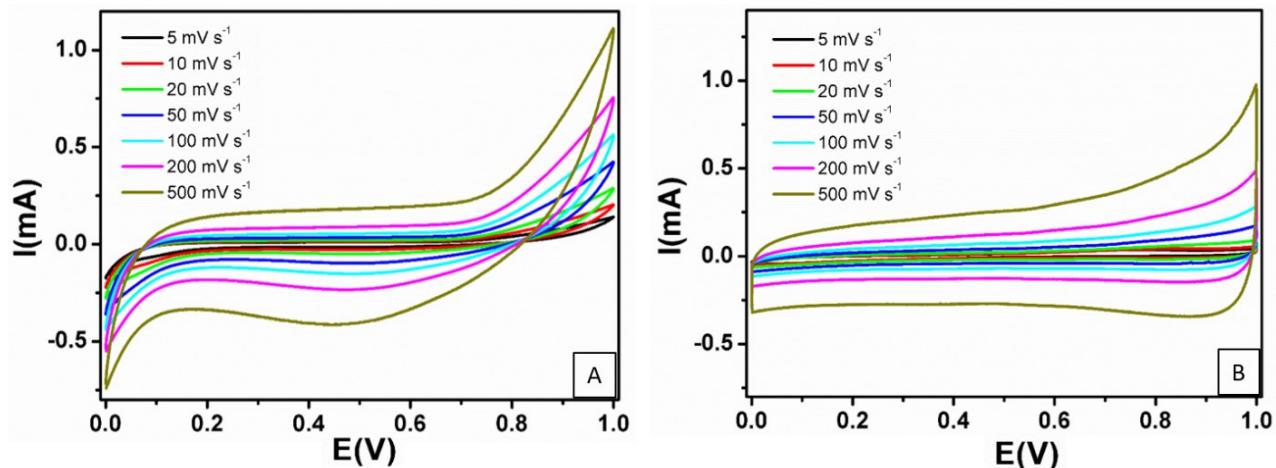
**Fig. S15** HRMS spectrum of NDI-2DP



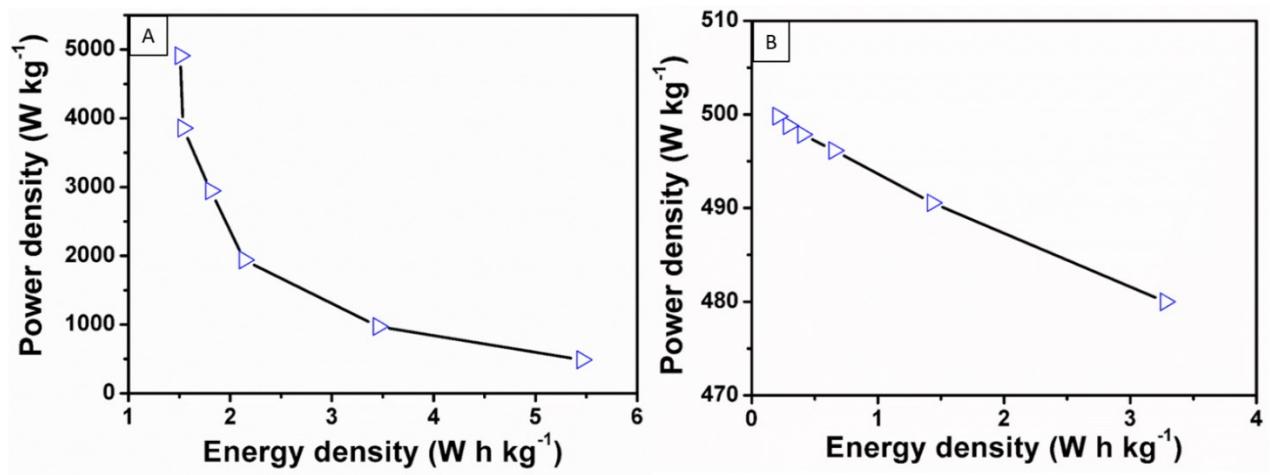
**Fig. S16** Thermogravimetric (TGA) of (a) NDI-2DP, (b) NDI-1DP and (c) NDI.



**Fig. S17 Three-electrode system:** Scan rate dependent cyclic voltammograms of (A) **NDI-1DP** and (B) **NDI**.



**Fig. S18 Two electrode solid-state symmetric capacitor:** Scan rate dependent cyclic voltammograms of (A) **NDI-1DP** and (B) **NDI**.



**Fig. S19** Ragone plot for the (A) **NDI-1DP** and (B) **NDI** symmetric cell device.

**Table S1.** Comparison of the electrochemical performances of Naphthalenediimide derivatives with other literature.

Naphthalene diimide derivatives	Electrolyte	Single electrode-CV			CD			Cycle No. @ Cs (A g <sup>-1</sup> ) retention (%)	Ref No.
		Volta ge (V)	Scan rate (mV s <sup>-1</sup> )	C <sub>s</sub> (F g <sup>-1</sup> )	Voltage (V)	Curr. Density (A g <sup>-1</sup> )	C <sub>s</sub> (F g <sup>-1</sup> )		
TPA-1Th-NDI	1 M TEATFB in 1:1 PC and DMC	-2.0 to 0	10	-----	0 to 2	0.1 mA	22	500 at 0.1 mA >90%	1
(NIBDZ)	1 M H <sub>3</sub> PO <sub>4</sub>	0 to 1	100	-----	0 to 1	0.5	66.56	***	2
P2 P(NDI2OD -OThCNPV)	0.5 M H <sub>2</sub> SO <sub>4</sub>	-0.7 to 0.5	10	-----	-0.7 to 0.5	0.5	124	5000 at 0.5 100%	3
P1 P(NDI2OD -OThPV)	0.5 M H <sub>2</sub> SO <sub>4</sub>	-0.7 to 0.5	10	-----	-0.7 to 0.5	0.5	84	-----	3
P(NDI2OD -T2)	0.5 M H <sub>2</sub> SO <sub>4</sub>	-0.7 to 0.5	10	-----	-0.7 to 0.5	<sup>a</sup> 0.5	61	-----	3
P(NDI-Alt-BDT)	1 M PC-LiClO <sub>4</sub>	-0.2 to 1.1	10	-----	-1.0 to 1	<sup>b</sup> 0.5	80	-----	4
P(NDI-r-BDT)	1 M PC-LiClO <sub>4</sub>	-2.0 to 2.0	5	-----	-1.0 to 1.0	0.5	44	-----	4
NDI-2DP/CP	1 M H <sub>2</sub> SO <sub>4</sub>	0 – 1.0	5	202.5	0 to 1	0.5	195.9	10000 at 3 mA cm <sup>-2</sup> 96 %	This work

Tetraethylammonium tetrafluoroborate (TEATFB),  
Propylene carbonate (PC)  
Dimethylcarbonate (DMC)

## References

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- 2 A. Roy, S. Mondal, A. Halder, A. Banerjee, D. Ghoshal, A. Paul and S. Malik, *Eur. Poly. J.*, 2017, **93**, 448-457.
- 3 S. Sharma, R. Soni, S. Kurugot and S. K. Asha, *Macromolecules*, 2018, **51**, 954–965.
- 4 S. Sharma, R. Soni, S. Kurugot and S. K. Asha, *J. Phys. Chem. C*, 2019, **123**, 2084–2093.