

Electronic Supplementary Information (ESI)

Partial sulfidation for constructing Cu₂O-CuS heterostructures realizing enhanced electrochemical glucose sensing

*Sai Zhang,[‡] Xiaoming Mou,[‡] Zhao Cui, Changmin Hou, Wenlong Yang, * Hongtao Gao, and Xiliang Luo**

S. Zhang, X. M. Mou, Z. Cui, C. M. Hou, Dr. W. L. Yang, Prof. H. T. Gao, Prof. X. L. Luo

Key Laboratory of Optic-electric Sensing and Analytical Chemistry for Life Science, MOE,

Key Laboratory of Analytical Chemistry for Life Science in Universities of Shandong,

College of Chemistry and Molecular Engineering,

Qingdao University of Science and Technology,

Qingdao, Shandong, 266042, P. R. China

E-mail: wlyang@qust.edu.cn; xiliangluo@qust.edu.cn

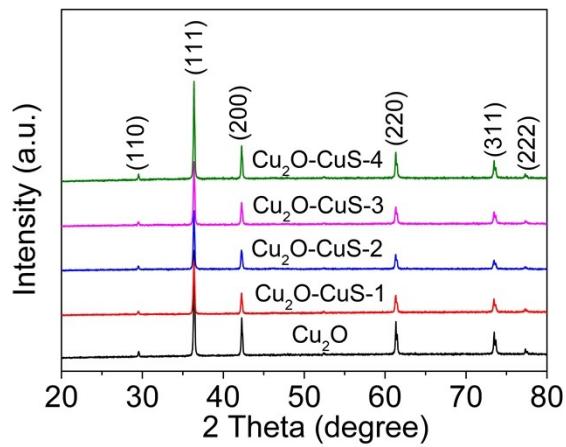


Fig. S1 XRD patterns of Cu₂O cubes and different Cu₂O-CuS hybrid structures.

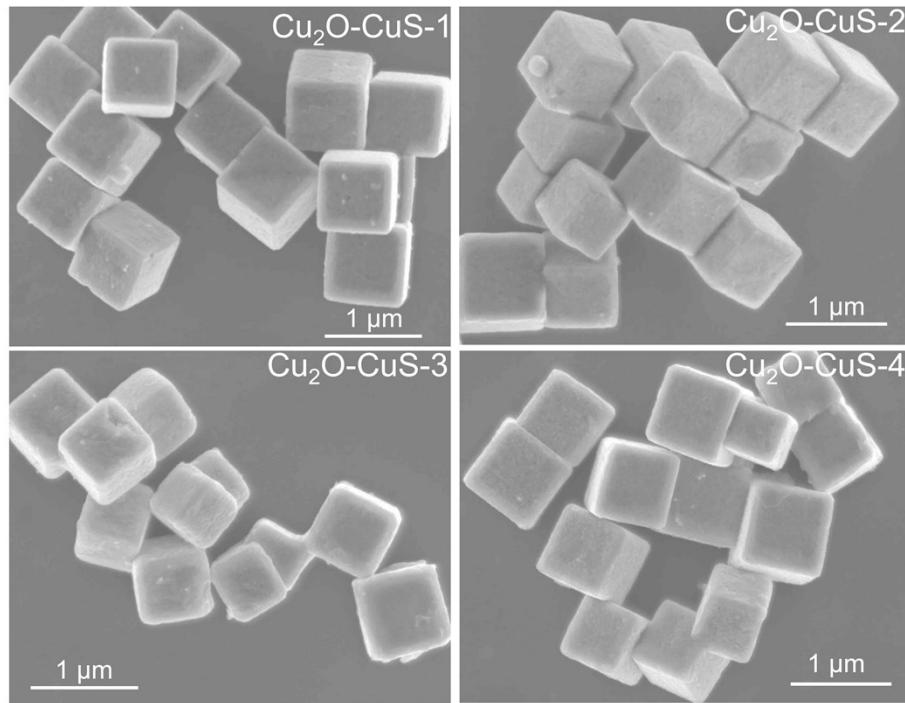


Fig. S2 SEM images of different Cu₂O-CuS hybrid structures.

Cu₂O-CuS-1, Cu₂O-CuS-2, Cu₂O-CuS-3, and Cu₂O-CuS-4 were prepared with 20 mg, 30 mg, 40 mg, and 50 mg thioacetamide (TAA), respectively.

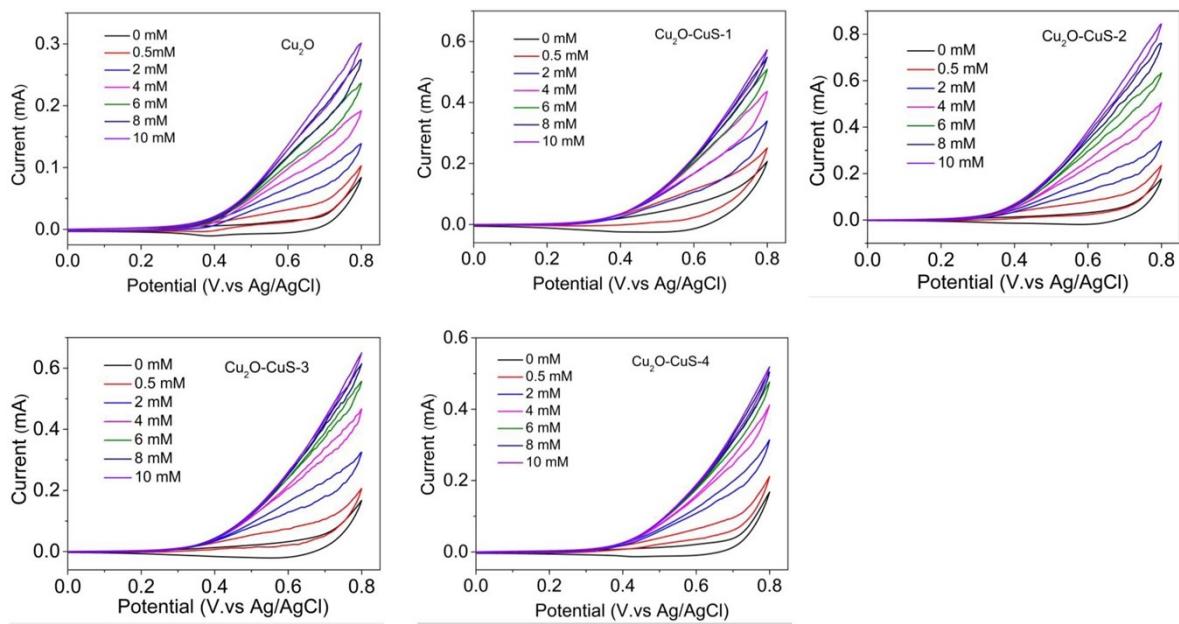


Fig. S3 CV curves of pure Cu_2O cubes and various $\text{Cu}_2\text{O}-\text{CuS}$ hybrid structures measured in 0.1 M NaOH solution with different concentrations of glucose at a scan rate of 50 mV s^{-1} .

Table S1 Comparison of the electrochemical performance for glucose sensing between the $\text{Cu}_2\text{O}-\text{CuS}$ -2 hybrid electrode and other transition-metal-based electrodes recently reported.

Electrode	Linear range (mM)	Detection limit (μM)	Sensitivity ($\mu\text{A}\cdot\text{mM}^{-1}\cdot\text{cm}^{-2}$)	Reference
$\text{Cu}_2\text{O}-\text{CuS-2}$	0.01–1	0.46	1876	This work
$\text{Cu}_2\text{O}/\text{Cu}$	0.05–6.75	37	62.29	Int. J. Electrochem. Sci. 2012, 7, 12587–12600.
Cu_2O MSs/S-MWCNTs	0.005–7	1.46	581.89	ACS Appl. Nano Mater. 2020, 3, 4788–4798.
$\text{Cu}_2\text{O}/$ graphene	0.3–3.3	3.3	285	Biosens. Bioelectron. 2013, 45, 206–212.
spiky Cu/CuO NW array	0.01–7	10	1210	ACS Omega 2019, 4, 12222–12229.
rGO-Pt NW	0.032–1.89	4.6	56.11	Int. J. Electrochem. Sci. 2018, 13, 4817–4826.
Co_3N NW/TM	0.0001–2.5	0.05	3325.6	Sens. Actuators, B 2018, 255, 1254–1261.
$\text{W}_{18}\text{O}_{49}$	0.06–1.6	0.02	167	ACS Biomater. Sci. Eng. 2020, 6, 1909–1919.
NiO/TiO_2	0.005–12.1	1.0	252.0	Sci. Total Environ. 2015, 502, 70–79.

Table S2 DFT data for the formation energies of electroactive Cu(III) species.

Species	Energy (eV)	Formation energy (eV)
Cu ₂ O	-61431.0703	5.824285
Cu ₂ O-OH	-61456.2941	
CuS-Cu ₂ O	-77222.8434	3.066185
CuS-Cu ₂ O-OH	-77250.8253	
OH*	-31.048085	--

Table S3 DFT data for the adsorption energies of glucose on the surface of catalysts.

Species	Energy (eV)	Adsorption energy (eV)
Cu ₂ O	-581493.4961	-4.6189
Cu ₂ O-C ₆ H ₁₂ O ₆	-601175.6299	
CuS-Cu ₂ O	-582448.3601	-7.9101
CuS-Cu ₂ O-C ₆ H ₁₂ O ₆	-602133.7851	
C ₆ H ₁₂ O ₆	-19677.5149	--