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## supplementary

## 2D g- $C_3N_4$ as Bifunctional Photocatalyst for Co-catalyst and Sacrificial Agent-Free Photocatalytic $N_2$ Fixation and Dye

## Photodegradation

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Catalysts	Cocatalyst	Light source	Sacrificial	The rate of the NH <sub>3</sub> production	Ref
g-C <sub>3</sub> N <sub>4</sub>	Sulfur- doped	Xenon lamp (500-W)	Methanol	5.99 mM h <sup>-1</sup> gcat <sup>-1</sup>	[S1]
g-C <sub>3</sub> N <sub>4</sub>	HCl	Sodium lamp (250 W, 400–800 nm)	Ethanol	6.32 mg L <sup>-1</sup> h <sup>-1</sup> gcat <sup>-1</sup>	[S2]
g-C <sub>3</sub> N <sub>4</sub>	rGO	Sodium lamp (250 W, 400–800 nm)	EDTA- 2Na	9.276 mg L <sup>-1</sup> h <sup>-1</sup> gcat <sup>-1</sup>	[83]
g-C <sub>3</sub> N <sub>4</sub>	КОН	Xe lamp (300 W)	Methanol	3.632 mmol h <sup>-1</sup> g <sup>-1</sup>	[S4]
g-C <sub>3</sub> N <sub>4</sub> / Fe <sub>2</sub> O <sub>3</sub>	Non	Xenon lamp (300 W)	Ethanol	47.9 mg/L/h	[85]
g-C <sub>3</sub> N <sub>4</sub> / ZnMo/CdS	Non	Sodium lamp (250 W, 400–800 nm)	Ethanol	1.47 mg L <sup>-1</sup> h <sup>-1</sup> gcat <sup>-1</sup>	[S6]
V-g-C <sub>3</sub> N <sub>4</sub> / Ag <sub>2</sub> CO <sub>3</sub>	Non	Sodium lamp (250 W, 400–800 nm) with NaNO <sub>2</sub> solution filter	Ethanol or Methanol	4.5 mg·L <sup>-1</sup> h <sup>-1</sup> gcat <sup>-1</sup>	[S7]
g-C <sub>3</sub> N <sub>4</sub>	Non	Sodium lamp(200W,400- 800nm)	Ethanol	3.01 mg L <sup>-1</sup> h <sup>-1</sup> gcat <sup>-1</sup>	[S8]
V-g-C <sub>3</sub> N <sub>4</sub>	Non	Xe lamp with UV cutoff filter (300 W, $\lambda > 420$ nm)	Methanol	1.24 mmol/h per 1g of V- g-C <sub>3</sub> N <sub>4</sub>	[S9]
r-GO @PMo <sub>10</sub> V <sub>2</sub>	Non	Xenon lamp (300 W)	Non	130.3 μmol L <sup>-1</sup> h <sup>-1</sup>	[S10]
2D g-C <sub>3</sub> N <sub>4</sub>	Non	Sunlight	Non	3.9 mg/L/h	Current work

Table S1: Comparison of  $NH_3$  production rate on 2D g-C<sub>3</sub>N<sub>4</sub> with reported relevant catalysts.

\* Nitrogen vacancies (V)



Fig. S1. Second order kinetic rate of bulk  $g-C_3N_4$  and 2D  $g-C_3N_4$  of MB.

Table S2. Kinetic k constant and  $R^2$  for first order reaction for MB and MG, and second order reaction for MB.

Sample	First-order kinetic rate (k1) constant (min <sup>-1</sup> )	R <sup>2</sup>	Second order kinetic rate (k2) constant L/mg/min	R <sup>2</sup>
MB				
$2D g-C_3N_4$	0.43463	0.88833	0.11179	0.68702
bulk g-C <sub>3</sub> N <sub>4</sub>	0.31967	0.93027	0.0623	0.81244
MG				
$2D g-C_3N_4$	0.5133	0.8808		
bulk g-C <sub>3</sub> N <sub>4</sub>	0.38103	0.92414		

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