

Supporting Information

Manganese(II) Schiff-base-mediated reversible deactivation controlled radical polymerization of vinyl acetate

Patricia K. Hashimoto,^a Larissa F. Oliveira,^a Beatriz A. Riga-Rocha,^a Antonio E. H.
Machado,^b Vinicius T. Santana,^c Otaciro R. Nascimento,^d Valdemiro P. Carvalho-Jr^a and
Beatriz E. Goi^{a*}

^a*Faculdade de Ciências e Tecnologia, UNESP – Univ. Estadual Paulista, CEP 19060-900,
Presidente Prudente, SP, Brazil*

^b*Instituto de Química, Universidade Federal de Uberlândia, P.O. Box 593, Uberlândia
38400-089, Minas Gerais, Brazil*

^c*CEITEC – Central European Institute of Technology, Brno University of Technology,
Purkyňova 123, 61200 Brno, Czech Republic*

^d*Instituto de Física de São Carlos, USP Universidade de São Paulo, CEP 13563-120, São
Carlos, SP, Brazil.*

*Correspondence to: B. E. Goi (E-mail: beatriz.goi@unesp.br)

Experimental

General Procedure for the Preparation of Schiff-Base Ligands (1a = cyclopentyl, 1b = cyclohexyl and 1c = cycloheptyl)

To prepare the Schiff-base ligands **1a-1c**, a solution of salicylaldehyde (4 mmol; 0.49 g) in methanol (20 mL) was slowly added over a solution of the appropriate amine (4 mmol; 0.34, 0.39 and 0.45 g for cyclopentylamine, cyclohexylamine and cycloheptylamine, respectively) in the same solvent (20 mL). The mixture was stirred at room temperature for 16 h and the product was obtained as a yellowish orange oil.

Ligand (*E*)-2-((cyclopentylimino)methyl)phenol: Yield: 88%; Refractive index 1.5626; (a) UV-Vis: $\lambda_{\max(n)}$ (nm), $\epsilon_{\max(n)}$ [$M^{-1} \text{ cm}^{-1}$]: $\lambda_{\max(1)}$ (317), $\epsilon_{\max(1)}$ [5320]; (b) IR (KBr, cm^{-1}): $\nu_{\text{C=N}}$ (1626), $\nu_{\text{C-O}}$ (1277); (c) ^1H NMR: (CDCl_3 , 400 MHz): 13.68 (s, 1H, OH), 8.34 (s, 1H, CH=N), 7.28 (ddd, $^3J_{\text{b,a}} = ^3J_{\text{b,c}} = 7.5$ Hz, $^4J_{\text{b,d}} = 1.9$ Hz, 1H, salicyl-ring), 7.23 (dd, $^3J_{\text{d,c}} = 7.5$ Hz, $^4J_{\text{d,b}} = 1.9$ Hz, 1H, salicyl-ring), 6.94 (dd, $^3J_{\text{a,b}} = 7.5$ Hz, $^4J_{\text{a,c}} = 1.1$ Hz, 1H, salicyl-ring), 6.86 (ddd, $^3J_{\text{c,d}} = ^3J_{\text{c,b}} = 7.5$ Hz, $^4J_{\text{c,a}} = 1.1$ Hz, 1H, salicyl-ring), 3.75-3.82 (m, 1H, $\text{CH}^{\text{Pentyl}}$), 1.95-1.81 (m, 4H, $\text{CH}_2^{\text{Pentyl}}$), 1.65-1.76 (m, 4H, $\text{CH}_2^{\text{Pentyl}}$). ^{13}C NMR (CDCl_3) δ 162.3, 161.3, 131.8, 130.9, 118.4, 116.9, 70, 34.7, 24.5.

Ligand (*E*)-2-((cyclohexylimino)methyl)phenol: Yield: 94%; Refractive index 1.5678; (a) UV-Vis: $\lambda_{\max(n)}$ (nm), $\epsilon_{\max(n)}$ [$M^{-1} \text{ cm}^{-1}$]: $\lambda_{\max(1)}$ (316,8), $\epsilon_{\max(1)}$ [7770]; (b) IR (KBr, cm^{-1}): $\nu_{\text{C=N}}$ (1629), $\nu_{\text{C-O}}$ (1274), (c) ^1H NMR (CDCl_3 , 400 MHz): 13.82 (s, 1H, OH), 8.36 (s, 1H, CH=N), 7.28 (ddd, $^3J_{\text{b,c}} = 7.5$ Hz, $^3J_{\text{b,a}} = 7.2$ Hz, $^4J_{\text{b,d}} = 1.9$ Hz, 1H, salicyl-ring), 7.23 (dd, $^3J_{\text{d,c}} = 7.5$ Hz, $^4J_{\text{d,b}} = 1.9$ Hz, 1H, salicyl-ring), 6.94 (dd, $^3J_{\text{a,b}} = 7.2$ Hz, $^4J_{\text{a,c}} = 1.1$ Hz, 1H, salicyl-ring), 6.86 (ddd, $^3J_{\text{c,b}} = ^3J_{\text{c,d}} = 7.5$ Hz, $^4J_{\text{c,a}} = 1.1$ Hz, 1H, salicyl-ring), 3.20-3.27 (m, 1H, CH^{Hexyl}), 1.78-1.86 (m, 4H, $\text{CH}_2^{\text{Hexyl}}$), 1.50-1.68 (m, 3H, $\text{CH}_2^{\text{Hexyl}}$), 1.43-1.25 (m, 3H, $\text{CH}_2^{\text{Hexyl}}$). ^{13}C NMR (CDCl_3) δ 162.1, 161.4, 131.9, 131.1, 118.9, 118.3, 117.04, 77.3, 77.02, 76.7, 67.4, 25.5, 24.3.

Ligand (*E*)-2-((cycloheptylimino)methyl)phenol: Yield: 91%; Refractive index 1.5652; (a) UV-Vis: $\lambda_{\max(n)}$ (nm), $\epsilon_{\max(n)}$ [$M^{-1} \text{ cm}^{-1}$]: $\lambda_{\max(1)}$ (316.2), $\epsilon_{\max(1)}$ [10000]; (b) IR (KBr, cm^{-1}): $\nu_{\text{C=N}}$ (1630), $\nu_{\text{C-O}}$ (1270); (c) ^1H NMR (CDCl_3 , 400 MHz): 13.82 (s, 1H, OH), 8.36 (s, 1H, CH=N), 7.28 (ddd, $^3J_{\text{b,c}} = 7.5$ Hz, $^3J_{\text{b,a}} = 7.2$ Hz, $^4J_{\text{b,d}} = 1.9$ Hz, 1H,

salicyl-ring), 7.23 (dd, $^3J_{d,c} = 7.5$ Hz, $^4J_{d,b} = 1.9$ Hz, 1H, salicyl-ring), 6.94 (dd, $^3J_{a,b} = 7.5$ Hz, $^4J_{a,c} = 1.1$ Hz, 1H, salicyl-ring), 6.86 (ddd, $^3J_{c,b} = ^3J_{c,d} = 7.5$ Hz, $^4J_{c,a} = 1.1$ Hz, 1H, salicyl-ring), 3.37-3.46 (m, 1H, CH^{Heptyl}), 1.74-1.85 (m, 4H, CH₂^{Heptyl}), 1.50-1.68 (m, 8H, CH₂^{Heptyl}). ¹³C NMR (CDCl₃) δ 161.66, 161.39, 131.92, 131.02, 118.91, 118.29, 116.99, 77.34, 77.08, 76.83, 70.12, 36.40, 28.54, 24.24.

Characterization

The bidentate Schiff-bases were prepared from condensation of salicylaldehyde with three different cycloalkylamines in methanol. The ¹H NMR spectrum showed two important observations: the completion of the condensation observed by the appearance of a singlet peak at 8.3 ppm belonging to the imine proton (HC=N) and the complete loss of the peak around 5 ppm corresponding to the amine protons. The other peaks observed in the ¹H NMR spectrum are as follows: the singlet in the down-filled, at 13.8 ppm, observed in the spectra of **1a-1c**, is assignable to the hydrogen of the OH. Also, for all ligands, the peak group assigned to the CH from salicylaldehyde is in the range of δ 6.8–7.3 ppm. In the higher field, the peaks relative to parts of the ligands of the *N*-cycloalkyl group are observed from 3.2 to 3.8 ppm for CH-N and from 1.4 to 1.9 ppm for CH₂.

Figure S1 Absorption spectra of the manganese(II) complexes in degassed CH₂Cl₂ solution at room temperature ([complex] = 0.2 mM).

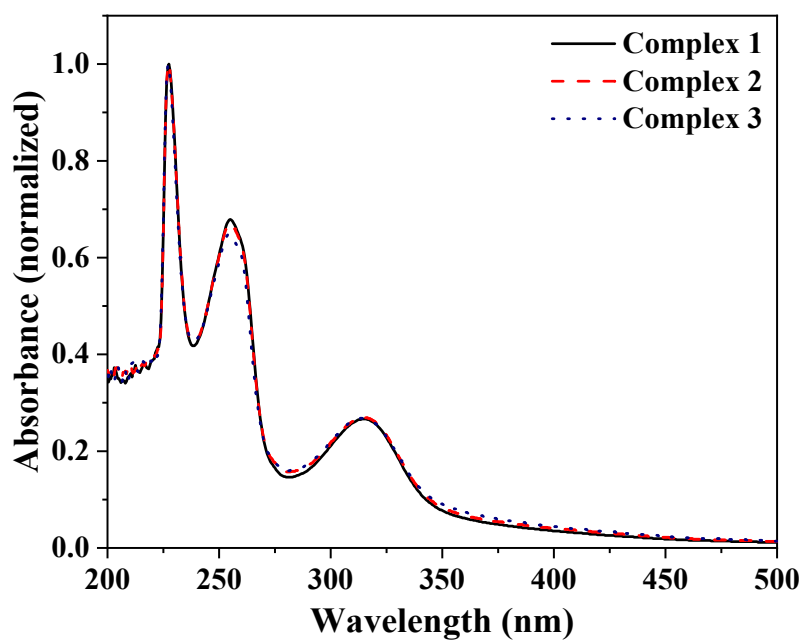


Table S1 Cyclic voltammetry results for complexes 1-3

Complex	Scan rate (mV s ⁻¹)	E_{pa} (V)	E_{pc} (V)	$\Delta E_p^{[a]}$ (V)	I_{pa}/I_{pc}
1	100	-0.326	-0.741	0.415	1.481
	200	-0.200	-0.755	0.555	1.818
	300	-0.108	-0.762	0.654	1.843
2	100	-0.226	-0.762	0.536	1.430
	200	-0.123	-0.805	0.682	1.861
	300	-0.083	-0.868	0.785	1.957

^[a] $\Delta E_p = E_{pa} - E_{pc}$.

Figure S2 GPC traces as a function of time for OMRP of VAc mediated by complexes **1-3**; $[VAc]/[AIBN]/[Mn] = 542/3.25/1$ with complex in bulk at 55 °C.

