

## Electronic supplementary information

### Syntheses, structures and magnetic properties in two isostructural dicyanamide-bridged 2D polymers †

Andrii Kliuikov<sup>a</sup>, Oleksandr Bukrynov<sup>b</sup>, Erik Čižmár<sup>a</sup>, Lucia Váhovská<sup>c</sup>, Svitlana Vitushkina<sup>b</sup>, Erika Samoľová<sup>d</sup>, Ivan Potočňák<sup>e\*</sup>

<sup>a</sup> *P. J. Šafárik University in Košice, Faculty of Science, Institute of Physics, Park Angelinum 9, SK-041 54 Košice, Slovakia*

<sup>b</sup> *V.N. Karazin Kharkiv National University, Faculty of Chemistry, Department of Applied Chemistry, Svobody sq. 4, UA-61022 Kharkiv, Ukraine*

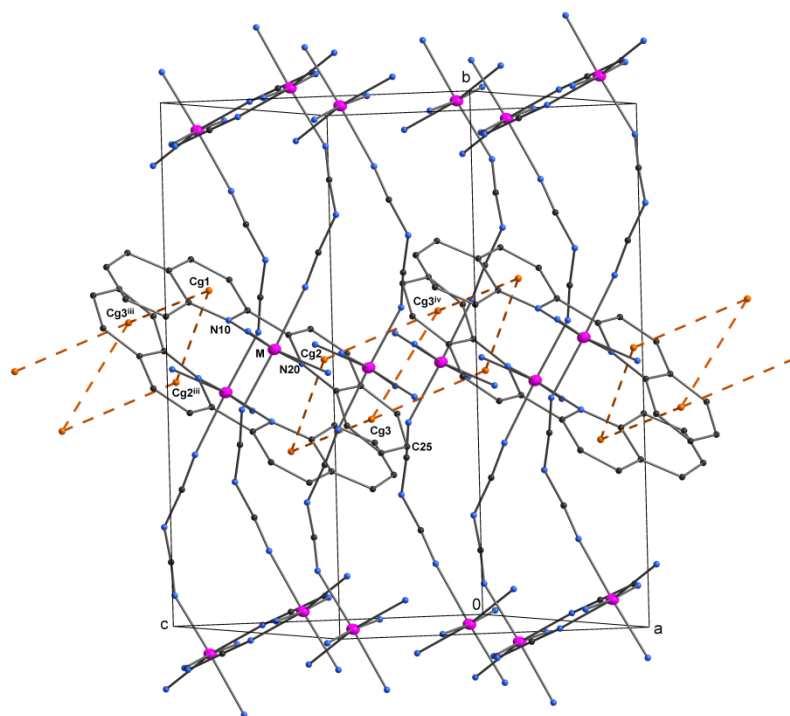
<sup>c</sup> *University of Veterinary Medicine and Pharmacy in Košice, Department of Chemistry, Biochemistry and Biophysics, Komenského 73, SK-041 84 Košice, Slovakia*

<sup>d</sup> *Institute of Physics of the Czech Academy of Sciences, Na Slovance 2, 182 21 Prague 8, Czech Republic*

<sup>e</sup> *P. J. Šafárik University in Košice, Faculty of Science, Institute of Chemistry, Department of Inorganic Chemistry, Moyzesova 11, SK-041 54 Košice, Slovakia*

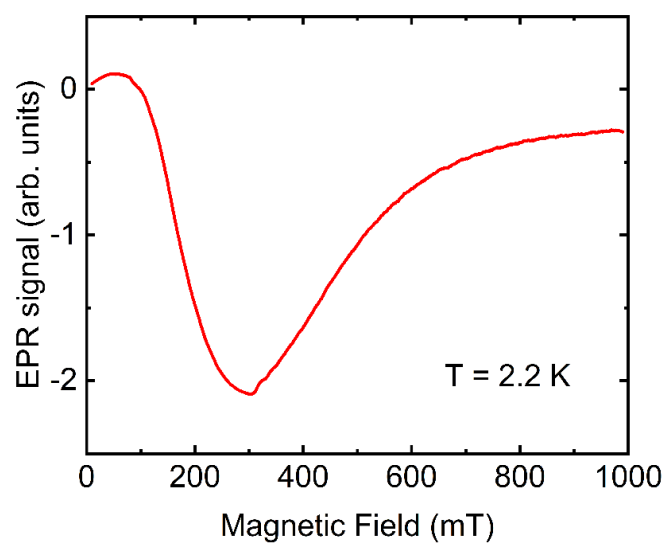
---

† Low-dimensional compounds containing cyanido groups. Part XXXVI.

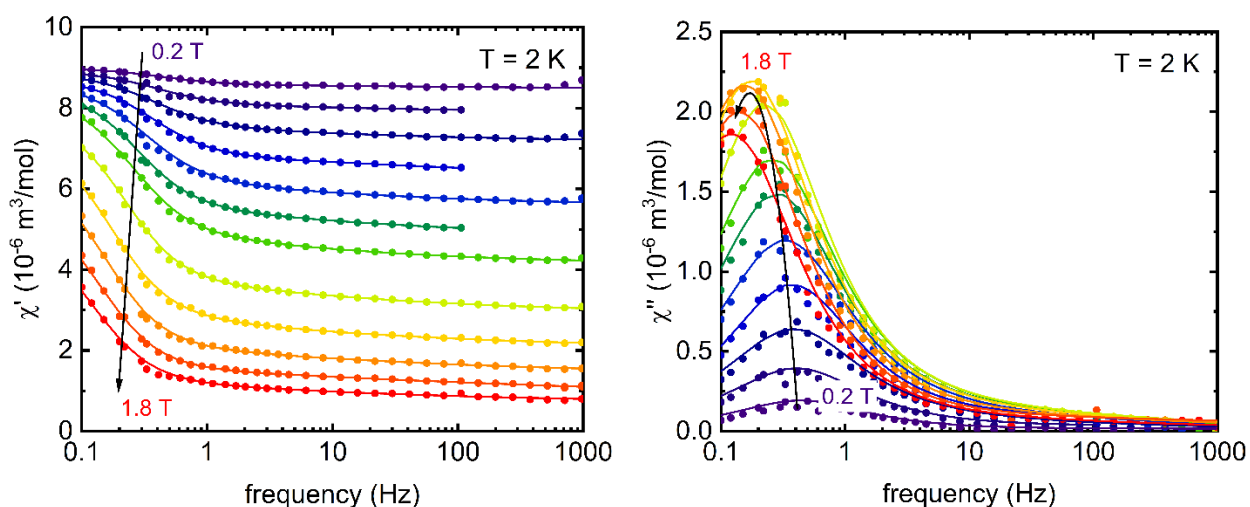


**Fig S1**  $\pi$ - $\pi$  interactions (orange dashed lines) between aromatic rings in **1** and **2**. Hydrogen atoms not involved in hydrogen bonds are omitted for clarity.

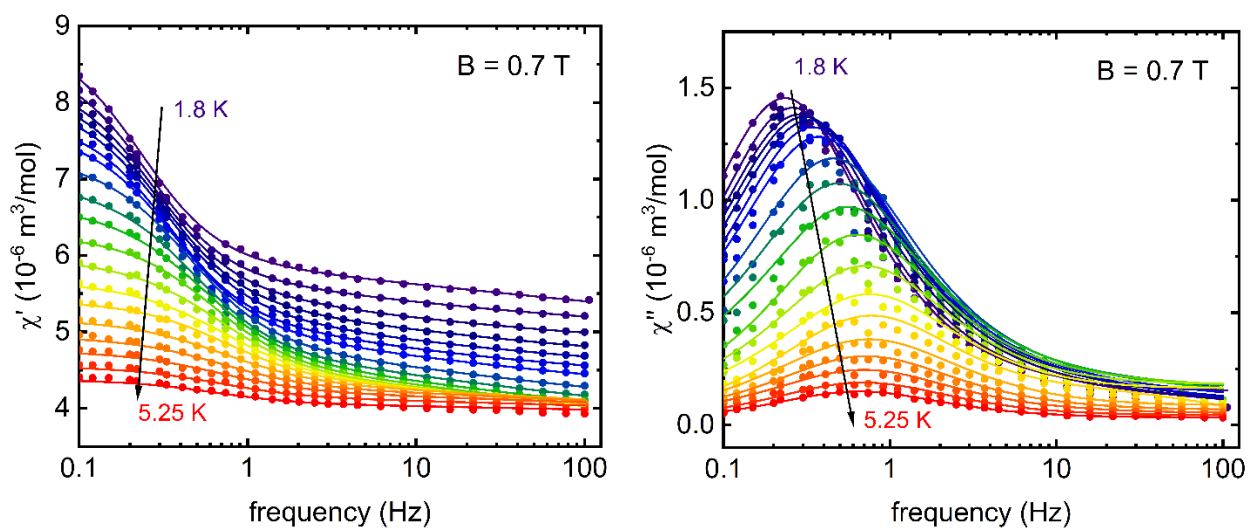
Symmetry codes: iii =  $-x, 1 - y, 1 - z$ ; iv =  $1 - x, 1 - y, 1 - z$ .



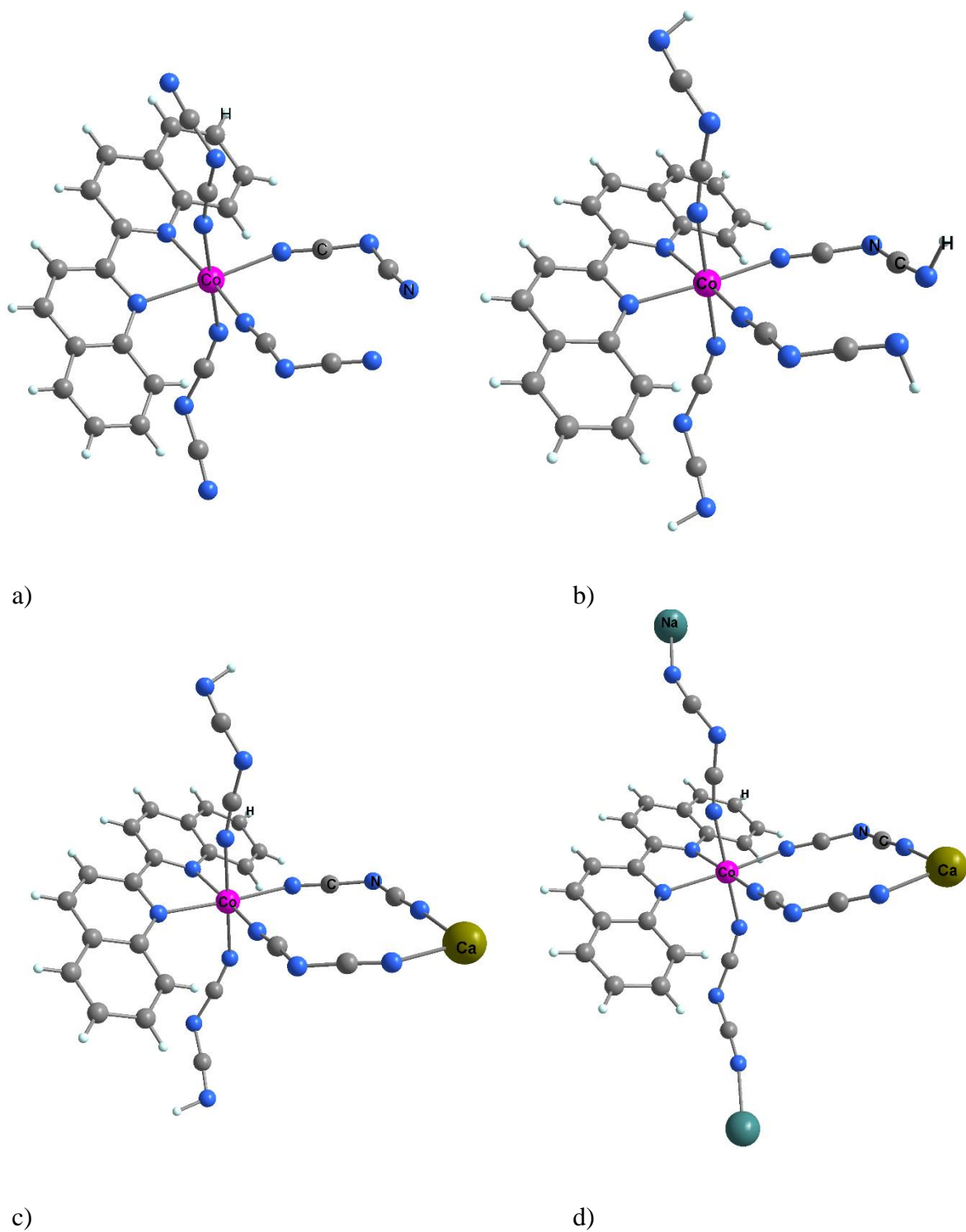
**Fig. S2** X-band EPR spectrum of **1** measured at 2.2 K.



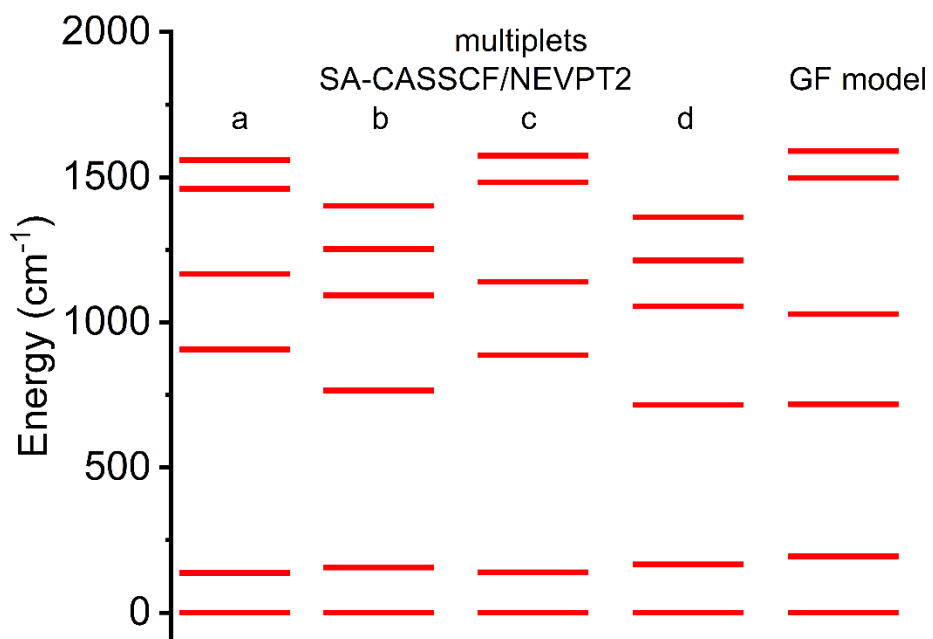
**Fig. S3** Frequency dependence of real (left panel) and imaginary (right panel) component of AC susceptibility of **1** measured at different applied DC fields measured at 2 K including the fit of the modified Debye model (an additional Debye function was included to take into account a very small contribution at higher frequencies).



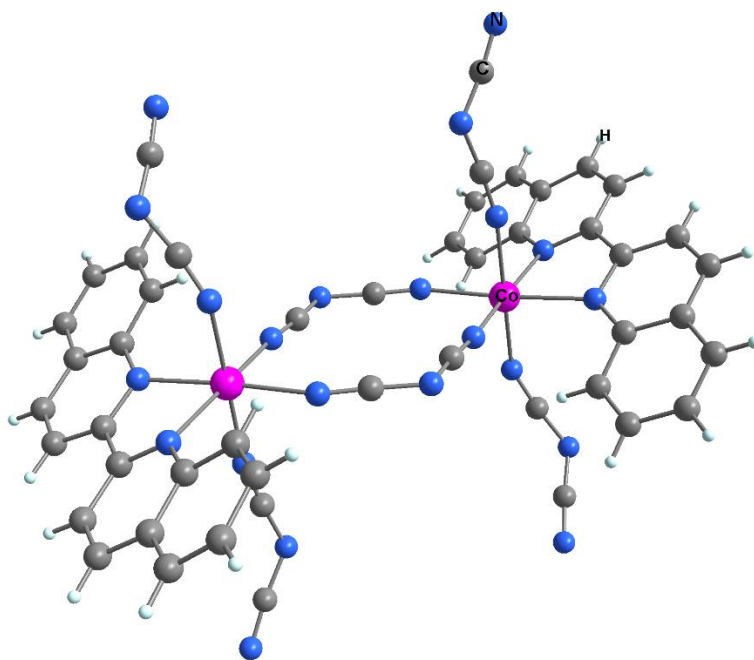
**Fig. S4** Frequency dependence of real (left panel) and imaginary (right panel) component of AC susceptibility of **1** measured at an applied DC field 0.7 T in the temperature range 1.8 - 5.25 K including the fit of the modified Debye model (an additional Debye function was included to take into account a very small contribution at higher frequencies).



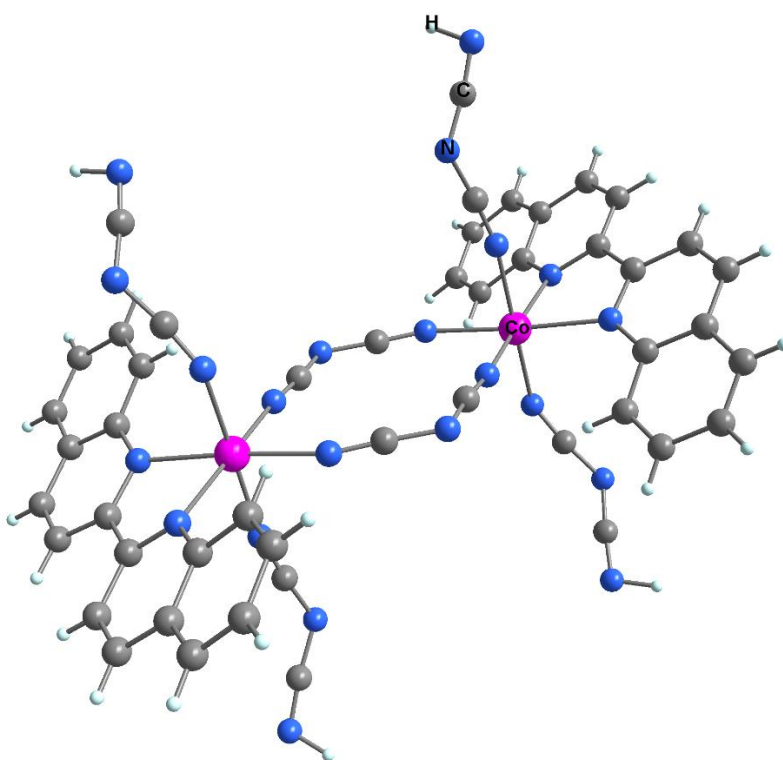
**Fig. S5** Structure of all fragments used for the SA-CASSCF/NEVPT2 calculations of ZFS parameters in compound **1**, a)  $[\text{Co}(\text{biq})(\text{dca})_4]^{2-}$ , b)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_4]^{2+}$ , c)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_2\text{Ca}]^{2+}$ , d)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{Na}_2\text{Ca}]^{2+}$ , the same type of fragments was used for compound **2**.



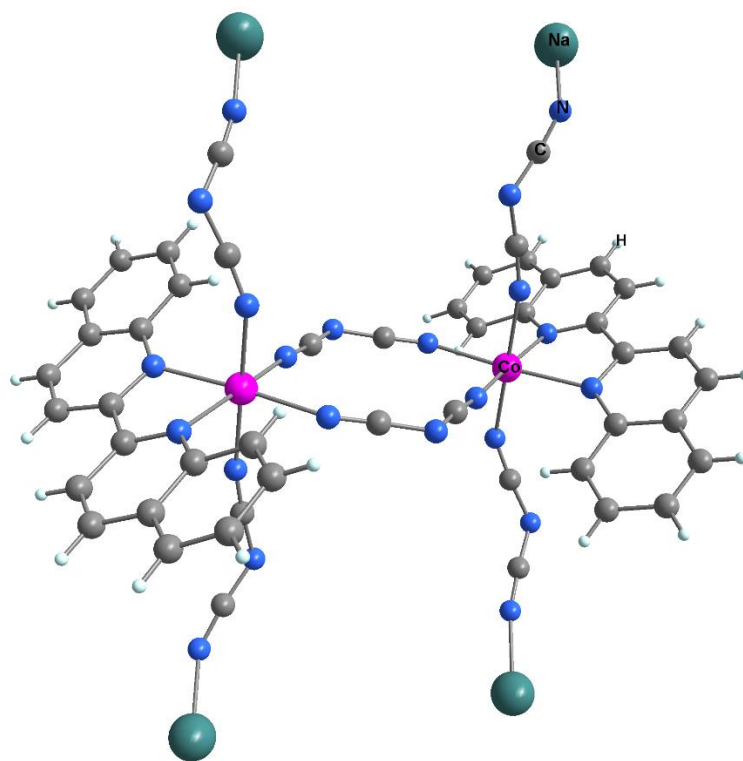
**Fig. S6** The energies of 6 lowest multiplets as obtained from SA-CASSCF/NEVPT2 calculations for: a)  $[\text{Co}(\text{biq})(\text{dca})_4]^{2-}$ , b)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_4]^{2+}$ , c)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_2\text{Ca}]^{2+}$ , d)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{Na}_2\text{Ca}]^{2+}$ . Energies obtained using Griffith-Figgis (GF) Hamiltonian Eq. 3 with the set of the parameters obtained from the fit of  $\mu_{\text{eff}}(\text{T})$  are included.



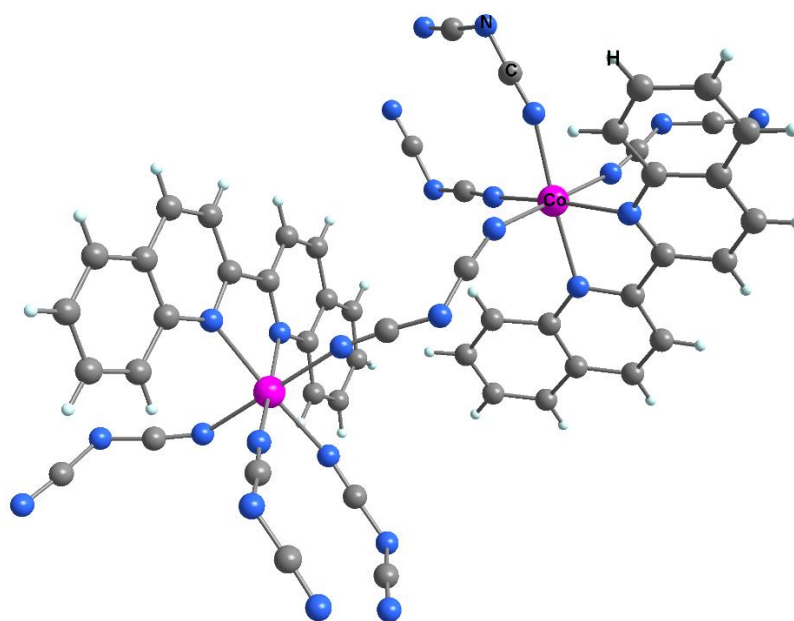
a)



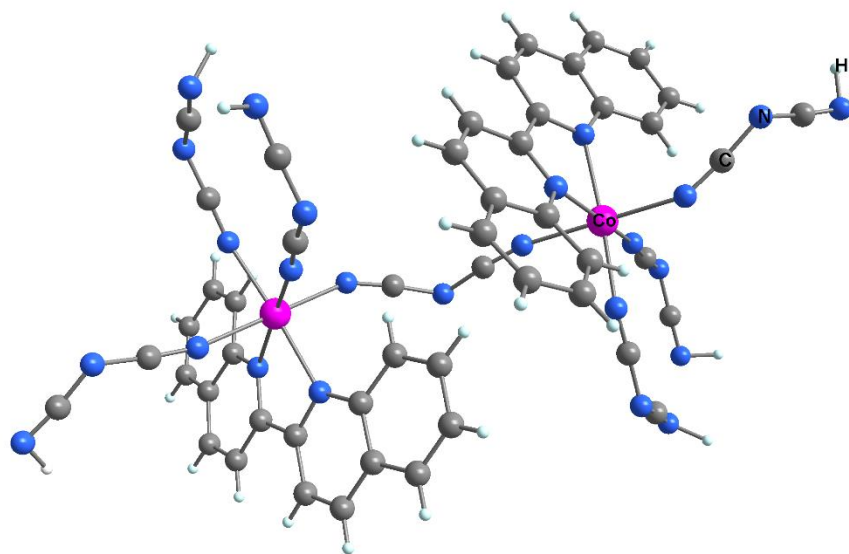
b)



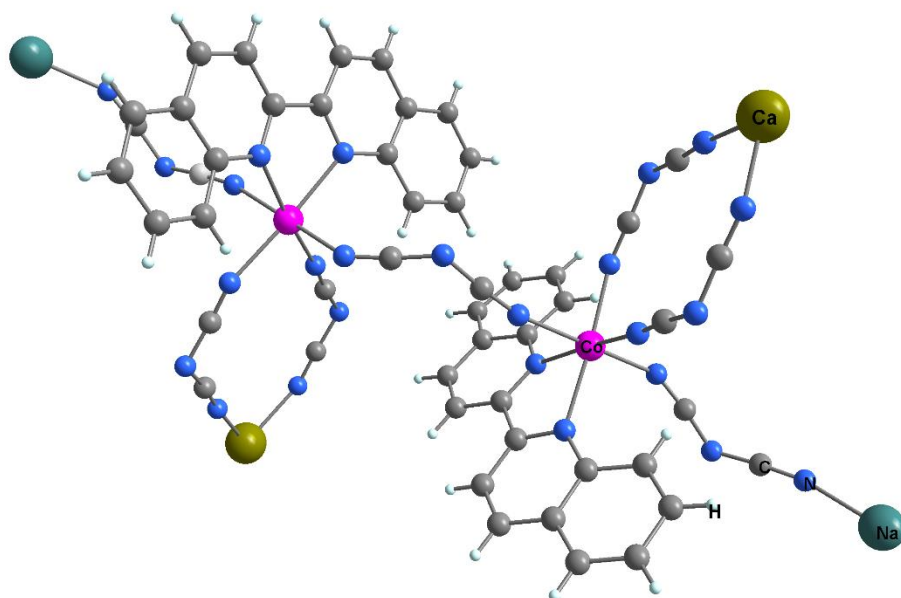
c)



d)



e)



f)

**Fig. S7** Structure of all fragments used for the BS DFT calculations of two different exchange paths in compound **1**, a)  $[\text{Co}(\text{biq})(\text{dca})_2-(\mu_{1,5}\text{-dca})_2\text{-Co}(\text{biq})(\text{dca})_2]^{2-}$ , b)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{H}_2-(\mu_{1,5}\text{-dca})_2\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{H}_2]^{2+}$ , c)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{Na}_2-(\mu_{1,5}\text{-dca})_2\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{Na}_2]^{2+}$ , d)  $[(\text{dca})\text{-Co}(\text{biq})(\text{dca})_2-(\mu_{1,5}\text{-dca})\text{-Co}(\text{biq})(\text{dca})_2-(\text{dca})]^{3-}$ , e)  $[\text{H}(\mu_{1,5}\text{-dca})\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{H}_2-(\mu_{1,5}\text{-dca})\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{H}_2-(\mu_{1,5}\text{-dca})\text{H}]^{3+}$ , f)  $[\text{Na}(\mu_{1,5}\text{-dca})\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{Ca}-(\mu_{1,5}\text{-dca})\text{-Co}(\text{biq})(\mu_{1,5}\text{-dca})_2\text{Ca}-(\mu_{1,5}\text{-dca})\text{Na}]^{3+}$ , the same type of fragments was used for compound **2**.



**Table S1** Comparison of magnetic parameters of other dca bridged polymeric complexes showing slow magnetic relaxation.

Compound	$U_{\text{eff}}/\text{cm}^{-1}$ (applied field/T)	$\tau_0/s$	magnetic interactions	Ref.
$[\text{Co}(\text{atz})_2(\text{dca})_2]_n$ easy-axis anisotropy	5.1 (0.1)	$1.7 \times 10^{-6}$	neglected	Palion-Gazda 2015
$[\text{Co}(\text{dca})_2(\text{NH}_2\text{pyz})_2]_n \cdot \text{H}_2\text{O}$ easy-plane anisotropy	24.5 – 28.2 (0.25) 1.62 – 2.37	$1.19 – 2.5 \times 10^{-7}$ $3.0 – 5.7 \times 10^{-5}$	neglected	Palion-Gazda 2019
$[\text{Co}_3(\text{dca})_6(\text{HOPyz})_5(\text{H}_2\text{O})_2]_n$ easy-plane anisotropy	11.6 – 20.1 (0.2) 0.33 – 1.26	$1.73 – 14.1 \times 10^{-7}$ $1.96 – 4.3 \times 10^{-5}$	$zJ = -0.3 \text{ cm}^{-1}$	Palion-Gazda 2019
$[\text{Co}(\text{bim})_2(\text{dca})_2]_n$ easy-axis anisotropy	5.33 (0.1)	$1.54 \times 10^{-6}$	neglected	Switliczka 2016
$[\text{Co}(\text{bmim})_2(\text{dca})_2]_n$ easy-axis anisotropy	13.81 (0.1)	$0.63 \times 10^{-6}$	$\Theta = -0.07 \text{ cm}^{-1}$	Switliczka 2016
$[\text{Co}(\text{pypz})(\text{dca})(\text{H}_2\text{O})]_n \cdot (\text{dca})$ easy-axis anisotropy	103 (0.4)	$1.2 \times 10^{-11}$	$zJ = -0.006 \text{ cm}^{-1}$	Switlicka 2020

*atz* = 2-amino-1,3,5- triazine; *pyz* = pyrazine; *bim* = 1-benzylimidazole; *bmim* = 1-benzyl-2-methylimidazole; *pypz* = 2,6-bis(pyrazol-1-yl)pyridine

J. Palion-Gazda, T. Klemens, B. Machura, J. Vallejo, F. Lloret and M. Julve, *Dalton Trans.*, 2015, **44**, 2989-2992.

J. Palion-Gazda, K. Choroba, B. Machura, A. Świtlicka, R. Kruszynski, J. Cano, F. Lloret and M. Julve, *Dalton Trans.*, 2019, **48**, 17266-17280.

A. Świtlicka-Olszewska, J. Palion-Gazda, T. Klemens, B. Machura, J. Vallejo, J. Cano, F. Lloret, M. Julve, *Dalton Trans.*, 2016, **45**, 10181–10193.

A. Świtlicka, B. Machura, M. Penkala, A. Bieńko, D. C. Bieńko, J. Titiš, C. Rajnák, R. Boča, A. Ozarowski, *Inorg. Chem. Front.*, 2020, **7**, 2637-2650.

**Table S2** The energies (in  $\text{cm}^{-1}$ ) of 6 lowest multiplets as obtained from SA-CASSCF/NEVPT2 calculations for: a)  $[\text{Co}(\text{biq})(\text{dca})_4]^{2+}$ , b)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_4]^{2+}$ , c)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{H}_2\text{Ca}]^{2+}$ , d)  $[\text{Co}(\text{biq})(\mu_{1,5}\text{-dca})_4\text{Na}_2\text{Ca}]^{2+}$ . Energies obtained using GF Hamiltonian Eq. 3 with the set of the parameters obtained from the fit of the  $\mu_{\text{eff}}(\text{T})$  are included.

multiplet	a	b	c	d	GF model
1	0	0	0	0	0
2	137.5	155.8	166.7	139.4	193.8
3	906.9	765.2	716.1	887.2	717.8
4	1166.9	1093.4	1055.8	1140.0	1028.8
5	1460.2	1252.4	1214.1	1482.8	1497.1
6	1558.9	1401.0	1361.9	1574.4	1590.1

**Table S3** Comparison of the exchange coupling  $J_{BS1}$  and  $J_{BS2}$  in **1** obtained from BS DFT calculations for different binuclear model fragments of the polymeric structure.

	B3LYP	TPSSh	PBE0
	$J_{BS1}$ (cm <sup>-1</sup> )		
[Co( <i>biq</i> )(dca) <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Co( <i>biq</i> )(dca) <sub>2</sub> ] <sup>2-</sup>	-0.44	-0.59	-0.46
[Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> ] <sup>2+</sup>	-0.36	-1.03	-0.41
[Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Na <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Na <sub>2</sub> ] <sup>2+</sup>	-0.60	-1.12	-0.91
	$J_{BS2}$ (cm <sup>-1</sup> )		
[(dca)-Co( <i>biq</i> )(dca) <sub>2</sub> -(μ <sub>1,5</sub> -dca)-Co( <i>biq</i> )(dca) <sub>2</sub> -(dca)] <sup>3-</sup>	-0.34	-0.49	-0.45
[H(μ <sub>1,5</sub> -dca)-Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca)-Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca)H] <sup>3+</sup>	-0.56	-0.84	-0.40
[Na(μ <sub>1,5</sub> -dca)-Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Ca-(μ <sub>1,5</sub> -dca)-Co( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Ca-(μ <sub>1,5</sub> -dca)Na] <sup>3+</sup>	-0.46	-0.71	-0.40

**Table S4** Comparison of the exchange coupling  $J_{BS1}$  and  $J_{BS2}$  in **2** obtained from BS DFT calculations for different binuclear model fragments of the polymeric structure.

	B3LYP	TPSSh	PBE0
	$J_{BS1}$ (cm <sup>-1</sup> )		
[Ni( <i>biq</i> )(dca) <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Ni( <i>biq</i> )(dca) <sub>2</sub> ] <sup>2-</sup>	-0.70	-1.00	0.06
[Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> ] <sup>2+</sup>	-0.48	-0.41	0.31
[Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Na <sub>2</sub> -(μ <sub>1,5</sub> -dca) <sub>2</sub> -Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Na <sub>2</sub> ] <sup>2+</sup>	0.15	-0.46	0.33
	$J_{BS2}$ (cm <sup>-1</sup> )		
[(dca)-Ni( <i>biq</i> )(dca) <sub>2</sub> -(μ <sub>1,5</sub> -dca)-Ni( <i>biq</i> )(dca) <sub>2</sub> -(dca)] <sup>3-</sup>	0.36	0.53	1.05
[H(μ <sub>1,5</sub> -dca)-Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca)-Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> H <sub>2</sub> -(μ <sub>1,5</sub> -dca)H] <sup>3+</sup>	0.45	0.65	1.11
[Na(μ <sub>1,5</sub> -dca)-Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Ca-(μ <sub>1,5</sub> -dca)-Ni( <i>biq</i> )(μ <sub>1,5</sub> -dca) <sub>2</sub> Ca-(μ <sub>1,5</sub> -dca)Na] <sup>3+</sup>	0.38	0.57	1.22