

Supporting information

An effective photocatalytic hydrogen evolution strategy based on tunable band gap $(\text{CuIn})_x\text{Zn}_{2(1-x)}\text{S}_2$ combined with amorphous molybdenum sulfide

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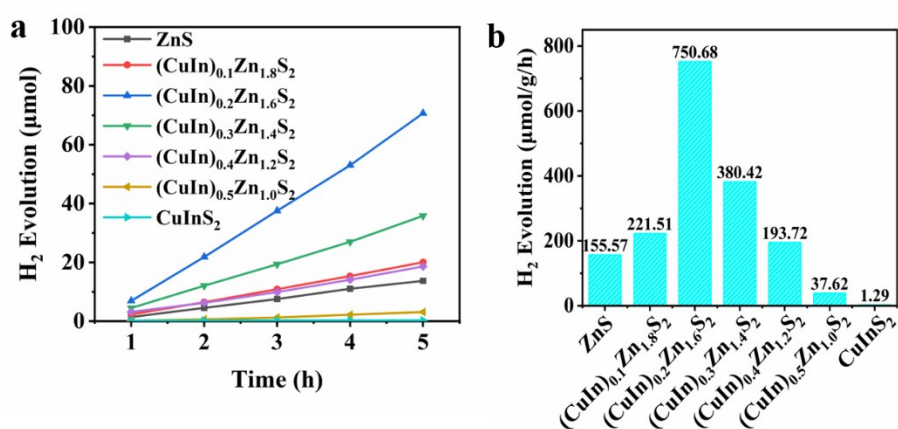


Fig. S1 (a) Time-dependent photocatalytic H₂ evolution on $(\text{CuIn})_x\text{Zn}_{2(1-x)}\text{S}_2$; (b) Comparison of the H₂ evolution rate under visible light irradiation on $(\text{CuIn})_x\text{Zn}_{2(1-x)}\text{S}_2$.

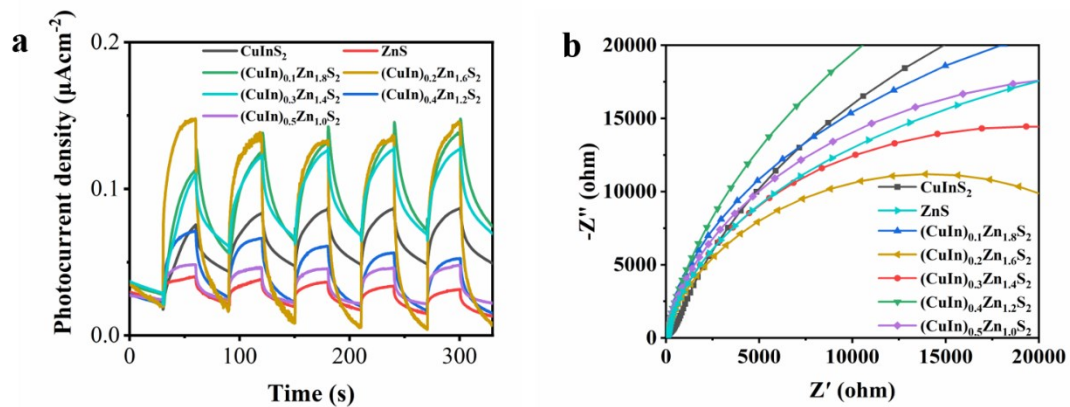


Fig. S2 The photocurrent response and EIS of the $(\text{CuIn})_x\text{Zn}_{2(1-x)}\text{S}_2$ solid solution.

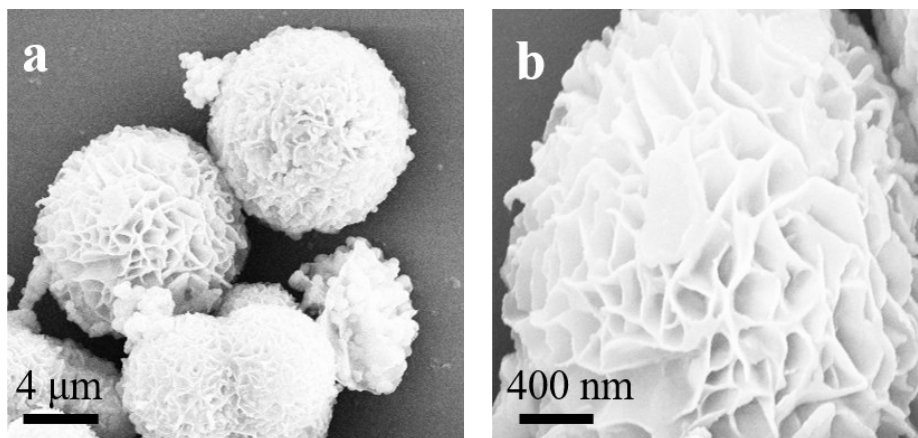


Fig. S3 SEM images of CIZS.

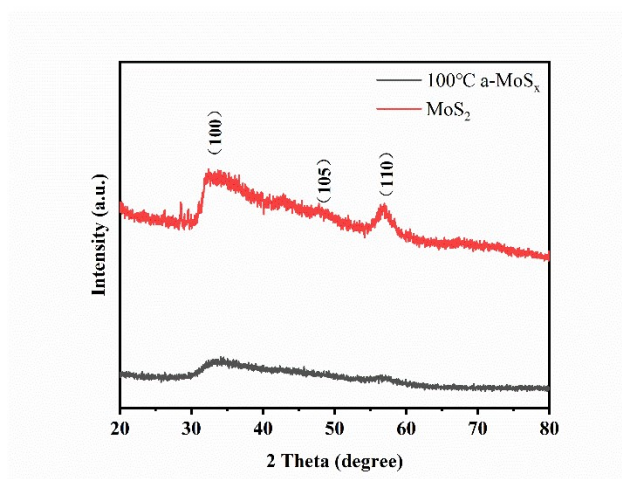


Fig. S4 The XRD patterns of the individual MoS_2 and a-MoS_x .

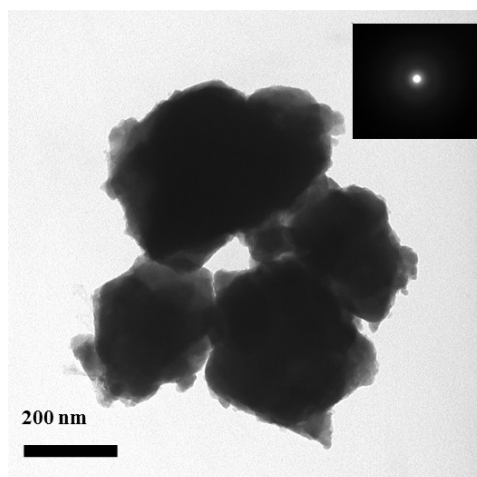


Fig. S5 TEM image of a-MoS_x and the electron diffraction spectrum of the selected region.

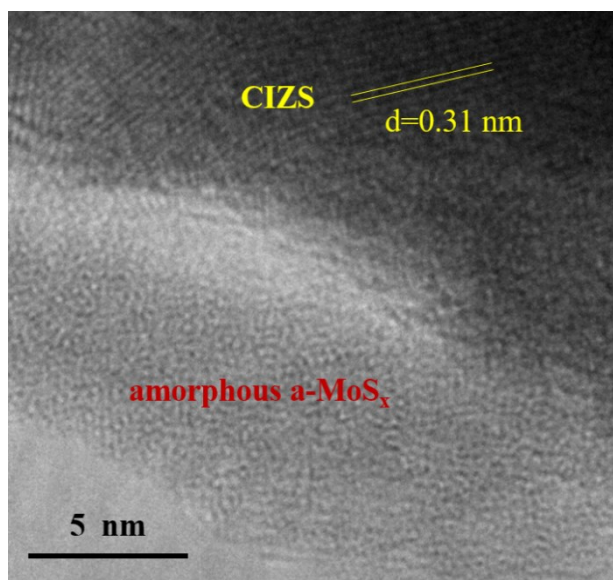


Fig. S6 The high-resolution TEM image of a-MoS_x/CIZS.

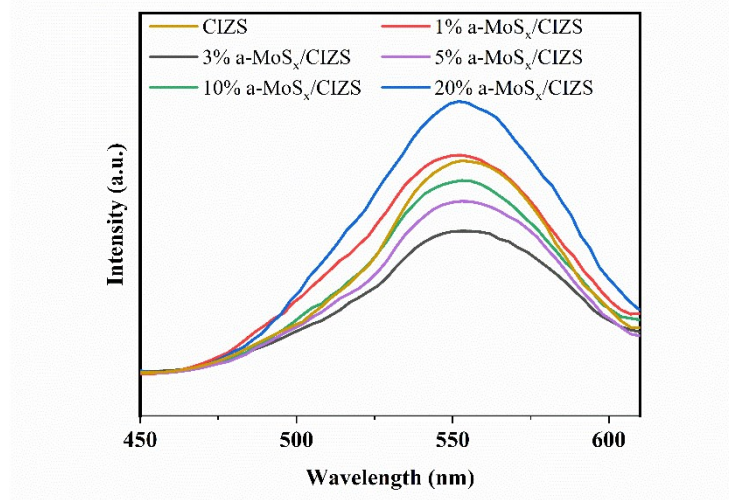


Fig. S7 PL spectra of a-MoS_x/CIZS (excitation wavelength 380 nm).

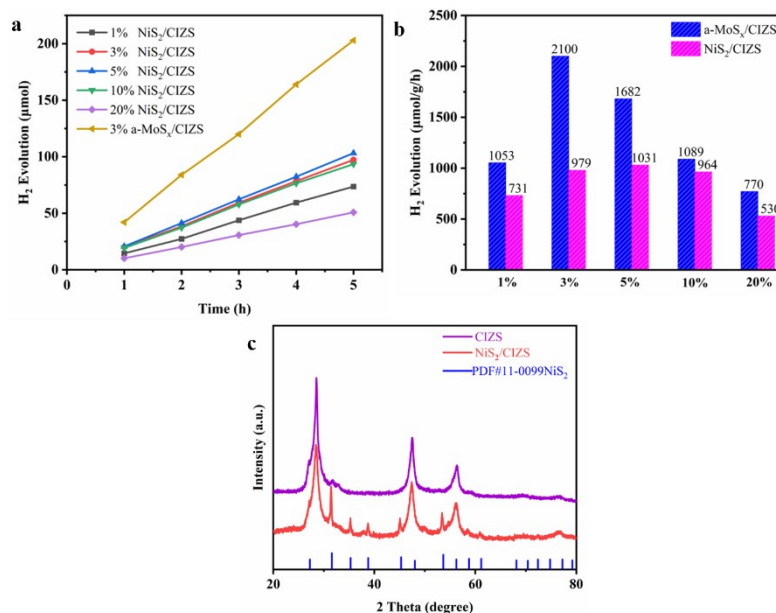


Fig. S8 (a-b)H₂ evolution rates of a-MoS_x/CIZS and NiS₂/CIZS;(c) XRD patterns of CIZS and NiS₂/CIZS.

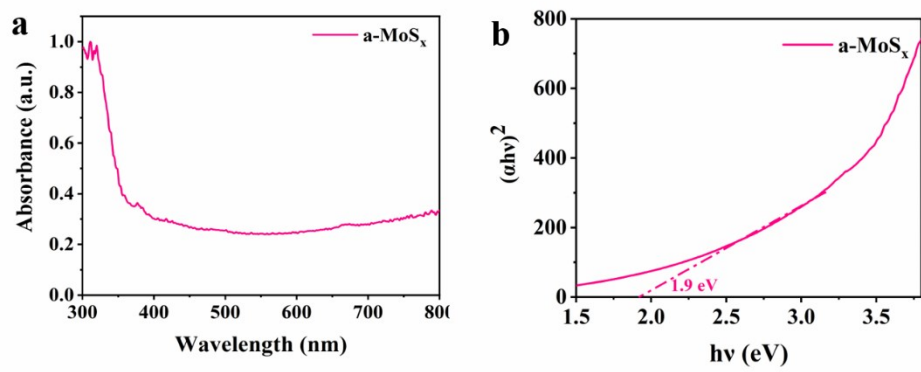


Fig. S9 UV-vis absorption spectrum and Tauc plot of a-MoS_x.

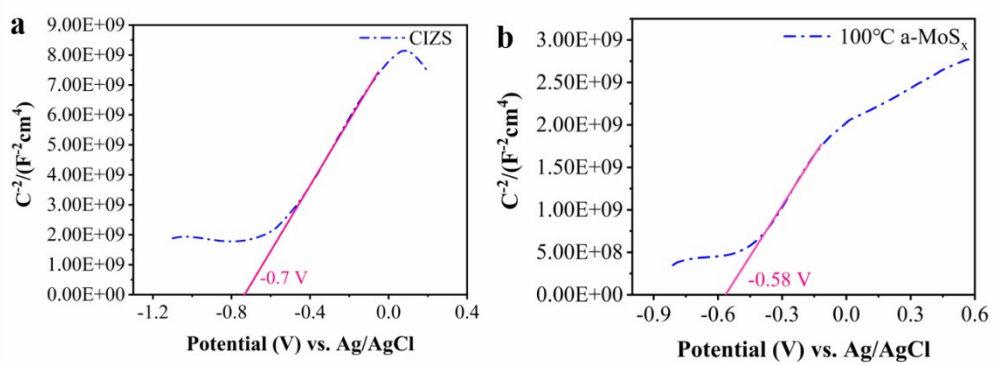


Fig. S10 Mott-Schottky plots of CIZS and a-MoS_x.