

Probing Catalytic Activity of Highly Efficient Sulfonic Acid Fabricated Cobalt Ferrite Magnetic Nanoparticles for Clean and Scalable Synthesis of Dihydro, Spiro and bis Quinazolinones

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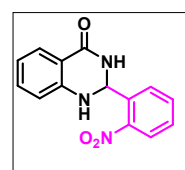
Materials and Methods

Most of the chemicals and reagents used during the experiment were procured from Sigma-Aldrich, Fluka, Merck, Spectrochem, and all of them were of analytical grade and used as such without any further purification. All the reactions were monitored using Thin-layer chromatography (TLC), which were performed on Merck precoated silica gel aluminium plated (60F254) under UV light.

Characterization Techniques. The synthesized nanoparticles acting as nanocatalyst were characterized using aid of different techniques to assure their complete magnetization, functionalization. Bruker diffractometer (D8 Discover) was used to acquire X-ray diffraction (XRD) pattern at room temperature within 2θ range 5–80° (scanning rate = 2°/min, λ = 0.15406

nm, 40 kV, 40 mA). The TEM analysis of synthesized nanoparticles were carried out on FEI TECHNAI (model number G² T20) transmission electron microscope (operated at 200 kV). SEM studies were carried out using Carl Zeiss, India (Jeol Japan Mode: JSM 6610LV). N₂ adsorption-desorption isotherm and pore size distribution was measured by Micromeritics Instrument, Gemini Model: 2380. Magnetization measurements M (T, H) of nanoparticles were determined using vibrating sample magnetometer (model number EV-9, Microsense, ADE). The Fourier transform infrared spectra (FT-IR) of the nanoparticles were recorded using a PerkinElmer Spectrum 2000. ¹H NMR spectra of the synthesized compounds were recorded on 400 MHz JEOL NMR spectrometer in deuterated dimethylsulfoxide (DMSO) and chloroform (CDCl₃). Chemical shifts (δ-values) have been reported in parts per million.

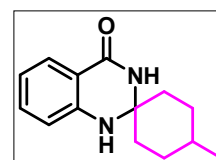
Table S1: Crystal data and structure refinement for compound **3c**.



CCDC Number	2033523
Molecular Formula	C ₁₄ H ₁₁ N ₃ O ₃
Molecular weight	269.08
Temperature	295.4
Radiation	Mo K α (λ = 0.71073)
Crystal system	Monoclinic
Space group	C2/c
Unit cell dimensions	a/Å 20.0038(9) b/Å 10.8076(4) c/Å 12.4320(5) α /° 90 β /° 109.8180(10) γ /° 90
Volume /Å ³	2528.53(18)

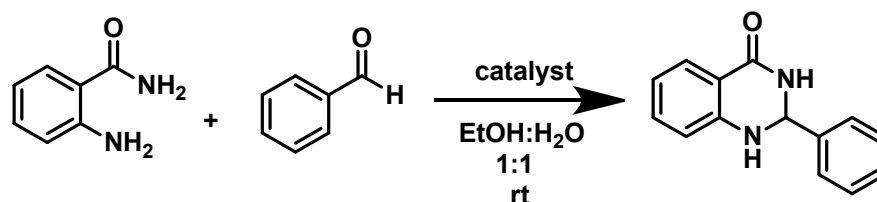
Z	50
Density ($\rho_{\text{calc}}/\text{cm}^3$)	1.409
Absorption coefficient (μ/mm^{-1})	0.102
F(000)	1112.6
2 θ range for data collection	4.32 to 56.58
Index ranges	$-26 \leq h \leq 26, -14 \leq k \leq 14, -16 \leq l \leq 16$
Reflections collected	20839
Independent reflections	3136 [$R_{\text{int}} = 0.0206, R_{\text{sigma}} = 0.0188$]
Data/restraints/parameters	3136/0/182
Goodness-of-fit on F^2	1.303
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0850, wR_2 = 0.2335$
Final R indexes [all data]	$R_1 = 0.1382, wR_2 = 0.2821$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	0.32/-0.80

Table S2: Crystal data and structure refinement for compound **5e**.



CCDC Number	2043509
Molecular Formula	C ₁₄ H ₁₈ N ₂ O
Molecular weight	230.30
Temperature	293
Radiation	Mo K α ($\lambda = 0.71073$)
Crystal system	orthorhombic
Space group	P1

Unit cell dimensions	$a/\text{\AA}$ 10.223(8) $b/\text{\AA}$ 11.333(12) $c/\text{\AA}$ 21.64(2) $\alpha/^\circ$ 90 $\beta/^\circ$ 90 $\gamma/^\circ$ 90
Volume / \AA^3	2507(4)
Z	50
Density ($\rho_{\text{calc}}/\text{cm}^3$)	1.222
Absorption coefficient (μ/mm^{-1})	0.078
F(000)	994.4
2 θ range for data collection	4.4 to 56.58
Index ranges	$-10 \leq h \leq 11$, $-14 \leq k \leq 15$, $-28 \leq l \leq 17$
Reflections collected	12964
Independent reflections	12500 [$R_{\text{int}} = 0.2383$, $R_{\text{sigma}} = 0.1422$]
Data/restraints/parameters	12500/3/1233
Goodness-of-fit on F^2	1.343
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.1484$, $wR_2 = 0.3604$
Final R indexes [all data]	$R_1 = 0.2422$, $wR_2 = 0.4592$
Largest diff. peak/hole / $e \text{\AA}^{-3}$	0.49/-0.65

Table S3: Calculation of green chemistry metrics for compound **31**.

F.W.	136.06	106.04	224.09
mmol	1.0	1.0	0.98
amount	0.136	0.106	0.220

S. No.	Parameters	Formula	Characteristics	Ideal Value	Calculated value for compound 31
1.	Environmental (E) factor	[Total mass of raw materials - the total mass of product]/mass of product	E-factor signifies the total amount of waste generated in a chemical reaction.	0	$[(0.136 + 0.106) - 0.220] / 0.220 = 0.10$
2.	Process mass intensity (PMI)	\sum (mass of stoichiometric reactants)/[mass of stoichiometric product]	PMI takes into account reaction efficiency, stoichiometry, amount of solvent and all reagent used in the chemical reaction.	1	$(0.136 + 0.106) / 0.220 = 1.1$
3.	Reaction mass efficiency (RME %)	$[\text{mass of product} / \sum (\text{mass of stoichiometric reactants})] \times 100$	RME accounts into atom economy, chemical yield and stoichiometry.	100 %	$[0.220 / (0.136 + 0.106)] \times 100 = 90.90\%$
4.	Atom economy (AE %)	$[\text{MW of product}] \div \sum (\text{MW of stoichiometric reactants}) \times 100$	Atom economy signifies the percentage of atoms wasted in chemical	100 %	$[(224.09) / (136.06 + 106.04)] \times 100 = 92.6\%$

		100	reaction. Higher the value of AE, greener is the reaction. Maximum value of atom economy is 100% which indicates that all the atoms present in reactants lies in the product.		
5.	Carbon efficiency (CE %)	[Amount of carbon in product/ Total carbon present in reactants] x 100	CE signifies the percentage of carbons in the reactants that is left in the product.	100 %	$[1 \times 14 / (1 \times 7 + 1 \times 7)] \times 100$ $= [14 / (7+7)] \times 100$ $= 100\%$

¹H and ¹³C NMR chemical shifts (3a-3n, 5a-5f, 7 and 9)

(3a)

¹H NMR (DMSO-d₆, 400 MHz), δ_{H} (ppm): 8.31 (s, 1H), 7.55 (d, J = 8.4 Hz, 3H), 7.40 (d, J = 8.4 Hz, 2H), 7.21 (t, 1H), 7.12 (s, 1H), 6.70 (d, J = 8.1 Hz, 1H), 6.64 (t, 1H), 5.71 (s, 1H)

¹³C NMR (DMSO-d₆, 101 MHz), δ_{C} (ppm): 164.01, 148.15, 141.61, 133.94, 131.77, 129.62, 127.89, 122.10, 117.81, 115.45, 114.98, 66.30.

(3b)

¹H NMR (CDCl₃ + DMSO-d₆, 400 MHz), δ_{H} (ppm): 7.92 (d, J = 7.7 Hz, 1H), 7.53 (d, J = 8.2 Hz, 2H), 7.41 (d, J = 8.1 Hz, 2H), 7.34 (t, 1H), 6.90 (t, 1H), 6.67 (d, J = 8.0 Hz, 1H), 6.29 (s, 1H), 5.89 (s, 1H), 4.37 (s, 1H).

¹³C NMR (DMSO-d₆, 101 MHz), δ_{C} (ppm): 164.00, 148.17, 141.20, 133.92, 133.48, 129.28, 128.83, 127.89, 117.80, 115.46, 114.98, 66.26.

(3c)

¹H NMR (DMSO-d₆, 400 MHz), δ_{H} (ppm): 8.19 (s, 1H), 8.03 (d, J = 8.1 Hz, 1H), 7.82 (d, J = 7.8 Hz, 1H), 7.75 (t, 1H), 7.60 (q, J = 7.8 Hz, 2H), 7.20-7.24 (m, 1H), 6.97 (s, 1H), 6.66-6.74 (m, 2H), 6.30 (s, 1H).

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 163.90, 148.17, 147.65, 136.42, 134.45, 134.09, 130.43, 129.46, 127.84, 125.25, 118.21, 117.09, 115.43, 115.03, 62.68.

(3d)

^1H NMR (DMSO- d_6 , 400 MHz), δ_H (ppm): 8.50 (s, 1H), 8.22 (d, $J = 8.7$ Hz, 2H), 7.71 (d, $J = 8.5$ Hz, 2H), 7.57 (d, $J = 7.8$ Hz, 1H), 7.31 (s, 1H), 7.20-7.24 (m, 1H), 6.73 (d, $J = 8.1$ Hz, 1H), 6.65 (t, 1H), 5.88 (s, 1H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 163.82, 149.86, 147.92, 147.77, 134.10, 128.56, 127.93, 124.12, 117.99, 115.41, 115.07, 65.77

(3e)

^1H NMR ($\text{CDCl}_3 + \text{DMSO-}d_6$, 400 MHz), δ_H (ppm): 8.24 (s, 1H), 7.66 (m, 3H), 7.49 (d, $J = 8.2$ Hz, 1H), 7.26 (t, 1H), 7.04 (s, 1H), 6.70-6.77 (m, 2H), 6.12 (s, 1H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 169.54, 152.25, 141.46, 139.78, 138.70, 137.98, 134.57, 134.17, 132.72, 132.39, 123.20, 119.70, 68.67.

(3f)

^1H NMR (DMSO- d_6 , 400 MHz), δ_H (ppm): 8.22 (s, 1H), 7.61 (d, $J = 9.0$ Hz, 1H), 7.37 (d, $J = 8.0$ Hz, 2H), 7.18-7.26 (m, 3H), 7.05 (s, 1H), 6.74 (d, $J = 8.0$ Hz, 1H), 6.67 (t, 1H), 5.71 (s, 1H), 2.30 (s, 3H).

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 164.07, 148.35, 139.13, 138.15, 133.69, 129.25, 127.78, 127.24, 117.50, 115.46, 114.86, 66.82, 21.19

(3g)

^1H NMR (DMSO- d_6 , 400 MHz), δ_H (ppm): 8.34 (s, 1H), 7.90 (d, $J = 8.2$ Hz, 2H), 7.54-7.57 (m, 3H), 7.18-7.22 (m, 2H), 6.70 (d, $J = 8.2$ Hz, 1H), 6.63 (t, 1H), 5.78 (s, 1H), 3.44 (brs, 1H).

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 167.54, 163.96, 148.22, 146.72, 133.96, 129.92, 127.90, 127.28, 117.80, 115.42, 114.98, 66.50.

(3h)

^1H NMR (DMSO- d_6 , 400 MHz), δ_H (ppm): 8.36 (s, 1H), 7.91 (s, 1H), 7.86-7.89 (m, 3H), 7.67 (d, $J = 8.7$, 1H), 7.61 (d, $J = 7.8$, 1H), 7.48 (m, 2H), 7.19-7.23 (t, 1H), 6.74 (d, $J = 7.8$ Hz, 1H), 6.64 (t, 1H), 5.90 (s, 1H), 3.86 (brs, 1H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_c (ppm): 164.22, 148.40, 139.39, 133.97, 133.44, 132.98, 128.67, 128.50, 128.11, 127.95, 126.94, 126.35, 125.34, 117.74, 115.43, 114.82, 67.26.

(3i)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): 8.43 (s, 1H), 7.59 (d, $J = 7.7$ Hz, 1H), 7.41 (d, $J = 4.9$ Hz, 1H), 7.23 (t, 1H), 7.10 (d, $J = 8.3$ Hz, 1H), 6.94 (m, 1H), 6.74 (d, $J = 8.1$ Hz, 1H), 6.67 (t, 1H), 5.99 (s, 1H).

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): 163.68, 147.78, 146.98, 133.91, 127.85, 127.00, 126.41, 126.23, 118.06, 115.64, 115.25, 63.10.

(3j)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.10 (d, $J = 7.7$ Hz, 1H), 8.04 (s, 1H), 7.79 (d, $J = 7.6$ Hz, 2H), 7.51-7.59 (m, 2H), 7.04-7.10 (m, 2H), 6.35 (s, 1H), 6.13 (s, 1H), 5.12 (s, 1H)

^{13}C NMR (CDCl_3 + DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 169.60, 152.35, 147.49, 138.70, 137.05, 135.22, 132.89, 130.86, 127.34, 123.39, 119.85, 119.59, 72.34.

(3k)

^1H NMR (CDCl_3 , 400 MHz), δ_{H} (ppm): δ 7.93 (dd, $J = 7.8$, 1H), 7.30-7.35 (m, 1H), 7.19 (s, 1H), 7.03 (dd, $J = 8.2$, 2.0 Hz, 1H), 6.85-6.91 (m, 2H), 6.66 (d, $J = 8.0$ Hz, 1H), 5.83 (s, 1H), 5.79 (s, 1H), 4.37 (s, 1H), 3.90 (s, 6H)

^{13}C NMR (CDCl_3 , 101 MHz), δ_{C} (ppm): δ 164.93, 150.51, 149.66, 147.40, 134.11, 130.93, 128.80, 120.25, 119.77, 115.70, 114.64, 111.10, 109.82, 69.11, 56.10

(3l)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.27 (s, 1H), 7.59 (d, $J = 7.7$ Hz, 1H), 7.47 (d, $J = 6.9$ Hz, 2H), 7.31-7.37 (m, 3H), 7.19-7.23 (m, 1H), 6.73 (d, $J = 8.1$ Hz, 1H), 6.64 (t, 1H), 5.73 (s, 1H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 164.15, 148.41, 142.16, 133.85, 128.98, 128.86, 127.89, 127.40, 117.65, 115.49, 114.94, 67.10.

(3m)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.28 (s, 1H), 7.61 (d, $J = 7.8$ Hz, 1H), 7.52 (dd, $J = 8.5$, 5.5 Hz, 2H), 7.23 (dt, $J = 20.2$, 8.7 Hz, 2H), 6.67-6.76 (m, 2H), 5.77 (d, $J = 1.8$ Hz, 1H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 164.18, 163.76, 148.14, 135.21, 133.98, 127.84, 117.83, 115.46, 115.25, 114.93, 66.27.

(3n)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): 8.77 (d, $J = 9.3$ Hz, 1H), 8.38 (s, 1H), 8.29 (s, 4H), 8.23 (d, $J = 9.3$ Hz, 1H), 8.19 (d, $J = 2.9$ Hz, 1H), 8.09 (s, 1H), 7.72 (d, $J = 7.7$ Hz, 1H), 7.25-7.29 (m, 1H), 7.19 (s, 1H), 6.83 (s, 1H), 6.74-6.78 (m, 2H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 164.66, 149.21, 133.88, 133.57, 131.76, 131.28, 130.69, 128.96, 128.36, 128.13, 128.04, 127.88, 126.94, 126.69, 126.09, 125.97, 125.28, 124.73, 124.44, 124.16, 117.96, 115.71, 115.12, 66.08.

(5a)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 7.94 (s, 1H), 7.50 (d, $J = 7.7$ Hz, 1H), 7.15 (t, 1H), 6.69 (d, $J = 8.0$ Hz, 1H), 6.60 (s, 1H), 6.55 (t, 1H), 1.78-1.89 (m, 4H), 1.46 (s, 10H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 163.54, 147.38, 133.66, 127.58, 116.78, 114.89, 114.79, 71.87, 36.09, 28.21, 24.61, 21.27

(5b)

^1H NMR (CDCl_3 , 400 MHz), δ_{H} (ppm): δ 7.84 (d, $J = 7.7$, 1H), 7.26 (t, 1H), 6.79 (t, 1H), 6.63 (d, $J = 8.0$ Hz, 1H), 6.48 (s, 1H), 4.38 (s, 1H), 1.81 (s, 4H), 1.50-1.66 (m, 4H), 1.45 (dd, $J = 11.2, 5.4$ Hz, 2H)

^{13}C NMR (CDCl_3 , 101 MHz), δ_{C} (ppm): δ 164.33, 145.71, 133.97, 128.46, 118.82, 115.32, 114.88, 68.48, 37.88, 24.76, 22.01.

(5c)

^1H NMR (CDCl_3 , 400 MHz), δ_{H} (ppm): δ 7.84 (d, $J = 7.7$, 1H), 7.23-7.27 (m, 1H), 6.77 (t, 2n), 6.61 (d, $J = 8.1$ Hz, 1H), 4.44 (s, 1H), 2.00 (m, 4H), 1.48-1.60 (m, 8H)

^{13}C NMR (CDCl_3 , 101 MHz), δ_{C} (ppm): δ 164.20, 145.76, 133.88, 128.35, 118.67, 115.30, 114.89, 72.57, 41.58, 29.09, 21.63.

(5d)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.08 (s, 1H), 7.58 (d, $J = 9.0$ Hz, 1H), 7.19-7.23 (m, 1H), 6.70 (d, $J = 8.0$ Hz, 1H), 6.63 (t, 1H), 1.79 (s, 4H), 1.67 (s, 4H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 163.90, 147.97, 133.45, 127.70, 117.00, 115.05, 114.81, 77.66, 22.46

(5e)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 7.83 (s, 1H), 7.52 (d, $J = 7.7$ Hz, 1H), 7.16 (t, 1H), 6.83 (d, $J = 8.1$ Hz, 1H), 6.51-6.64 (m, 2H), 1.87 (d, $J = 10.3$ Hz, 2H), 1.27-1.44 (m, 7H), 0.85 (s, 3H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 163.85, 147.12, 127.54, 116.99, 115.20, 115.02, 114.83, 114.74, 68.12, 37.05, 31.48, 29.84, 29.20, 22.19

(5f)

^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.11-8.17 (m, 1H), 7.68-7.79 (m, 1H), 7.54 (dd, $J = 7.8, 1.8$ Hz, 1H), 7.19-7.24 (m, 1H), 6.78 (q, 1H), 6.62 (t, 1H), 3.96 (t, 2H), 3.75-3.85 (m, 2H), 1.80 (dd, $J = 13.5, 4.8$ Hz, 4H)

^{13}C NMR (DMSO- d_6 , 101 MHz), δ_{C} (ppm): δ 169.66, 163.62, 153.69, 146.76, 134.39, 128.22, 117.99, 115.67, 114.78, 67.12, 42.51, 36.57

(7)

Proton-NMR (400 MHz, CHLOROFORM- D) δ_{H} (ppm): δ 7.80 (s, 2n), 7.54 (d, $J = 6.3$ Hz, 2n), 7.20 (t, $J = 8.5$ Hz, 2n), 6.75-6.69 (2n), 6.60-6.64 (m, 2n), 6.42 (s, 2n), 1.75-1.90 (m, 8n)

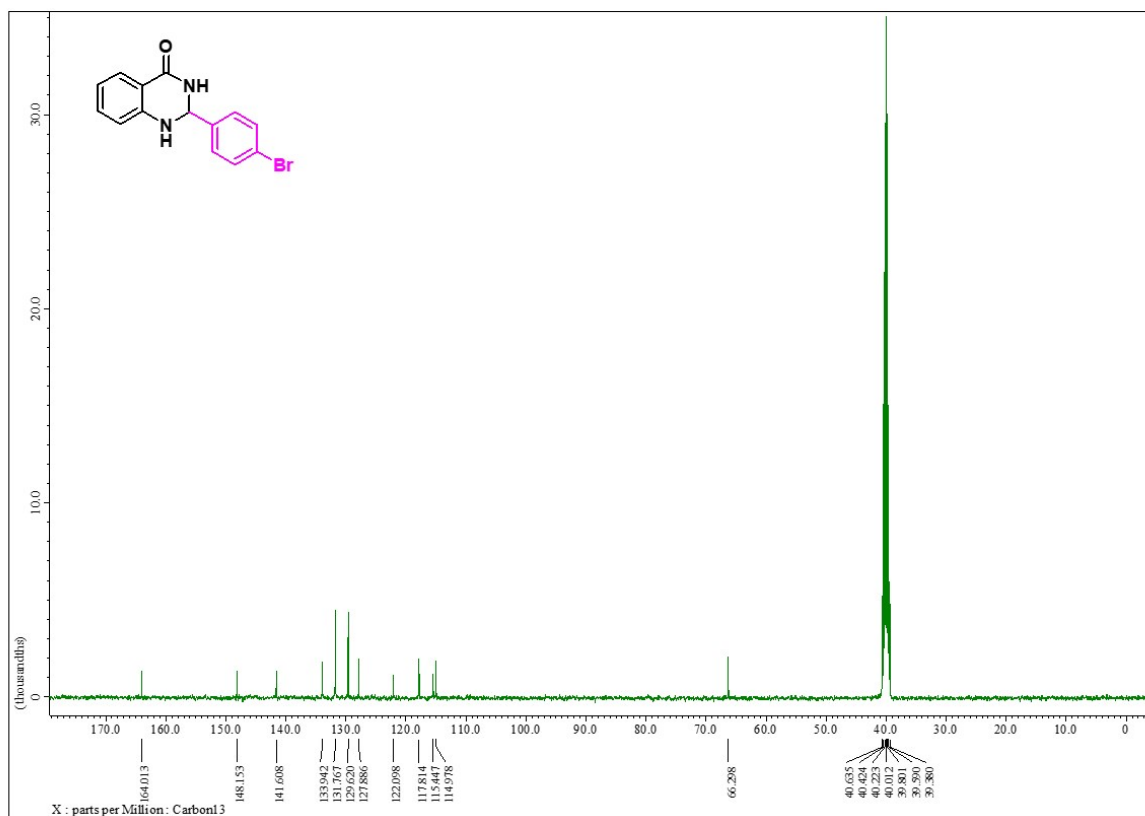
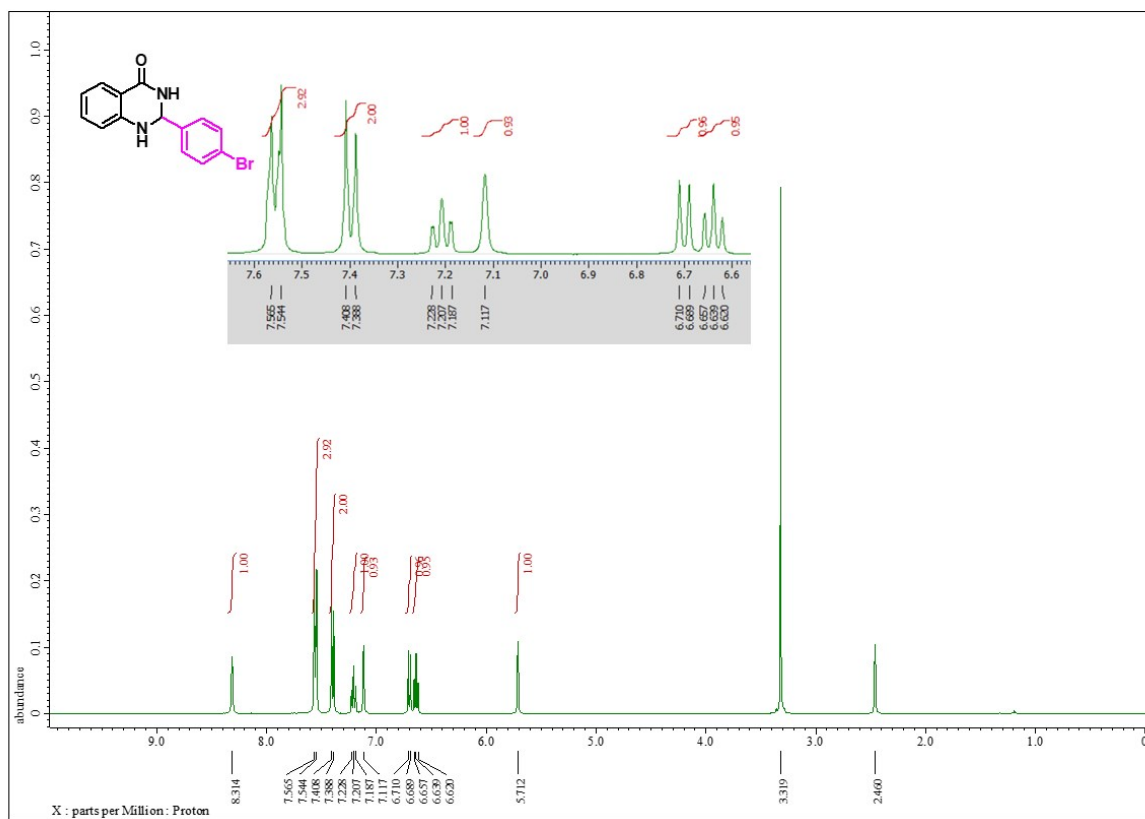
Carbon13-NMR (DMSO- D_6 , 100 MHz) δ_{C} (ppm): δ 163.80, 146.97, 133.87, 127.85, 117.56, 115.38, 115.32, 67.26, 32.58.

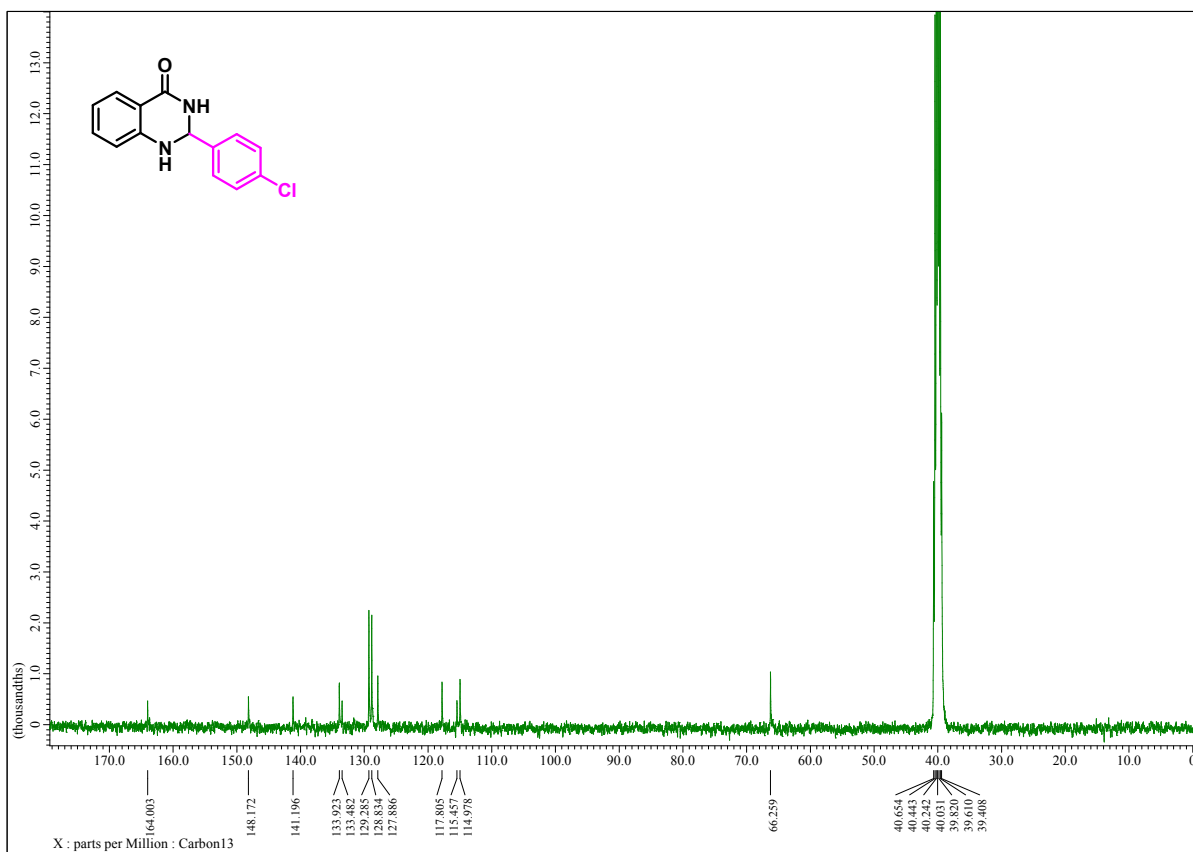
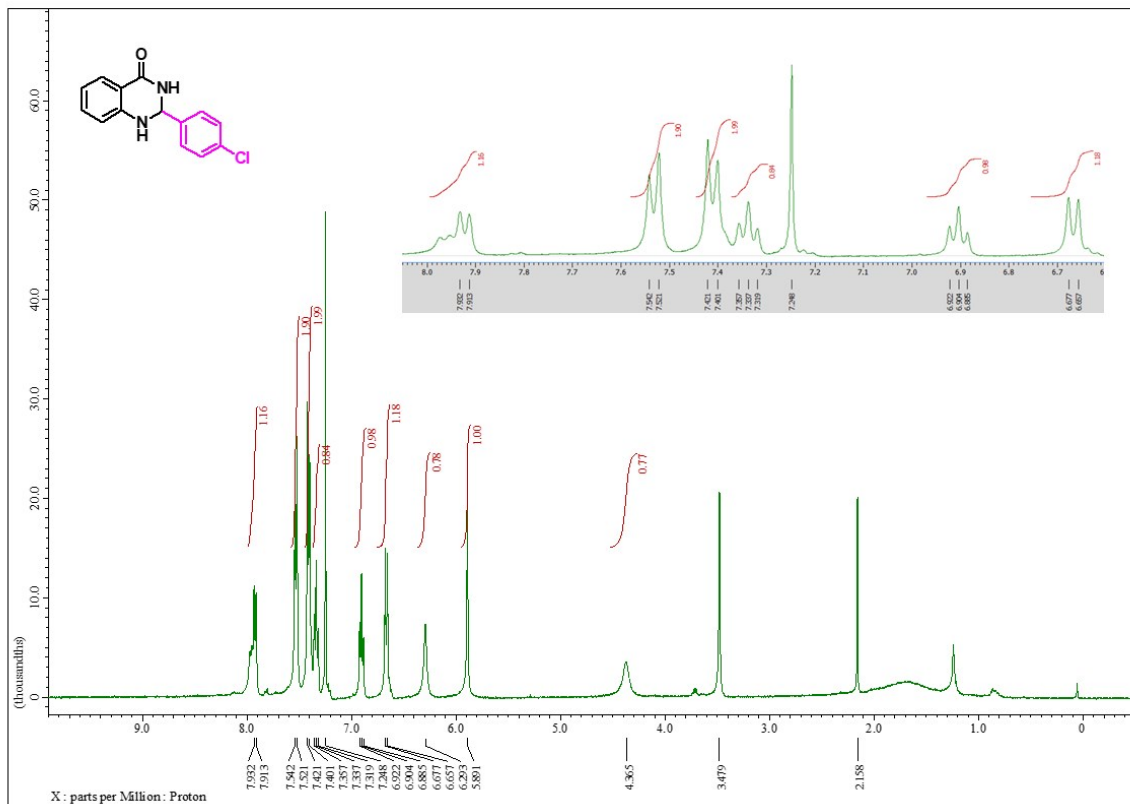
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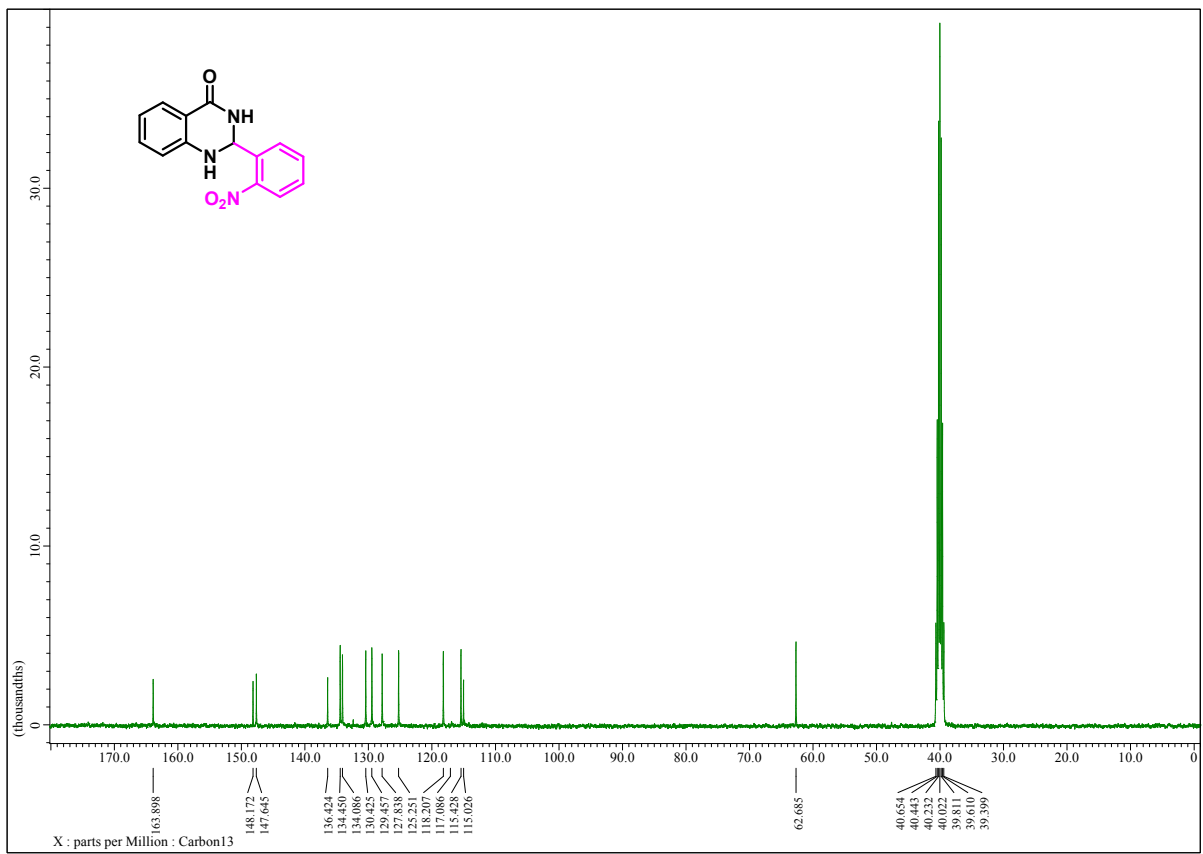
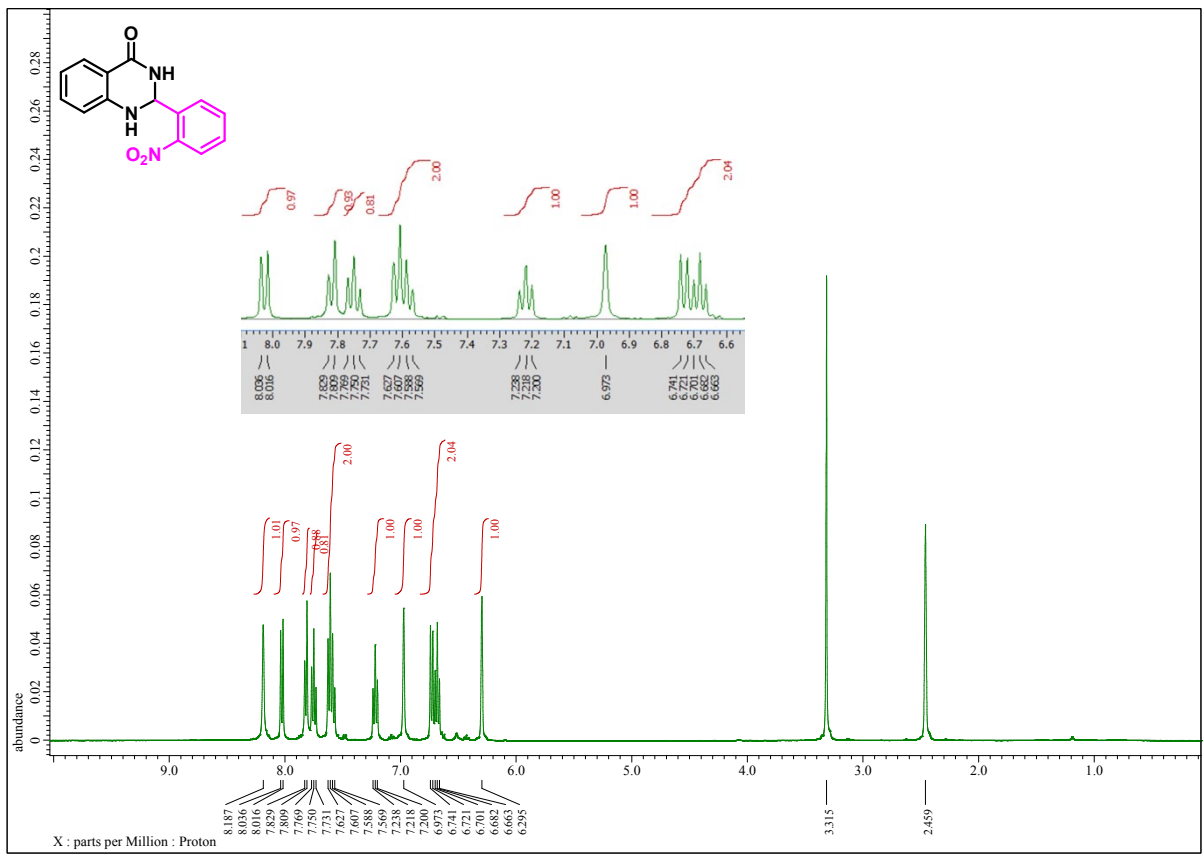
^1H NMR (DMSO- d_6 , 400 MHz), δ_{H} (ppm): δ 8.33 (s, 2H), 7.54 (d, $J = 6.6$ Hz, 2H), 7.43 (s, 4H), 7.19 (t, 2H), 7.12 (s, 2H), 6.68 (d, $J = 8.1$ Hz, 2H), 6.61 (t, 2H), 5.69 (s, 2H)

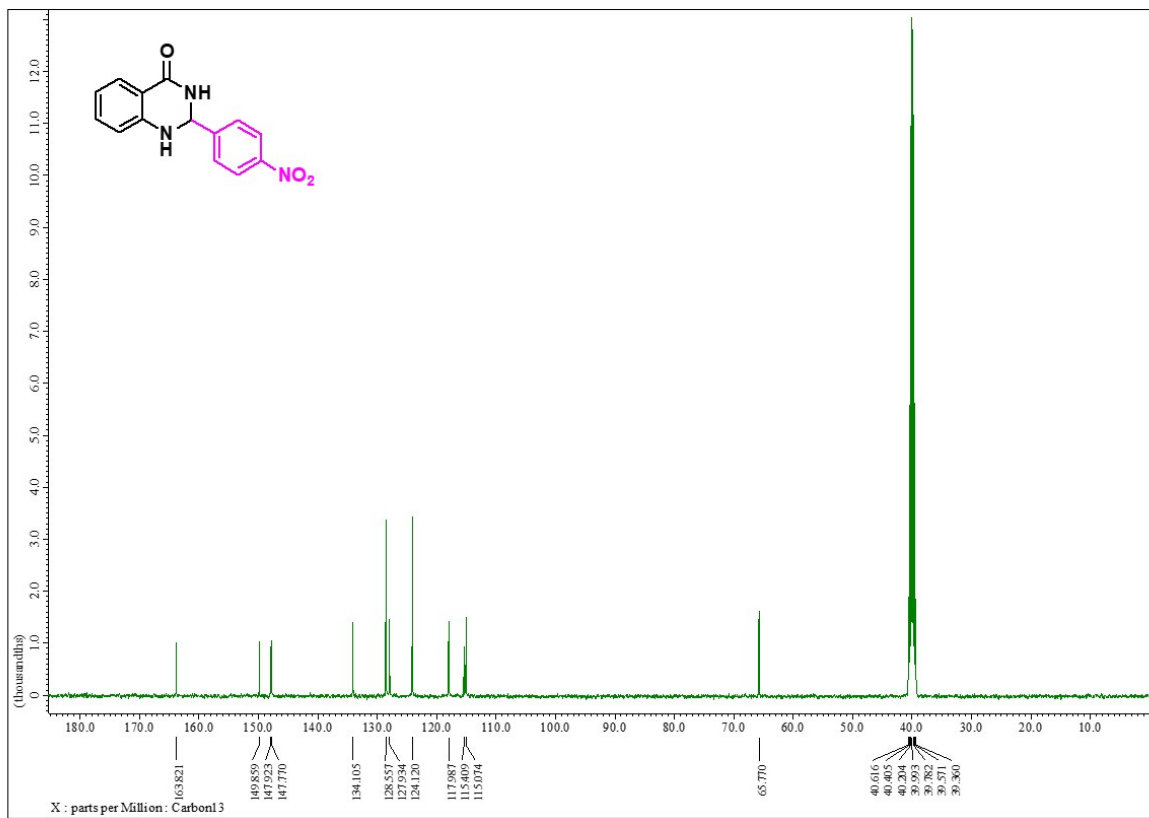
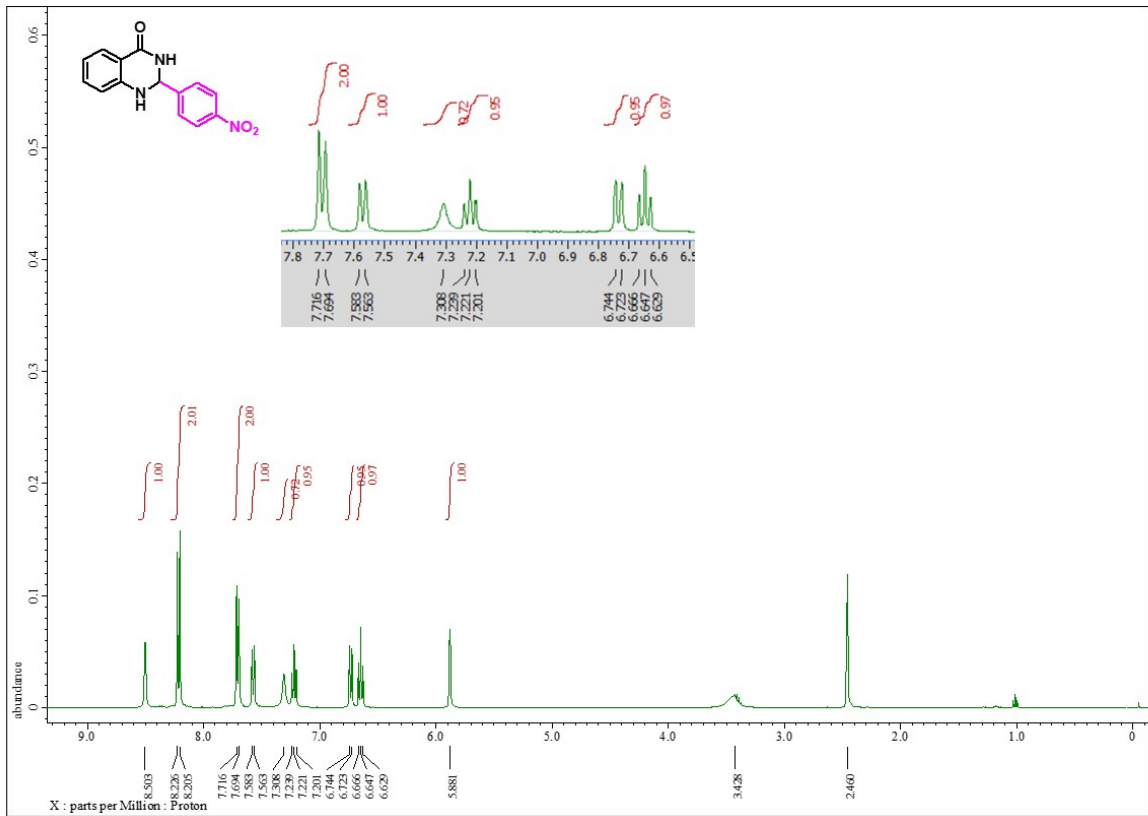
^{13}C NMR (DMSO- d_6 , 100 MHz), δ_{C} (ppm): δ 164.09, 148.20, 142.49, 133.87, 127.87, 127.19, 117.63, 115.49, 114.96, 66.42.

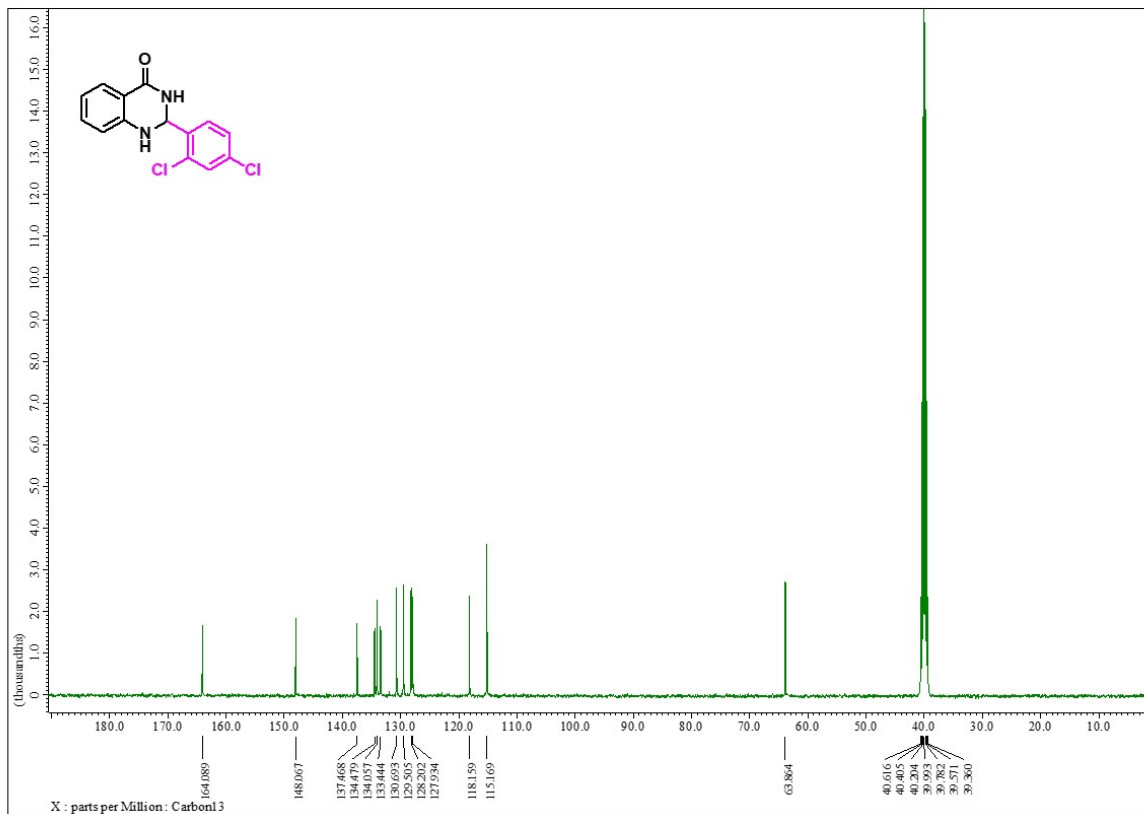
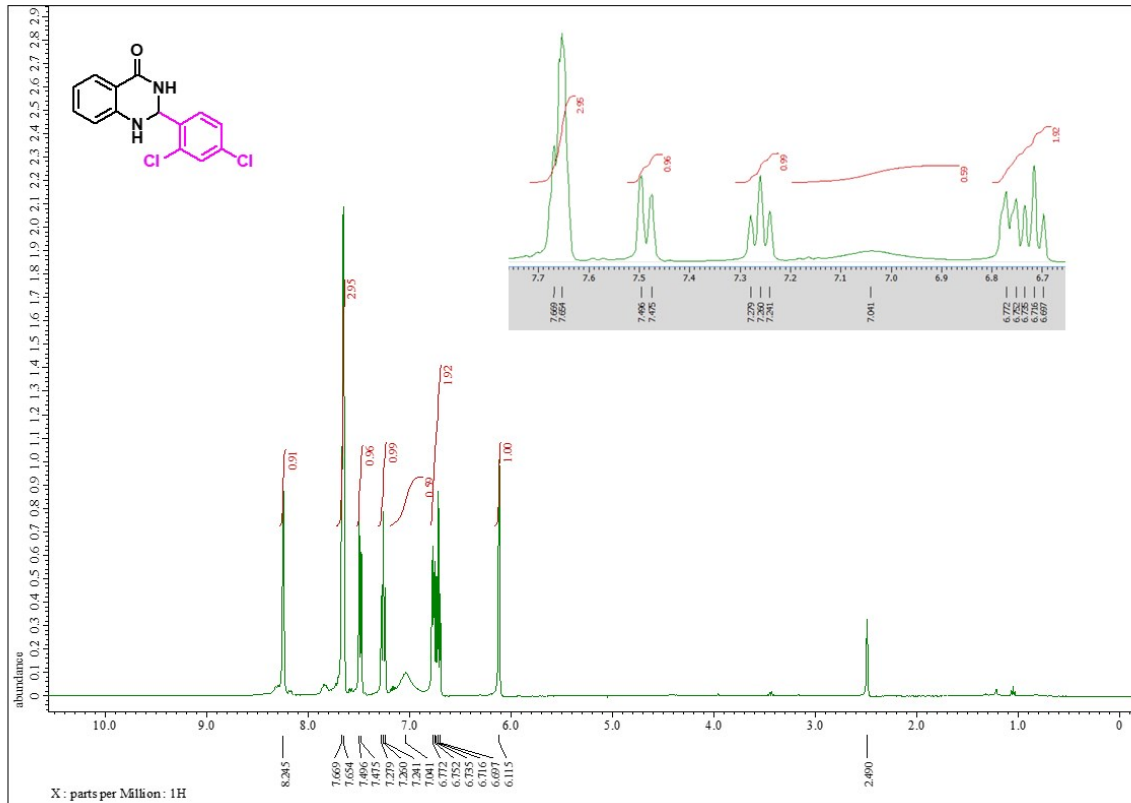
¹H and ¹³C NMR spectra

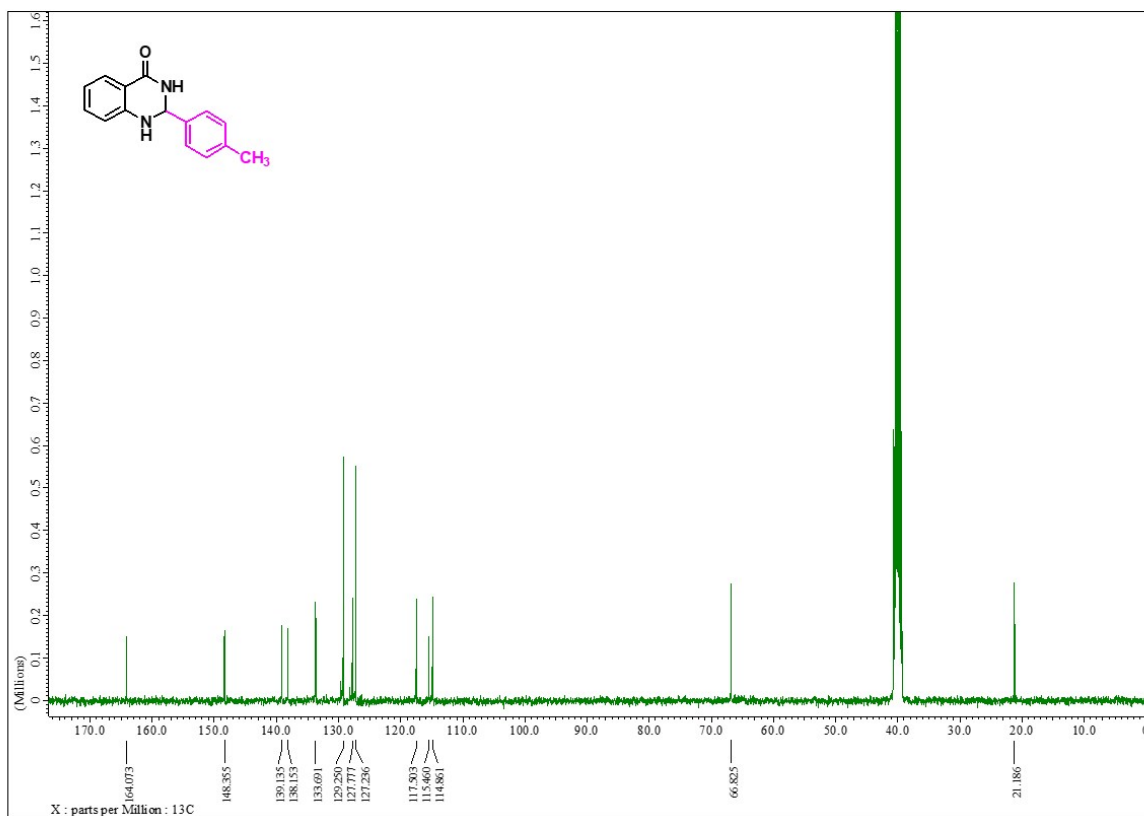
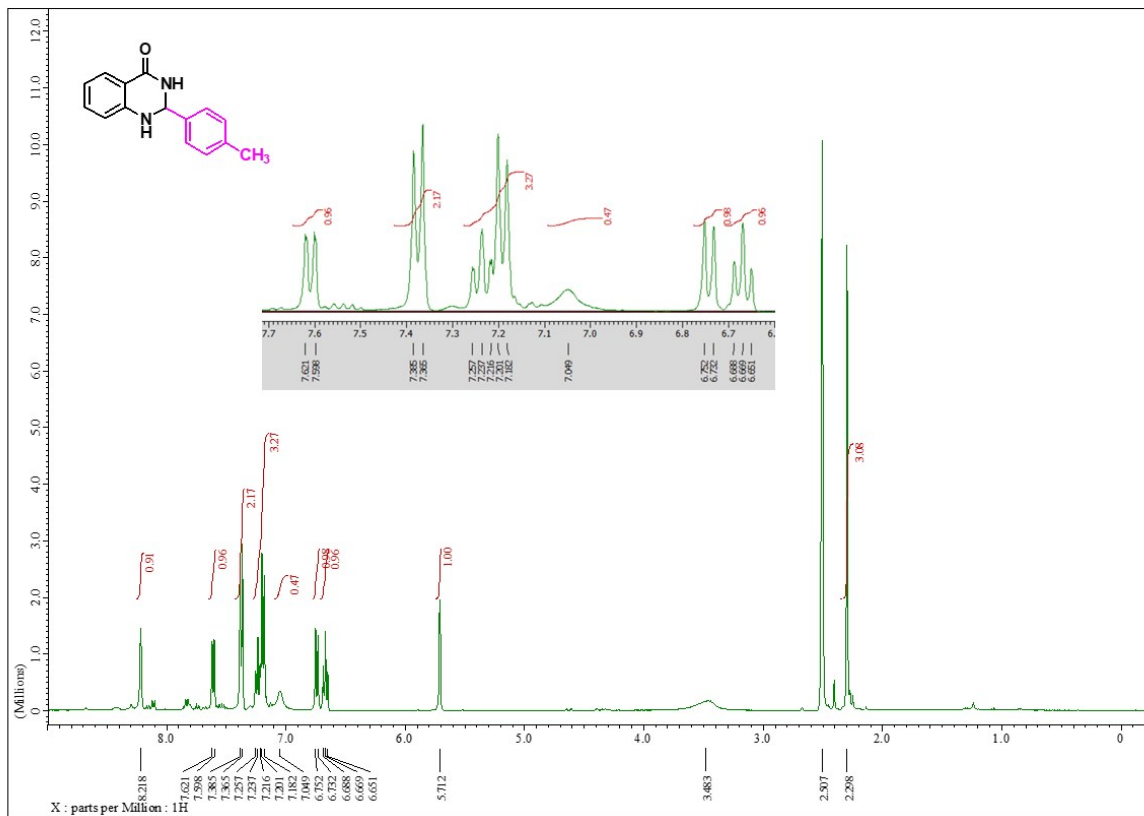


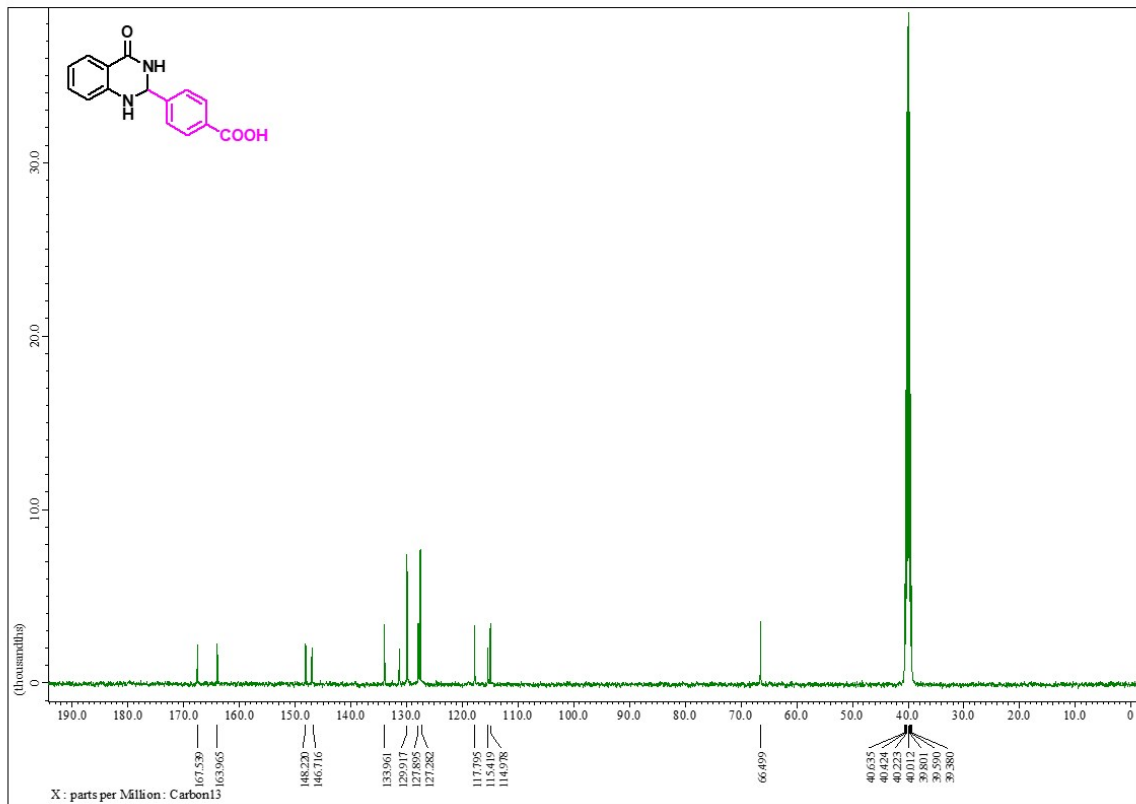
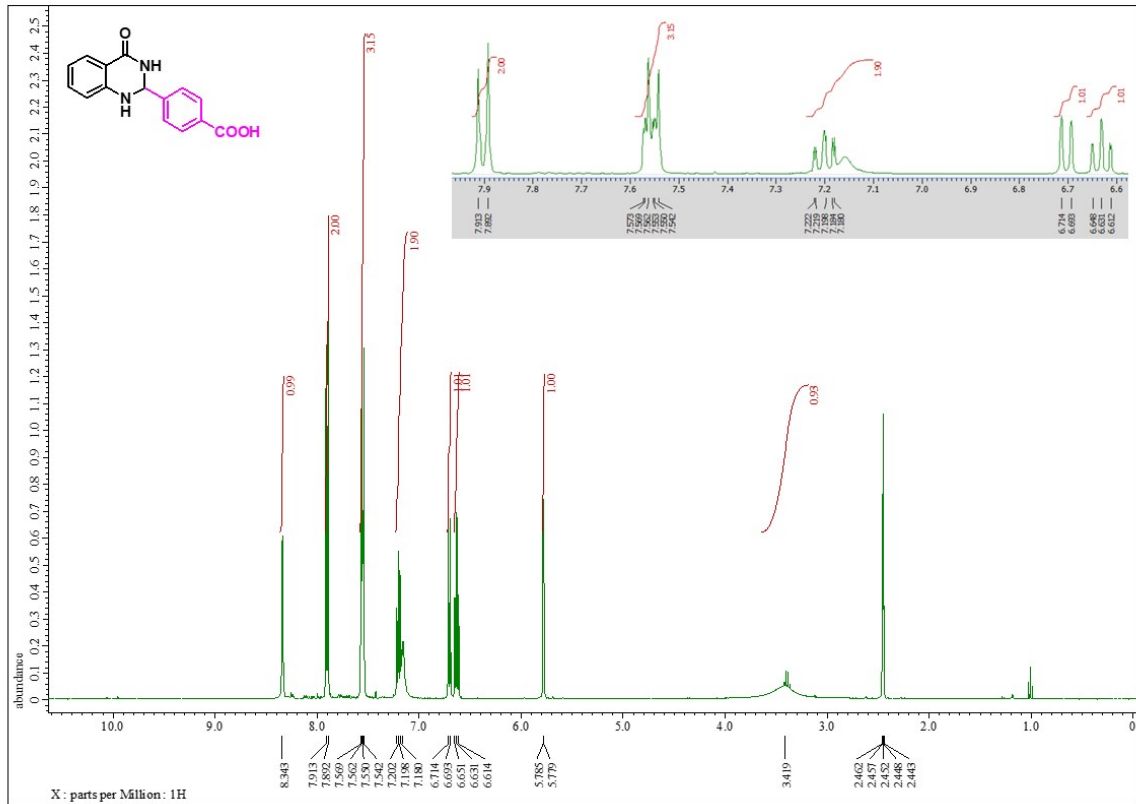


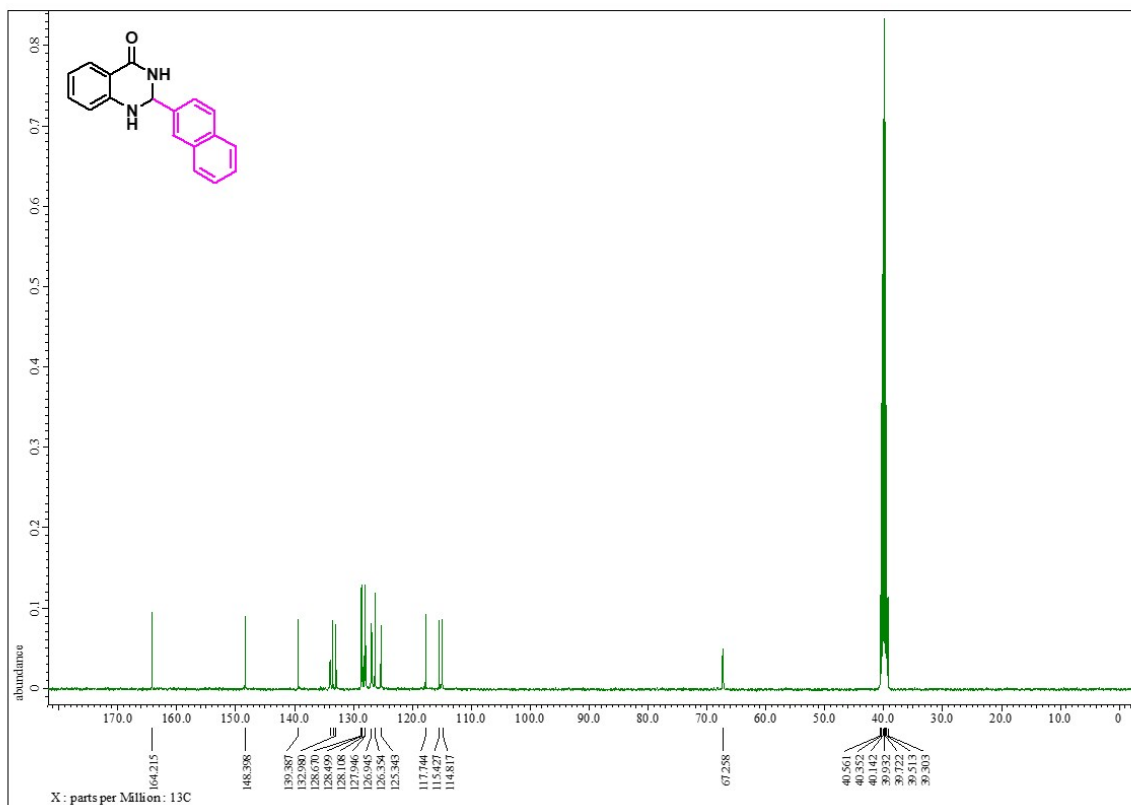
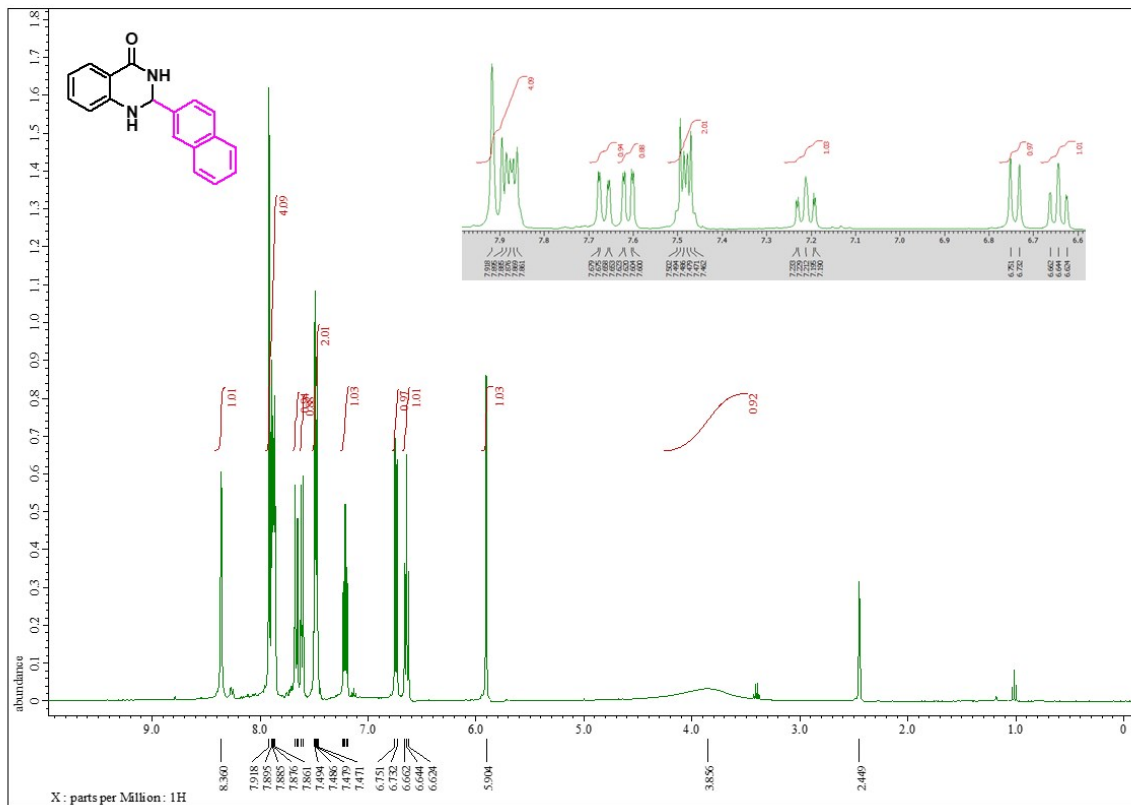


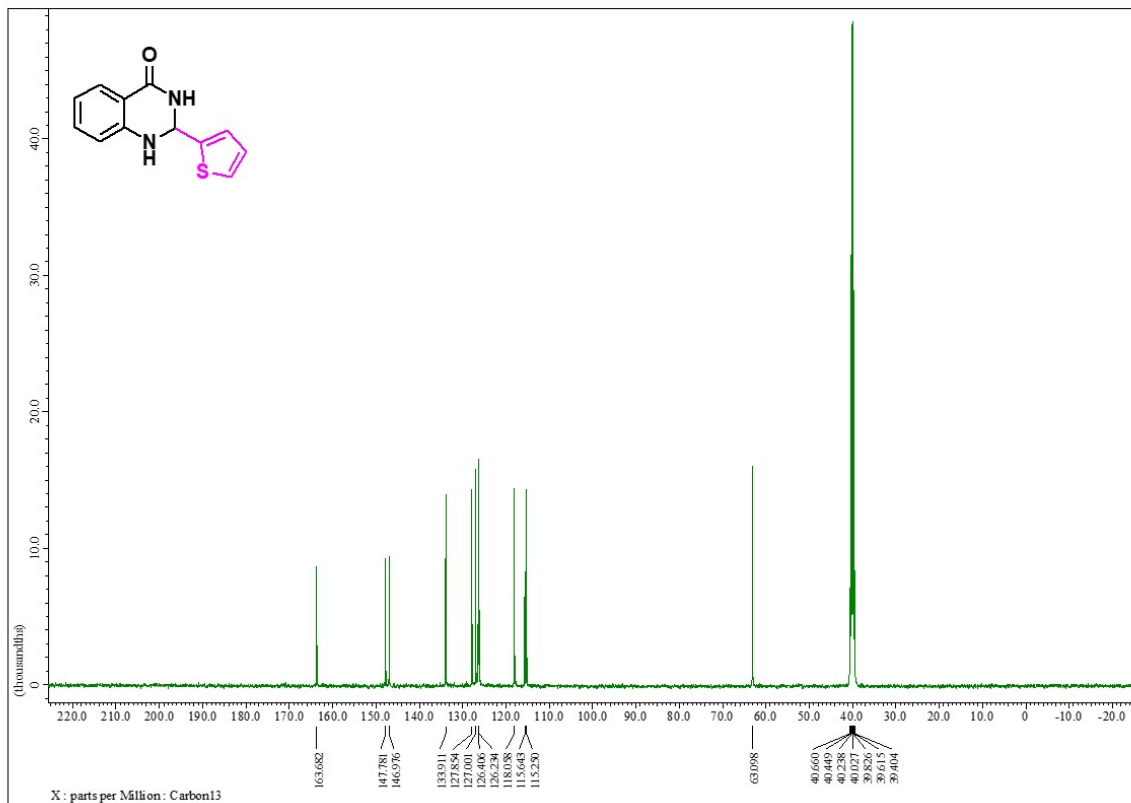
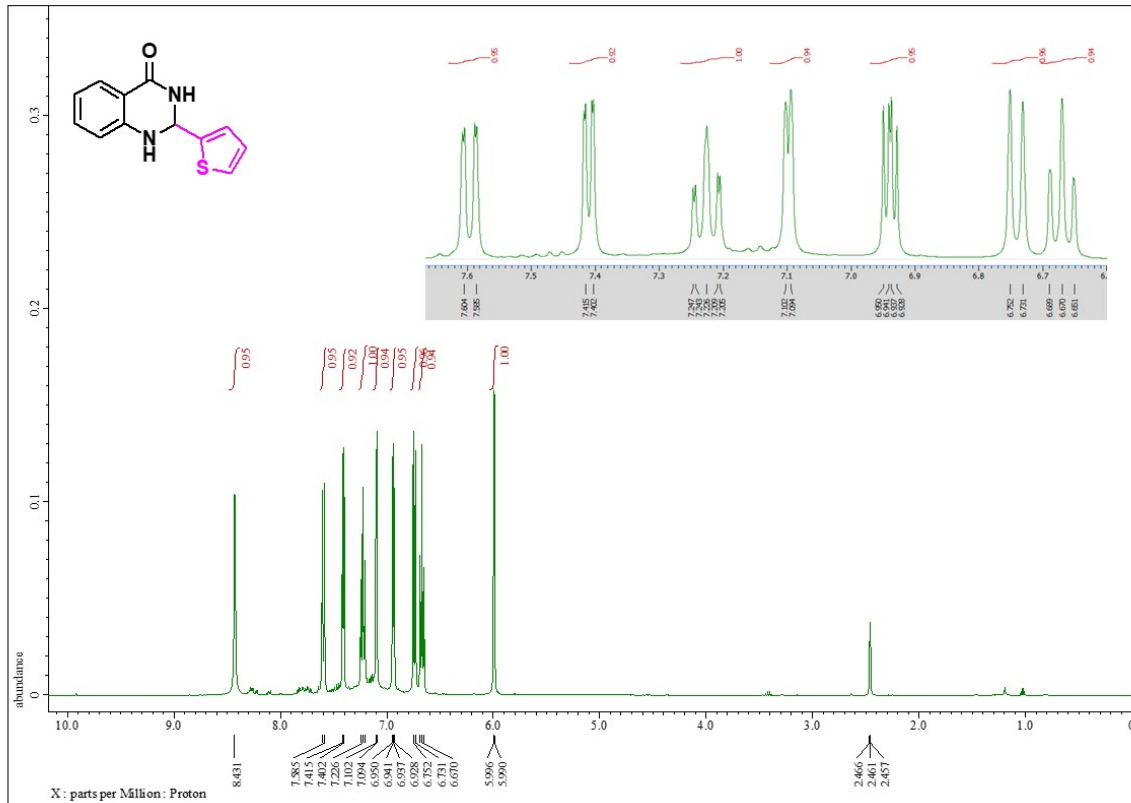


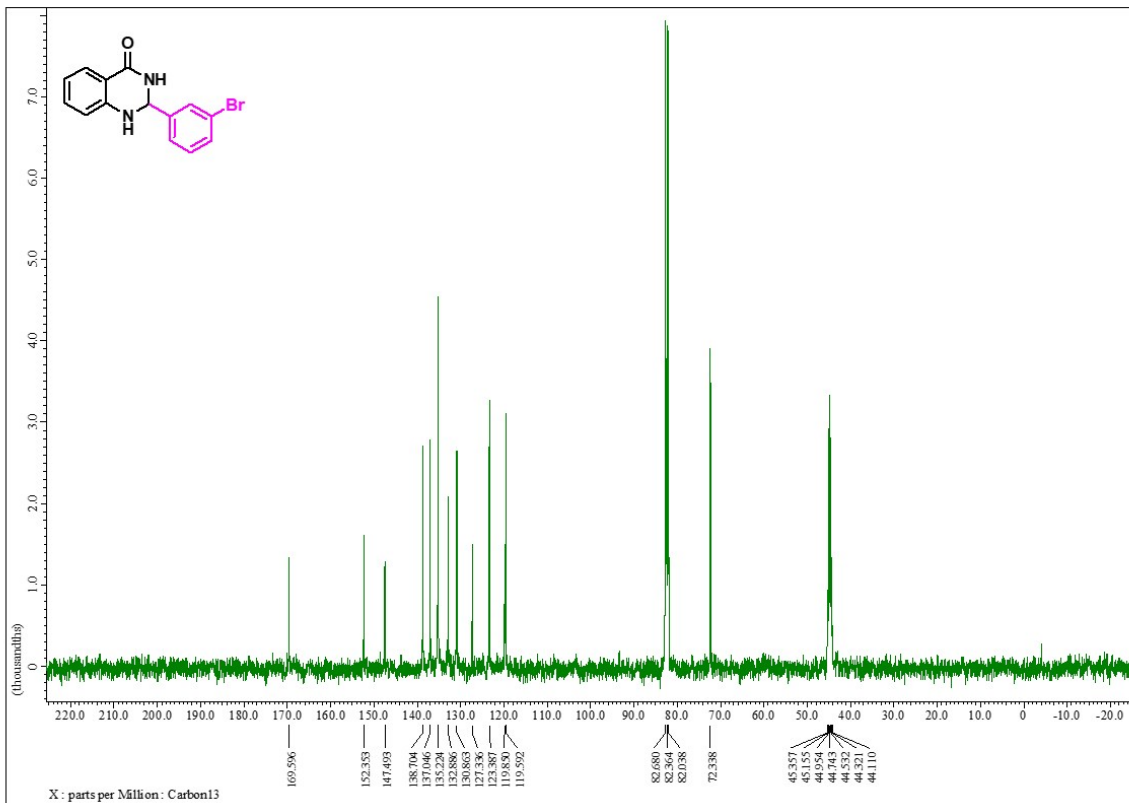
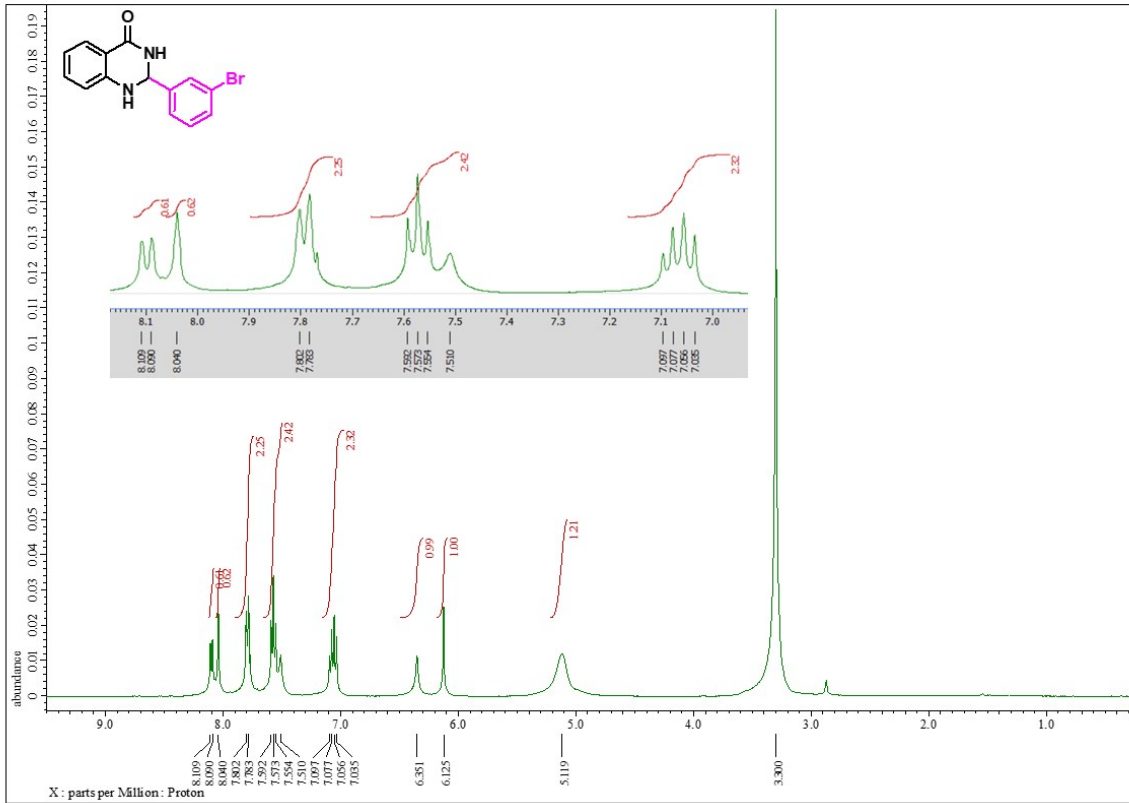


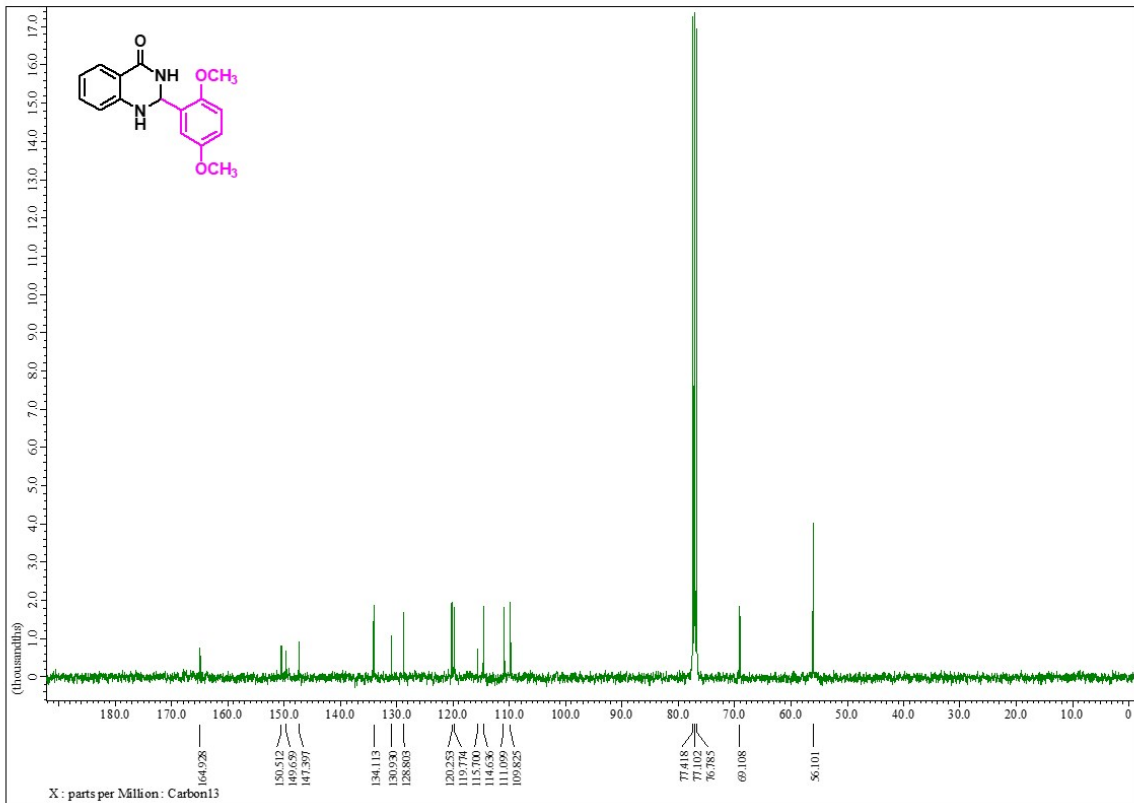
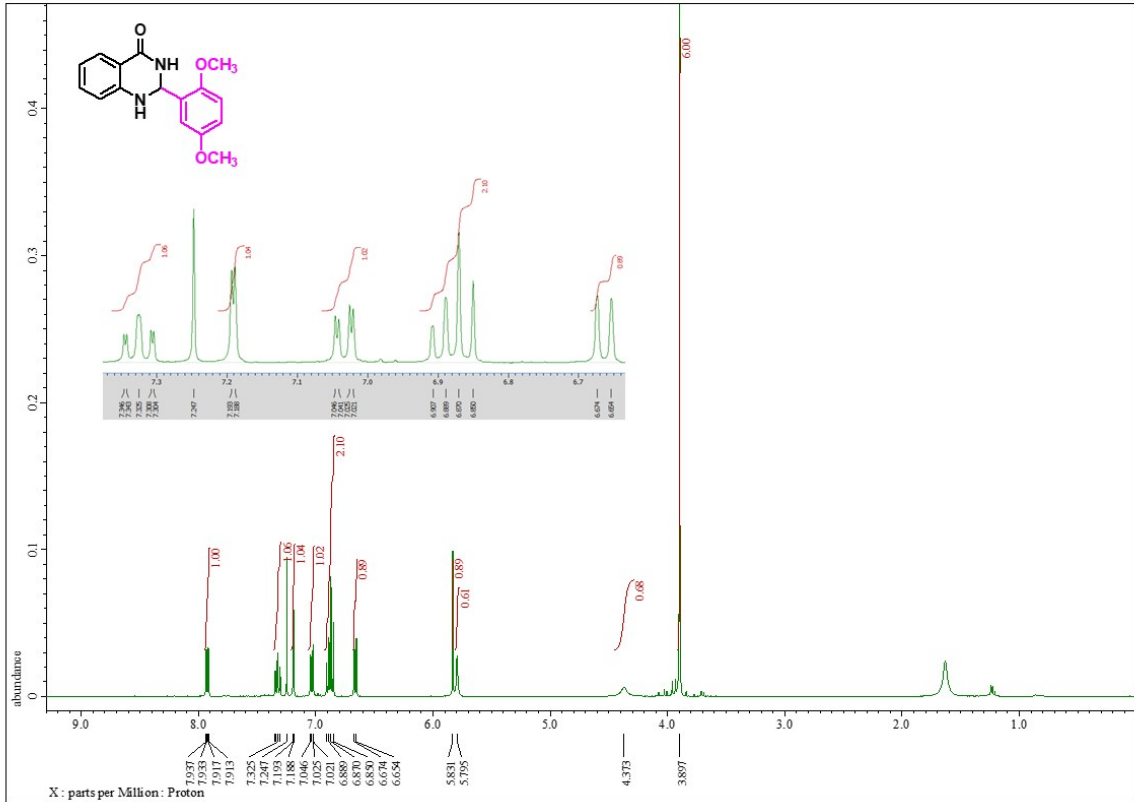


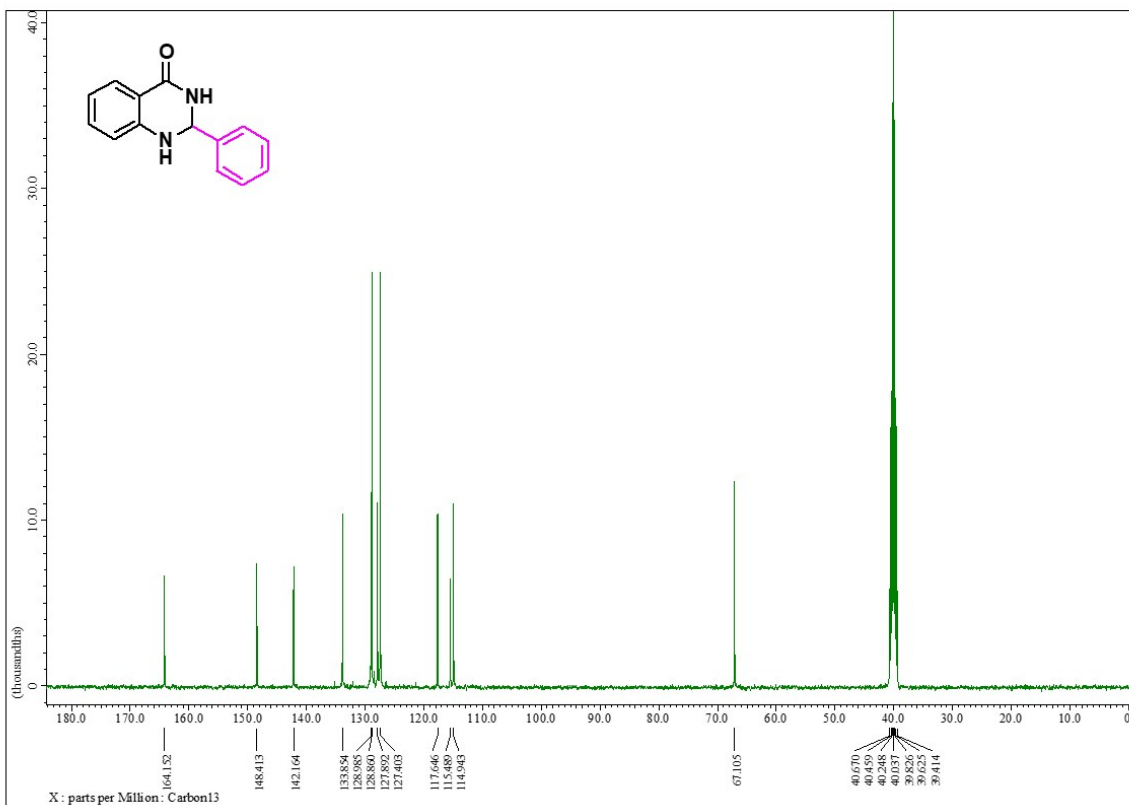
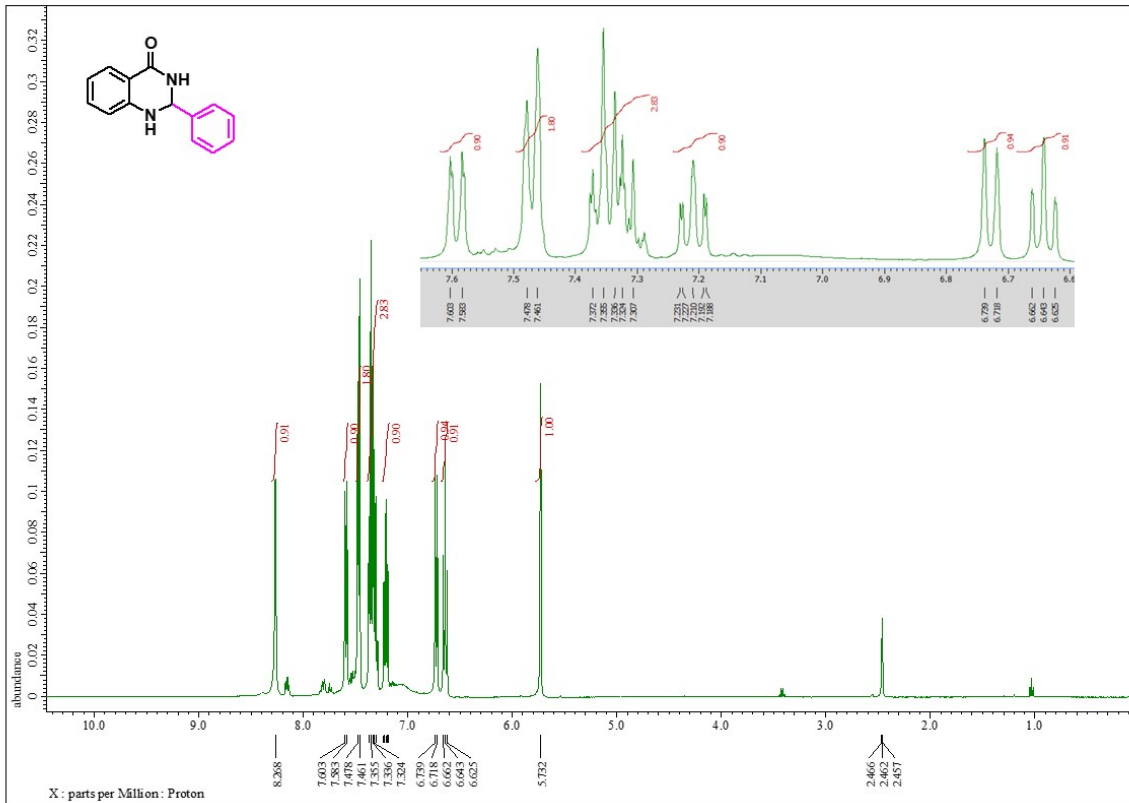


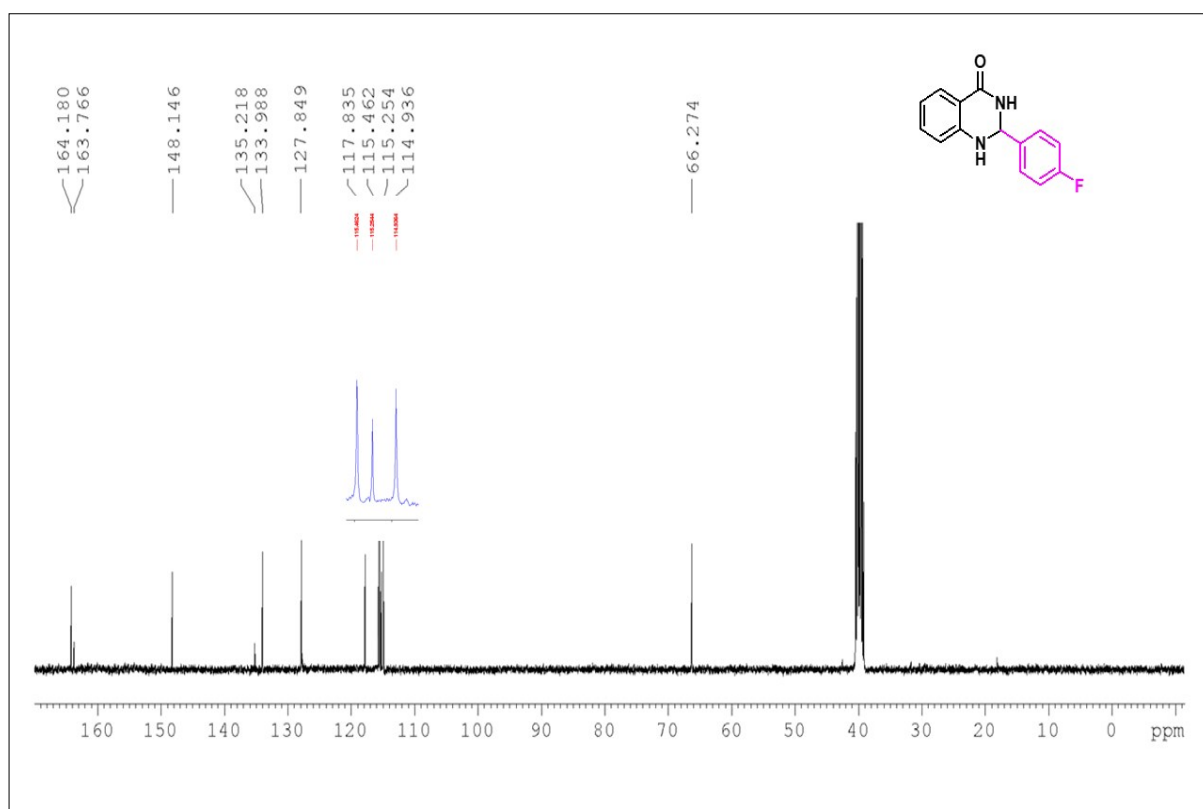
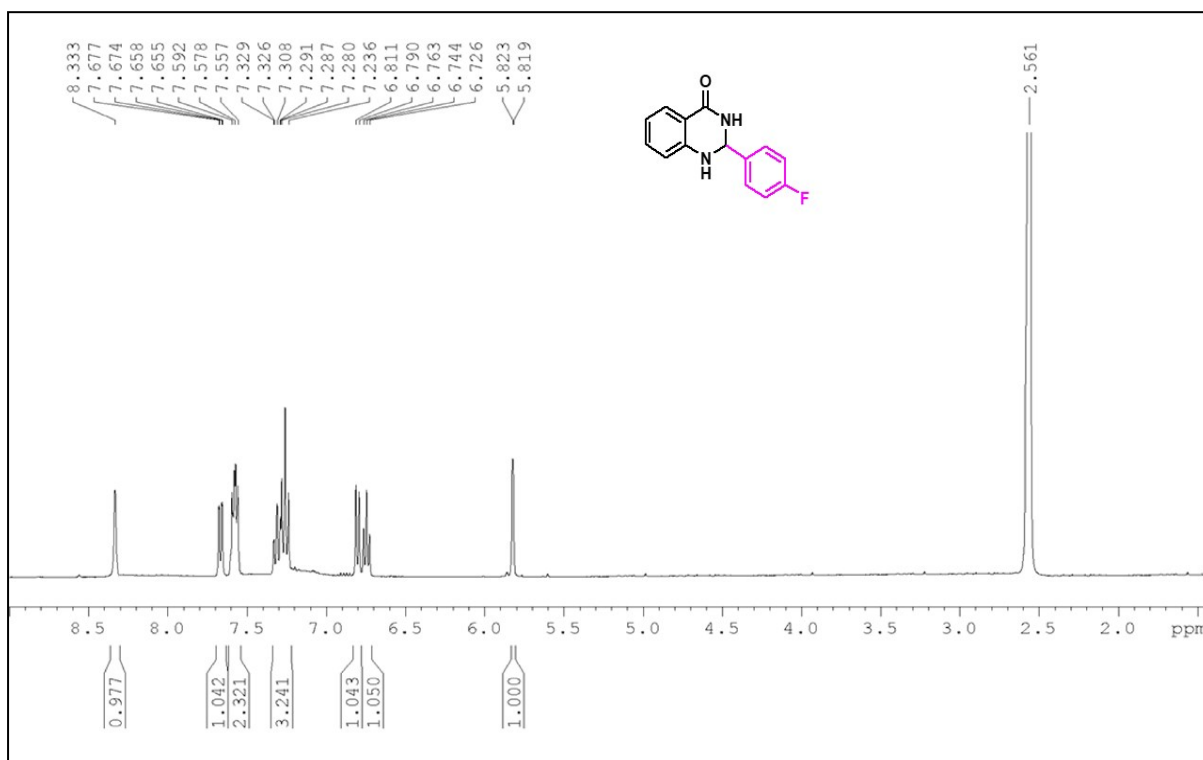


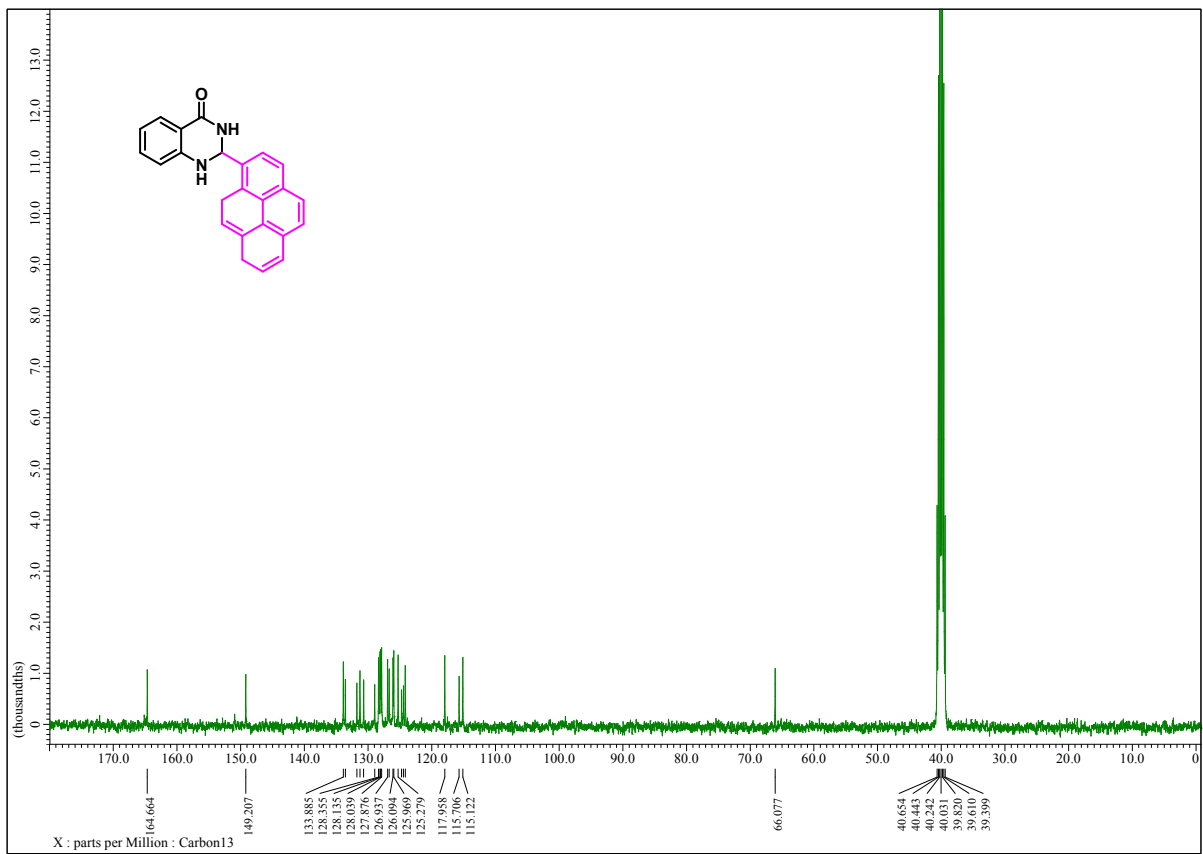
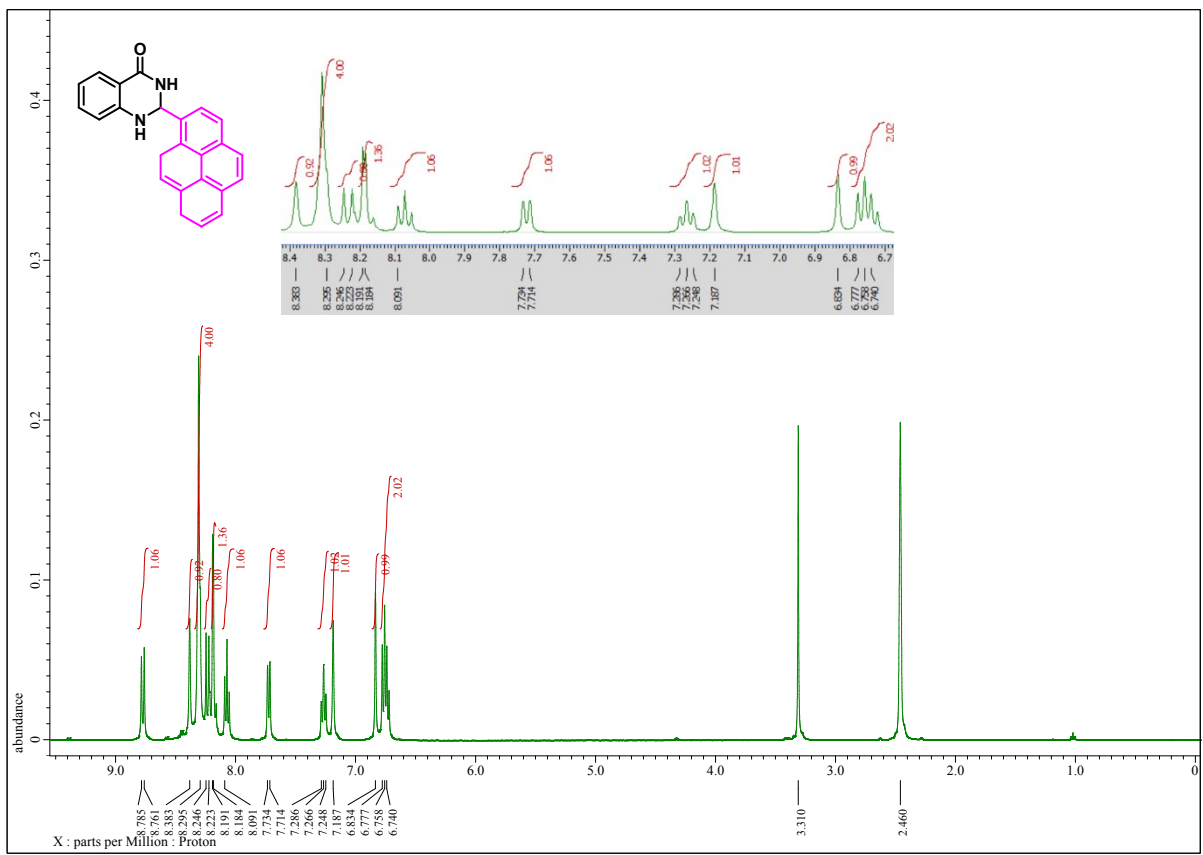


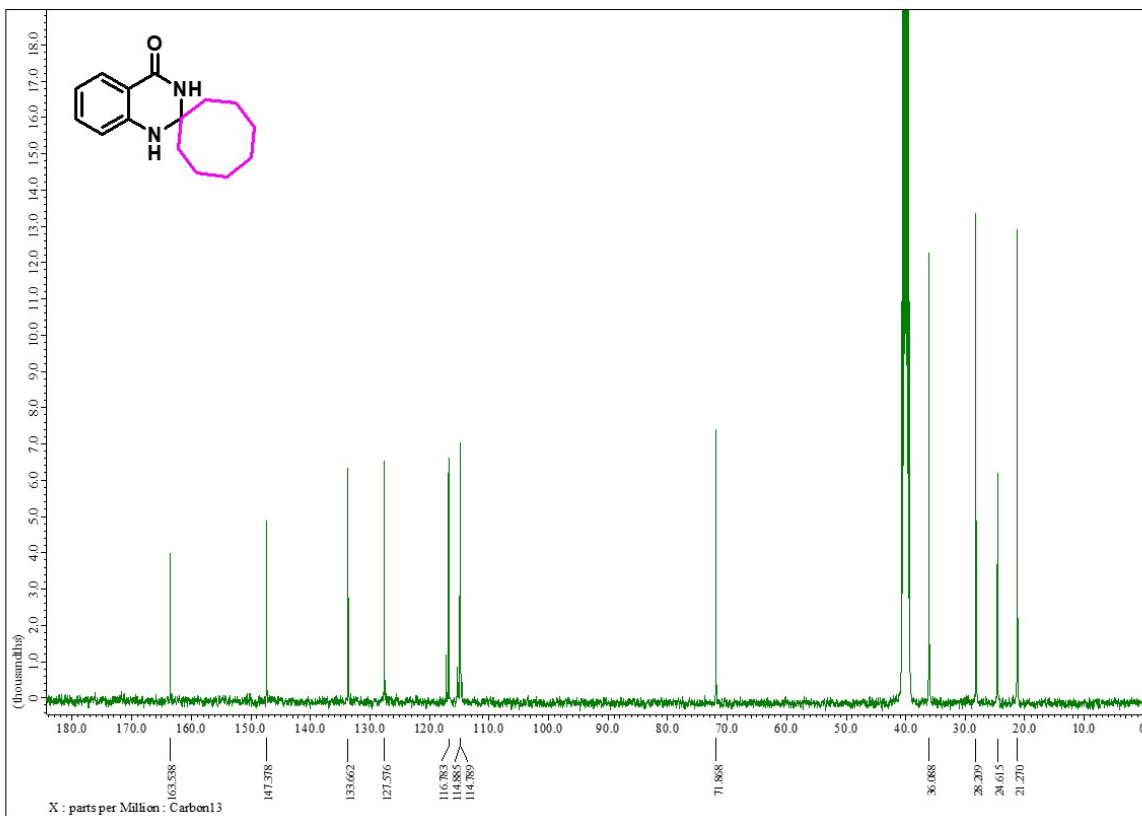
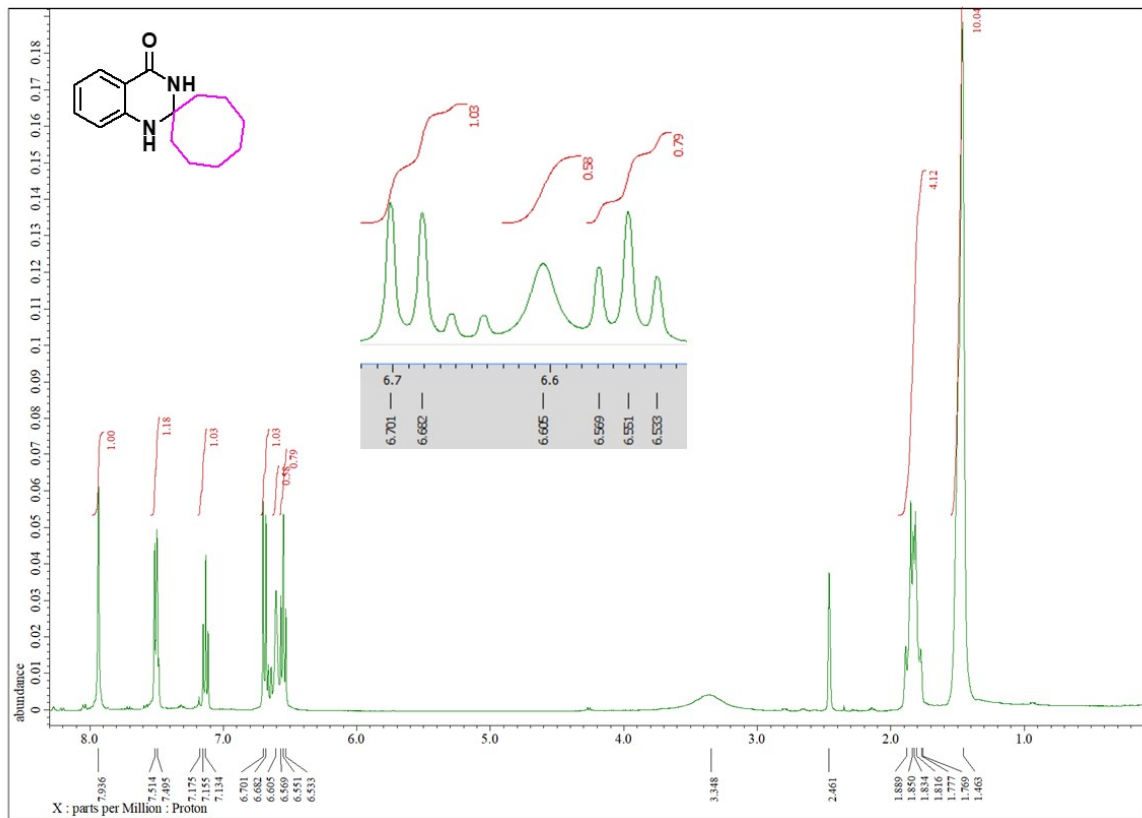


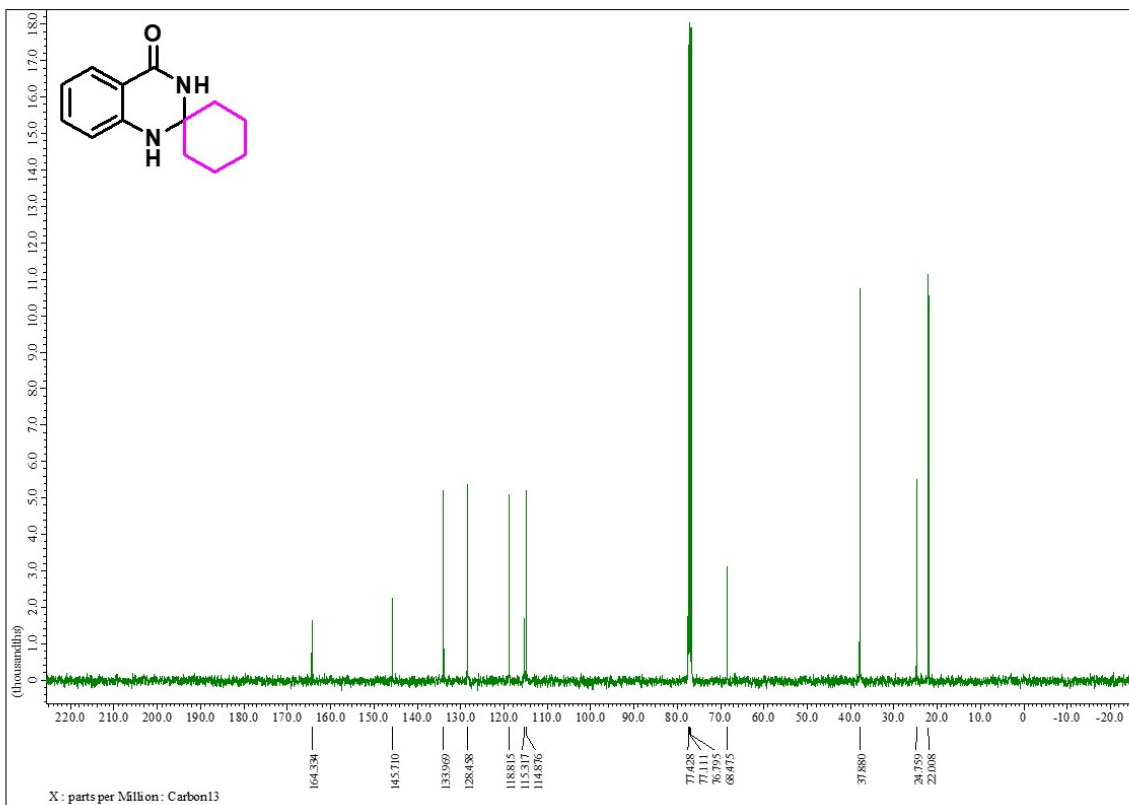
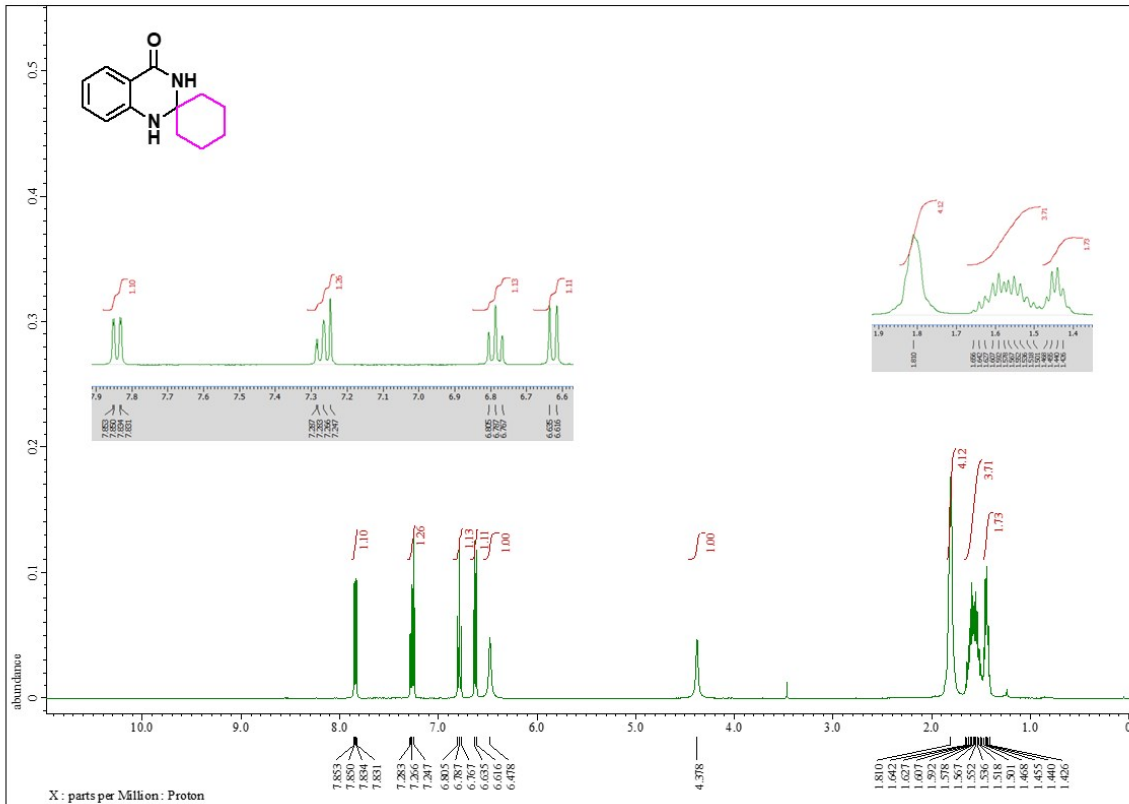


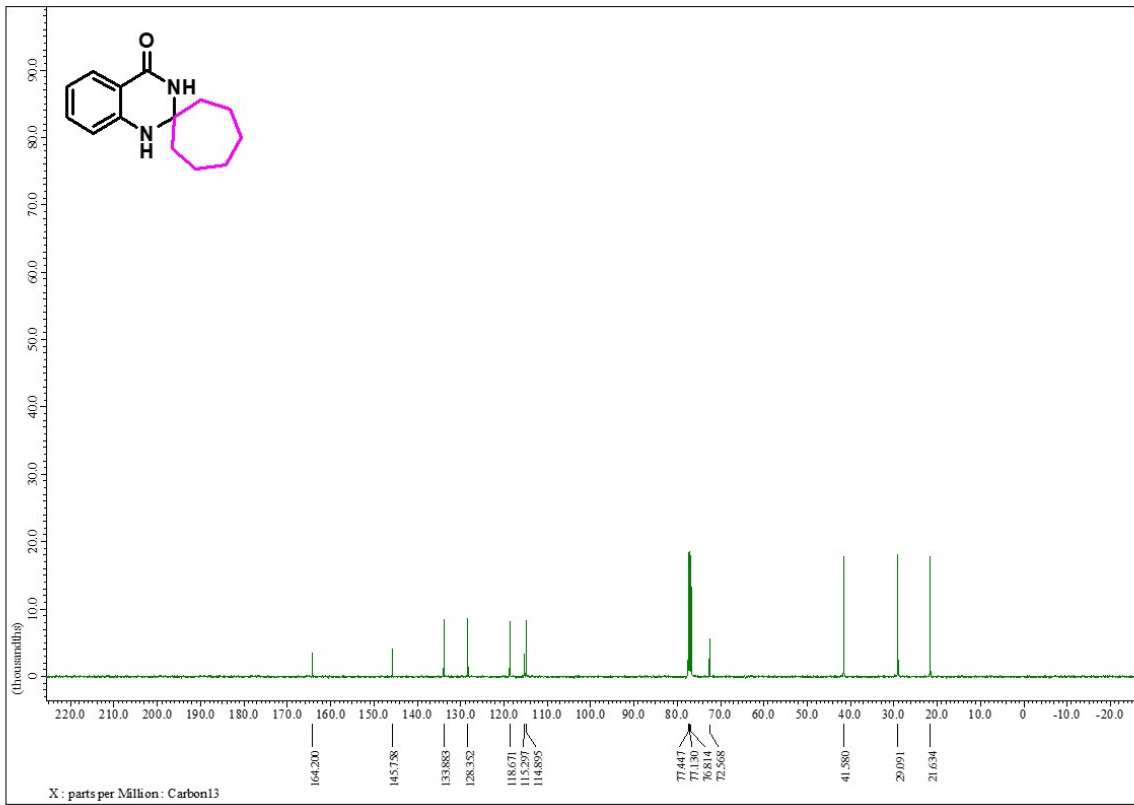
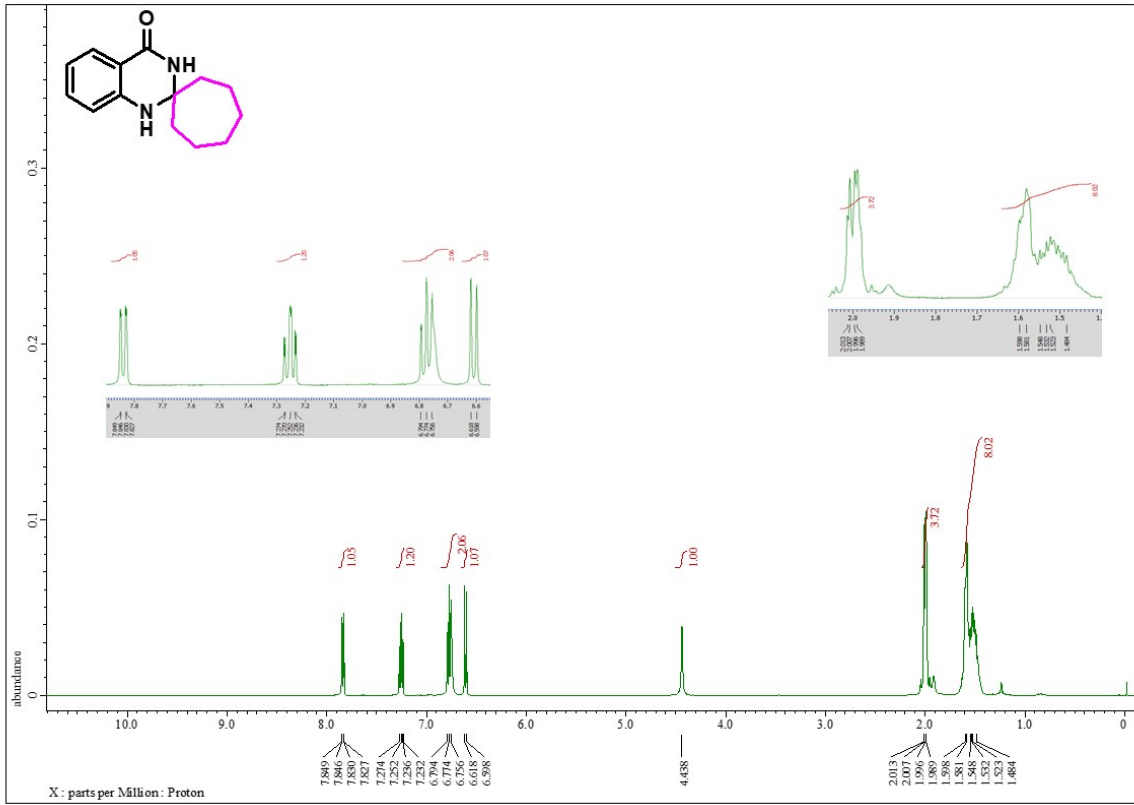


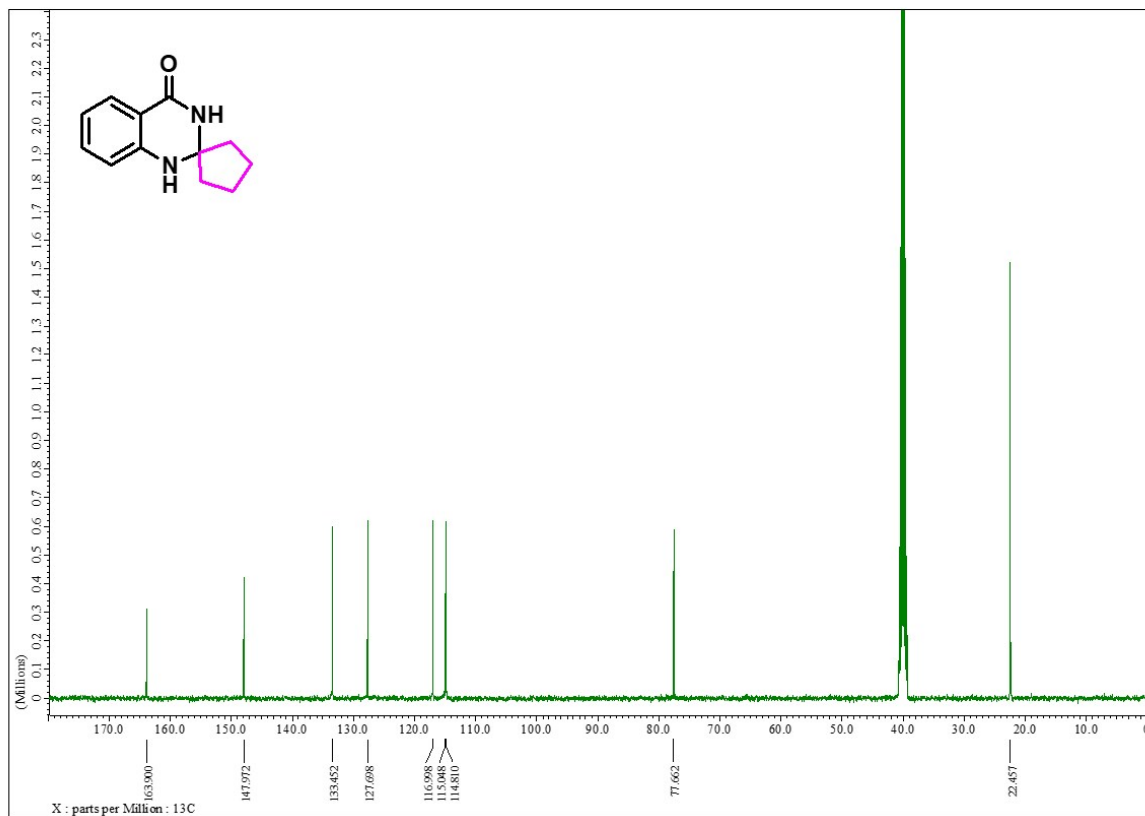
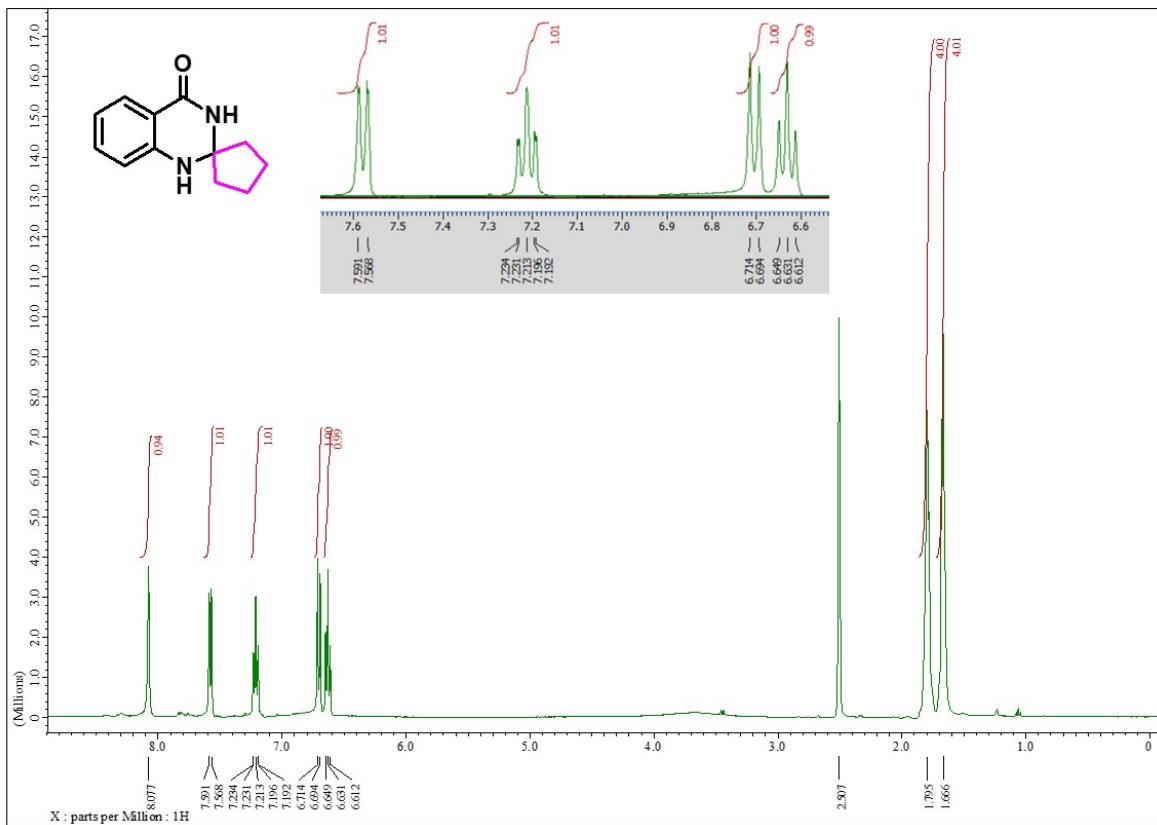


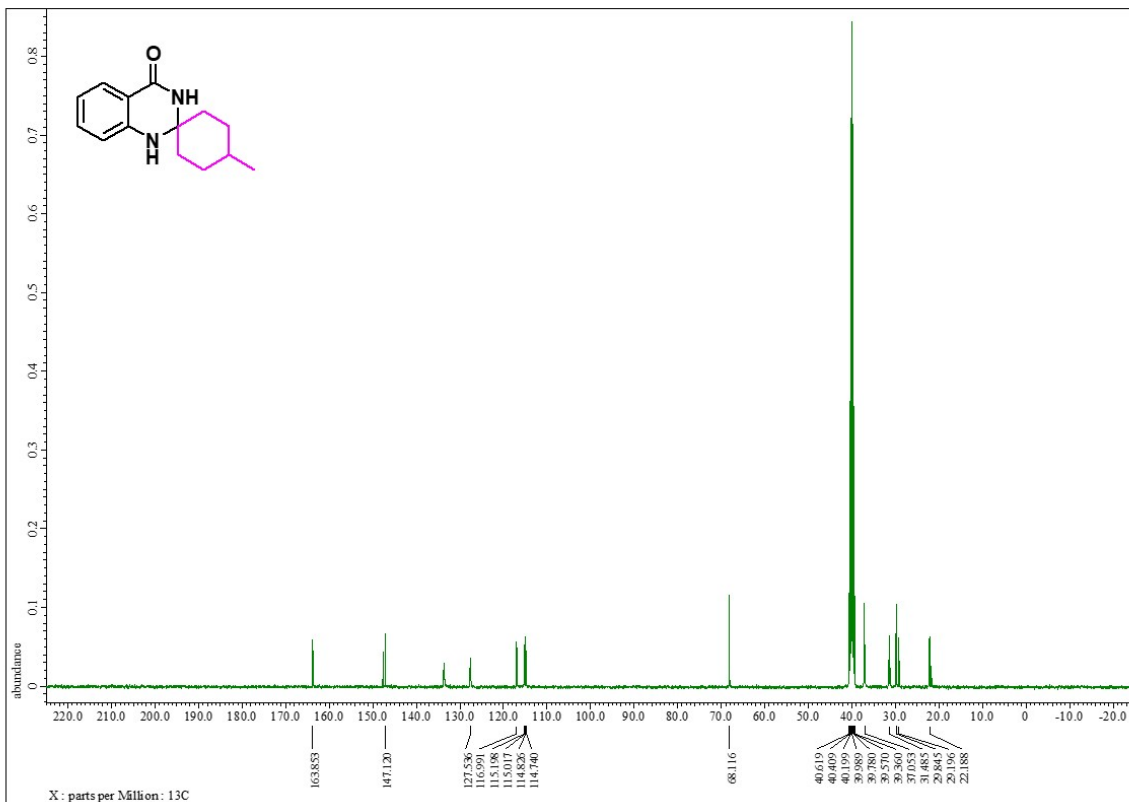
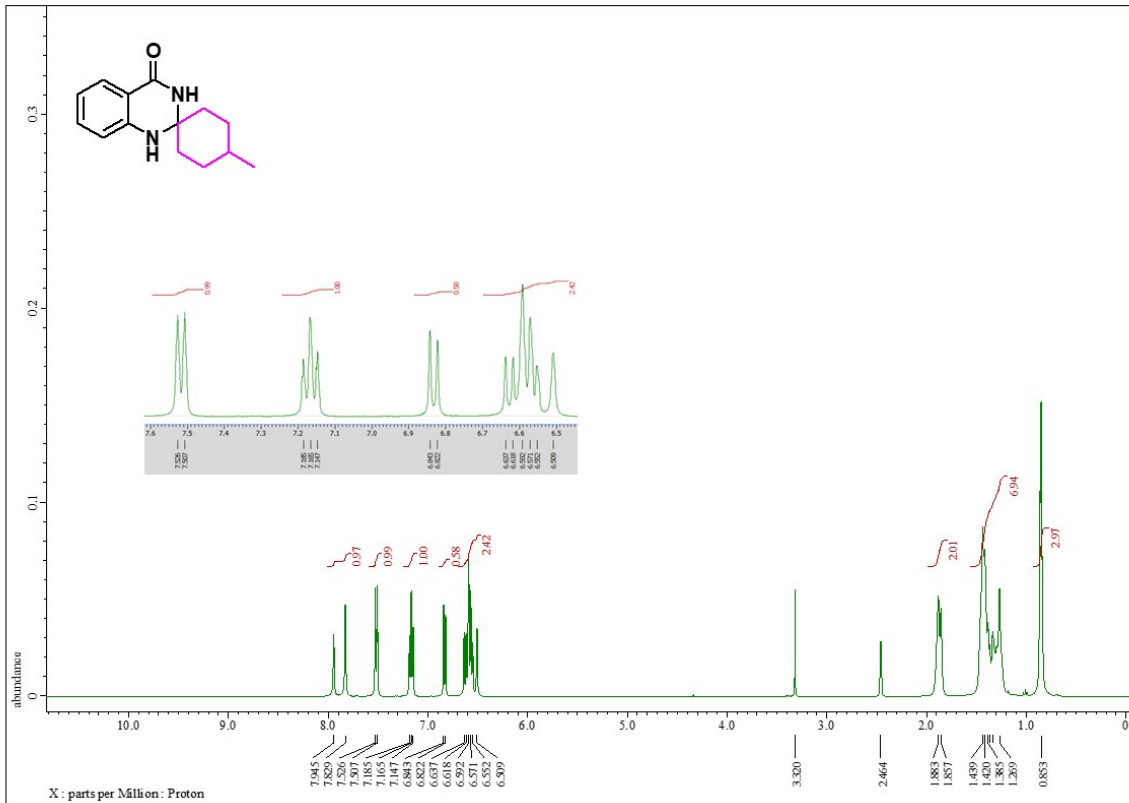


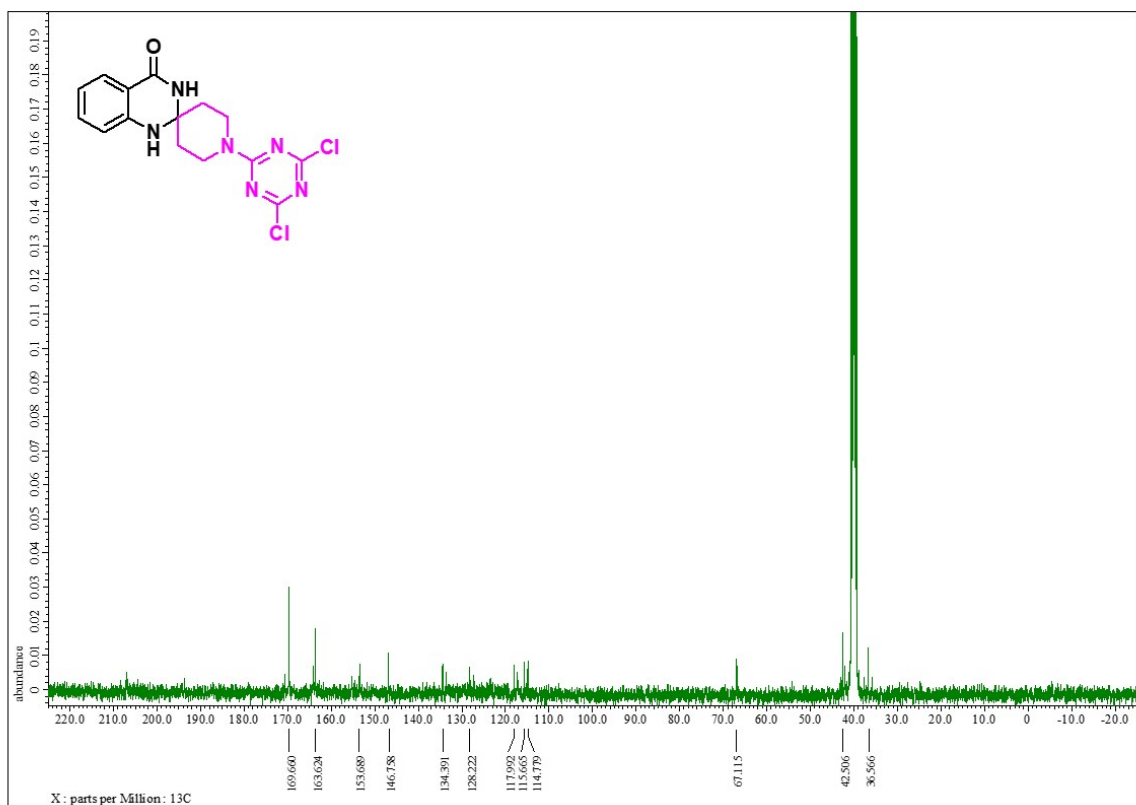
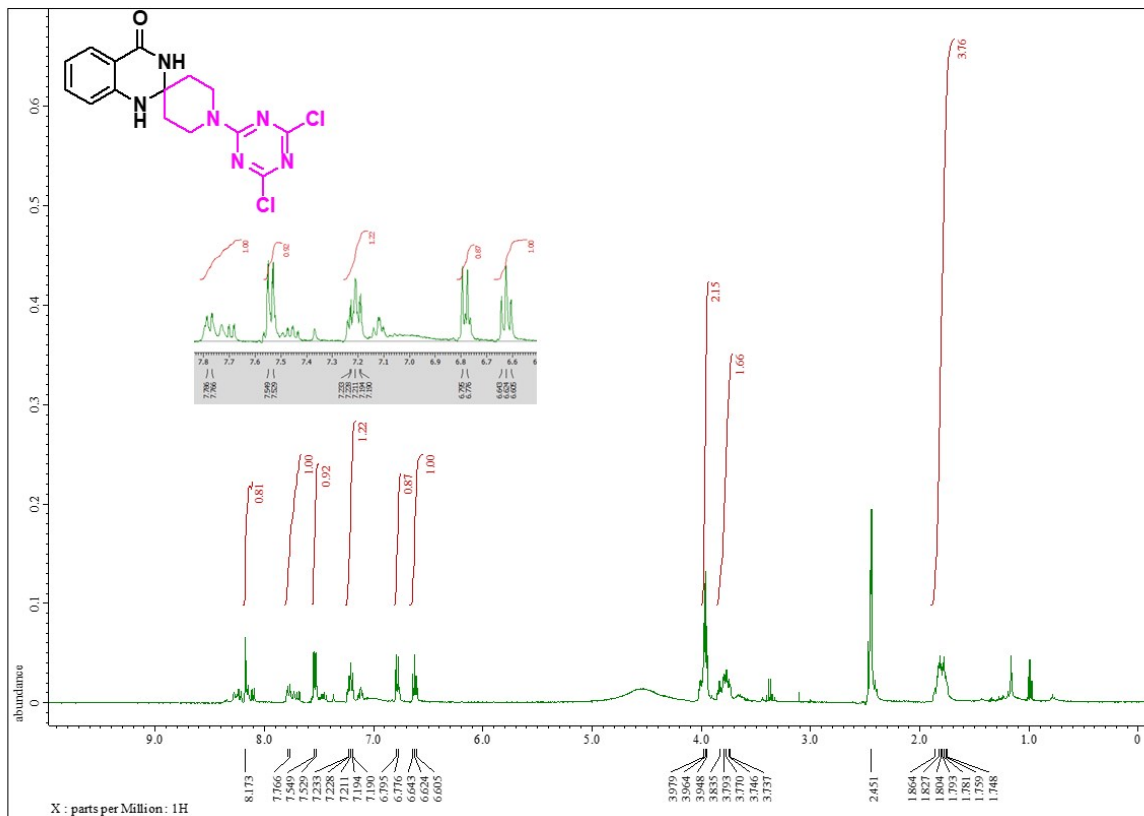


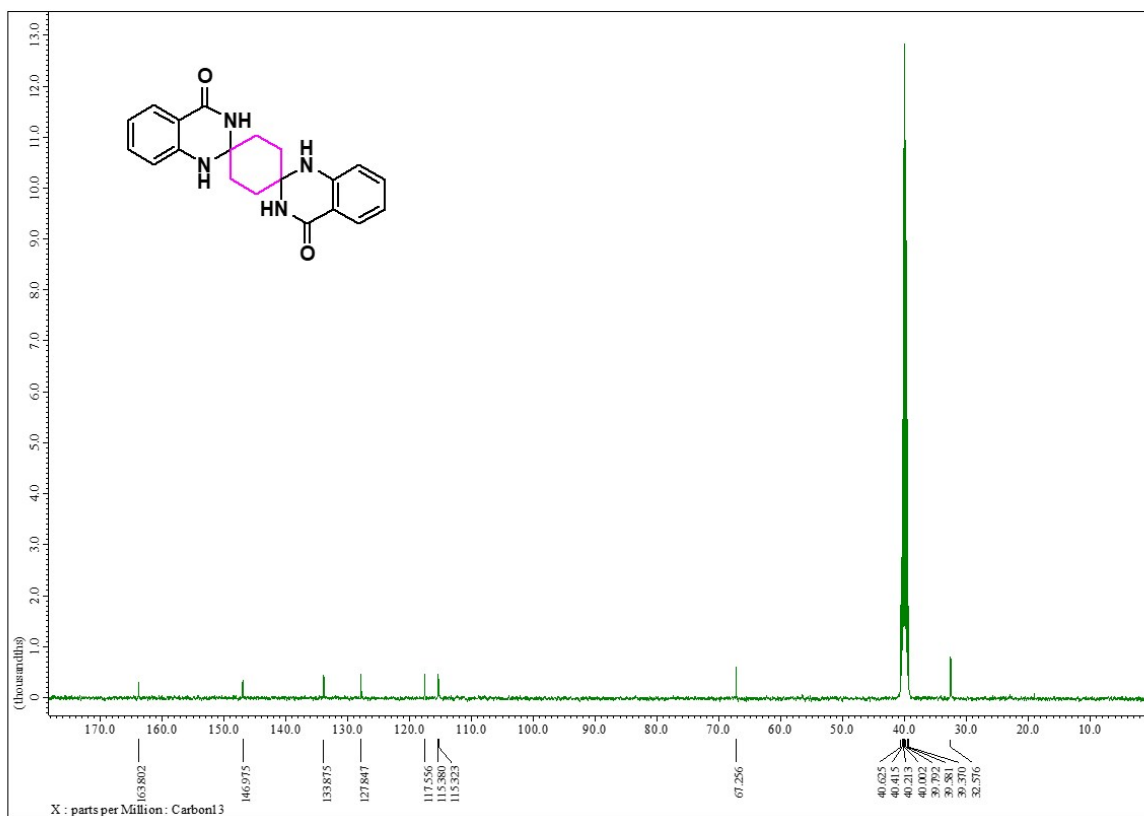
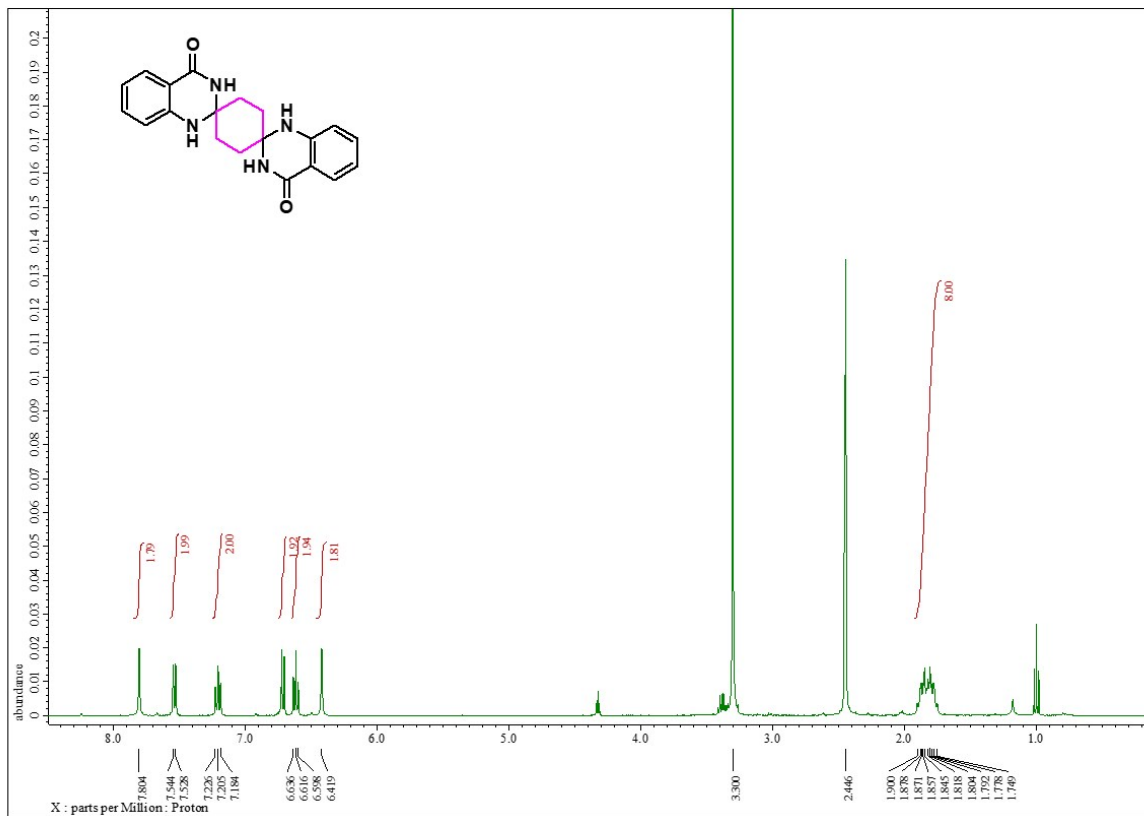


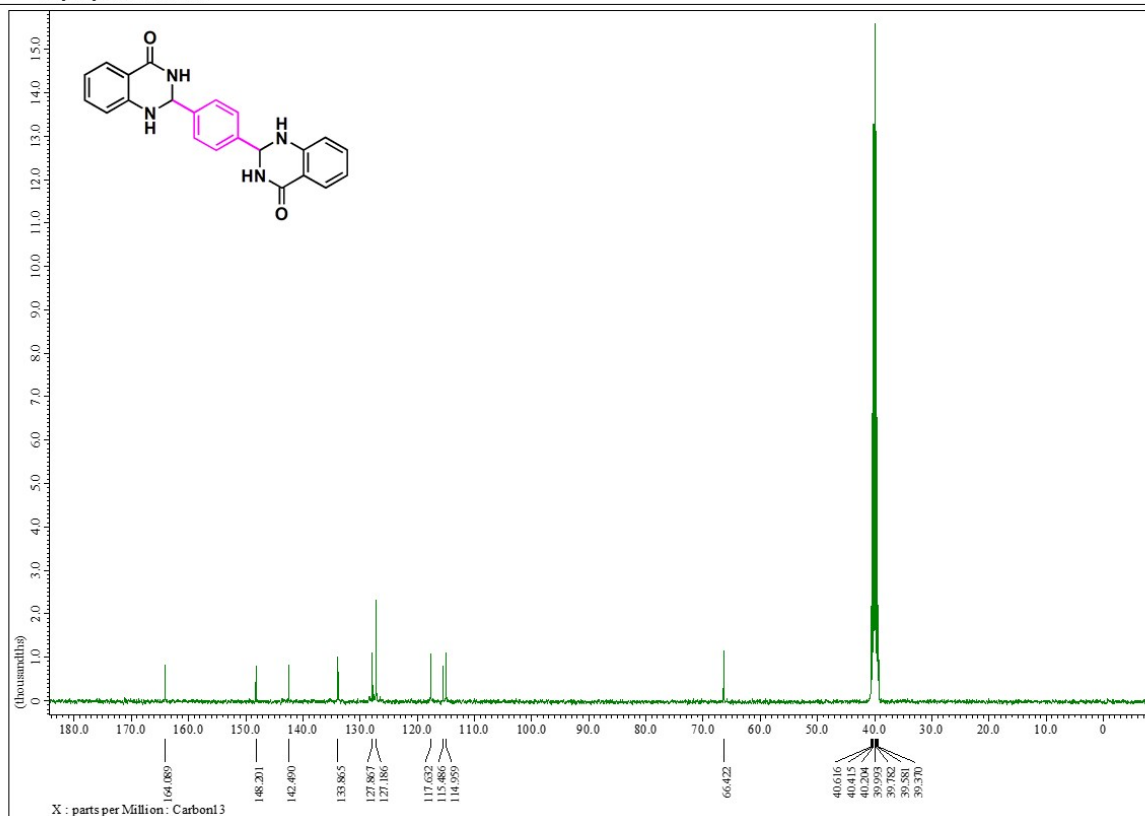
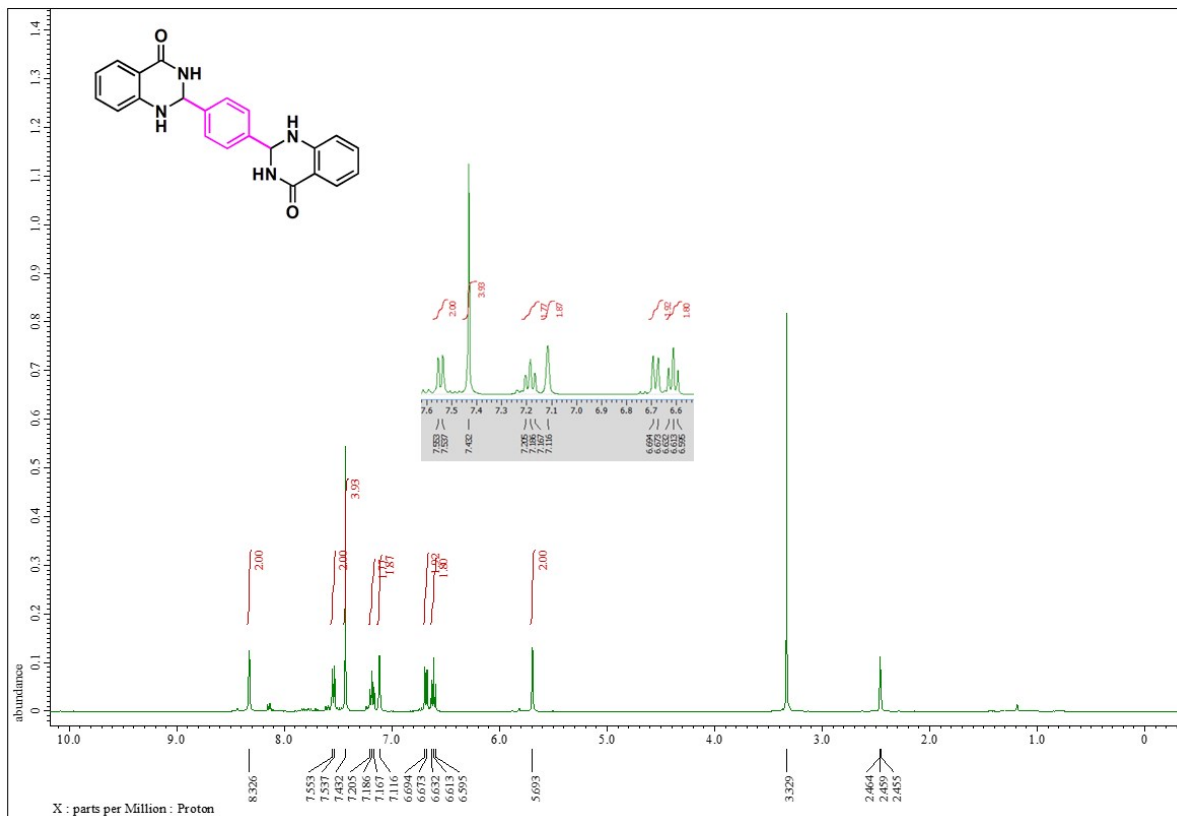








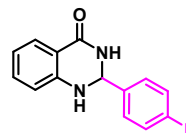




5. Mass spectra of some selected compounds

Qualitative Compound Report

Data File	QC-15.d	Sample Name	QC-15
Sample Type	Sample	Position	P1-A8
Instrument Name	Instrument 1	User Name	
Acq Method	Damo JK.m	Acquired Time	05-08-2019 14:03:09
IRM Calibration Status	Success	DA Method	Default.m
Comment			



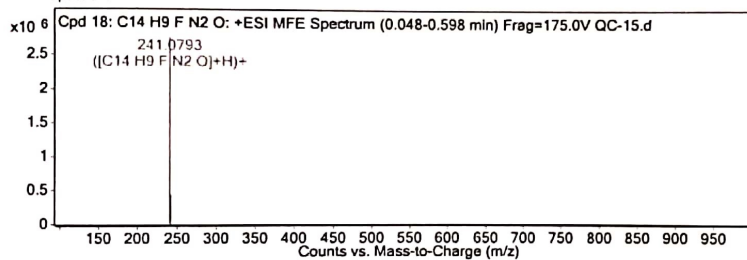
Sample Group		Info.
Acquisition SW	6200 series TOF/6500 series	
Version	Q-TOF B.05.01 (B5125.1)	

Compound Table

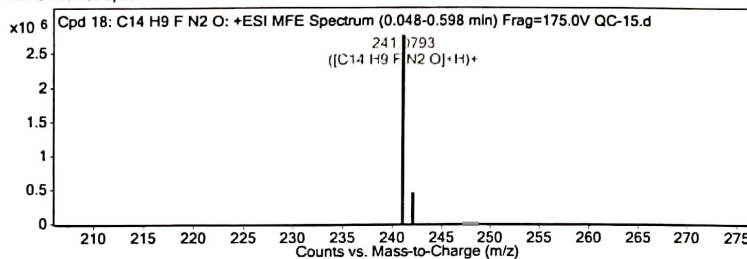
Compound Label	RT	Mass	Formula	MFG Formula	MFG Diff (ppm)	DB Formula
Cpd 18: C14 H9 F N2 O	0.109	240.072	C14 H9 F N2 O	C14 H9 F N2 O	-8.81	C14 H9 F N2 O

Compound Label	m/z	RT	Algorithm	Mass
Cpd 18: C14 H9 F N2 O	241.0793	0.109	Find by Molecular Feature	240.072

MFE MS Spectrum



MFE MS Zoomed Spectrum



MS Spectrum Peak List

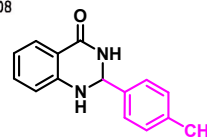
m/z	z	Abund	Formula	Ion
241.0793	1	2761729.5	C14 H9 F N2 O	(M+H)+
242.0823	1	438520.15	C14 H9 F N2 O	(M+H)+

--- End Of Report ---

Qualitative Compound Report

Data File	QC-8.d	Sample Name	QC-8
Sample Type	Sample	Position	P1-B7
Instrument Name	Instrument 1	User Name	
Acq Method	Damo JK.m	Acquired Time	29-07-2019 12:10:08
IRM Calibration Status	Success	DA Method	Default.m
Comment			

Sample Group		Info.
Acquisition SW	6200 series TOF/6500 series	
Version	Q-TOF B.05.01 (B5125.1)	

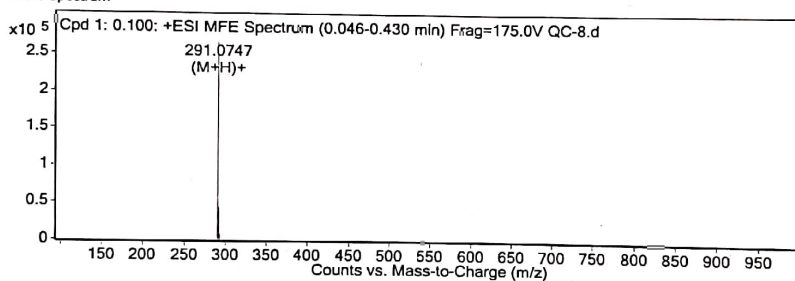


Compound Table

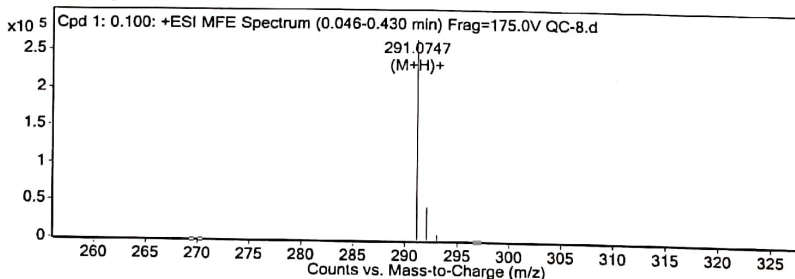
Compound Label	RT	Mass	Formula	MFG Formula	DB Formula
Cpd 1: 0.100	0.1	290.0676	C13 H6 N8 O	C13 H6 N8 O	C13 H6 N8 O

Compound Label	m/z	RT	Algorithm	Mass
Cpd 1: 0.100	291.0747	0.1	Find by Molecular Feature	290.0676

MFE MS Spectrum



MFE MS Zoomed Spectrum



MS Spectrum Peak List

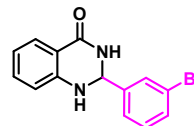
m/z	z	Abund	Ion
291.0747	1	264994.72	(M+H)+
292.0775	1	44133.07	(M+H)+
293.0835	1	8402.72	(M+H)+

--- End Of Report ---

Qualitative Compound Report

Data File	QC-16.d	Sample Name	QC-16
Sample Type	Sample	Position	P1-D6
Instrument Name	Instrument 1	User Name	
Acq Method	Damo JK.m	Acquired Time	17-07-2019 17:15:22
IRM Calibration Status	Success	DA Method	Default.m
Comment			

Sample Group Info.
 Acquisition SW 6200 series TOF/6500 series
 Version Q-TOF B.05.01 (B5125.1)

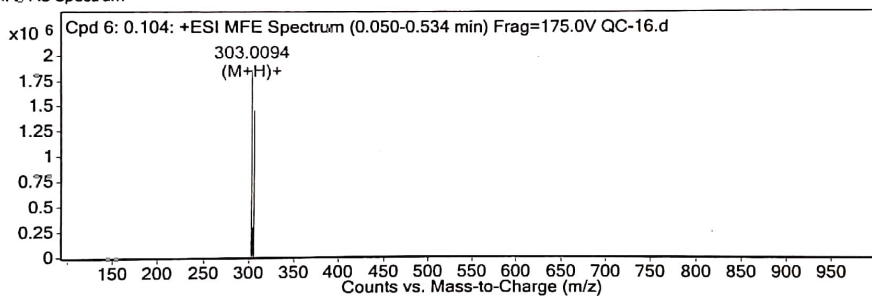


Compound Table

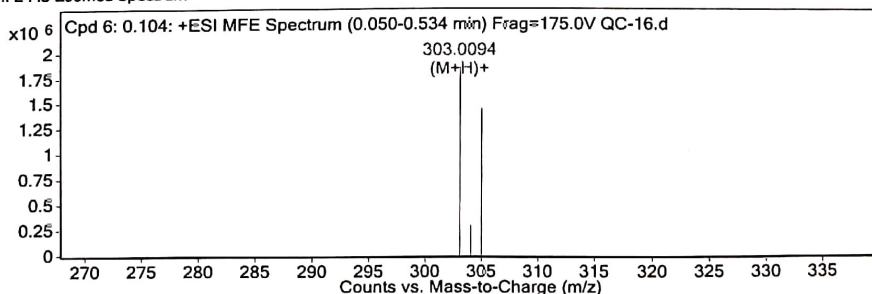
Compound Label	RT	Mass	MFG Formula
Cpd 6: 0.104	0.104	302.0021	<none>

Compound Label	m/z	RT	Algorithm	Mass
Cpd 6: 0.104	303.0094	0.104	Find by Molecular Feature	302.0021

MFE MS Spectrum



MFE MS Zoomed Spectrum



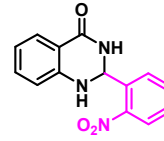
MS Spectrum Peak List

m/z	z	Abund	Ion
303.0094	1	1858237.38	(M+H)+
304.0124	1	289825.91	(M+H)+
305.0109	1	1453844.52	(M+H)+

--- End Of Report ---

Qualitative Compound Report

Data File	QC-3.d	Sample Name	QC-3
Sample Type	Sample	Position	P1-D1
Instrument Name	Instrument 1	User Name	
Acq Method	Damo JK.m	Acquired Time	14-06-2019 13:28:58
IRM Calibration Status	Success	DA Method	Default.m
Comment			



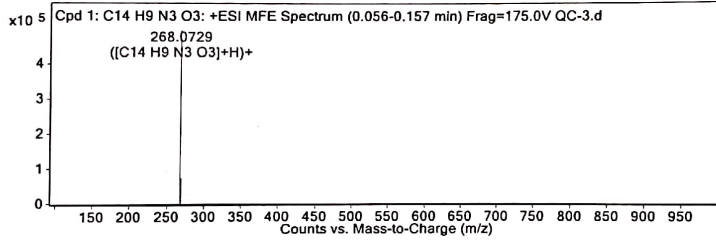
Sample Group		Info.
Acquisition SW	6200 series TOF/6500 series	
Version	Q-TOF B.05.01 (B5125.1)	

Compound Table

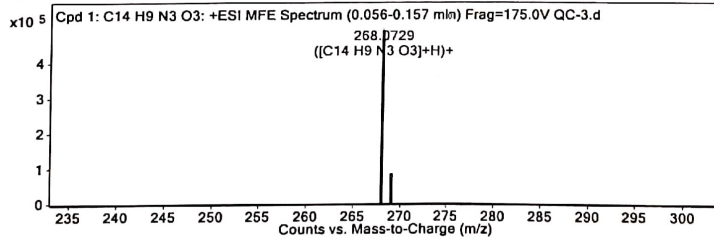
Compound Label	RT	Mass	Formula	MFG Formula	MFG Diff (ppm)	DB Formula
Cpd 1: C14 H9 N3 O3	0.091	267.0656	C14 H9 N3 O3	C14 H9 N3 O3	-4.4	C14 H9 N3 O3

Compound Label	m/z	RT	Algorithm	Mass
Cpd 1: C14 H9 N3 O3	268.0729	0.091	Find by Molecular Feature	267.0656

MFE MS Spectrum



MFE MS Zoomed Spectrum



MS Spectrum Peak List

m/z	z	Abund	Formula	Ion
268.0729	1	497907.19	C14 H9 N3 O3	(M+H)+
269.0757	1	74185.73	C14 H9 N3 O3	(M+H)+

— End Of Report —

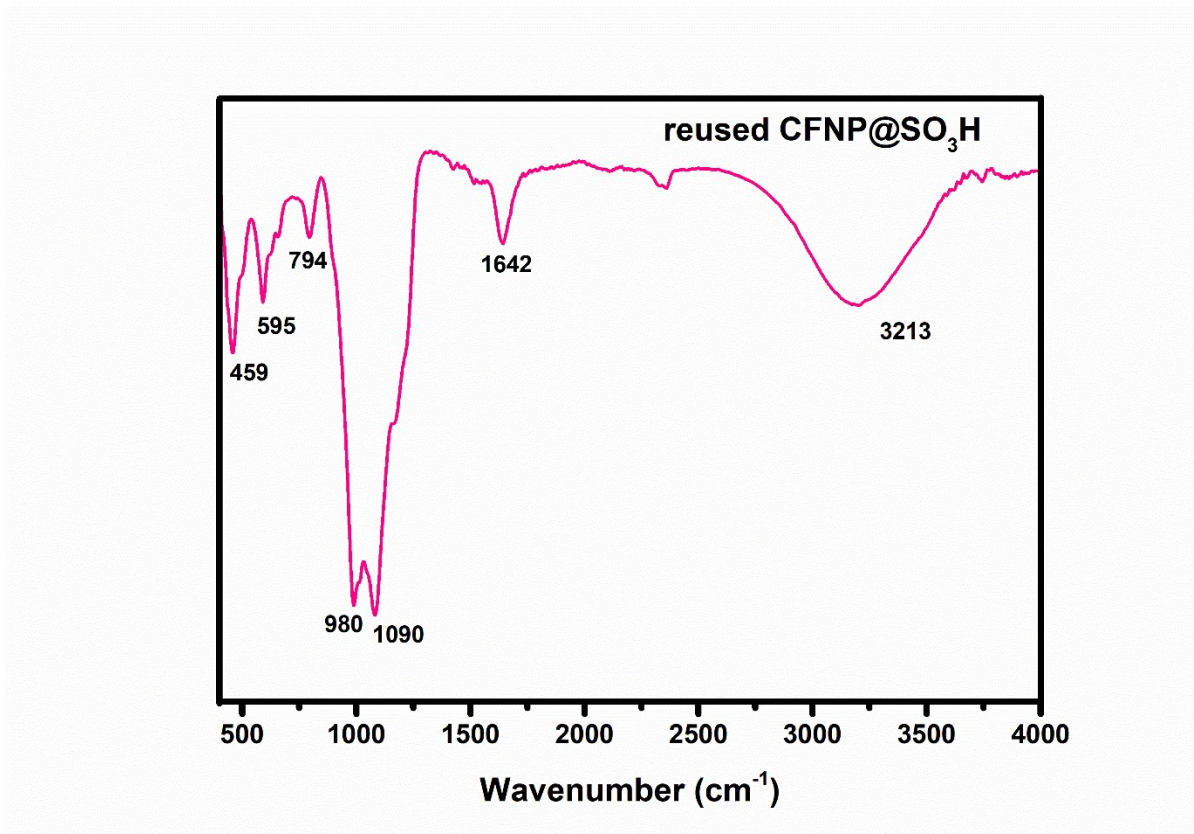


Figure S1: FT-IR spectra of reused catalyst (reused CFNP@SO₃H) after 6 serial runs.

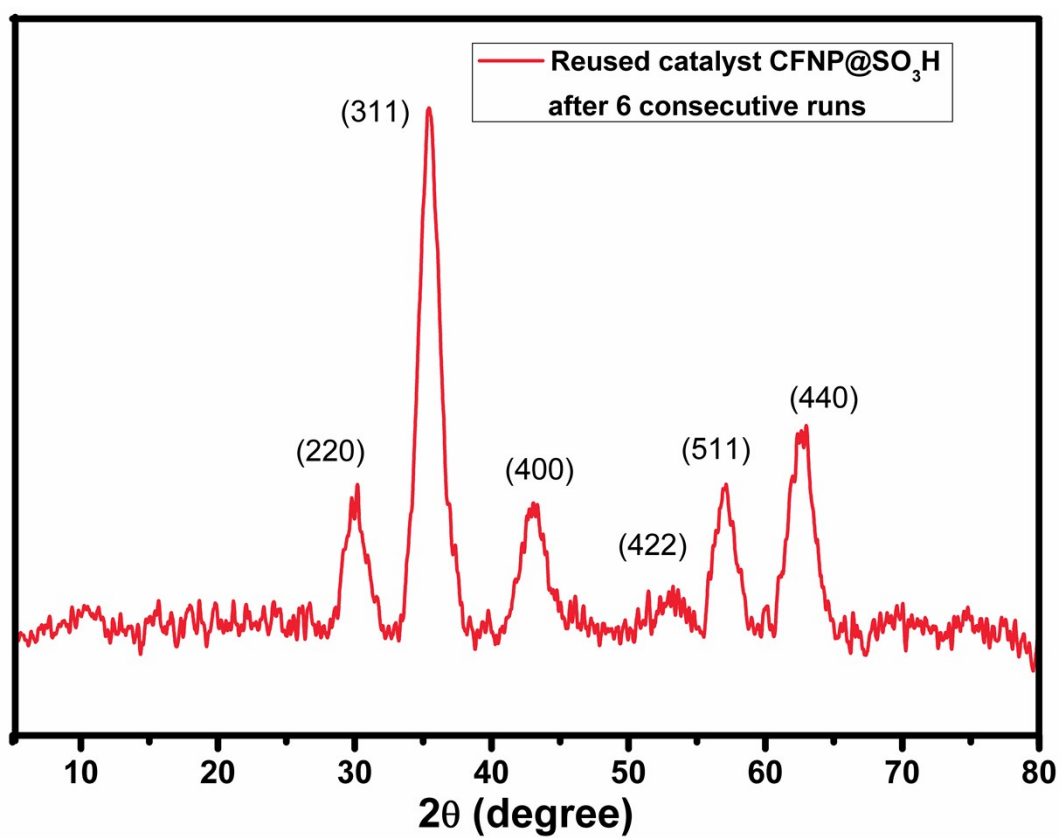


Figure S2: Powder XRD analysis of reused catalyst (reused CFNP@SO₃H) after 6 serial runs.

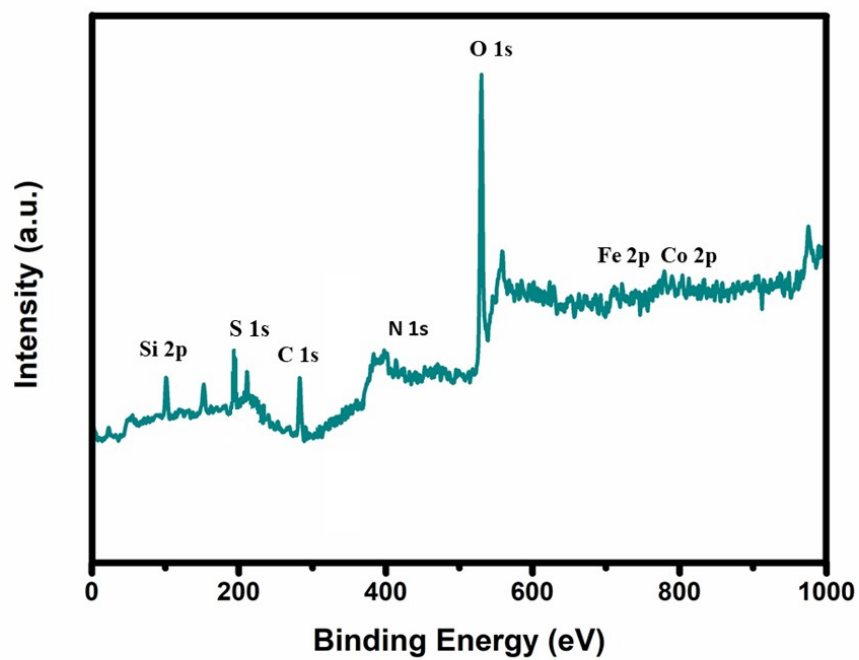


Figure S3: XPS survey spectra of catalyst (CFNP@SO₃H).

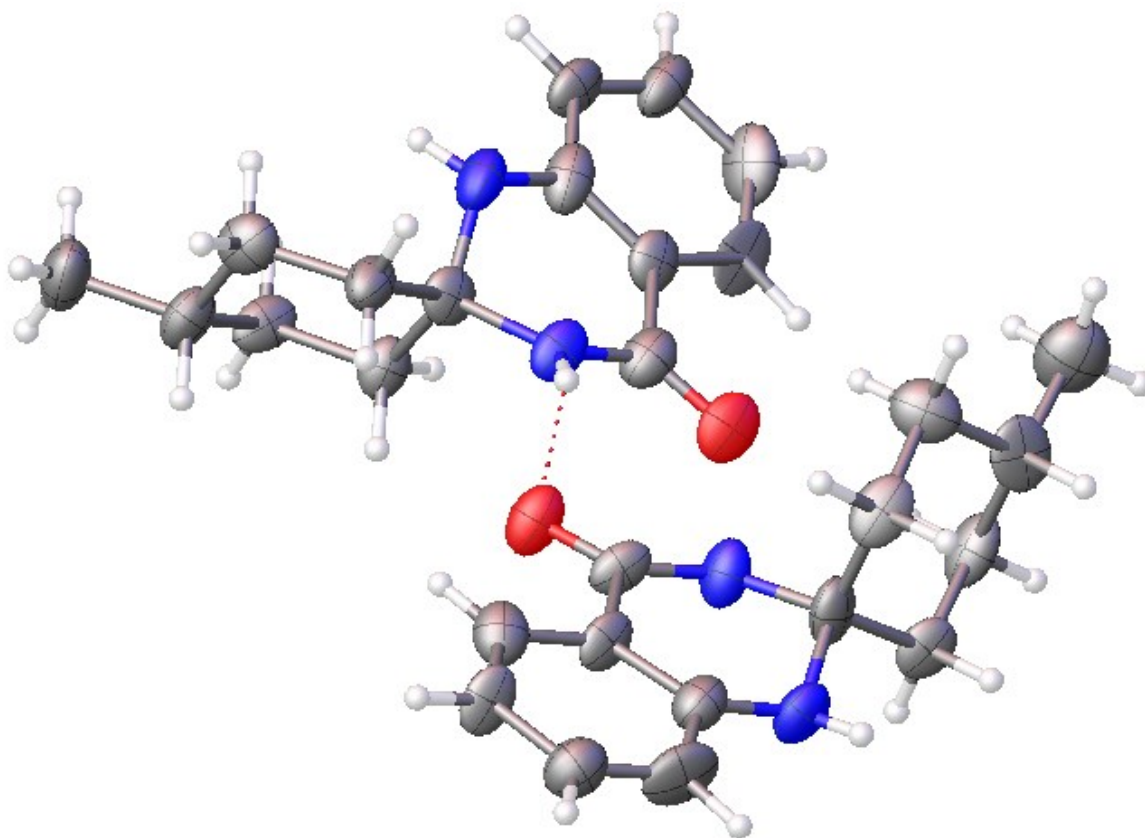


Figure S4: Crystal structure of compound 5e showing H-bonding.

Calculation of the Turn over frequency (TOF) of catalyst

1000 mg (1 g) catalyst (CFNP@SO₃H) contains 2.48 mmol free sulfonic acid.

To determine the turn over number of the catalyst we had tested the synthesis of quinazolinone (3l) by taking 1 mmol of anthranilamide and 1 mmol of benzaldehyde in presence of 18 mg of catalyst at room temperature.

As the yield of product formation is 98%,

The turn over number (TON) of 3l is = $\frac{\text{mmol of product}}{\text{mmol of acid site present in catalyst}}$

1 g catalyst holds 2.48 mmol of acid site, so 18 mg of catalyst holds = 0.04464 mmol acid sites.

The turn over number is = $\frac{0.98}{0.04464} = 21.95$ (as the yield of product is 98%, mmol of product = 0.98)

And turn over frequency (TOF) of 3l is = $\frac{\text{Turn over number}}{\text{reaction time}} = \frac{21.95}{4/60} \text{ h}^{-1} = 329.25 \text{ h}^{-1}$

Table S4: Calculation of TOF (h⁻¹) for the synthesized products.

Product	Yield (%)	TOF (h ⁻¹)
3a	97	325.94
3b	96	322.58
3c	92	247.31
3d	94	252.69
3e	94	252.69
3f	96	322.58
3g	90	201.61
3h	88	197.13
3i	90	403.23
3j	93	312.50
3k	93	249.99
3l	98	329.25
3m	90	201.61
3n	96	258.06

5a	88	197.13
5b	97	315.86
5c	90	302.42
5d	95	255.37
5e	97	325.94
5f	88	295.70