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Supplementary Information

Pt Distribution- Controlled Ni-Pt Nanocrystals via Alcohol Reduction Technique for Oxygen Reduction Reaction

Kaneyuki Taniguchi,^a Jhon L. Cuya Huaman,^a Daichi Iwata,^a Shun Yokoyama,^b Takatoshi Matsumoto,^c Kazumasa Suzuki,^a Hiroshi Miyamura^a and Balachandran Jeyadevan^a*

^aDepartment of Materials Science, The University of Shiga Prefecture, Hikone, Shiga 522-8533, Japan.

^bGraduate School of Environmental Studies, Tohoku University, Sendai 980-8579, Japan.

^cInstitute of Multidisciplinary Research for Advanced Materials, Tohoku University, Sendai 980-8577, Japan.

*Corresponding author. E-mail: jeyadevan.b@mat.usp.ac.jp

Tel: +81-749-28-8352 Fax: +81-749-28-8486



Fig. S1 XRD patterns of Ni-Pt alloy NPs synthesized using different Ni:Pt precursor ratios.



Fig. S2 Cyclic voltammograms of (a) commercial Pt catalyst, (b) $Ni_{93}Pt_7/C$, (c) $Ni_{86}Pt_{14}/C$, (d) $Ni_{74}Pt_{26}/C$, (e) $Ni_{50}Pt_{50}/C$ and (f) $Ni_{27}Pt_{73}/C$.



Fig. S3 LSV curves of C-supported Pt and Ni-Pt alloy NPs with varying Pt concentrations.



Fig. S4 XPS spectra of Pt 4f for various compositions of Ni-Pt nanoparticles with different compositions.



Fig. S5 Cyclic voltammograms of (a) commercial Pt and (b) Pt coated Ni-Pt alloy NPs measured at different numbers of cycles.



Fig. S6 HRTEM images of Ni-Pt alloy NPs (a) before and (b) after Pt coating.



Fig. S7 XRD patterns of Ni-Pt nanoparticles before and after Pt coating by adding OAm and Pt precursor (OAm/Pt ratio = 5.0) mixed for 6h at 160 °C.



Fig. S8 LSV curves of (a) C-supported Pt and Ni-Pt alloy NPs (b) before coating and (c) after coating measured after 8000 cycles of catalytic treatment.