Supporting Information for

## Ag/AgCl Nanoparticles Embedded Porous TiO<sub>2</sub>: Defect Formation triggered by light irradiation

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Fig. S1. XRD pattern of (a) Ag/AgCl-pTiO<sub>2</sub> without aging (b) Ag/AgCl-pTiO<sub>2</sub> (c) Ag/AgCl-pTiO<sub>2</sub> before UV irradiation



Fig. S2. SEM images of (a,b) Cl-pTiO<sub>2</sub> (c,d) Ag/AgCl-pTiO<sub>2</sub>



Fig. S3. SEM-EDS mapping images of Ag/AgCl-pTiO<sub>2</sub>



## Fig. S4. EDS analysis of Ag/AgCl-pTiO $_2$



Element	Atomic %	
0	73.84	
C1	0.50	
Ti	25.65	
Total:	100.00	

Fig. S5. EDS analysis of Cl-pTiO<sub>2</sub>



Fig. S6. SEM-EDS mapping images of Ag/AgCl-pTiO<sub>2</sub> without aging



Fig. S7. Cl 2p XPS spectrum of the Cl-doped porous  $\rm TiO_2$ 

Element	Chemical composition of Ag/AgCl-pTiO <sub>2</sub> (Atomic %)		
	Nominal	EDS	XPS
Ti	28.9	23.7	26.7
0	57.8	73.9	66.9
Ag	5	2	4.6
C1	1.4	0.4	1.8

Table S1. Nominal and measured chemical composition of Ag/AgCl-pTiO\_2  $\,$ 



Fig. S8. Adsorption capability of Ag/AgCl-pTiO $_2$  vs Cl-pTiO $_2$  for rhodamine B



Fig. S9. Adsorption capability of Ag/AgCl-pTiO<sub>2</sub> vs Cl-pTiO<sub>2</sub> for methylene blue



Fig. S10. Photocatalytic activity of commercial anatase TiO<sub>2</sub>, pTiO<sub>2</sub>, Cl-pTiO<sub>2</sub> for rhodamine B degradation



Fig. S11. XRD patterns of (a) pTiO<sub>2</sub> (b) Cl-pTiO<sub>2</sub> and (c) anatase TiO<sub>2</sub> (JCPDS: 00-021-1272)



Fig. S12. Adsorption capabilities of Ag/AgCl-pTiO<sub>2</sub>-0.1, Ag/AgCl-pTiO<sub>2</sub>-0.5, Ag/AgCl-pTiO<sub>2</sub> for methylene blue



Fig. S13. XPS data Ti of Ag/AgCl-pTiO<sub>2</sub> after MB adsorption



Fig. S14. XPS data Ag of Ag/AgCl-pTiO<sub>2</sub> after MB adsorption



Fig. S15. Photocatalytic degradation efficiency of aged Ag/AgCl-TiO<sub>2</sub> and unaged Ag/AgCl- $pTiO_2$  for methylene blue



Fig. S16. Cycling photocatalytic performance for rhodamine B degradation of Ag/AgCl-pTiO<sub>2</sub> after light irradiation (405 nm) for 60 minutes.