

**Supporting Information**

**Achiral and chiral NNN-pincer nickel complexes with oxazolinyl backbone: application in transfer hydrogenation of ketones**

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## Contents

|                                    | Page # |
|------------------------------------|--------|
| 1. X-ray structure determination   | S3     |
| 2. References                      | S19    |
| 3. NMR spectra of <b>2a-2f</b>     | S20    |
| 4. NMR spectra of ligands          | S25    |
| 5. NMR spectra of nickel complexes | S31    |
| 6. NMR spectra of <b>6a</b>        | S37    |

## 1. X-ray structure determination

X-ray intensity data measurement of compounds **4a**, **4d** and **4f** was carried out on a Bruker D8 VENTURE Kappa Duo PHOTON II CPAD diffractometer equipped with Incoatech multilayer mirrors optics. The intensity measurements were carried out with Mo micro-focus sealed tube diffraction source ( $\text{MoK}_\alpha = 0.71073 \text{ \AA}$ ) at 100(2) K temperature. The X-ray generator was operated at 50 kV and 1.4 mA. A preliminary set of cell constants and an orientation matrix were calculated from three matrix sets of 36 frames (each matrix run consists of 12 frames). Data were collected with  $\omega$  scan width of 0.5° at different settings of  $\varphi$  and  $2\theta$  with a frame time of 10-20 sec depending on the diffraction power of the crystals keeping the sample-to-detector distance fixed at 5.00 cm. The X-ray data collection was monitored by APEX3 program (Bruker, 2016).<sup>S1</sup> All the data were corrected for Lorentzian, polarization and absorption effects using SAINT and SADABS programs (Bruker, 2016). Using the APEX3 (Bruker) program suite, the structure was solved with the ShelXS-97 (Sheldrick, 2008)<sup>S2</sup> structure solution program, using direct methods. The model was refined with a version of ShelXL-2018/3 (Sheldrick, 2015)<sup>S3</sup> using Least Squares minimization. All the hydrogen atoms were placed in a geometrically idealized position and constrained to ride on its parent atoms. An ORTEP III<sup>S4</sup> view of the compounds was drawn with 50% probability displacement ellipsoids, and H atoms are shown as small spheres of arbitrary radii. Crystal data for the structures have been deposited in the Cambridge Crystallographic Data Center with numbers (compound numbers) CCDC-2074093 (**4a**), CCDC-2074094 (**4d**) and CCDC-2074095 (**4f**).

A single crystal of a compound **4a** with molecular formula  $\text{C}_{15}\text{H}_{20}\text{ClN}_3\text{NiO}_2$ , approximate crystal dimensions 0.041 mm x 0.138 mm x 0.150 mm, was used for the X-ray crystallographic analysis. The frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm. The integration of the data using an orthorhombic unit cell yielded a total of 41734 reflections to a maximum  $\theta$  angle of 30.02° (0.71 Å resolution), of which 4658 were independent (average redundancy 8.960, completeness = 99.7%,  $R_{int} = 3.29\%$ ,  $R_{sig} = 1.65\%$ ) and 4192 (90.00%) were greater than  $2\sigma(F^2)$ . The final cell constants of  $a = 16.793(3) \text{ \AA}$ ,  $b = 9.9261(18) \text{ \AA}$ ,  $c = 19.198(4) \text{ \AA}$ , volume =  $3200.1(10) \text{ \AA}^3$ , are based upon the refinement of the XYZ-centroids of 1269 reflections above  $20 \sigma(I)$  with  $4.851^\circ < 2\theta$

< 58.65°. Data were corrected for absorption effects using the multi-scan method (SADABS). The ratio of minimum to maximum apparent transmission was 0.908. The calculated minimum and maximum transmission coefficients (based on crystal size) are 0.8190 and 0.9450. The structure was solved and refined using the Bruker SHELXTL Software Package, using the space group *Pbca*, with *Z* = 8 for the formula unit, C<sub>15</sub>H<sub>20</sub>ClN<sub>3</sub>NiO<sub>2</sub>. The final anisotropic full-matrix least-squares refinement on *F*<sup>2</sup> with 201 variables converged at *R*1 = 2.31%, for the observed data and *wR*2 = 5.75% for all data. The goodness-of-fit (*S*) was 1.028. The largest peak in the final difference electron density synthesis was 0.456 e<sup>-</sup>/Å<sup>3</sup> and the largest hole was -0.461 e<sup>-</sup>/Å<sup>3</sup> with an RMS deviation of 0.055 e<sup>-</sup>/Å<sup>3</sup>. On the basis of the final model, the calculated density was 1.530 g/cm<sup>3</sup> and *F*(000), 1536 e<sup>-</sup>.

A single crystal of a compound **4d** with molecular formula C<sub>22</sub>H<sub>26</sub>ClN<sub>3</sub>NiO<sub>2</sub>, approximate crystal dimensions 0.030 mm x 0.150 mm x 0.210 mm, was used for the X-ray crystallographic analysis. The integration of the data using an orthorhombic unit cell yielded a total of 44424 reflections to a maximum  $\theta$  angle of 34.63° (0.63 Å resolution), of which 8381 were independent (average redundancy 5.301, completeness = 95.2%, *R*<sub>int</sub> = 3.62%, *R*<sub>sig</sub> = 3.49%) and 7601 (90.69%) were greater than 2σ(*F*<sup>2</sup>). The final cell constants of *a* = 9.9695(4) Å, *b* = 10.0137(3) Å, *c* = 21.4431(7) Å, volume = 2140.70(13) Å<sup>3</sup>, are based upon the refinement of the XYZ-centroids of reflections above 20 σ(*I*). The calculated minimum and maximum transmission coefficients (based on crystal size) are 0.8090 and 0.9690. The structure was solved and refined using the Bruker SHELXTL Software Package, using the space group *P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>*, with *Z* = 4 for the formula unit, C<sub>22</sub>H<sub>26</sub>ClN<sub>3</sub>NiO<sub>2</sub>. The final anisotropic full-matrix least-squares refinement on *F*<sup>2</sup> with 264 variables converged at *R*1 = 2.64%, for the observed data and *wR*2 = 5.85% for all data. The goodness-of-fit (*S*) was 1.036. The largest peak in the final difference electron density synthesis was 0.254 e<sup>-</sup>/Å<sup>3</sup> and the largest hole was -0.369 e<sup>-</sup>/Å<sup>3</sup> with an RMS deviation of 0.052 e<sup>-</sup>/Å<sup>3</sup>. On the basis of the final model, the calculated density was 1.423 g/cm<sup>3</sup> and *F*(000), 960 e<sup>-</sup>. The absolute configuration (Flack parameter, *x* = 0.008(4)) is established by the structure determination of **4d** containing a chiral reference fragment of known absolute configuration and confirmed by anomalous dispersion effects in diffraction measurements on the crystals.

A single crystal of a compound **4f** with molecular formula C<sub>19</sub>H<sub>28</sub>ClN<sub>3</sub>NiO<sub>2</sub>, approximate crystal dimensions 0.050 mm x 0.170 mm x 0.200 mm, was used for the X-ray crystallographic analysis. The integration of the data using an orthorhombic unit cell yielded a total of 36549 reflections to a maximum  $\theta$  angle of 30.52° (0.70 Å resolution), of which 6183 were independent (average redundancy 5.911, completeness = 99.9%,  $R_{\text{int}} = 2.40\%$ ,  $R_{\text{sig}} = 2.49\%$ ) and 6036 (97.62%) were greater than  $2\sigma(F^2)$ . The final cell constants of  $a = 9.7315(3)$  Å,  $b = 9.8437(3)$  Å,  $c = 21.1840(7)$  Å, volume = 2029.30(11) Å<sup>3</sup>, are based upon the refinement of the XYZ-centroids of reflections above  $20\sigma(I)$ . The calculated minimum and maximum transmission coefficients (based on crystal size) are 0.8090 and 0.9470. The structure was solved and refined using the Bruker SHELXTL Software Package, using the space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, with Z = 4 for the formula unit, C<sub>19</sub>H<sub>28</sub>ClN<sub>3</sub>NiO<sub>2</sub>. The final anisotropic full-matrix least-squares refinement on  $F^2$  with 244 variables converged at  $R1 = 1.87\%$ , for the observed data and  $wR2 = 4.83\%$  for all data. The goodness-of-fit ( $S$ ) was 1.031. The largest peak in the final difference electron density synthesis was 0.305 e<sup>-</sup>/Å<sup>3</sup> and the largest hole was -0.296 e<sup>-</sup>/Å<sup>3</sup> with an RMS deviation of 0.042 e<sup>-</sup>/Å<sup>3</sup>. On the basis of the final model, the calculated density was 1.390 g/cm<sup>3</sup> and  $F(000)$ , 896 e<sup>-</sup>. The absolute configuration (Flack parameter, x = 0.023(8)) is established by the structure determination of **4f** containing a chiral reference fragment of known absolute configuration and confirmed by anomalous dispersion effects in diffraction measurements on the crystals.

**Table S1.** Crystal data of complexes **4a**, **4d** and **4f**.

|   | <b>4a</b>   | <b>4d</b>   | <b>4f</b>   |
|---|---|---|---|
| Formula   | C <sub>15</sub> H <sub>20</sub> ClN <sub>3</sub> NiO <sub>2</sub> | C <sub>22</sub> H <sub>26</sub> ClN <sub>3</sub> NiO <sub>2</sub> | C <sub>19</sub> H <sub>28</sub> ClN <sub>3</sub> NiO <sub>2</sub> |
| Molecular weight  | 368.50  | 458.62  | 424.60  |
| Crystal Size, mm  | 0.158 × 0.138 × 0.041   | 0.21 × 0.15 × 0.03  | 0.20 × 0.17 × 0.05  |
| Temp. (K)   | 100(2)  | 100(2)  | 100(2)  |
| Wavelength (Å)  | 0.71073   | 0.71073   | 0.71073   |
| Crystal Syst.   | orthorhombic  | orthorhombic  | orthorhombic  |
| Space Group   | <i>Pbca</i>   | <i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>                   | <i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>                   |
| <i>a</i> /Å   | 16.793(3)   | 9.9695(4)   | 9.7315(3)   |
| <i>b</i> /Å   | 9.9261(18)  | 10.0137(3)  | 9.8437(3)   |
| <i>c</i> /Å   | 19.198(4)   | 21.4431(7)  | 21.1840(7)  |
| <i>V</i> /Å <sup>3</sup>                                  | 3200.2(10)  | 2140.70(13)   | 2029.30(11)   |
| <i>Z</i>  | 8   | 4   | 4   |
| <i>D</i> <sub>calc</sub> /g cm <sup>-3</sup>              | 1.530   | 1.423   | 1.390   |
| $\mu$ /mm <sup>-1</sup>                                   | 1.389   | 1.054   | 1.105   |
| <i>F</i> (000)  | 1536  | 960   | 896   |
| <i>Ab. Correct.</i>                                       | multi-scan  | multi-scan  | multi-scan  |
| <i>T</i> <sub>min</sub> / <i>T</i> <sub>max</sub>         | 0.819/0.945   | 0.809/0.969   | 0.809/0.947   |
| 2 <i>θ</i> <sub>max</sub>                                 | 60  | 69.26   | 62  |
| Total reflns.   | 41734   | 44424   | 36549   |
| Unique reflns.  | 4658  | 8381  | 6183  |
| Obs. reflns.  | 4192  | 7601  | 6036  |
| <i>h</i> , <i>k</i> , <i>l</i> (min, max)                 | (-23, 23), (-12, 13),<br>(-25, 27)                                | (-14, 15), (-15, 15),<br>(-34, 33)                                | (-13, 13), (-14, 14),<br>(-30, 30)                                |
| R <sub>int</sub> / R <sub>sig</sub>                       | 0.0329 / 0.0165   | 0.0362 / 0.0349   | 0.0240 / 0.0249   |
| No. of parameters   | 201   | 264   | 245   |
| <i>RI</i> [ <i>I</i> > 2σ( <i>I</i> )]                    | 0.0231  | 0.0264  | 0.0187  |
| w <i>R</i> 2 [ <i>I</i> > 2σ( <i>I</i> )]                 | 0.0552  | 0.0557  | 0.0481  |
| <i>RI</i> [all data]                                      | 0.0272  | 0.0338  | 0.0193  |
| w <i>R</i> 2 [all data]                                   | 0.0575  | 0.0585  | 0.0483  |
| goodness-of-fit   | 1.028   | 1.036   | 1.030   |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (eÅ <sup>-3</sup> ) | +0.456, -0.461  | +0.254, -0.369  | +0.305, -0.295  |
| CCDC No.  | 2074093   | 2074094   | 2074095   |

**Table S2.** Bond lengths (Å) for **4a**.

|          |            |          |            |
|----------|------------|----------|------------|
| Ni1-N1   | 1.8780(10) | Ni1-N2   | 1.8806(10) |
| Ni1-N3   | 1.9514(10) | Ni1-Cl1  | 2.2025(4)  |
| O1-C3    | 1.3434(13) | O1-C2    | 1.4733(14) |
| O2-C10   | 1.2323(14) | N1-C3    | 1.2890(14) |
| N1-C1    | 1.4811(14) | N2-C10   | 1.3630(14) |
| N2-C9    | 1.4064(14) | N3-C11   | 1.4880(14) |
| N3-C12   | 1.4984(15) | N3-C14   | 1.5091(15) |
| C1-C2    | 1.5283(16) | C1-H1A   | 0.99       |
| C1-H1B   | 0.99       | C2-H2A   | 0.99       |
| C2-H2B   | 0.99       | C3-C4    | 1.4570(15) |
| C4-C5    | 1.4033(15) | C4-C9    | 1.4145(15) |
| C5-C6    | 1.3854(17) | C5-H5    | 0.95       |
| C6-C7    | 1.3959(17) | C6-H6    | 0.95       |
| C7-C8    | 1.3889(16) | C7-H7    | 0.95       |
| C8-C9    | 1.4107(16) | C8-H8    | 0.95       |
| C10-C11  | 1.5101(16) | C11-H11A | 0.99       |
| C11-H11B | 0.99       | C12-C13  | 1.5209(18) |
| C12-H12A | 0.99       | C12-H12B | 0.99       |
| C13-H13A | 0.98       | C13-H13B | 0.98       |
| C13-H13C | 0.98       | C14-C15  | 1.5187(18) |
| C14-H14A | 0.99       | C14-H14B | 0.99       |
| C15-H15A | 0.98       | C15-H15B | 0.98       |

**Table S3.** Bond angles ( $^{\circ}$ ) for **4a**.

|             |            |              |            |
|-------------|------------|--------------|------------|
| N1-Ni1-N2   | 91.05(4)   | N1-Ni1-N3    | 167.95(4)  |
| N2-Ni1-N3   | 85.93(4)   | N1-Ni1-Cl1   | 92.46(3)   |
| N2-Ni1-Cl1  | 165.90(3)  | N3-Ni1-Cl1   | 93.28(3)   |
| C3-O1-C2    | 106.55(8)  | C3-N1-C1     | 108.27(9)  |
| C3-N1-Ni1   | 126.43(8)  | C1-N1-Ni1    | 124.61(7)  |
| C10-N2-C9   | 120.35(9)  | C10-N2-Ni1   | 115.52(7)  |
| C9-N2-Ni1   | 123.53(7)  | C11-N3-C12   | 110.18(9)  |
| C11-N3-C14  | 108.65(9)  | C12-N3-C14   | 107.05(9)  |
| C11-N3-Ni1  | 105.64(7)  | C12-N3-Ni1   | 111.95(7)  |
| C14-N3-Ni1  | 113.33(7)  | N1-C1-C2     | 102.94(9)  |
| N1-C1-H1A   | 111.2      | C2-C1-H1A    | 111.2      |
| N1-C1-H1B   | 111.2      | C2-C1-H1B    | 111.2      |
| H1A-C1-H1B  | 109.1      | O1-C2-C1     | 103.99(9)  |
| O1-C2-H2A   | 111.0      | C1-C2-H2A    | 111.0      |
| O1-C2-H2B   | 111.0      | C1-C2-H2B    | 111.0      |
| H2A-C2-H2B  | 109.0      | N1-C3-O1     | 116.53(10) |
| N1-C3-C4    | 125.89(10) | O1-C3-C4     | 117.58(9)  |
| C5-C4-C9    | 120.86(10) | C5-C4-C3     | 119.24(10) |
| C9-C4-C3    | 119.89(10) | C6-C5-C4     | 120.57(11) |
| C6-C5-H5    | 119.7      | C4-C5-H5     | 119.7      |
| C5-C6-C7    | 119.07(11) | C5-C6-H6     | 120.5      |
| C7-C6-H6    | 120.5      | C8-C7-C6     | 121.04(11) |
| C8-C7-H7    | 119.5      | C6-C7-H7     | 119.5      |
| C7-C8-C9    | 120.94(11) | C7-C8-H8     | 119.5      |
| C9-C8-H8    | 119.5      | N2-C9-C8     | 122.09(10) |
| N2-C9-C4    | 120.42(10) | C8-C9-C4     | 117.43(10) |
| O2-C10-N2   | 128.05(10) | O2-C10-C11   | 119.97(10) |
| N2-C10-C11  | 111.92(9)  | N3-C11-C10   | 111.86(9)  |
| N3-C11-H11A | 109.2      | C10-C11-H11A | 109.2      |

|               |       |               |            |
|---------------|-------|---------------|------------|
| N3-C11-H11B   | 109.2 | C10-C11-H11B  | 109.2      |
| H11A-C11-H11B | 107.9 | N3-C12-C13    | 115.14(10) |
| N3-C12-H12A   | 108.5 | C13-C12-H12A  | 108.5      |
| N3-C12-H12B   | 108.5 | C13-C12-H12B  | 108.5      |
| H12A-C12-H12B | 107.5 | C12-C13-H13A  | 109.5      |
| C12-C13-H13B  | 109.5 | H13A-C13-H13B | 109.5      |
| C12-C13-H13C  | 109.5 | H13A-C13-H13C | 109.5      |
| H13B-C13-H13C | 109.5 | N3-C14-C15    | 113.91(10) |
| N3-C14-H14A   | 108.8 | C15-C14-H14A  | 108.8      |
| N3-C14-H14B   | 108.8 | C15-C14-H14B  | 108.8      |
| H14A-C14-H14B | 107.7 | C14-C15-H15A  | 109.5      |
| C14-C15-H15B  | 109.5 | H15A-C15-H15B | 109.5      |
| C14-C15-H15C  | 109.5 | H15A-C15-H15C | 109.5      |
| H15B-C15-H15C | 109.5 |               |            |

**Table S4.** Torsion angles ( $^{\circ}$ ) for **4a**.

|                |             |               |             |
|----------------|-------------|---------------|-------------|
| N2-Ni1-N1-C3   | -21.00(10)  | N3-Ni1-N1-C3  | -96.3(2)    |
| C11-Ni1-N1-C3  | 145.34(9)   | N2-Ni1-N1-C1  | 148.34(9)   |
| N3-Ni1-N1-C1   | 73.1(2)     | C11-Ni1-N1-C1 | -45.33(8)   |
| N1-Ni1-N2-C10  | -150.51(8)  | N3-Ni1-N2-C10 | 17.82(8)    |
| C11-Ni1-N2-C10 | 105.06(13)  | N1-Ni1-N2-C9  | 38.35(9)    |
| N3-Ni1-N2-C9   | -153.33(9)  | C11-Ni1-N2-C9 | -66.08(16)  |
| C3-N1-C1-C2    | -11.31(11)  | Ni1-N1-C1-C2  | 177.72(7)   |
| C3-O1-C2-C1    | -10.34(11)  | N1-C1-C2-O1   | 12.76(11)   |
| C1-N1-C3-O1    | 5.33(13)    | Ni1-N1-C3-O1  | 176.10(7)   |
| C1-N1-C3-C4    | -173.85(10) | Ni1-N1-C3-C4  | -3.08(16)   |
| C2-O1-C3-N1    | 3.55(13)    | C2-O1-C3-C4   | -177.20(9)  |
| N1-C3-C4-C5    | -160.97(11) | O1-C3-C4-C5   | 19.85(15)   |
| N1-C3-C4-C9    | 19.54(16)   | O1-C3-C4-C9   | -159.63(10) |
| C9-C4-C5-C6    | -2.36(16)   | C3-C4-C5-C6   | 178.16(10)  |

|                |             |                |             |
|----------------|-------------|----------------|-------------|
| C4-C5-C6-C7    | -0.17(17)   | C5-C6-C7-C8    | 1.97(17)    |
| C6-C7-C8-C9    | -1.25(17)   | C10-N2-C9-C8   | -26.69(15)  |
| Ni1-N2-C9-C8   | 144.04(9)   | C10-N2-C9-C4   | 156.29(10)  |
| Ni1-N2-C9-C4   | -32.97(14)  | C7-C8-C9-N2    | -178.33(10) |
| C7-C8-C9-C4    | -1.23(16)   | C5-C4-C9-N2    | -179.84(10) |
| C3-C4-C9-N2    | -0.37(15)   | C5-C4-C9-C8    | 3.01(15)    |
| C3-C4-C9-C8    | -177.51(10) | C9-N2-C10-O2   | -14.88(17)  |
| Ni1-N2-C10-O2  | 173.67(10)  | C9-N2-C10-C11  | 167.90(9)   |
| Ni1-N2-C10-C11 | -3.55(12)   | C12-N3-C11-C10 | -90.45(11)  |
| C14-N3-C11-C10 | 152.57(9)   | Ni1-N3-C11-C10 | 30.65(10)   |
| O2-C10-C11-N3  | 163.61(10)  | N2-C10-C11-N3  | -18.92(13)  |
| C11-N3-C12-C13 | 60.69(13)   | C14-N3-C12-C13 | 178.67(10)  |
| Ni1-N3-C12-C13 | -56.55(12)  | C11-N3-C14-C15 | -69.66(12)  |
| C12-N3-C14-C15 | 171.38(10)  | Ni1-N3-C14-C15 | 47.45(11)   |

**Table S5.** Bond lengths (Å) for **4d**.

|         |            |         |            |
|---------|------------|---------|------------|
| Ni1-N2  | 1.8826(12) | Ni1-N1  | 1.8904(12) |
| Ni1-N3  | 1.9611(12) | Ni1-C11 | 2.1896(4)  |
| O1-C3   | 1.3505(18) | O1-C2   | 1.455(2)   |
| O2-C10  | 1.2306(19) | N1-C3   | 1.2862(19) |
| N1-C1   | 1.4953(19) | N2-C10  | 1.356(2)   |
| N2-C9   | 1.4114(19) | N3-C11  | 1.486(2)   |
| N3-C14  | 1.506(2)   | N3-C12  | 1.509(2)   |
| C1-C16  | 1.522(2)   | C1-C2   | 1.527(2)   |
| C1-H1   | 1.0        | C2-H2A  | 0.99       |
| C2-H2AB | 0.99       | C3-C4   | 1.446(2)   |
| C4-C5   | 1.402(2)   | C4-C9   | 1.408(2)   |
| C5-C6   | 1.380(3)   | C5-H5   | 0.95       |
| C6-C7   | 1.384(3)   | C6-H6   | 0.95       |

|          |          |          |          |
|----------|----------|----------|----------|
| C7-C8    | 1.380(2) | C7-H7    | 0.95     |
| C8-C9    | 1.405(2) | C8-H8    | 0.95     |
| C10-C11  | 1.509(2) | C11-H11A | 0.99     |
| C11-H11B | 0.99     | C12-C13  | 1.513(2) |
| C12-H12A | 0.99     | C12-H12B | 0.99     |
| C13-H13A | 0.98     | C13-H13B | 0.98     |
| C13-H13C | 0.98     | C14-C15  | 1.515(3) |
| C14-H14A | 0.99     | C14-H14B | 0.99     |
| C15-H15A | 0.98     | C15-H15B | 0.98     |
| C15-H15C | 0.98     | C16-C17  | 1.514(2) |
| C16-H16A | 0.99     | C16-H16B | 0.99     |
| C17-C18  | 1.387(2) | C17-C22  | 1.388(2) |
| C18-C19  | 1.394(2) | C18-H18  | 0.95     |
| C19-C20  | 1.382(3) | C19-H19  | 0.95     |
| C20-C21  | 1.379(3) | C20-H20  | 0.95     |
| C21-C22  | 1.388(2) | C21-H21  | 0.95     |
| C22-H22  | 0.95     |          |          |

**Table S6.** Bond angles ( $^{\circ}$ ) for **4d**.

|            |            |            |            |
|------------|------------|------------|------------|
| N2-Ni1-N1  | 91.69(5)   | N2-Ni1-N3  | 85.96(5)   |
| N1-Ni1-N3  | 172.85(6)  | N2-Ni1-Cl1 | 171.32(4)  |
| N1-Ni1-Cl1 | 91.73(4)   | N3-Ni1-Cl1 | 91.55(4)   |
| C3-O1-C2   | 106.20(13) | C3-N1-C1   | 107.70(11) |
| C3-N1-Ni1  | 126.58(10) | C1-N1-Ni1  | 124.64(10) |
| C10-N2-C9  | 119.11(13) | C10-N2-Ni1 | 115.07(10) |
| C9-N2-Ni1  | 125.74(11) | C11-N3-C14 | 108.05(13) |
| C11-N3-C12 | 110.45(13) | C14-N3-C12 | 105.85(12) |
| C11-N3-Ni1 | 106.44(9)  | C14-N3-Ni1 | 116.01(10) |
| C12-N3-Ni1 | 110.02(10) | N1-C1-C16  | 108.30(13) |

|               |            |               |            |
|---------------|------------|---------------|------------|
| N1-C1-C2      | 102.02(13) | C16-C1-C2     | 114.04(14) |
| N1-C1-H1      | 110.7      | C16-C1-H1     | 110.7      |
| C2-C1-H1      | 110.7      | O1-C2-C1      | 104.43(13) |
| O1-C2-H2A     | 110.9      | C1-C2-H2A     | 110.9      |
| O1-C2-H2AB    | 110.9      | C1-C2-H2AB    | 110.9      |
| H2A-C2-H2AB   | 108.9      | N1-C3-O1      | 116.56(14) |
| N1-C3-C4      | 127.03(13) | O1-C3-C4      | 116.40(13) |
| C5-C4-C9      | 120.55(15) | C5-C4-C3      | 119.03(15) |
| C9-C4-C3      | 120.41(14) | C6-C5-C4      | 120.82(17) |
| C6-C5-H5      | 119.6      | C4-C5-H5      | 119.6      |
| C5-C6-C7      | 118.83(16) | C5-C6-H6      | 120.6      |
| C7-C6-H6      | 120.6      | C8-C7-C6      | 121.34(17) |
| C8-C7-H7      | 119.3      | C6-C7-H7      | 119.3      |
| C7-C8-C9      | 121.07(16) | C7-C8-H8      | 119.5      |
| C9-C8-H8      | 119.5      | C8-C9-C4      | 117.35(14) |
| C8-C9-N2      | 121.96(14) | C4-C9-N2      | 120.63(13) |
| O2-C10-N2     | 128.00(15) | O2-C10-C11    | 118.85(15) |
| N2-C10-C11    | 113.12(14) | N3-C11-C10    | 112.33(13) |
| N3-C11-H11A   | 109.1      | C10-C11-H11A  | 109.1      |
| N3-C11-H11B   | 109.1      | C10-C11-H11B  | 109.1      |
| H11A-C11-H11B | 107.9      | N3-C12-C13    | 115.16(13) |
| N3-C12-H12A   | 108.5      | C13-C12-H12A  | 108.5      |
| N3-C12-H12B   | 108.5      | C13-C12-H12B  | 108.5      |
| H12A-C12-H12B | 107.5      | C12-C13-H13A  | 109.5      |
| C12-C13-H13B  | 109.5      | H13A-C13-H13B | 109.5      |
| C12-C13-H13C  | 109.5      | H13A-C13-H13C | 109.5      |
| H13B-C13-H13C | 109.5      | N3-C14-C15    | 114.59(14) |
| N3-C14-H14A   | 108.6      | C15-C14-H14A  | 108.6      |
| N3-C14-H14B   | 108.6      | C15-C14-H14B  | 108.6      |

|               |            |               |            |
|---------------|------------|---------------|------------|
| H14A-C14-H14B | 107.6      | C14-C15-H15A  | 109.5      |
| C14-C15-H15B  | 109.5      | H15A-C15-H15B | 109.5      |
| C14-C15-H15C  | 109.5      | H15A-C15-H15C | 109.5      |
| H15B-C15-H15C | 109.5      | C17-C16-C1    | 113.98(13) |
| C17-C16-H16A  | 108.8      | C1-C16-H16A   | 108.8      |
| C17-C16-H16B  | 108.8      | C1-C16-H16B   | 108.8      |
| H16A-C16-H16B | 107.7      | C18-C17-C22   | 118.57(14) |
| C18-C17-C16   | 122.13(14) | C22-C17-C16   | 119.30(14) |
| C17-C18-C19   | 120.99(16) | C17-C18-H18   | 119.5      |
| C19-C18-H18   | 119.5      | C20-C19-C18   | 119.47(17) |
| C20-C19-H19   | 120.3      | C18-C19-H19   | 120.3      |
| C21-C20-C19   | 120.12(16) | C21-C20-H20   | 119.9      |
| C19-C20-H20   | 119.9      | C20-C21-C22   | 120.06(16) |
| C20-C21-H21   | 120.0      | C22-C21-H21   | 120.0      |
| C17-C22-C21   | 120.71(15) | C17-C22-H22   | 119.6      |
| C21-C22-H22   | 119.6      |               |            |

**Table S7.** Torsion angles ( $^{\circ}$ ) for **4d**.

|               |             |               |             |
|---------------|-------------|---------------|-------------|
| N2-Ni1-N1-C3  | 12.98(14)   | C11-Ni1-N1-C3 | -159.04(14) |
| N2-Ni1-N1-C1  | -153.65(12) | C11-Ni1-N1-C1 | 34.34(12)   |
| N1-Ni1-N2-C10 | 153.65(12)  | N3-Ni1-N2-C10 | -19.58(12)  |
| N1-Ni1-N2-C9  | -29.69(13)  | N3-Ni1-N2-C9  | 157.07(13)  |
| C3-N1-C1-C16  | -106.75(15) | Ni1-N1-C1-C16 | 62.01(15)   |
| C3-N1-C1-C2   | 13.86(17)   | Ni1-N1-C1-C2  | -177.39(12) |
| C3-O1-C2-C1   | 15.0(2)     | N1-C1-C2-O1   | -17.14(18)  |
| C16-C1-C2-O1  | 99.38(16)   | C1-N1-C3-O1   | -5.07(19)   |
| Ni1-N1-C3-O1  | -173.55(11) | C1-N1-C3-C4   | 175.61(17)  |
| Ni1-N1-C3-C4  | 7.1(3)      | C2-O1-C3-N1   | -6.8(2)     |
| C2-O1-C3-C4   | 172.62(16)  | N1-C3-C4-C5   | 161.32(18)  |

|                 |             |                 |             |
|-----------------|-------------|-----------------|-------------|
| O1-C3-C4-C5     | -18.0(2)    | N1-C3-C4-C9     | -17.5(3)    |
| O1-C3-C4-C9     | 163.20(15)  | C9-C4-C5-C6     | 1.1(3)      |
| C3-C4-C5-C6     | -177.69(18) | C4-C5-C6-C7     | 0.7(3)      |
| C5-C6-C7-C8     | -1.4(3)     | C6-C7-C8-C9     | 0.2(3)      |
| C7-C8-C9-C4     | 1.5(3)      | C7-C8-C9-N2     | 178.76(17)  |
| C5-C4-C9-C8     | -2.2(2)     | C3-C4-C9-C8     | 176.60(16)  |
| C5-C4-C9-N2     | -179.45(16) | C3-C4-C9-N2     | -0.7(2)     |
| C10-N2-C9-C8    | 26.8(2)     | Ni1-N2-C9-C8    | -149.69(14) |
| C10-N2-C9-C4    | -156.03(15) | Ni1-N2-C9-C4    | 27.4(2)     |
| C9-N2-C10-O2    | 14.3(3)     | Ni1-N2-C10-O2   | -168.77(15) |
| C9-N2-C10-C11   | -167.65(14) | Ni1-N2-C10-C11  | 9.24(18)    |
| C14-N3-C11-C10  | -149.88(13) | C12-N3-C11-C10  | 94.77(16)   |
| Ni1-N3-C11-C10  | -24.63(16)  | O2-C10-C11-N3   | -170.54(15) |
| N2-C10-C11-N3   | 11.3(2)     | C11-N3-C12-C13  | -57.84(19)  |
| C14-N3-C12-C13  | -174.56(15) | Ni1-N3-C12-C13  | 59.38(17)   |
| C11-N3-C14-C15  | 69.60(18)   | C12-N3-C14-C15  | -172.06(16) |
| Ni1-N3-C14-C15  | -49.75(19)  | N1-C1-C16-C17   | 174.16(12)  |
| C2-C1-C16-C17   | 61.35(18)   | C1-C16-C17-C18  | 48.5(2)     |
| C1-C16-C17-C22  | -132.29(15) | C22-C17-C18-C19 | -2.0(3)     |
| C16-C17-C18-C19 | 177.22(17)  | C17-C18-C19-C20 | -0.7(3)     |
| C18-C19-C20-C21 | 2.5(3)      | C19-C20-C21-C22 | -1.6(3)     |
| C18-C17-C22-C21 | 2.8(2)      | C16-C17-C22-C21 | -176.36(15) |
| C20-C21-C22-C17 | -1.1(3)     |                 |             |

**Table S8.** Bond lengths ( $\text{\AA}$ ) for **4f**.

|        |            |         |            |
|--------|------------|---------|------------|
| Ni1-N2 | 1.8860(12) | Ni1-N1  | 1.8887(12) |
| Ni1-N3 | 1.9569(13) | Ni1-Cl1 | 2.1952(4)  |
| O1-C3  | 1.3509(18) | O1-C2   | 1.452(2)   |
| O2-C10 | 1.2275(19) | N1-C3   | 1.2899(19) |

|           |            |           |            |
|-----------|------------|-----------|------------|
| N1-C1     | 1.4890(18) | N2-C10    | 1.3668(19) |
| N2-C9     | 1.4078(18) | N3-C11    | 1.486(2)   |
| N3-C12    | 1.504(2)   | N3-C14    | 1.510(2)   |
| C1-C16    | 1.524(2)   | C1-C2     | 1.533(2)   |
| C1-H1     | 1.0        | C2-H2A    | 0.99       |
| C2-H2AB   | 0.99       | C3-C4     | 1.447(2)   |
| C4-C5     | 1.404(2)   | C4-C9     | 1.408(2)   |
| C5-C6     | 1.380(2)   | C5-H5     | 0.95       |
| C6-C7     | 1.394(3)   | C6-H6     | 0.95       |
| C7-C8     | 1.387(2)   | C7-H7     | 0.95       |
| C8-C9     | 1.409(2)   | C8-H8     | 0.95       |
| C10-C11   | 1.507(2)   | C11-H11A  | 0.99       |
| C11-H11B  | 0.99       | C12-C13   | 1.518(2)   |
| C12-H12A  | 0.99       | C12-H12B  | 0.99       |
| C13-H13A  | 0.98       | C13-H13B  | 0.98       |
| C13-H13C  | 0.98       | C14-C15   | 1.507(3)   |
| C14-H14A  | 0.99       | C14-H14B  | 0.99       |
| C15-H15A  | 0.98       | C15-H15B  | 0.98       |
| C15-H15C  | 0.98       | C16-C17   | 1.538(2)   |
| C16-H16A  | 0.99       | C16-H16B  | 0.99       |
| C17-C19A  | 1.409(6)   | C17-C18   | 1.494(3)   |
| C17-C19B  | 1.593(4)   | C17-H17   | 1.0        |
| C18-H18A  | 0.98       | C18-H18B  | 0.98       |
| C18-H18C  | 0.98       | C19A-H19A | 0.98       |
| C19A-H19B | 0.98       | C19A-H19C | 0.98       |
| C19B-H19C | 0.98       | C19B-H19D | 0.98       |
| C19B-H19E | 0.98       |           |            |

**Table S9.** Bond angles ( $^{\circ}$ ) for **4f**.

|             |            |            |            |
|-------------|------------|------------|------------|
| N2-Ni1-N1   | 92.04(5)   | N2-Ni1-N3  | 85.95(5)   |
| N1-Ni1-N3   | 173.47(6)  | N2-Ni1-C11 | 169.78(4)  |
| N1-Ni1-C11  | 91.43(4)   | N3-Ni1-C11 | 91.60(4)   |
| C3-O1-C2    | 106.28(12) | C3-N1-C1   | 108.02(12) |
| C3-N1-Ni1   | 126.06(10) | C1-N1-Ni1  | 125.55(10) |
| C10-N2-C9   | 119.50(13) | C10-N2-Ni1 | 114.88(10) |
| C9-N2-Ni1   | 125.46(10) | C11-N3-C12 | 110.19(13) |
| C11-N3-C14  | 107.96(13) | C12-N3-C14 | 106.47(13) |
| C11-N3-Ni1  | 107.07(10) | C12-N3-Ni1 | 109.88(9)  |
| C14-N3-Ni1  | 115.23(11) | N1-C1-C16  | 110.56(12) |
| N1-C1-C2    | 101.68(12) | C16-C1-C2  | 113.31(13) |
| N1-C1-H1    | 110.3      | C16-C1-H1  | 110.3      |
| C2-C1-H1    | 110.3      | O1-C2-C1   | 104.18(12) |
| O1-C2-H2A   | 110.9      | C1-C2-H2A  | 110.9      |
| O1-C2-H2AB  | 110.9      | C1-C2-H2AB | 110.9      |
| H2A-C2-H2AB | 108.9      | N1-C3-O1   | 116.06(13) |
| N1-C3-C4    | 127.12(13) | O1-C3-C4   | 116.80(13) |
| C5-C4-C9    | 120.92(14) | C5-C4-C3   | 118.26(14) |
| C9-C4-C3    | 120.81(13) | C6-C5-C4   | 120.76(15) |
| C6-C5-H5    | 119.6      | C4-C5-H5   | 119.6      |
| C5-C6-C7    | 118.90(15) | C5-C6-H6   | 120.5      |
| C7-C6-H6    | 120.5      | C8-C7-C6   | 121.03(15) |
| C8-C7-H7    | 119.5      | C6-C7-H7   | 119.5      |
| C7-C8-C9    | 121.11(15) | C7-C8-H8   | 119.4      |
| C9-C8-H8    | 119.4      | C4-C9-N2   | 120.73(13) |
| C4-C9-C8    | 117.26(14) | N2-C9-C8   | 121.98(13) |
| O2-C10-N2   | 127.84(16) | O2-C10-C11 | 118.99(14) |
| N2-C10-C11  | 113.15(13) | N3-C11-C10 | 112.20(12) |

|                |          |                |            |
|----------------|----------|----------------|------------|
| N3-C11-H11A    | 109.2    | C10-C11-H11A   | 109.2      |
| N3-C11-H11B    | 109.2    | C10-C11-H11B   | 109.2      |
| H11A-C11-H11B  | 107.9    | N3-C12-C13     | 114.72(13) |
| N3-C12-H12A    | 108.6    | C13-C12-H12A   | 108.6      |
| N3-C12-H12B    | 108.6    | C13-C12-H12B   | 108.6      |
| H12A-C12-H12B  | 107.6    | C12-C13-H13A   | 109.5      |
| C12-C13-H13B   | 109.5    | H13A-C13-H13B  | 109.5      |
| C12-C13-H13C   | 109.5    | H13A-C13-H13C  | 109.5      |
| H13B-C13-H13C  | 109.5    | C15-C14-N3     | 114.95(17) |
| C15-C14-H14A   | 108.5    | N3-C14-H14A    | 108.5      |
| C15-C14-H14B   | 108.5    | N3-C14-H14B    | 108.5      |
| H14A-C14-H14B  | 107.5    | C14-C15-H15A   | 109.5      |
| C14-C15-H15B   | 109.5    | H15A-C15-H15B  | 109.5      |
| C14-C15-H15C   | 109.5    | H15A-C15-H15C  | 109.5      |
| H15B-C15-H15C  | 109.5    | C1-C16-C17     | 112.70(13) |
| C1-C16-H16A    | 109.1    | C17-C16-H16A   | 109.1      |
| C1-C16-H16B    | 109.1    | C17-C16-H16B   | 109.1      |
| H16A-C16-H16B  | 107.8    | C19A-C17-C18   | 120.4(3)   |
| C19A-C17-C16   | 117.5(3) | C18-C17-C16    | 111.51(16) |
| C18-C17-C19B   | 104.3(2) | C16-C17-C19B   | 108.50(17) |
| C19A-C17-H17   | 100.9    | C18-C17-H17    | 100.9      |
| C16-C17-H17    | 100.9    | C17-C18-H18A   | 109.5      |
| C17-C18-H18B   | 109.5    | H18A-C18-H18B  | 109.5      |
| C17-C18-H18C   | 109.5    | H18A-C18-H18C  | 109.5      |
| H18B-C18-H18C  | 109.5    | C17-C19A-H19A  | 109.5      |
| C17-C19A-H19B  | 109.5    | H19A-C19A-H19B | 109.5      |
| C17-C19A-H19C  | 109.5    | H19A-C19A-H19C | 109.5      |
| H19B-C19A-H19C | 109.5    | C17-C19B-H19C  | 109.5      |
| C17-C19B-H19D  | 109.5    | H19C-C19B-H19D | 109.5      |

|                |       |                |       |
|----------------|-------|----------------|-------|
| C17-C19B-H19E  | 109.5 | H19C-C19B-H19E | 109.5 |
| H19D-C19B-H19E | 109.5 |                |       |

**Table S10.** Torsion angles ( $^{\circ}$ ) for **4f**.

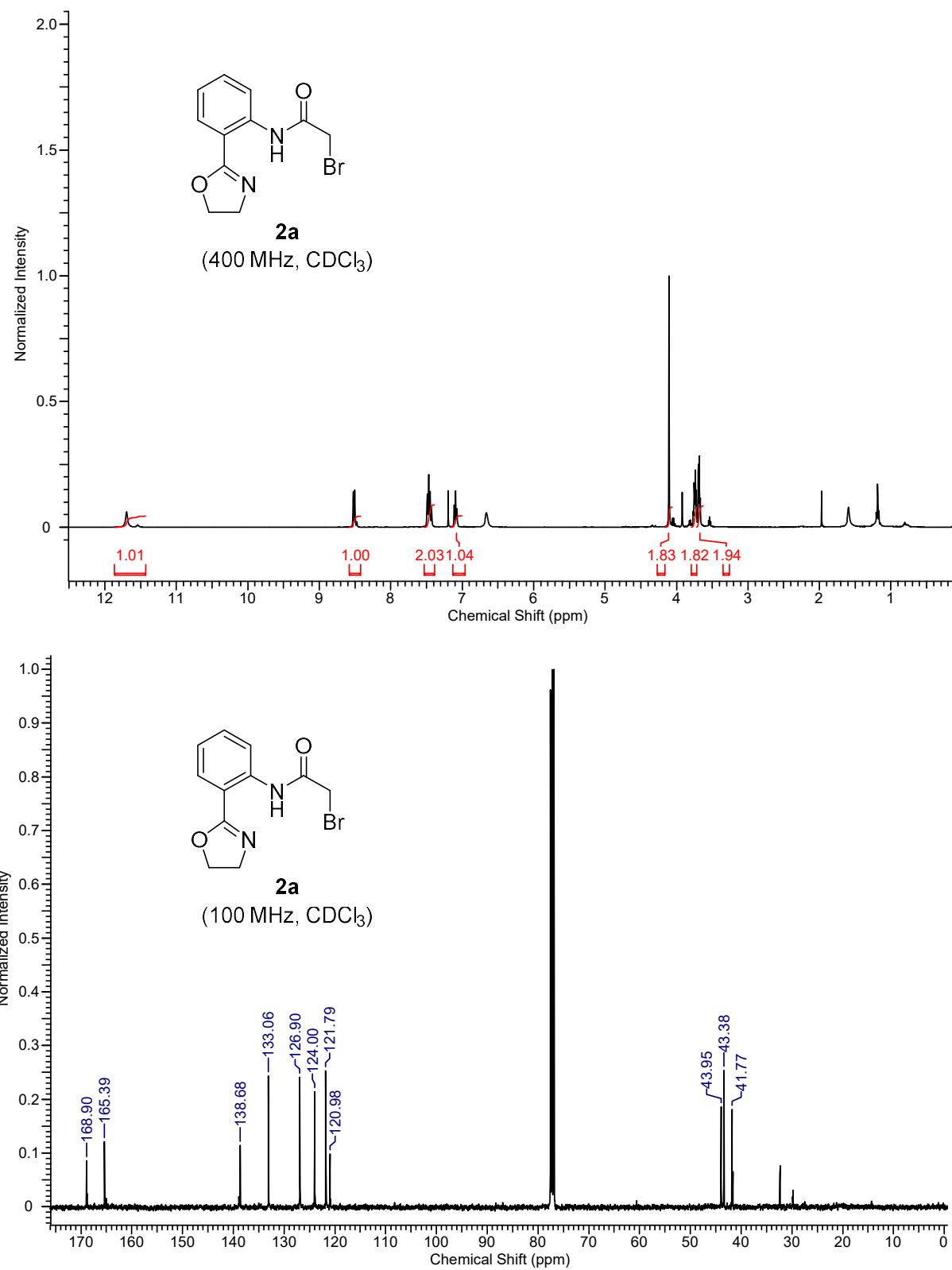
|                |             |                |             |
|----------------|-------------|----------------|-------------|
| N2-Ni1-N1-C3   | 15.65(13)   | C11-Ni1-N1-C3  | -154.74(12) |
| N2-Ni1-N1-C1   | -156.46(12) | C11-Ni1-N1-C1  | 33.15(11)   |
| N1-Ni1-N2-C10  | 154.58(11)  | N3-Ni1-N2-C10  | -19.20(11)  |
| C11-Ni1-N2-C10 | -95.6(2)    | N1-Ni1-N2-C9   | -30.14(12)  |
| N3-Ni1-N2-C9   | 156.08(12)  | C11-Ni1-N2-C9  | 79.7(3)     |
| C3-N1-C1-C16   | -105.69(14) | Ni1-N1-C1-C16  | 67.62(15)   |
| C3-N1-C1-C2    | 14.91(15)   | Ni1-N1-C1-C2   | -171.78(10) |
| C3-O1-C2-C1    | 16.96(16)   | N1-C1-C2-O1    | -18.91(15)  |
| C16-C1-C2-O1   | 99.74(14)   | C1-N1-C3-O1    | -4.95(18)   |
| Ni1-N1-C3-O1   | -178.21(10) | C1-N1-C3-C4    | 176.66(14)  |
| Ni1-N1-C3-C4   | 3.4(2)      | C2-O1-C3-N1    | -8.20(18)   |
| C2-O1-C3-C4    | 170.37(13)  | N1-C3-C4-C5    | 163.55(15)  |
| O1-C3-C4-C5    | -14.8(2)    | N1-C3-C4-C9    | -15.8(2)    |
| O1-C3-C4-C9    | 165.85(13)  | C9-C4-C5-C6    | 1.0(2)      |
| C3-C4-C5-C6    | -178.31(15) | C4-C5-C6-C7    | 0.2(3)      |
| C5-C6-C7-C8    | -0.8(3)     | C6-C7-C8-C9    | 0.2(2)      |
| C5-C4-C9-N2    | -179.29(14) | C3-C4-C9-N2    | 0.0(2)      |
| C5-C4-C9-C8    | -1.5(2)     | C3-C4-C9-C8    | 177.78(14)  |
| C10-N2-C9-C4   | -158.53(14) | Ni1-N2-C9-C4   | 26.39(19)   |
| C10-N2-C9-C8   | 23.8(2)     | Ni1-N2-C9-C8   | -151.27(12) |
| C7-C8-C9-C4    | 0.9(2)      | C7-C8-C9-N2    | 178.65(14)  |
| C9-N2-C10-O2   | 15.0(2)     | Ni1-N2-C10-O2  | -169.38(14) |
| C9-N2-C10-C11  | -166.37(13) | Ni1-N2-C10-C11 | 9.21(17)    |
| C12-N3-C11-C10 | 95.44(15)   | C14-N3-C11-C10 | -148.66(15) |
| Ni1-N3-C11-C10 | -24.03(16)  | O2-C10-C11-N3  | -170.56(15) |

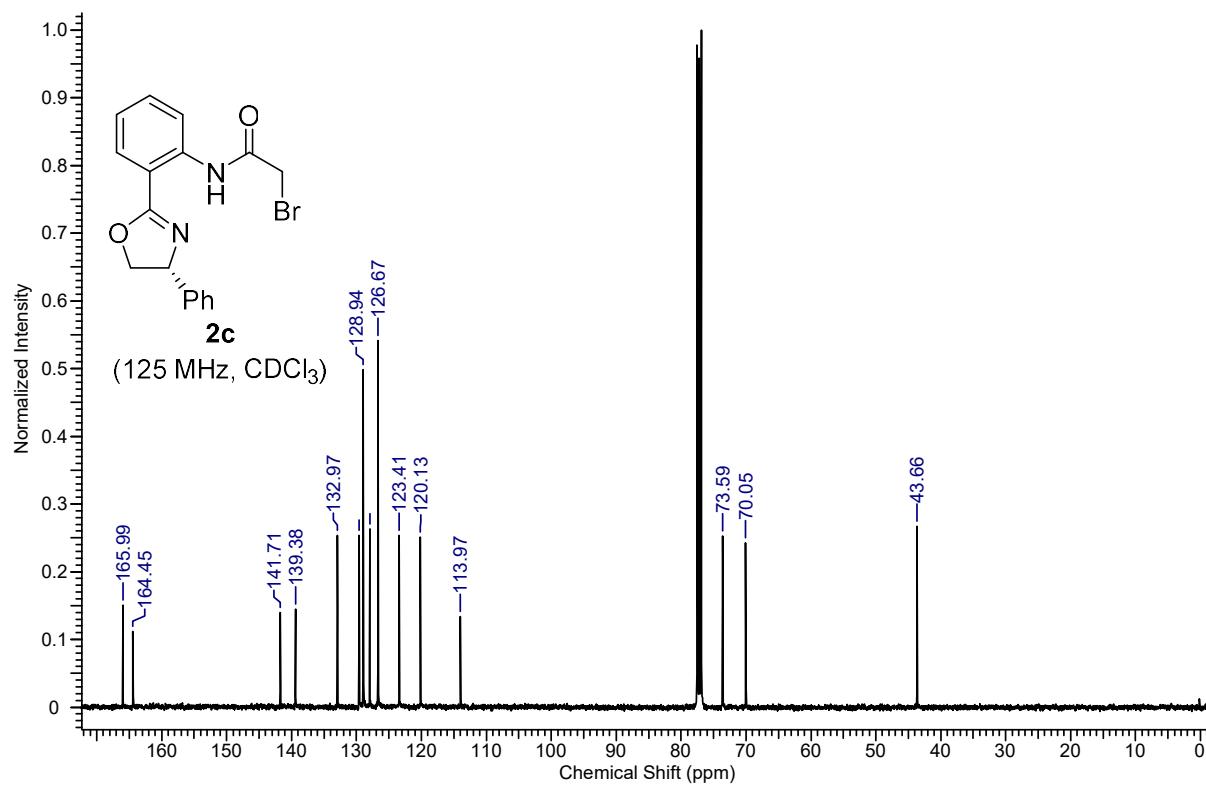
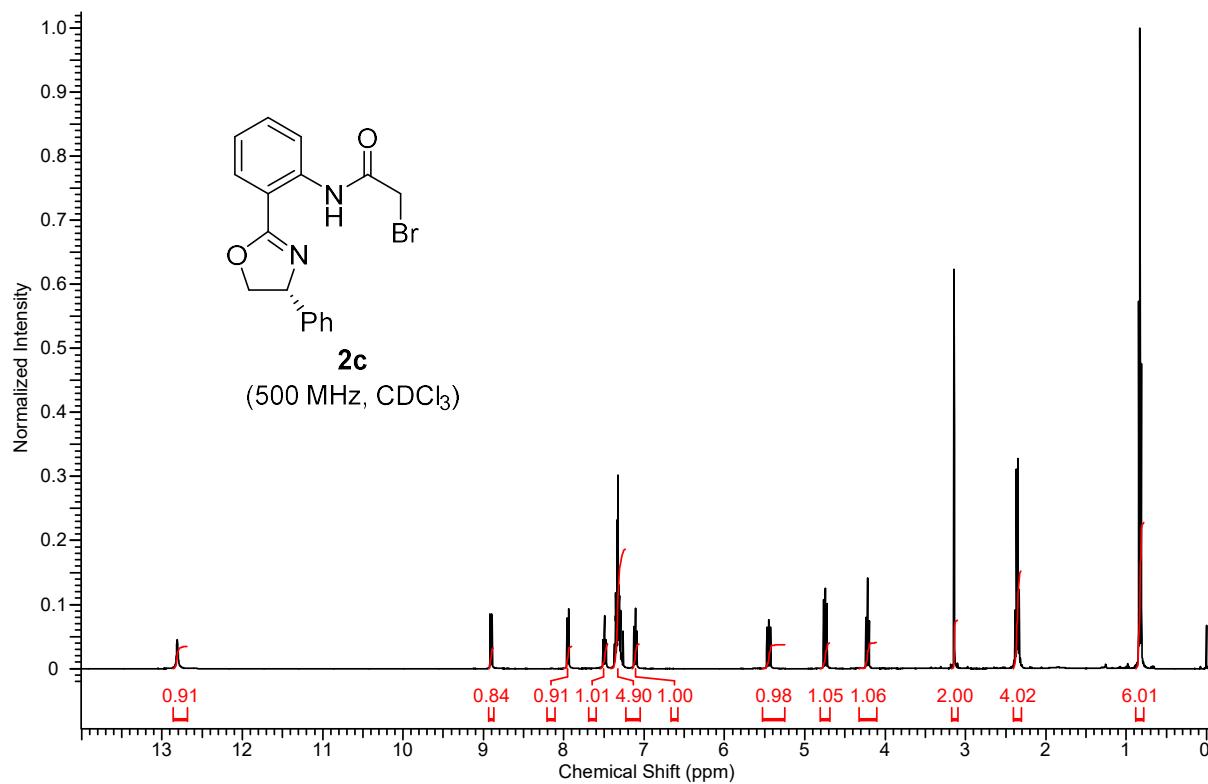
|                |             |                 |             |
|----------------|-------------|-----------------|-------------|
| N2-C10-C11-N3  | 10.7(2)     | C11-N3-C12-C13  | -57.95(17)  |
| C14-N3-C12-C13 | -174.78(14) | Ni1-N3-C12-C13  | 59.80(15)   |
| C11-N3-C14-C15 | 69.87(19)   | C12-N3-C14-C15  | -171.83(15) |
| Ni1-N3-C14-C15 | -49.73(19)  | N1-C1-C16-C17   | -172.40(13) |
| C2-C1-C16-C17  | 74.21(16)   | C1-C16-C17-C19A | 52.5(4)     |
| C1-C16-C17-C18 | -162.43(17) | C1-C16-C17-C19B | 83.3(2)     |

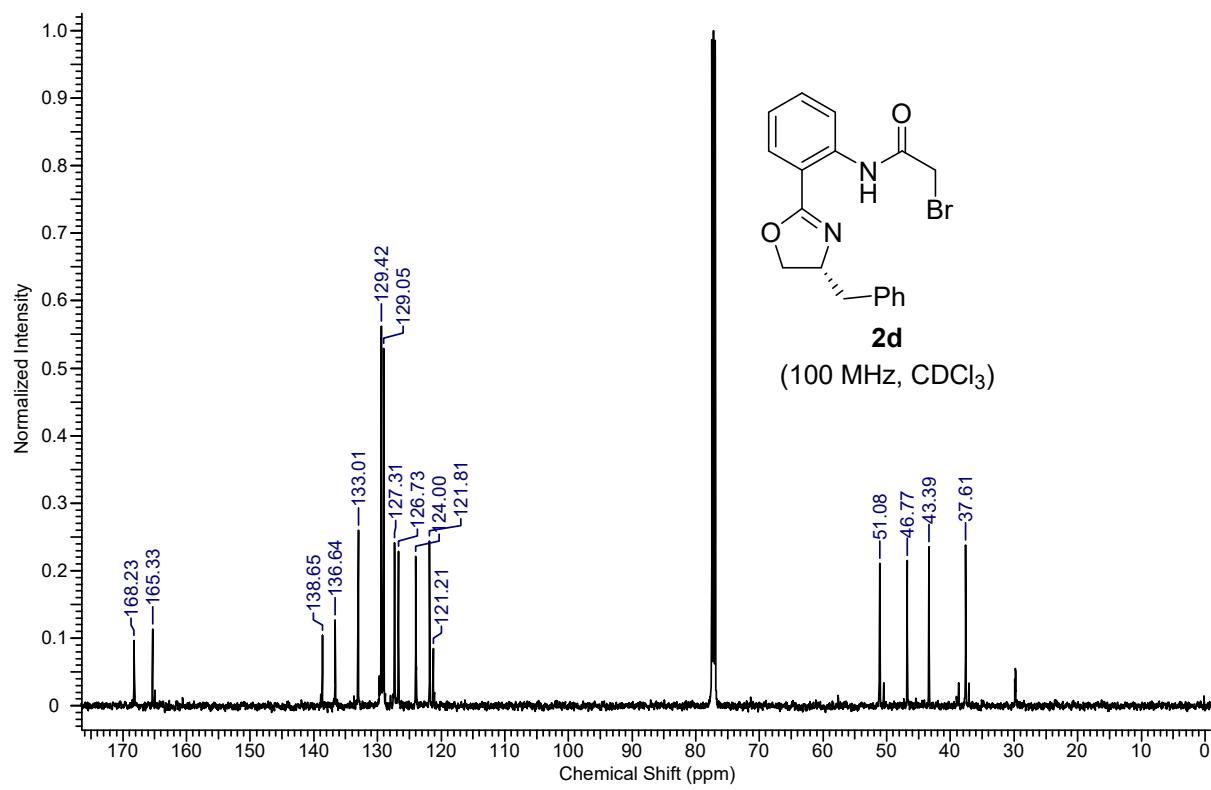
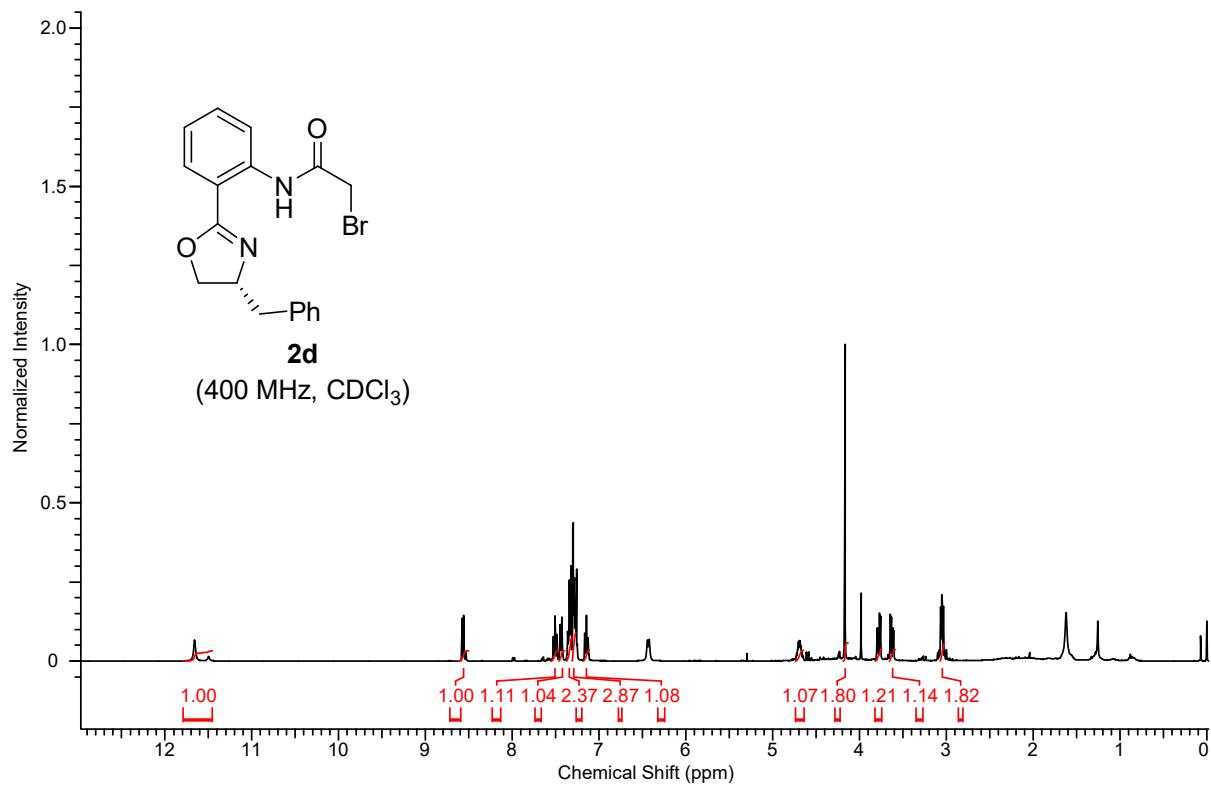
## 2. References

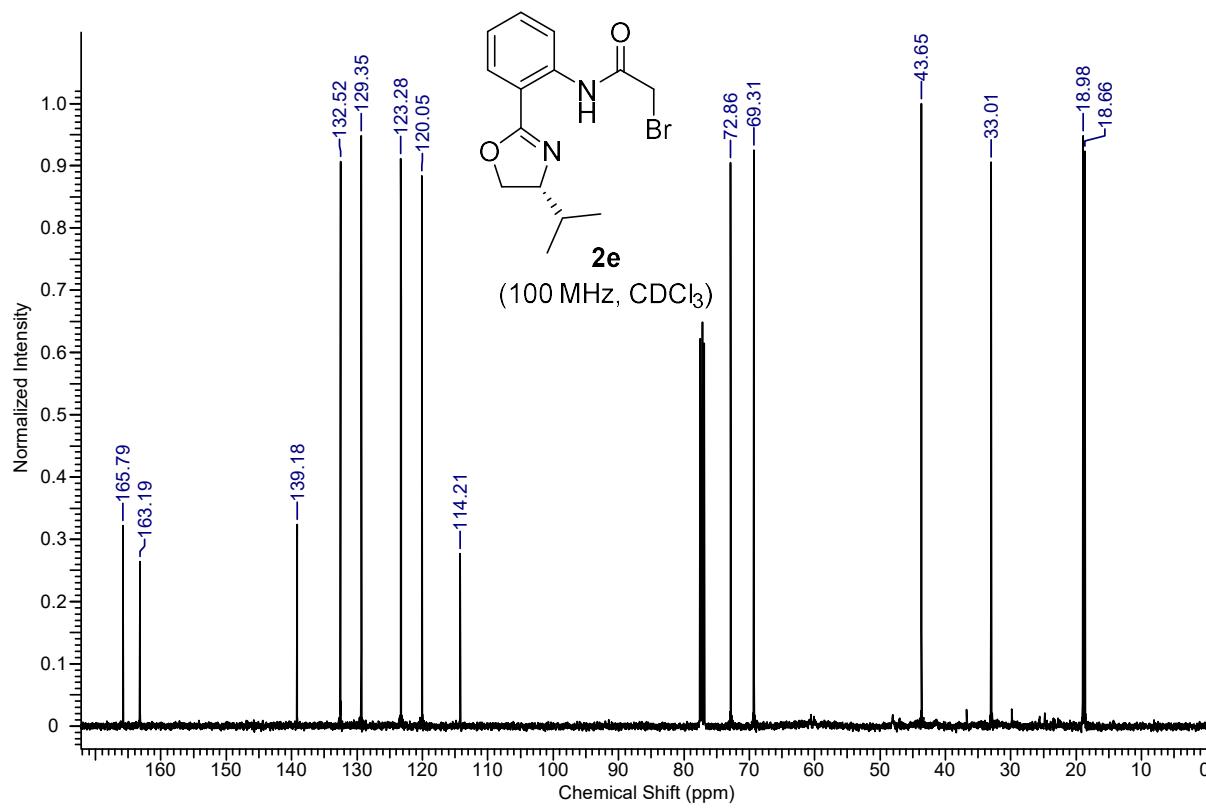
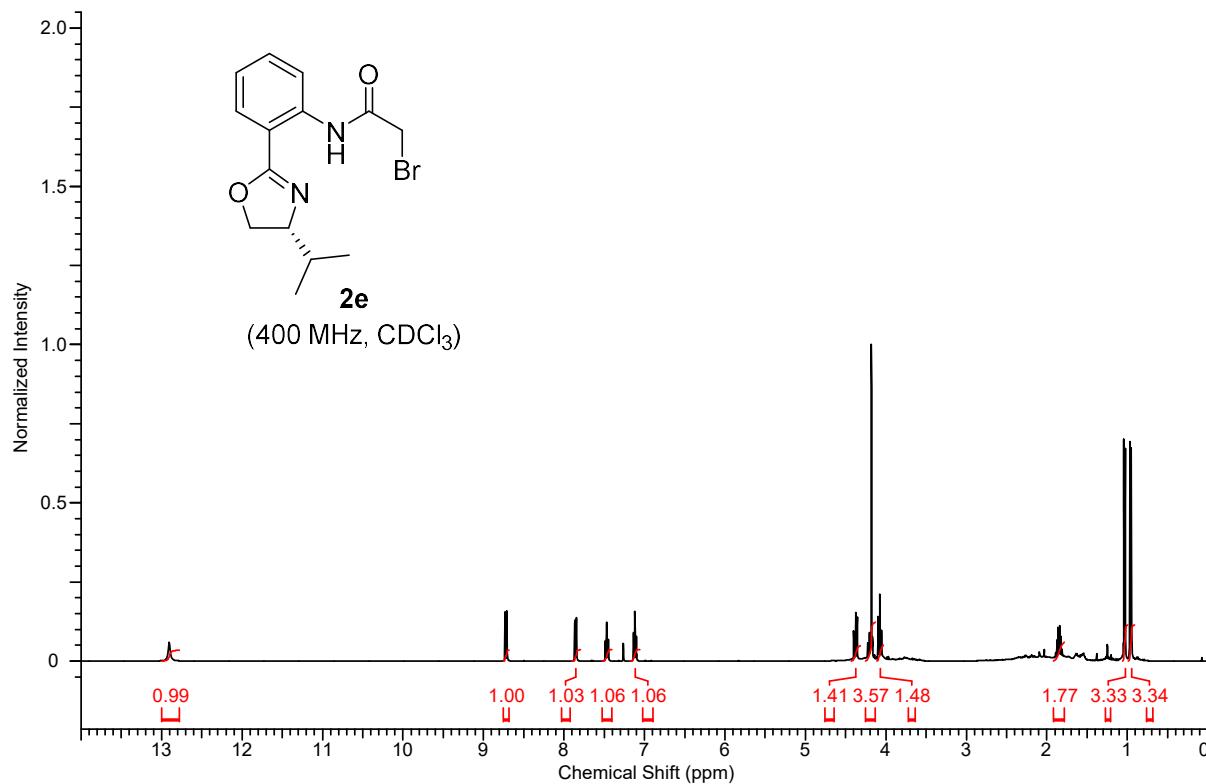
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- S2. G. M. Sheldrick, *Acta Crystallogr.*, 2008, **A64**, 112.
- S3. G. M. Sheldrick, *Acta Crystallogr.*, 2015, **C71**, 3–8.
- S4. L. J. Farrugia, *J. Appl. Crystallogr.* 2012, **45**, 849–854.

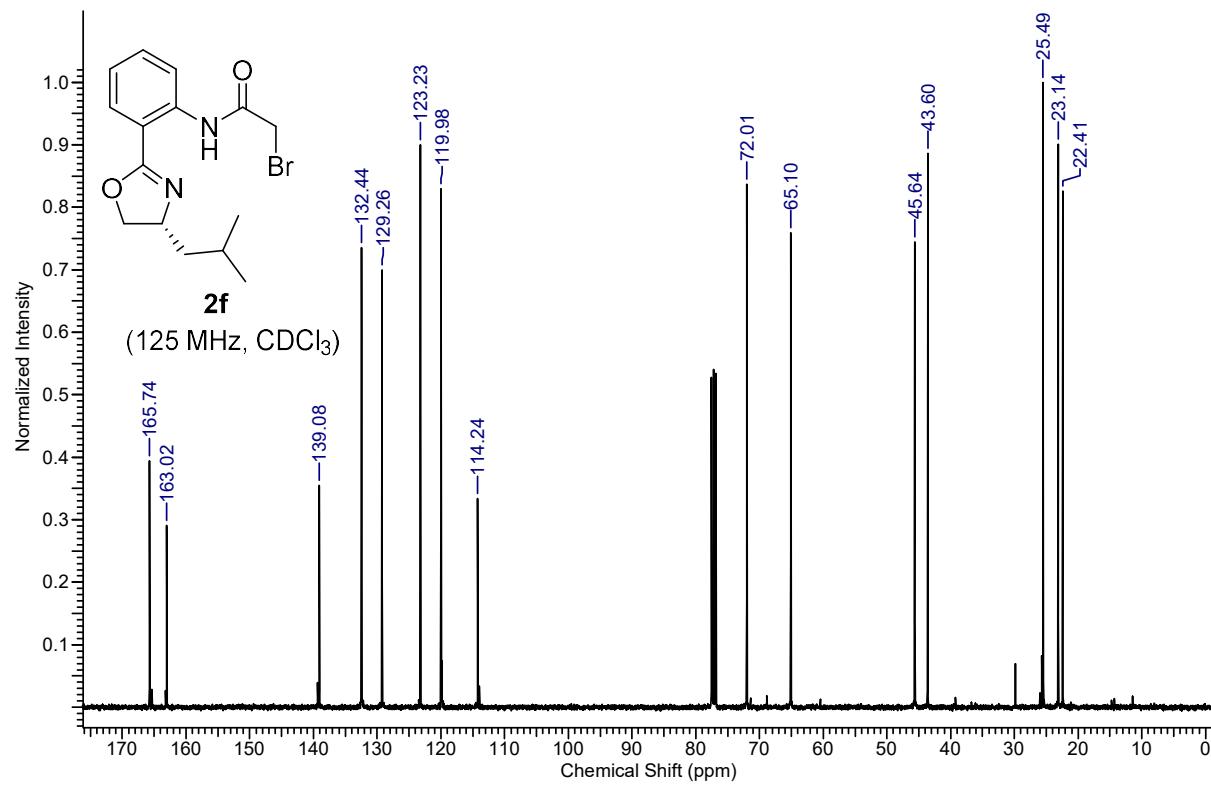
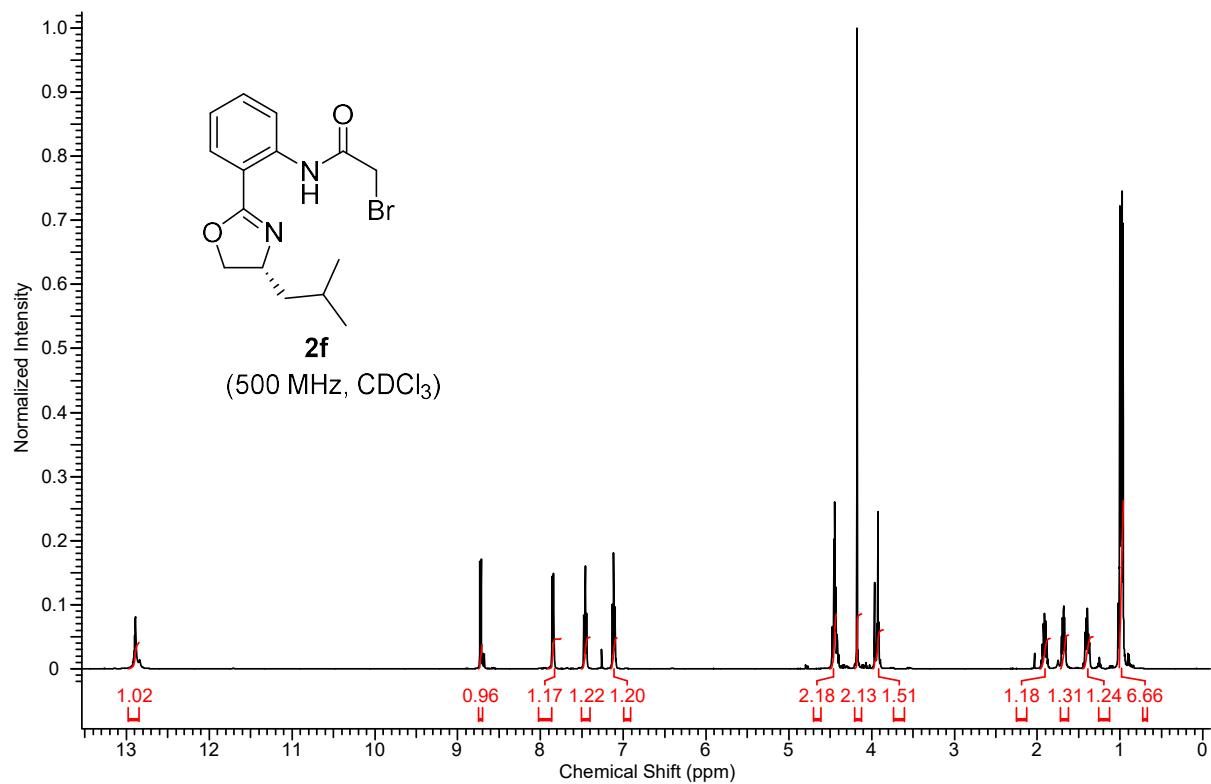
**3.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Spectra of 2a-2f**



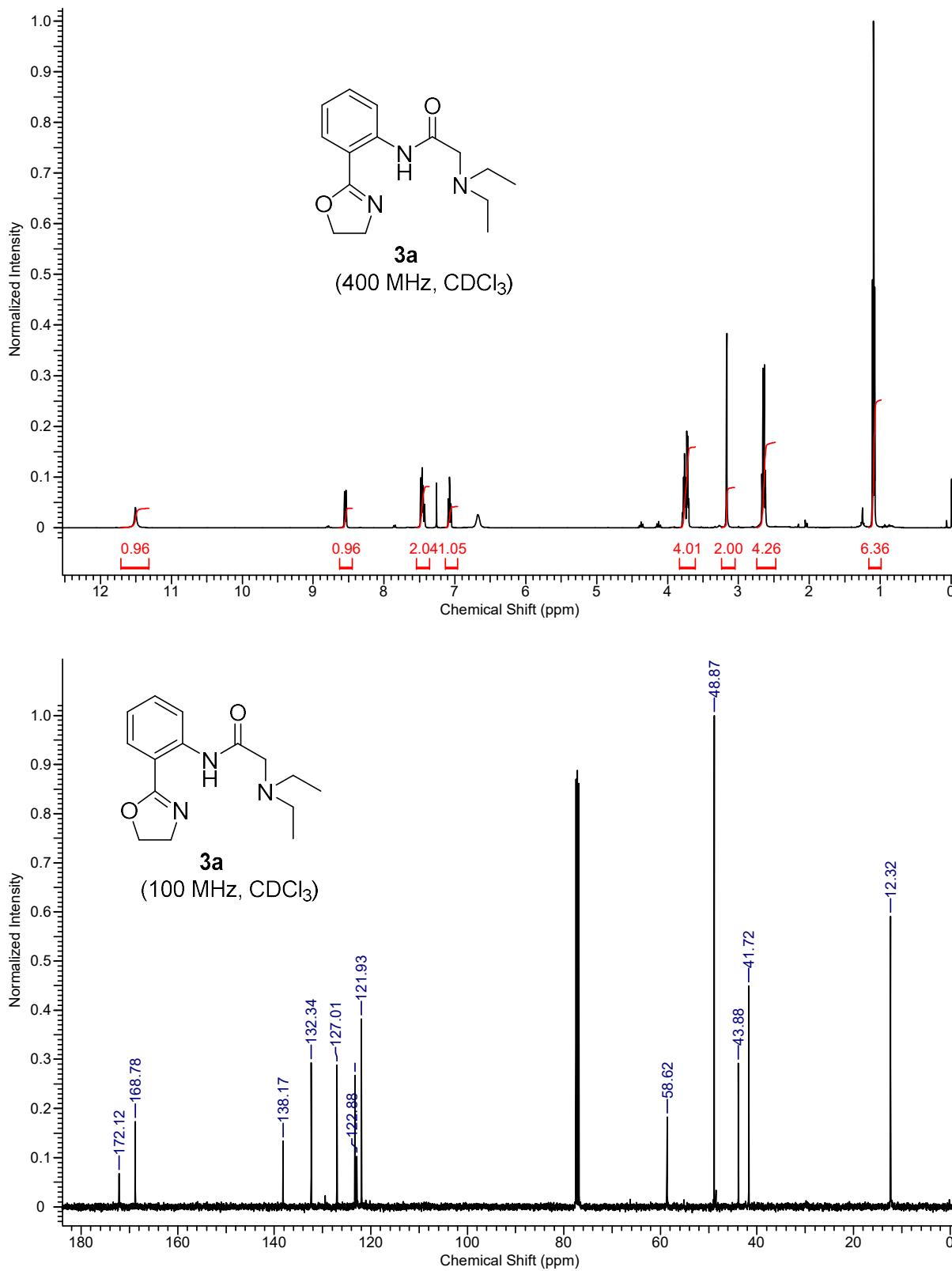


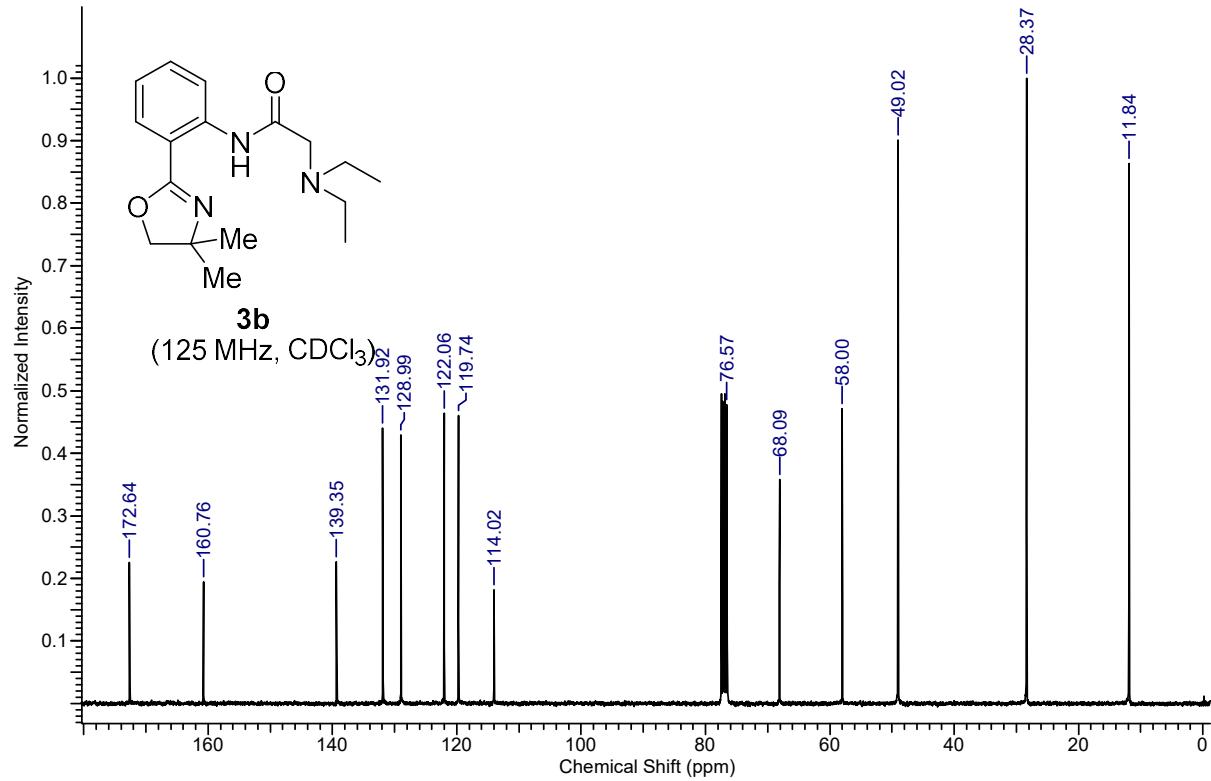
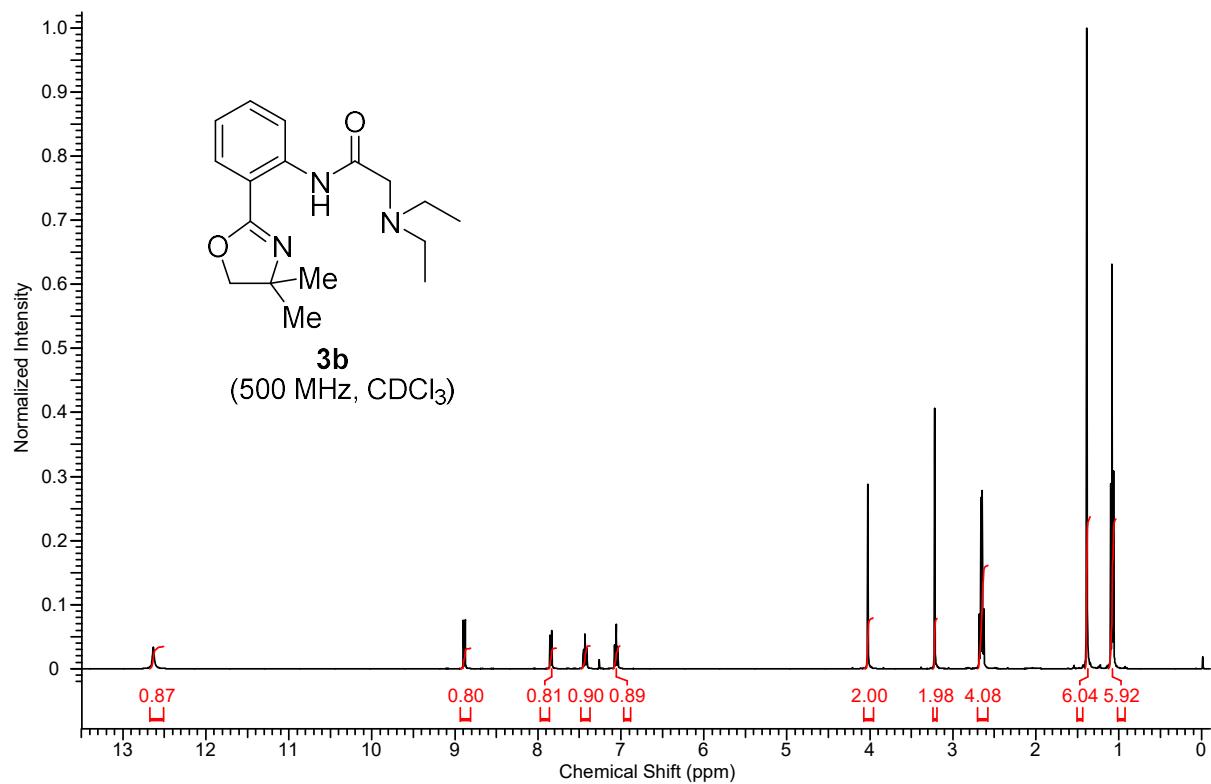


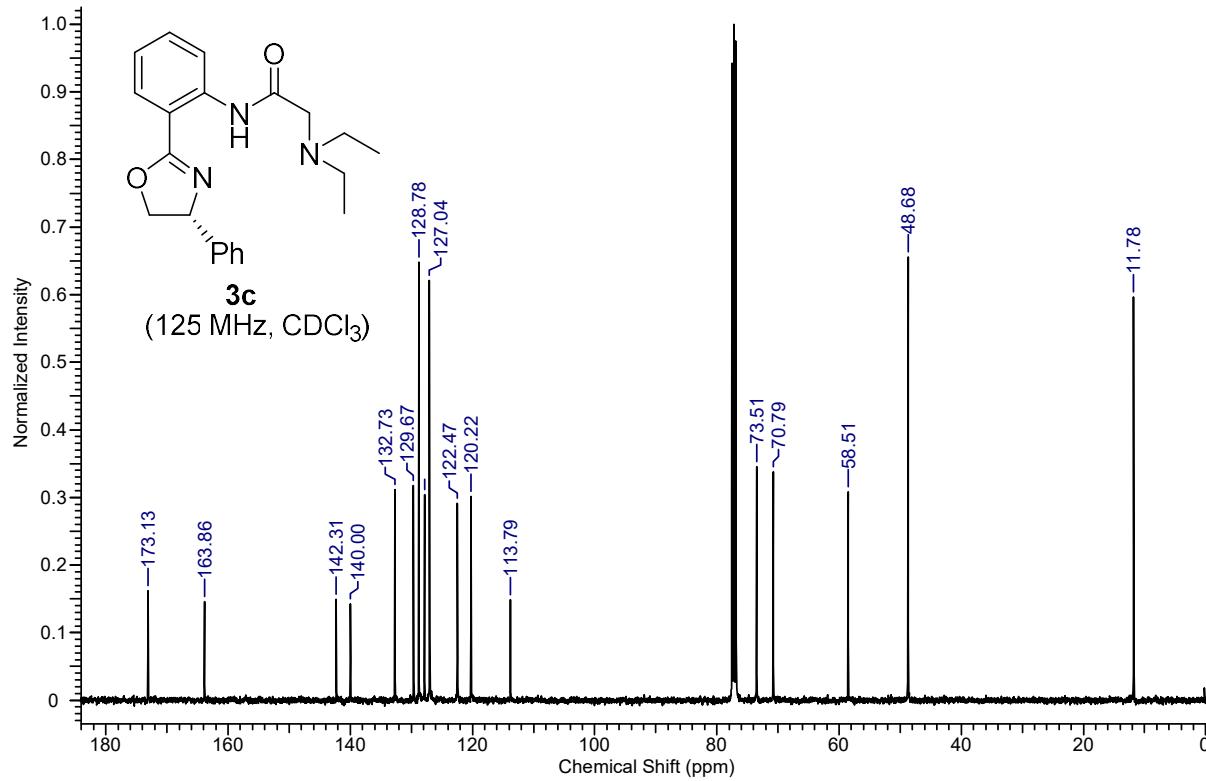
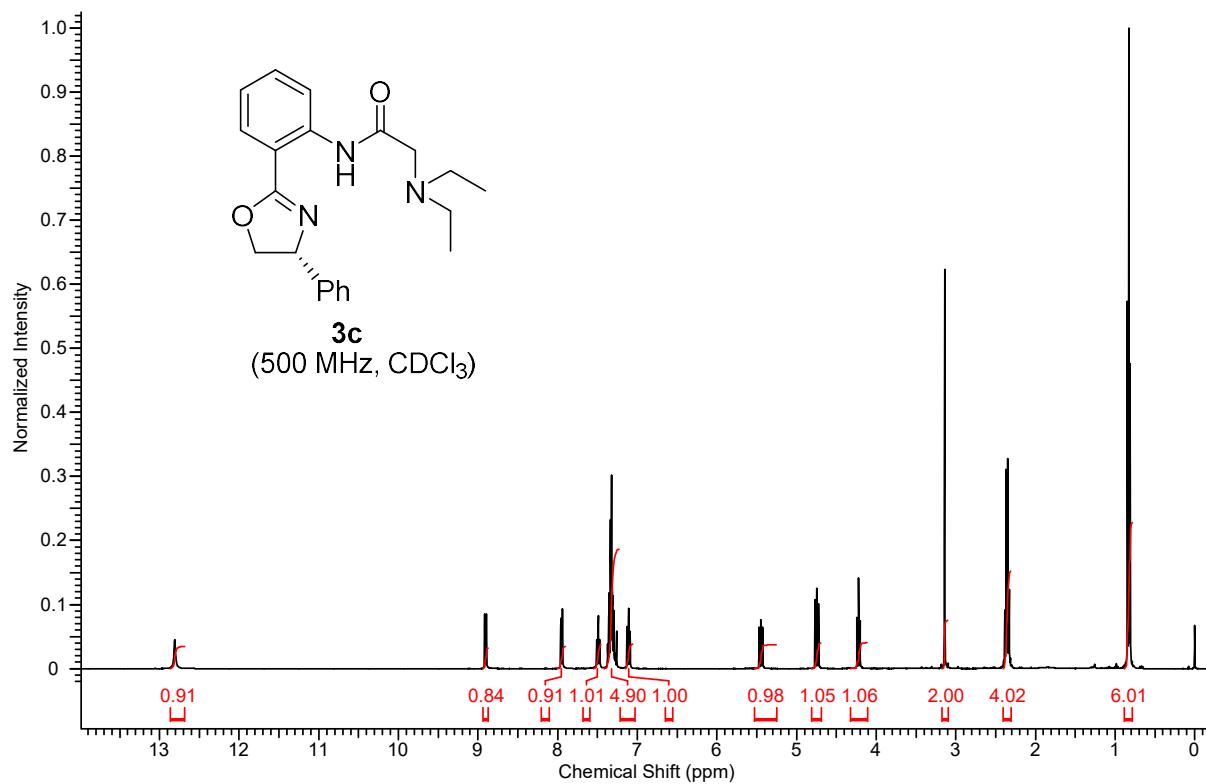


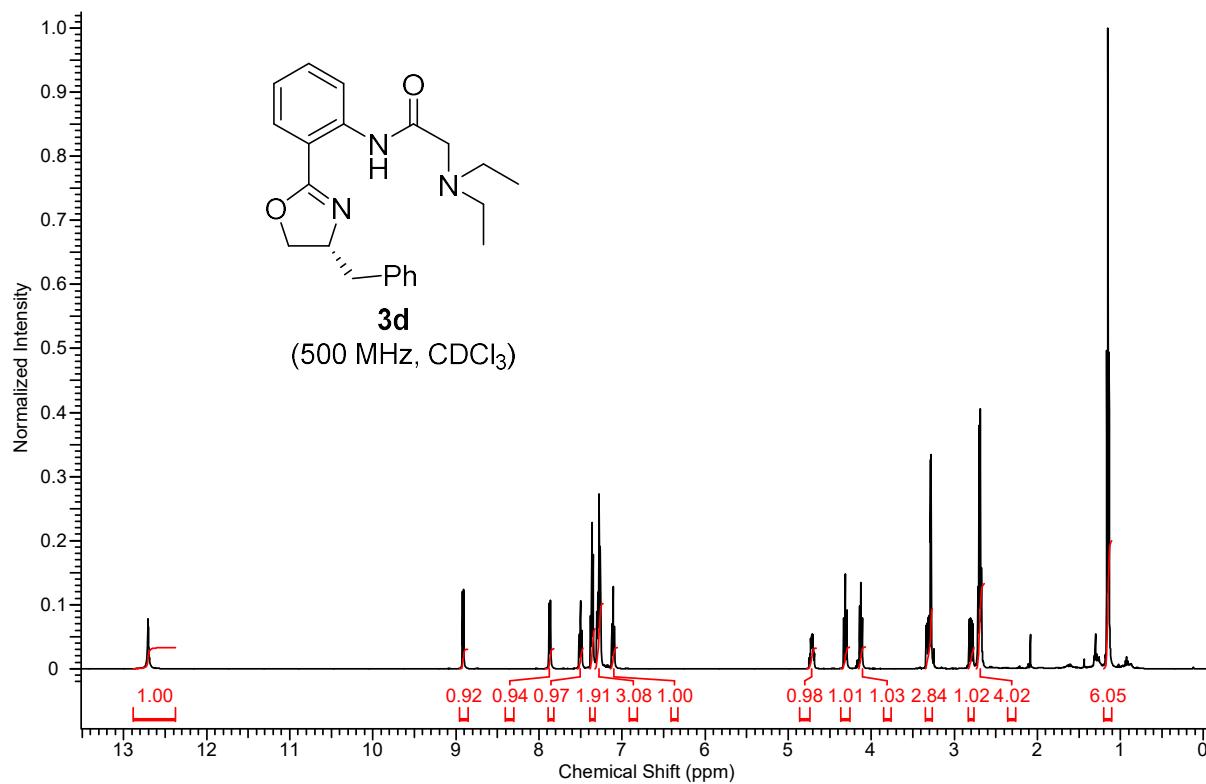


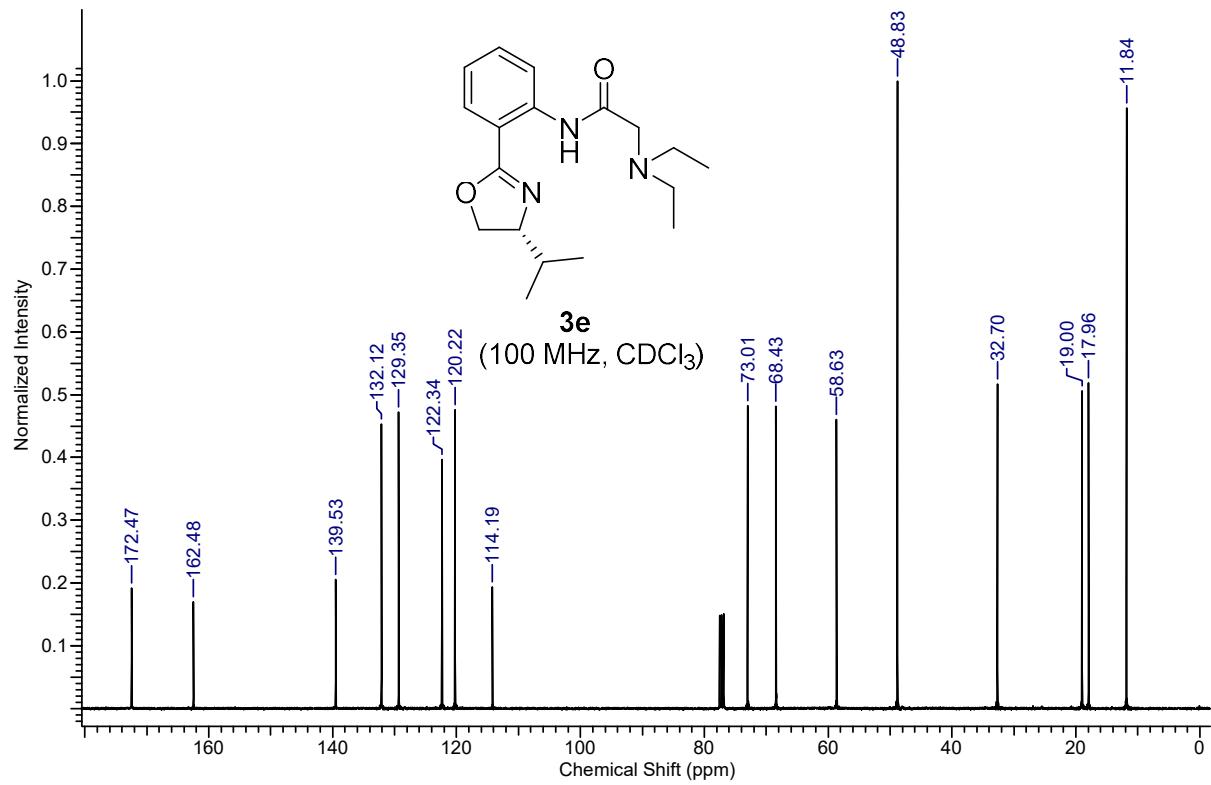
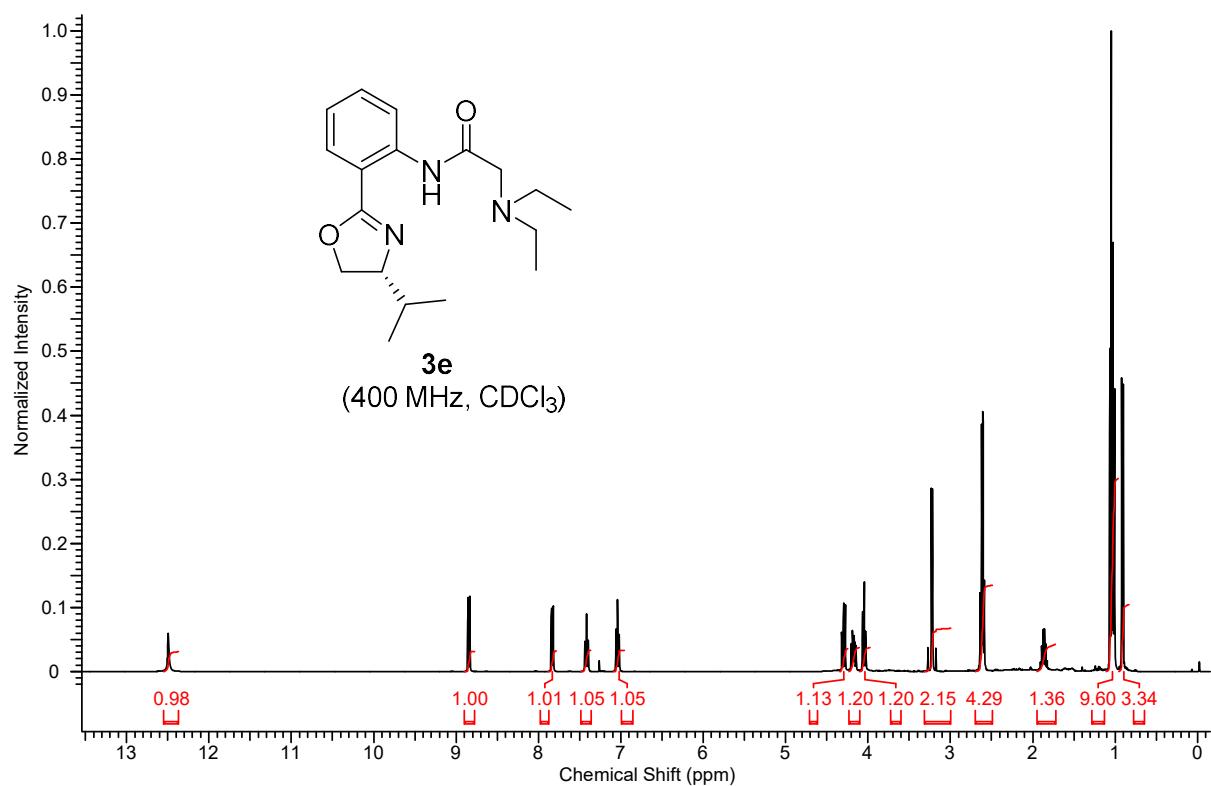
#### 4. $^1\text{H}$ and $^{13}\text{C}$ NMR Spectra of Ligands

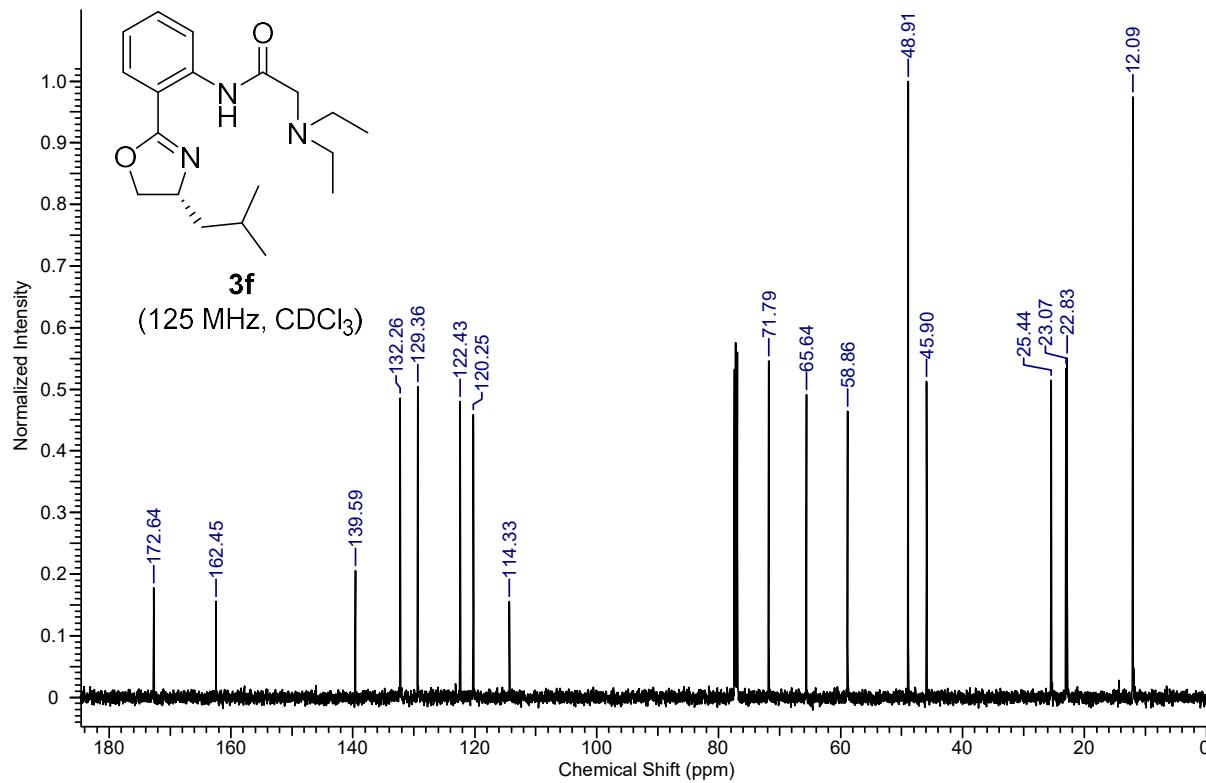
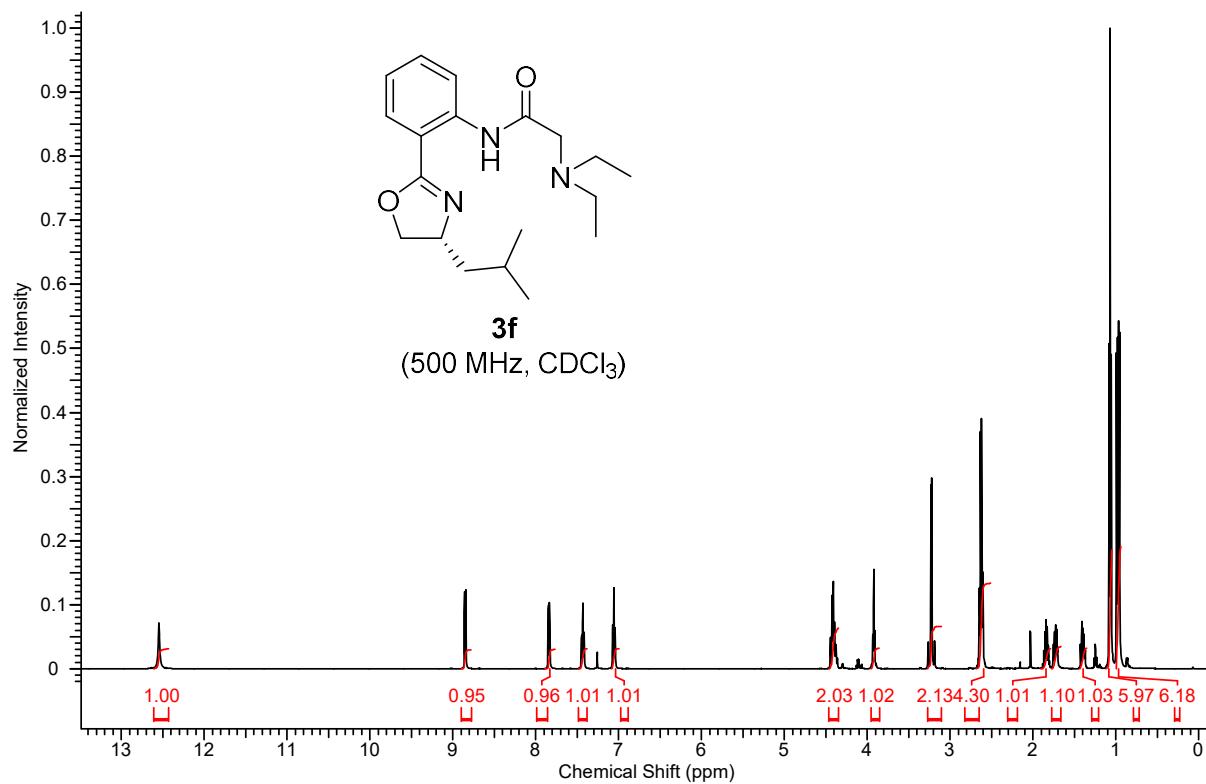




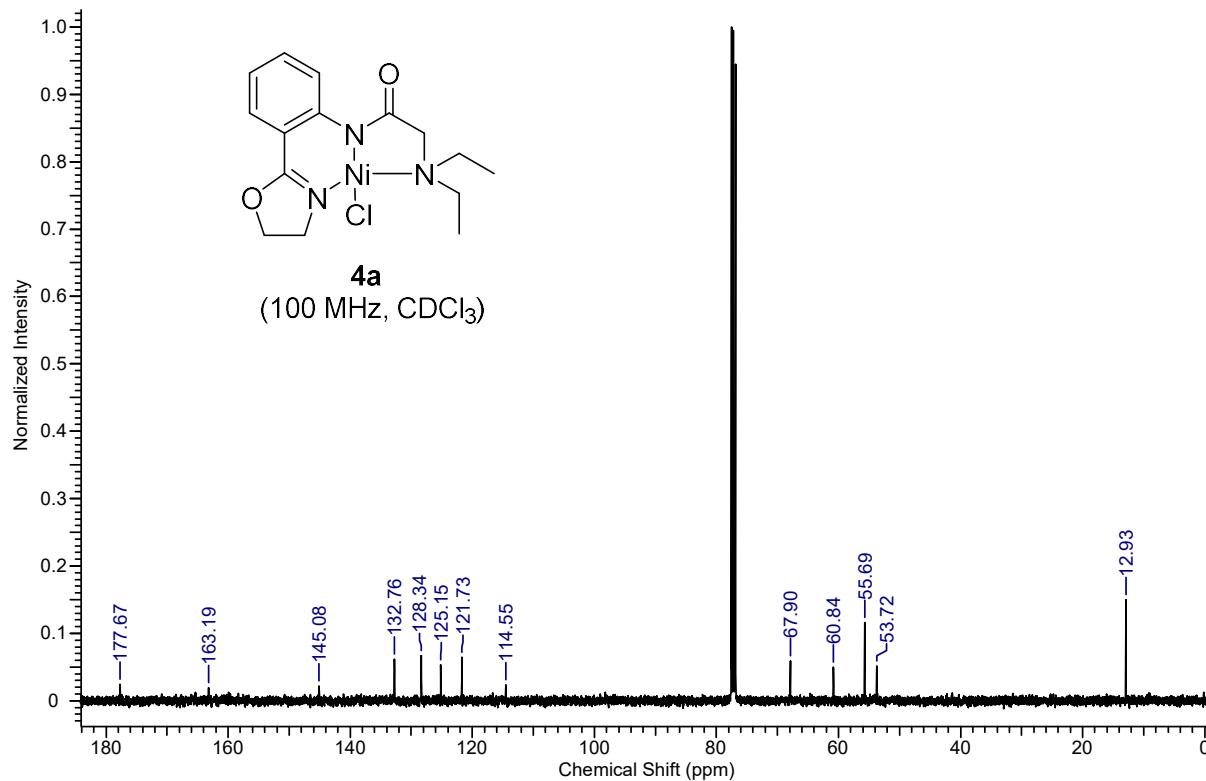
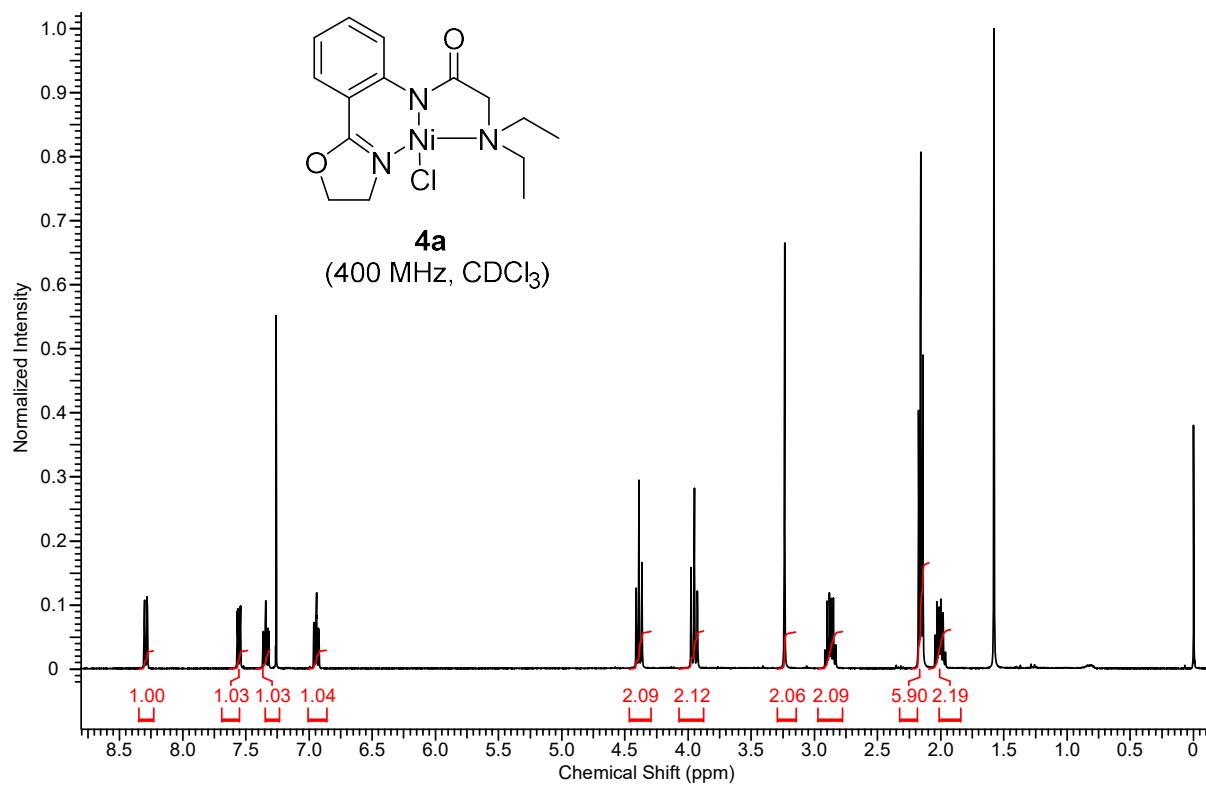


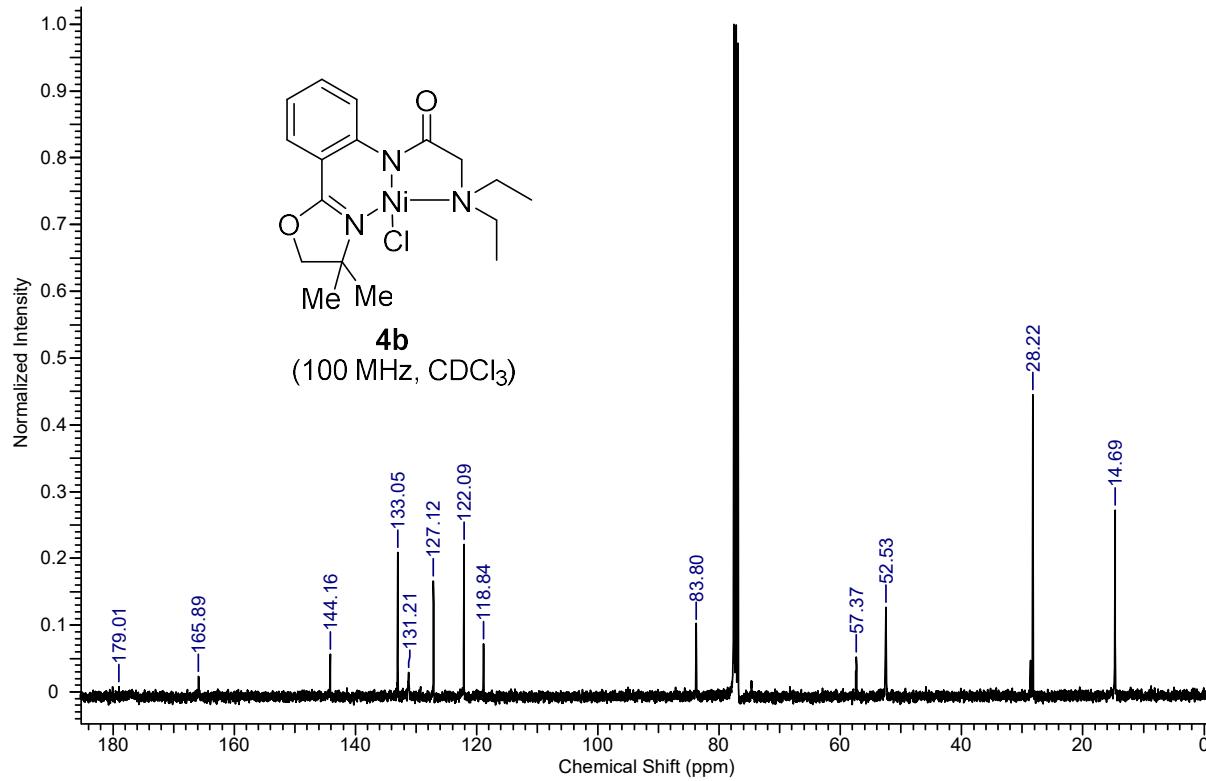
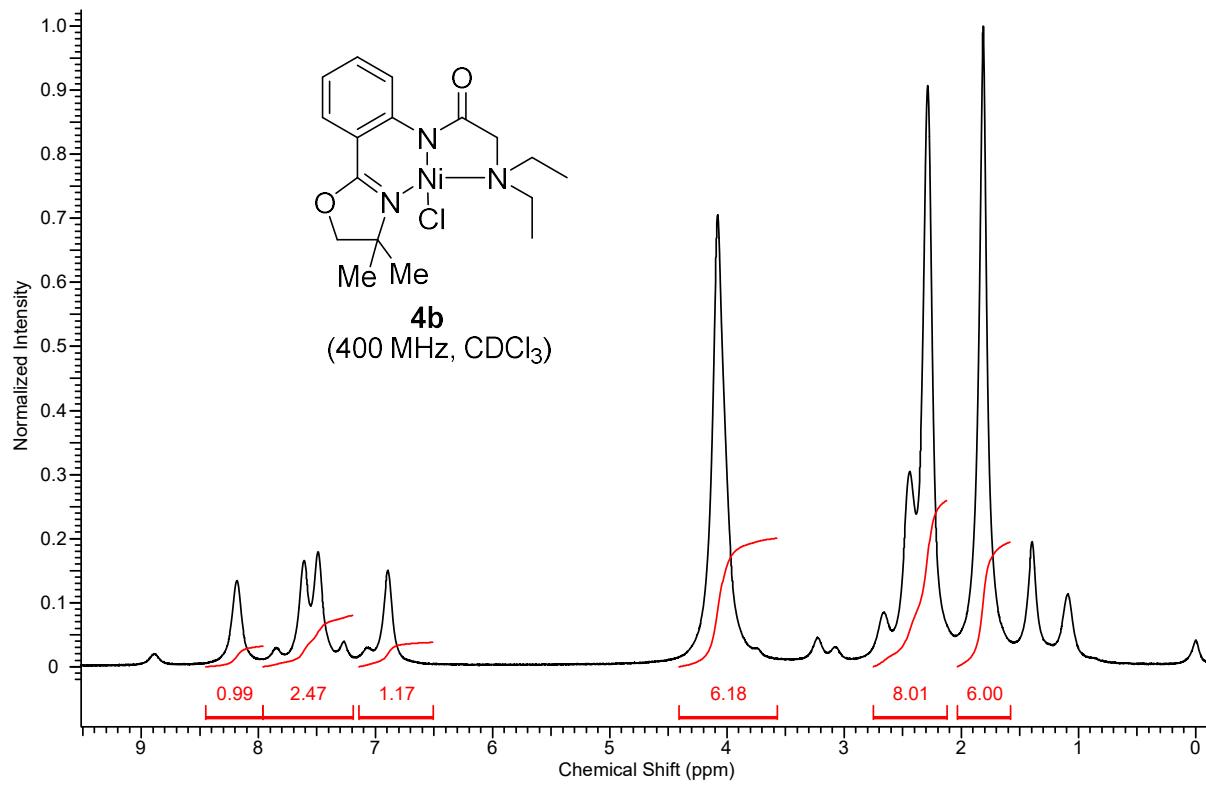


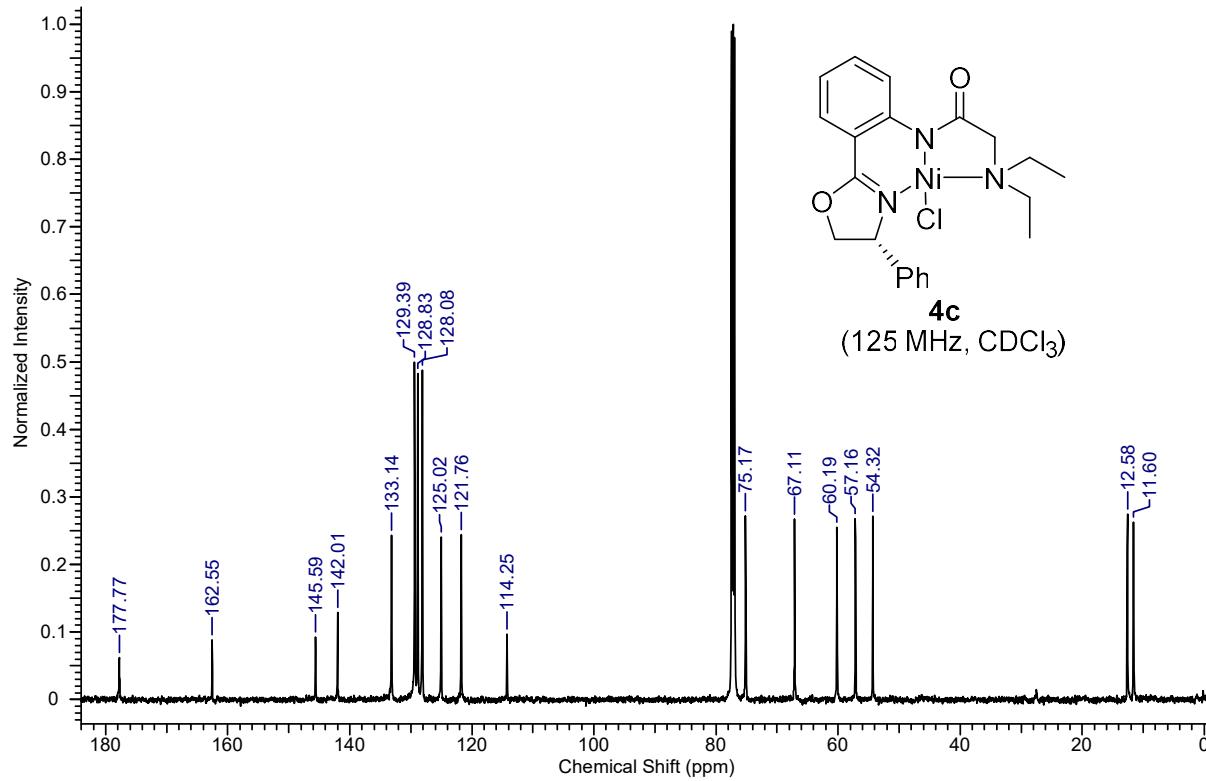
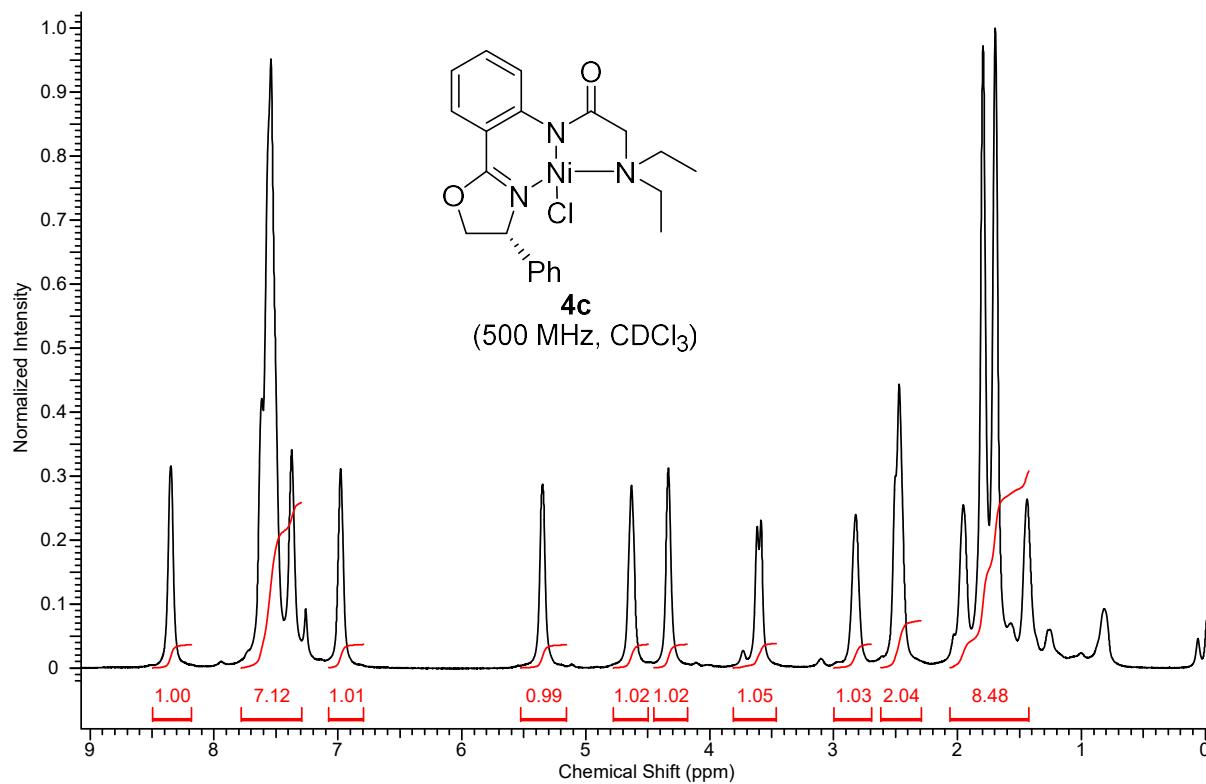


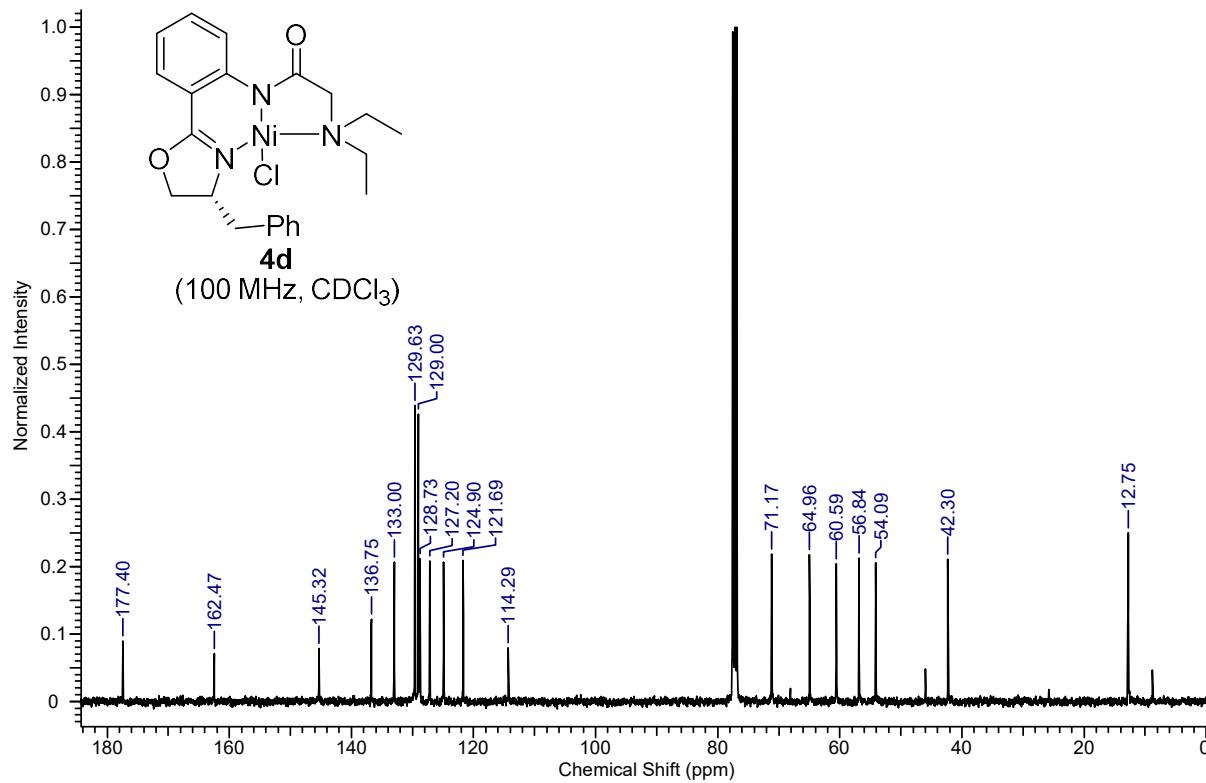
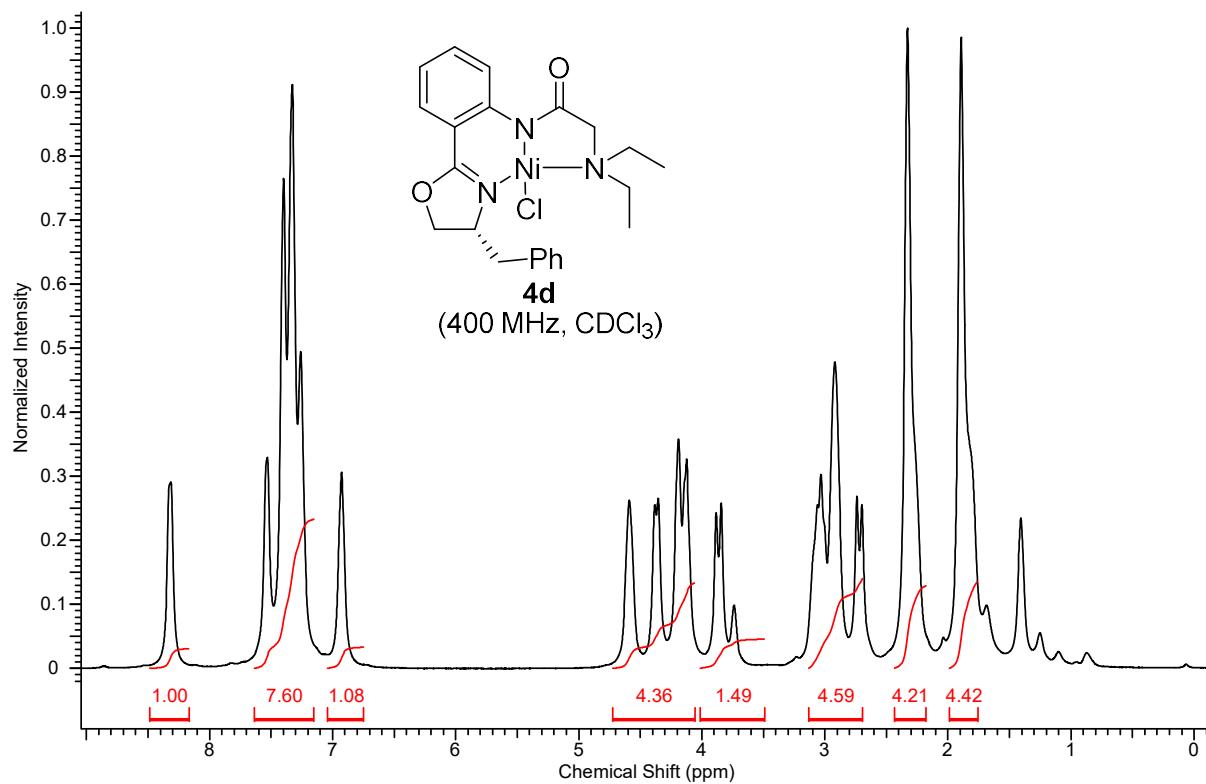


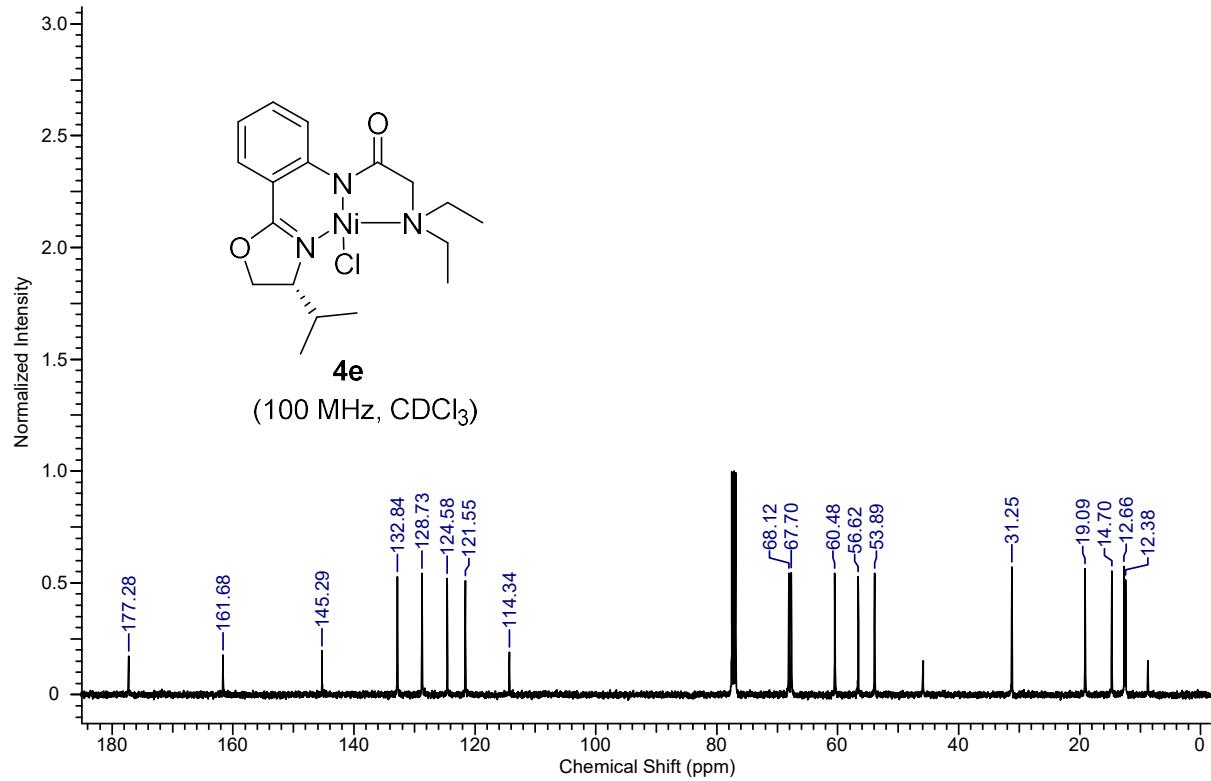
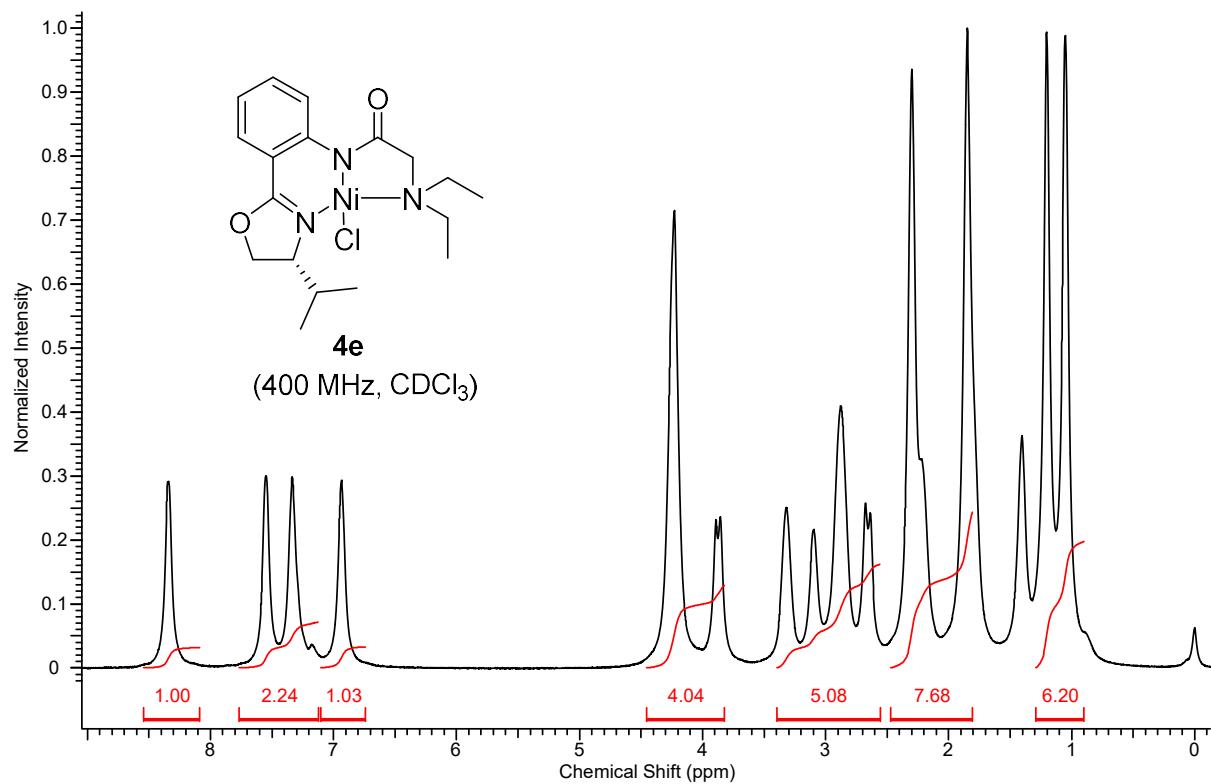
**5.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Spectra of Nickel Complexes**

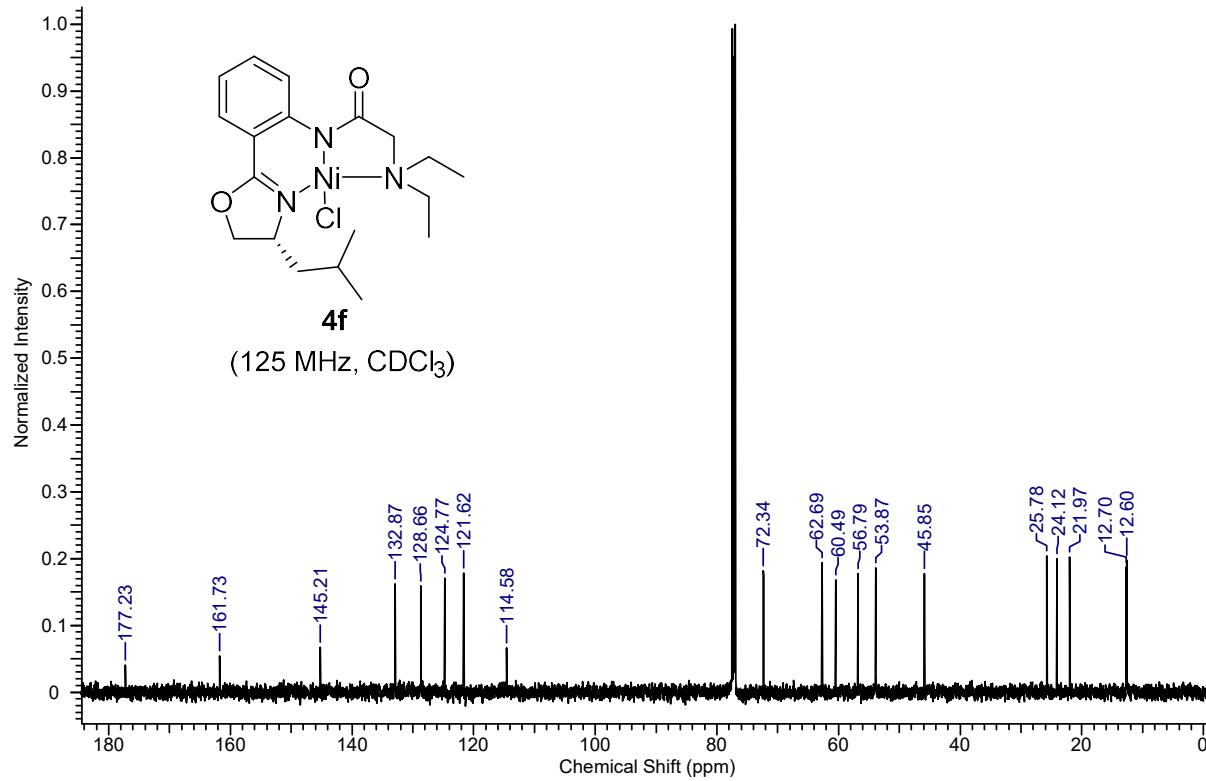
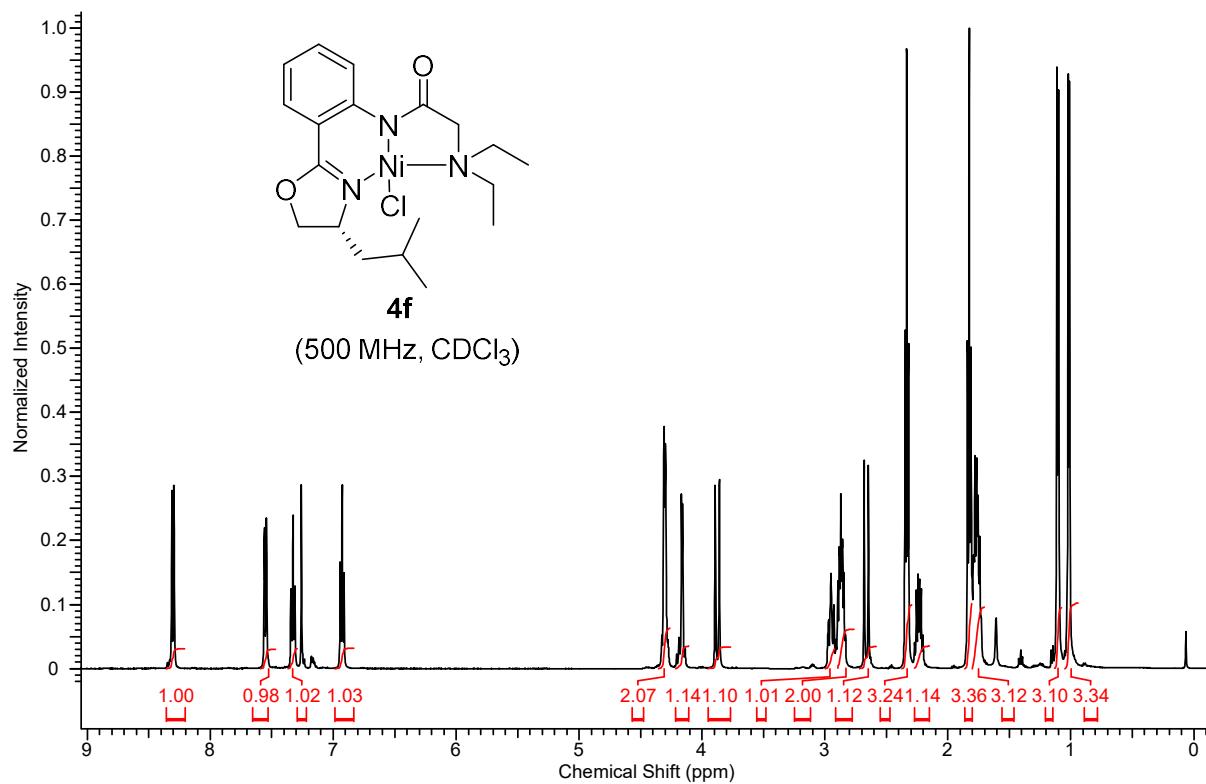












6.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Spectra of 6a

