

Supporting Information

Reactivity and mechanisms of hydridic hydrogen of B-H in ammonia borane towards acetic acids: the ammonia B-monoacyloxy boranes

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1. Supporting Results

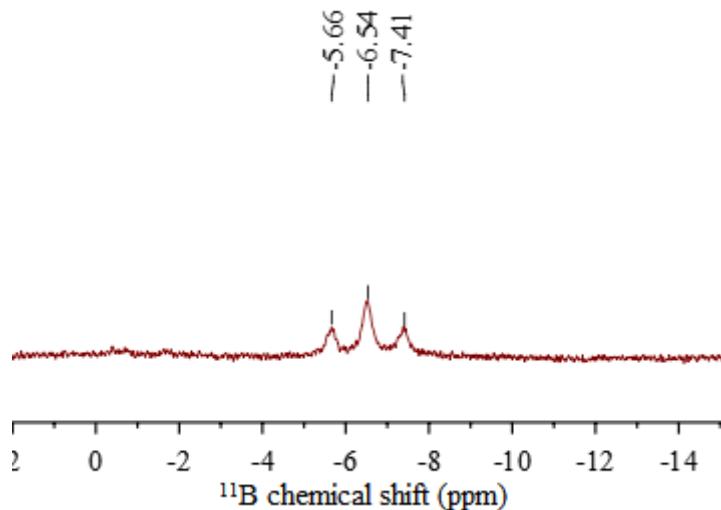


Figure S1a. ^{11}B NMR spectrum of $\text{NH}_3\text{BH}_2\text{OOCCH}_3$ in CD_3CN .

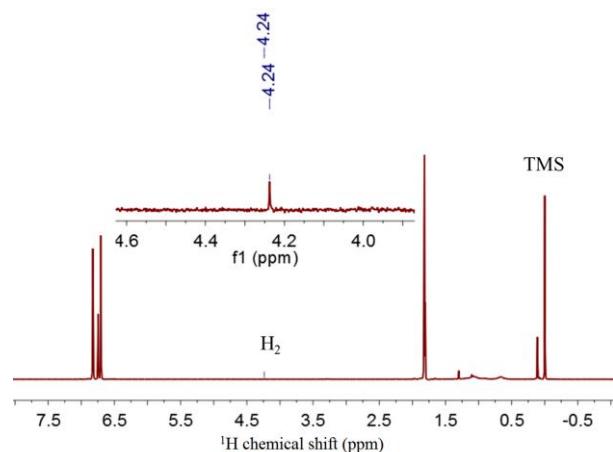


Figure S1b. ^1H NMR spectrum of the product H_2 in d_8 -Toluene.

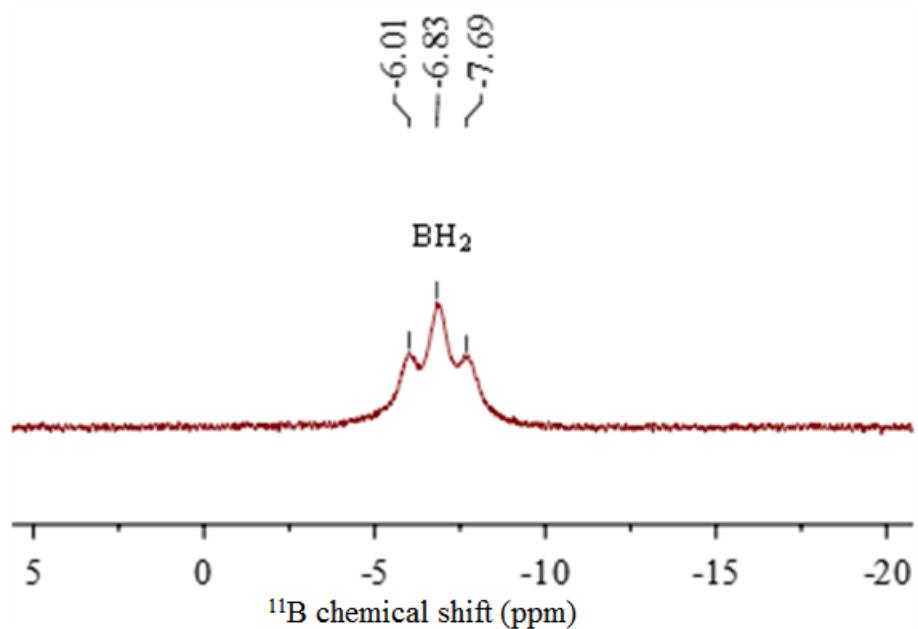


Figure S2a. ¹¹B NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**) in CD_3CN .

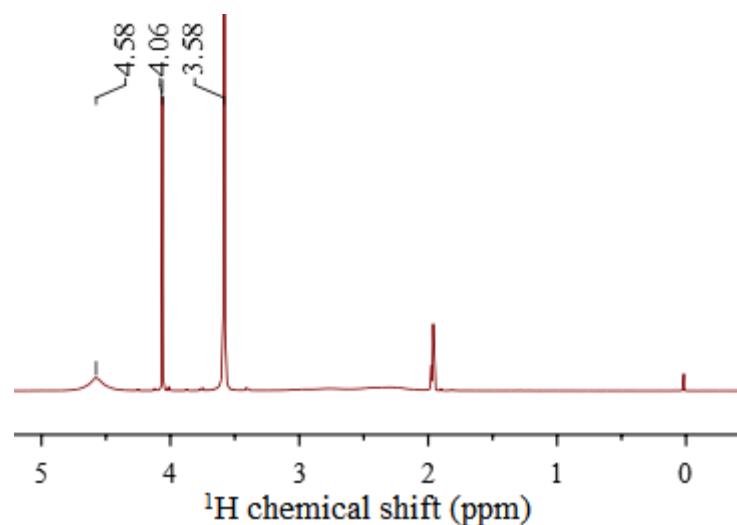


Figure S2b. ¹H NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**) in CD_3CN .

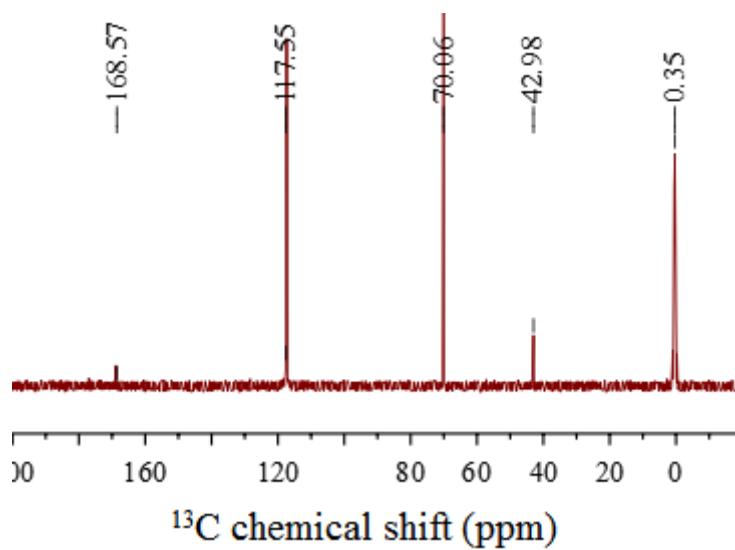


Figure S2c. ¹³C NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**) in CD_3CN .

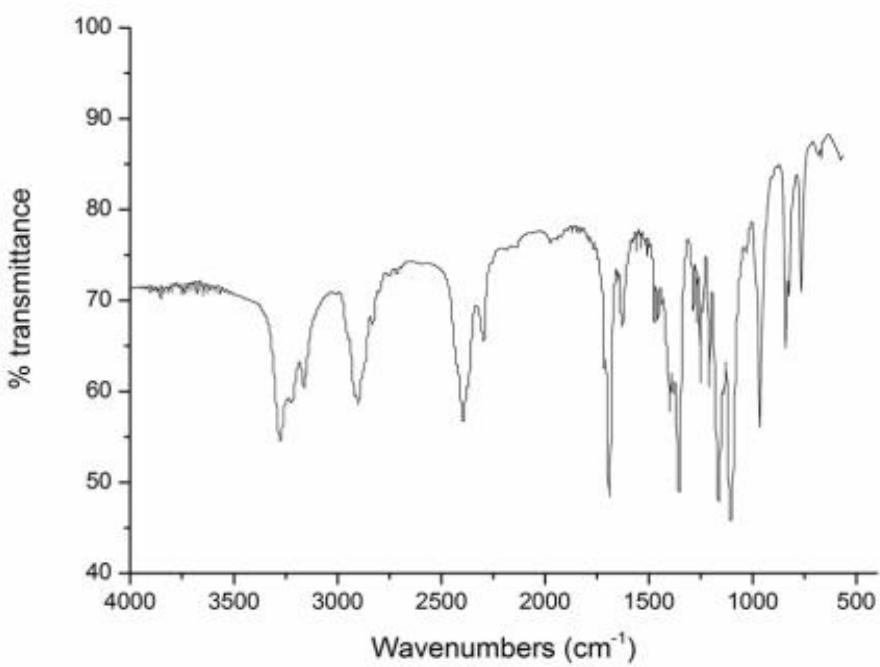


Figure S2d. FTIR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**).

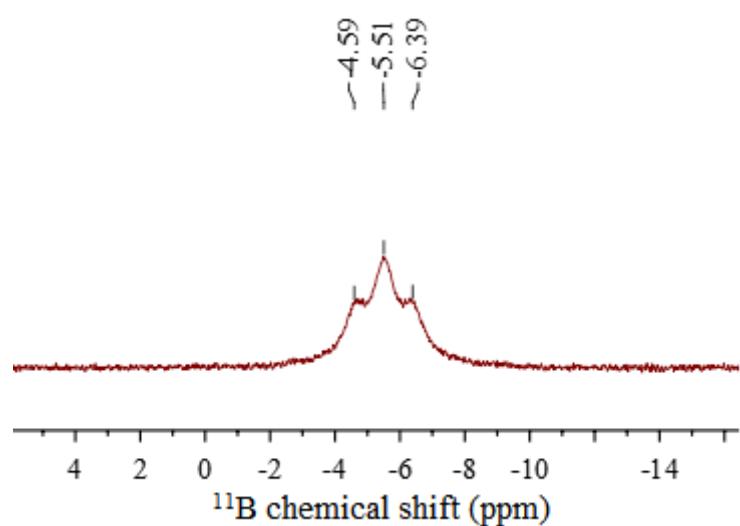


Figure S3a. ¹¹B NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2 \text{C}_{12}\text{H}_{24}\text{O}_6$ (**2**) in CD_3CN .

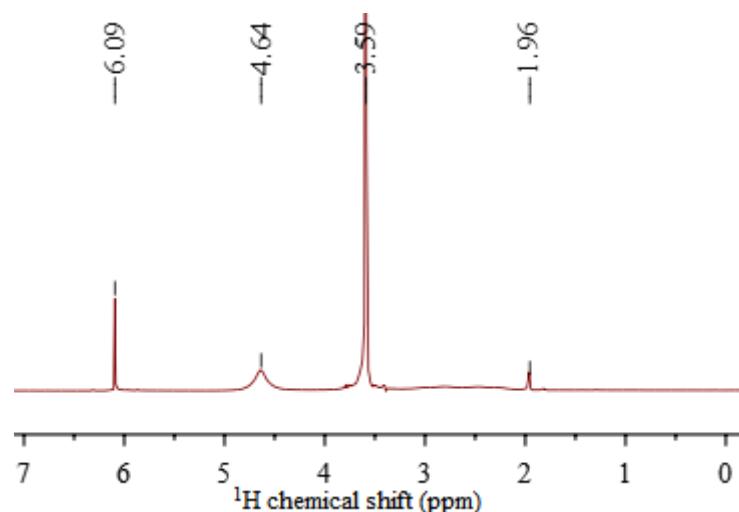


Figure S3b. ¹H NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2 \text{C}_{12}\text{H}_{24}\text{O}_6$ (**2**) in CD_3CN .

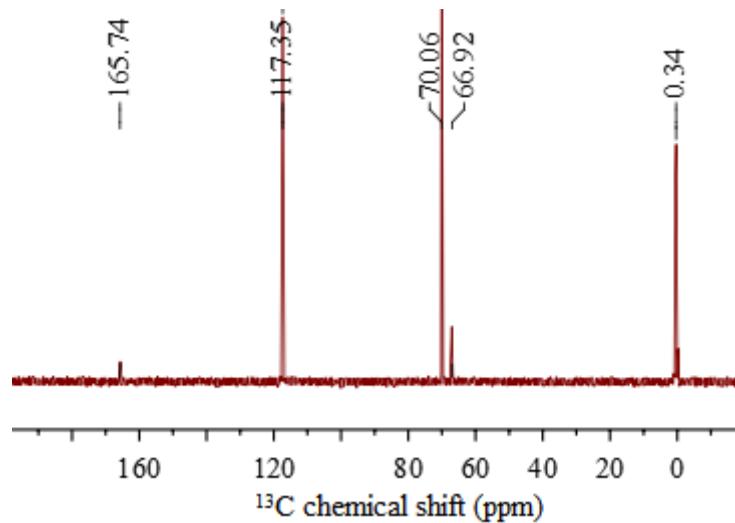


Figure S3c. ¹³C NMR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2 \text{ C}_{12}\text{H}_{24}\text{O}_6$ (**2**) in CD_3CN .

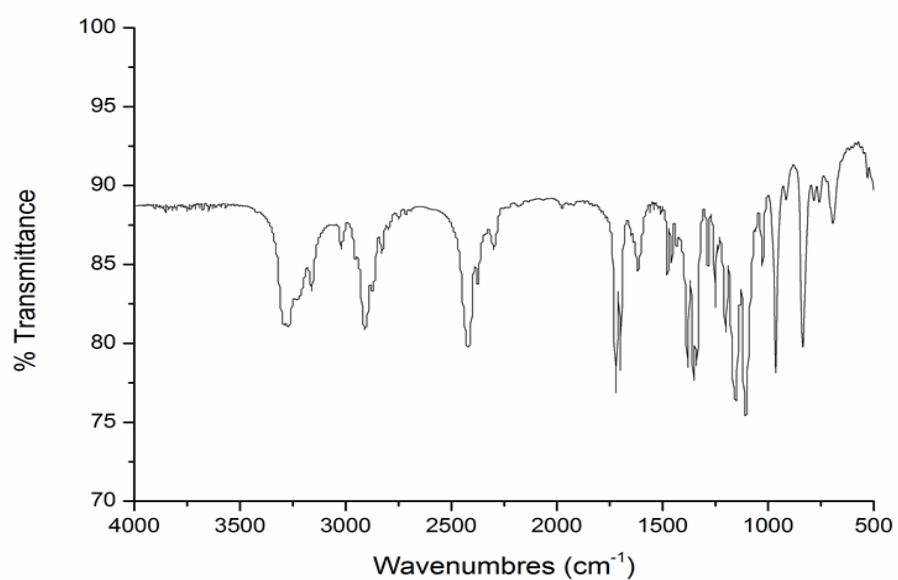


Figure S3d. FTIR spectrum of $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2 \text{ C}_{12}\text{H}_{24}\text{O}_6$ (**2**).

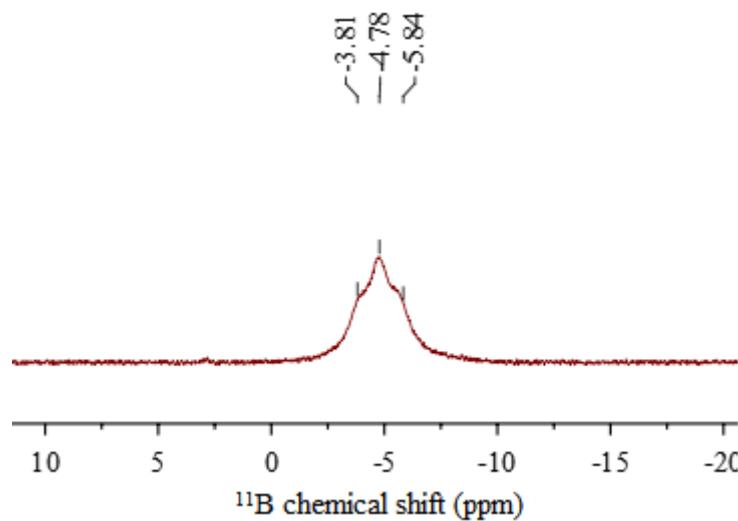


Figure S4. ^{11}B NMR spectrum of $\text{NH}_3\text{BH}_2\text{OOCCCl}_3$ in CD_3CN .

Table S1. The chemical shifts of B in the ammonia B-monoacyloxy boranes.

| Ammonia boranes | B-monoacyloxy | $\delta (\text{BH}_2)$, ppm |
|---|---------------|------------------------------|
| $\text{NH}_3\text{BH}_2\text{OOCCH}_3$ | | -6.5 |
| $\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ | | -5.7 |
| $\text{NH}_3\text{BH}_2\text{OOCCHCl}_2$ | | -5.5 |
| $\text{NH}_3\text{BH}_2\text{OOCCCl}_3$ | | -4.8 |

Table S2. Crystallographic data for $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**) and $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**2**).

| Complexes | $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}$ O_6 | $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ |
|--|---|--|
| Chemical formula | $\text{C}_8\text{H}_{19}\text{BClNO}_5$ | $\text{C}_8\text{H}_{18}\text{BCl}_2\text{NO}_5$ |
| Fw (g/mol) | 255.50 | 289.94 |
| T (K) | 103.0 (2) | 153.0 (2) |
| Wavelength (Å) | 0.71073 | 0.71073 |
| cryst syst | Monoclinic | Monoclinic |
| Space group | $\text{P}2_1/\text{n}$ | $\text{P}2_1/\text{n}$ |
| a , (Å) | 10.312 (2) | 8.616 (1) |
| b , (Å) | 12.187 (2) | 11.914 (2) |
| c , (Å) | 10.604 (2) | 13.693 (2) |
| α , (deg) | 90 | 90 |
| β , (deg) | 100.686(6) | 99.807(3) |
| γ , (deg) | 90 | 90 |
| V , (Å ³) | 1309.6(4) | 1385.1(3) |
| Z | 4 | 4 |
| D calc, (g/cm ³) | 1.296 | 1.390 |
| μ (mm ⁻¹) | 0.296 | 0.476 |
| F(000) | 544.0 | 608.0 |
| 2θ range, (deg) | 6.06-62.16 | 5.90-55.00 |
| reflns collected | 21063 | 13711 |
| indep reflns/ R_{int} | 4181/0.1584 | 3186/0.0703 |
| Data/restraints/parameters | 4181/0/145 | 3186/694/313 |
| GOF on F^2 | 0.983 | 0.907 |
| Coverage of independent reflections | 99.1% | 99.9% |
| Refinement method | Full-matrix least-squares on F^2 | Full-matrix least-squares on F^2 |
| R_{1,wR_2} [$I > 2\sigma(I)$] ^[a] | 0.0842, 0.1926 | 0.0557, 0.1271 |
| R_{1,wR_2} (all data) ^[a] | 0.1320, 0.2281 | 0.1074, 0.1465 |
| Largest diff. peak/hole (eÅ ⁻³) | 0.69/-0.64 | 0.467/-0.467 |

^[a] $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR_2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$

Table S3. Bond lengths (Å) and bond angles (°) of $2\text{NH}_3\text{BH}_2\text{OOCCH}_2\text{Cl}$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**1**).

| Bond lengths (Å) | Bond angles (°) | | |
|------------------|-----------------|-----------------|-----------|
| B(1)-N(1) | 1.581(3) | O(1)-B(1)-N(1) | 102.66(5) |
| B(1)-O(1) | 1.505(3) | C(1)-C(2)-Cl(1) | 117.5(2) |
| C(1)-C(2) | 1.505(4) | O(2)-C(1)-O(1) | 125.5(3) |
| C(2)-Cl(1) | 1.779(3) | O(2)-C(1)-C(2) | 116.9(2) |
| C(2)-O(1) | 1.303(3) | O(1)-C(1)-C(2) | 117.5(2) |
| C(1)-O(2) | 1.217(3) | C(1)-O(1)-B(1) | 118.1(2) |

Table S4. Bond lengths (Å) and bond angles (°) of $2\text{NH}_3\text{BH}_2\text{OOCCHCl}_2$ $\text{C}_{12}\text{H}_{24}\text{O}_6$ (**2**).

| Bond lengths (Å) | Bond angles (°) | | |
|------------------|-----------------|------------------|----------|
| B(1)-N(1) | 1.590(7) | O(1)-B(1)-N(1) | 109.1(5) |
| B(1)-O(1) | 1.513(7) | C(1)-C(2)-Cl(2) | 109.0(6) |
| C(1)-C(2) | 1.536(9) | C(1)-C(2)-Cl(1) | 110.1(5) |
| C(2)-Cl(2) | 1.724(7) | Cl(2)-C(2)-Cl(1) | 113.2(4) |
| C(2)-Cl(1) | 1.750(7) | O(2)-C(1)-O(1) | 126.9(7) |
| C(1)-O(2) | 1.205(9) | O(2)-C(1)-C(2) | 123.6(7) |
| C(1)-O(1) | 1.304(8) | O(1)-C(1)-C(2) | 109.5(7) |
| | | C(1)-O(1)-B(1) | 120.2(5) |

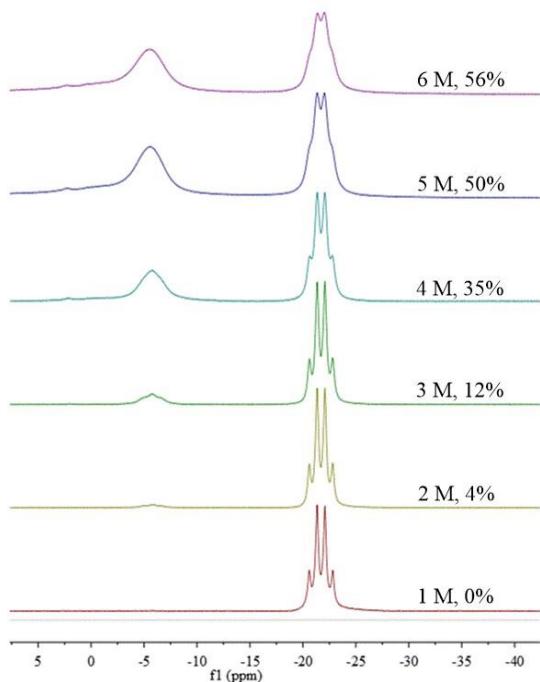


Figure S5. ^{11}B NMR spectra of the reaction mixture of AB with CH_3COOH with different concentration in 14 h ($c_{\text{AB}} = 1\text{--}6 \text{ M}$, the conversion of AB: %).

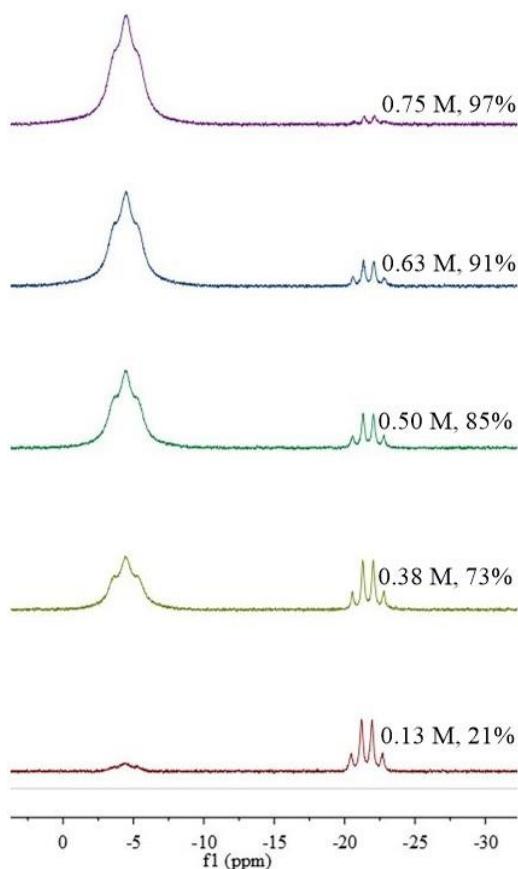


Figure S6. ^{11}B NMR spectra of the reaction mixture of AB with CCl_3COOH at a ratio of 1:1 ($c_{\text{AB}} = 0.13\text{--}0.75 \text{ M}$, the conversion of AB: %).

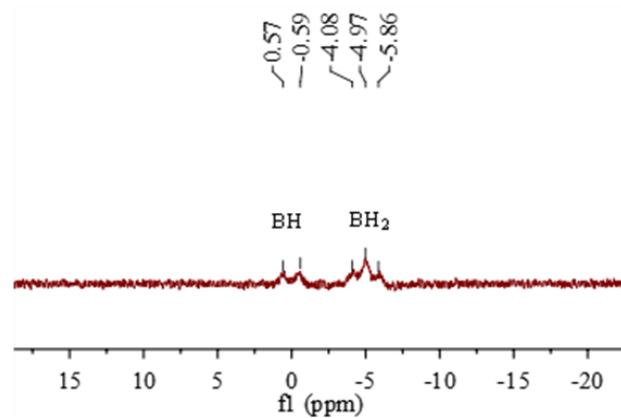


Figure S7. ¹¹B NMR spectrum of the reaction mixture of AB with excess ClCH₂COOH over a period of 10 d.

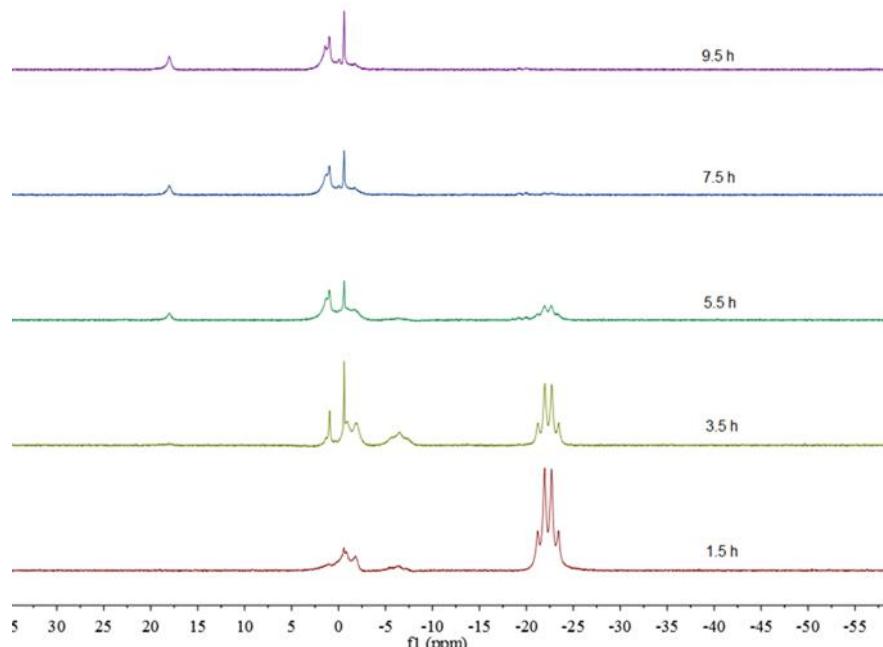


Figure S8. ¹¹B NMR spectra of the reaction of AB with CH₃COOH at 70 °C (n_{AB} : n_{acid} to 1:1, $c_{AB} = 3 \text{ M}$).

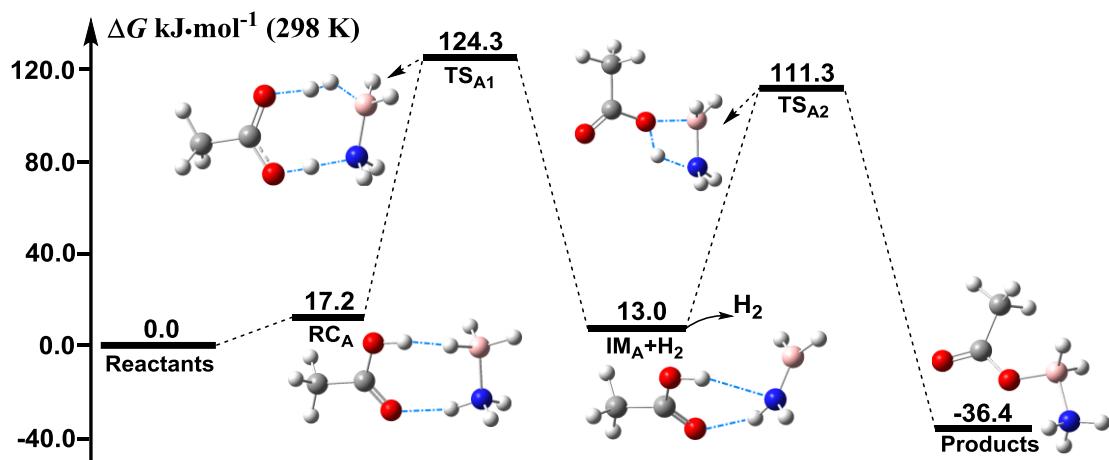


Figure S9a. Energy profile of the reaction of AB with CH_3COOH .

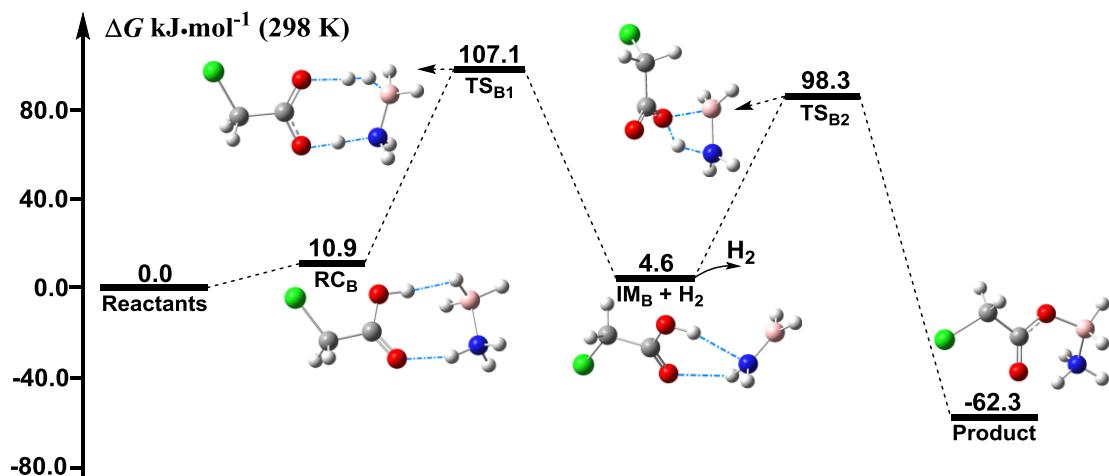


Figure S9b. Energy profile of the reaction of AB with CH_2ClCOOH .

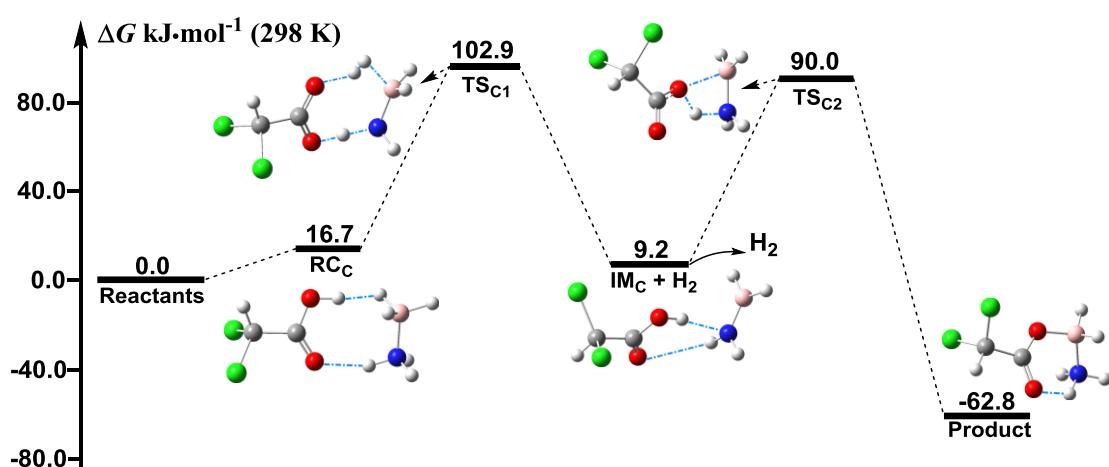


Figure S9c. Energy profile of the reaction of AB with CHCl₂COOH.

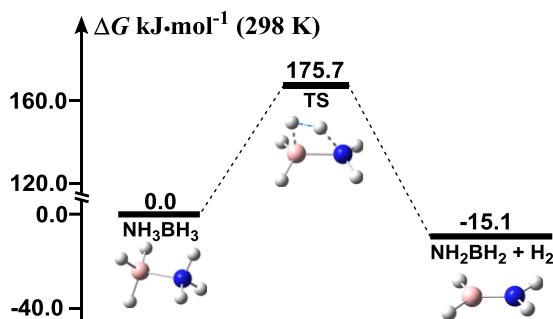


Figure S10. Energy profile of the decomposition of AB to produce hydrogen and NH_2BH_2 .

Table S5. The relative energies $\{\Delta(E+\text{ZPE})/(\text{kJ}\cdot\text{mol}^{-1})\}$, enthalpies $\{\Delta H(298\text{ K})/(\text{kJ}\cdot\text{mol}^{-1})\}$, free energy changes $\{\Delta G(298\text{ K})/(\text{kJ}\cdot\text{mol}^{-1})\}$ and entropy $\{S(298\text{ K})(\text{J}\cdot\text{mol}^{-1})\}$ for the reactions of AB with CH_3COOH .

| Species | ΔE | $\Delta(E + \text{ZPE})$ | ΔH | ΔG | S |
|--------------------------------------|------------|--------------------------|------------|------------|-------|
| $\text{AB} + \text{CH}_3\text{COOH}$ | 0.0 | 0.0 | 0.0 | 0.0 | 538.1 |
| RC_A | -32.2 | -25.9 | -26.4 | 17.2 | 392.5 |
| TS_{A1} | 92.5 | 78.2 | 75.3 | 124.3 | 374.5 |
| $\text{IM}_A + \text{H}_2$ | 27.2 | 2.1 | 10.0 | 13.0 | 527.6 |
| TS_{A2} | 127.2 | 92.5 | 96.2 | 111.3 | 487.0 |
| Product | -36.8 | -54.8 | -50.2 | -36.4 | 491.6 |

Table S6. The relative energies $\{\Delta(E + \text{ZPE})/(\text{kJ}\cdot\text{mol}^{-1})\}$, enthalpies $\{\Delta H(298\text{ K})/(\text{kJ}\cdot\text{mol}^{-1})\}$, free energy changes $\{\Delta G(298\text{ K})/(\text{kJ}\cdot\text{mol}^{-1})\}$ and entropy $\{S(298\text{ K})(\text{J}\cdot\text{mol}^{-1})\}$ for the reactions of AB with CH_2ClCOOH .

| Species | ΔE | $\Delta(E + \text{ZPE})$ | ΔH | ΔG | S |
|--|------------|--------------------------|------------|------------|-------|
| $\text{AB} + \text{CH}_2\text{ClCOOH}$ | 0.0 | 0.0 | 0.0 | 0.0 | 545.2 |
| RC_B | -32.6 | -26.8 | -24.7 | 10.9 | 425.9 |
| TS_{B1} | 78.7 | 64.4 | 64.0 | 107.1 | 400.4 |
| $\text{IM}_B + \text{H}_2$ | 23.4 | -2.5 | 8.0 | 4.6 | 556.1 |
| TS_{B2} | 119.2 | 83.7 | 90.8 | 98.3 | 519.7 |
| Product | -61.9 | -79.1 | -72.8 | -62.3 | 510.4 |

Table S7. The relative energies $\{\Delta(E + ZPE)/(kJ \cdot mol^{-1})\}$, enthalpies $\{\Delta H(298 K)/(kJ \cdot mol^{-1})\}$, free energy changes $\{\Delta G(298 K)/(kJ \cdot mol^{-1})\}$ and entropy $\{S(298 K)(J \cdot mol^{-1})\}$ for the reactions of AB with $CHCl_2COOH$.

| Species | ΔE | $\Delta(E + ZPE)$ | ΔH | ΔG | S |
|-------------------|------------|-------------------|------------|------------|-------|
| AB + $CHCl_2COOH$ | 0.0 | 0.0 | 0.0 | 0.0 | 595.8 |
| RC_C | -33.5 | -27.2 | -27.2 | 16.7 | 448.1 |
| TS_{C1} | 70.7 | 56.5 | 54.0 | 102.9 | 431.8 |
| $IM_C + H_2$ | 23.4 | -2.1 | 5.9 | 9.2 | 585.3 |
| TS_{C2} | 108.0 | 72.0 | 77.4 | 90.0 | 554.0 |
| Product | -66.5 | -83.3 | -79.1 | -62.8 | 541.0 |

Table S8. The relative energies $\{\Delta(E + ZPE)/(kJ \cdot mol^{-1})\}$, enthalpies $\{\Delta H(298 K)/(kJ \cdot mol^{-1})\}$, free energy changes $\{\Delta G(298 K)/(kJ \cdot mol^{-1})\}$ and entropy $\{S(298 K)(J \cdot mol^{-1})\}$ for the reactions of AB with CCl_3COOH .

| Species | ΔE | $\Delta(E + ZPE)$ | ΔH | ΔG | S |
|------------------|------------|-------------------|------------|------------|-------|
| AB + CCl_3COOH | 0.0 | 0.0 | 0.0 | 0.0 | 619.2 |
| RC_D | -34.3 | -28.5 | -28.5 | 14.6 | 474.0 |
| TS_{D1} | 64.0 | 49.4 | 47.3 | 94.6 | 459.8 |
| $IM_D + H_2$ | 23.4 | -3.4 | 5.4 | 5.4 | 618.0 |
| TS_{D2} | 99.6 | 63.6 | 69.0 | 80.3 | 580.7 |
| Product | -69.5 | -86.6 | -82.0 | -66.5 | 567.4 |

Table S9. The relative energies $\{\Delta(E + ZPE)/(kJ \cdot mol^{-1})\}$, enthalpies $\{\Delta H(298 K)/(kJ \cdot mol^{-1})\}$, free energy changes $\{\Delta G(298 K)/(kJ \cdot mol^{-1})\}$ and entropy $\{S(298 K) (J \cdot mol^{-1})\}$ for the reactions of AB release hydrogen without catalyst.

| Species | ΔE | $\Delta(E + ZPE)$ | ΔH | ΔG | S |
|------------------|------------|-------------------|------------|------------|-------|
| AB | 0.0 | 0.0 | 0.0 | 0.0 | 248.5 |
| TS | 192.5 | 175.3 | 174.9 | 175.7 | 245.2 |
| $NH_2BH_2 + H_2$ | 41.4 | 10.5 | 17.6 | -15.1 | 358.6 |

Table S10. The total rate constants for the reactions of AB with acids.

| T/K | k_r | k_1 | k_2 | k_3 | k_4 |
|-----|----------|----------|----------|----------|----------|
| 258 | 1.95E-20 | 8.70E-31 | 2.94E-27 | 1.54E-26 | 6.03E-25 |
| 278 | 1.02E-18 | 1.09E-29 | 2.55E-26 | 9.65E-26 | 3.05E-24 |
| 298 | 4.79E-17 | 1.00E-28 | 1.70E-25 | 4.83E-25 | 1.27E-23 |
| 318 | 1.84E-15 | 7.08E-28 | 9.09E-25 | 2.01E-24 | 4.47E-23 |
| 338 | 5.52E-14 | 4.02E-27 | 4.06E-24 | 7.16E-24 | 1.38E-22 |
| 358 | 1.26E-12 | 1.91E-26 | 1.56E-23 | 2.24E-23 | 3.79E-22 |
| 378 | 2.22E-11 | 7.75E-26 | 5.26E-23 | 6.29E-23 | 9.46E-22 |
| 398 | 3.05E-10 | 2.76E-25 | 1.59E-22 | 1.61E-22 | 2.18E-21 |

The r is the decomposition of AB into AoB and H_2 , and the 1-4 are the reactions of AB react with CH_3COOH , $CH_2ClCOOH$, $CHCl_2COOH$ and CCl_3COOH , respectively.

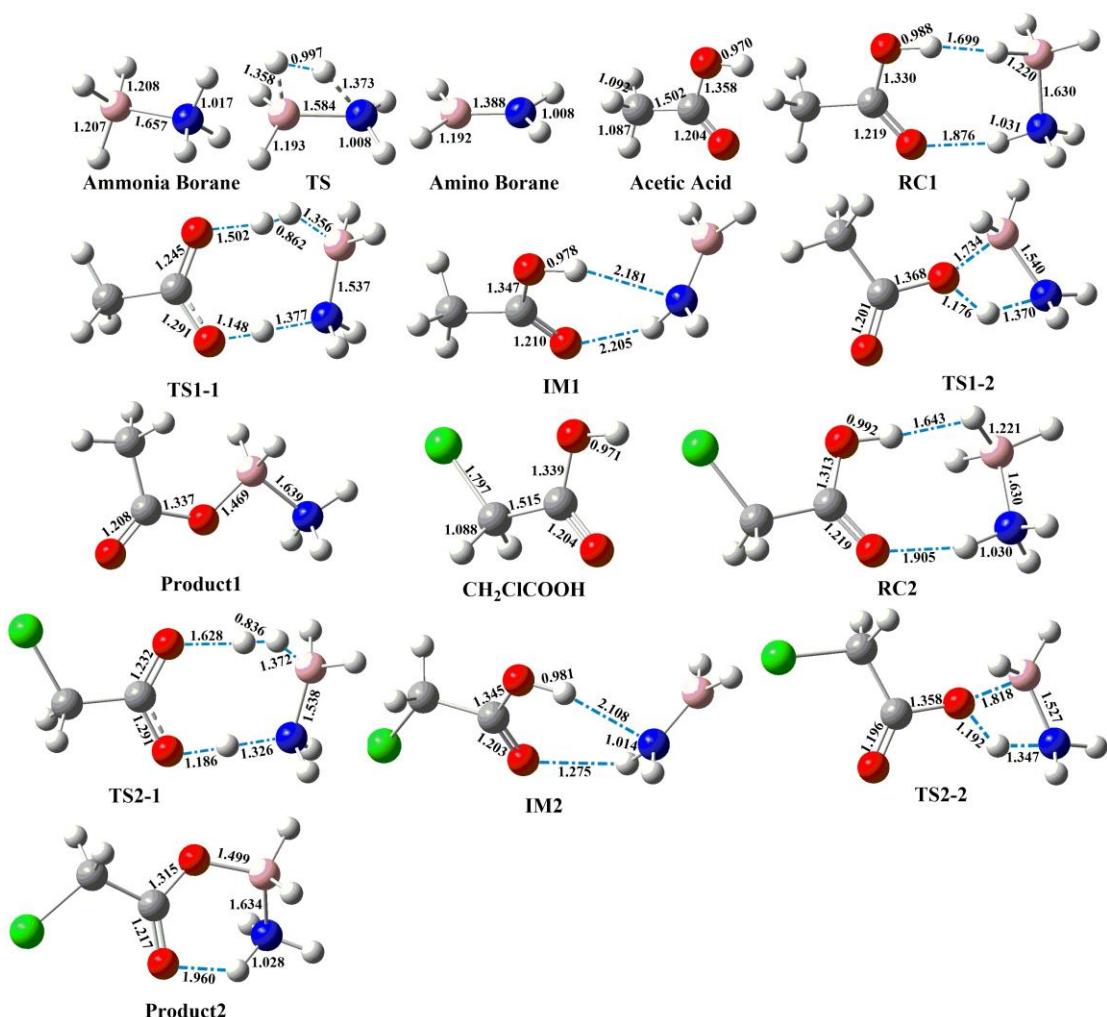


Figure S11a. The optimized geometry constructions of the reactions for AB, AB with acetic acid and AB with CH₂ClCOOH.

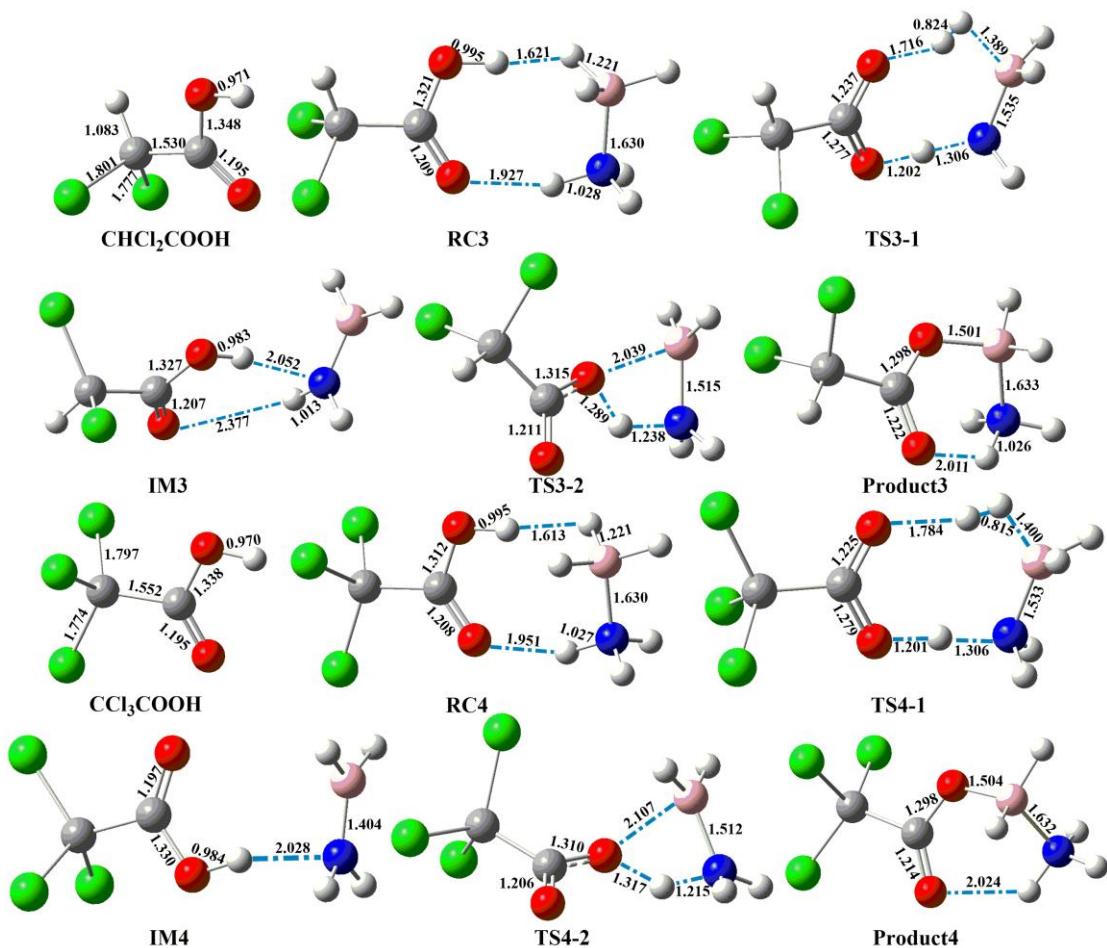


Figure S11b. The optimized geometry constructions of the reactions for AB with CHCl_2COOH and AB with CCl_3COOH .

2. The optimized cartesian coordinates of all species

| Ammonia Borane | | | | CH₃COOH | | | |
|-------------------------------------|-------------|-------------|-------------|---------------------------|-------------|-------------|-------------|
| N | 0.72745900 | 0.00001600 | 0.00000000 | C | -0.08853400 | 0.12698900 | -0.00000500 |
| H | 1.09314700 | -0.73111300 | -0.60538100 | O | -0.62190700 | 1.20624200 | -0.00000100 |
| H | 1.09313000 | -0.15863800 | 0.93587600 | O | -0.79628700 | -1.03164100 | -0.00000700 |
| B | -0.92948300 | 0.00009900 | 0.00016500 | H | -1.73486500 | -0.78603300 | 0.00005200 |
| H | -1.24146900 | 0.89752800 | 0.74500600 | C | 1.39216500 | -0.12774300 | -0.00000200 |
| H | -1.24116700 | -1.09462100 | 0.40439100 | H | 1.66707300 | -0.71260100 | 0.87978300 |
| H | 1.09300100 | 0.88978400 | -0.33066000 | H | 1.66702700 | -0.71310300 | -0.87946400 |
| H | -1.24144000 | 0.19645100 | -1.15006000 | H | 1.92453300 | 0.81945000 | -0.00026600 |
| TS | | | | RC1 | | | |
| N | 0.76209200 | 0.00000000 | 0.01356800 | N | 2.68138300 | -0.78397500 | 0.10922600 |
| H | 1.26116400 | 0.83823000 | -0.24032500 | H | 1.73157100 | -1.12874800 | -0.03357400 |
| H | 1.26116400 | -0.83823000 | -0.24032600 | H | 2.95418200 | -1.00655000 | 1.06361000 |
| B | -0.81291400 | 0.00000000 | -0.15199500 | B | 2.75987200 | 0.83795100 | -0.17438200 |
| H | -1.26869200 | -1.03187000 | -0.53981700 | H | 3.88809600 | 1.17018300 | 0.08602300 |
| H | -1.26869200 | 1.03187000 | -0.53981600 | H | 1.95409600 | 1.33679400 | 0.58637500 |
| H | -0.13228800 | 0.00000000 | 1.05476100 | H | 3.30298300 | -1.28965000 | -0.51693600 |
| H | -1.12273100 | 0.00000000 | 1.17052200 | H | 2.46512400 | 0.97681400 | -1.33530500 |
| | | | | C | -1.55343500 | 0.19222000 | 0.02000500 |
| NH₂BH₂ | | | | O | -1.94386200 | 1.32639200 | 0.09358700 |
| N | 0.00000000 | 0.00000000 | 0.61065500 | O | -0.23372500 | -0.12923400 | 0.03787200 |
| H | 0.00000000 | 0.84353800 | 1.16284600 | H | 0.29703900 | 0.69266200 | 0.10351500 |
| H | 0.00000000 | -0.84353800 | 1.16284600 | C | -2.42272900 | -1.02975300 | -0.10227200 |
| B | 0.00000000 | 0.00000000 | -0.77775100 | H | -2.20523800 | -1.54613700 | -1.03958000 |
| H | 0.00000000 | -1.04237500 | -1.35576100 | H | -2.21083000 | -1.72411200 | 0.71300800 |
| H | 0.00000000 | 1.04237500 | -1.35576100 | H | -3.46838300 | -0.73525100 | -0.07788300 |

| TS1-1 | | | | H | -3.20565300 | 0.78551800 | 0.29329700 |
|--------------|-------------|-------------|-------------|-----------------|-------------|-------------|-------------|
| N | -2.01671600 | -0.78732300 | 0.01456500 | H | -3.06657100 | -0.64595600 | -0.76721100 |
| H | -0.65027100 | -0.94633200 | -0.05594100 | H | -2.87693100 | -0.83239600 | 0.97131900 |
| H | -2.44975200 | -1.26564200 | -0.76978600 | | | | |
| B | -2.41464200 | 0.69436900 | 0.11322100 | TS1-2 | | | |
| H | -3.39657700 | 1.02419700 | -0.49288100 | N | 2.31230500 | -0.37445500 | 0.15100800 |
| H | -1.64260000 | 1.42640700 | -0.72822100 | H | 1.09813500 | -0.74144000 | -0.36750800 |
| H | -2.30287400 | -1.28600900 | 0.85152100 | H | 3.10640100 | -0.41364500 | -0.47520800 |
| H | -2.30817300 | 1.14565100 | 1.21370400 | B | 1.46954200 | 0.91503400 | 0.15500700 |
| C | 1.12027300 | 0.05749900 | -0.02427400 | H | 1.78389500 | 1.74571100 | -0.63802300 |
| O | 0.49138500 | -1.06805000 | -0.08232000 | H | 2.55746200 | -0.77574300 | 1.04712600 |
| O | 0.56539900 | 1.17230000 | -0.03963900 | H | 1.06996400 | 1.25253300 | 1.22821700 |
| H | -0.91101000 | 1.26396400 | -0.30158300 | C | -1.03883200 | -0.26316600 | -0.07168100 |
| C | 2.61966200 | -0.04160500 | 0.07248600 | O | 0.20315900 | -0.01037900 | -0.58539200 |
| H | 3.07044800 | 0.94737900 | 0.05136700 | O | -1.39773700 | -1.38612100 | 0.15679300 |
| H | 2.88669900 | -0.55094900 | 1.00073500 | C | -1.86493200 | 0.97732000 | 0.12226600 |
| H | 3.00045100 | -0.64860400 | -0.75057500 | H | -2.85342100 | 0.70637100 | 0.48425100 |
| | | | | H | -1.37271700 | 1.64026600 | 0.83658100 |
| IM1 | | | | H | -1.94436100 | 1.51904200 | -0.82225500 |
| N | 2.25236600 | 0.34993000 | -0.39747200 | | | | |
| H | 0.41929300 | -0.82663600 | -0.29692400 | Product1 | | | |
| H | 2.45821900 | 0.36961800 | -1.38591100 | N | 2.57372300 | -0.23754100 | 0.00031800 |
| B | 3.06260200 | -0.32029100 | 0.51741200 | H | 2.59194100 | -0.83992200 | 0.82198000 |
| H | 3.96946100 | -0.97970900 | 0.11369300 | H | 2.59273900 | -0.84001000 | -0.82125200 |
| H | 1.50746800 | 0.98589900 | -0.13231500 | B | 1.20619900 | 0.66505700 | -0.00021500 |
| H | 2.80904100 | -0.23679700 | 1.67853100 | H | 1.24822200 | 1.32264900 | -1.01365000 |
| C | -1.21392600 | 0.08746300 | 0.00314800 | H | 3.40921900 | 0.34274600 | 0.00067500 |
| O | -0.53125400 | -1.04903800 | -0.23675000 | H | 1.24741000 | 1.32244100 | 1.01337800 |
| O | -0.67732200 | 1.16821400 | 0.09936300 | C | -1.12698300 | -0.25498000 | 0.00004500 |
| C | -2.69028900 | -0.15763100 | 0.13349500 | O | 0.20148800 | -0.40698800 | -0.00070200 |

| | | | | | | | |
|-----------------------------|-------------|-------------|-------------|--------------|-------------|-------------|-------------|
| O | -1.85588800 | -1.21885900 | 0.00019200 | H | -1.79508100 | 1.20859800 | 1.19797100 |
| C | -1.64285200 | 1.16737100 | 0.00014200 | H | -1.86953400 | 1.64184600 | -0.51344800 |
| H | -2.72981000 | 1.15637600 | -0.00007900 | Cl | -2.93588800 | -0.42838400 | -0.07139600 |
| H | -1.27650400 | 1.70259800 | 0.87911700 | | | | |
| H | -1.27606600 | 1.70306200 | -0.87836300 | TS2-1 | | | |
| | | | | N | -3.04958800 | -0.47787600 | 0.01949600 |
| CH₂ClCOOH | | | | H | -1.79763000 | -0.91035000 | -0.04256800 |
| C | -1.04471400 | -0.06543500 | 0.00035600 | H | -3.56916900 | -0.85509100 | -0.76797100 |
| O | -2.10561900 | -0.63491400 | -0.00869500 | B | -3.13662600 | 1.05336500 | 0.12750700 |
| O | -0.90992900 | 1.26709400 | 0.00521400 | H | -4.03697100 | 1.58103900 | -0.46115900 |
| H | -1.80579800 | 1.64144000 | -0.00025800 | H | -2.27513800 | 1.62439400 | -0.77550400 |
| C | 0.25126600 | -0.84951300 | 0.01034700 | H | -3.43726100 | -0.90692500 | 0.85496600 |
| H | 0.25762300 | -1.47205300 | 0.90246700 | H | -2.90888900 | 1.47496200 | 1.21834300 |
| H | 0.25566900 | -1.49782600 | -0.86298300 | C | 0.17788800 | -0.32007100 | -0.03543200 |
| Cl | 1.77515100 | 0.10357000 | -0.00444700 | O | -0.67373000 | -1.28948400 | -0.07042200 |
| | | | | O | -0.09638000 | 0.88115100 | -0.05924000 |
| RC2 | | | | H | -1.63775500 | 1.34002000 | -0.31511500 |
| N | 3.41598700 | 0.45999000 | -0.10153800 | C | 1.60843100 | -0.82873800 | 0.04901500 |
| H | 2.51092500 | 0.94981900 | -0.08230200 | H | 1.72868900 | -1.38762700 | 0.97498900 |
| H | 3.85944800 | 0.64219100 | -0.99844100 | H | 1.79687900 | -1.50024400 | -0.78564200 |
| B | 3.16998800 | -1.13300800 | 0.13890900 | Cl | 2.85943800 | 0.45750900 | 0.01655600 |
| H | 4.22655500 | -1.69549300 | 0.01778700 | | | | |
| H | 2.39768300 | -1.47732700 | -0.74131200 | IM2 | | | |
| H | 4.01296100 | 0.85292200 | 0.62161600 | N | -3.14198700 | -0.52084700 | -0.39319700 |
| H | 2.69584600 | -1.23651000 | 1.24483300 | H | -1.56163800 | 0.86444800 | -0.22906000 |
| C | -0.24025900 | 0.38794900 | 0.01321000 | H | -3.38589600 | -0.45539800 | -1.37126500 |
| O | 0.62831500 | 1.24057400 | -0.05488800 | B | -3.99599800 | -0.07458300 | 0.61704700 |
| O | -0.04019200 | -0.90954300 | -0.01861500 | H | -4.98334400 | 0.52033900 | 0.31738600 |
| H | 0.92681300 | -1.11295300 | -0.10700800 | H | -2.33420800 | -1.11006300 | -0.22429200 |
| C | -1.68013200 | 0.82976900 | 0.18317000 | H | -3.69485900 | -0.27415400 | 1.75169100 |

| | | | | | | | |
|-----------------|-------------|-------------|-------------|-----------------------------|-------------|-------------|-------------|
| C | 0.22439900 | 0.24276900 | -0.08214300 | H | 2.84938200 | 0.79133300 | 1.34925700 |
| O | -0.66150600 | 1.25046100 | -0.17211500 | H | 3.16706100 | -1.49954700 | 0.23668100 |
| O | -0.07087700 | -0.92301100 | -0.09338300 | C | 0.04960300 | -0.13530000 | -0.11141200 |
| C | 1.62255200 | 0.81801700 | 0.03679300 | O | 1.02550400 | -0.98321500 | 0.13009200 |
| H | 1.81911100 | 1.46597900 | -0.81528300 | O | 0.18612800 | 1.03470300 | -0.41816000 |
| H | 1.68517100 | 1.41590500 | 0.94453400 | C | -1.29467600 | -0.82791500 | 0.04393600 |
| Cl | 2.89452500 | -0.43485500 | 0.09938300 | H | -1.43851900 | -1.50226000 | -0.79925600 |
| | | | | H | -1.30208300 | -1.41317300 | 0.95983100 |
| TS2-2 | | | | Cl | -2.68832400 | 0.29976300 | 0.08692800 |
| N | 3.19229900 | 0.16482900 | -0.33921800 | | | | |
| H | 2.12281100 | 0.60975500 | 0.34734300 | CHCl₂COOH | | | |
| H | 3.33799700 | 0.61057300 | -1.23632200 | C | 1.14646000 | -0.32213100 | 0.07149500 |
| B | 2.26361300 | -1.04632800 | -0.28711700 | O | 1.35268800 | -0.93432400 | 1.07672800 |
| H | 1.66180100 | -1.28271300 | -1.28718600 | O | 2.10371800 | 0.15305400 | -0.75034600 |
| H | 4.06174700 | 0.11688600 | 0.17671400 | H | 2.96259600 | -0.06400200 | -0.35300400 |
| H | 2.57196700 | -1.92016700 | 0.45567700 | C | -0.22738300 | 0.06251700 | -0.48000300 |
| C | -0.05622700 | 0.45458800 | 0.26779200 | H | -0.19082600 | 0.20967300 | -1.55263300 |
| O | 1.15371000 | 0.00880200 | 0.69302800 | Cl | -1.43230100 | -1.19810900 | -0.13863800 |
| O | -0.25663200 | 1.58806700 | -0.05853600 | Cl | -0.68166800 | 1.64882500 | 0.24132200 |
| C | -1.07350700 | -0.66966200 | 0.30983000 | | | | |
| H | -0.73104000 | -1.48903600 | -0.31978000 | RC3 | | | |
| H | -1.15887700 | -1.03346600 | 1.33245000 | N | -3.57885700 | 0.42165000 | -0.88448500 |
| Cl | -2.70169300 | -0.17755900 | -0.24708500 | H | -4.04929600 | 1.31995600 | -0.96000900 |
| | | | | H | -3.91900900 | -0.16755800 | -1.64056500 |
| Product2 | | | | B | -3.83740200 | -0.25661100 | 0.57424500 |
| N | 2.81244400 | 0.77713600 | 0.33213300 | H | -5.01064200 | -0.50300900 | 0.66900600 |
| H | 2.05994400 | 1.40531100 | 0.02177400 | H | -3.44820100 | 0.54281600 | 1.39135100 |
| H | 3.70771400 | 1.09986500 | -0.02545800 | H | -2.57048000 | 0.56909500 | -1.01993200 |
| B | 2.43971300 | -0.69368700 | -0.27381700 | H | -3.18424900 | -1.28835800 | 0.57133100 |
| H | 2.54971600 | -0.58162700 | -1.46705800 | C | -0.14315400 | -0.08234700 | 0.26151200 |

| | | | | | | | |
|--------------|-------------|-------------|-------------|-----------------|-------------|-------------|-------------|
| O | -0.67699300 | 0.44740400 | -0.68466000 | B | -4.28048900 | 0.60580000 | -0.54565700 |
| O | -0.74679600 | -0.74426900 | 1.23183000 | H | -4.17021600 | 1.68474100 | -0.05559500 |
| H | -1.72791800 | -0.78257100 | 1.07313700 | H | -4.93539000 | 0.41241700 | -1.52078300 |
| C | 1.37101400 | -0.08872700 | 0.48297700 | H | -1.67882600 | -0.11589000 | -0.65704700 |
| H | 1.61385300 | -0.34811800 | 1.50583400 | C | -0.17392500 | -0.12715500 | 0.47743900 |
| Cl | 2.07679200 | -1.37207600 | -0.56145900 | O | -0.70234900 | -0.03194900 | -0.73583600 |
| Cl | 2.07368400 | 1.51270000 | 0.14297200 | O | -0.79268500 | -0.29286900 | 1.50085100 |
| | | | | C | 1.34636200 | -0.01299000 | 0.49473700 |
| TS3-1 | | | | H | 1.68732200 | -0.08846400 | 1.51917800 |
| N | -3.15730900 | 0.18827600 | -0.95404300 | Cl | 2.08648300 | -1.37136500 | -0.41595200 |
| H | -3.34255700 | 1.12459500 | -1.30268300 | Cl | 1.87325300 | 1.58251100 | -0.13028600 |
| H | -3.48976500 | -0.45907800 | -1.66306800 | | | | |
| B | -3.82559800 | -0.01034700 | 0.41329900 | TS3-2 | | | |
| H | -4.81943100 | -0.67729900 | 0.43455700 | N | 3.33904700 | -0.20081500 | -0.37967500 |
| H | -3.83074900 | 0.95813500 | 1.10557500 | H | 2.23868100 | -0.76751900 | -0.35069200 |
| H | -1.85511000 | 0.10243300 | -0.89732300 | H | 3.83466400 | -0.30149200 | 0.49851500 |
| H | -3.20275100 | -1.05952000 | 1.07713000 | B | 2.43626300 | 1.00436200 | -0.54970700 |
| C | -0.21661700 | -0.07938300 | 0.34695800 | H | 2.20130800 | 1.63314100 | 0.42384200 |
| O | -0.65597900 | 0.04809700 | -0.84505500 | H | 3.91884400 | -0.47831100 | -1.16266200 |
| O | -0.88941900 | -0.23557600 | 1.37257500 | H | 2.28831900 | 1.41206500 | -1.64910200 |
| H | -2.56504900 | -0.54693100 | 1.17066000 | C | 0.25716400 | -0.62909100 | 0.58595800 |
| C | 1.30477900 | -0.03822900 | 0.52219900 | O | 0.99945000 | -0.44032900 | -0.48317800 |
| H | 1.55560900 | -0.12707200 | 1.57099500 | O | 0.60906500 | -1.19991800 | 1.59447800 |
| Cl | 2.06548200 | -1.43209100 | -0.32075000 | C | -1.17543400 | -0.08973700 | 0.50155400 |
| Cl | 1.97058700 | 1.52762300 | -0.05096400 | H | -1.68695000 | -0.27890700 | 1.43629900 |
| | | | | Cl | -1.18666000 | 1.68951700 | 0.24406100 |
| IM3 | | | | Cl | -2.09028200 | -0.94834800 | -0.78555600 |
| N | -3.58330800 | -0.46470700 | 0.02247800 | | | | |
| H | -3.71575300 | -1.41704300 | -0.28804800 | Product3 | | | |
| H | -3.09141000 | -0.40187700 | 0.90610800 | N | 3.24741300 | -0.38978300 | -0.07204700 |

| | | | | | | | |
|----------------------------|-------------|-------------|-------------|--------------|-------------|-------------|-------------|
| H | 2.89209300 | -0.43697300 | 0.88956400 | H | -1.88296900 | 1.24586300 | 0.20232900 |
| H | 4.22652500 | -0.11543100 | -0.06586900 | H | -4.45998900 | -1.08635700 | 0.91394000 |
| B | 2.34448700 | 0.72716100 | -0.84944700 | H | -5.18392700 | 1.20423000 | -0.02064500 |
| H | 2.58714300 | 1.77123700 | -0.30544000 | C | -0.44543900 | 0.00415600 | 0.06900900 |
| H | 3.17753400 | -1.31549300 | -0.49023500 | O | -0.88990800 | 1.23570300 | 0.14500500 |
| H | 2.61821800 | 0.67242800 | -2.01456300 | O | -1.10831600 | -1.00522300 | 0.06242000 |
| C | 0.41053800 | 0.07679800 | 0.50662500 | H | -3.37671300 | 1.25782000 | 0.81086500 |
| O | 0.91011500 | 0.32650000 | -0.66459900 | C | 1.10810000 | -0.03303900 | -0.00927700 |
| O | 1.02897300 | -0.06748400 | 1.55110100 | Cl | 1.63525600 | 0.89909800 | -1.44895600 |
| C | -1.11024400 | -0.08314600 | 0.53101300 | Cl | 1.76627900 | 0.71366100 | 1.48651400 |
| H | -1.43758900 | -0.22068500 | 1.55323400 | Cl | 1.68441800 | -1.70936200 | -0.14057500 |
| Cl | -1.93492800 | 1.37377500 | -0.10556200 | | | | |
| Cl | -1.58464300 | -1.56768300 | -0.37284700 | TS4-1 | | | |
| | | | | N | -3.59272200 | -0.75869000 | -0.03754500 |
| CCl₃COOH | | | | H | -2.29395000 | -0.86359100 | 0.05622100 |
| C | 1.26219000 | 0.47181700 | -0.00017600 | H | -3.82622200 | -1.23167300 | -0.90638500 |
| O | 1.58313100 | 1.62242800 | -0.00026800 | B | -4.07911200 | 0.69474700 | -0.08060000 |
| O | 2.10725500 | -0.56590600 | -0.00018000 | H | -4.00064500 | 1.21414100 | -1.14836300 |
| H | 3.00800000 | -0.20500300 | -0.00022100 | H | -2.75646900 | 1.37470500 | 0.33726300 |
| C | -0.21114600 | -0.01715000 | -0.00001400 | H | -4.01011100 | -1.30217700 | 0.71282900 |
| Cl | -1.31319400 | 1.37311700 | -0.00025400 | H | -5.05242400 | 0.95979700 | 0.56304400 |
| Cl | -0.48582300 | -1.00972500 | -1.47204600 | C | -0.49805600 | 0.15082300 | 0.04263500 |
| Cl | -0.48553400 | -1.00899100 | 1.47259000 | O | -1.00661100 | 1.26502900 | 0.00628700 |
| | | | | O | -1.09980200 | -0.97486200 | 0.11874700 |
| RC4 | | | | H | -3.37110100 | 1.44168000 | 0.86891500 |
| N | -4.02664200 | -0.78887300 | 0.04309500 | C | 1.05887400 | 0.01596700 | -0.00461000 |
| H | -3.05592500 | -1.12334000 | 0.03584100 | Cl | 1.85283900 | 1.60553400 | -0.10965400 |
| H | -4.52796000 | -1.23870600 | -0.71879700 | Cl | 1.60379300 | -0.82067800 | 1.49291500 |
| B | -4.04466200 | 0.83315500 | -0.11902600 | Cl | 1.50465800 | -0.96590700 | -1.44479900 |
| H | -3.55409500 | 1.06853000 | -1.19657300 | | | | |

| | | | | | | | |
|--------------|-------------|-------------|-------------|-----------------|-------------|-------------|-------------|
| IM4 | | | | C | 0.50741000 | 0.67266400 | 0.13881000 |
| N | -4.15616100 | -0.59927600 | -0.00104500 | O | 0.65943900 | 1.82699500 | -0.17519600 |
| H | -2.12781000 | -0.57621500 | 0.00006700 | O | 1.42263300 | -0.12142400 | 0.63545900 |
| H | -4.21889500 | -1.15324100 | -0.84397100 | C | -0.88752000 | -0.02326900 | -0.00239500 |
| B | -4.39769000 | 0.78407500 | 0.00152100 | Cl | -2.07852600 | 1.06369900 | -0.75346400 |
| H | -4.48739300 | 1.34815400 | -1.04077600 | Cl | -0.71294800 | -1.50226800 | -1.01469700 |
| H | -4.21823600 | -1.15630100 | 0.83991400 | Cl | -1.45184000 | -0.48286500 | 1.64470700 |
| H | -4.48617600 | 1.34445200 | 1.04593600 | | | | |
| C | -0.45999700 | 0.31572900 | -0.00026100 | Product4 | | | |
| O | -0.90382200 | 1.42706500 | -0.00115400 | N | 3.52275200 | 0.49402300 | 0.29719500 |
| O | -1.17107000 | -0.80800200 | 0.00059900 | H | 3.51316100 | 0.45935300 | 1.31491200 |
| C | 1.06475800 | 0.00722300 | 0.00008000 | H | 2.98809500 | 1.31462800 | -0.00755000 |
| Cl | 1.99761100 | 1.51655800 | 0.00019700 | B | 2.82317700 | -0.81102400 | -0.38913900 |
| Cl | 1.45965700 | -0.94722300 | 1.47221700 | H | 2.97742800 | -0.67494600 | -1.57269400 |
| Cl | 1.45982900 | -0.94712900 | -1.47217500 | H | 4.48855200 | 0.58122800 | -0.00921700 |
| | | | | H | 3.31890900 | -1.78877400 | 0.09292800 |
| TS4-2 | | | | C | 0.64301000 | 0.28794400 | -0.21936300 |
| N | 3.70847800 | 0.34717100 | 0.18031400 | O | 1.03211900 | 1.39716600 | -0.52230000 |
| H | 2.61002900 | 0.44479200 | 0.69088200 | O | 1.36776600 | -0.76827900 | -0.01278500 |
| H | 3.95080900 | 1.20019200 | -0.31148500 | C | -0.87468700 | 0.00391600 | 0.00781300 |
| B | 3.01191400 | -0.74438700 | -0.60004000 | Cl | -1.85000500 | 1.45100700 | -0.33485700 |
| H | 2.59569700 | -0.45815800 | -1.66796700 | Cl | -1.39215500 | -1.33533900 | -1.06768500 |
| H | 4.43228400 | 0.07962900 | 0.83718000 | Cl | -1.10315200 | -0.47312600 | 1.73177400 |
| H | 3.15266500 | -1.85125800 | -0.21247300 | | | | |