Improved Plasmonic Au-Ag/TiO₂/rGO Photocatalyst through Entire Visible Range Absorption, Charge Separation and High Adsorption Ability.

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ELECTRONIC SUPPLEMENTARY INFORMATION

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Figure S1. ¹⁹F NMR spectra in d8-THF of the reaction by-product C_6F_5 - C_6F_5 formed during the decomposition of complex $[Au_2Ag_2(C_6F_5)_4(OEt_2)_2]_n$ under reflux in the presence of PVP polymer, confirming a reductive elimination mechanism.



Figure S2. Time-resolved UV-Vis spectra of the AuAg-PVP NPs formation. The orange arrow shows the red-shift of the maximum absorption with time.



Figure S3. XRD spectrum of AuAg PVP NPs.



Figure S4. Size histograms of TiO_2 NPs in nanohybrids **1A-3A** (up) and **1B-3B** (bottom).



Figure S5. Tauc plots for solid samples of P25 TiO_2 NPs (black), **1A** (red), **1B** (green), **2A** (blue), **2B** (cyan), **3A** (magenta) and **3B** (ochre).



Figure S6. Raman spectrum for nanohybrid 3A



Figure S7. Raman spectrum for nanohybrid **3B**



Figure S8. General XPS survey of AuAg-TiO₂ NP nanohybrid **1A**



Figure S9. High-resolution $Ag_{3d5/2-3/2}$ XPS spectra of: **1A** (a), **1B** (b), **3A** (c) and **3B** (d) nanohybrids.



Figure S10. High-resolution $Ti_{2p3/2-1/2}$ XPS spectra of: **1A** (a), **1B** (b), **3A** (c) and **3B** (d) nanohybrids.



Figure S11. Adsorption of a methylene blue (MB) solution (5 mg/L concentration) under visible light (low-power white LED sources $(4 \times 10 \text{ W})$) for a 2 mg sample of rGO.



Figure S12. Kinetics of the photocatalysts (40 mg) under visible light. The fitting results are represented assuming a pseudo-first order reaction



Figure S13. Kinetics of the photocatalysts (15 mg) under visible light. The fitting results are represented assuming a pseudo-first order reaction



Figure S14. Kinetics of the photocatalysts (15 mg) under sunlight. The fitting results are represented assuming a pseudo-first order reaction