

# A coumarin-based reversible fluorescent probe for Cu<sup>2+</sup> and S<sup>2-</sup> and its applicability in vivo and organisms imaging

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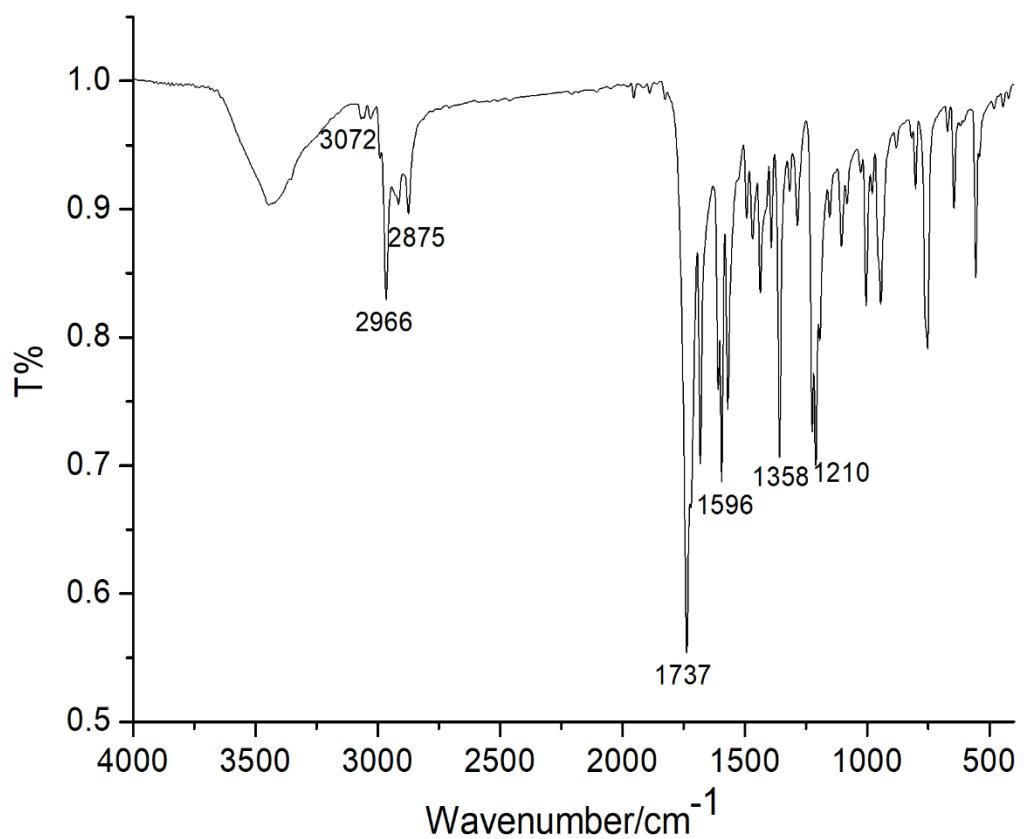
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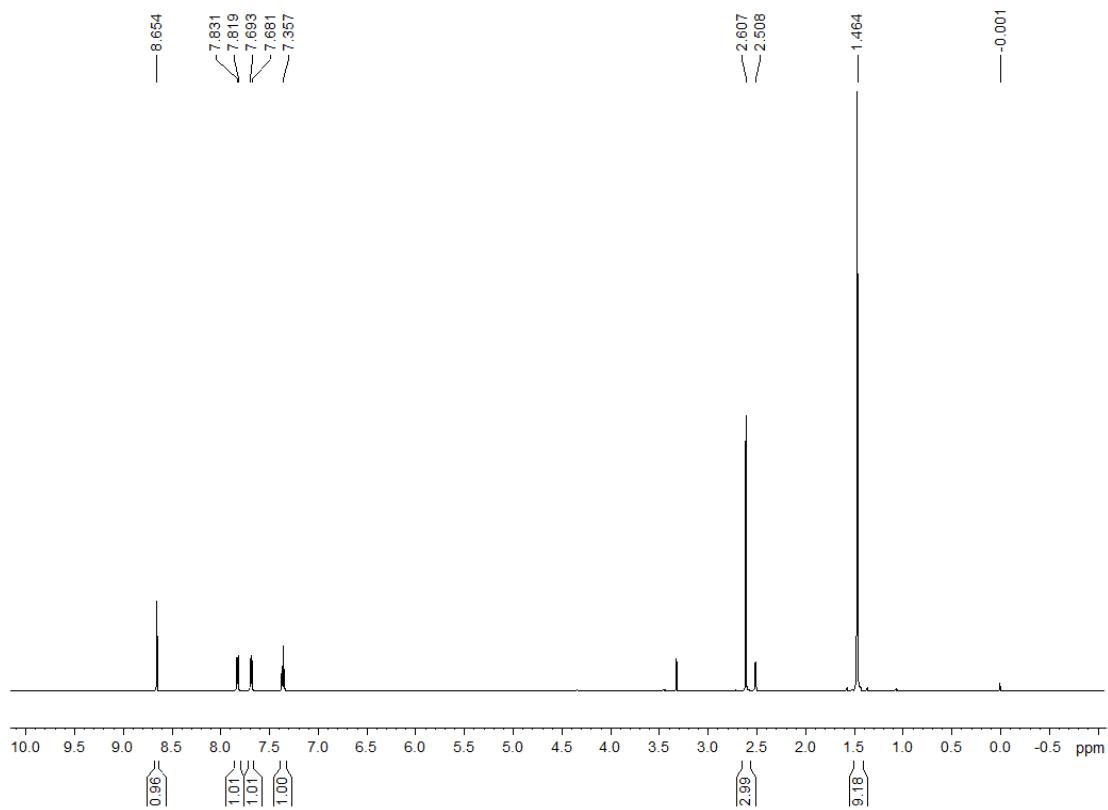
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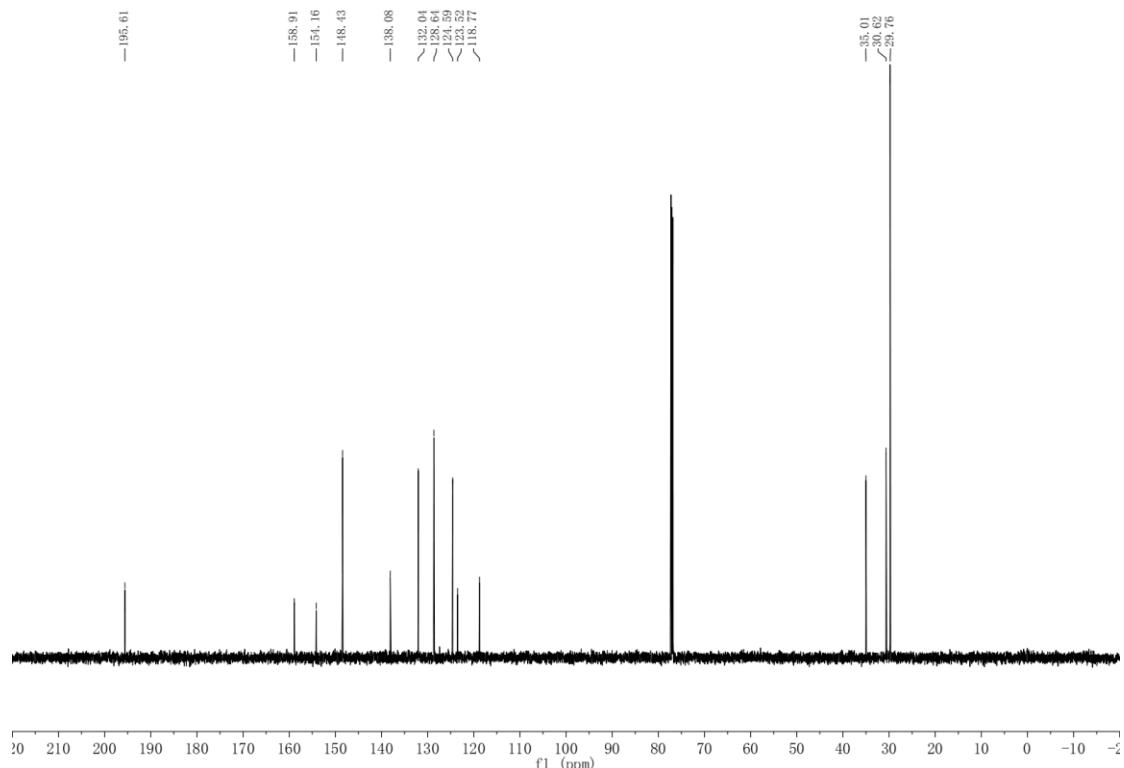
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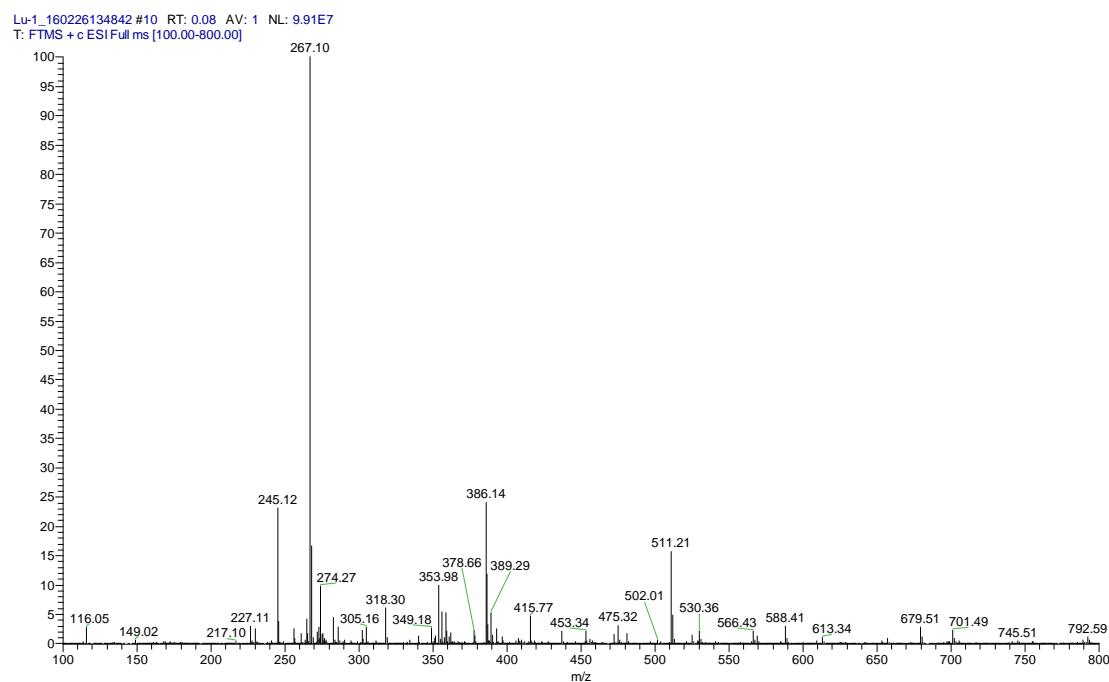
**Fig. S1.** IR spectrum of 3-acetyl-8-tert-butyl coumarin (**BuCAC**).



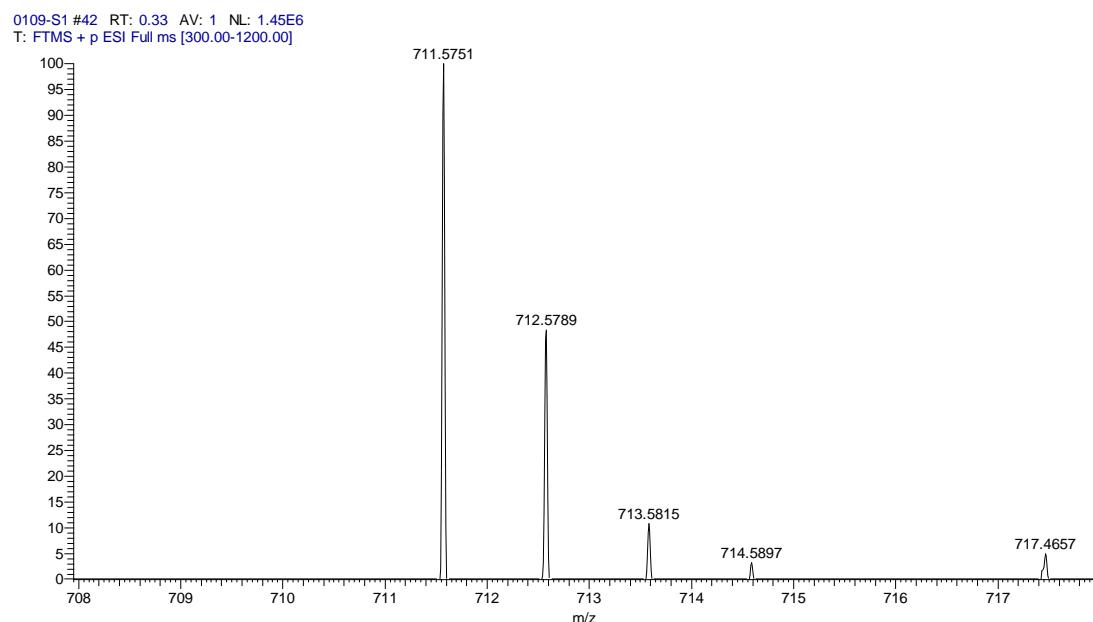
**Fig. S2.** <sup>1</sup>H NMR spectrum of 3-acetyl-8-tert-butyl coumarin (**BuCAC**).



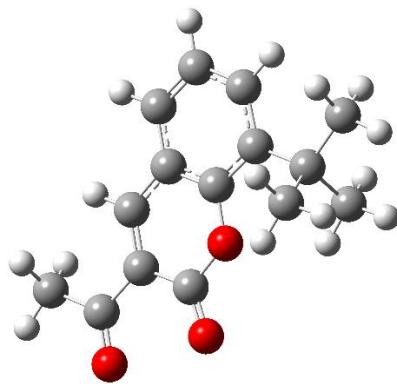
**Fig. S3.** <sup>13</sup>C NMR spectrum of 3-acetyl-8-tert-butyl coumarin (**BuCAC**).



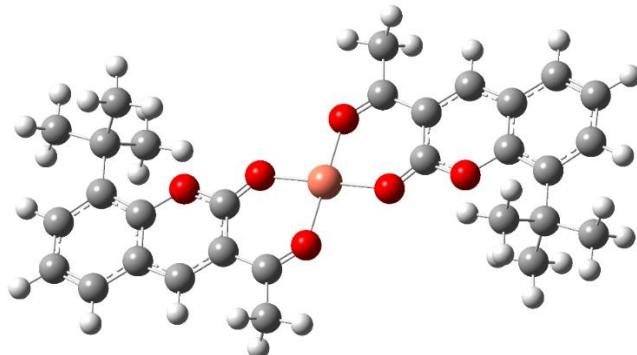
**Fig. S4.** 3-acetyl-8-tert-butyl coumarin (**BuCAC**).



**Fig. S5.** Experimental mass spectrum of **BuCAC-Cu<sup>2+</sup>**.



**Fig. S6** The optimized molecular structure of BuCAC.



**Fig. S7** The optimized molecular structure of BuCAC- $\text{Cu}^{2+}$ .

**Table S1** Compared with other currently reported probe for  $\text{Cu}^{2+}$  and  $\text{S}^{2-}$ .

Structure	Limit of detection		Ref
	$\text{Cu}^{2+}$	$\text{S}^{2-}$	
	$6.5 \times 10^{-8} \text{M}$	$2.9 \times 10^{-7} \text{M}$	39
	$1.1 \times 10^{-7} \text{M}$	$2.2 \times 10^{-7} \text{M}$	40
	$3.1 \times 10^{-7} \text{M}$	$1.9 \times 10^{-7} \text{M}$	41
	$1.9 \times 10^{-7} \text{M}$	$4.4 \times 10^{-7} \text{M}$	42
	$1.5 \times 10^{-6} \text{M}$	No data	43
	$3.03 \times 10^{-7} \text{M}$	$1.7 \times 10^{-7} \text{M}$	This work

**Table S2** The HOMO and LUMO distributions of **BuCAC** and **BuCAC-Cu<sup>2+</sup>**.

	<b>BuCAC</b>	<b>BuCAC-Cu<sup>2+</sup></b>	
		$\alpha$ orbital	$\beta$ orbital
LUMO (eV)	-2.636	-8.743	-9.449
HOMO (eV)	-6.826	-12.248	-12.245
Gap(eV)	4.190	3.505	2.796

#### **Ethical Statement of Animal Experiment**

All animal procedures were performed in accordance with the Guidelines for Care and Use of Laboratory Animals of "Nanjing Forestry" University and experiments were approved by the Animal Ethics Committee of "Nanjing Forestry University".