

Supporting Information

CoO_x/UiO-66 and NiO/UiO-66 Heterostructures with UiO-66 Frameworks for Enhanced Oxygen Evolution Reactions

Victor Charles,^{a,b,#} Yong Yang,^{a,#} Menglei Yuan,^{a,b} Jitao Zhang,^f Yaling Li,^f Jingxian Zhang,^{a,b} Tongkun Zhao,^{a,b} Zhanjun Liu^c, Bin Li^{d*}, Guangjin Zhang,^{a,b,e*}

^aCAS Key Laboratory of Green Process and Engineering, Institute of Process Engineering, Chinese Academy of Sciences, 100190, Beijing, China. E-mail: zhanggj@ipe.ac.cn

^bCenter of Materials Science and Optoelectronics Engineering, University of Chinese Academy of Sciences, 100049, Beijing, China

^cCAS Key Laboratory of Carbon Materials, Institute of Coal Chemistry, Chinese Academy of Sciences, Taiyuan, 030001, China

^dZhengzhou Tobacco Research Institute of CNTC, No 2 Fengyang Street, Zhengzhou High-Tech Development District, Henan, China, 450001, E-mail: lib@ztri.com.cn

^eChemistry and Chemical Engineering Guangdong Laboratory, Shantou, 515031 China

^fBeijing Engineering Research Center of Printed Electronics, Beijing Institute of Graphic Communication, Beijing, China

#Equal contribution

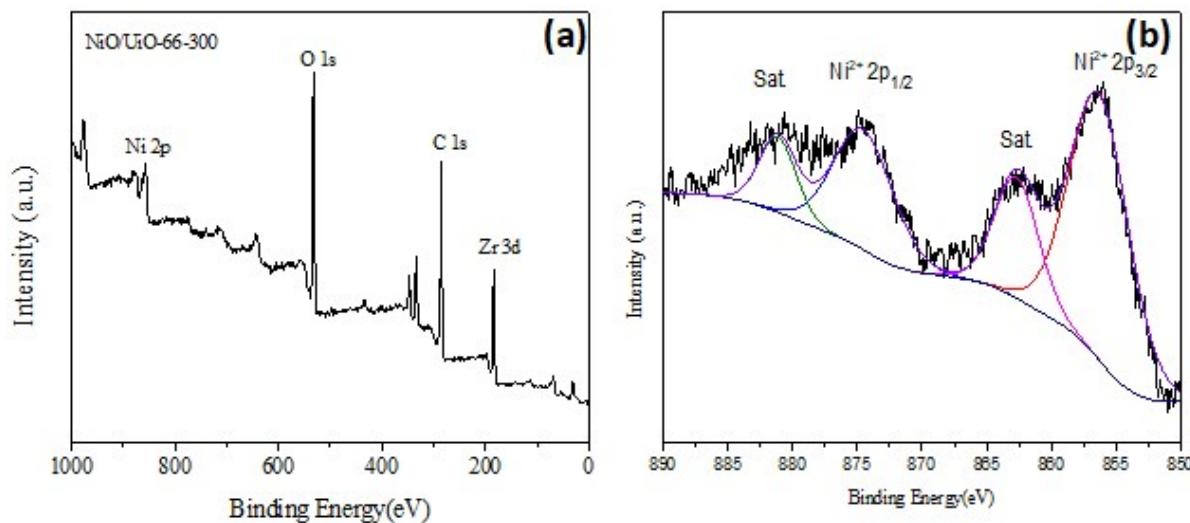
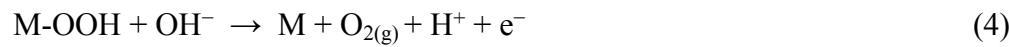


Figure S1. (a) Survey spectrum of NiO/UiO-66-300 and (b) XPS spectra of Ni 2p

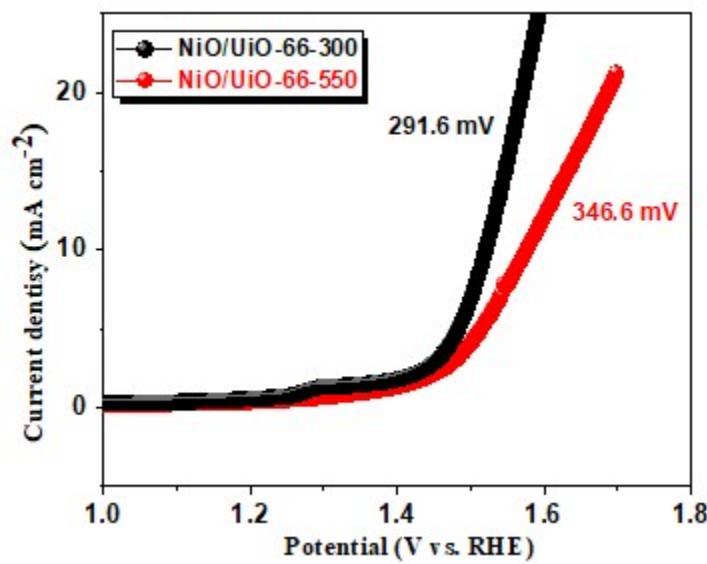


Figure S2. OER polarization curves of NiO/UiO-66-300, and NiO/UiO-66-550

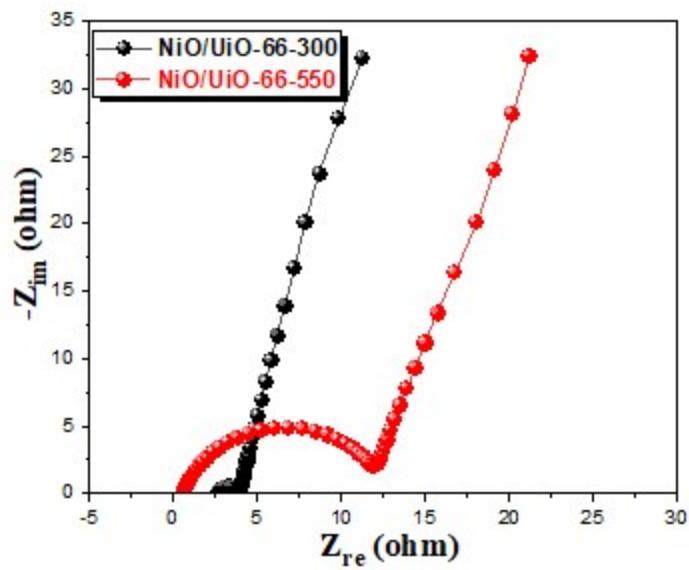


Figure S3. EIS of NiO/UiO-66-300, and NiO/UiO-66-550

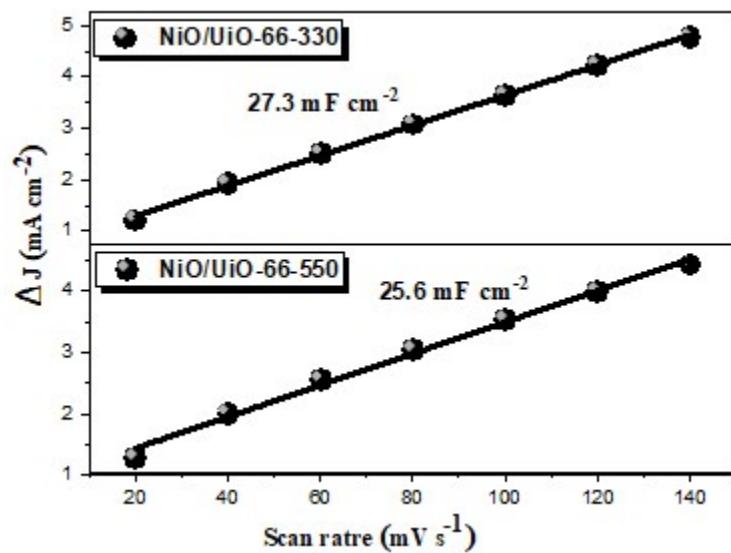


Figure S4. C_{di} of NiO/UiO-66-300, and NiO/UiO-66-550

Table S1. OER performance comparisons of some recently published heterostructure catalysts

Catalysts	Mass loading (Mg/cm ²)	Electrolyte	Overpotential at 10 mA cm ⁻² (mV vs. RHE)	References
CoO _x /UiO-66-300	0.3	1 M KOH	283	This work
NiO/UiO-66-300	0.3	1 M KOH	291	This work
2.5Fe-NiCoP/PBA HNCs	0.2	1 M KOH	290	¹
RuO ₂ /CeO ₂ heterostructure	0.28	1 M KOH	350	²
Co ₃ O ₄ /Fe ₂ O ₃ nanocubes	3.0	1 M KOH	310	³
Ultrathin Co ₃ O ₄ nanomeshes	~0.34	1 M KOH	307	⁴
Fe-CoOOH/graphene	0.2	1 M KOH	330	⁵
MoS _x - encapsulated Co(OH) ₂ nanosheets	0.2	0.1 M KOH	350	⁶
Co-BPDC/Co-BDC heterostructure	0.28	1 M KOH	335	⁷
Ni-BDC/Ni(OH) ₂ heterostructure	-	1 M KOH	320	⁸
Se-(CoFe)S ₂ heterostructure	-	1 M KOH	281	⁹
Ni-MOF/LDH heterostructure	-	1 M KOH	220	¹⁰
Ni ₃ S ₂ -Co ₉ S ₈ heterostructure	-	1 M KOH	294	¹¹
Co ₉ S ₈ /Ni ₃ S ₂ /NF heterostructure	-	1 M KOH	227	¹²
Ni ₃ S ₂ @Co(OH) ₂ /NF heterostructure	-	1 M KOH	290	¹³
NCS-0.5/NF heterostructure	-	1 M KOH	340	¹⁴
NiO/NiS heterostructure	-	1 M KOH	209	¹⁵

Supplementary References

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