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## Supplementary information

## Defective Sr<sub>0.9</sub>Mo<sub>0.9</sub>O<sub>3-8</sub> perovskites with *exsolved* Ni nanoparticles as high-performance composite anodes for solid-oxide fuel cells

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**Fig. S1.** Thermal evolution of the NPD patterns for Ni-Sr<sub>0.90</sub>Mo<sub>0.90</sub>O<sub>3-δ</sub> between 25 and 850 °C. No crystallographic transitions were detected in the entire temperature range.



**Fig. S2.** Observed (red crosses), calculated (black full line) and difference (blue line) NPD profiles for the Ni-Sr<sub>0.9</sub>Mo<sub>0.9</sub>O<sub>3- $\delta$ </sub> composite at a) 300 °C and b) 600 °C, refined in the *Pm-3m* (No. 221) space group. A second and third set of Bragg reflections correspond to minor amounts of oxidized scheelite phase (s.g. *I4*<sub>1</sub>/*a*) and exsolved metallic Ni (s.g. *Fm-3m*) respectively.





Element	Obs. (% weight)	Calc.
0	13.94	21.90
Ni	10.35	2.67
Sr	31.82	36.00
Мо	43.89	39.41
Total	100.00	100.00

**Fig. S3.** a) Typical particle of Sr<sub>0.9</sub>MoO<sub>3-δ</sub> matrix with exsolved Ni particles adhered at the surface. b) Histogram of the diameter of Ni nanoparticles in correlation with the frequency. The average size is 30-35 nm. c) EDX spectrum showing the major occurrence of Sr, Mo and Ni. The Table below indicates the % weight of Sr, Mo, Ni and O when focusing on a Ni nanoparticle; the rest of the elements are also patent.