

New Journal of Chemistry

Supporting Information

Coordination compounds containing 2-pyridylselenium ligands: synthesis, structural characterization, and antibacterial evaluation

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Table S1. Crystallographic and structure refinement data for complexes **1**, **2**, **3**, **5** and **6**.

Complex	1	2	3	5	6
Formula	C ₁₁ H ₁₁ Cl ₂ CoN ₃ Se	C ₁₁ H ₁₁ Br ₂ CoN ₃ Se	C ₁₁ H ₁₁ Cl ₂ CuN ₃ Se	C ₁₁ H ₁₁ Cl ₂ N ₃ SeZn	C ₁₀ H ₁₀ N ₄ OReSe ₂ Cl ₂ C ₃ H ₇ NO
CCDC n°	2075668	2075669	2075670	2075671	2075672
F.W. (g·mol ⁻¹)	394.02	482.94	398.63	400.46	654.89
Crystal system	Triclinic	Monoclinic	Orthorhombic	Triclinic	Orthorhombic
Space group	<i>P</i> -1	<i>P</i> 2 ₁ / <i>n</i>	<i>Pna</i> 2 ₁	<i>P</i> -1	<i>Pbca</i>
<i>a</i> (Å)	7.6910(4)	8.3309(6)	13.8873(8)	7.6948(8)	7.0855(16)
<i>b</i> (Å)	7.7691(4)	12.1666(9)	12.4918(7)	7.7787(8)	16.597(5)
<i>c</i> (Å)	12.7167(6)	14.2649(12)	7.9670(4)	12.7007(14)	31.064(9)
α (°)	97.9940(10)	90	90	98.027(3)	90
β (°)	91.522(2)	93.019(3)	90	91.224(3)	90
γ (°)	112.2890(10)	90	90	112.402(3)	90
T (K)	296(2)	296(2)	295(2)	296(2)	100(2)
V (Å ³)	693.63(6)	1443.87(19)	1382.09(13)	693.64(13)	3653.2(17)
Z	2	4	4	2	8
<i>r</i> calc. (g·cm ⁻³)	1.887	2.222	1.916	1.920	2.381
μ (mm ⁻¹)	2.219	4.907	2.413	2.507	10.801
<i>F</i> (000)	386	916	780	392	2448
Collected reflections	55879	19972	41077	10908	44575
Independent reflections [<i>R</i> _{int}]	4225 [0.0427]	4433 [0.1132]	4120 [0.0578]	2690 [0.0475]	9869 [0.0361]
<i>R</i> ₁ [<i>I</i> > 2σ(<i>I</i>)]	0.0394	0.0491	0.0361	0.0493	0.0189
w <i>R</i> ₂ [<i>I</i> > 2σ(<i>I</i>)]	0.0937	0.0746	0.0815	0.1082	0.0374
<i>R</i> ₁ (all data) ^[a]	0.0499	0.1199	0.0505	0.0588	0.0241
w <i>R</i> ₂ (all data) ^[b]	0.0980	0.0898	0.0872	0.1146	0.0400
GOOF on <i>F</i> ²	1.174	0.998	1.045	1.086	1.166
Largest diff. peak and hole (e.Å ⁻³)	0.703 and -0.565	0.535 and -0.579	0.403 and -0.451	0.590 and -0.382	0.538 and -0.917

^[a]*R*₁ = $\sum |F_o| - |F_c| | / \sum |F_o|$; ^[b]w*R*₂ = { $\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2$ }^{1/2}

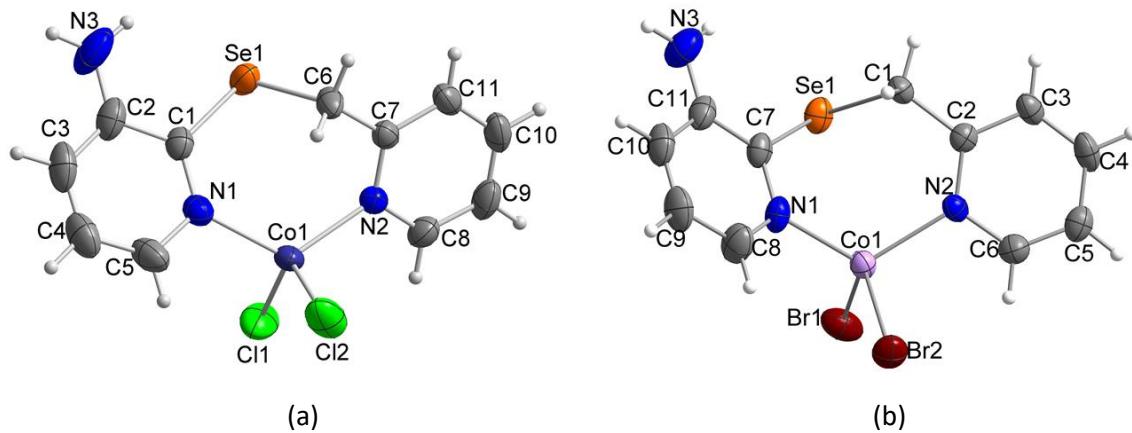


Figure S1. Ellipsoid representations (50% probability) of $[\text{CoCl}_2(\text{L})]$ (1) (a) and $[\text{CoBr}_2(\text{L})]$ (2) (b).

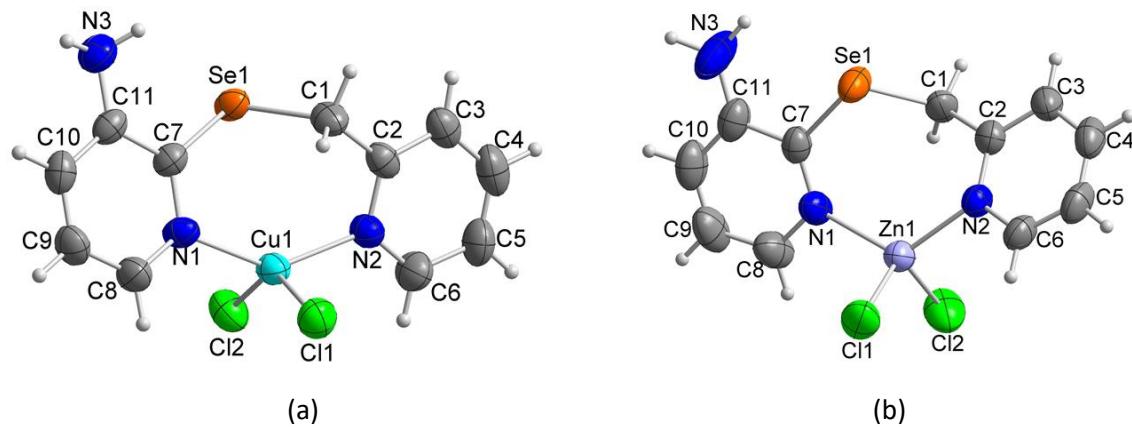


Figure S2. Ellipsoid representations (50% probability) of $[\text{CuCl}_2(\text{L})]$ (3) (a) and $[\text{ZnCl}_2(\text{L})]$ (5) (b).

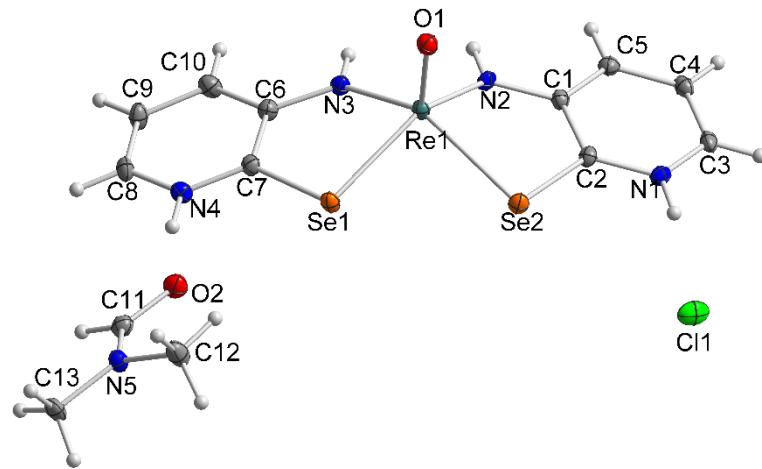


Figure S3. Ellipsoid representations (50% probability) of $[\text{ReO}(\text{apySe})_2]\text{Cl}\cdot\text{DMF}$ (6).

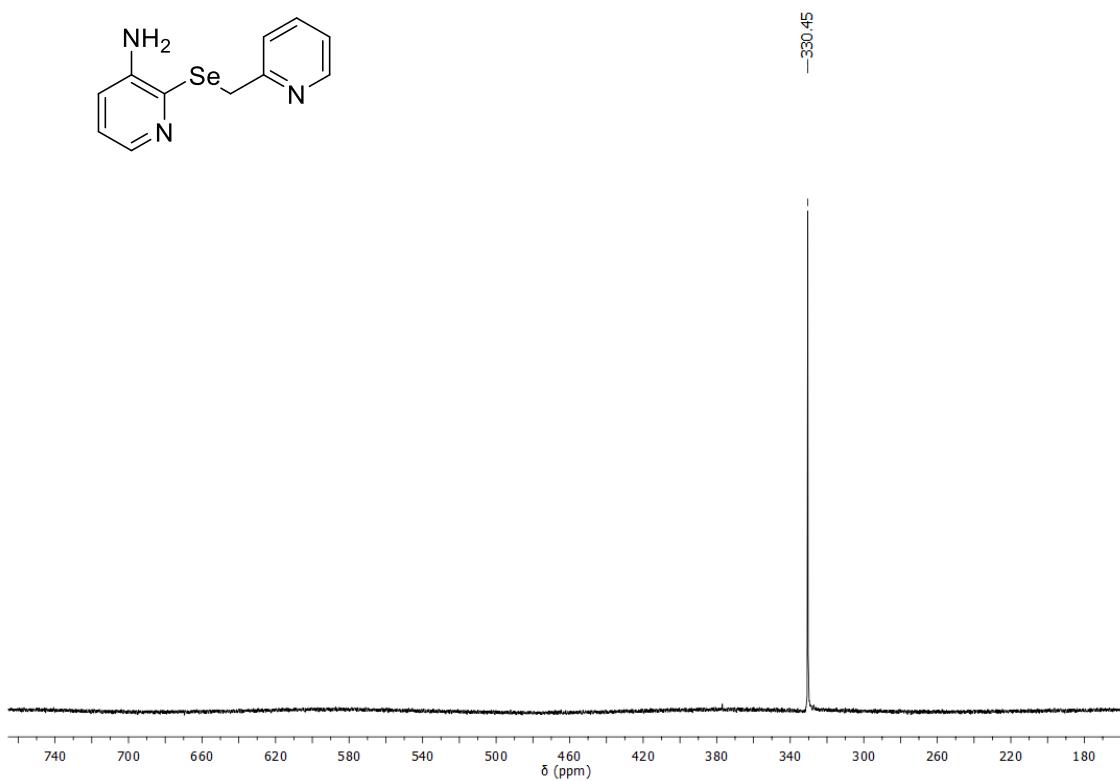


Figure S4. ^{77}Se NMR (76 MHz, CDCl_3) spectrum of Ligand L.

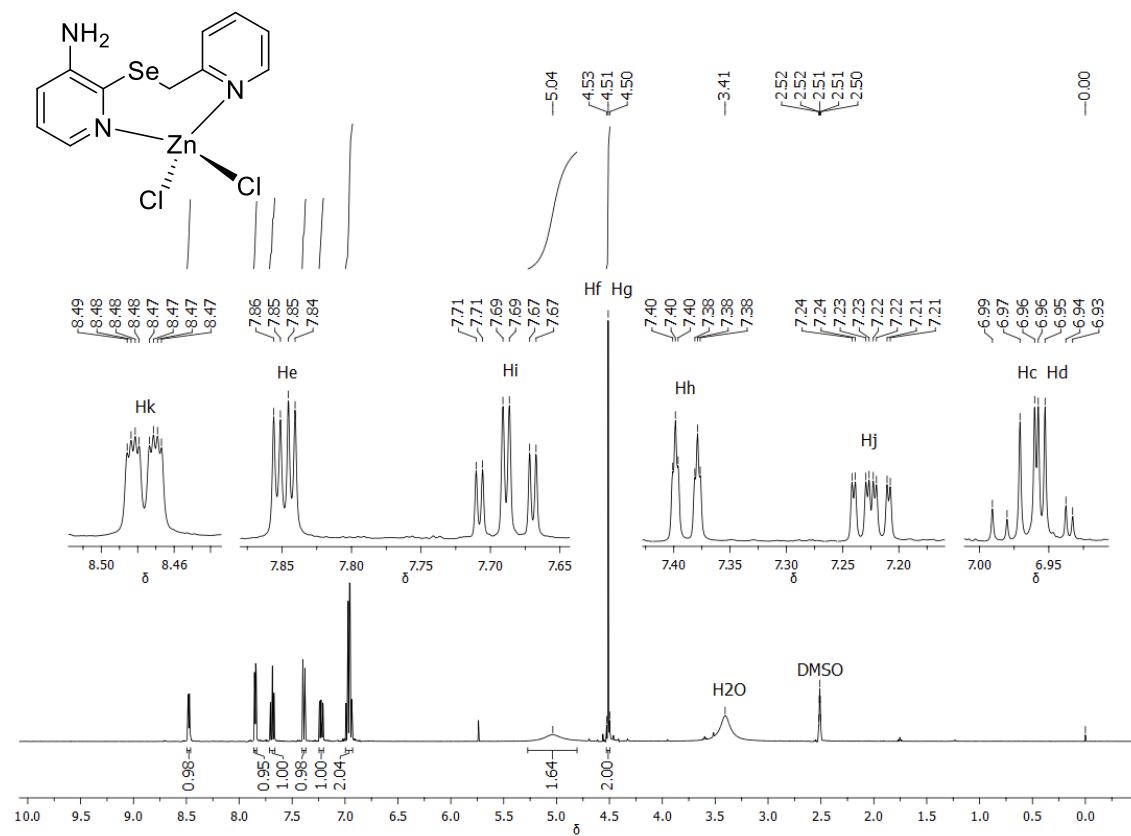


Figure S5. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) spectrum of complex 5.

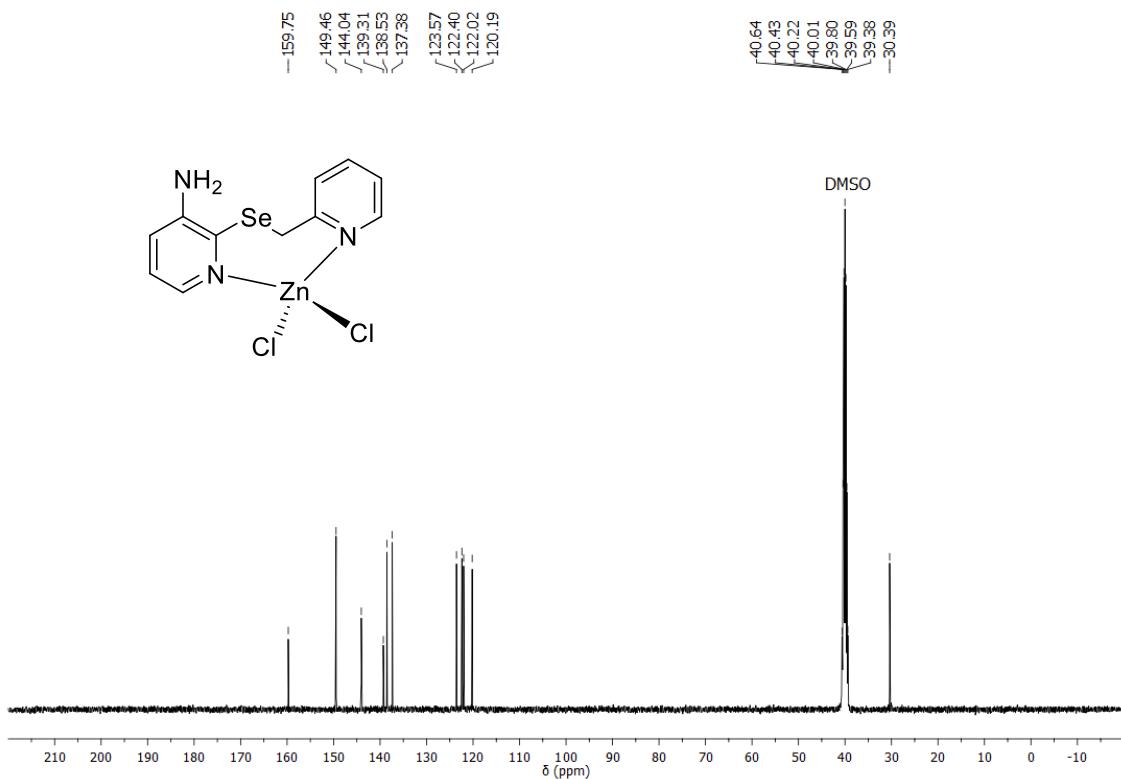


Figure S6. ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) spectrum of complex 5.

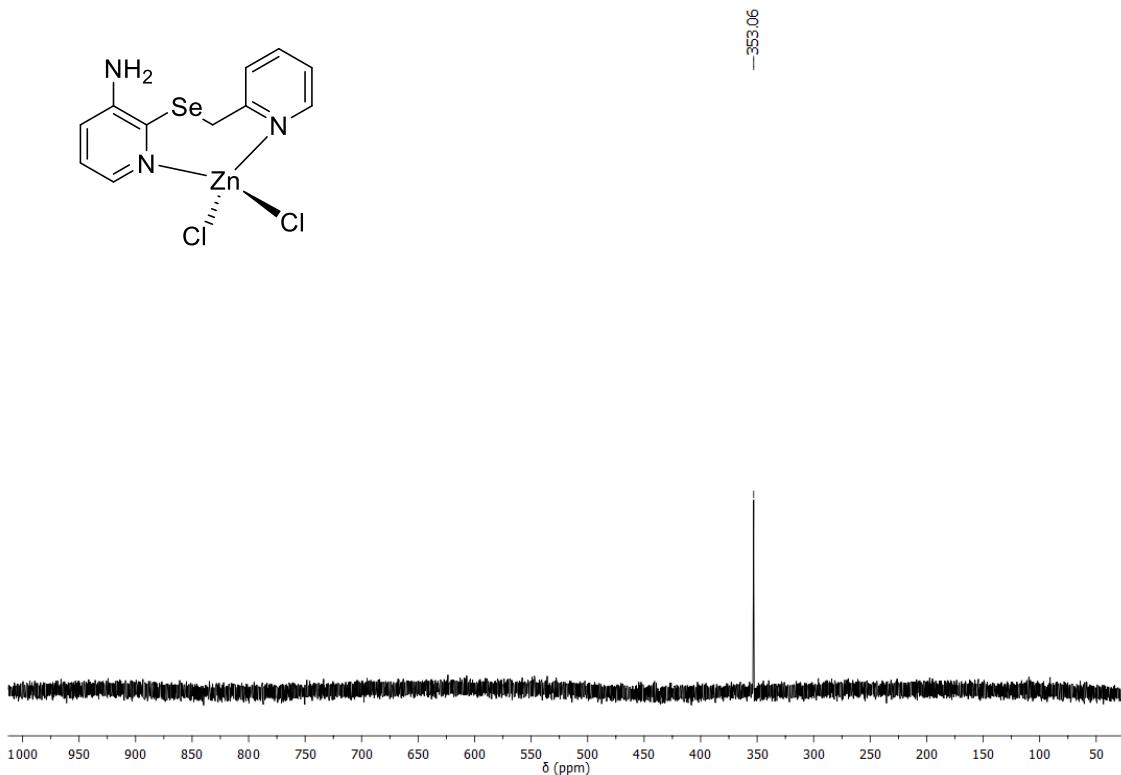


Figure S7. ^{77}Se NMR (76 MHz, $\text{DMSO}-d_6$) spectrum of complex 5.

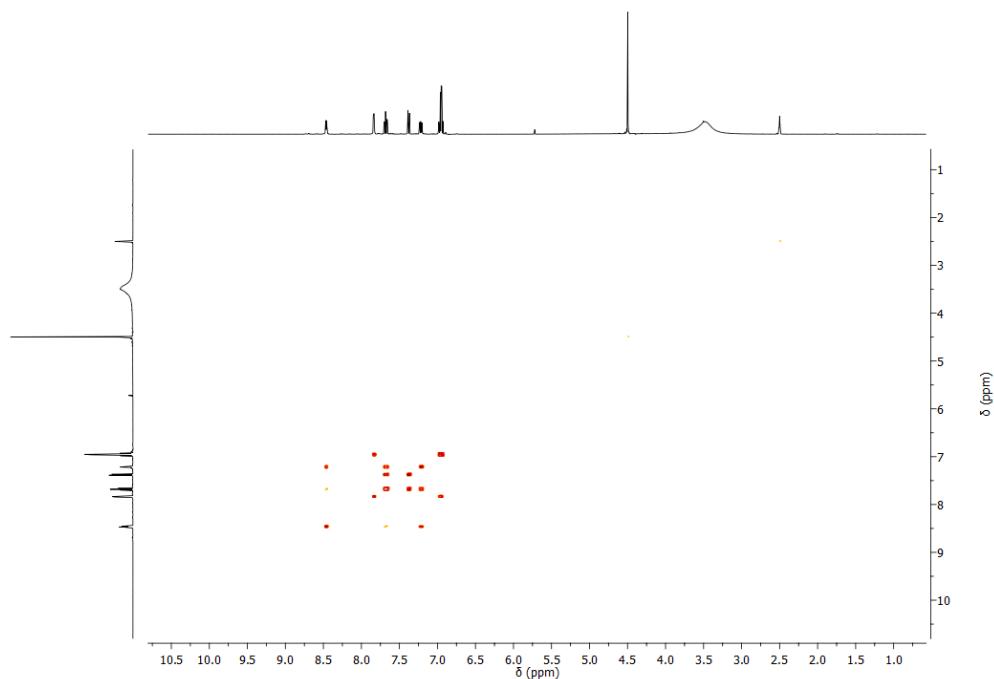


Figure S8. 2D COSY NMR spectrum of complex 5 (DMSO-*d*₆).

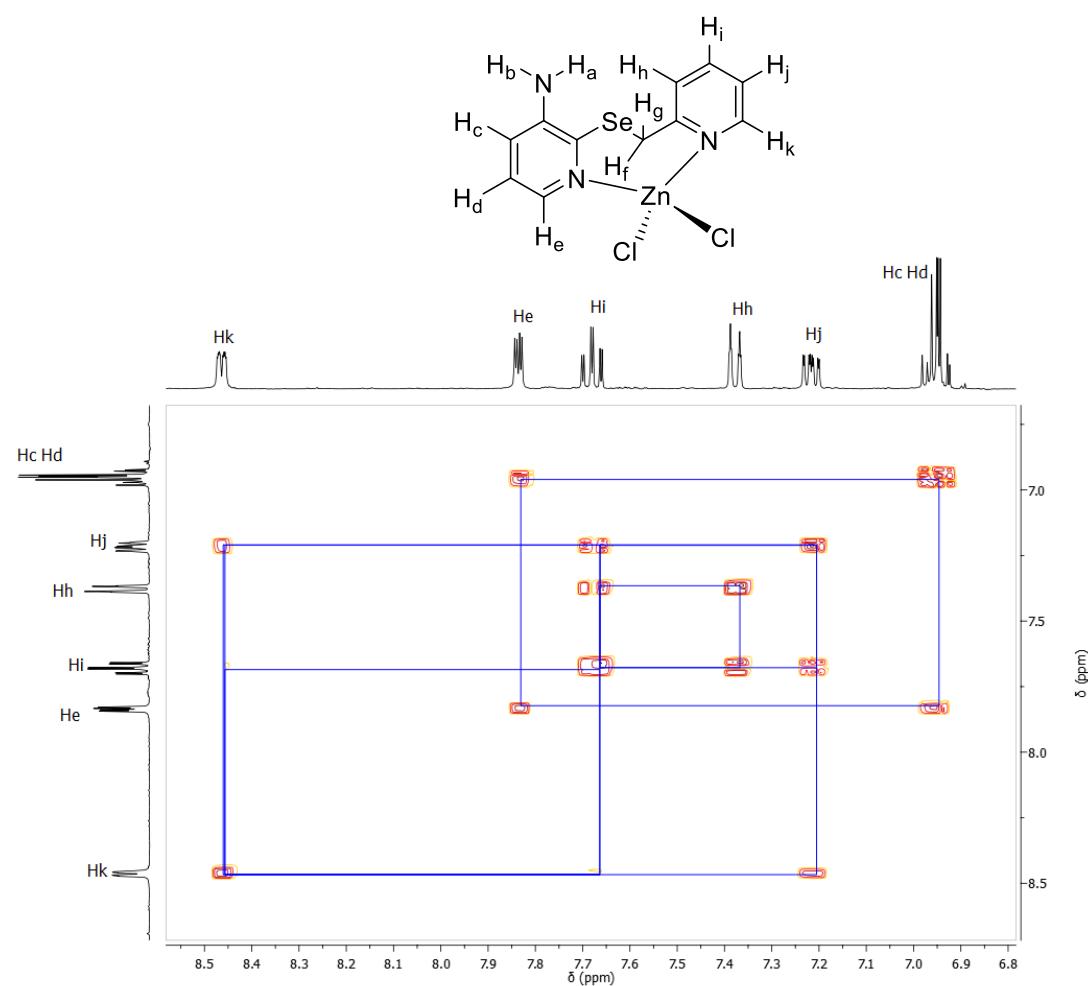


Figure S9. 2D COSY expansion spectrum of complex 5 (DMSO-*d*₆).

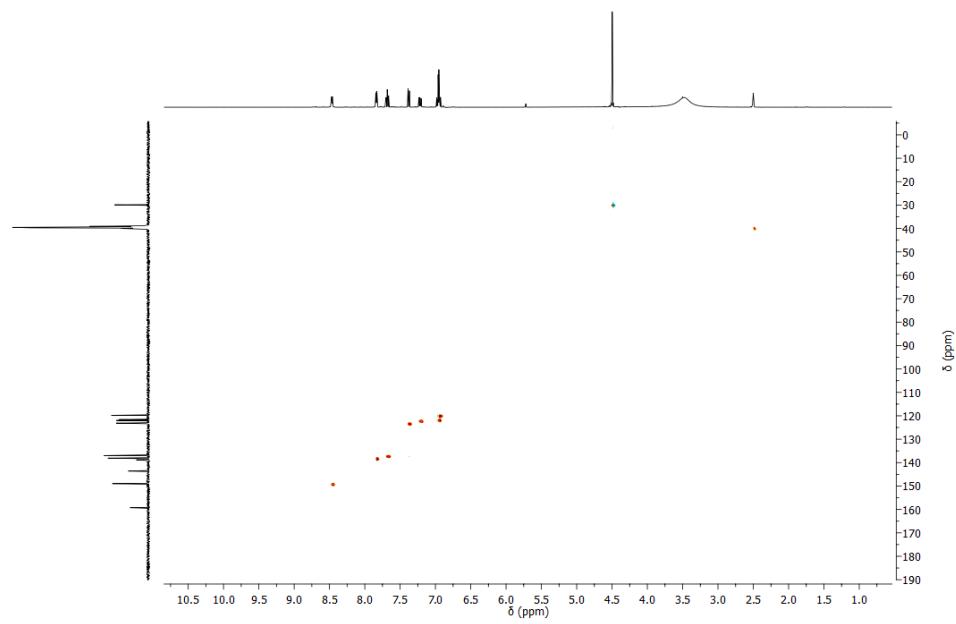


Figure S10. 2D HMQC NMR spectrum of complex 5 (DMSO-*d*₆).

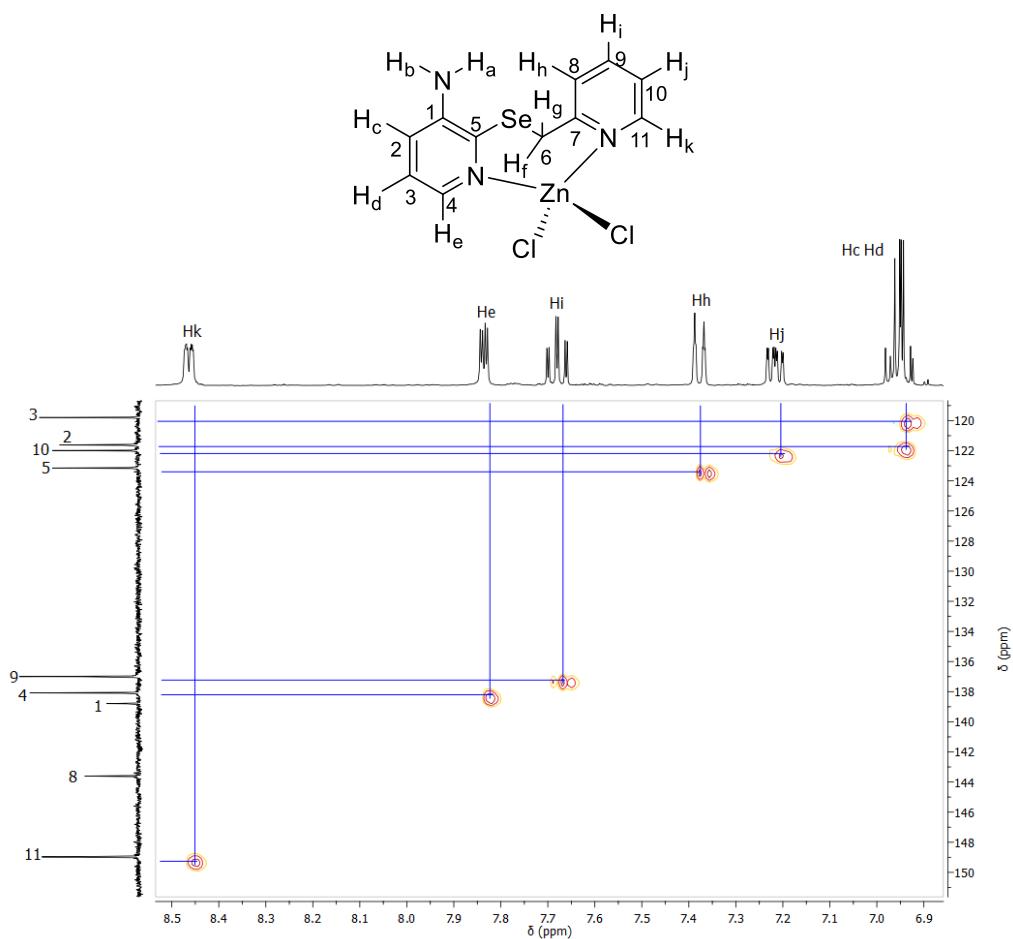


Figure S11. 2D HMQC expansion spectrum of complex 5 (DMSO-*d*₆).

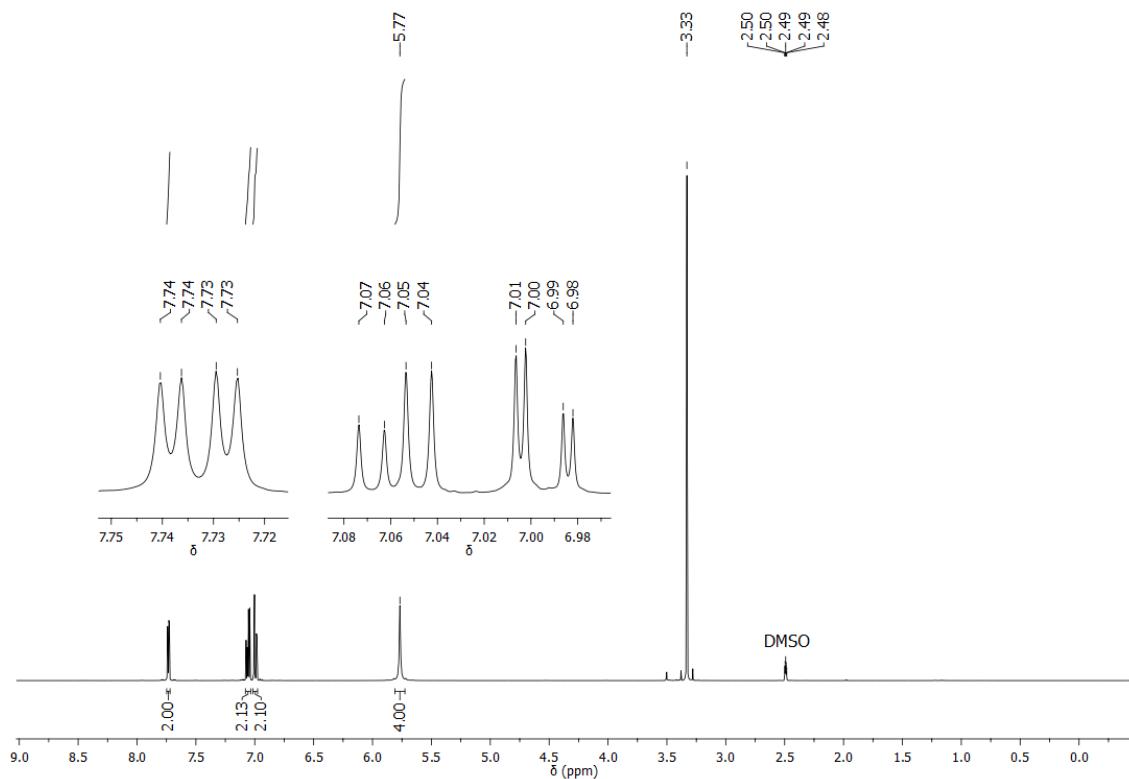


Figure S12. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) spectrum of $(\text{apySe})_2$.

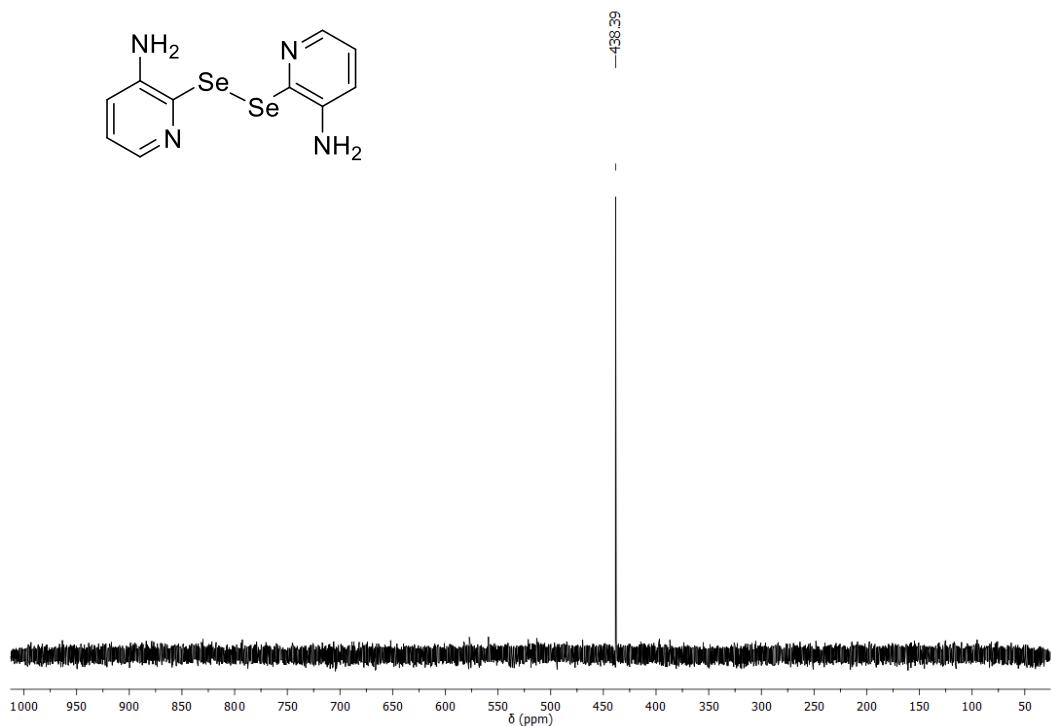


Figure S13. ^{77}Se NMR (76 MHz, $\text{DMSO}-d_6$) spectrum of $(\text{apySe})_2$.

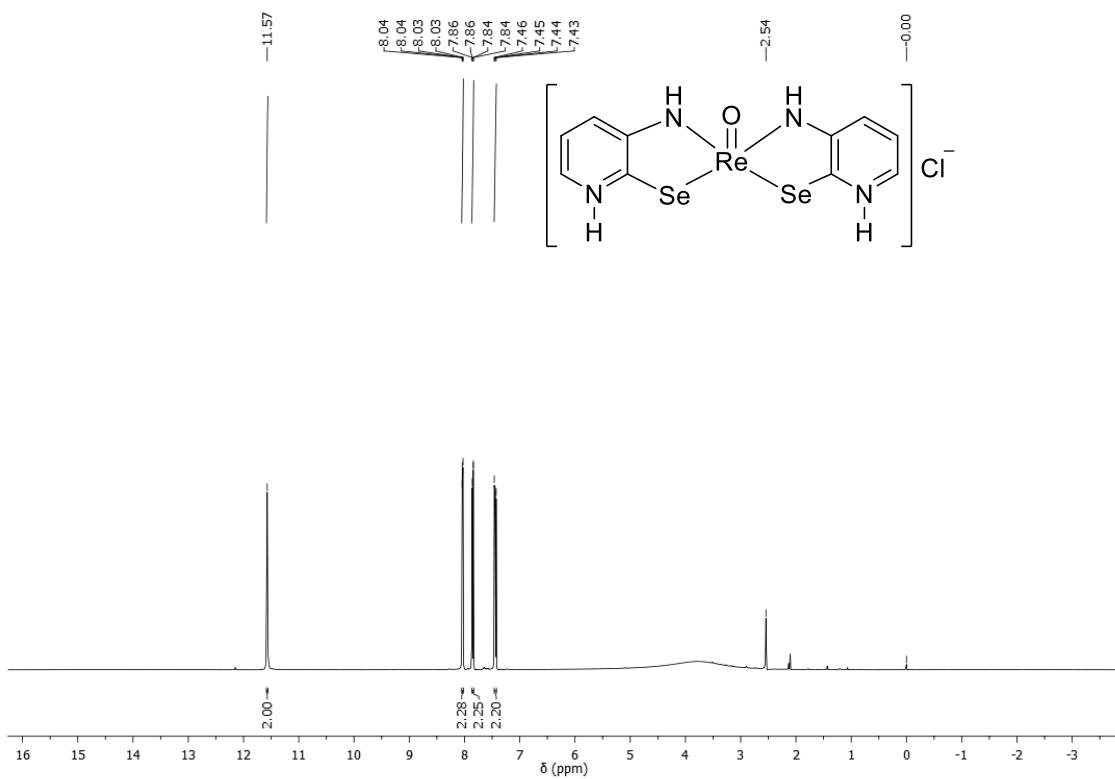


Figure S14. ^1H NMR (400 MHz, DMSO-*d*₆) spectrum of complex 6.

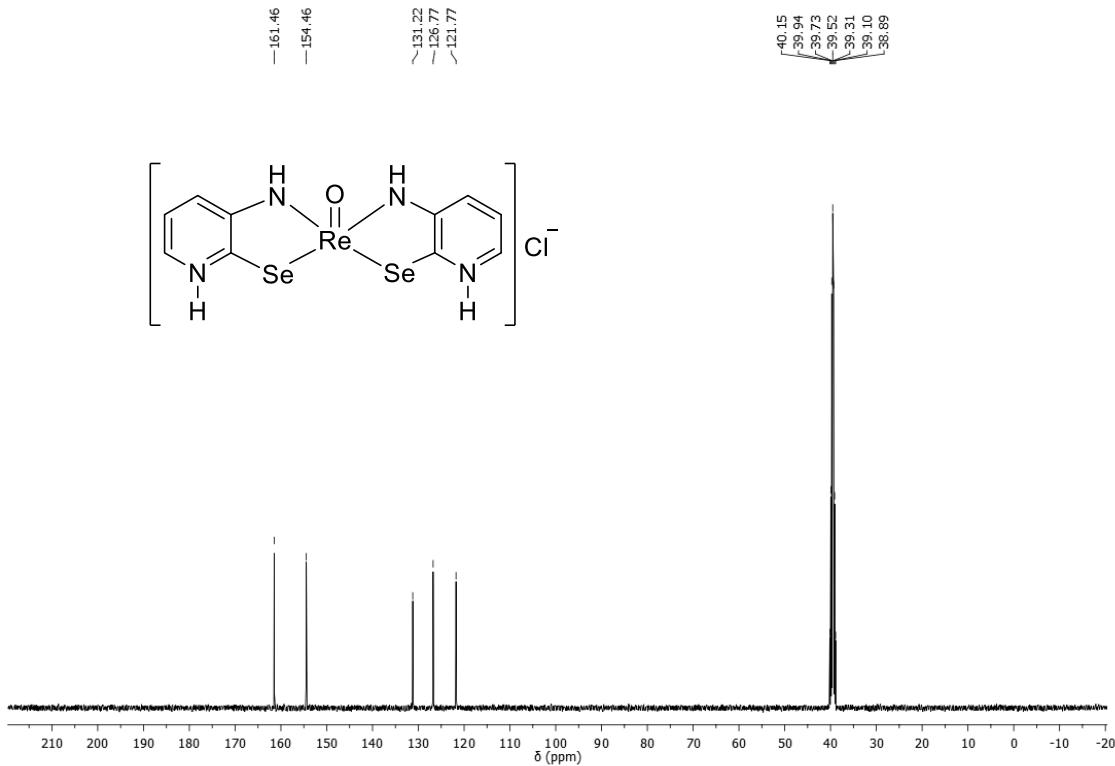


Figure S15. ^{13}C NMR (100 MHz, DMSO-*d*₆) spectrum of complex 6.

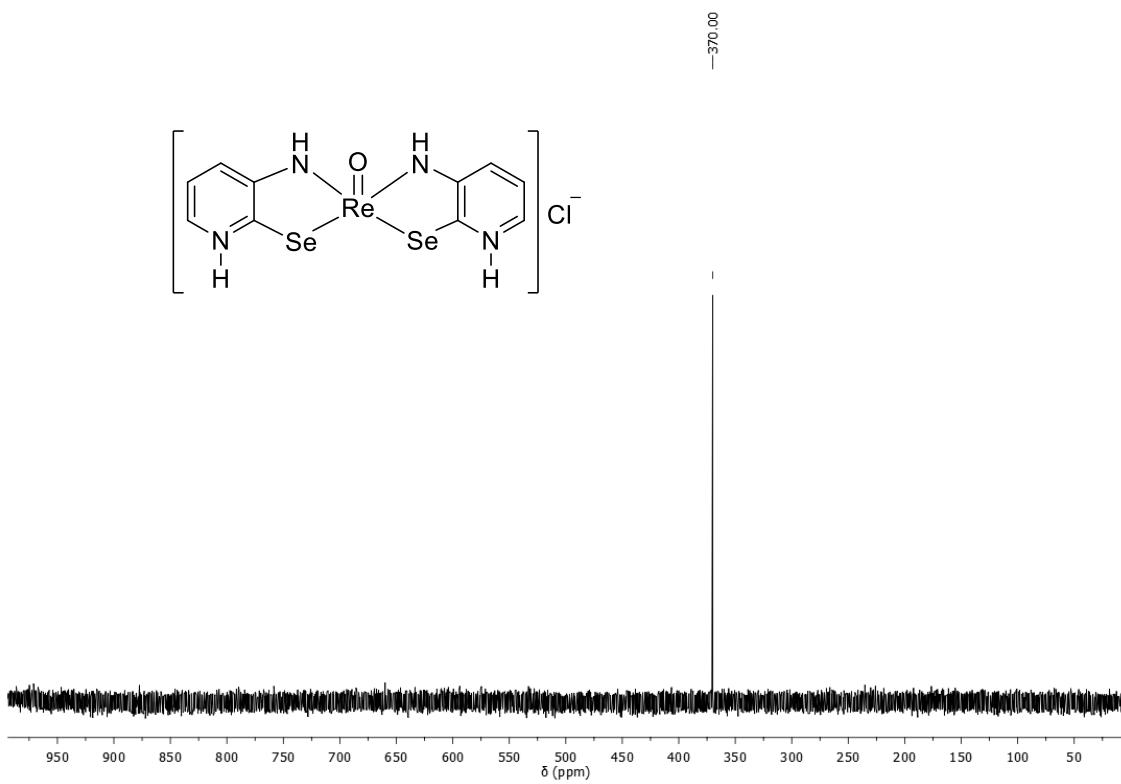


Figure S16. ^{77}Se NMR (76 MHz, $\text{DMSO}-d_6$) spectrum of complex 6.

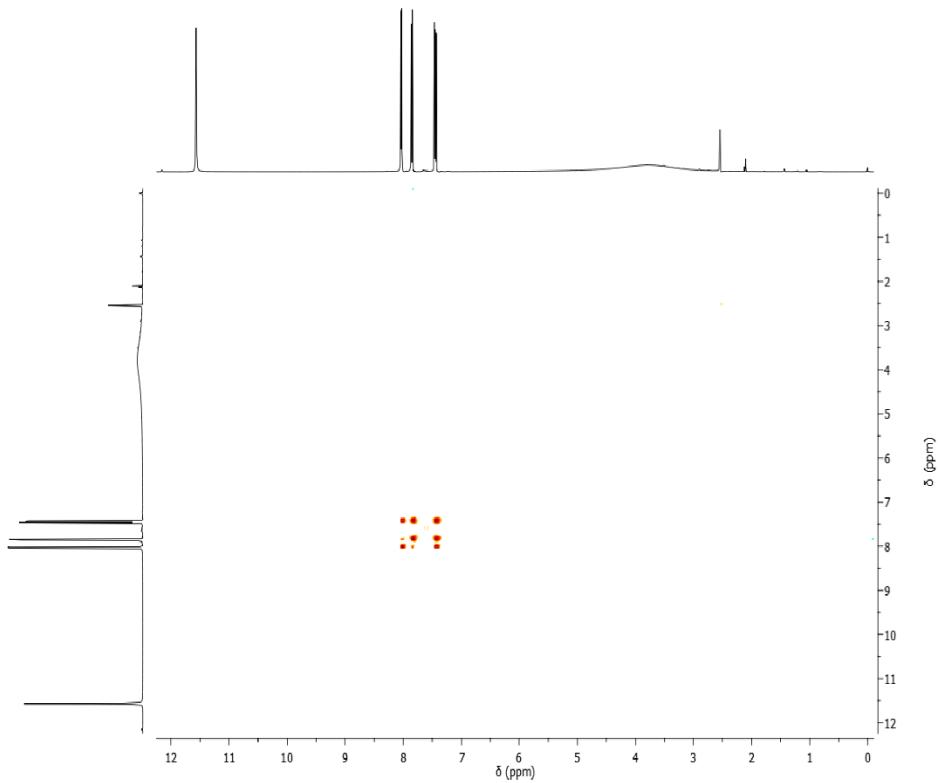


Figure S17. 2D COSY NMR spectrum of complex 6 ($\text{DMSO}-d_6$).

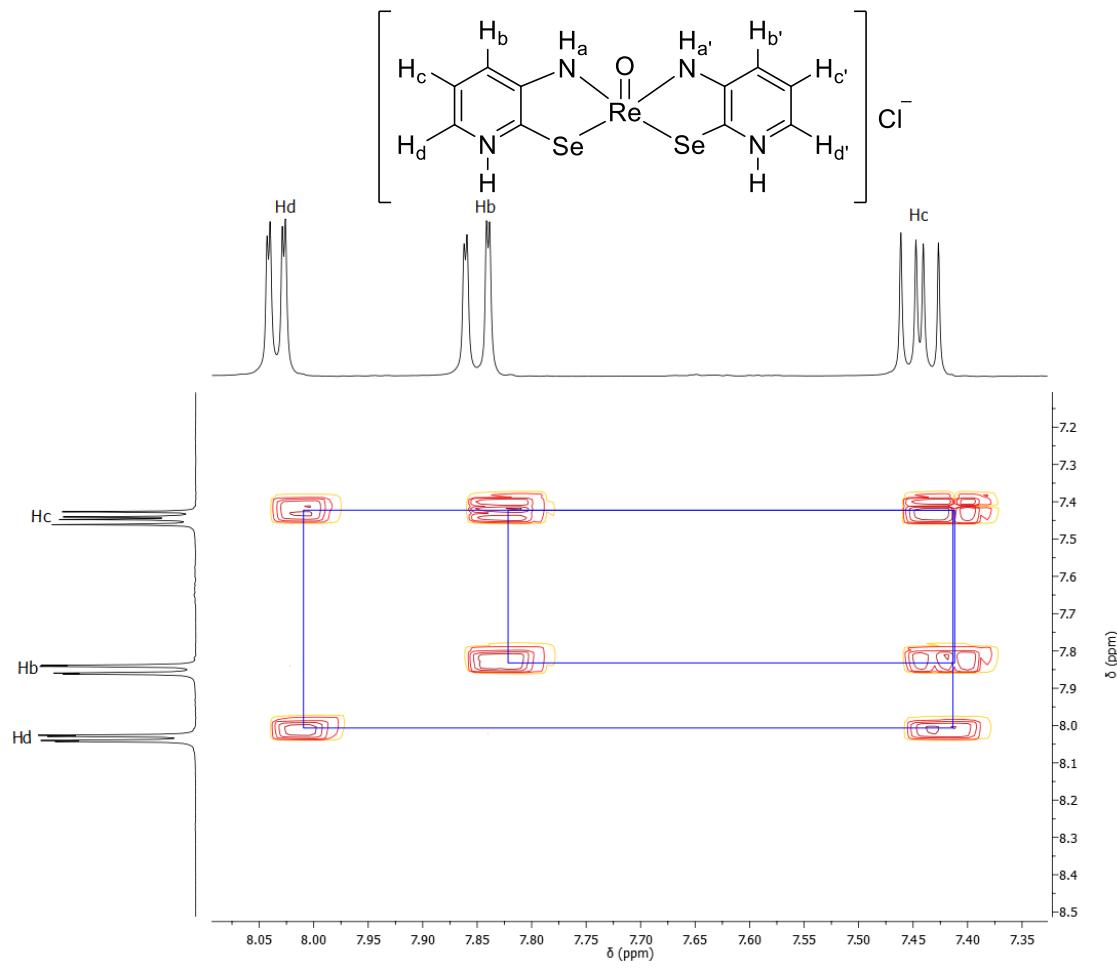


Figure S18. 2D COSY expansion spectrum of complex **6** ($\text{DMSO}-d_6$).

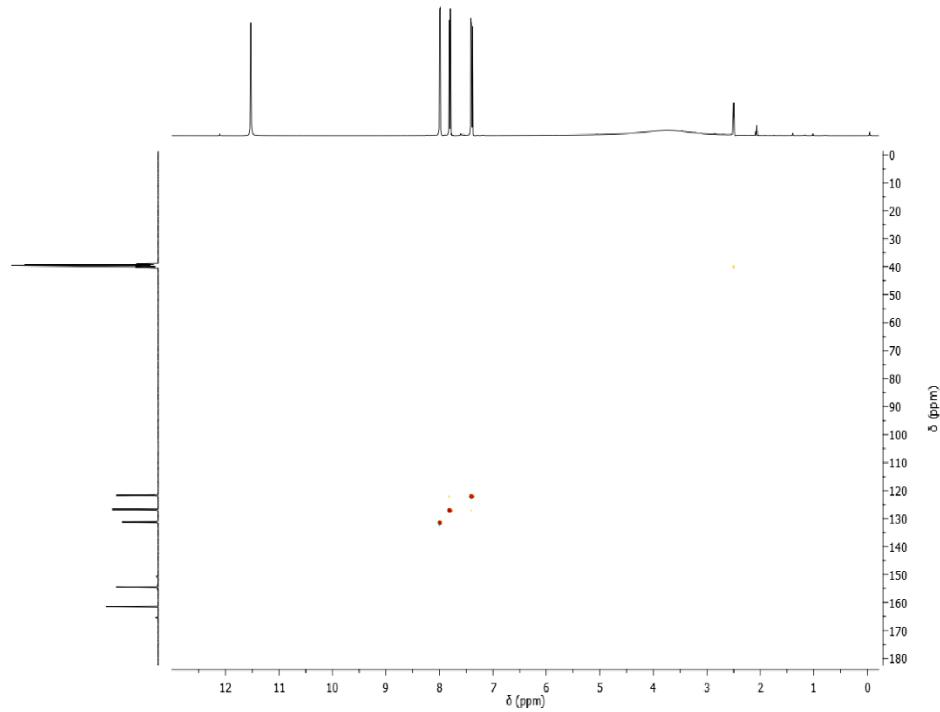


Figure S19. 2D HSQC NMR spectrum of complex **6** ($\text{DMSO}-d_6$).

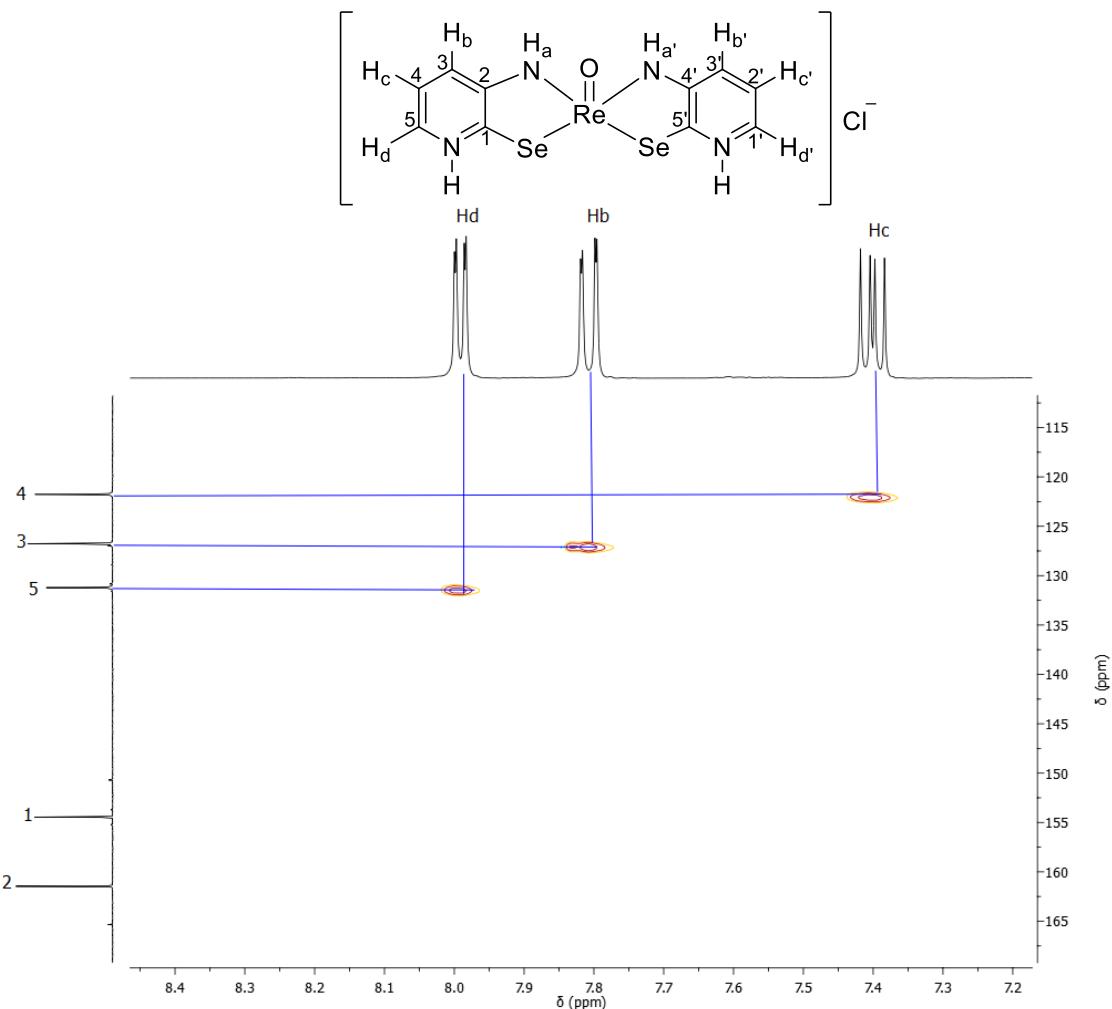


Figure S20. 2D HSQC expansion spectrum of complex **6** (DMSO- d_6).

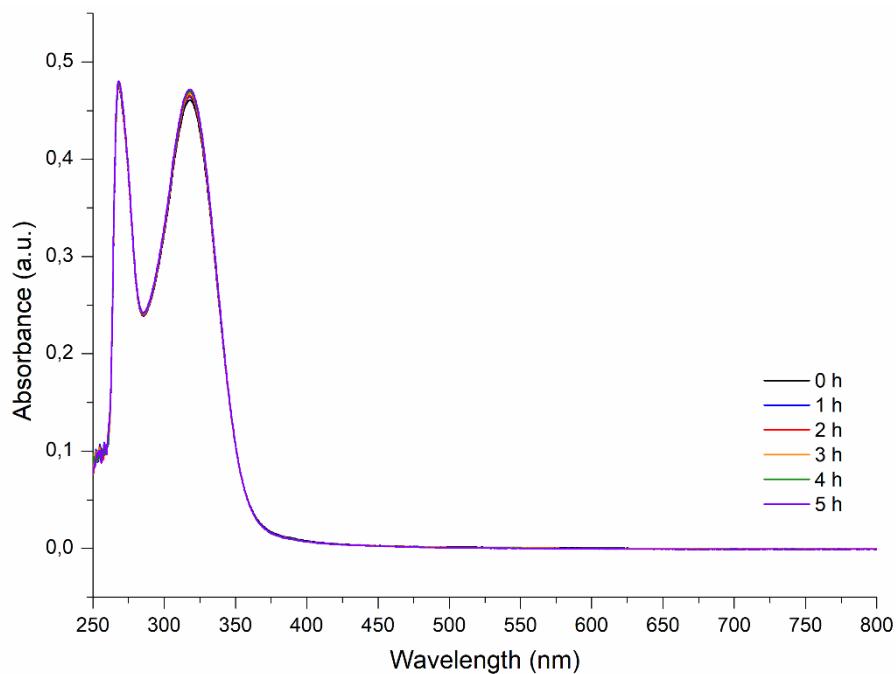


Figure S21. Electronic UV–Vis spectra for **L** in $H_2O/DMSO$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[] = 1 \times 10^{-4}$ M.

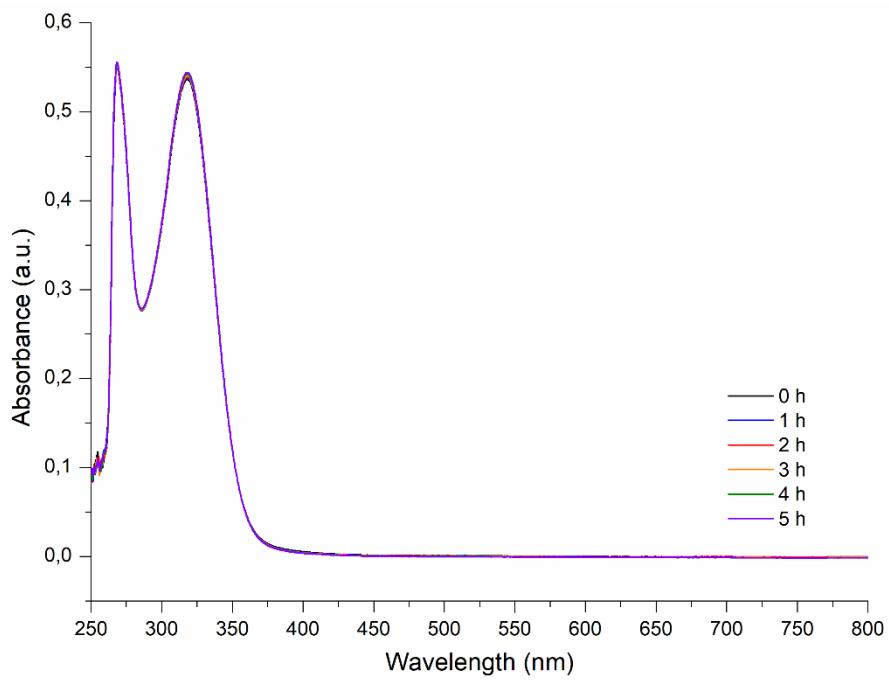


Figure S22. Electronic UV–Vis spectra for **1** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[\] = 1 \times 10^{-4} \text{ M}$.

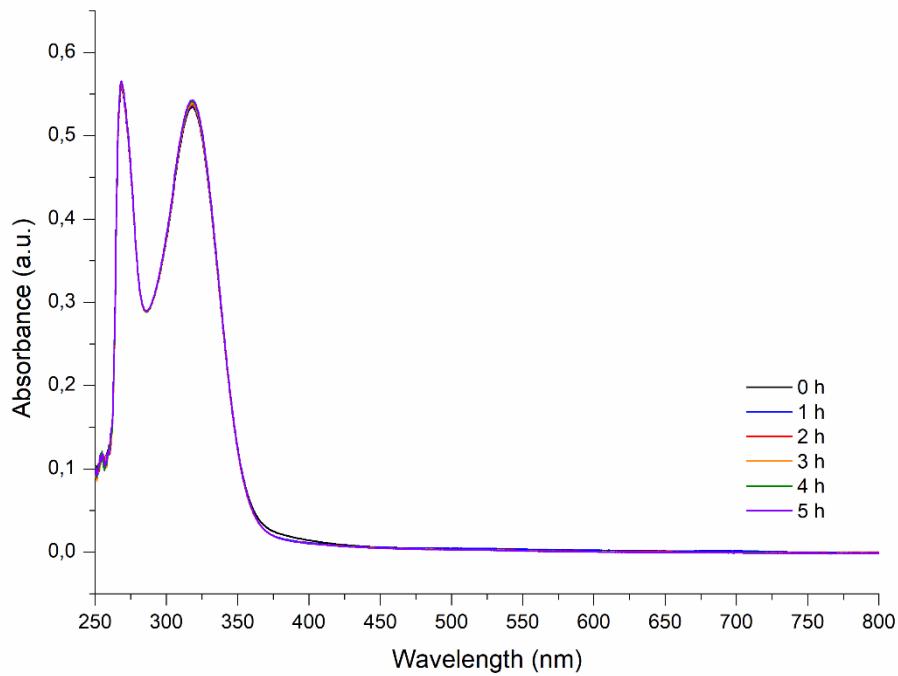


Figure S23. Electronic UV–Vis spectra for **2** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[\] = 1 \times 10^{-4} \text{ M}$.

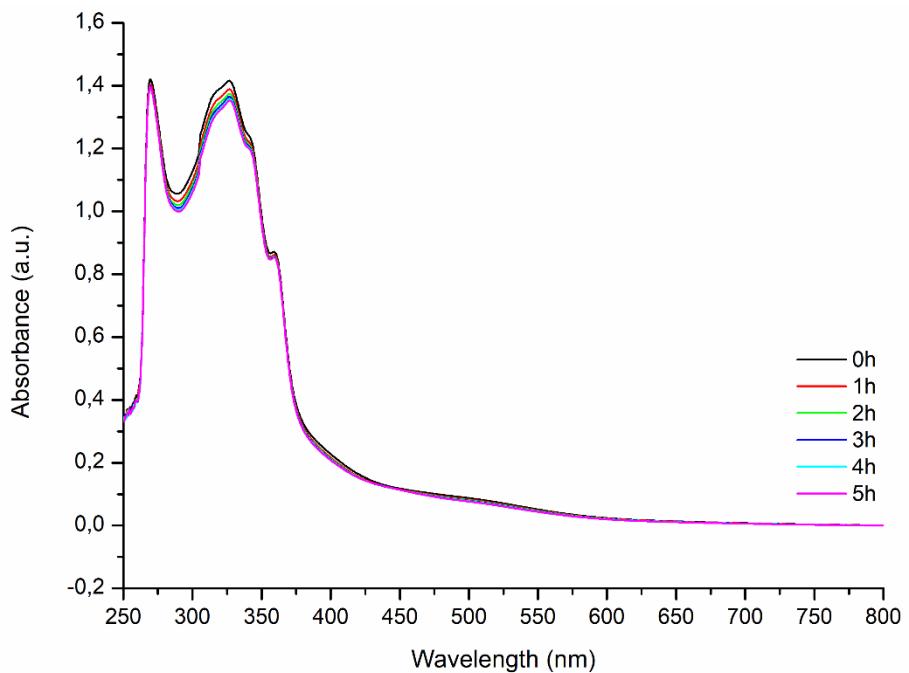


Figure S24. Electronic UV–Vis spectra for **3** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[] = 1 \times 10^{-4} \text{ M}$.

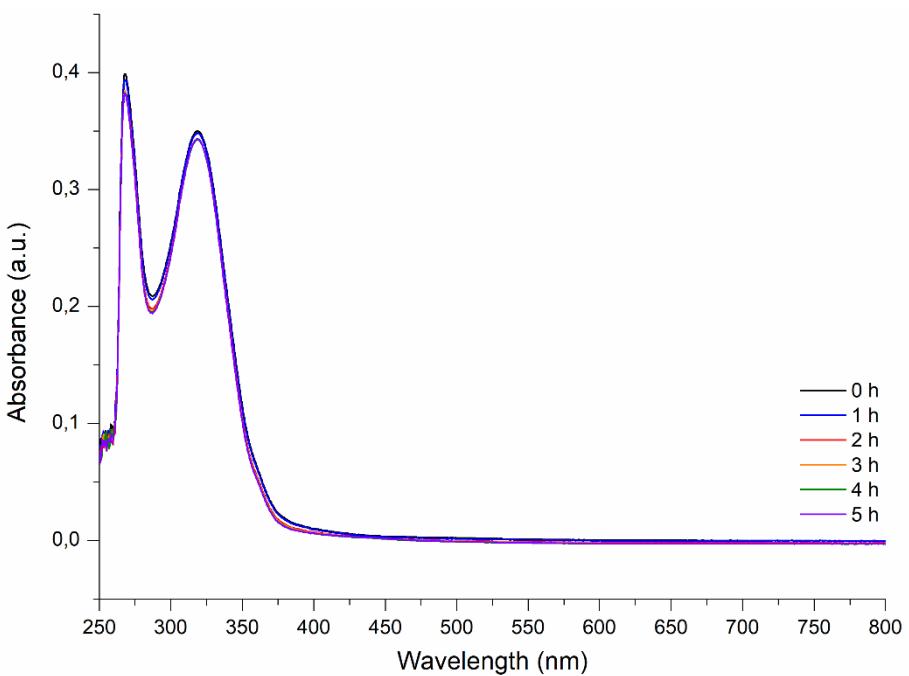


Figure S25. Electronic UV–Vis spectra for **4** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[] = 1 \times 10^{-4} \text{ M}$.

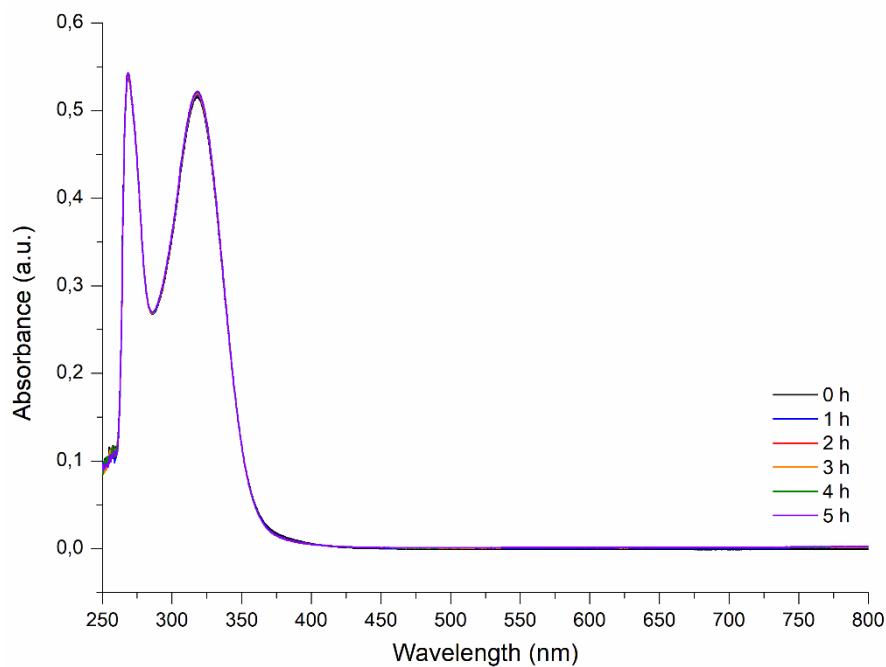


Figure S26. Electronic UV–Vis spectra for **5** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[] = 1 \times 10^{-4} \text{ M}$.

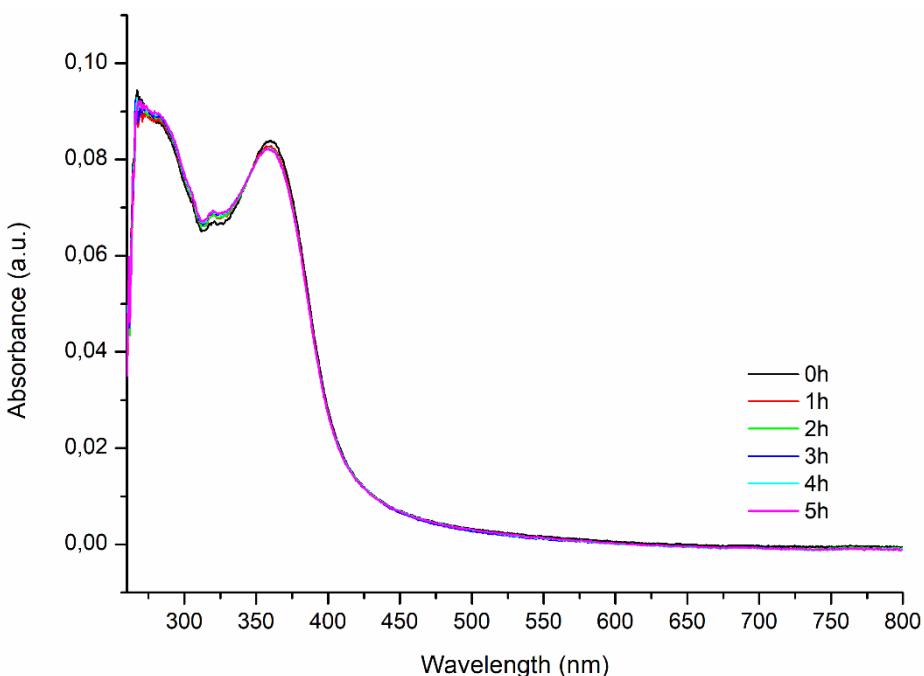


Figure S27. Electronic UV–Vis spectra for **6** in $\text{H}_2\text{O}/\text{DMSO}$ (4:1, v/v) solution, measured every 1 hour (total 5 hours); $[] = 1 \times 10^{-4} \text{ M}$.