

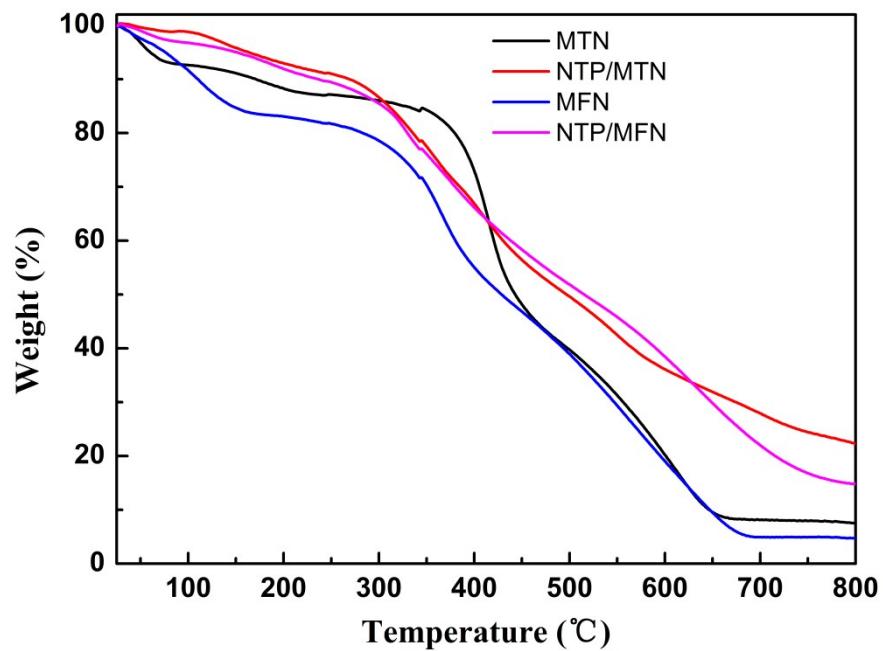
## Supporting Information

### Influence of Chemical Composition on Proton Conductivity of Microporous Organic Polymers Entrapped in Nitrilotrimethylphosphonic Acid

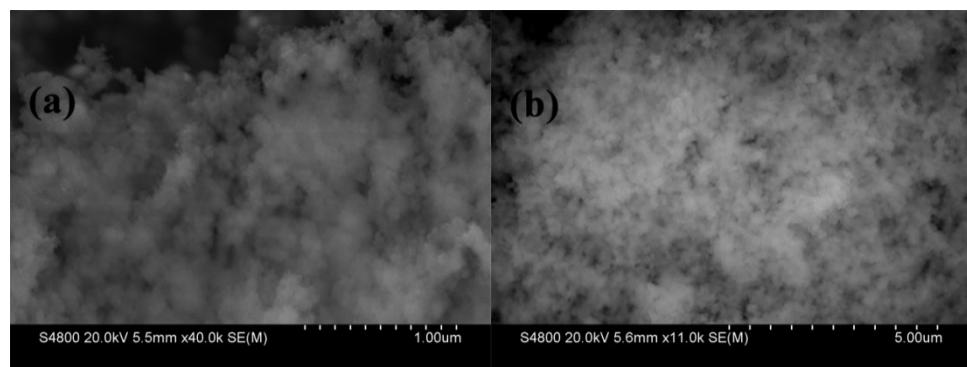
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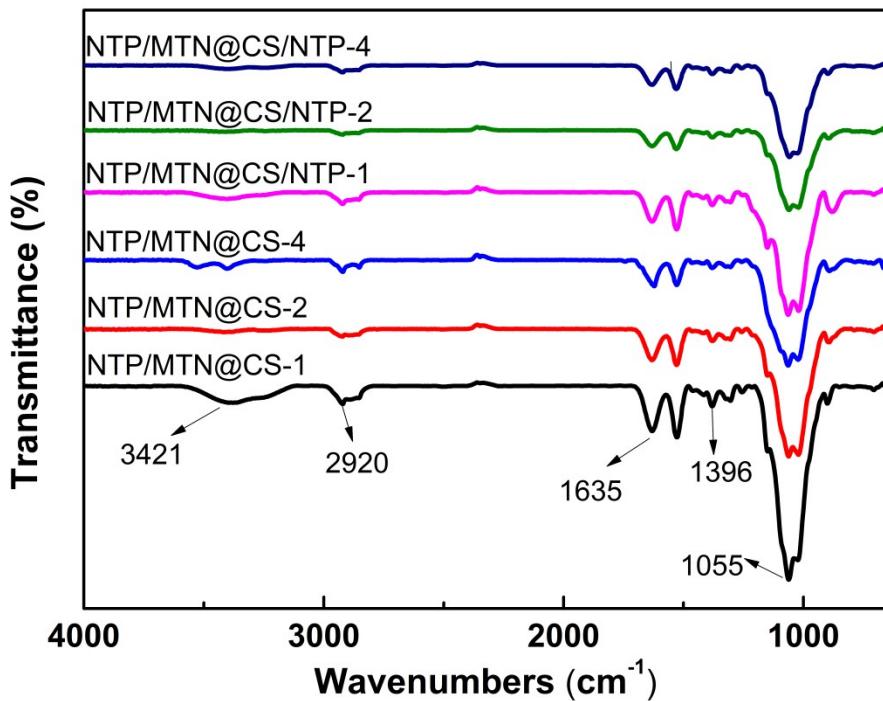
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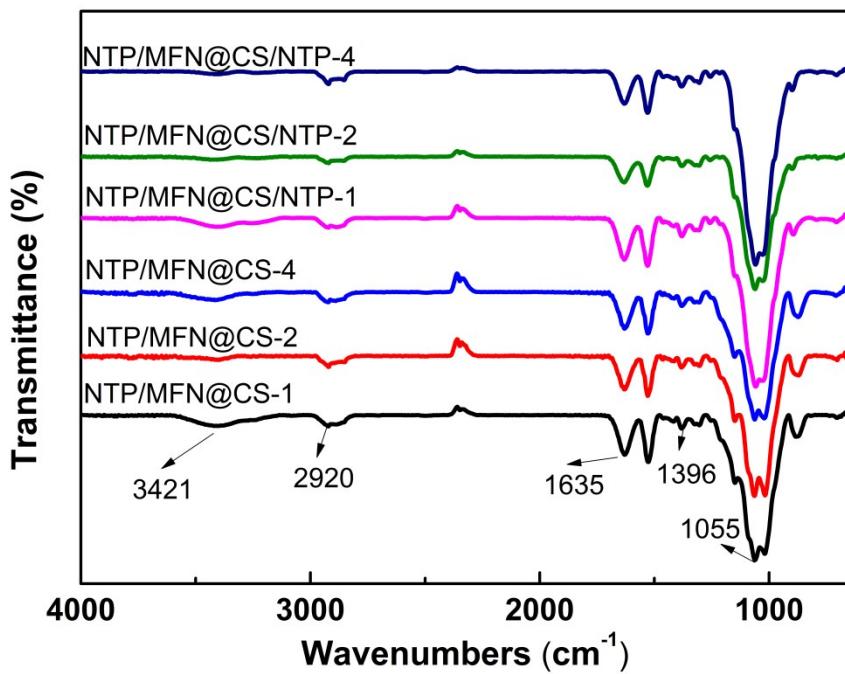
**Fig. S1** TGA curves of MTN, MFN, NTP/MTN and NTP/MFN.



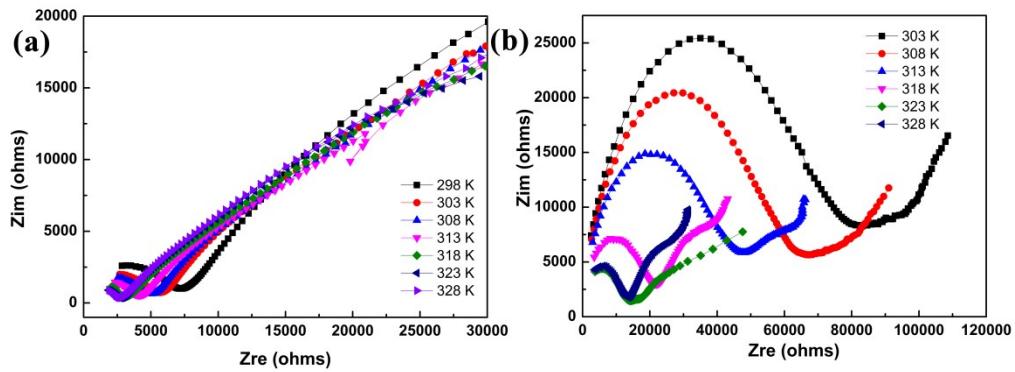
**Fig. S2** SEM images of (a) MTN and (b) NTP/MTN.



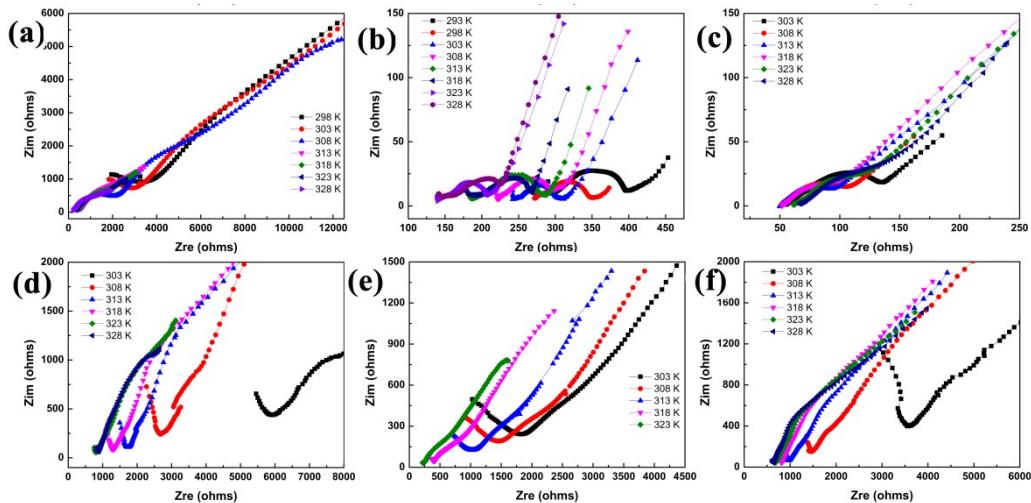
**Fig. S3** FT-IR spectra of NTP/MTN membranes.



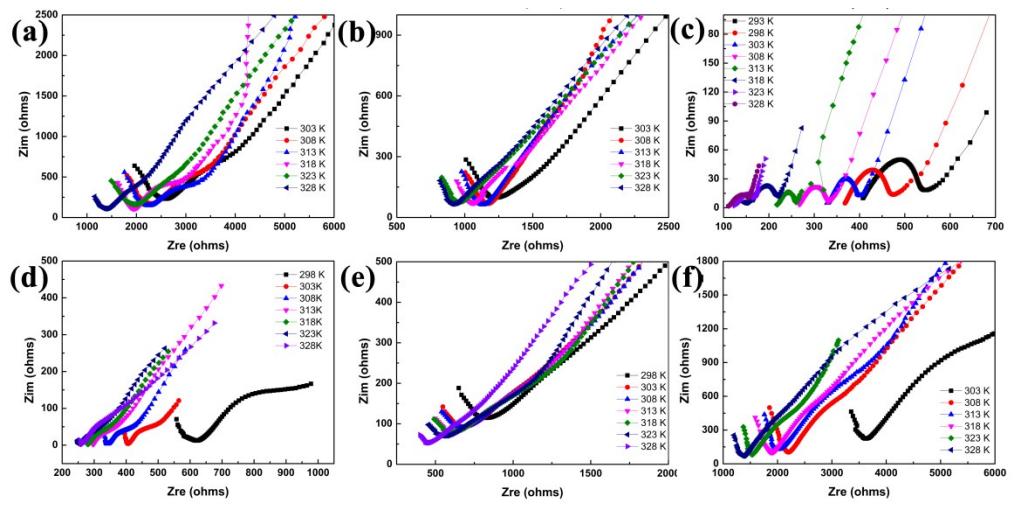
**Fig. S4** FT-IR spectra of NTP/MFN membranes.



**Fig. S5** Nyquist plots of CS membranes at (a) ~98% RH and (b) ~76% RH.



**Fig. S6** Nyquist plots of (a) NTP/MFN@CS-1, (b) NTP/MFN@CS-2 and (c) NTP/MFN@CS-4 (d) NTP/MFN@CS/NTP-1, (e) NTP/MFN@CS/NTP-2 and (f) NTP/MFN@CS/NTP-4 at ~98% RH.



**Fig. S7** Nyquist plots of (a) NTP/MTN@CS-1, (b) NTP/MTN@CS-2 and (c) NTP/MTN@CS-4 (d) NTP/MTN@CS/NTP-1, (e) NTP/MTN@CS/NTP-2 and (f) NTP/MTN@CS/NTP-4 at ~98% RH.

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**Table S1** CHN elemental analysis and ICP of MTN, NTP/MTN, MFN and NTP/MFN.

Samples	C (%)	H (%)	N (%)	S (%)	P (%)
MFN	33.83	4.78	46.13	0	0.0003
NTP/MFN	28.67	4.74	33.83	0	0.085
MTN	45.12	3.33	30.14	4.13	0.0002
NTP/MTN	32.44	3.35	23.62	3.01	0.083

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**Table S2** Porosity parameters of MTN, NTP/MTN, MFN and NTP/MFN.

Samples	$S_{BET}$ ( $\text{cm}^2 \text{ g}^{-1}$ )	Pore volume ( $\text{cm}^3 \text{ g}^{-1}$ )
MFN	230.07	0.38
NTP/MFN	42.33	0.11
MTN	785.09	1.13
NTP/MTN	185.07	0.69

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**Table S3** Ionic-exchange capability (IEC) and activation energy ( $E_a$ ) of membranes.

Sample	IEC (meq g <sup>-1</sup> )	$E_a$ (eV)	$\sigma$ (S cm <sup>-1</sup> )
MFN	0.15	/	/
NTP/MFN	6.58	/	/
MTN	0.045	/	/
NTP/MTN	5.78	/	/
NTP/MFN@CS-1	3.20	0.88	$1.9 \times 10^{-2}$
NTP/MFN@CS-2	4.40	0.83	$4.6 \times 10^{-2}$
NTP/MFN@CS-4	3.14	0.20	$5.5 \times 10^{-3}$
NTP/MFN@CS/NTP-1	0.53	0.70	$9.1 \times 10^{-4}$
NTP/MFN@CS/NTP-2	0.25	0.98	$3.4 \times 10^{-3}$
NTP/MFN@CS/NTP-4	0.32	0.58	$5.3 \times 10^{-4}$
NTP/MTN@CS-1	1.19	0.21	$1.1 \times 10^{-3}$
NTP/MTN@CS-2	1.25	0.14	$1.3 \times 10^{-3}$
NTP/MTN@CS-4	0.95	0.35	$3.4 \times 10^{-3}$
NTP/MTN@CS/NTP-1	1.19	0.21	$9.2 \times 10^{-3}$
NTP/MTN@CS/NTP-2	0.56	0.18	$1.1 \times 10^{-3}$
NTP/MTN@CS/NTP-4	0.34	0.48	$3.8 \times 10^{-4}$
CS	0.20	0.32	$1.1 \times 10^{-4}$

**Table S4** Proton conductivity of COF and membranes under ambient conditions.

Materials	Condition	$\sigma/\text{S cm}^{-1}$	Reference
I	55 °C, 99% RH	$6.17 \times 10^{-2}$	[1]
SZrTi	90°C, 100% RH	$2.9 \times 10^{-3}$	[2]
Fe-NH <sub>3</sub> -72h	80 °C, 95% RH	$1.8 \times 10^{-3}$	[3]
c-PBI-30	200 °C, 100% RH	$2.53 \times 10^{-1}$	[4]
FJU-80	80 °C, 98% RH	$1.05 \times 10^{-4}$	[5]
COF-1-Li	40 °C, 98% RH	$2.7 \times 10^{-2}$	[6]
COF-1-Na	40 °C, 98% RH	$2.5 \times 10^{-2}$	[6]
NH <sub>4</sub> PO <sub>3</sub> /MO <sub>2</sub>	175 °C, 99% RH	$8.5 \times 10^{-3}$	[7]
BaCe <sub>(0.85-x)</sub> Co <sub>x</sub> Gd <sub>0.15</sub> O <sub>3-δ</sub>	200 °C, 99% RH	$4.81 \times 10^{-3}$	[8]
Nafion/BP3-1.0	50 °C, 80% RH	$8.5 \times 10^{-2}$	[9]
SPS/POM-BC-30	25 °C, 100% RH	$5.3 \times 10^{-2}$	[10]
FJU-106	70 °C, 99% RH	$1.8 \times 10^{-2}$	[11]
h-BN	55 °C, 99% RH	$6.17 \times 10^{-2}$	[12]
Am3-sNCC-5	40 °C, 100% RH	$4.3 \times 10^{-2}$	[13]
NUS-10(R)@PVDF-50	25 °C, 100% RH	$5.16 \times 10^{-3}$	[14]
NTP/MFN@CS-2	50 °C, 98% RH	$4.6 \times 10^{-2}$	This work

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