

Supporting Information

Cycloaddition of carbon dioxide and epoxides catalyzed by rare earth metal complexes bearing Trost ligand

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Solid state structures of complexes 2-5

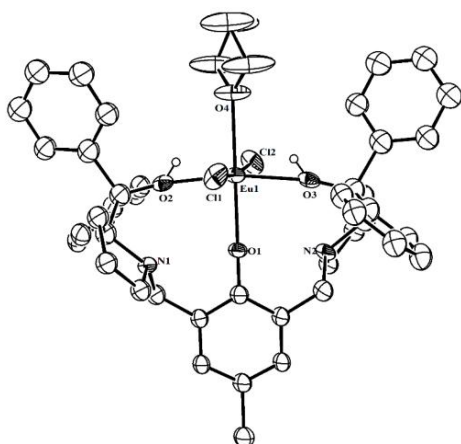


Fig. S1 Solid state structure of complex **2**. Thermal ellipsoids are drawn at the 30% probability level, for clarity, all other hydrogen atoms and solvent molecules except alcohol hydroxyl hydrogen. Selected bond lengths [Å] and bond angles [deg]: Eu(1)-O(1) 2.176(8), Eu(1)-O(2) 2.196(18), Eu(1)-O(3) 2.196(19), Eu(1)-O(4) 2.442(19), Eu(1)-Cl(1) 2.727(6), Eu(1)-Cl(2) 2.727(6); O(1)-Eu(1)-O(2) 85.1(4), O(1)-Eu(1)-O(3) 85.1(4), O(1)-Eu(1)-O(4) 180.0(4), O(2)-Eu(1)-O(3) 170.2(8), O(2)-Eu(1)-O(4) 94.9(4), O(3)-Eu(1)-O(4) 94.9(4), Cl(1)-Eu(1)-Cl(2) 172.3(3), O(1)-Eu(1)-Cl(1) 93.9(2), O(1)-Eu(1)-Cl(2) 93.9(2), O(2)-Eu(1)-Cl(1) 94.1(4), O(2)-Eu(1)-Cl(2) 86.6(4), O(3)-Eu(1)-Cl(1) 94.1(4), O(3)-Eu(1)-Cl(2) 86.6(4), O(4)-Eu(1)-Cl(1) 86.1(2), O(4)-Eu(1)-Cl(2) 86.1(2).

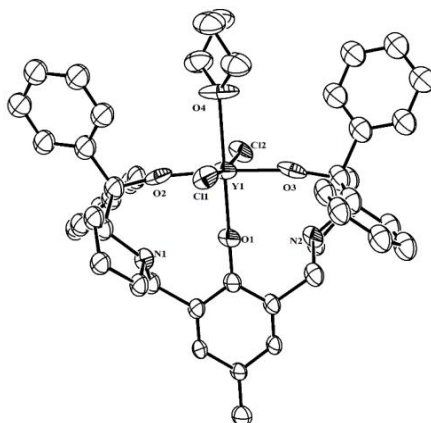


Fig. S2 Solid state structure of complex **3**. Thermal ellipsoids are drawn at the 30% probability level, for clarity, all other hydrogen atoms and solvent molecules except alcohol hydroxyl hydrogen. Selected bond lengths [Å] and bond angles [deg]: Y(1)-O(1) 2.098(14), Y(1)-O(2) 2.086(15), Y(1)-O(3) 2.086(15), Y(1)-O(4) 2.422(17), Y(1)-Cl(1) 2.679(5), Y(1)-Cl(2) 2.679(5); O(1)-Y(1)-O(2) 87.7(4), O(1)-Y(1)-O(3) 87.7(4), O(1)-Y(1)-O(4) 180.0, O(2)-Y(1)-O(3) 175.3(7), O(2)-Y(1)-O(4) 92.3(4), Cl(1)-Y(1)-Cl(2) 171.2(3), O(1)-Y(1)-Cl(1) 94.39(13), O(1)-Y(1)-Cl(2) 94.39(13),

O(2)-Y(1)-Cl(1) 93.7(4), O(2)-Y(1)-Cl(2) 86.7(4), O(3)-Y(1)-Cl(1) 93.7(4),
O(3)-Y(1)-Cl(2) 86.7(4), O(4)-Y(1)-Cl(1) 85.61(13), O(4)-Y(1)-Cl(2) 85.61(13).

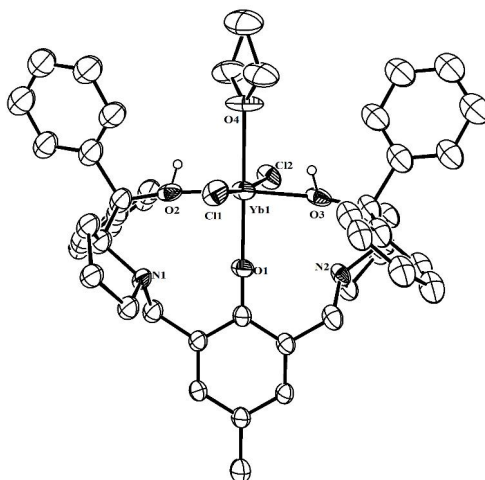


Fig. S3 Solid state structure of complex **4**. Thermal ellipsoids are drawn at the 30% probability level, for clarity, all other hydrogen atoms and solvent molecules except alcohol hydroxyl hydrogen. Selected bond lengths [Å] and bond angles [deg]: Yb1-O1 2.126(11), Yb1-O4 2.369(11), Yb1-O2 2.075(11), Yb1-Cl1 2.651(4), Yb1-O3 2.075(11), Yb1-Cl2 2.651(4), O1-Yb1-O2 86.9(3), O1-Yb1-Cl1 93.53(10), O1-Yb1-O3 86.9(3), O1-Yb1-Cl2 93.53(10), O1-Yb1-O4 180.0, O2-Yb1-Cl1 94.4(3), O2-Yb1-O3 173.8(5), O2-Yb1-Cl2 86.0(3), O2-Yb1-O4 93.1(3), O3-Yb1-Cl1 94.3(3), O3-Yb1-O4 93.1(3), O3-Yb1-Cl2 86.0(3), Cl1-Yb1-Cl2 172.9(2), O4-Yb1-Cl1 86.47(10), O4-Yb1-Cl2 86.47(10).

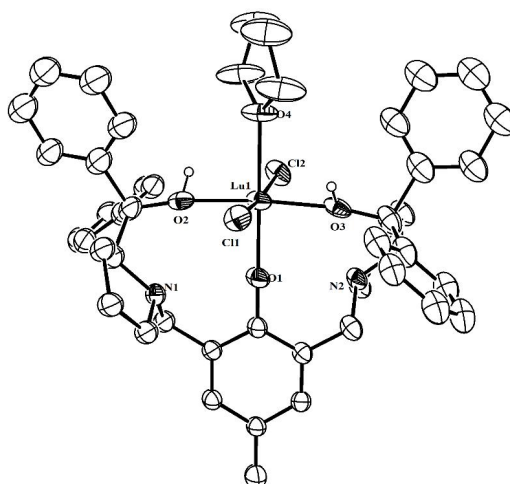


Fig. S4 Solid state structure of complex **5**. Thermal ellipsoids are drawn at the 30% probability level, for clarity, all other hydrogen atoms and solvent molecules except alcohol hydroxyl hydrogen. Selected bond lengths [Å] and bond angles [deg]: Lu1-O1 2.125(8), Lu1-O4 2.359(9), Lu1-O2 2.093(8), Lu1-Cl1 2.629(3), Lu1-O3 2.093(8), Lu1-Cl2 2.629(3), O1-Lu1-O2 86.9(3), O1-Lu1-Cl1 93.60(8), O1-Lu1-O3 86.9(3), O1-Lu1-Cl2 93.60(8), O1-Lu1-O4 180.0, O2-Lu1-Cl1 86.51(19), O2-Lu1-O3

173.8(5), O2-Lu1-Cl2 93.81(19), O2-Lu1-O4 93.1(3), O3-Lu1-Cl1 86.51(19), O3-Lu1-O4 93.1(3), O3-Lu1-Cl2 93.81(19), Cl1-Lu1-Cl2 172.81(15), O4-Lu1-Cl1 86.5(5), O4-Lu1-Cl2 86.3(5).

Crystallographic data for complexes 1-5

Complexes	1	2	3
Empirical formula	C ₄₇ H ₅₁ Cl ₂ N ₂ O ₄ Sm·THF	C ₅₅ H ₆₉ Cl ₂ N ₂ O ₆ Eu·2THF	C ₄₇ H ₅₃ Cl ₂ N ₂ O ₄ Y·THF
Formula weight	929.15	1072.80	869.72
Temperature/K	296.15	296(2)	120.01
Crystal system	orthorhombic	tetragonal	tetragonal
Space group	P2 ₁ 2 ₁ 2 ₁	P4 ₃ 2 ₁ 2	P4 ₃ 2 ₁ 2
a/Å	12.9665(12)	12.9908(4)	12.8764(6)
b/Å	12.9771(12)	12.9908(4)	12.8764(6)
c/Å	30.056(3)	30.0245(16)	29.6614(17)
$\alpha/^\circ$	90	90	90
$\beta/^\circ$	90	90	90
$\gamma/^\circ$	90	90	90
Volume/Å ³	5057.4(8)	5067.0(4)	4917.9(5)
Z	4	4	4
ρ_{calc} g/cm ³	1.220	1.406	1.175
μ/mm^{-1}	1.305	1.361	1.334
F(000)	1900.0	2225.0	1816.0
Crystal size/mm ³	0.3/0.2/0.2	0.3/0.2/0.2	0.3/0.2/0.2
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
Reflections collected	139369	96947	80557
Independent reflections	9582 [R _{int} = 0.1725]	5853 [R _{int} = 0.0516]	4188 [R _{int} = 0.1297]
Goodness-of-fit on F ²	1.037	1.077	1.040
Final R indexes [I $\geq$ 2 σ (I)]	R ₁ = 0.1025, wR ₂ = 0.2417	R ₁ = 0.1259, wR ₂ = 0.2777	R ₁ = 0.1392, wR ₂ = 0.3606
Largest diff. peak/hole / e Å ⁻³	2.73/-2.49	5.21/-3.56	1.30/-1.35
Flack parameter	0.122(9)	0.079(9)	0.080(11)

Complexes	4	5
Empirical formula	C ₄₇ H ₅₃ Cl ₂ N ₂ O ₄ Yb·THF	C ₅₅ H ₆₉ Cl ₂ N ₂ O ₆ Lu·2THF
Formula weight	953.85	1100.04
Temperature/K	296.15	296.15
Crystal system	tetragonal	tetragonal
Space group	P4 ₃ 2 ₁ 2	P4 ₃ 2 ₁ 2
a/Å	12.9011(5)	12.8865(5)
b/Å	12.9011(5)	12.8865(5)
c/Å	29.9105(16)	29.8955(17)
$\alpha/^\circ$	90	90
$\beta/^\circ$	90	90
$\gamma/^\circ$	90	90
Volume/Å ³	4978.3(5)	4964.5(4)
Z	4	4
ρ_{calc} g/cm ³	1.273	1.4717
μ/mm^{-1}	2.025	2.148
F(000)	1940.0	2264.8
Crystal size/mm ³	0.3/0.25/0.2	0.2/0.2/0.1
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
Reflections collected	111202	97816
Independent reflections	4736 [R _{int} = 0.0896]	5742 [R _{int} = 0.0933]
Goodness-of-fit on F ²	1.048	1.130
Final R indexes [I >= 2 σ (I)]	R ₁ = 0.0671, wR ₂ = 0.1542	R ₁ = 0.0709, wR ₂ = 0.1555
Largest diff. peak/hole / e Å ⁻³	2.00/-1.43	3.77/-2.99
Flack parameter	0.038(7)	0.06(2)

The IR spectra of complexes 2-5.

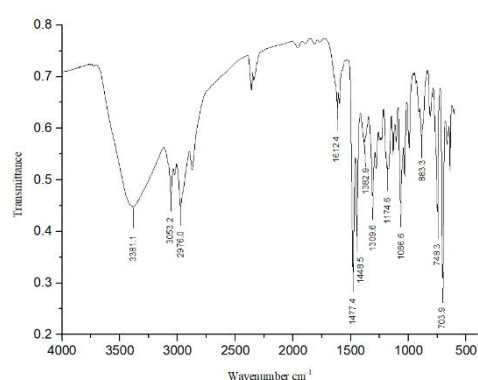


Fig. S5 IR spectrum of complex 2

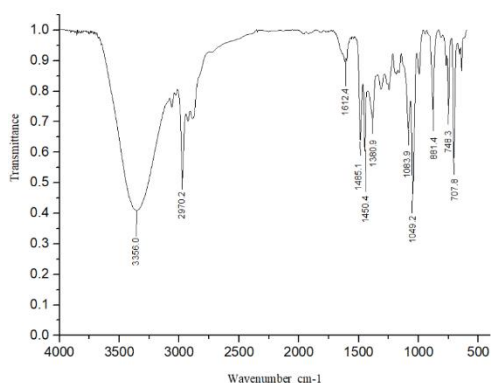


Fig. S6 IR spectrum of complex 3

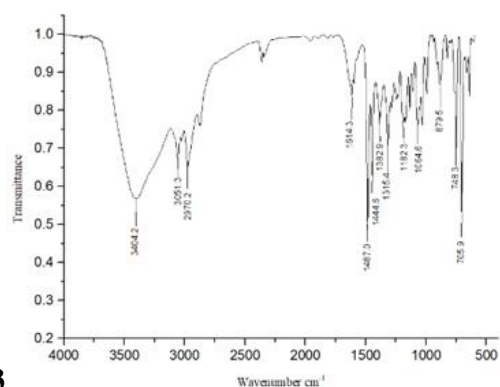


Fig. S7 IR spectrum of complex 4

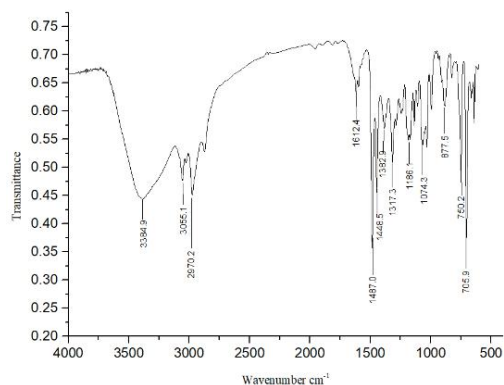


Fig. S8 IR spectrum of complex 5

NMR spectra of complexes 3 and 5

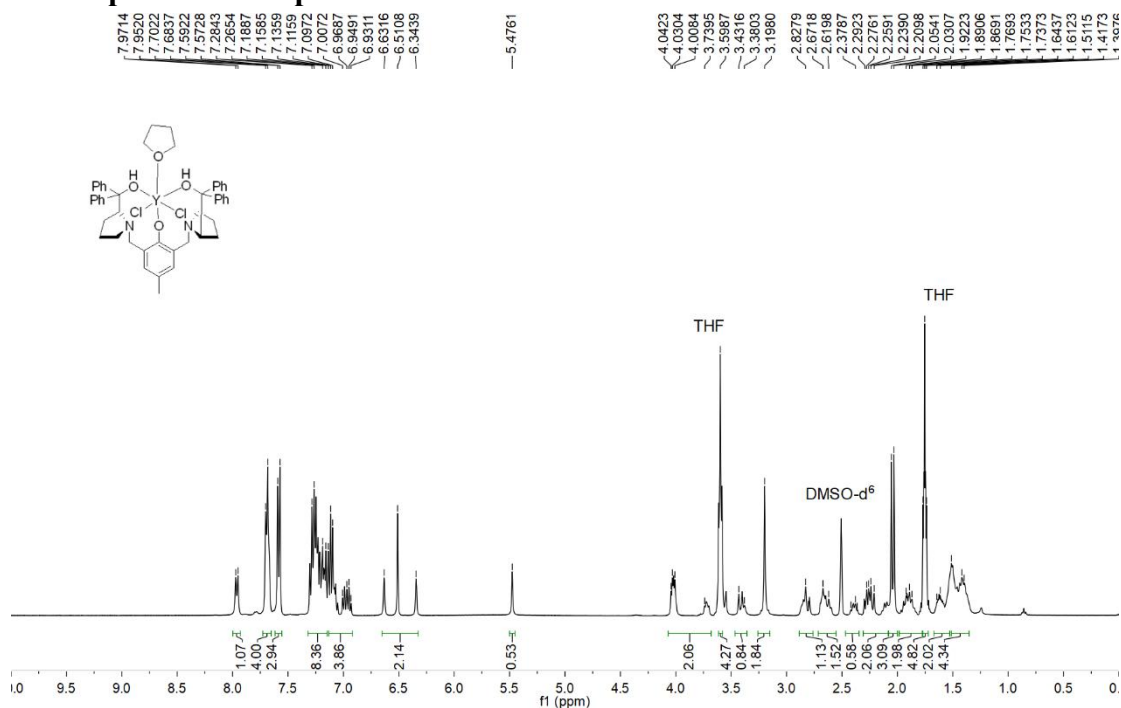


Fig. S9 ^1H NMR spectrum of complex 3 in DMSO-d_6

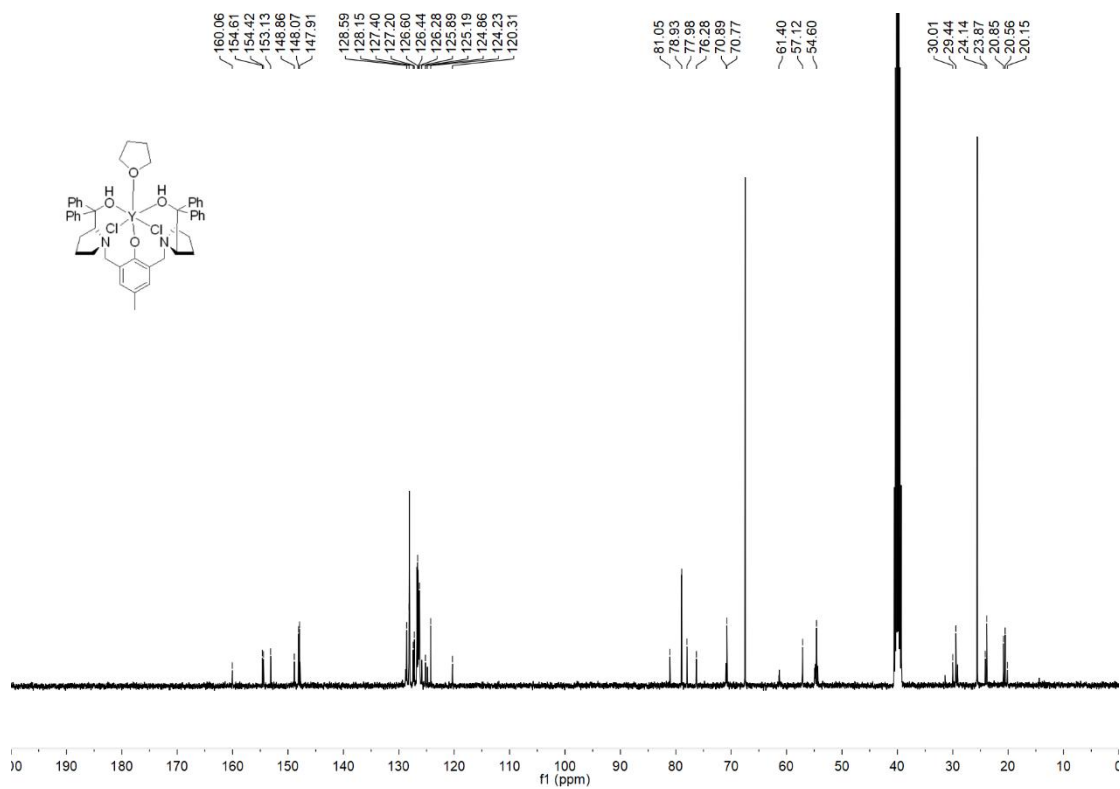


Fig. S10 ¹³C NMR spectrum of complex **3** in DMSO-*d*₆

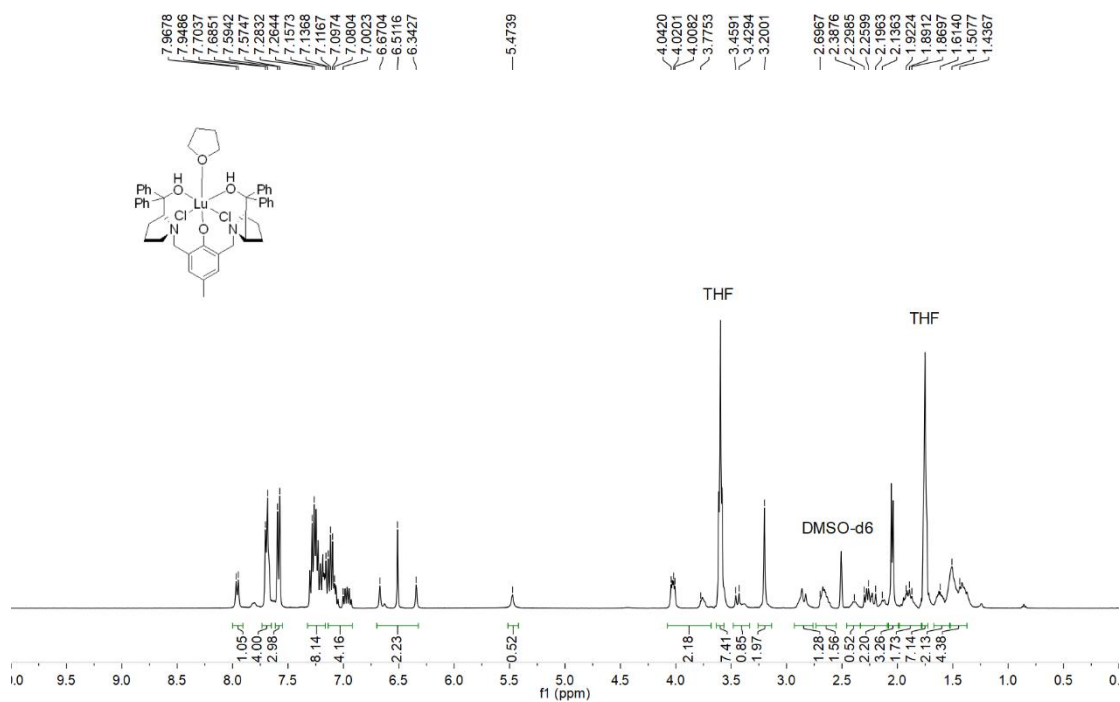


Fig. S11 ¹H NMR spectrum of complex **5** in DMSO-*d*₆

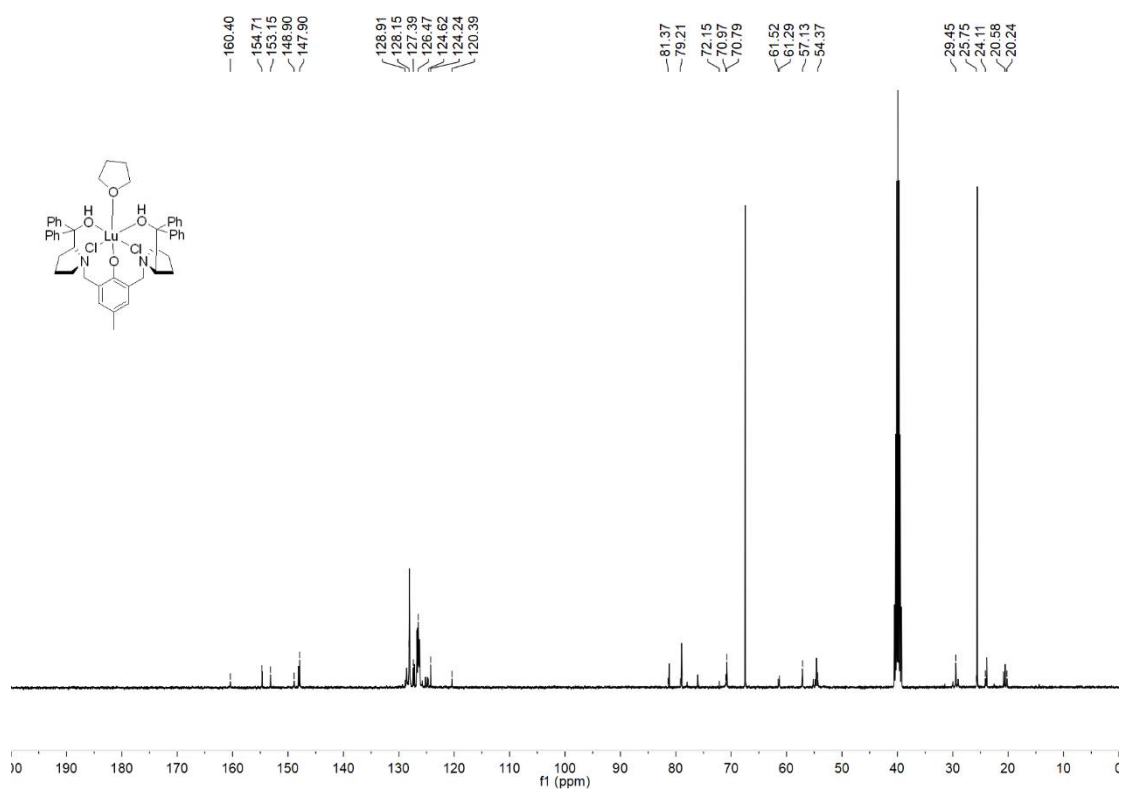


Fig. S12 ^{13}C NMR spectrum of complex **5** in $\text{DMSO}-d_6$

NMR spectra of the cyclic carbonates

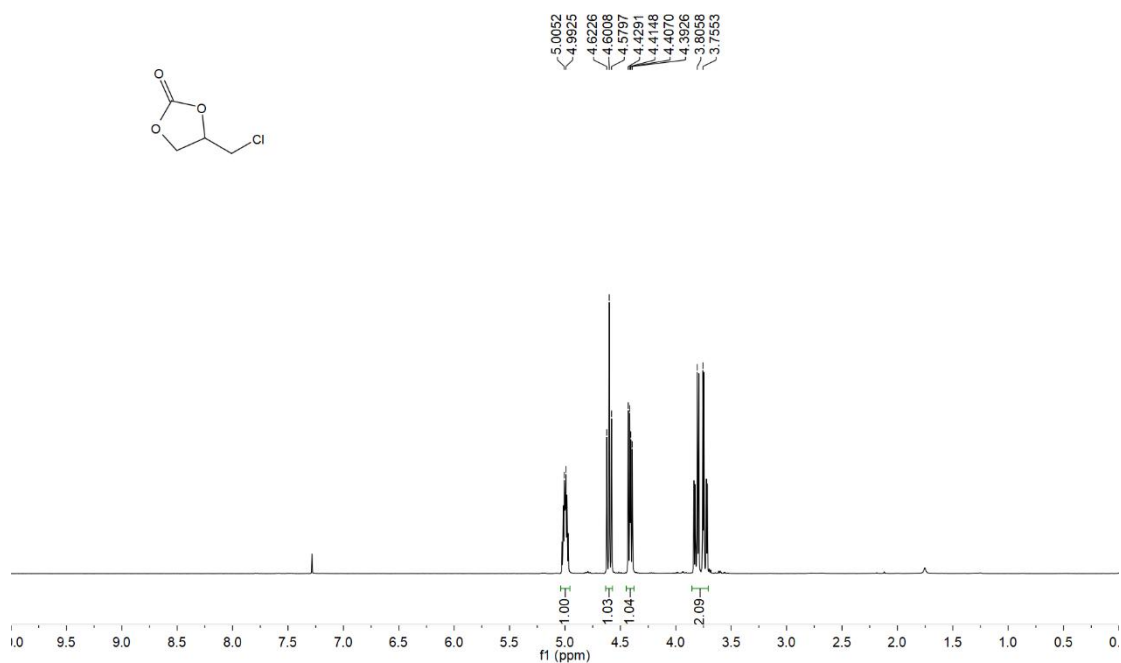


Fig. S13 ^1H NMR spectrum of **7a** in CDCl_3

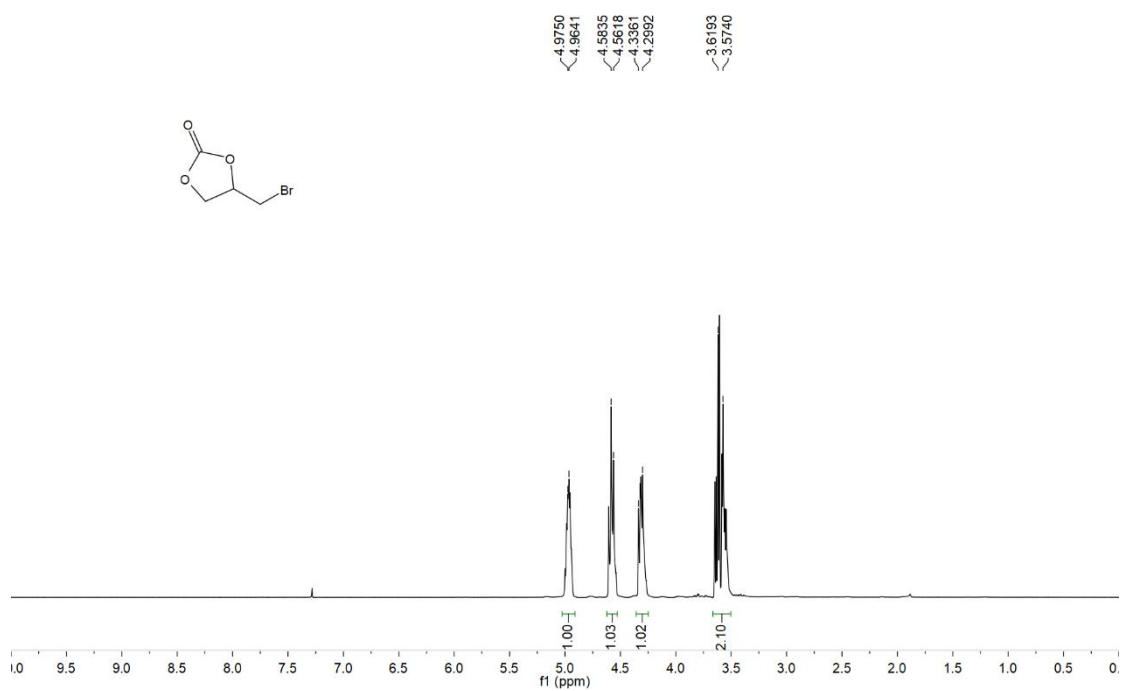


Fig. S14 ¹H NMR spectrum of **7b** in CDCl₃

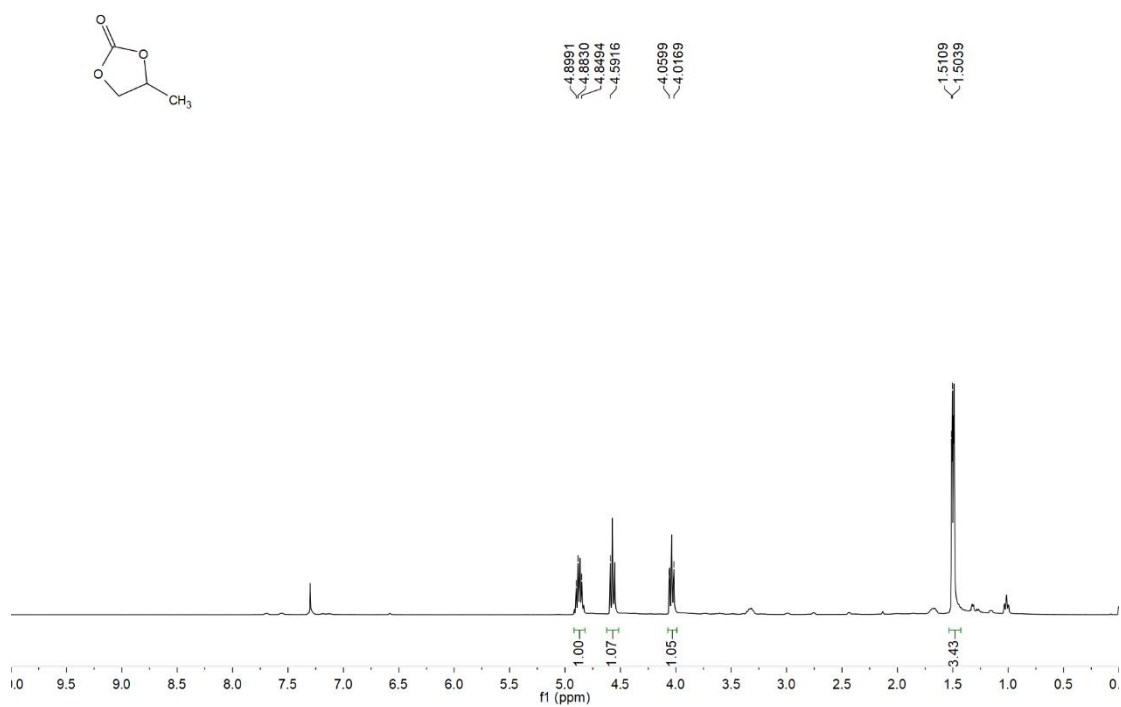


Fig. S15 ¹H NMR spectrum of **7c** in CDCl₃

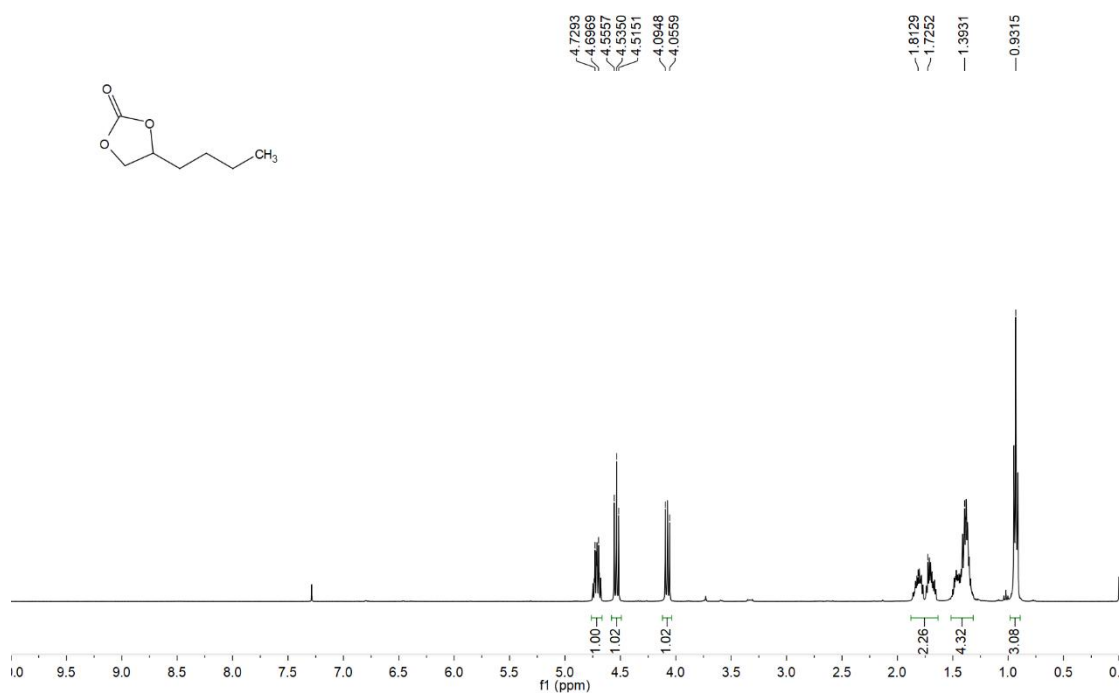


Fig. S16 ¹H NMR spectrum of **7d** in CDCl₃

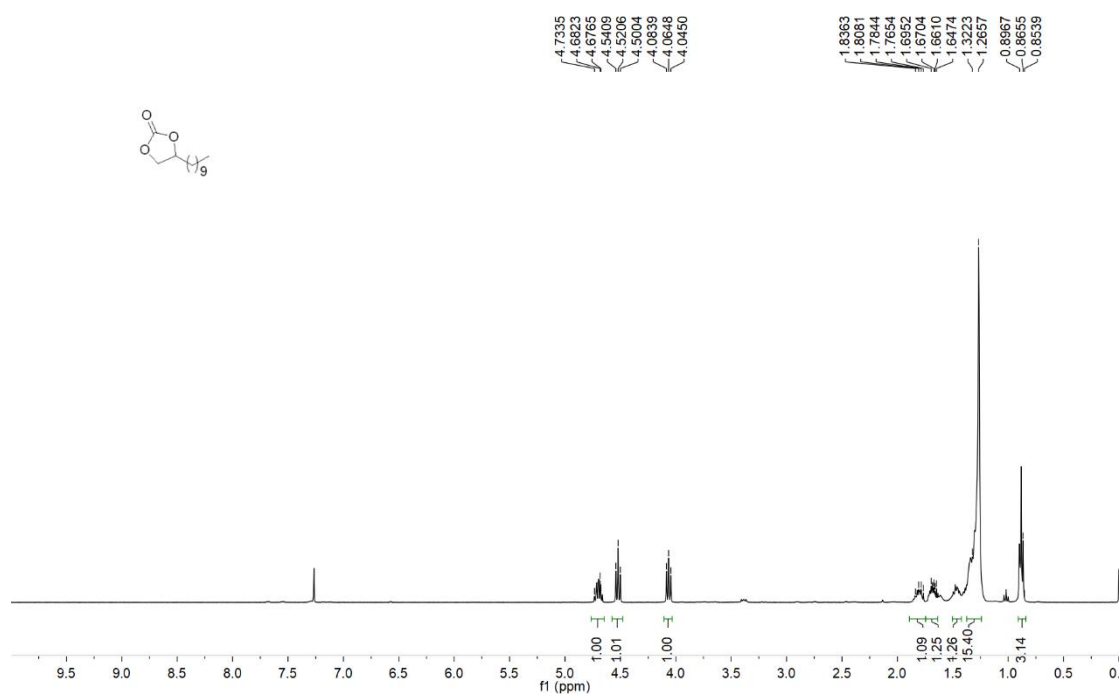


Fig. S17 ¹H NMR spectrum of **7e** in CDCl₃

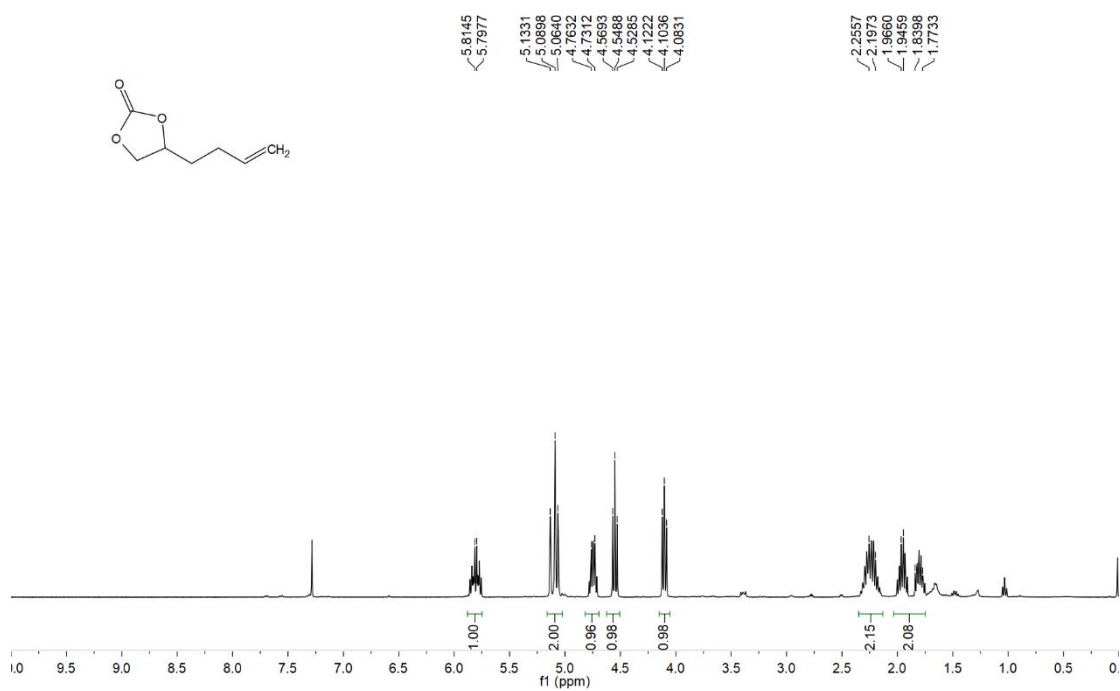


Fig. S18 ¹H NMR spectrum of **7f** in CDCl₃

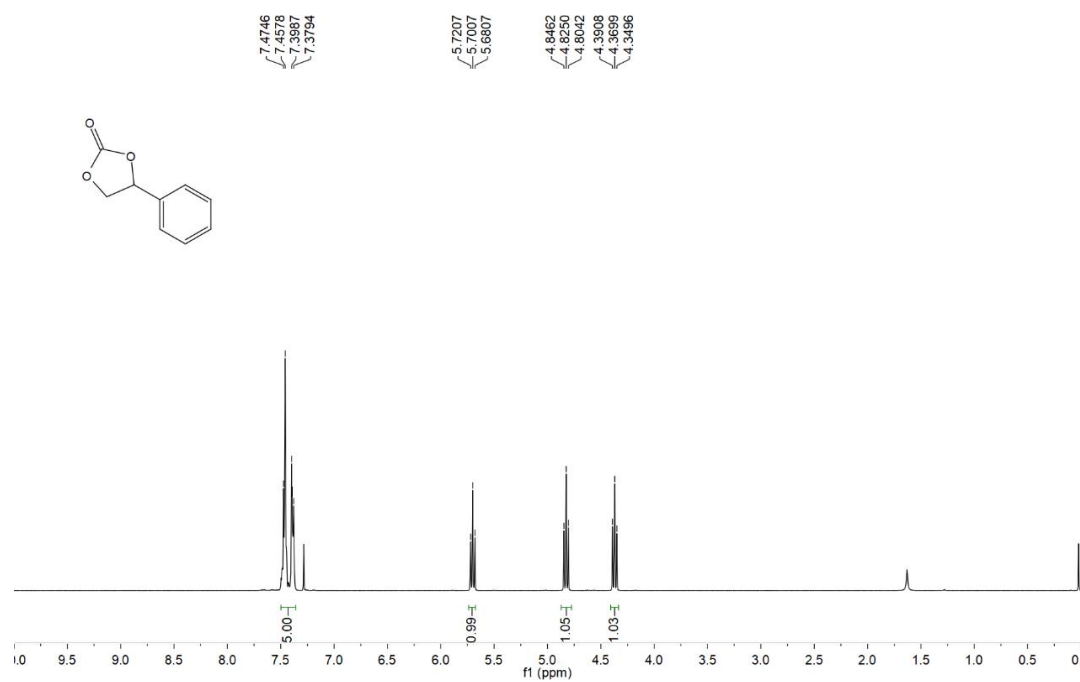


Fig. S19 ¹H NMR spectrum of **7g** in CDCl₃

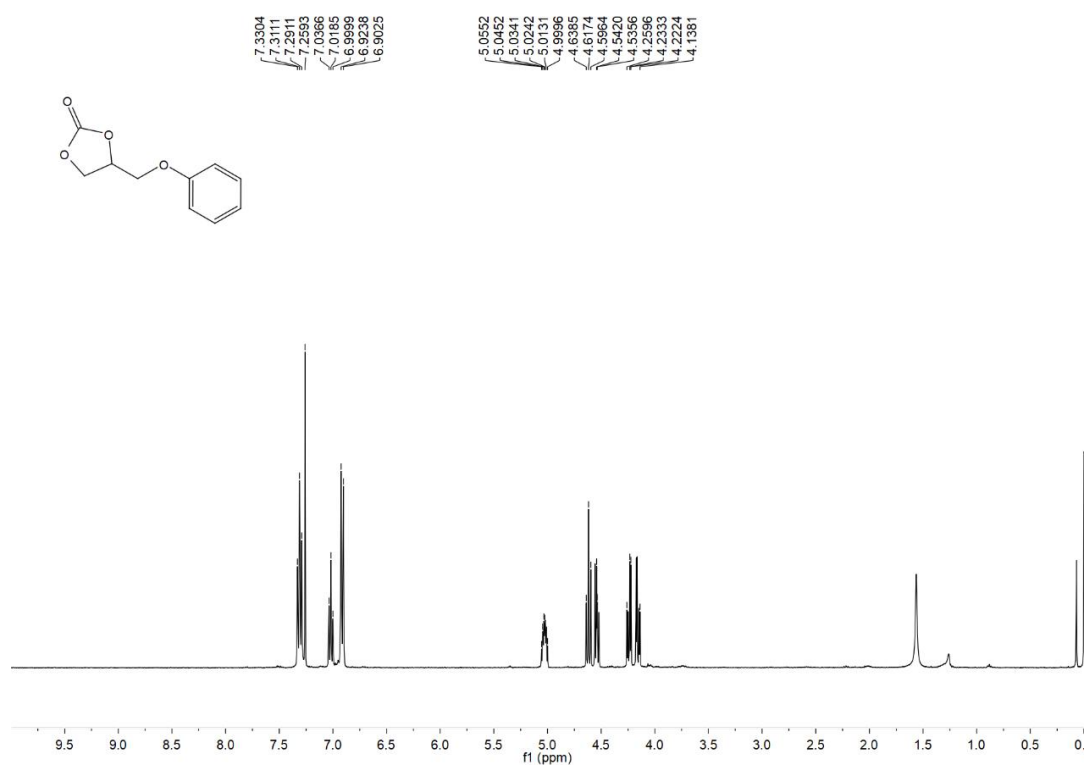


Fig. S20 ¹H NMR spectrum of **7h** in CDCl₃

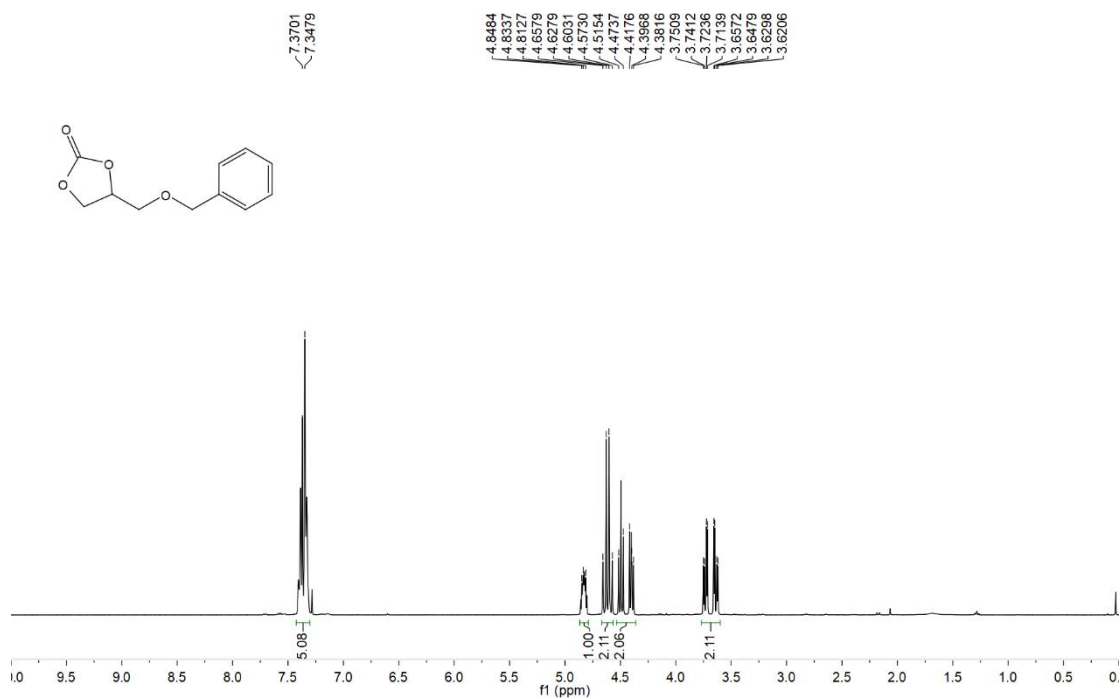


Fig. S21 ¹H NMR spectrum of **7i** in CDCl₃

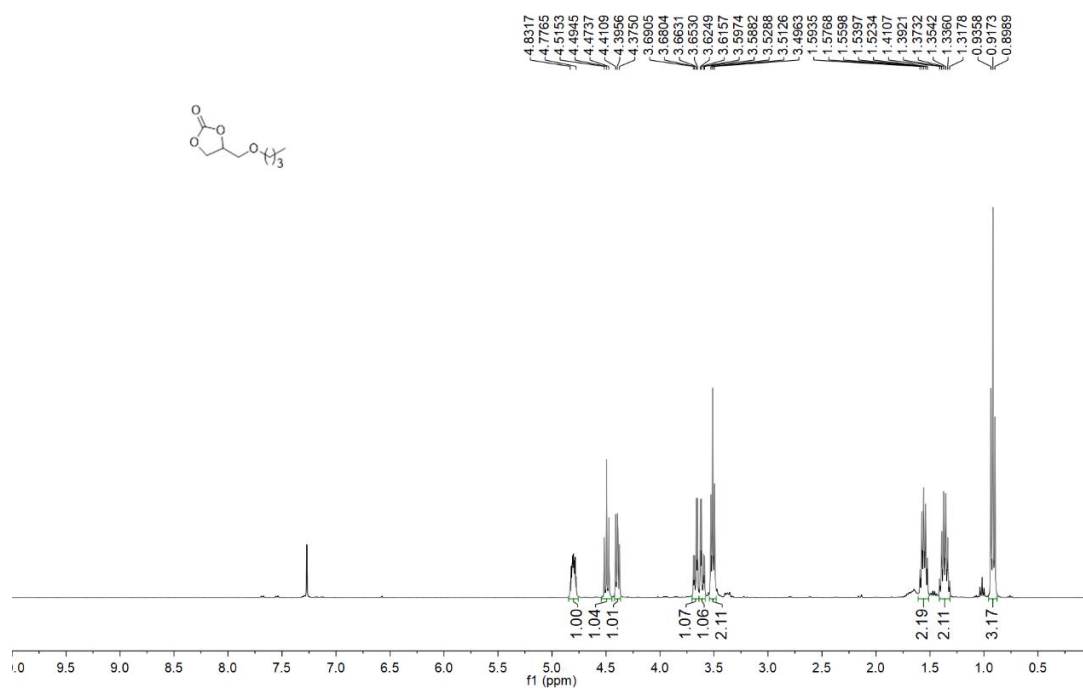


Fig. S22 ¹H NMR spectrum of **7j** in CDCl₃

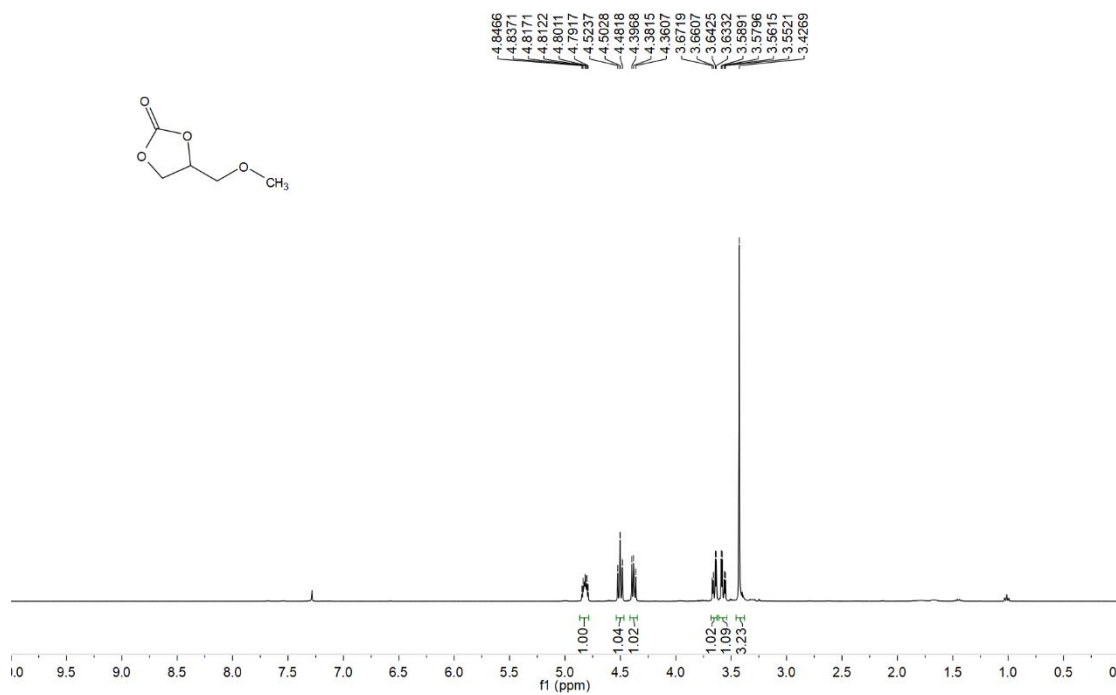


Fig. S23 ¹H NMR spectrum of **7k** in CDCl₃

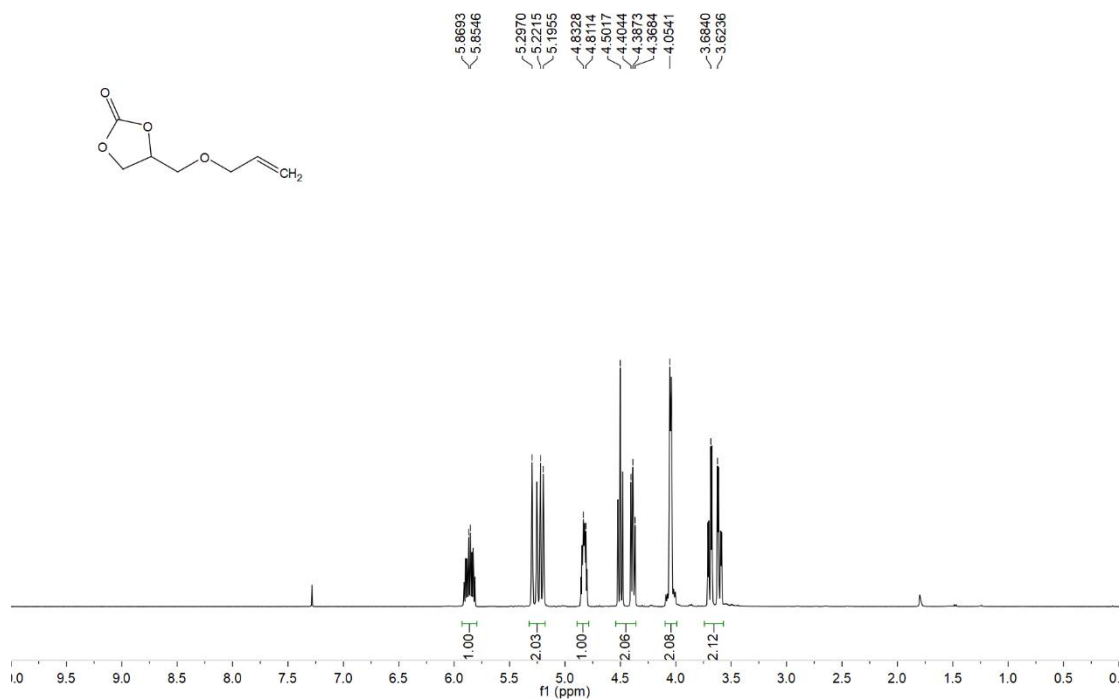


Fig. S24 ¹H NMR spectrum of **7l** in CDCl₃

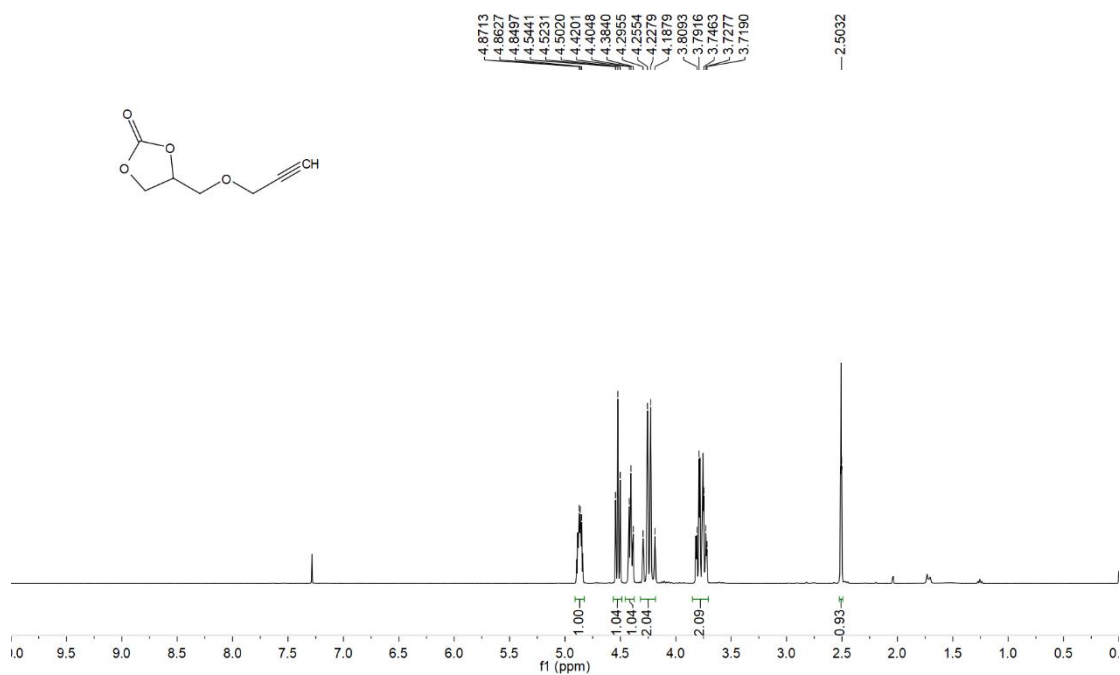


Fig. S25 ¹H NMR spectrum of **7m** in CDCl₃

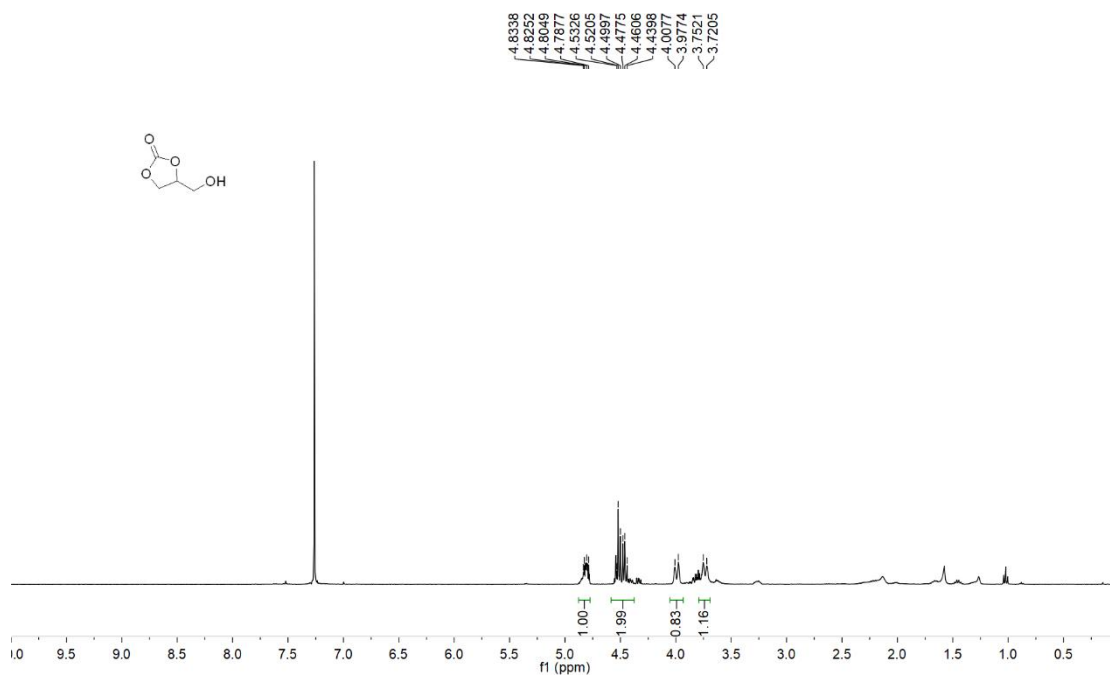


Fig. S26 ¹H NMR spectrum of **7n** in CDCl₃

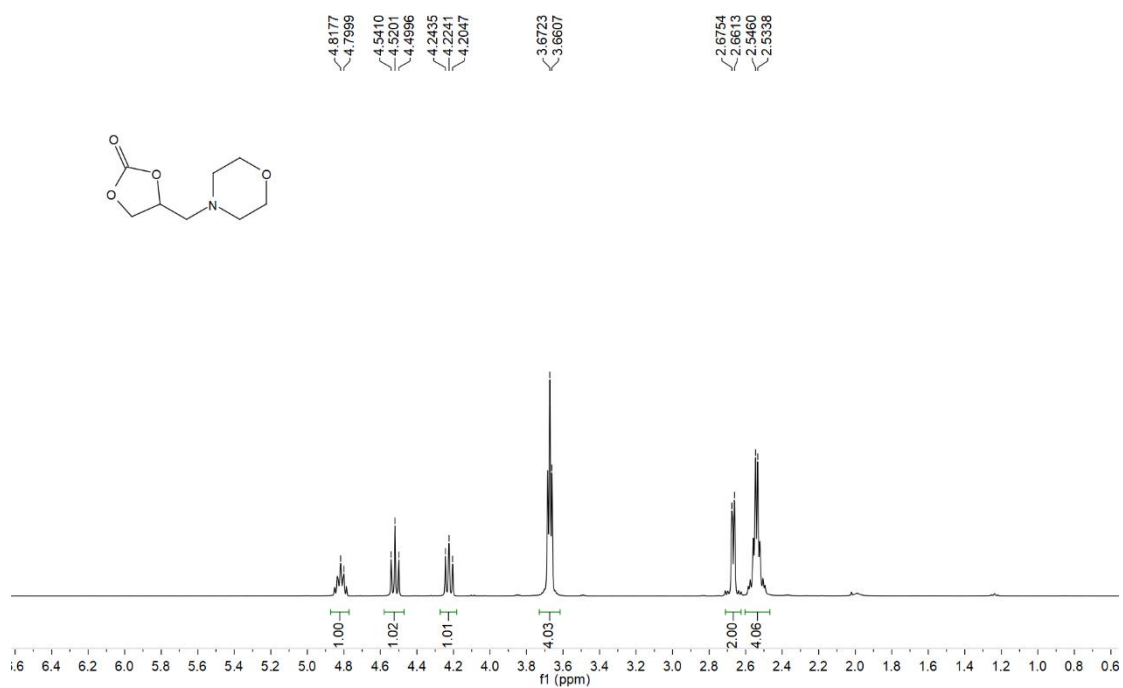


Fig. S27 ¹H NMR spectrum of **7o** in CDCl₃

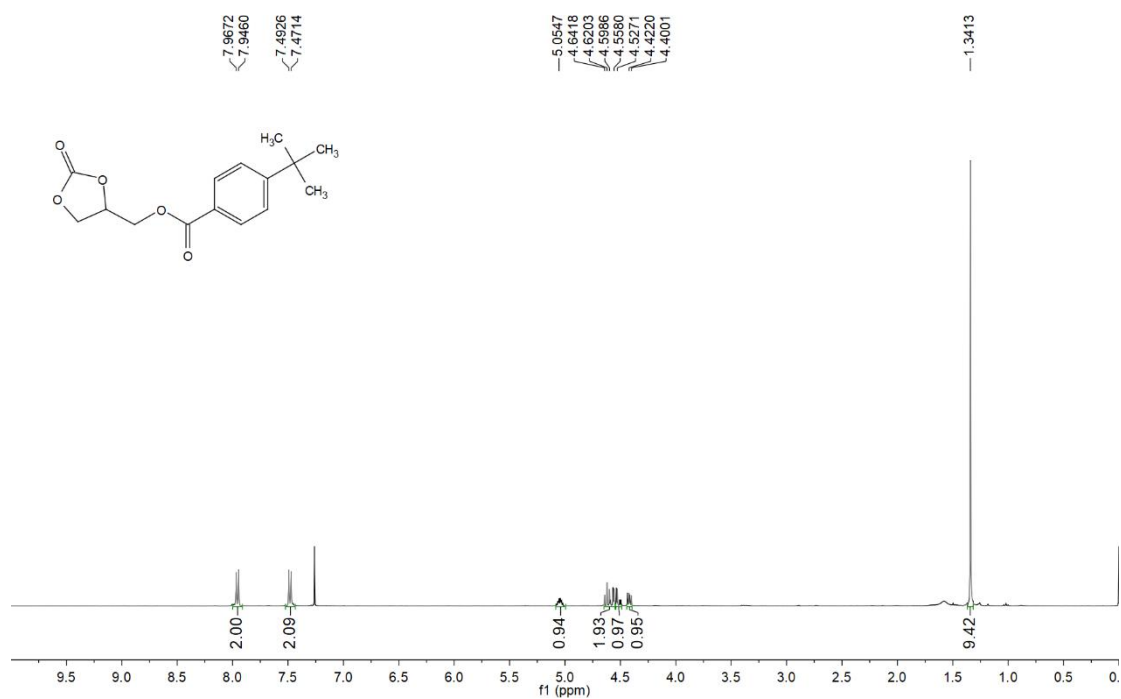


Fig. S28 ¹H NMR spectrum of **7p** in CDCl₃

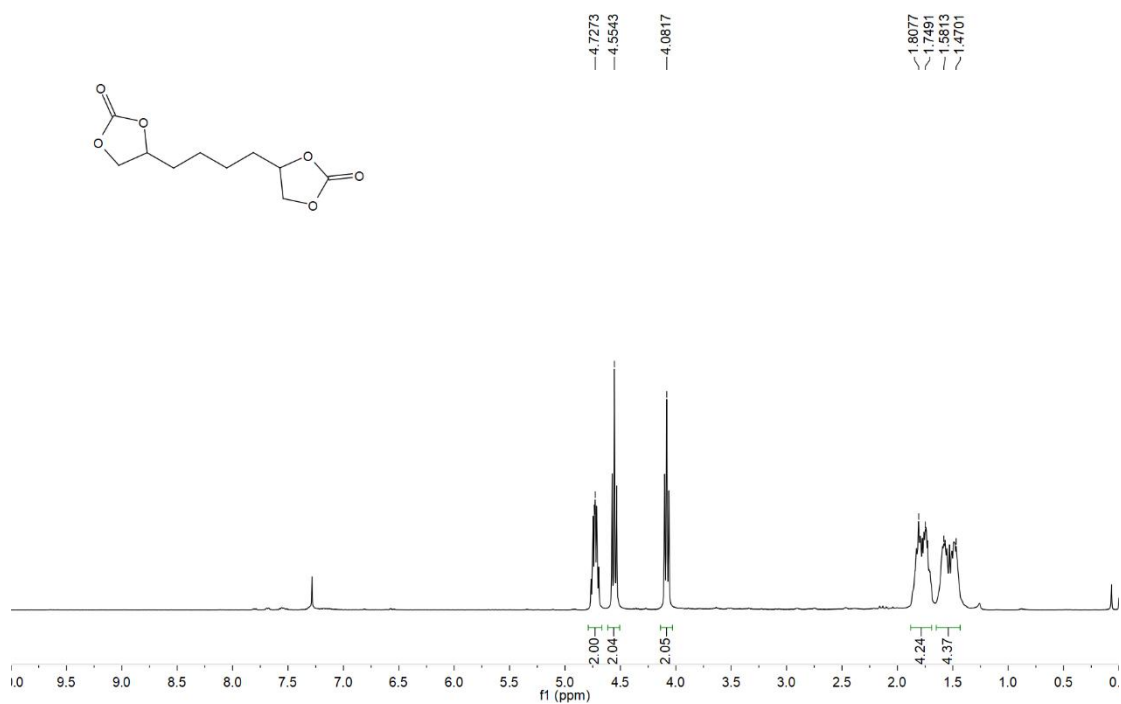


Fig. S29 ¹H NMR spectrum of **7q** in CDCl₃

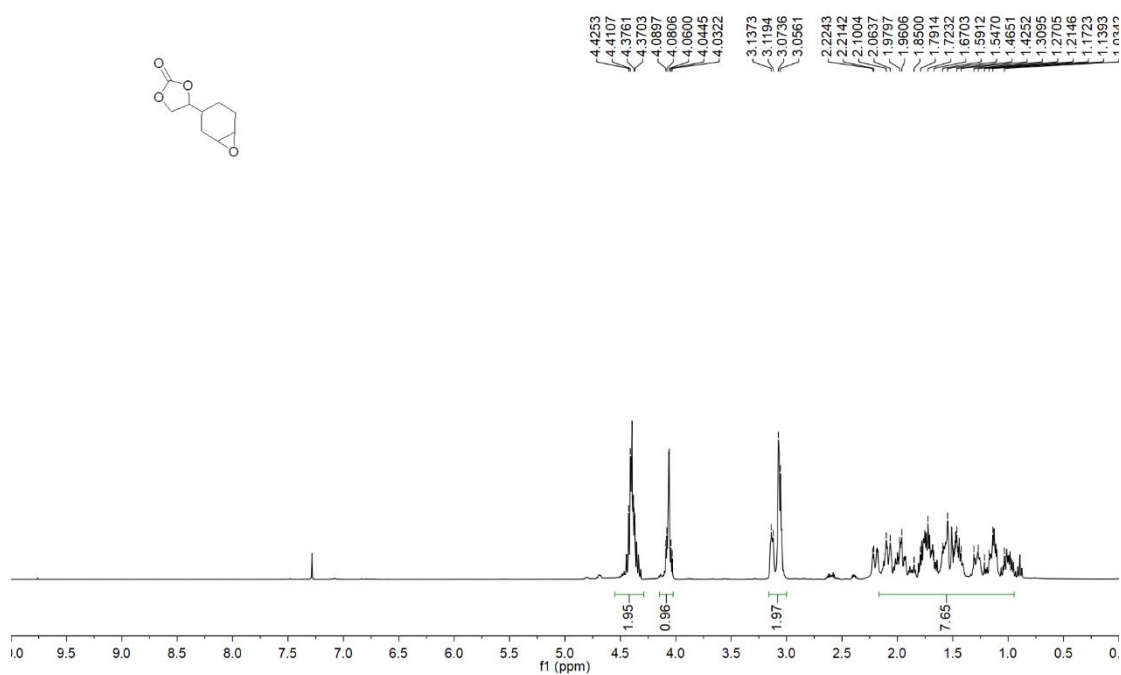


Fig. S30 ^1H NMR spectrum of **7r** in CDCl_3

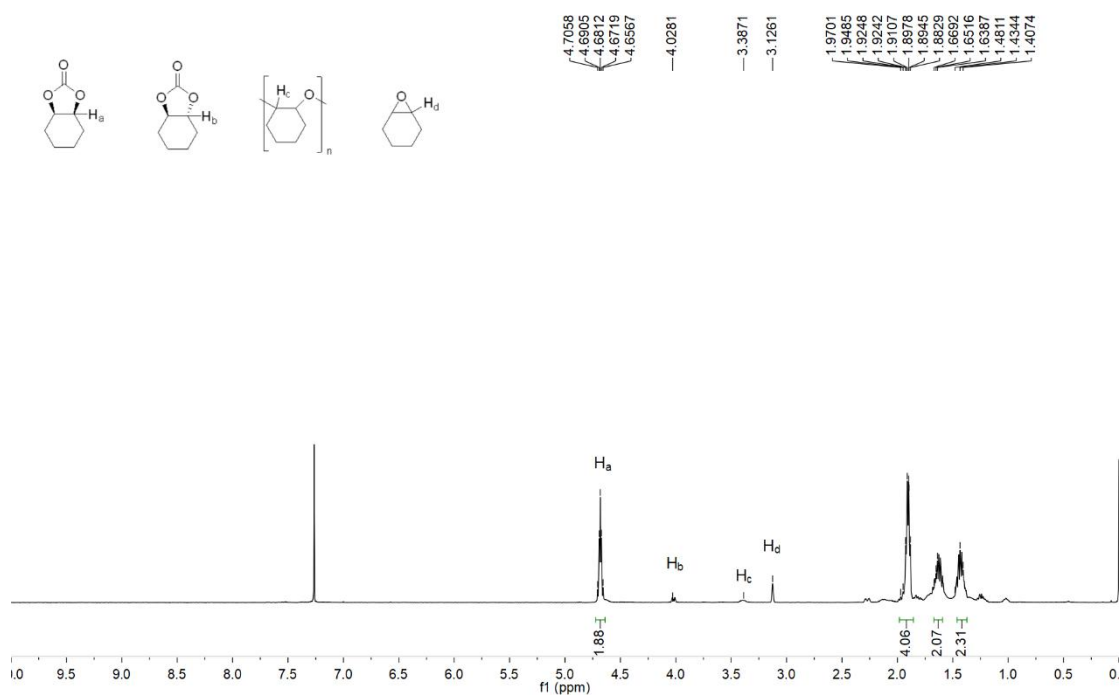


Fig. S31 ^1H NMR spectrum of **cis-9a** in CDCl_3

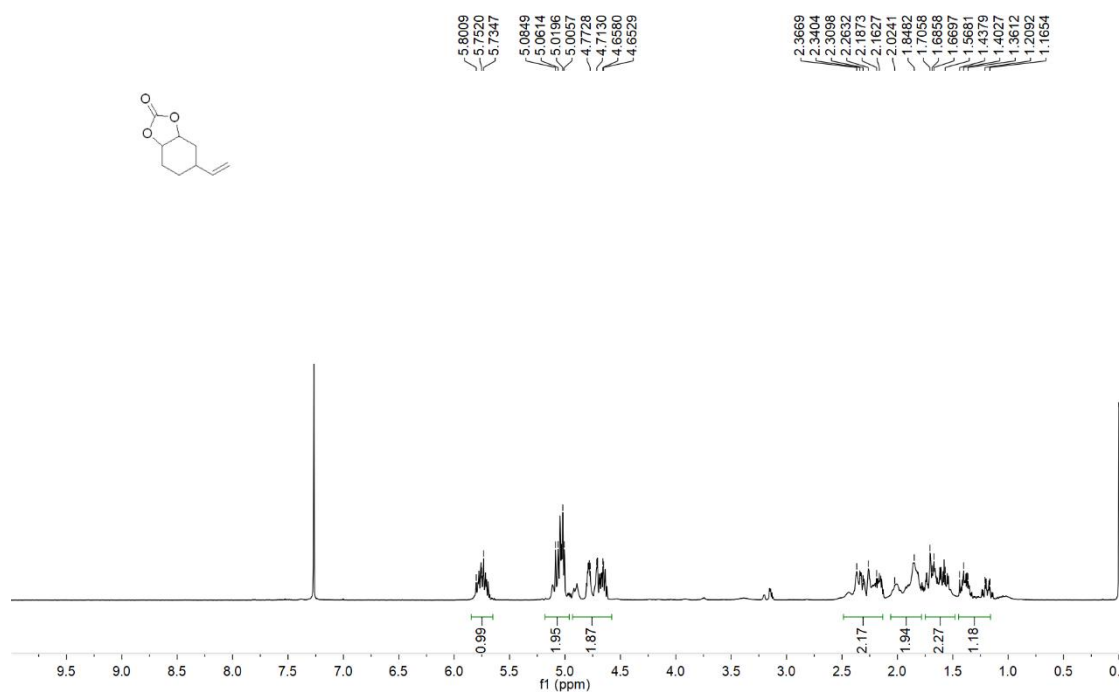


Fig. S32 ¹H NMR spectrum of **9b** in CDCl₃

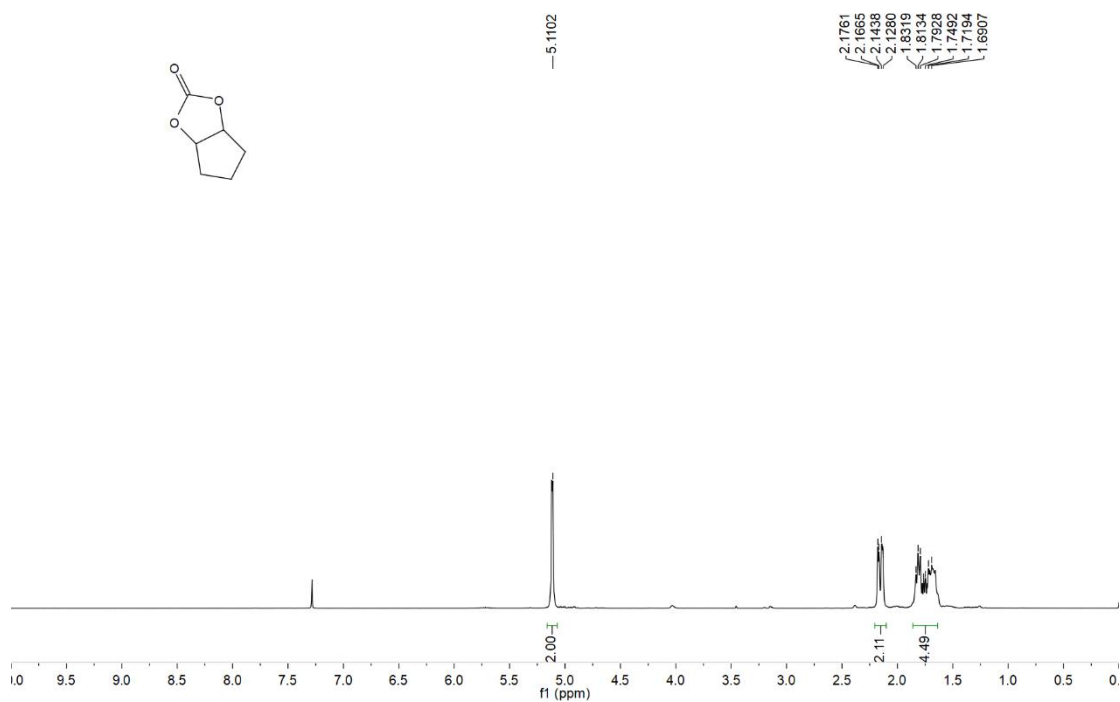


Fig. S33 ¹H NMR spectrum of **9c** in CDCl₃

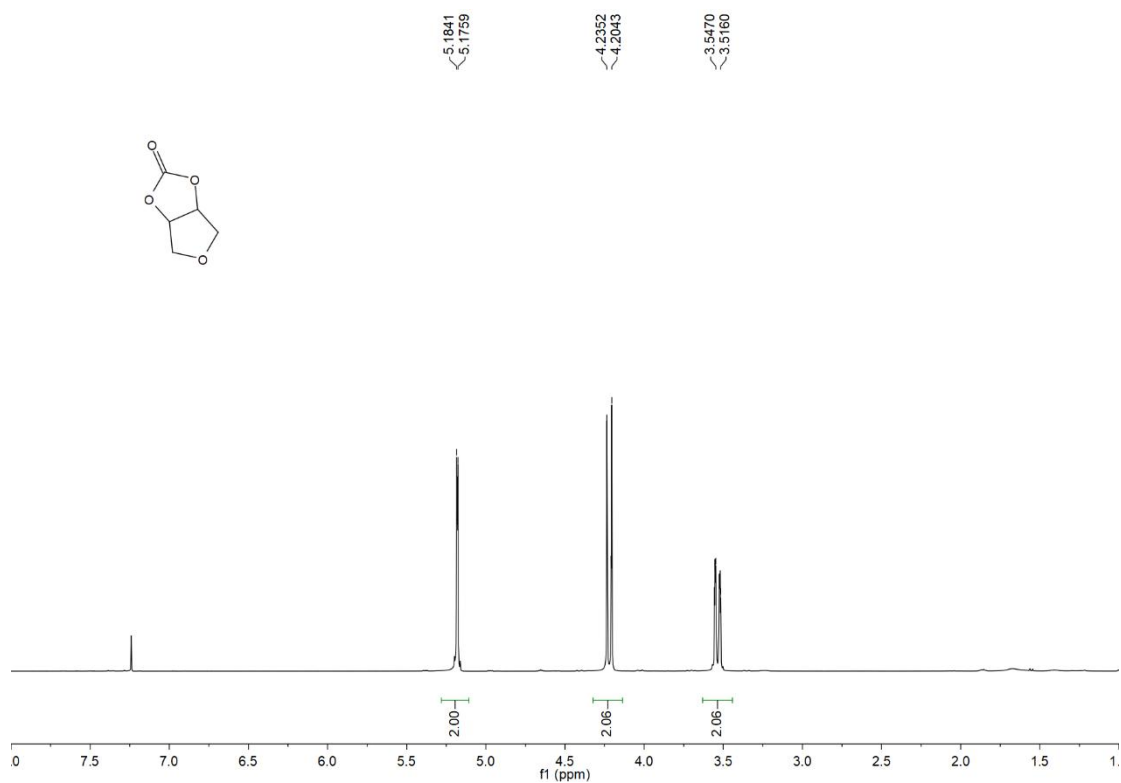


Fig. S34 ¹H NMR spectrum of **9d** in CDCl₃

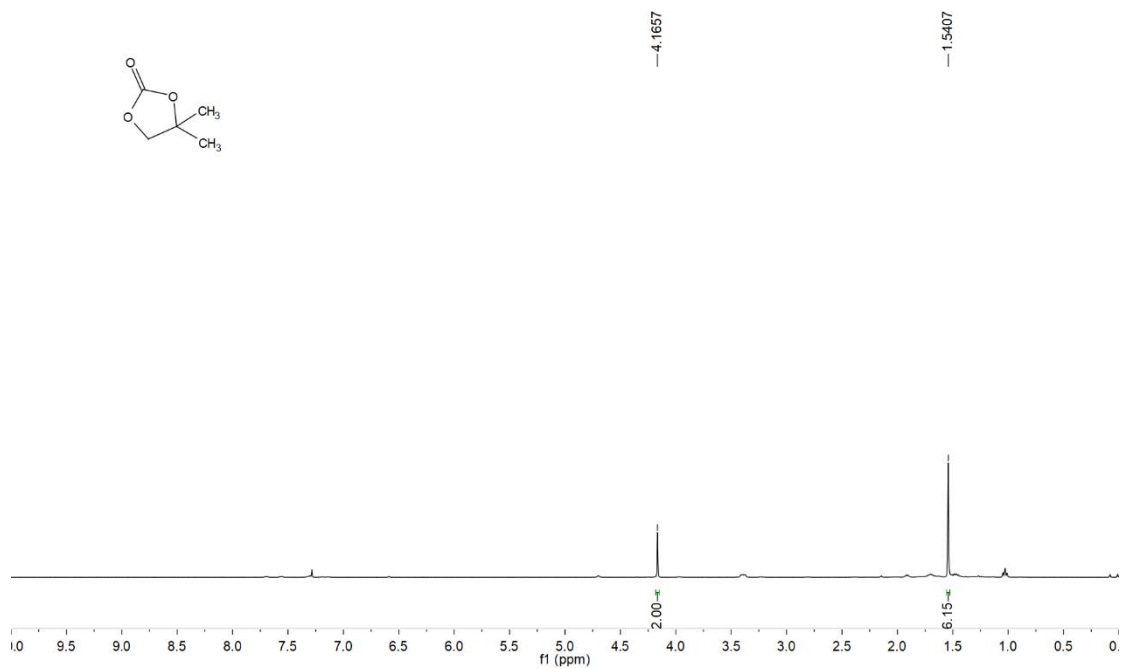


Fig. S35 ¹H NMR spectrum of **9e** in CDCl₃

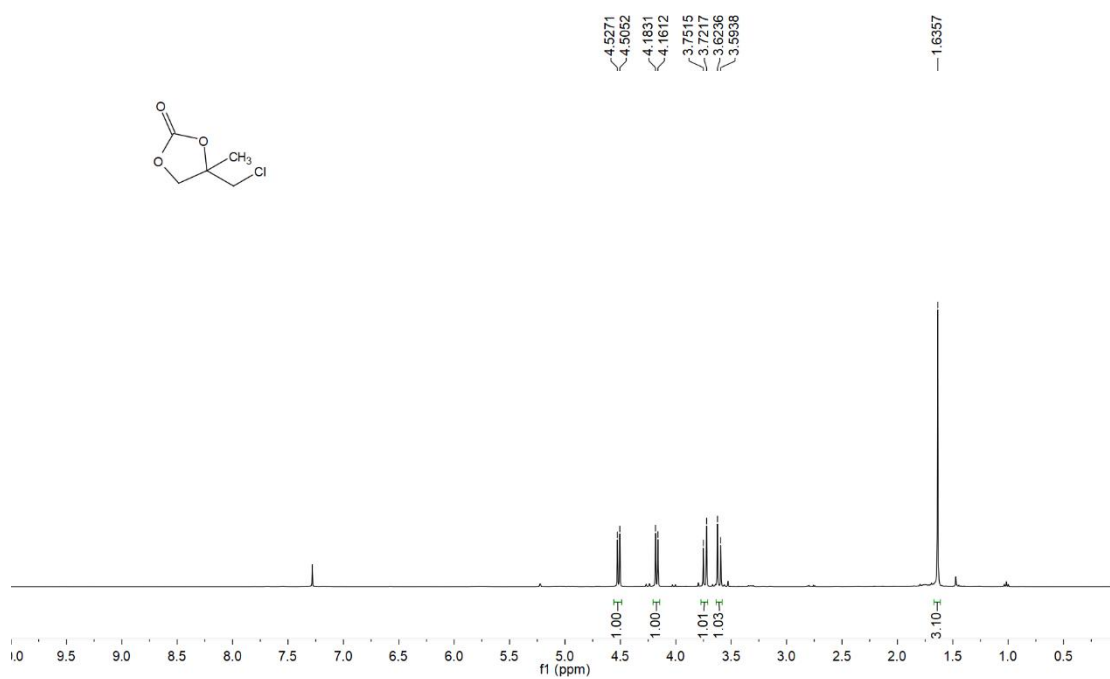


Fig. S36 ¹H NMR spectrum of **9f** in CDCl₃